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## Supplementary Material

## Rational design of MOFs@Au@COFs catalyst with electron synergy for

## reduction of 4-nitrophenol

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Fig.S1 EDS images of NH<sub>2</sub>-MIL-88B(Fe)@Au@COFs and analysis of C, O, Au, Fe and N elements



Fig.S2 UV-vis spectra of (a) NH<sub>2</sub>-MIL-88B(Fe)@COFs and (b) NH<sub>2</sub>-MIL-88B(Fe)@Au@COFs



Fig.S3 TGA analysis of (a) NH<sub>2</sub>-MIL-88B(Fe) @COFs and (b) NH<sub>2</sub>-MIL-88B(Fe)



Before reaction at adding catalyst After reaction at separating catalyst
Fig.S4 The change of the solution color before and after the reaction at the
amplification experiment process



Fig.S5 FT-IR spectra of fresh catalyst and recycle catalyst



Fig.S6 TEM images of (a) NH<sub>2</sub>-MIL-88B(Fe)@Au



Fig.S7 EDS images of NH<sub>2</sub>-MIL-88B(Fe)@Au and analysis of C, O, Au, Fe and N

elements



Fig.S8 UV-vis absorption spectra: (a) catalytic reduction of 4-nitrophenol by NH<sub>2</sub>-MIL-88B(Fe)@Au@COFs; (b) the catalytic reduction of 4-nitrophenol by NH<sub>2</sub>-MIL-88B(Fe)@Au; (c) the catalytic reduction of 4-nitrophenol by NaBH<sub>4</sub> in the absence of catalyst.



**Fig.S9** Reusability of synthesized catalyst [NH2-MIL-88B(Fe)@Au@COFs] for the reduction of 4NP in the presence of NaBH4.