

Supporting Information

Hollow Metal Composite Phosphides Derived from MOFs as Highly Efficient and Durable Bifunctional Electrocatalyst for Water Splitting

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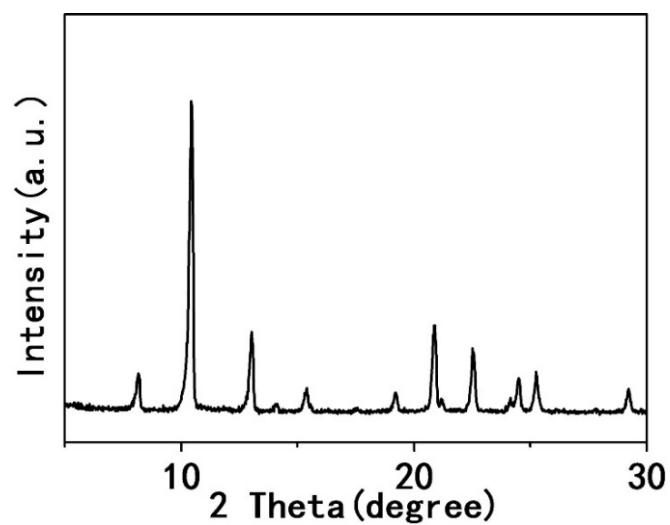


Figure S1. XRD patterns of Fe-MIL-88A

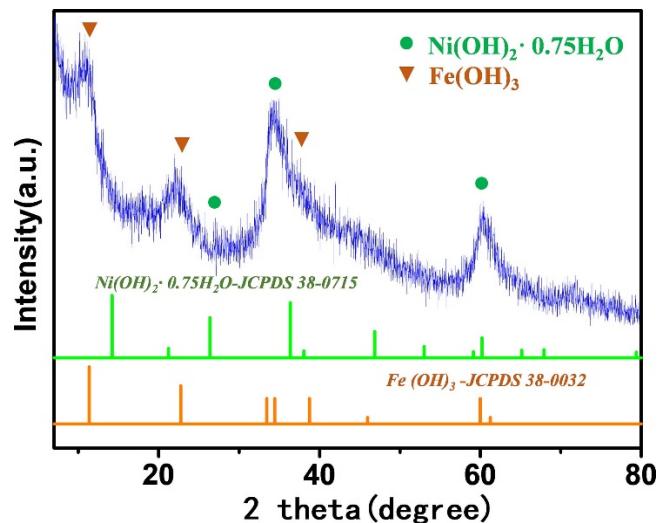


Figure S2. XRD patterns of Fe(OH)₃@Ni(OH)₂·0.75H₂O

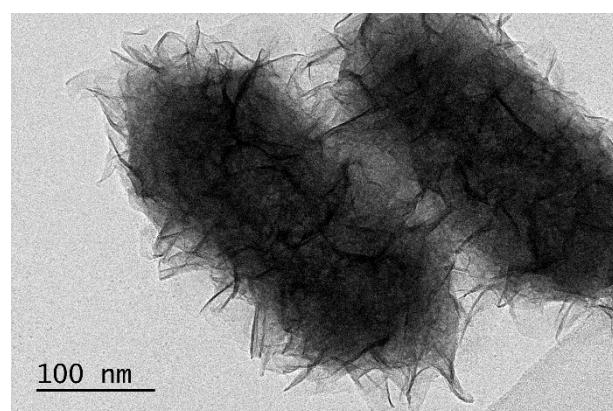


Figure S3. TEM image of Fe(OH)₃@Ni(OH)₂·0.75H₂O

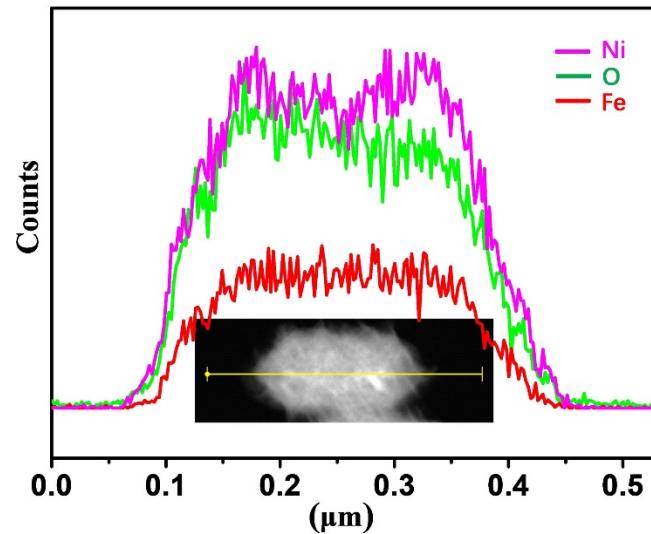


Figure S4. TEM -EDS elemental line-scan of $\text{Fe(OH)}_3 @ \text{Ni(OH)}_2 \cdot 0.75\text{H}_2\text{O}$

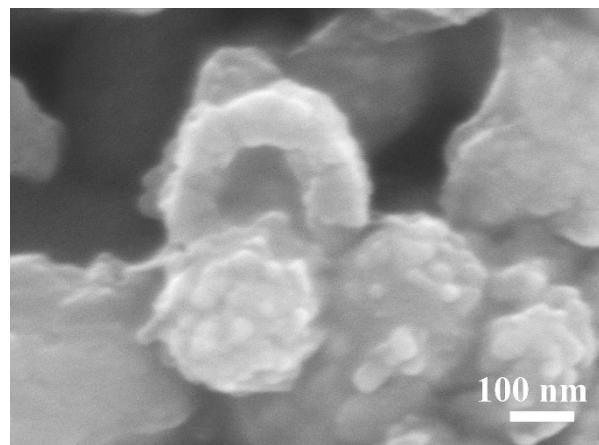


Figure S5. SEM image of a broken $\text{Fe}_2\text{P} @ \text{Ni}_2\text{P}/\text{Ni}_5\text{P}_4$ nanospindle

Table S1. Comparison of HER and OER catalytic performance of hollow Fe₂P@Ni₅P₄/Ni₂P nanospindles and other reported metal phosphides electrocatalysts in acidic and basic conditions.

Materials	Electrolyte	Catalyst loading	Overpotential (η_{10} , mV)	Long-time stability	Ref
Ni ₂ P@NC	OER 1M KOH	1.0 mg cm ⁻²	320	10h (η_{10})	43
Ni ₁₂ P ₅ @NC	OER 1M KOH	1.0 mg cm ⁻²	330	-	43
CoP@NC	OER 1M KOH	1.0 mg cm ⁻²	348	-	43
Co ₂ P@NC	OER 1M KOH	1.0 mg cm ⁻²	383	-	43
FeP@NC	OER 1M KOH	1.0 mg cm ⁻²	409	-	43
Fe ₂ P@NC	OER 1M KOH	1.0 mg cm ⁻²	415	-	43
NiFeP	OER 1M KOH	7.0 mg cm ⁻²	231	20h (η_{100})	47
Fe ₂ P/Ni ₂ P	OER 1M KOH	5.0 mg cm ⁻²	185	100h (η_{20})	48
Fe ₂ P–Ni ₂ P	OER 1M KOH	0.085 mg cm ⁻²	317	20h (η_{10})	49
Co-doped Ni ₅ P ₄	HER 0.5M H ₂ SO ₄	0.05 mg cm ⁻²	310	18h (η_{10})	50
Ni ₂ P@C	HER 1M KOH	0.64 mg cm ⁻²	148	24h (η_{10})	51
Ni ₂ P@C	OER 1M KOH	0.64 mg cm ⁻²	328	24h (η_{20})	51
Ni ₅ P ₄ @FeP	OER 1M KOH	-	205	50h (η_{50})	52
Fe ₂ P@Ni ₅ P ₄ /Ni ₂ P	HER 1M KOH	1.0 mg cm ⁻²	376 (η_{20})	110h (η_{50})	This work
	OER 1M KOH	1.0 mg cm ⁻²	230	49h (η_{20})	