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#### **Electronic Supporting Information (ESI) for**

Silver(I) complexes containing heteroleptic diorganochalcogen(II) ligands

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## NMR spectra

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**Figure S4.** <sup>77</sup>Se NMR spectra (CDCl<sub>3</sub>, 400 MHz) of **1** (a), **3** (b) and **5** (c)



Figure S5. UV-Vis spectra of compounds 5, 8 and 10 (dichloromethane,  $5 \cdot 10^{-5}$  M).



Figure S6. UV-Vis spectra of compounds 6, 9 and 11 (dichloromethane,  $5 \cdot 10^{-5}$  M).



Figure S7. Thermal ellipsoids representation at 50% probability of 6. Hydrogen atoms are omitted for clarity.



**Figure S8.** Dimeric association in **6** [symmetry equivalent position (2-x, 1-y, 2-z) is given by "prime"]. Hydrogen atoms not involved in intermolecular interactions are omitted for clarity. O2…H13' 2.50 Å.



**Figure S9.** Best view of the supramolecular leader-like chain of dimers build through  $\pi \cdots \pi$  interactions in the crystal of **5** (only hydrogen atoms involved in intermolecular interactions are shown). Symmetry equivalent atoms (2-x, 1-y, 2-z), (x, -1+y, z) and (x, 1+y, z) are given by "prime" "a", and "b", respectively. Cg1…Cg2 3.50 Å.



Figure S10. Best view of the supramolecular chain of dimers build through C–H···π interactions in the crystal of 6 (only hydrogen atoms involved in intermolecular interactions are shown). Symmetry equivalent atoms (2-x, 1-y, 2-z), (1-x, 1/2+y, 3/2-z) and (1+x, 1/2-y, 1/2+z) are given by "prime" "double prime", and "triple prime", respectively. O2···H13' 2.50 Å and C–H19A''···Cg1 2.99 Å.

Compound	5	6	9
Empirical formula	$C_{20}H_{19}NO_2Se$	$C_{20}H_{19}NO_2S$	$C_{42}H_{38}Ag_2F_6N_2O_{10}S_4$
Formula weight	384.32	337.42	1188.72
<i>T</i> [K]	100.(2)	100.(2)	100.(2)
Crystal system	monoclinic	monoclinic	monoclinic
Space group	P21/c	P21/c	C2/c
<i>a</i> [Å]	18.3038(11)	18.681(2)	41.589(5)
<i>b</i> [Å]	5.2946(3)	5.2749(6)	13.9962(18)
<i>c</i> [Å]	18.3266(11)	17.908(2)	15.4588(17)
<i>α</i> [°]	90	90	90
β[°]	102.537(2)	104.319(4)	100.215(5)
γ[°]	90	90	90
V [Å <sup>3</sup> ]	1733.71(18)	1709.8(3)	8855.7(18)
Ζ	4	4	8
$\rho_{\text{calcd}}$ [g cm <sup>-3</sup> ]	1.472	1.311	1.783
Absorption coefficient [mm <sup>-1</sup> ]	2.177	0.201	1.159
Crystal size [mm]	0.057x0.087x0.12	0.089x0.092x0.122	0.012x0.085x 0.132
$\Theta$ range for data collection	2.28 to 28.33	2.2503 to 25.7198	2.3301 to 28.2410
Reflections collected	62141	36429	131001
Independent reflections	4317	4226	11010
	[R(int) = 0.1672]	[R(int) = 0.2359]	[R(int) = 0.0571]
Data/restraints/parameters	4317/0/221	36429/0/218	11010/0/597
Final R indices [I>2sigma(I)]	R1 = 0.0452	R1 = 0.0999	R1 = 0.0293
	wR2 = 0.0688	wR2 = 0.1527	wR2 = 0.0518
R indices (all data)	R1 = 0.0929	R1 = 0.2132	R1 = 0.0437
	wR2 = 0.0818	wR2 = 0.1900	wR2 = 0.0571
Goodness-of-fit on F <sup>2</sup>	1.088	1.064	1.057
Largest diff. peak/hole / e Å-3	0.551 / -0.480	0.850 / -0.301	0.492 / -0.625

 Table S2. Crystal data and details of data collection for 5, 6 and 9.







Figure S14.  $^1\text{H}$  NMR spectrum (CDCl<sub>3</sub>, 400.13 MHz, 20 °C) of 2











Figure S20. <sup>77</sup>Se{<sup>1</sup>H} NMR spectrum (CDCl<sub>3</sub>, 76.31 MHz, 20 °C) of 5







Figure S26.  ${}^{19}F{}^{1}H$  NMR spectrum (CDCl<sub>3</sub>, 376.49 MHz, 20 °C) of 7







**Figure S32**. <sup>19</sup>F{<sup>1</sup>H} NMR spectrum (acetone-*d*<sub>6</sub>, 564.68 MHz, 20 °C) of **9** 





-56 -57 -58 -59 -60 -61 -62 -63 -64 -65 -66 -67 -68 -69 -70 -71 -72 -73 -74 -75 -76 -77 -78 -79 -80 -81 -82 -83 -84 -85 -86 -87 -88 f1 (ppm)

Figure S36. <sup>19</sup>F{<sup>1</sup>H} NMR spectrum (acetone-*d*<sub>6</sub>, 564.68 MHz, 20 °C) of **10** 





Figure S40. <sup>19</sup>F{<sup>1</sup>H} NMR spectrum (acetone-*d*<sub>6</sub>, 564.68 MHz, 20 °C) of **11** 



-120 -122 -124 -126 -128 -130 -132 -134 -136 -138 -140 -142 -144 -146 -148 -150 -152 -154 -156 -158 -160 -162 -164 -16 f1 (ppm)

**Figure S41**. <sup>31</sup>P{<sup>1</sup>H} NMR spectrum (acetone- $d_6$ , 242.93 MHz, 20 °C) of **11**