

Supporting Information for
Facile synthesis of porous LiMn₂O₄ nano-cubes for ultra-stable
lithium-ion battery cathodes

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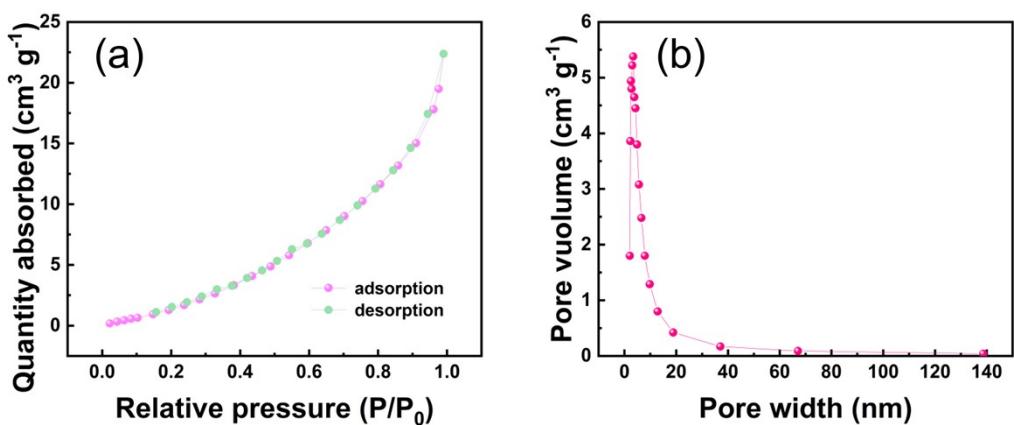


Fig. S1 (a) N2 adsorption-desorption isotherm of LMO; (b) Pore size distribution profiles

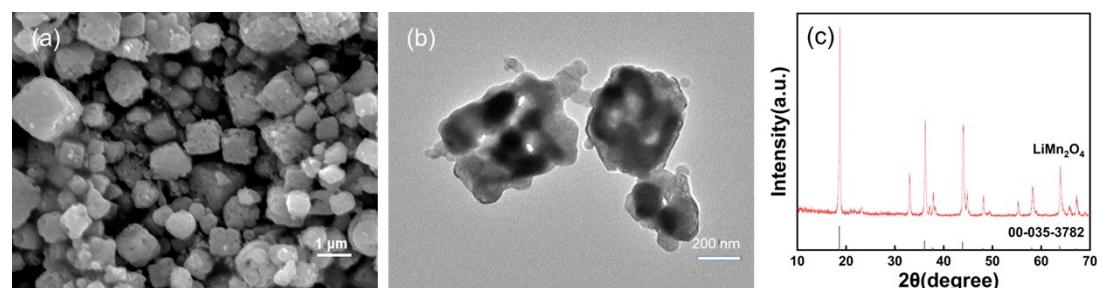


Fig. S2 (a) SEM, (b) TEM and (c) XRD of LMO after cycling

Table S1: Comparison of the cycling performance in this work with published literature

Active material	Synthesis route	Coating or doping	Charge-discharge rate (C)	Cycle number	Initial specific capacity (mAh g ⁻¹)	Capacity retention	Ref.
LiNi _{0.5} Mn _{1.5} O ₄	microwave-assisted	coating and doping	0.1	100	132	95.3%	[1]
LiNi _{0.02} Fe _{0.05} Mn _{1.93} O ₄	solid-state	doping	1	500	113.4	76.9%	[2]
Fe-LiMn ₂ O ₄	solid-state	doping	0.3	200	89.2	89%	[3]
LiMn ₂ O ₄	solid-state	none	1	100	115.7	68%	[4]
LiMn ₂ O ₄	solid-state	none	1	100	128.7	86.2%	[5]
LiMn ₂ O ₄	solid-state	none	0.2	50	119.1	87.2%	[6]
LiMn ₂ O ₄ @rGO	hydrothermal	coating	0.5	200	137.5	75%	[7]
LiMn ₂ O ₄	solid state	none	0.2	100	129.8	95%	[8]
Li _{0.09} K _{0.91} Mn ₂ O ₄	solid-state	doping	0.5	120	137	94.8%	[9]
LiMn ₂ O ₄ /graphene	solid-state	coating	10	160	121.9	82.9%	[10]
LiMn ₂ O ₄	hydrothermal	none	0.1	50	120	84.1%	[11]
LiMg _{0.05} Mn _{1.95} O ₄	molten-salt combustion	doping	1	100	122	86.4%	[12]
Li _{1.05} Al _{0.02} Mn _{1.98} F _{0.02} O _{3.98}	solid-state	doping	0.1	367	115.5	80%	[13]
LiMn ₂ O ₄	precipitation-freeze drying	none	2	500	103.3	87.1%	[14]
LiAl _{0.15} Mn _{1.85} O ₄	solid-state	doping	1	1000	103.3	72%	[15]
LiMn ₂ O ₄	microwave synthesis	none	0.5	100	118.4	74.7%	[16]
LiMn ₂ O ₄	solid-state	none	1	100	134	95.5%	[17]
LiMn ₂ O ₄	homogeneous precipitation	none	0.1	100	112.5	92.5	[18]
LiMn ₂ O ₄ /graphene	solid-state	coating	1	500	111.5	<80%	[19]
★This work	solid-state	none	1	500	104	90.91%	

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