

## Supplementary Information

### A DFT study on the stability and optoelectronic property of Pb/Sn/Ge-based MA<sub>2</sub>B(SCN)<sub>2</sub>I<sub>2</sub> perovskites

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Table S1. Relative energies for the hybrid structures of MA<sub>2</sub>Pb<sub>1-x</sub>Sn<sub>x</sub>(SCN)<sub>2</sub>I<sub>2</sub> (x = 0.25, 0.50, 0.75) with different doping sites.

System	Doping site	Relative energies (eV)
x = 0.25	1	0
	2	0.015
	3	0.011
	4	0.027
x = 0.50	1	0
	2	0.048
	3	0.006
	4	0.040
	5	0.037
	6	0.031
x = 0.75	1	0
	2	0.010
	3	0.026

Table S2. Atomic coordinates of H, C, N, S, I and Pb elements in MA<sub>2</sub>Pb(SCN)<sub>2</sub>I<sub>2</sub> structure.

			x	y	z	Wyckoff Position
<b>1</b>	H	H1	0.299	0.543	0.702	4b
<b>2</b>	H	H2	0.320	0.469	0.539	4b
<b>3</b>	H	H3	0.371	0.546	0.620	4b
<b>4</b>	H	H4	0.376	0.464	0.923	4b
<b>5</b>	H	H5	0.393	0.381	0.747	4b
<b>6</b>	H	H6	0.316	0.385	0.848	4b
<b>7</b>	H	H7	0.201	0.043	0.202	4b
<b>8</b>	H	H8	0.180	0.031	0.039	4b

<b>9</b>	H	H9	0.129	0.056	0.120	4b
<b>10</b>	H	H10	0.124	0.036	0.423	4b
<b>11</b>	H	H11	0.107	0.119	0.247	4b
<b>12</b>	H	H12	0.184	0.115	0.348	4b
<b>13</b>	H	H13	0.799	0.043	0.202	4b
<b>14</b>	H	H14	0.820	0.031	0.039	4b
<b>15</b>	H	H15	0.871	0.046	0.120	4b
<b>16</b>	H	H16	0.876	0.036	0.423	4b
<b>17</b>	H	H17	0.893	0.119	0.247	4b
<b>18</b>	H	H18	0.816	0.115	0.348	4b
<b>19</b>	H	H19	0.701	0.543	0.702	4b
<b>20</b>	H	H20	0.680	0.469	0.539	4b
<b>21</b>	H	H21	0.629	0.546	0.620	4b
<b>22</b>	H	H22	0.624	0.464	0.923	4b
<b>23</b>	H	H23	0.607	0.381	0.747	4b
<b>24</b>	H	H24	0.684	0.385	0.848	4b
<b>25</b>	H	H25	0.299	1.043	0.702	4b
<b>26</b>	H	H26	0.320	0.969	0.539	4b
<b>27</b>	H	H27	0.371	1.046	0.620	4b
<b>28</b>	H	H28	0.376	0.964	0.923	4b
<b>29</b>	H	H29	0.393	0.881	0.747	4b
<b>30</b>	H	H30	0.316	0.885	0.848	4b
<b>31</b>	H	H31	0.201	0.457	0.202	4b
<b>32</b>	H	H32	0.180	0.531	0.039	4b
<b>33</b>	H	H33	0.129	0.454	0.120	4b
<b>34</b>	H	H34	0.124	0.536	0.423	4b
<b>35</b>	H	H35	0.107	0.619	0.247	4b
<b>36</b>	H	H36	0.184	0.615	0.348	4b
<b>37</b>	H	H37	0.799	0.457	0.202	4b

<b>38</b>	H	H38	0.820	0.531	0.039	4b
<b>39</b>	H	H39	0.871	0.454	0.120	4b
<b>40</b>	H	H40	0.876	0.536	0.423	4b
<b>41</b>	H	H41	0.893	0.619	0.247	4b
<b>42</b>	H	H42	0.816	0.615	0.348	4b
<b>43</b>	H	H43	0.701	1.043	0.702	4b
<b>44</b>	H	H44	0.680	0.969	0.539	4b
<b>45</b>	H	H45	0.629	1.046	0.620	4b
<b>46</b>	H	H46	0.624	0.964	0.923	4b
<b>47</b>	H	H47	0.607	0.881	0.747	4b
<b>48</b>	H	H48	0.684	0.885	0.848	4b
<b>49</b>	C	C1	0.311	0.281	0.268	4b
<b>50</b>	C	C2	0.357	0.427	0.805	4b
<b>51</b>	C	C3	0.190	0.219	0.768	4b
<b>52</b>	C	C4	0.143	0.073	0.305	4b
<b>53</b>	C	C5	0.190	0.219	0.768	4b
<b>54</b>	C	C6	0.857	0.073	0.305	4b
<b>55</b>	C	C7	0.689	0.281	0.268	4b
<b>56</b>	C	C8	0.643	0.427	0.805	4b
<b>57</b>	C	C9	0.311	0.781	0.268	4b
<b>58</b>	C	C10	0.357	0.927	0.805	4b
<b>59</b>	C	C11	0.190	0.719	0.768	4b
<b>60</b>	C	C12	0.143	0.573	0.305	4b
<b>61</b>	C	C13	0.190	0.719	0.768	4b
<b>62</b>	C	C14	0.857	0.573	0.305	4b
<b>63</b>	C	C15	0.689	0.781	0.268	4b
<b>64</b>	C	C16	0.643	0.927	0.805	4b
<b>65</b>	N	N1	0.287	0.366	0.297	4b
<b>66</b>	N	N2	0.334	0.503	0.652	4b

<b>67</b>	N	N3	0.213	0.134	0.797	4b
<b>68</b>	N	N4	0.166	0.003	0.152	4b
<b>69</b>	N	N5	0.213	0.134	0.797	4b
<b>70</b>	N	N6	0.834	0.003	0.152	4b
<b>71</b>	N	N7	0.713	0.366	0.297	4b
<b>72</b>	N	N8	0.666	0.503	0.652	4b
<b>73</b>	N	N9	0.287	0.866	0.297	4b
<b>74</b>	N	N10	0.334	1.003	0.652	4b
<b>75</b>	N	N11	0.213	0.634	0.797	4b
<b>76</b>	N	N12	0.166	0.497	0.152	4b
<b>77</b>	N	N13	0.213	0.634	0.797	4b
<b>78</b>	N	N14	0.834	0.497	0.152	4b
<b>79</b>	N	N15	0.713	0.866	0.297	4b
<b>80</b>	N	N16	0.666	1.003	0.652	4b
<b>81</b>	S	S1	0.342	0.164	0.229	4b
<b>82</b>	S	S2	0.158	0.336	0.729	4b
<b>83</b>	S	S3	0.158	0.336	0.729	4b
<b>84</b>	S	S4	0.658	0.164	0.229	4b
<b>85</b>	S	S5	0.342	0.664	0.229	4b
<b>86</b>	S	S6	0.158	0.836	0.729	4b
<b>87</b>	S	S7	0.158	0.836	0.729	4b
<b>88</b>	S	S8	0.658	0.664	0.229	4b
<b>89</b>	I	I1	0.500	0.446	0.277	2a
<b>90</b>	I	I2	0.500	0.181	0.742	2a
<b>91</b>	I	I3	0.000	0.054	0.777	2a
<b>92</b>	I	I4	0.000	0.319	0.242	2a
<b>93</b>	I	I5	0.500	0.946	0.277	2a
<b>94</b>	I	I6	0.500	0.681	0.742	2a
<b>95</b>	I	I7	0.000	0.554	0.777	2a

<b>96</b>	I	I8	0.000	0.819	0.242	2a
<b>97</b>	Pb	Pb1	0.500	0.196	0.234	2a
<b>98</b>	Pb	Pb2	0.000	0.304	0.734	2a
<b>99</b>	Pb	Pb3	0.500	0.696	0.234	2a
<b>100</b>	Pb	Pb4	0.000	0.804	0.734	2a

Table S3. Average interatomic distances B-I and B-S (in Å) for MA<sub>2</sub>B(SCN)<sub>2</sub>I<sub>2</sub> (B = Pb, Sn, Ge) and MA<sub>2</sub>Pb<sub>1-x</sub>Sn<sub>x</sub>(SCN)<sub>2</sub>I<sub>2</sub> (x = 0.25, 0.50, 0.75) perovskites.

system	Apical			Equatorial			B-S		
	Pb-I	Sn-I	Ge-I	Pb-I	Sn-I	Ge-I	Pb-S	Sn-S	Ge-S
B = Pb	3.204			3.321			3.019		
B = Sn		3.173			3.322			2.947	
B = Ge			3.123			3.315			2.785
x = 0.25	3.204	3.165		3.338	3.336		3.010	2.954	
x = 0.50	3.202	3.199		3.349	3.343		3.012	2.953	
x = 0.75	3.216	3.175		3.349	3.343		3.012	2.959	

Table S4. Calculated bond angles of B-I-B (°) along the x-y plane for MA<sub>2</sub>B(SCN)<sub>2</sub>I<sub>2</sub> (B = Pb, Sn, Ge) and MA<sub>2</sub>Pb<sub>1-x</sub>Sn<sub>x</sub>(SCN)<sub>2</sub>I<sub>2</sub> (x = 0.25, 0.50, 0.75) perovskites.

system	Pb-I-Pb	Sn-I-Sn	Ge-I-Ge
B = Pb	164.710		
B = Sn		163.431	
B = Ge			159.498
x = 0.25	164.263		
x = 0.50	163.855	164.667	
x = 0.75		164.036	

Table S5. Total energy of decomposed fragments after optimized for MA<sub>2</sub>B(SCN)<sub>2</sub>I<sub>2</sub> (B = Pb, Sn, Ge) and MA<sub>2</sub>Pb<sub>1-x</sub>Sn<sub>x</sub>(SCN)<sub>2</sub>I<sub>2</sub> (x = 0.25, 0.50, 0.75).

System	Com.	E(eV)	Com.	E(eV)	Com.	E(eV)
B = Pb	2MAI	-84.918	Pb(SCN) <sub>2</sub>	-46.265	-	-
B = Sn	2MAI	-84.918	Sn(SCN) <sub>2</sub>	-46.374	-	-
B = Ge	2MAI	-84.918	Ge(SCN) <sub>2</sub>	-45.082	-	-

$x = 0.25$	2MAI	-84.918	$0.75\text{Pb}(\text{SCN})_2$	-34.699	$0.25\text{Sn}(\text{SCN})_2$	-11.594
$x = 0.50$	2MAI	-84.918	$0.50\text{Pb}(\text{SCN})_2$	-23.133	$0.50\text{Sn}(\text{SCN})_2$	-23.187
$x = 0.75$	2MAI	-84.918	$0.25\text{Pb}(\text{SCN})_2$	-11.566	$0.75\text{Sn}(\text{SCN})_2$	-34.781

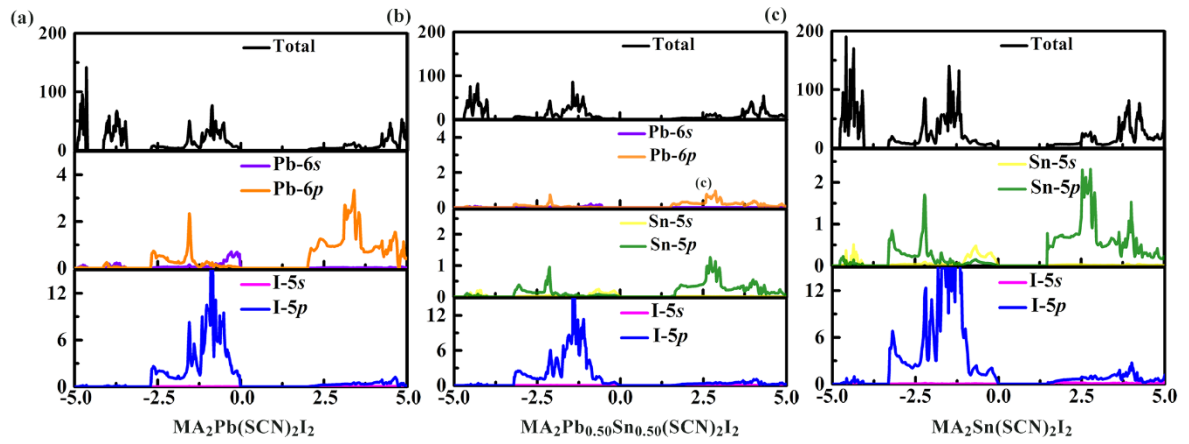


Figure S1. Respective PDOS of Pb, Sn and I atoms for (a)  $\text{MA}_2\text{Pb}(\text{SCN})_2\text{I}_2$ , (b)  $\text{MA}_2\text{Pb}_{0.50}\text{Sn}_{0.50}(\text{SCN})_2\text{I}_2$ , and (c)  $\text{MA}_2\text{Sn}(\text{SCN})_2\text{I}_2$  compounds.