

A new aza-crown macrocyclic fluorescence chemosensor (N_3O_2 donor atoms) for
magnesium ions in aqueous ethanol solution

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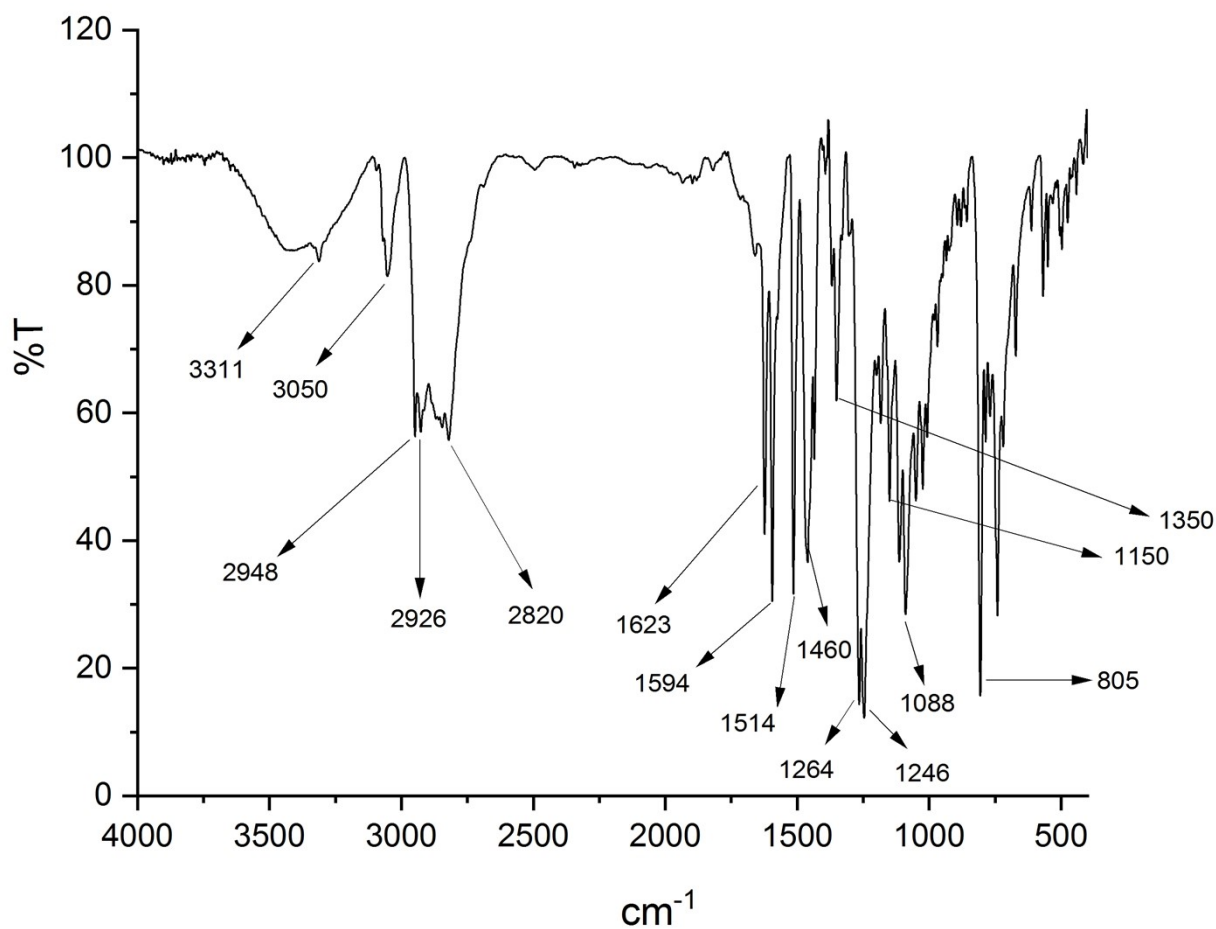


Fig. S1. IR spectrum of L

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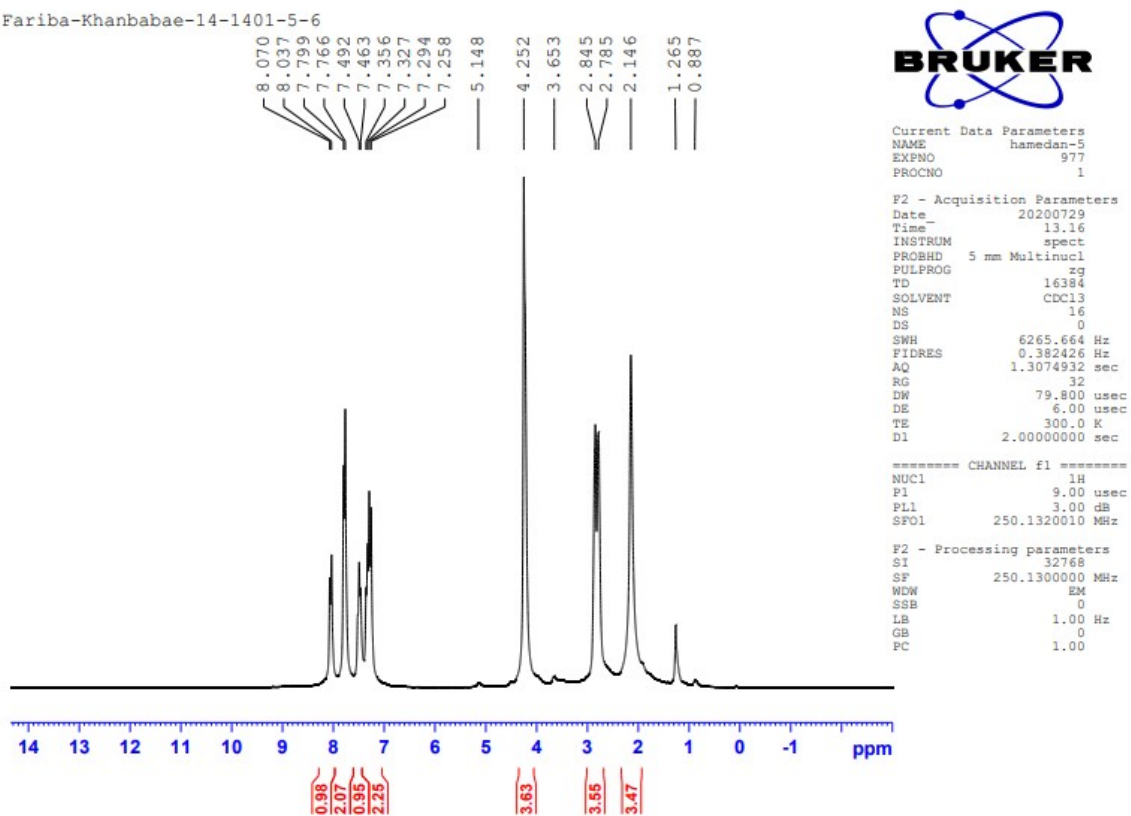


Fig. S2. ¹H NMR spectrum of L in CDCl₃.

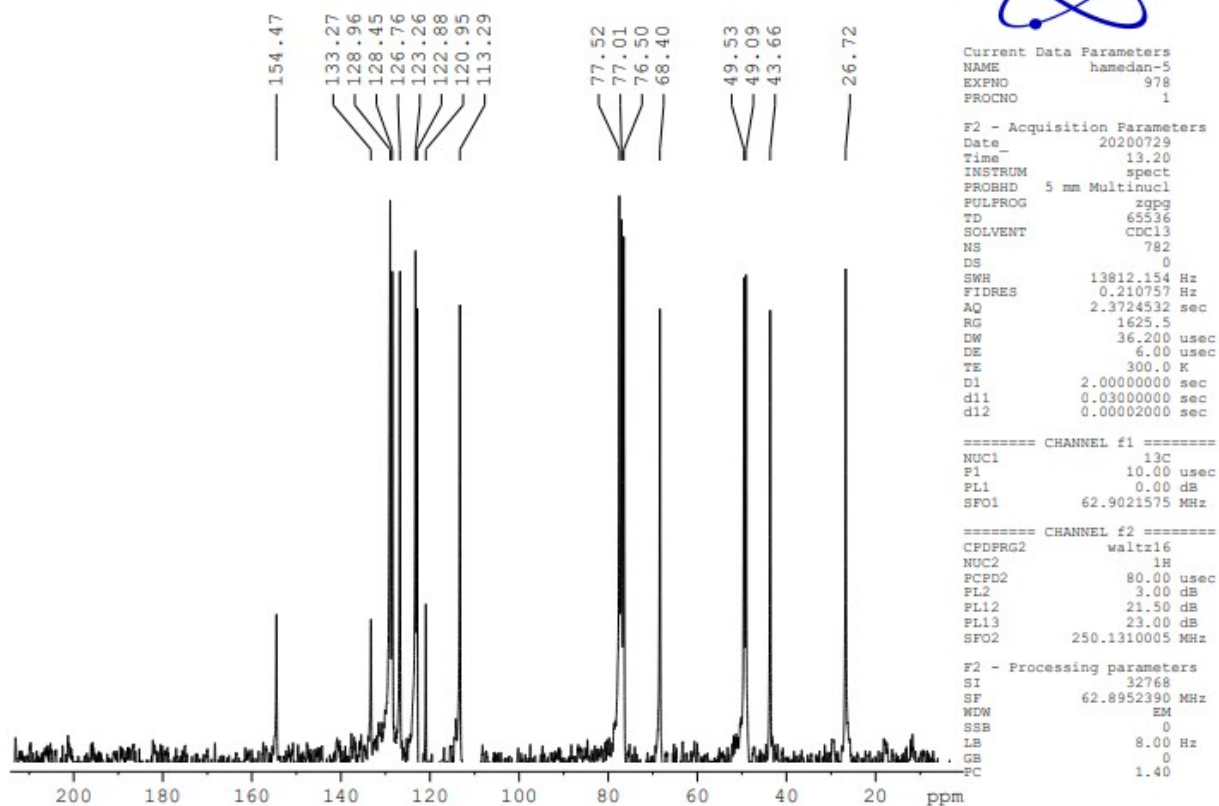


Fig. S3. ¹³CNMR spectrum of L in CDCl₃.

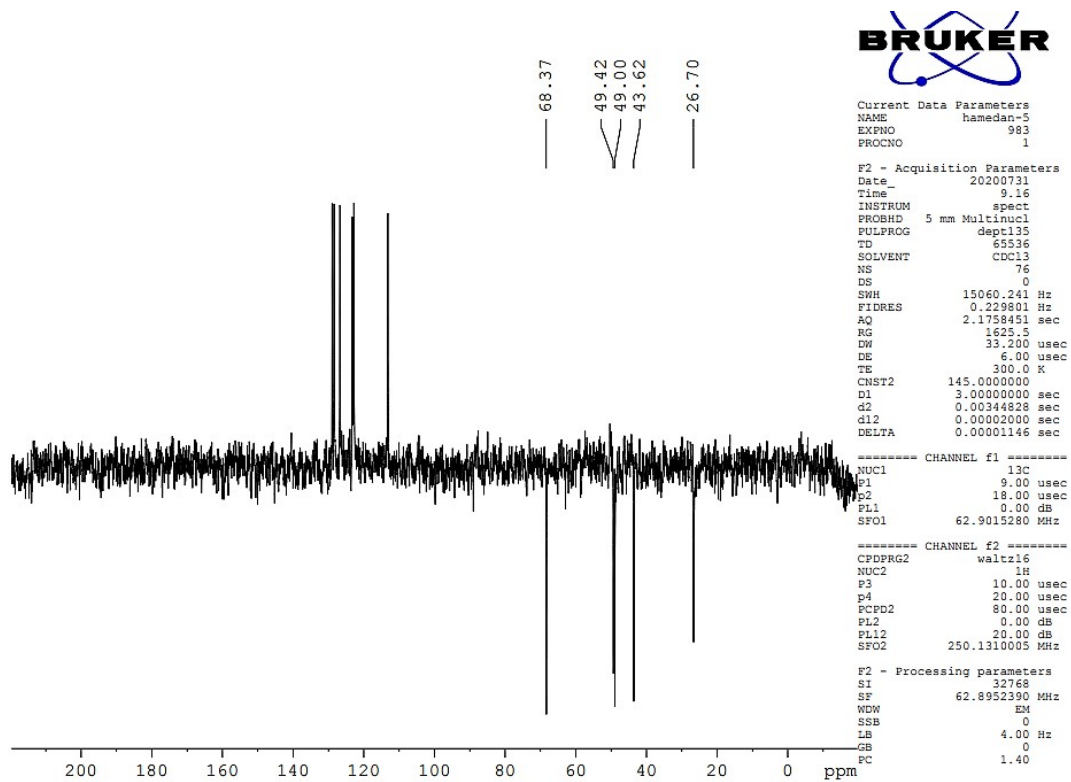


Fig. S4. DEPT spectrum of L in CDCl₃.

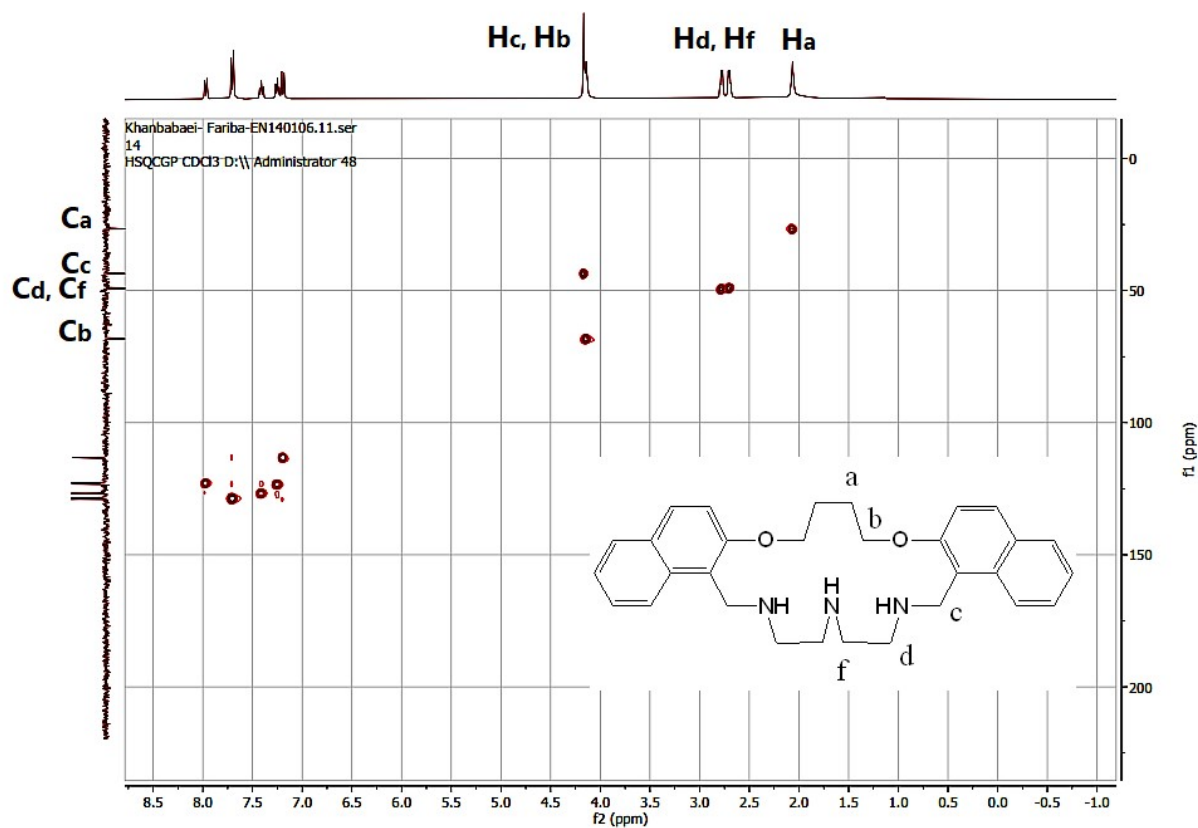


Fig. S5. HSQC NMR spectrum of L in CDCl₃ along with numbering scheme of the proton and carbon assignments for L.

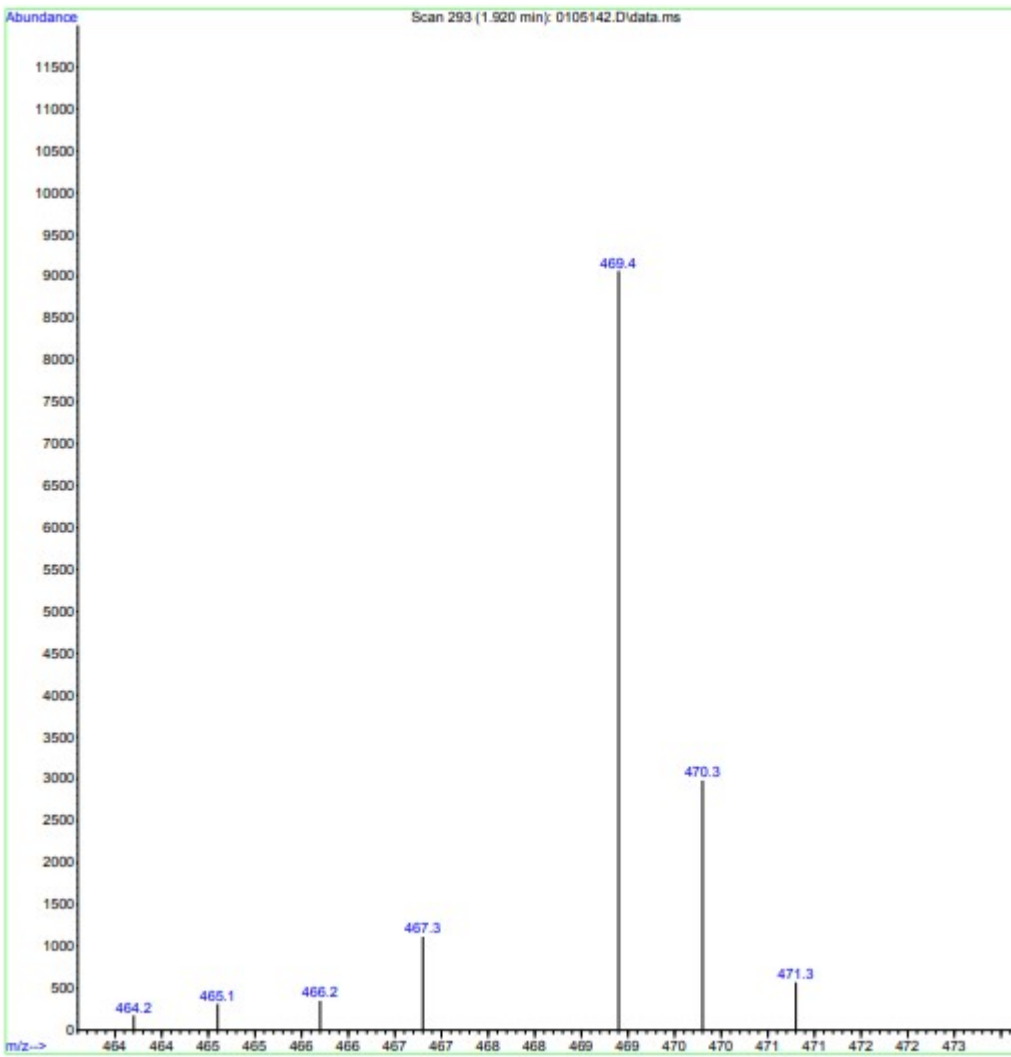


Fig. S6. Mass spectrum of L

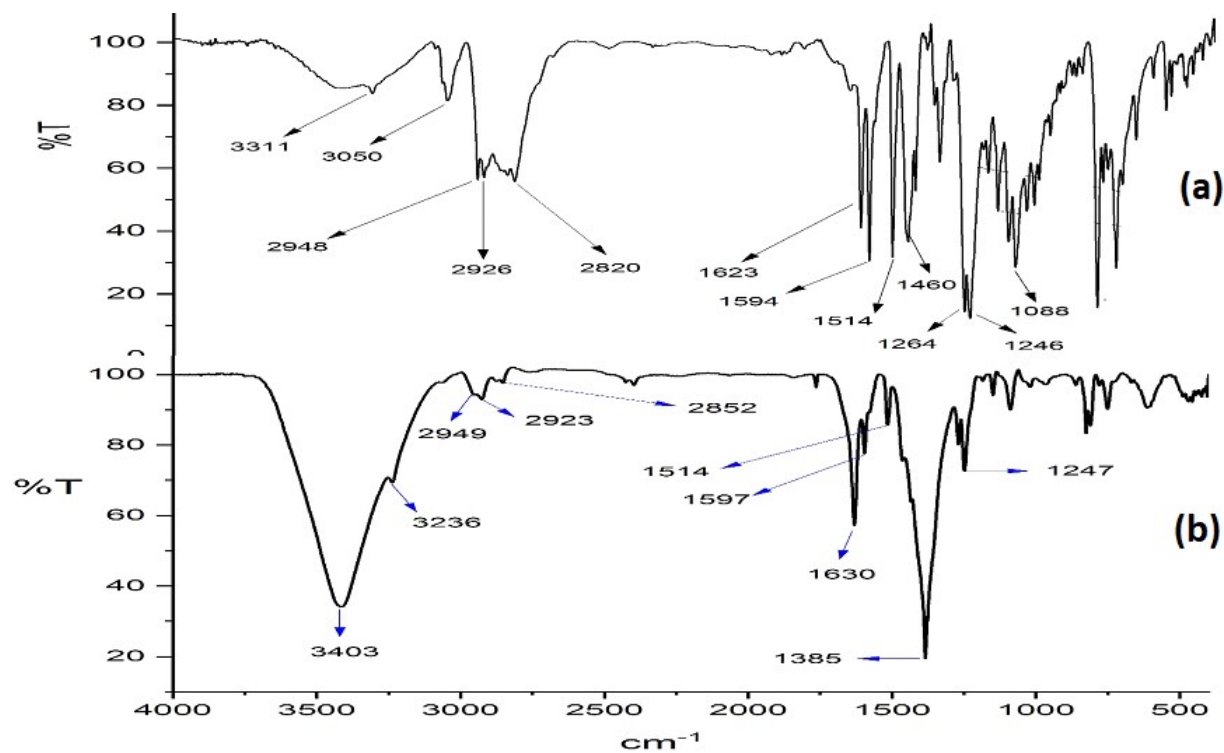


Fig. S7. IR spectra of L (a) L and (b) $[\text{MgL}]^{2+}$

Table S1. Selected bond lengths (Å) and angles (°) for $[\text{MgL}]^{2+}$ and L.

	$[\text{MgL}]^{2+}$		L
Bond length	(Å)	Distance	(Å)
Mg-O(1)	2.079	O(1)-N(1)	3.43
Mg-O(2)	2.110	O(1)-N(2)	5.16
Mg-N(1)	2.113	O(1)-N(3)	7.13
Mg-N(2)	2.200	O(2)-N(1)	5.76
Mg-N(3)	2.174	O(2)-N(2)	4.59
Bond angle	(°)	O(2)-N(3)	3.71
O(1)-Mg-O(2)	85.39	O(1)-O(2)	5.83

O(1)-Mg-N(1)	91.30	N(1)-N(2)	3.30
N(1)-Mg-N(2)	81.59	N(1)-N(3)	5.19
N(2)-Mg-N(3)	80.75	N(2)-N(3)	2.91
N(3)-Mg-O(2)	76.49		

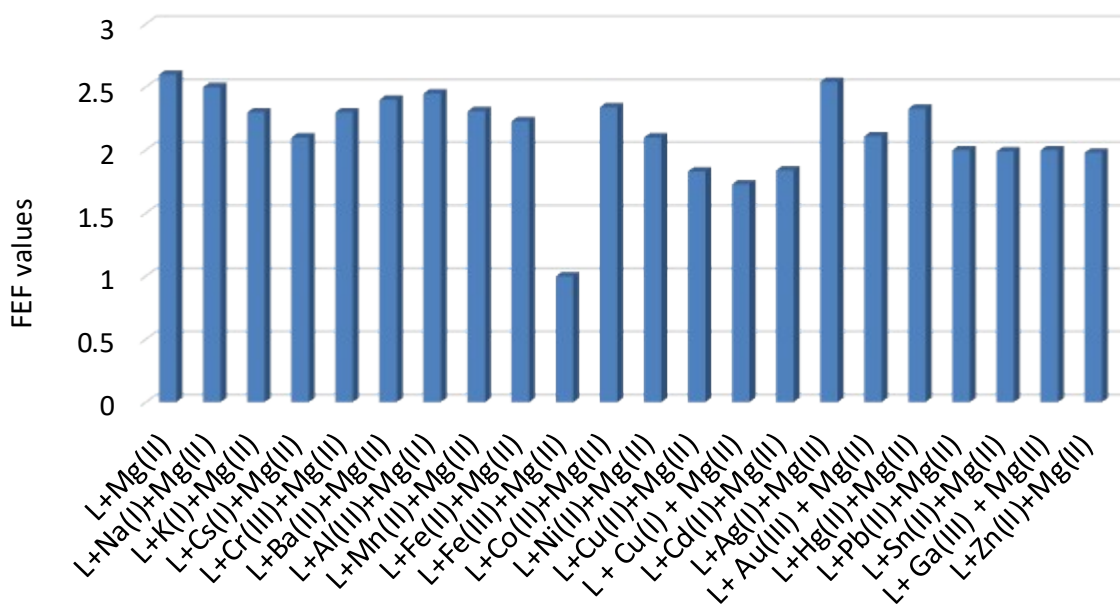


Fig. S8. The FEF value of (L + Mg²⁺) in the presence of various metal ions in EtOH–H₂O solution (9:1, v/v) at room temperature.

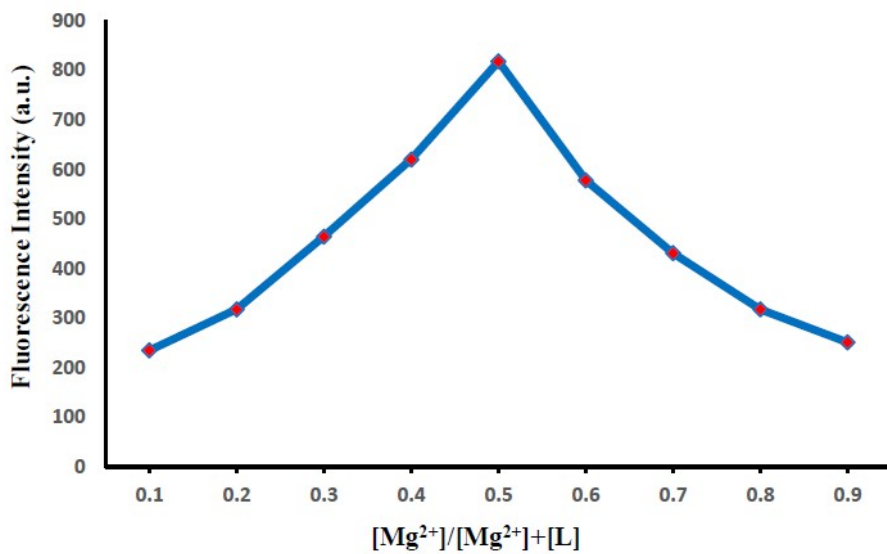


Fig. S9. Job's plot analysis of L and Mg^{2+} .

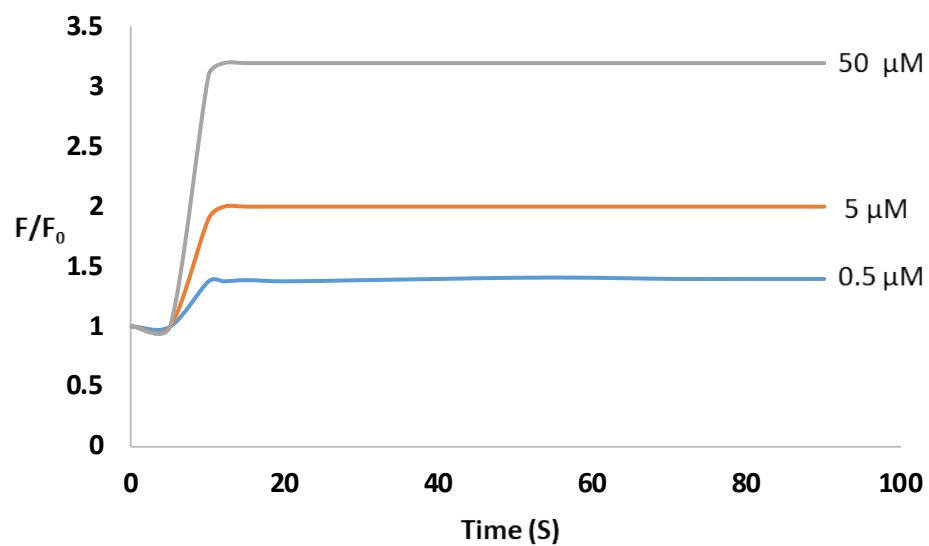


Fig. S10. Kinetics of the fluorescence enhancement of L (10 μM) in the presence of different concentrations of Mg^{2+} . Fluorescence intensity was recorded at 360 nm.

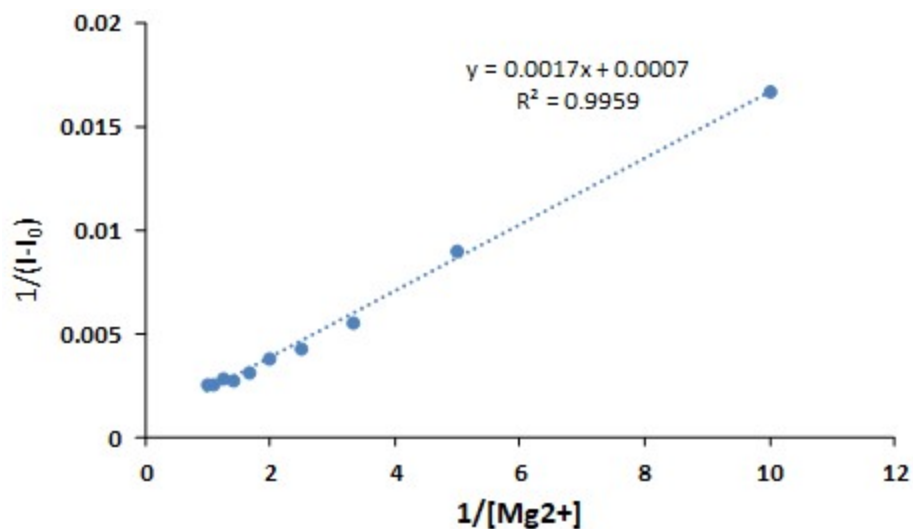


Fig. S11. Benesi-Hildebrand plot of L (10 μ M) with Mg^{2+} in EtOH/ H_2O (9:1, v/v) solution.

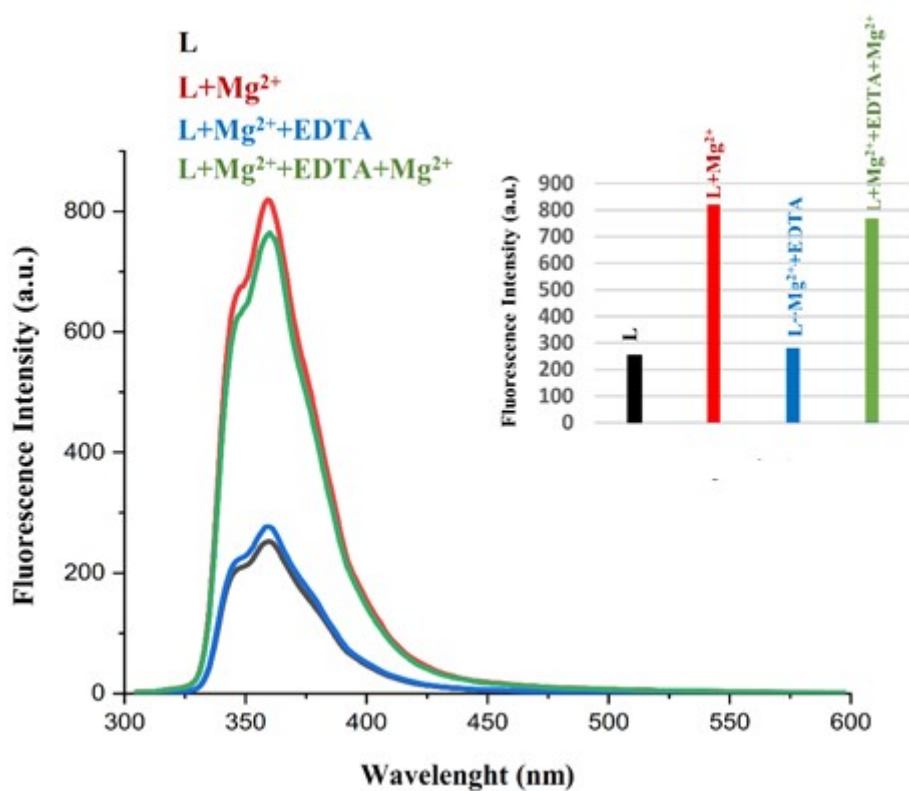


Fig. S12. Chemical reversibility behavior of the binding of L and $[MgL]^{2+}$ ions. Fluorescence intensity was recorded at 360 nm.