

## Supplementary Data

**An efficient photocatalytic system composed of  $\text{Ti}_3\text{C}_2$  quantum dots incorporated  $\text{TiO}_2$  nanosheets and  $\text{CuWO}_4$  nanoparticles: Fabrication and its photocatalytic activity for  $\text{H}_2$  production**

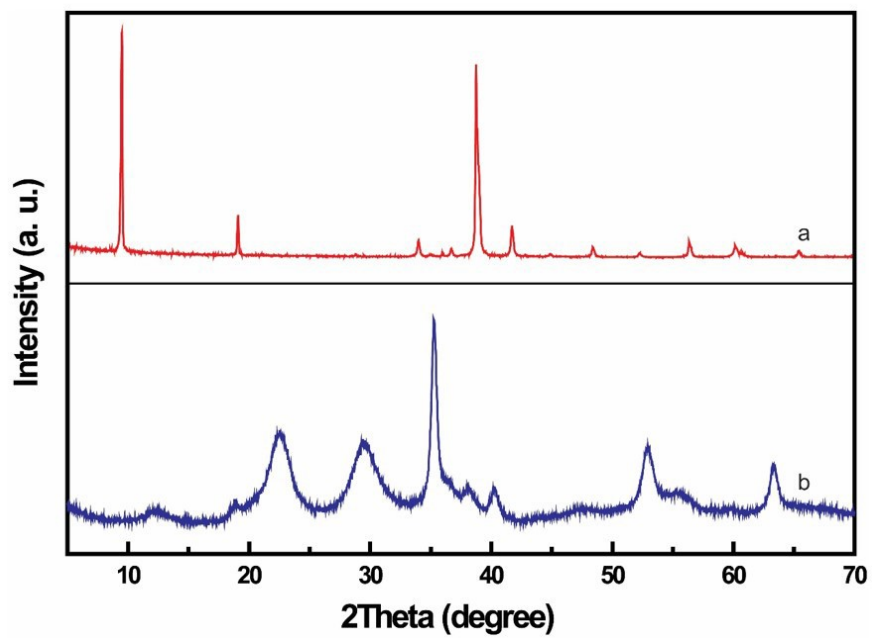
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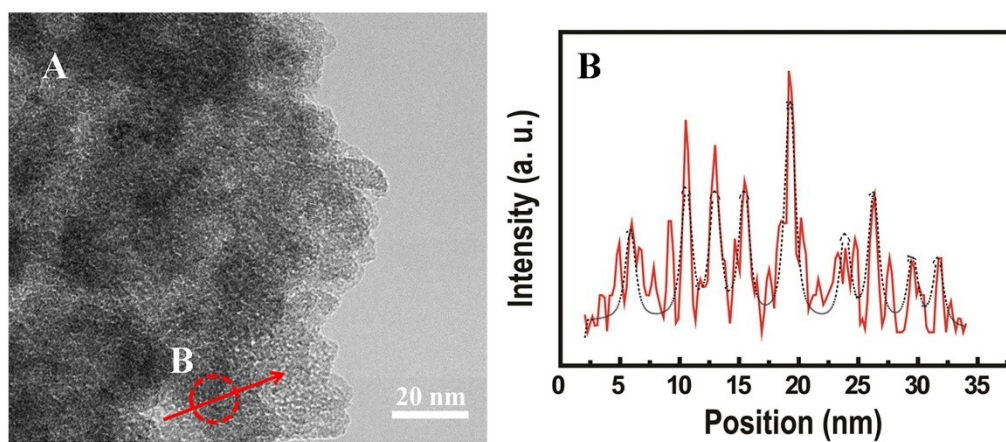
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Supplementary Fig. S1



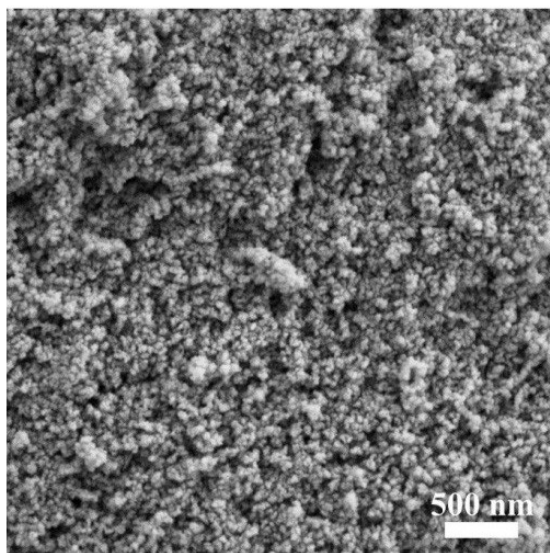
**Fig. S1.** XRD patterns of (a)  $\text{Ti}_3\text{AlC}_2$  and (b) the  $\text{CuWO}_4$  nanoparticles.

Supplementary Fig. S2



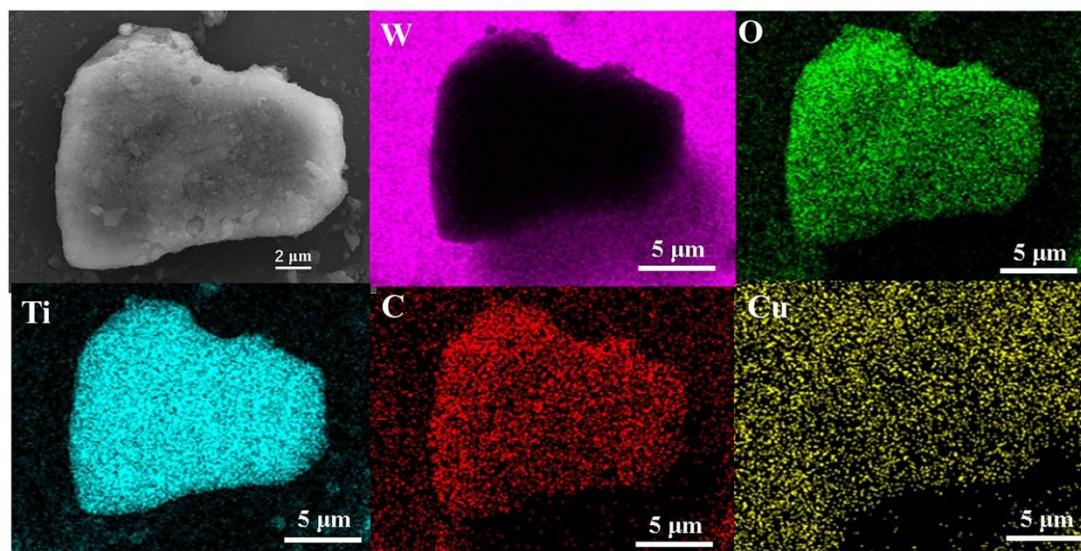
**Fig. S2.** (A) HRTEM image of 4% CuWO<sub>4</sub>-Ti<sub>3</sub>C<sub>2</sub>/TiO<sub>2</sub> and (B) C element TEM-EDS line profile along the red arrow shown in Fig. S2A (red) and curve-fitting analysis (black).

Supplementary Fig. S3



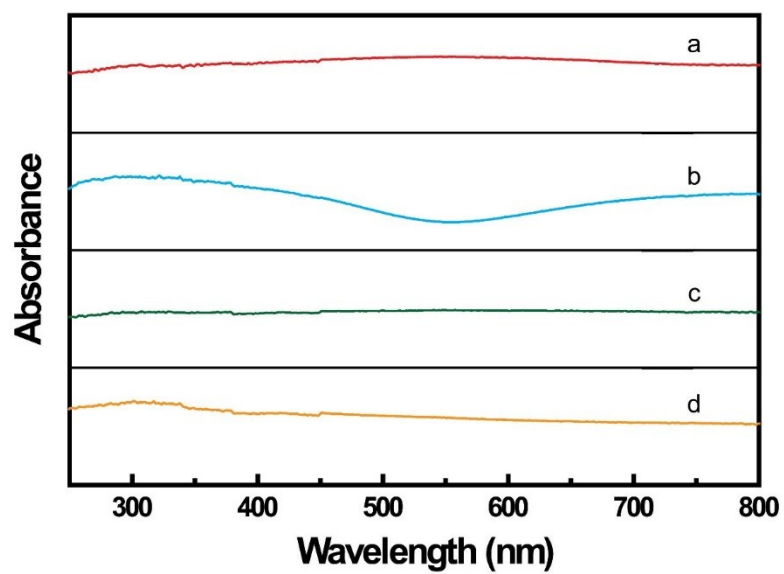
**Fig. S3.** SEM image of  $\text{CuWO}_4$  nanoparticles.

Supplementary Fig. S4



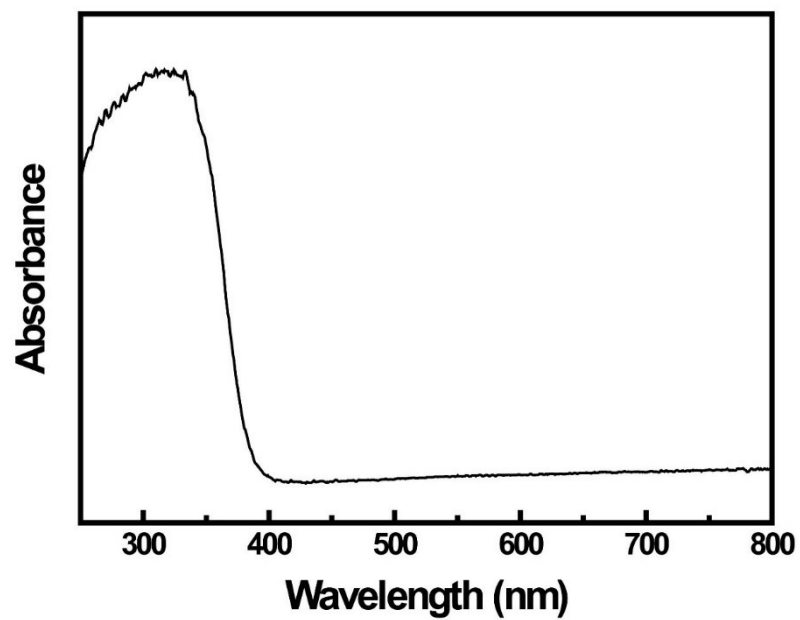
**Fig. S4.** Element mappings of 4%  $\text{CuWO}_4\text{-Ti}_3\text{C}_2/\text{TiO}_2$  (W, O, Ti, C and Cu).

Supplementary Fig. S5



**Fig. S5.** UV-vis spectra of (a) the  $\text{Ti}_3\text{C}_2$  nanosheets, (b) the  $\text{CuWO}_4$  nanoparticles, (c)  $\text{Ti}_3\text{C}_2/\text{TiO}_2$ , and (d) 4%  $\text{CuWO}_4\text{-Ti}_3\text{C}_2/\text{TiO}_2$ .

Supplementary Fig. S6



**Fig. S6.** UV-vis spectrum of P25.

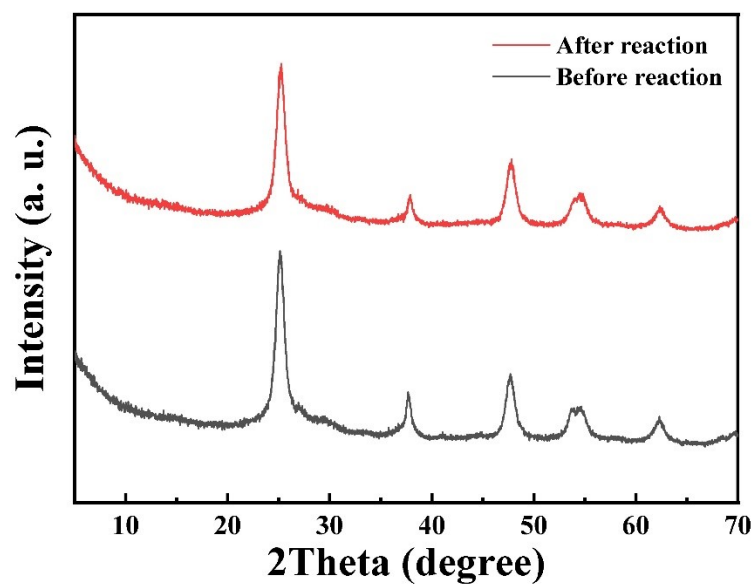
## Supplementary Table S1

**Table S1.** Photocatalytic activities of the photocatalysts reported previously.

<b>Photocatalyst</b>	<b>Rate of H<sub>2</sub> evolution (mmol g<sup>-1</sup> h<sup>-1</sup>)</b>	<b>Stable hydrogen production time (h)</b>	<b>Ref.</b>
CuWO <sub>4</sub> - Ti <sub>3</sub> C <sub>2</sub> /TiO <sub>2</sub>	3.65	14	This work
Pt-TiO <sub>2</sub>	0.16	5	[1]
Ti <sub>3</sub> C <sub>2</sub> -TiO <sub>2</sub> /Pt	1.60	12	[2]
Au/TiO <sub>2</sub>	0.36	4	[3]
Pd/TiO <sub>2</sub>	3.10	1	[4]
TiO <sub>2</sub> -Ti <sub>3</sub> C <sub>2</sub> /Ru	0.24	5	[5]
CuO/TiO <sub>2</sub>	2.00	3	[6]
Truxene/TiO <sub>2</sub>	21	5	[7]
MoS <sub>2</sub> /TiO <sub>2</sub>	1.38	4	[8]
CdS/Ni-MOF	2.51	3	[9]
CdS/CuS	0.30	4	[10]
Pt/IrO <sub>2</sub> /g-C <sub>3</sub> N <sub>4</sub>	2.47	2	[11]
WO <sub>3</sub> -MoS <sub>2</sub> -Pt	0.80	2	[12]
Zn-AgIn <sub>5</sub> S <sub>8</sub> /NiS	5.2	5	[13]

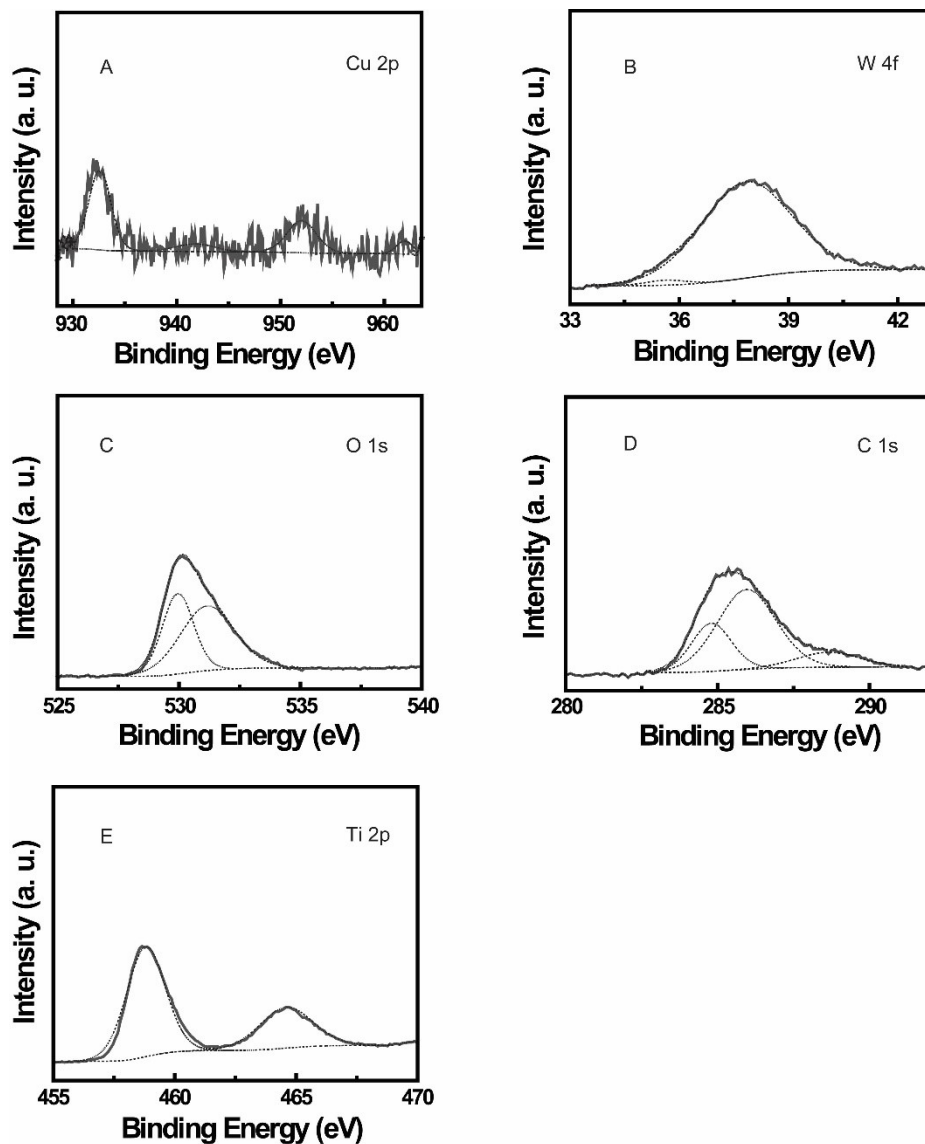


Supplementary Fig. S7



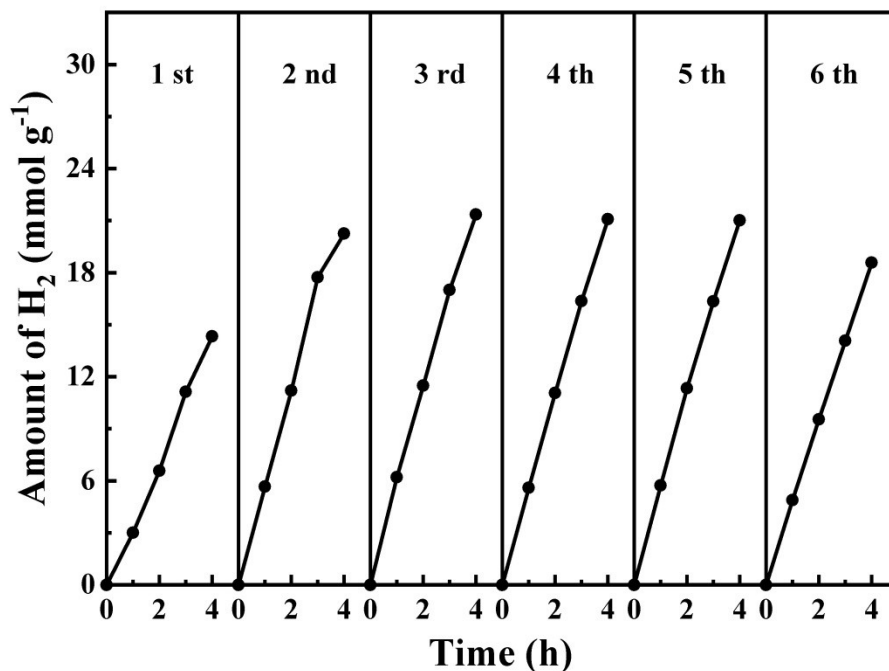
**Fig. S7.** XRD patterns of 4%  $\text{CuWO}_4\text{-Ti}_3\text{C}_2/\text{TiO}_2$  before and after photocatalytic reaction.

Supplementary Fig. S8



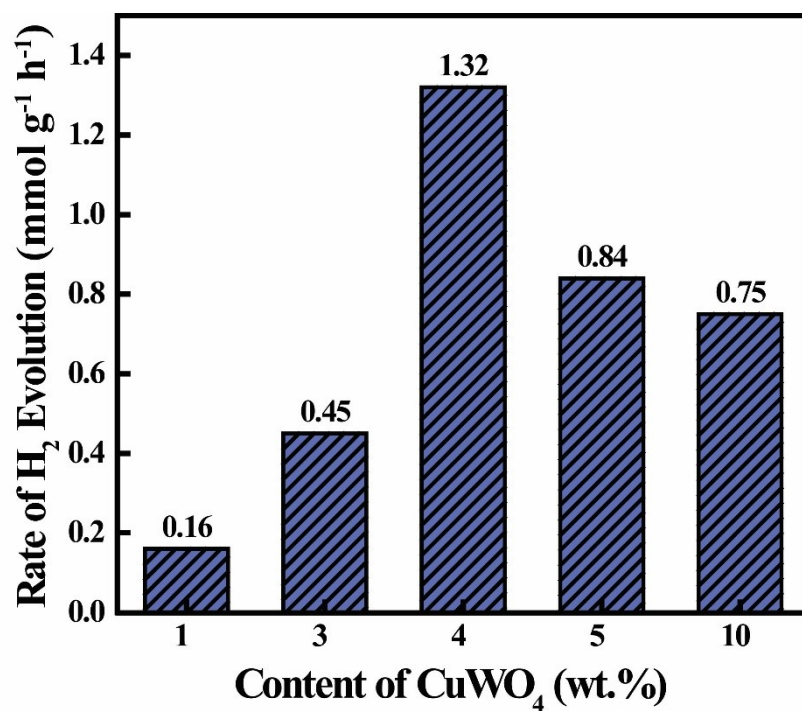
**Fig. S8.** XPS spectra of the used 4% CuWO<sub>4</sub>-Ti<sub>3</sub>C<sub>2</sub>/TiO<sub>2</sub>: (A) Cu 2p, (B) W 4f, (C) O 1s, (D) C 1s and (E) Ti 2p, high-resolution XPS spectra (solid) and curve-fitting analysis (dot line) of the states of Cu, W, O, C, and Ti.

Supplementary Fig. S9



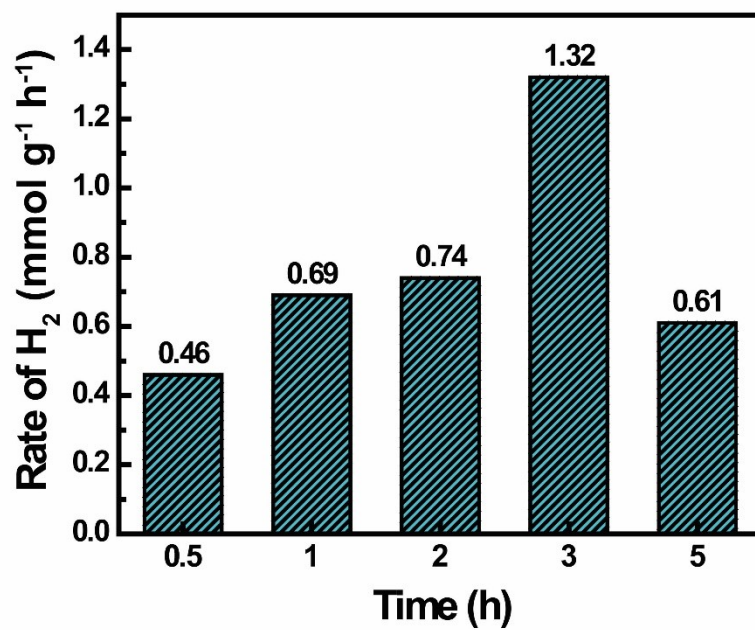
**Fig. S9.** Photostability of 4% CuWO<sub>4</sub>-Ti<sub>3</sub>C<sub>2</sub>/TiO<sub>2</sub> for photocatalytic hydrogen evolution (photocatalyst 10 mg; ethylene glycol aqueous solution 60 mL, 16.7vol.%; pH = 8; temperature 10°C; irradiation time between the two cycles 10 min).

Supplementary Fig. S10



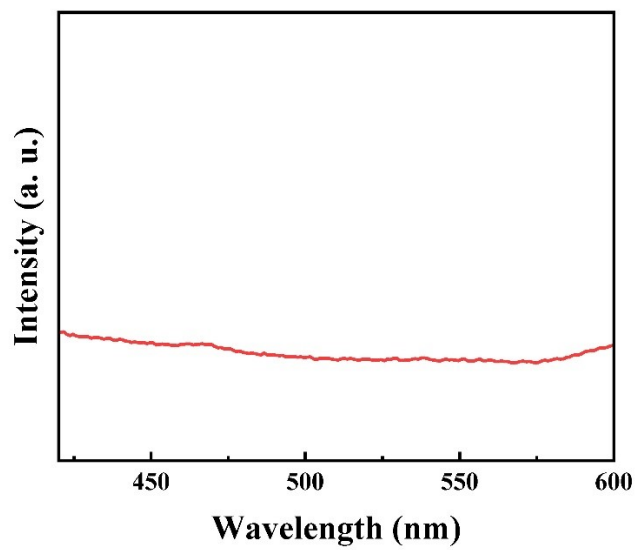
**Fig. S10.** Effect of the content of CuWO<sub>4</sub> nanoparticles on the photocatalytic activity of CuWO<sub>4</sub>-Ti<sub>3</sub>C<sub>2</sub>/TiO<sub>2</sub> (photocatalyst 10 mg; ethylene glycol aqueous solution 60 mL, 16.7vol.%; pH = 7; temperature 10°C; irradiation time 4 h).

Supplementary Fig. S11



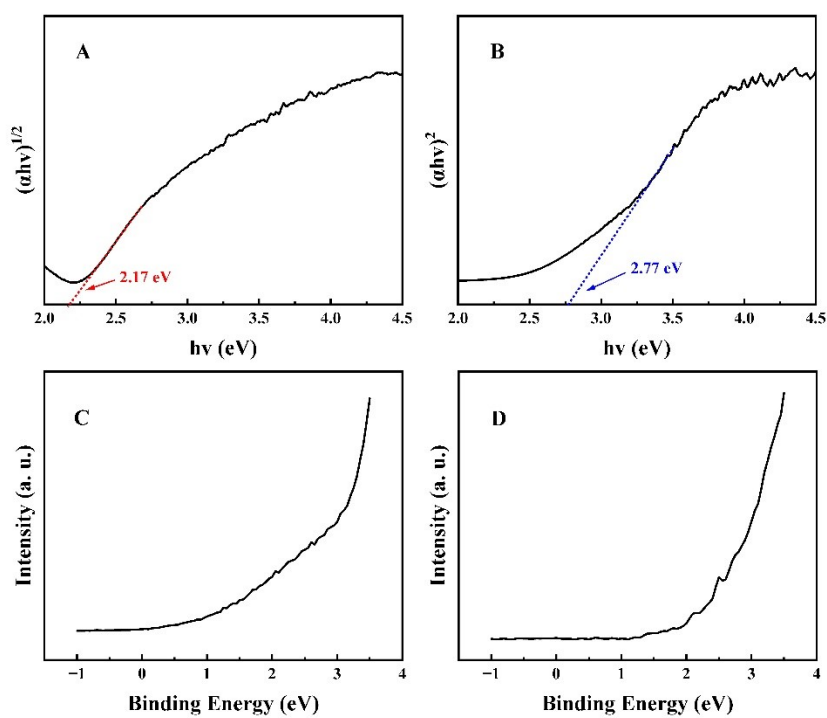
**Fig. S11.** Effect of oxidation time on the photocatalytic activity of 4% CuWO<sub>4</sub>-Ti<sub>3</sub>C<sub>2</sub>/TiO<sub>2</sub> (photocatalyst 10 mg; ethylene glycol aqueous solution 60 mL, 16.7vol.%; pH = 7; temperature 10°C; irradiation time 4 h).

Supplementary Fig. S12



**Fig. S12.** Fluorescence spectrum of the Ti<sub>3</sub>C<sub>2</sub> nanosheets (excitation wavelength 320 nm).

Supplementary Fig. S13



**Fig. S13.** Tauc plots of (A) the CuWO<sub>4</sub> nanoparticles and (B) the TiO<sub>2</sub> nanosheets, and UPS spectra of (C) the CuWO<sub>4</sub> nanoparticles and (D) the TiO<sub>2</sub> nanosheets.

## References

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