

Insight into the photocatalytic mechanism of the optimal x value in $\text{BiOBr}_x\text{I}_{1-x}$, $\text{BiOCl}_x\text{I}_{1-x}$ and $\text{BiOCl}_x\text{Br}_{1-x}$ series varying with pollutant type

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Figure S1 EDS spectra of $\text{BiOBr}_{0.5}\text{I}_{0.5}$ (a) and $\text{BiOBr}_{0.95}\text{I}_{0.05}$ (b)

Table S1 Apparent rate constant (k) of MG photodegradation on $\text{BiOBr}_x\text{I}_{1-x}$ with different x values

x values	0	0.05	0.25	0.4	0.5
k (min^{-1})	0.01286	0.01634	0.01803	0.02050	0.03145
x values	0.6	0.75	0.875	0.95	1
k (min^{-1})	0.01871	0.01723	0.01060	0.00661	0.00630

Table S2 Apparent rate constant (k) of TC photodegradation on $\text{BiOBr}_x\text{I}_{1-x}$ with different x values

x values	0	0.05	0.25	0.4	0.5
k (min^{-1})	0.00251	0.00452	0.00582	0.00735	0.01096
x values	0.6	0.75	0.875	0.95	1
k (min^{-1})	0.01193	0.02535	0.03283	0.03739	0.01048

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Figure S2 Photocatalytic activity of $\text{BiOBr}_x\text{I}_{1-x}$ for the degradation of 20 mg L^{-1} TC (a), 10 mg L^{-1} BPA (b), 10 mg L^{-1} MV (c) and 20 mg L^{-1} RhB (d) under visible light irradiation

Figure S3 Photocatalytic activity of $\text{BiOCl}_x\text{Br}_{1-x}$ for the degradation of 20 mg L^{-1} TC (a), 20 mg L^{-1} RhB (b), 20 mg L^{-1} MG (c) and 20 mg L^{-1} MV (d) under visible light irradiation

Figure S4 Photocatalytic activity of $\text{BiOCl}_x\text{I}_{1-x}$ for the degradation of 20 mg L^{-1} RhB (a), 20 mg L^{-1} TC (b), 30 mg L^{-1} MG (c) and 20 mg L^{-1} MV (d) under visible light irradiation

Figure S5 Time-resolved spectral changes of TC (a), RhB (b) and MG (c) over $\text{BiOBr}_x\text{I}_{1-x}$

Table S3 Equilibrium adsorption capacity (q_e) for MG adsorption by fresh and used $\text{BiOBr}_x\text{I}_{1-x}$

x values	0	0.25	0.5	0.75	0.95	1
q_e (Fresh, mg g^{-1})	32	35	29	28	25	21
q_e (Used, mg g^{-1})	20	21	19	16	14	13

Table S4 Equilibrium adsorption capacity (q_e) for TC adsorption by fresh and used $\text{BiOBr}_x\text{I}_{1-x}$

x values	0	0.25	0.5	0.75	0.95	1
q_e (Fresh, mg g^{-1})	15	20	27	28	26	23
q_e (Used, mg g^{-1})	5	11	18	19	18	14

Fig. S6 Photocatalytic activity of fresh $\text{BiOBr}_x\text{I}_{1-x}$ and $\text{BiOBr}_x\text{I}_{1-x}$ after desorption for the degradation of 20 mg L^{-1} MG (a) and 20 mg L^{-1} TC (b)

Fig. S7 Effect of contact time on the removal of 20 mg L^{-1} MG (a, b) and 20 mg L^{-1} TC (c, d) over fresh $\text{BiOBr}_x\text{I}_{1-x}$ (a, c) and used $\text{BiOBr}_x\text{I}_{1-x}$ (b, d)

Figure S8 Recycling experiments of TC and MG on $\text{BiOBr}_{0.95}\text{I}_{0.05}$ and $\text{BiOBr}_{0.5}\text{I}_{0.5}$, respectively

Figure S9 XRD pattern of $\text{BiOBr}_{0.5}\text{I}_{0.5}$ (a) and (b) $\text{BiOBr}_{0.95}\text{I}_{0.05}$ before and after cycles

Figure S10 Active species trapping experiments of 20 mg L^{-1} TC photodegradation over $\text{BiOBr}_{0.95}\text{I}_{0.05}$ under visible light irradiation

Figure S11 NBT transformation efficiency on BiOI , $\text{BiOBr}_{0.5}\text{I}_{0.5}$, $\text{BiOBr}_{0.95}\text{I}_{0.05}$ and BiOBr under visible light irradiation

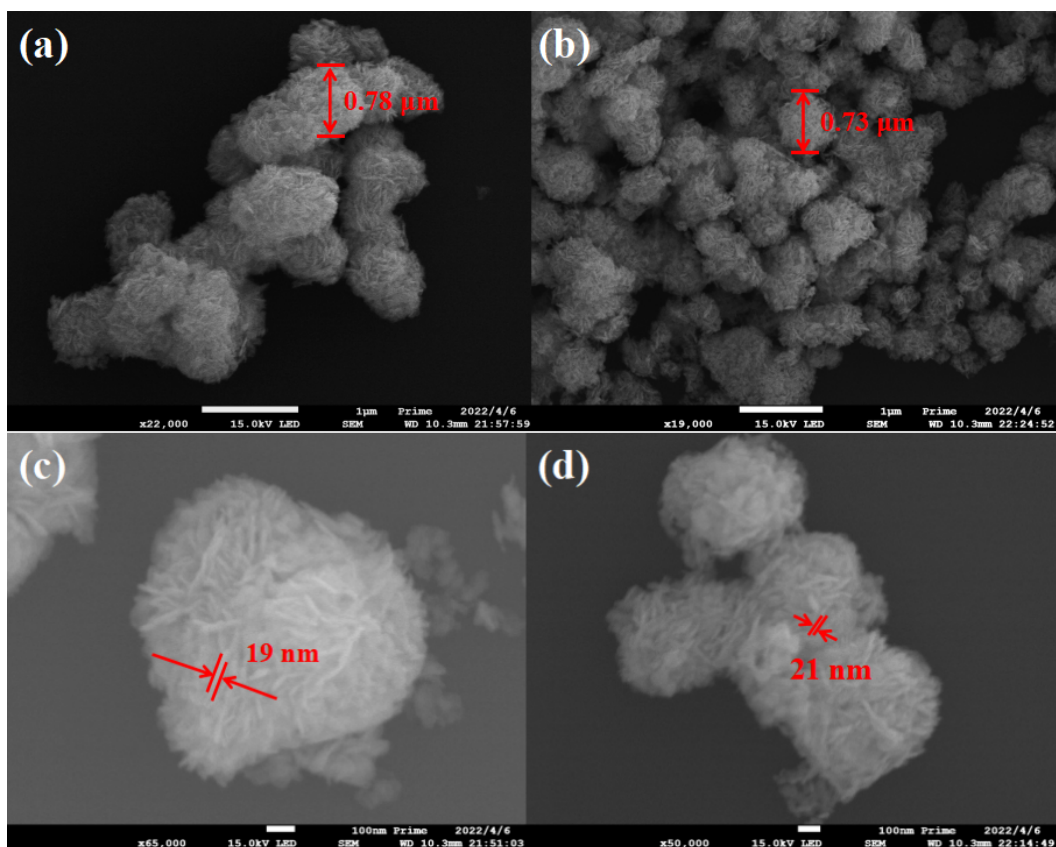


Figure S12 SEM images of $\text{BiOBr}_{0.5}\text{I}_{0.5}$ (a, c) and $\text{BiOBr}_{0.95}\text{I}_{0.05}$ (b, d)

Fig. S13 N_2 adsorption-desorption isotherms of $\text{BiOBr}_{0.25}\text{I}_{0.75}$ (a), $\text{BiOBr}_{0.5}\text{I}_{0.5}$ (b), $\text{BiOBr}_{0.75}\text{I}_{0.25}$ (c) and $\text{BiOBr}_{0.95}\text{I}_{0.05}$ (d), insets showing the corresponding pore size distribution curves and calculated surface areas and mean pore diameters

Fig. S14 Plots of $\Delta(\text{pH})$ vs. pH measured for $\text{BiOBr}_{0.25}\text{I}_{0.75}$ (a), $\text{BiOBr}_{0.5}\text{I}_{0.5}$ (b), $\text{BiOBr}_{0.75}\text{I}_{0.25}$ (c) and $\text{BiOBr}_{0.95}\text{I}_{0.05}$ (d)

Figure S15 Mott-Schottky curves of $\text{BiOBr}_{0.95}\text{I}_{0.05}$ at different frequency (a) and $\text{BiOBr}_x\text{I}_{1-x}$ samples with different x values (b)

Figure S16 UV-vis diffuse reflectance spectra (DRS) and the plots of the $(ah\nu)^{1/2}$ vs. photon energy ($h\nu$) of $\text{BiOCl}_{0.25}\text{Br}_{0.75}$, $\text{BiOCl}_{0.75}\text{Br}_{0.25}$ (a) and $\text{BiOCl}_{0.25}\text{I}_{0.75}$, $\text{BiOCl}_{0.8}\text{I}_{0.2}$ (b)

Figure S17 Mott-Schottky curves of $\text{BiOCl}_{0.25}\text{Br}_{0.75}$, $\text{BiOCl}_{0.75}\text{Br}_{0.25}$ (a) and $\text{BiOCl}_{0.25}\text{I}_{0.75}$, $\text{BiOCl}_{0.8}\text{I}_{0.2}$ (b)

Table S5 The fitting values of each component in the equivalent circuit diagram

x values	0	0.5	0.95	1
$R_1(\Omega)$	29.13	8.766	13.41	16.43
$R_2(\Omega)$	10334	2116	779.2	3694
CPE-T	3.94E-5	2.2E-3	1.64E-4	6.9E-4
CPE-P	0.86	0.35	0.71	0.79

Table S6 The band potentials and band gap energy of $\text{BiOCl}_{0.25}\text{Br}_{0.75}$, $\text{BiOCl}_{0.75}\text{Br}_{0.25}$, $\text{BiOCl}_{0.25}\text{I}_{0.75}$ and $\text{BiOCl}_{0.8}\text{I}_{0.2}$

Samples	E_g (eV)	E_{CB} (V)	E_{VB} (V)
$\text{BiOCl}_{0.25}\text{Br}_{0.75}$	2.48	-0.57	1.91
$\text{BiOCl}_{0.75}\text{Br}_{0.25}$	2.89	-0.61	2.28
$\text{BiOCl}_{0.25}\text{I}_{0.75}$	1.97	-0.18	1.79
$\text{BiOCl}_{0.8}\text{I}_{0.2}$	2.29	-0.11	2.18

Figure S18 Band structures of $\text{BiOCl}_{0.25}\text{I}_{0.75}$, $\text{BiOCl}_{0.8}\text{I}_{0.2}$, $\text{BiOCl}_{0.25}\text{Br}_{0.75}$, $\text{BiOCl}_{0.75}\text{Br}_{0.25}$ and the oxidation potential of MG, MV, TC and RhB