

Supporting Information

Designing Ferrimagnetic-Ferroelastic Multiferroic Semiconductor in FeMoClO₄ Nanosheet via Element Substitution

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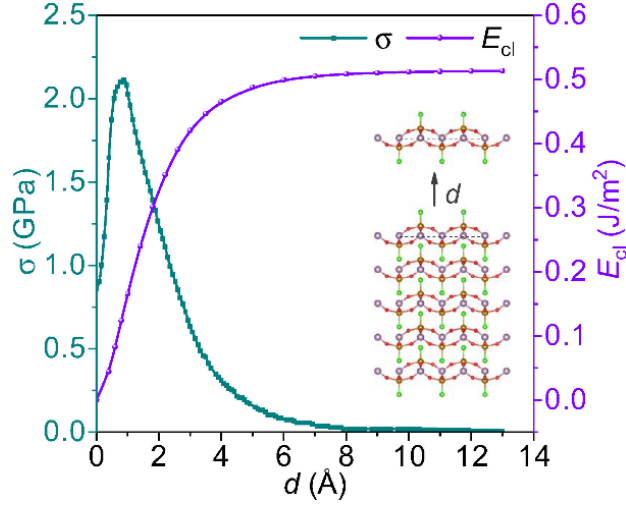


Figure S1. Exfoliation energy E_{cl} and cleavage strength σ as functions of separation distance d in the process of exfoliating one nanosheet from its bulk crystal.

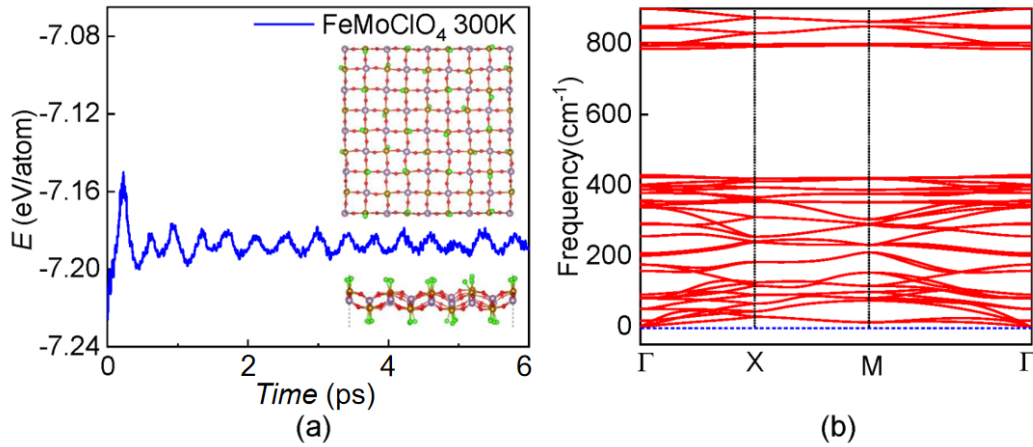


Figure S2. (a) *Ab initio* molecular dynamics (AIMD) simulation of total energy with time for FeMoClO_4 with a $P4/nmm$ space group at 300 K. The insets depict snapshots of $4 \times 4 \times 1$ supercell containing 224 atoms after 6 ps. (b) Phonon dispersions of the FeMoClO_4 nanosheet.

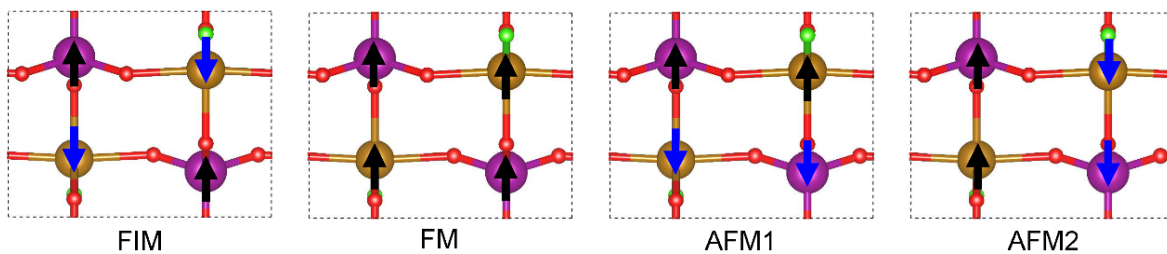


Figure S3. Four different magnetic states calculated for FeMnClO₄ nanosheet. They contain one ferrimagnetic (FIM) state, one ferromagnetic (FM) state, and two antiferromagnetic (AFM) states.

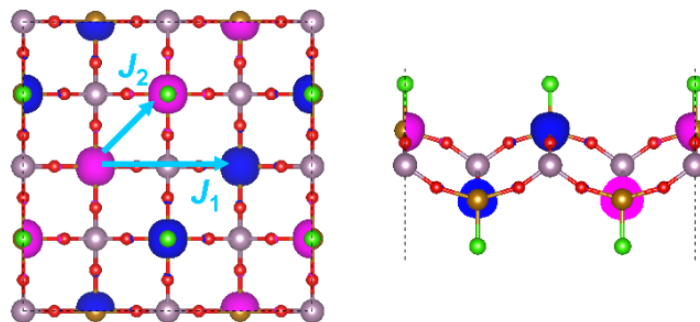


Figure S4. Spin density, nearest and next-nearest spin exchange (J_1 and J_2) paths for the FeMoClO₄ nanosheet. Pink (Blue) denotes spin-up (spin-down) component.

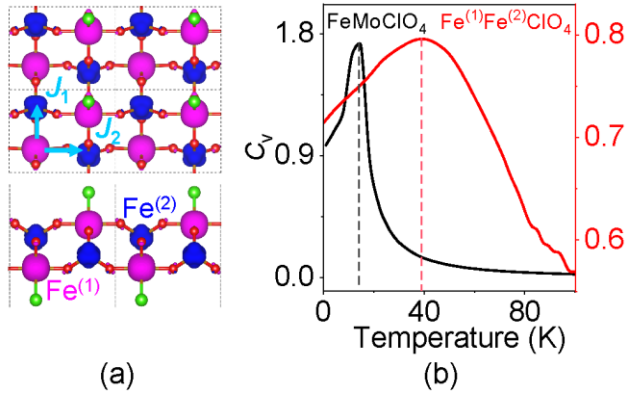


Figure S5. (a) Spin density, nearest and next-nearest spin exchange (J_1 and J_2) paths for 2D $\text{Fe}^{(1)}\text{Fe}^{(2)}\text{ClO}_4$. Pink (Blue) denotes spin-up (spin-down) component. The specific values of spin exchange parameters are shown in Table S4. After $\text{Fe}^{(2)6+}$ replaces Mo^{6+} , the calculated magnetic moment of $\text{Fe}^{(1)}\text{Fe}^{(2)}\text{ClO}_4$ is $6 \mu_B$ per unit cell, which is $2 \mu_B$ per unit cell lower than that of FeMnClO_4 . Noted that among the entire 3d transition metals, Mo^{6+} can be substituted by Cr^{6+} besides Fe^{6+} and Mn^{6+} , but FeCrClO_4 has an AFM ground state as well as FeMoClO_4 . Other transition metals that can not possess a +6 charge state are discarded. (b) Specific heat (C_v) as a function of temperature for FeMoClO_4 and $\text{Fe}^{(1)}\text{Fe}^{(2)}\text{ClO}_4$. The calculated Curie temperature ($T_c = 39 \text{ K}$) of $\text{Fe}^{(1)}\text{Fe}^{(2)}\text{ClO}_4$ is higher than 14 K of FeMoClO_4 , but lower than 127 K of FeMnClO_4 .

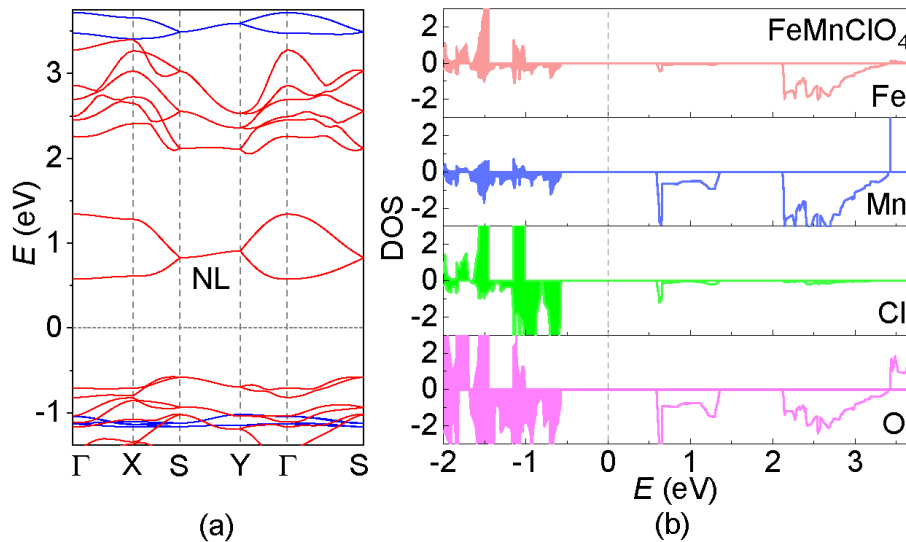


Figure S6. (a) Spin-polarized band structure and (b) projected density of states for the FeMnClO_4 nanosheet calculated by HSE06 functional.

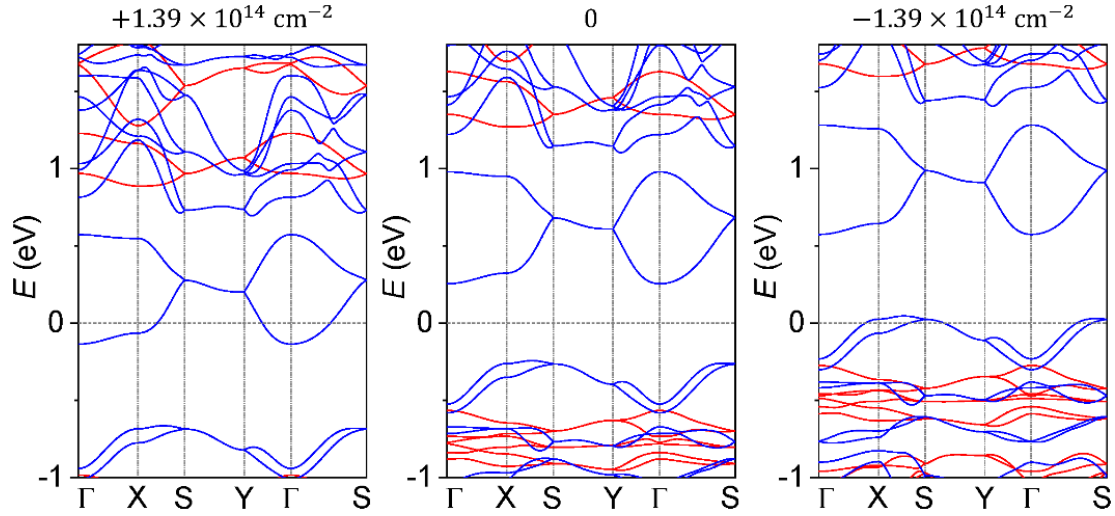


Figure S7. Band structure of FeMnClO₄ nanosheet for electron doping (left), pure (middle) and hole doping (right) with a carrier concentration of $\pm 1.39 \times 10^{14} \text{ cm}^{-2}$. Positive and negative values above the panel represent electron and hole doping, respectively.

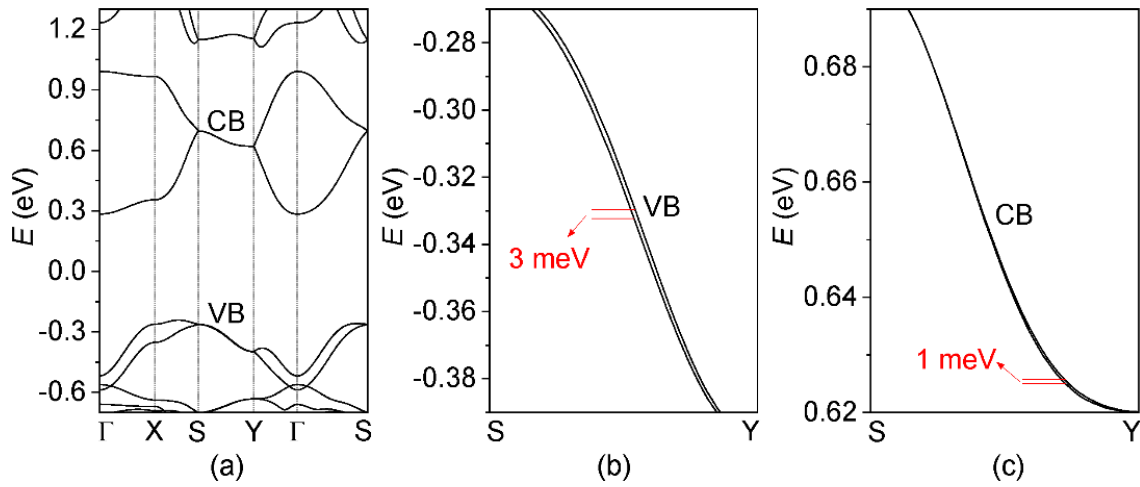


Figure S8. (a) Band structure of FeMnClO₄ nanosheet after considering the spin orbit coupling effect. (b, c) Topological nodal lines in valence band (VB) and conduction band (CB).

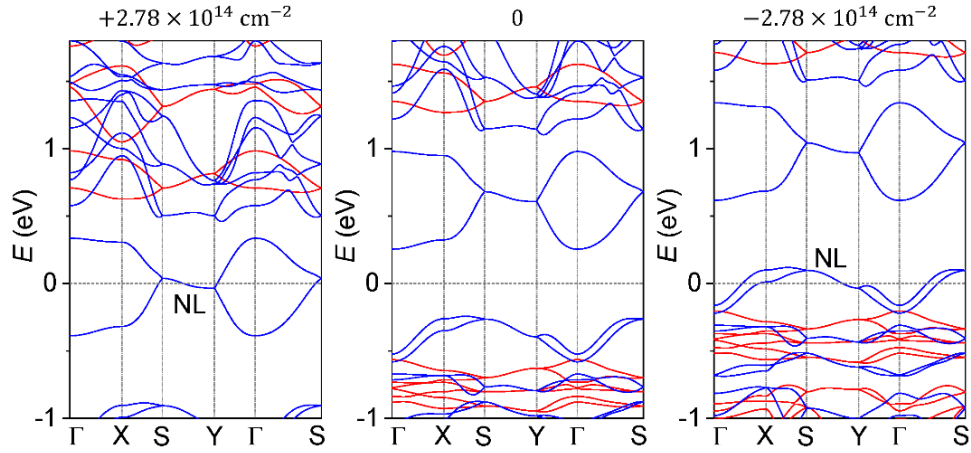


Figure S9. Band structure of FeMnClO₄ nanosheet for electron doping (left), pure (middle) and hole doping (right) with a carrier concentration of $2.78 \times 10^{14} \text{ cm}^{-2}$. Positive and negative values above the panel represent electron and hole doping, respectively.

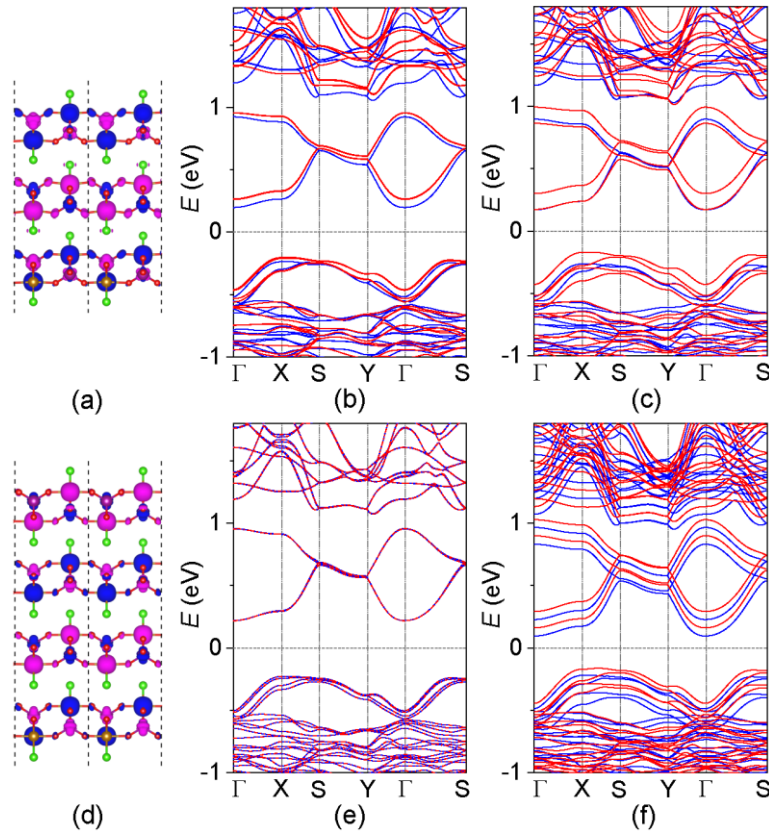


Figure S10. (a, d) Spin densities and (b, c, e, f) band structures of trilayer and four-layer FeMnClO₄, where the band structures in (c, f) panels are plotted under a perpendicular electric field of 0.01 V/Å.

Table S1. Calculated relative energies of ferromagnetic (FM) state to ferrimagnetic (FIM) state ΔE (eV per chemical formula) for FeMnClO_4 at different values of onsite Coulomb interaction U .

| ΔE (FeMnClO_4) | $U_{\text{Mn}=2}$ | $U_{\text{Mn}=3}$ | $U_{\text{Mn}=4}$ | $U_{\text{Mn}=5}$ | $U_{\text{Mn}=6}$ |
|-----------------------------------|-------------------|-------------------|-------------------|-------------------|-------------------|
| $U_{\text{Fe}=2}$ | 0.726 | 0.677 | 0.607 | 0.512 | 0.376 |
| $U_{\text{Fe}=3}$ | 0.696 | 0.656 | 0.588 | 0.495 | 0.394 |
| $U_{\text{Fe}=4}$ | 0.659 | 0.627 | 0.556 | 0.474 | 0.387 |
| $U_{\text{Fe}=5}$ | 0.621 | 0.593 | 0.523 | 0.451 | 0.374 |
| $U_{\text{Fe}=6}$ | 0.581 | 0.552 | 0.488 | 0.424 | 0.357 |

Table S2. Magnetic exchange parameters (in meV), magnetic anisotropy energies (meV per chemical formula), and axial (planar) anisotropy strength (meV) for the FeMoClO_4 nanosheet.

| FeMoClO_4 | J_1 | J_2 | $E_{[010]}-E_{[100]}$ | $E_{[001]}-E_{[100]}$ | D_{Fe} | E_{Fe} |
|--------------------|-------|-------|-----------------------|-----------------------|-----------------|-----------------|
| | 0.23 | 0.14 | 0 | 1.32 | 0.21 | 0.00 |

Table S3. Magnetic exchange parameters (in meV), magnetic anisotropy energies (meV per chemical formula), and axial (planar) anisotropy strength (meV) for the FeMnClO_4 nanosheet.

| FeMnClO_4 | J_1 | J_2 | $E_{[010]}-E_{[100]}$ | $E_{[001]}-E_{[100]}$ | D_{Fe} | E_{Fe} | D_{Mn} | E_{Mn} |
|--------------------|-------|-------|-----------------------|-----------------------|-----------------|-----------------|-----------------|-----------------|
| | 45.77 | 13.00 | 0.61 | 1.31 | 0.13 | 0.03 | 0.38 | 0.59 |

Table S4. Magnetic exchange parameters (in meV), magnetic anisotropy energies (meV per chemical formula), and axial (planar) anisotropy strength (meV) for the $\text{Fe}^{(1)}\text{Fe}^{(2)}\text{ClO}_4$ nanosheet.

| $\text{Fe}^{(1)}\text{Fe}^{(2)}\text{ClO}_4$ | J_1 | J_2 | $E_{[010]}-E_{[100]}$ | $E_{[001]}-E_{[100]}$ | $D_{\text{Fe}(1)}$ | $E_{\text{Fe}(1)}$ | $D_{\text{Fe}(2)}$ | $E_{\text{Fe}(2)}$ |
|--|-------|-------|-----------------------|-----------------------|--------------------|--------------------|--------------------|--------------------|
| | 18.47 | 1.98 | 0.48 | 1.21 | 0.10 | 0.01 | -0.14 | 0.06 |