

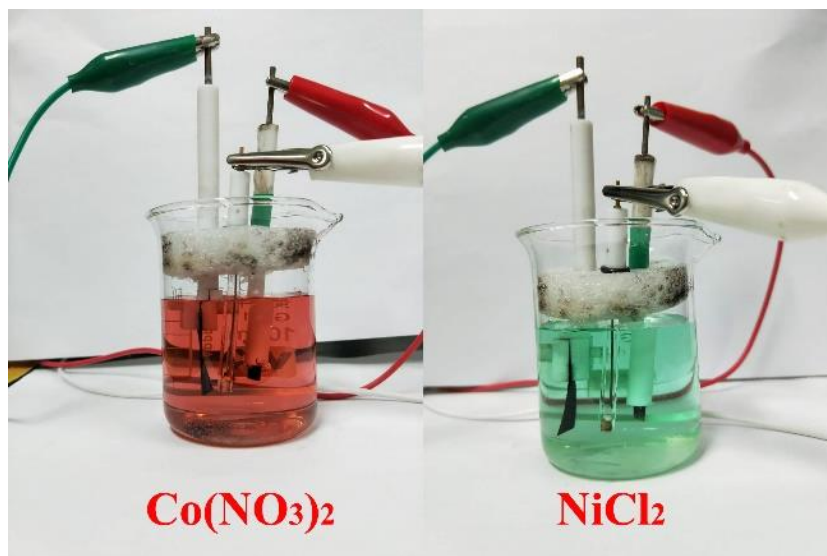
**In-situ Electrochemical Metal (Co, Ni) Oxide Deposition on MoS<sub>2</sub> Nanosheets for  
Highly Efficient Electrocatalytic Water Splitting**

Yueyao Zhong<sup>\*a</sup>, Shouzhi Wang<sup>b</sup>, Lei zhang<sup>b</sup>

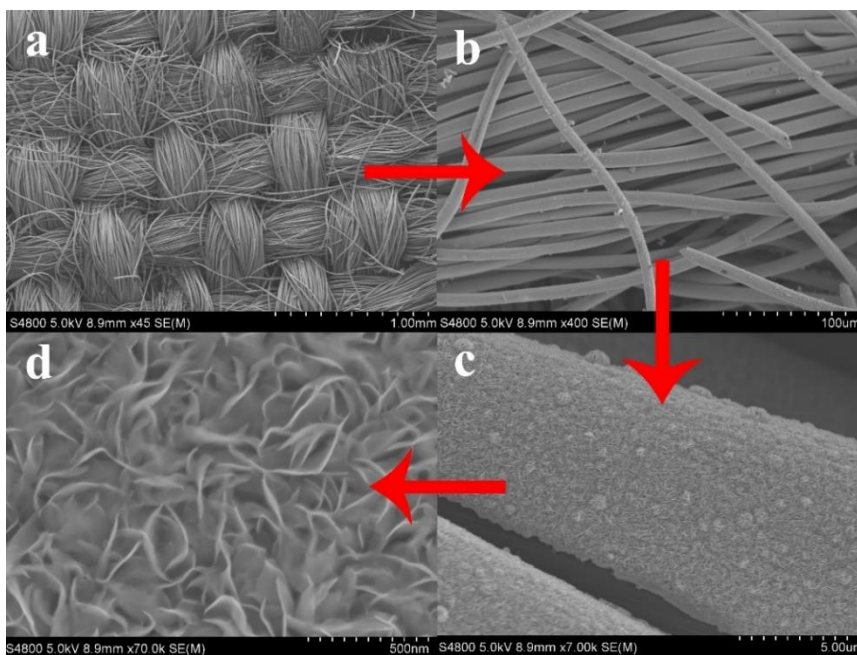
a School of Materials Science and Engineering, Shandong Jianzhu University, Jinan,  
250100, P. R. China

E-mail: zhongyueyao123@163.com

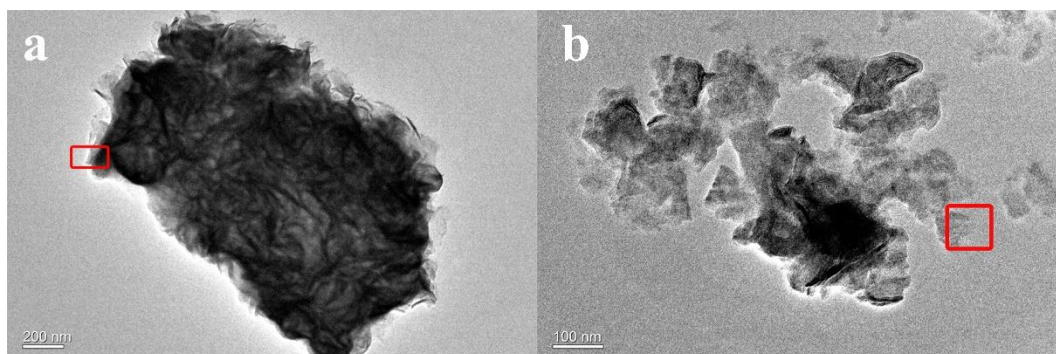
b Institute of Novel Semiconductors, State Key Lab of Crystal Materials, Shandong  
University, Jinan, 250100, P. R. China



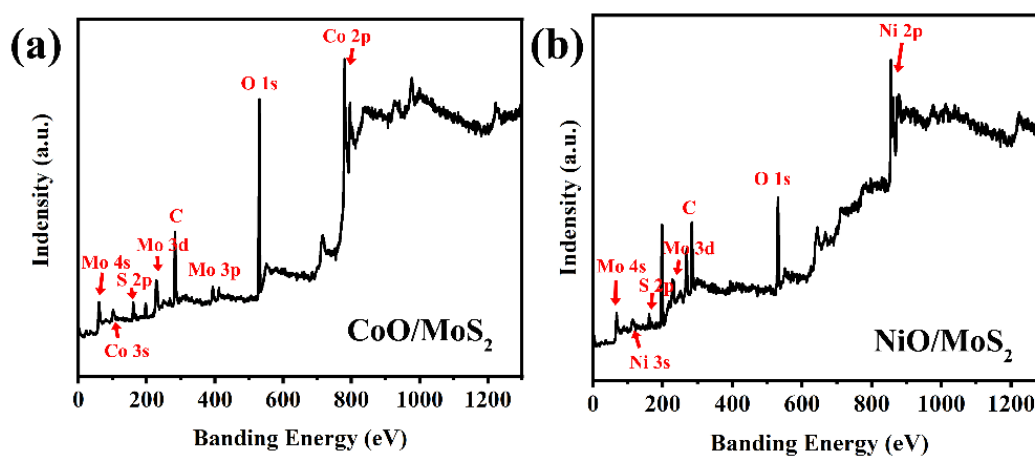
**Fig. S1** Preparation devices of Co/MoS<sub>2</sub>/CC and Ni/MoS<sub>2</sub>/CC.



**Fig. S2** The SEM images of MoS<sub>2</sub>/CC with different magnifications.



**Fig. S3** TEM images of (a) CoO/MoS<sub>2</sub>/CC, (b) NiO/MoS<sub>2</sub>/C.



**Fig. S4** XPS survey spectra of (a) CoO/MoS<sub>2</sub>/CC and (b) NiO/MoS<sub>2</sub>/CC.

**Table S1** Comparison of the electrocatalytic performance of CoO/MoS<sub>2</sub>/CC for HER with that of reported bifunctional MoS<sub>2</sub>-based electrocatalysts tested under similar conditions.

	Overpotential for -10 mA cm <sup>-2</sup> (mV)	Overpotential for -100 mA cm <sup>-2</sup> (mV)	Tafel slope (mV dec <sup>-1</sup> )	Ref.
CoO/MoS <sub>2</sub> /CC	93	162	57.9	This work
NiCo-MoS <sub>2</sub>	70	~	38.1	[1]
Co <sub>3</sub> O <sub>4</sub> @MoO <sub>2</sub> /CC	90	220	59.5	[2]
MoS <sub>2</sub> /FNS/FeNi foam	122	300	45.1	[3]
MoS <sub>2</sub> @NiOOH@C- MC	250	~	96	[4]
NiN@2M-MoS <sub>2</sub>	~	97	43.2	[5]

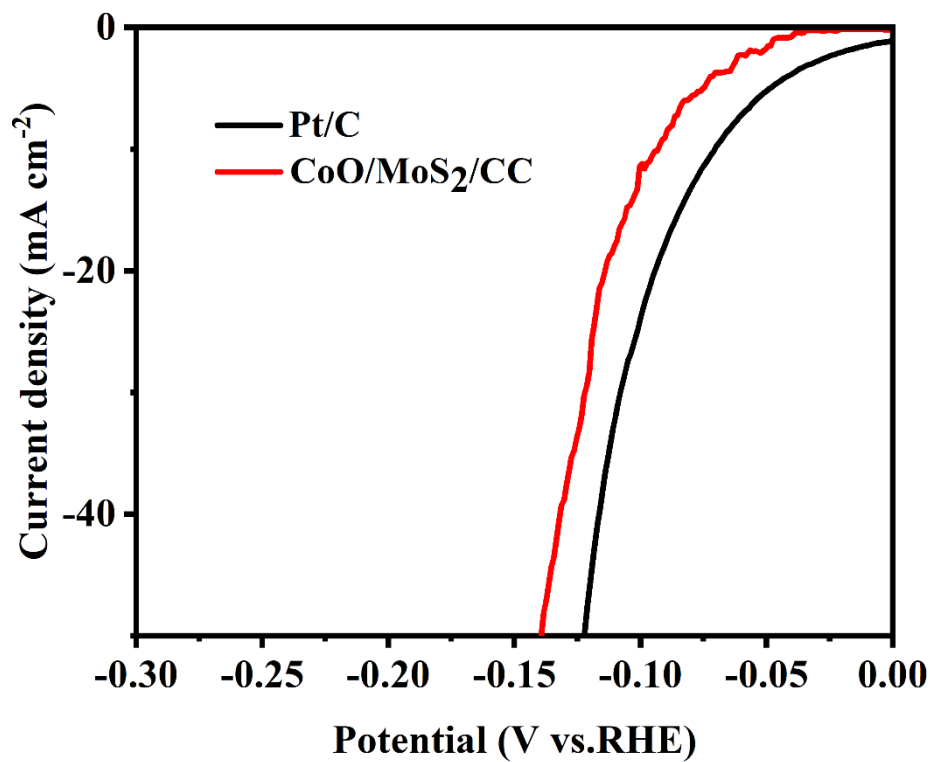


Fig. S5 Polarization curves of CoO/MoS<sub>2</sub>/CC and commercial 20 wt% Pt/C.

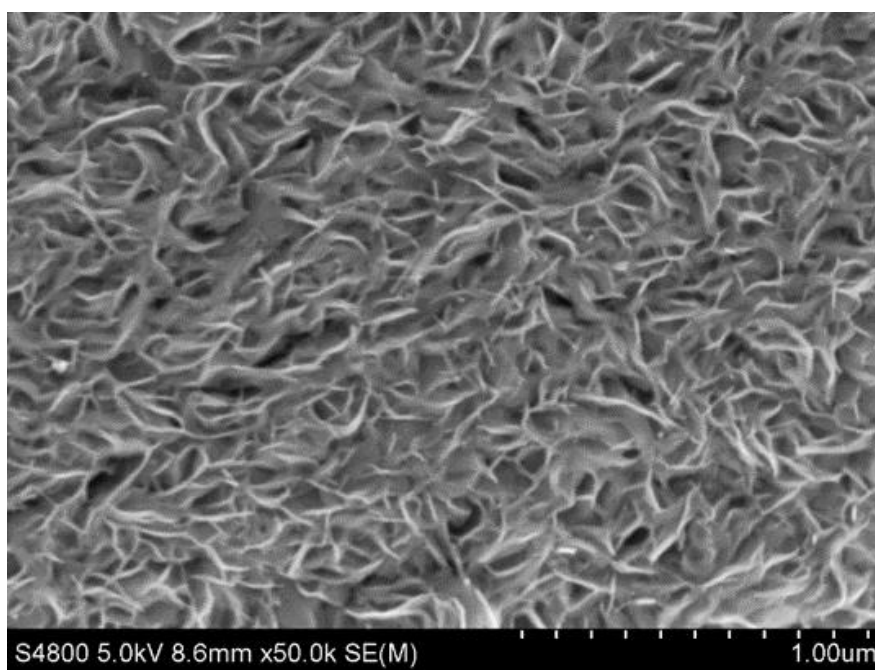
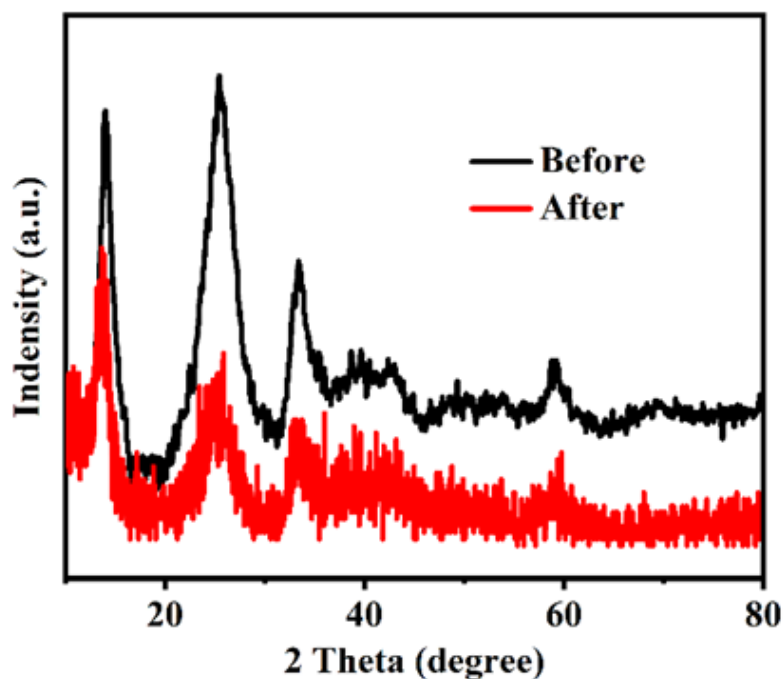


Fig. S6 SEM image of CoO/MoS<sub>2</sub>/CC after HER stability test carry out for 14 h.



**Fig. S7** XRD patterns of CoO/MoS<sub>2</sub>/CC before and after HER stability test carry out for 14 h.

**Table S2** Comparison of the electrocatalytic performance of CoO/MoS<sub>2</sub>/CC for OER with that of reported bifunctional MoS<sub>2</sub>-based electrocatalysts tested under similar conditions.

Samples	Overpotential for 10 mA cm <sup>-2</sup> (mV)	Overpotential for 50 mA cm <sup>-2</sup> (mV)	Tafel slope (mV dec <sup>-1</sup> )	Ref.
CoO/MoS <sub>2</sub> /CC8	210	450	282	This work
NiCo-MoS <sub>2</sub>	235	330	45.7	[1]
Co <sub>3</sub> O <sub>4</sub> @MoO <sub>2</sub> /CC	269	~	58	[2]
MoS <sub>2</sub> /FNS/FeNi foam	204	210	28.6	[3]
MoS <sub>2</sub> @NiOOH@C-MC	280	~	113	[4]
NiN@2M-MoS <sub>2</sub>	~	207	38.9	[5]

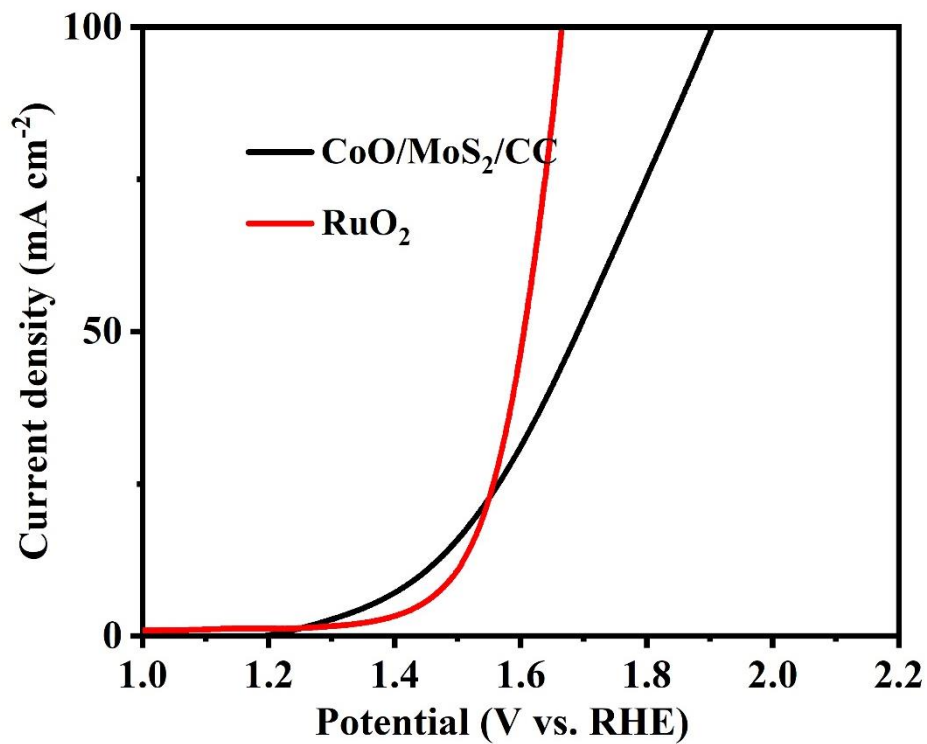


Fig. S8 Polarization curves of CoO/MoS<sub>2</sub>/CC and commercial RuO<sub>2</sub>.

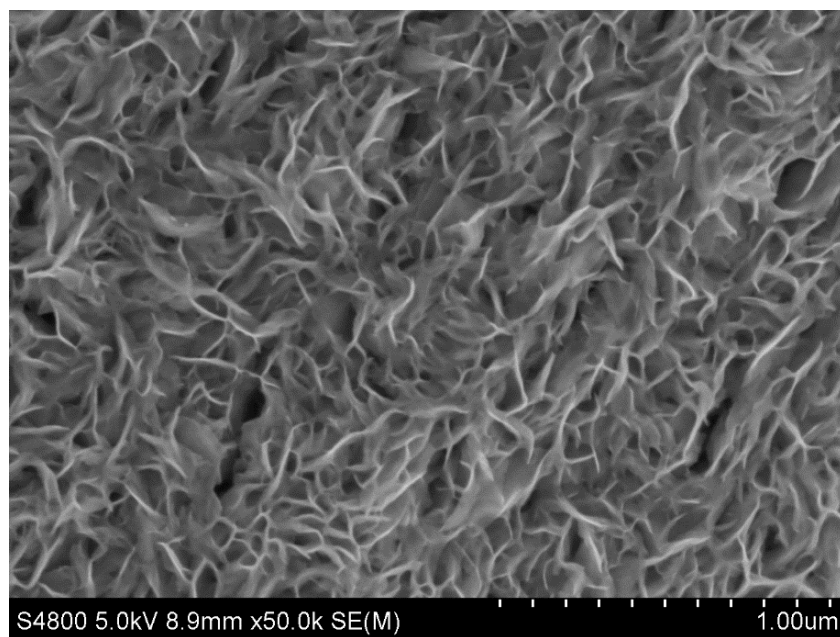
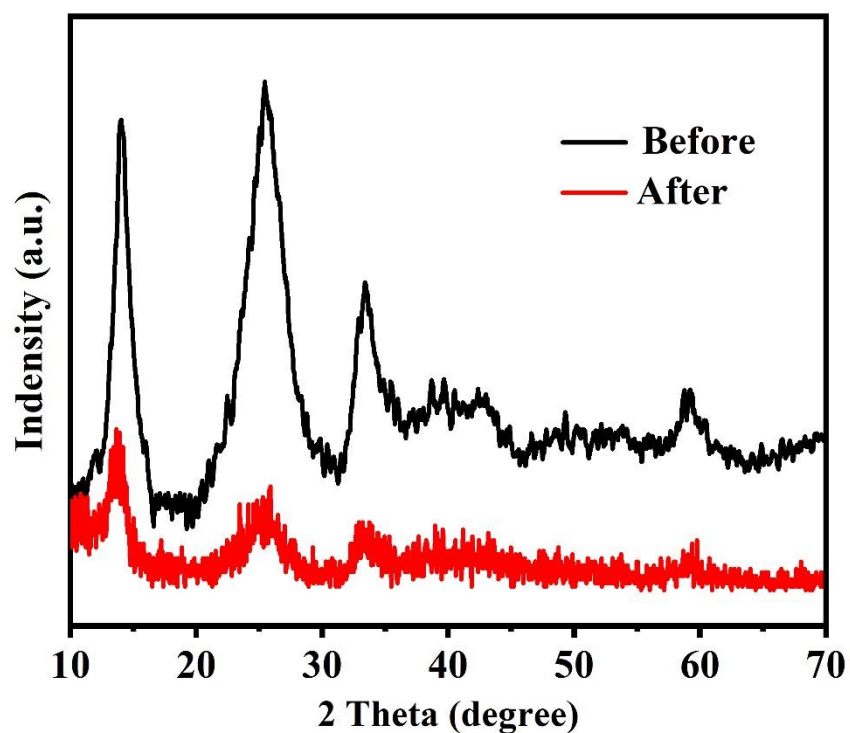


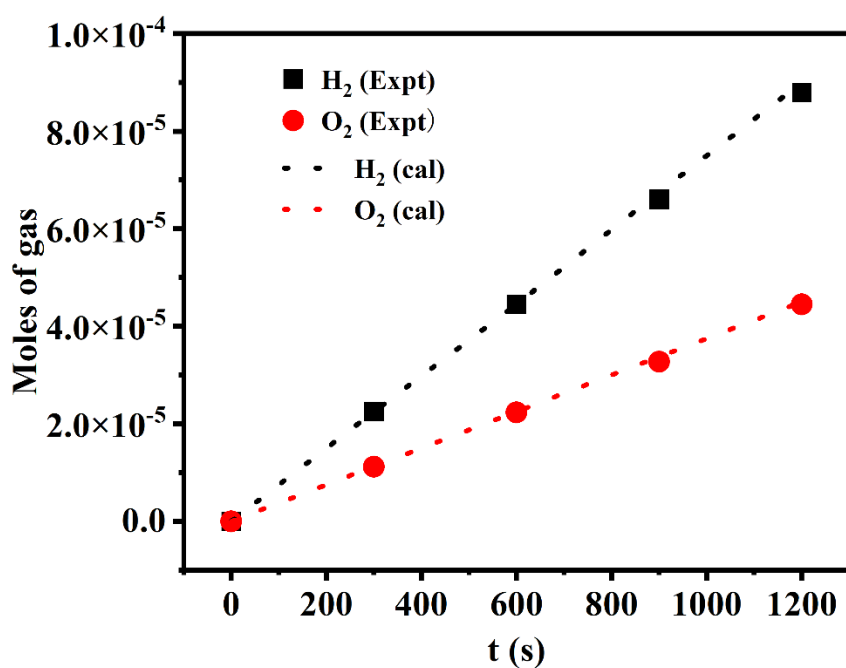
Fig. S9 SEM image of CoO/MoS<sub>2</sub>/CC after OER stability test carry out for 14 h.



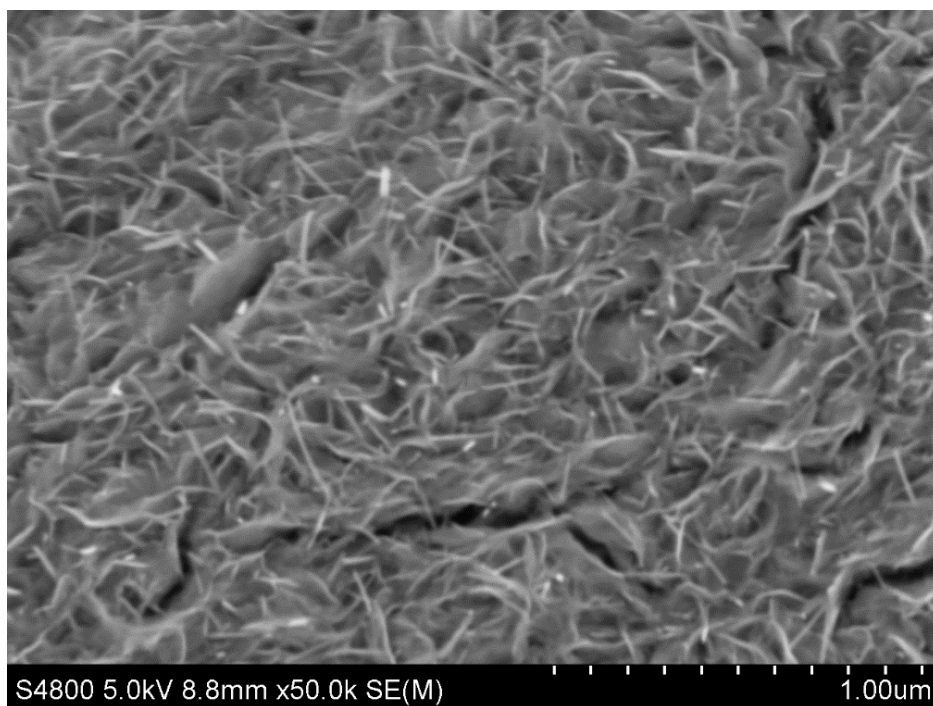
**Fig.S10** XRD patterns of CoO/MoS<sub>2</sub>/CC before and after OER stability test carry out for 14 h.

**Table S3** Comparison of the electrocatalytic performance of CoO/MoS<sub>2</sub>/CC for two-electrode overall water splitting with that of reported bifunctional MoS<sub>2</sub>-based electrocatalysts tested under similar conditions.

Samples	V <sub>10</sub> (V)	V <sub>20</sub> (V)	Durability test made	Ref.
CoO/MoS <sub>2</sub> /CC	1.65	1.75	12 h @ 10 mA cm <sup>-2</sup> (1.65 V)	This work
NiCo-MoS <sub>2</sub>	1.48	1.58	24 h @ 10 mA cm <sup>-2</sup> (1.49 V)	[1]
Co <sub>3</sub> O <sub>4</sub> @MoO <sub>2</sub> /CC	1.59	1.64	12 h @ 10 mA cm <sup>-2</sup> (1.65 V)	[2]
MoS <sub>2</sub> /FNS/FeNi foam	1.62	1.76	40 h @ 10 mA cm <sup>-2</sup> (1.62V)	[3]
NiN@2M-MoS <sub>2</sub>	1.58	1.59	300 h @ 1000 mA cm <sup>-2</sup> (1.65V)	[5]

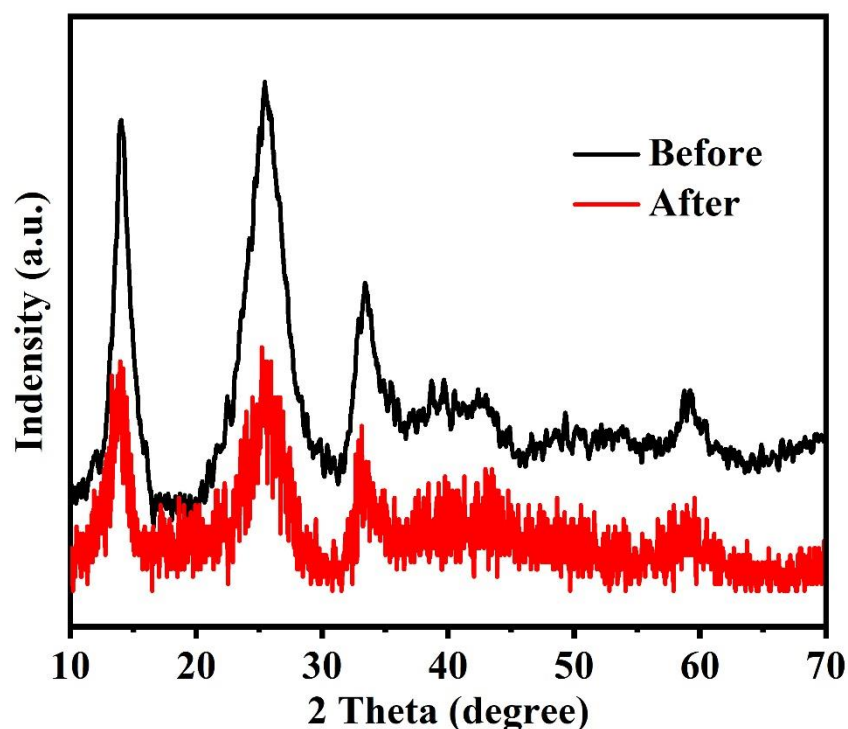


**Fig. S11** Faradic current efficiency for HER and OER of CoO/MoS<sub>2</sub>/CC electrode measured at 10 mA cm<sup>-2</sup> using conventional water displacement. The active electrode area of the cathode and anode was 1.5 cm<sup>2</sup>.



**Fig. S12** SEM image of CoO/MoS<sub>2</sub>/CC after 14 h chronopotentiometric curve test in a two-electrode configuration.





**Fig. S13** XRD patterns of CoO/MoS<sub>2</sub>/CC before and after 14 h chronopotentiometric curve test in a two-electrode configuration.

#### References:

- [1]. H. Li, S. Chen, X. Jia, B. Xu, H. Lin, H. Yang, L. Song, X. Wang, Amorphous nickel-cobalt complexes hybridized with 1T-phase molybdenum disulfide via hydrazine-induced phase transformation for water splitting, *Nat. Commun.* 2017, **8**, 15377.
- [2]. J. Liu, J. Wang, B. Zhang, Y. Ruan, H. Wan, X. Ji, K. Xu, D. Zha, L. Miao, J. Jiang, Mutually beneficial Co<sub>3</sub>O<sub>4</sub>@MoS<sub>2</sub> heterostructures as a highly efficient bifunctional catalyst for electrochemical overall water splitting, *J. Mater. Chem. A* 2018, **6**, 2067.
- [3]. Y. Wu, F. Li, W. Chen, Q. Xiang, Y. Ma, H. Zhu, P. Tao, C. Song, W. Shang, T. Deng, J. Wu, Coupling Interface Constructions of MoS<sub>2</sub>/Fe<sub>5</sub>Ni<sub>4</sub>S<sub>8</sub> Heterostructures for Efficient Electrochemical Water Splitting, *Adv. Mater.* 2018, **30**, 1803151.
- [4]. J. Liu, S. Zhao, C. Wang, Y. Ma, L. He, B. Liu, Z. Zhang, Catkin-derived mesoporous carbon-supported molybdenum disulfide and nickel hydroxyl oxide hybrid as a bifunctional electrocatalyst for driving overall water splitting, journal of colloid and interface science carbon support MoS<sub>2</sub> and NiOH bifunctional electrocatalyst, *J. Colloid Interf. Sci.* 2022, **608**, 1627.
- [5]. T. Wu, E. Song, S. Zhang, M. Luo, C. Zhao, W. Zhao, Ji. Liu, F. Huang, Engineering Metallic Heterostructure Based on Ni<sub>3</sub>N and 2M-MoS<sub>2</sub> for Alkaline Water Electrolysis with Industry-Compatible Current Density and Stability, *Adv. Mater.* 2022, **34**, 2108505.