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## Supporting Information Unveiling the reconstructed active phase of Ni<sub>3</sub>Se<sub>2</sub> model for water splitting

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Figure S1. The corresponding electrodeposition chronoamperometric curve for preparing the sample of  $Ni_3Se_2/FTO$ .



Figure S2. SEM images and EDS profiles of pristine  $Ni_3Se_2/FTO$ .



**Figure S3.** (a) Tafel plots derived from HER LSV curves of Ni<sub>3</sub>Se<sub>2</sub>/FTO and blank FTO substrate in 1.0 M KOH. (b) Scan rate dependence of the current densities ( $\Delta j = j_a - j_c$ ) for Ni<sub>3</sub>Se<sub>2</sub>/FTO at -0.65 V vs Ag/AgCl.



Figure S4. SEM images and EDS profiles of  $Ni_3Se_2$ /FTO after HER stability test.



**Figure S5.** (a) Tafel plots derived from OER LSV curves for the samples of Ni<sub>3</sub>Se<sub>2</sub>/FTO and blank FTO substrate in 1.0 M KOH. (b) Scan rate dependence of the current densities ( $\Delta j = j_a - j_c$ ) for Ni<sub>3</sub>Se<sub>2</sub>/FTO at 0.05 V vs Ag/AgCl.



Figure S6. SEM images and EDS profiles of  $Ni_3Se_2$ /FTO after OER stability test.



**Figure S7.** (a) Cyclic voltammetry (CV) curves toward OER for  $Ni_3Se_2/Ni$  foam and bare Ni foam. Considering the intense redox peak of Ni species, we show the CV curves rather than linear sweep voltammetry (LSV) curves. (b) LSV curves towards HER for  $Ni_3Se_2/Ni$  foam and bare Ni foam.