

## Electronic Supplementary Information

### **Ru doped WO<sub>3</sub> Enabling Efficient Hydrogen Oxidation Reaction in Alkaline Media**

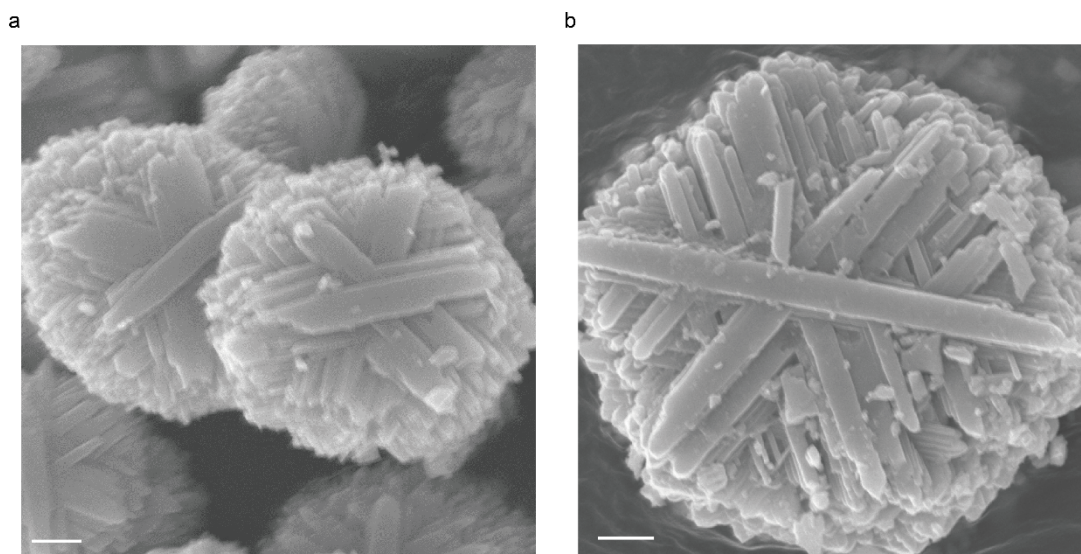
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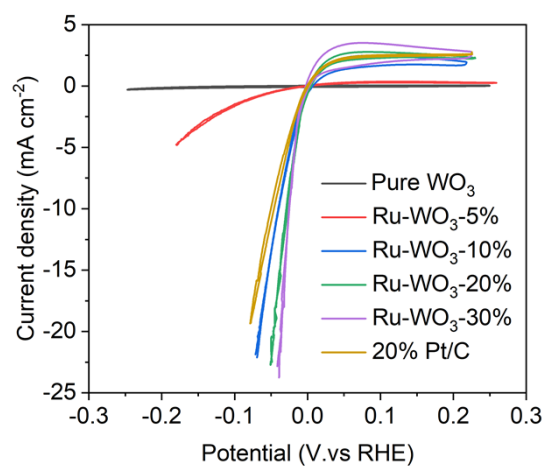
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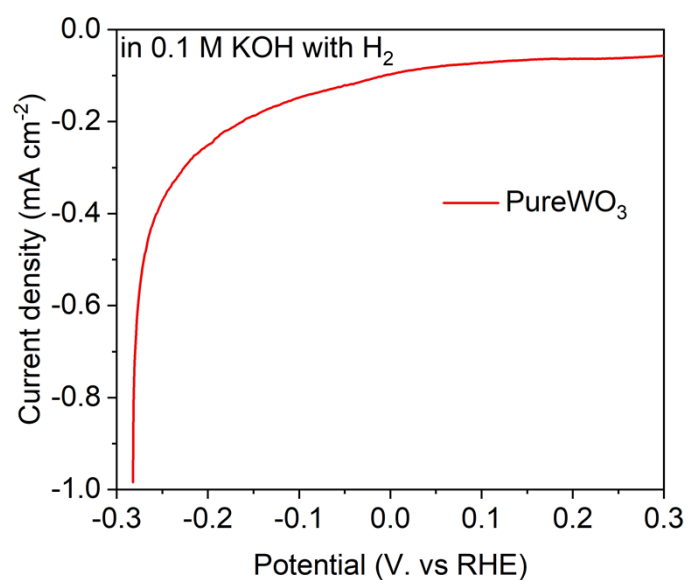
## Figures and Tables



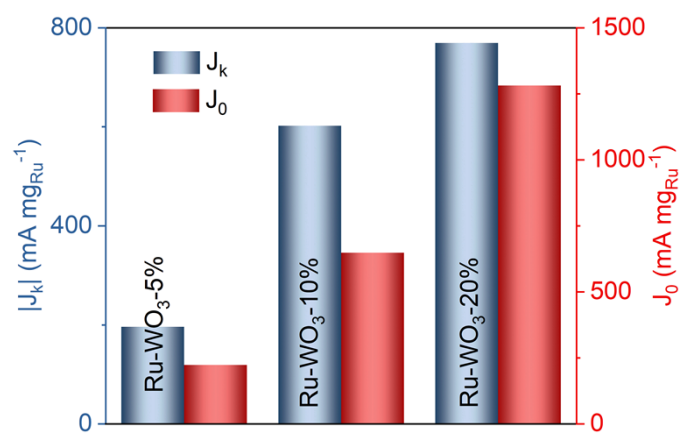
**Figure S1.** SEM images for Ru-WO<sub>3</sub> with different ratios at 200 nm scale. a) 5% Ru. b) 10% Ru.



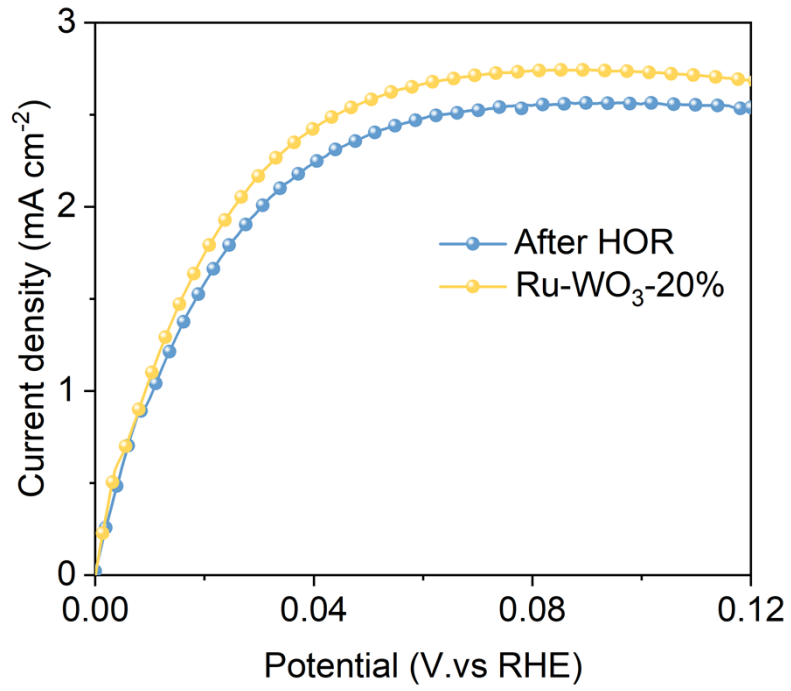
**Figure S2.** Corresponding cv curves for Ru-WO<sub>3</sub>, pure WO<sub>3</sub> and 20% Pt/C measured in 0.1 M KOH with H<sub>2</sub>.



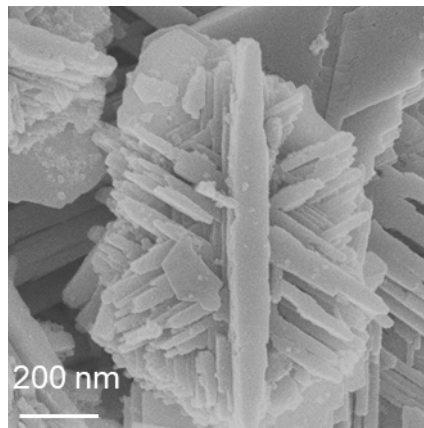
**Figure S3.** Polarization curve for pure WO<sub>3</sub> in 0.1 M KOH at a scan rate of 5 mV/s.



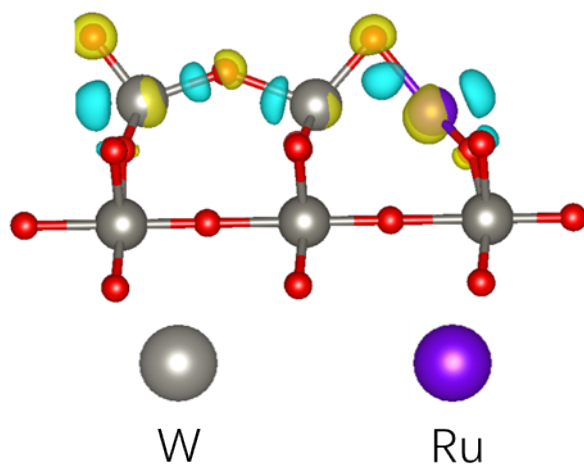
**Figure S4.** The kinetic current density and exchange current density normalized to the mass loading of Ru.



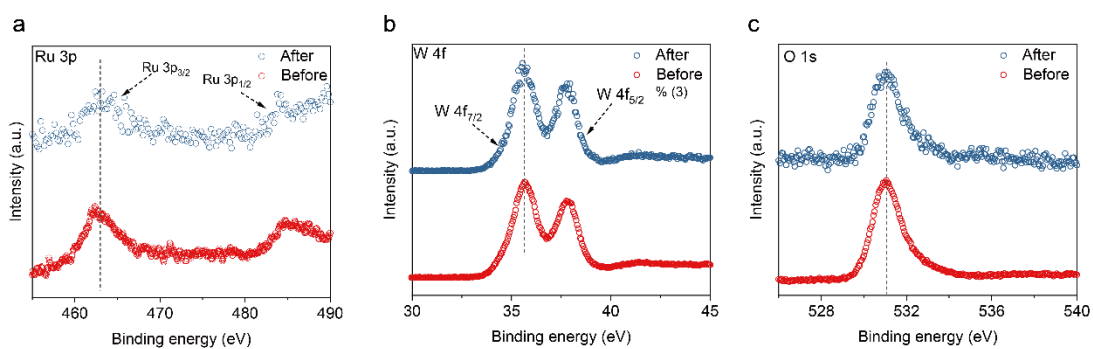
**Figure S5.** Polarization curves for Ru-WO<sub>3</sub>-20% before and after durability test at 0.1 V vs. RHE.



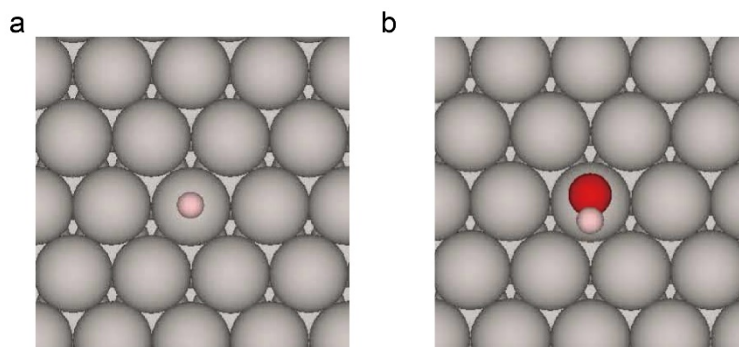
**Figure S6.** SEM image of Ru-WO<sub>3</sub>-20% after chronoamperometry test



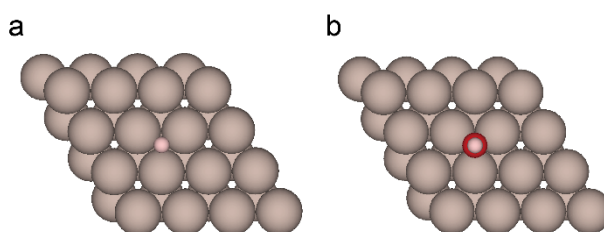
**Figure S7.** The electron density difference in Ru doped  $\text{WO}_3$ .



**Figure S8.** XPS spectra of Ru- $\text{WO}_3$ -20% after chronoamperometry test. a) Ru 3p XPS spectra. b) W 4f XPS spectra. c) O 1s XPS spectra.



**Figure S9.** a) H\* adsorption model on Pt (111) plane. b) OH\* adsorption model on Pt (111) plane.



**Figure S10.** a) H\* adsorption model on Ru (0001) plane. b) OH\* adsorption model on Ru (0001) plane.

**Table S1.** EDS results for Ru-WO<sub>3</sub>.

sample	W (at%)	Ru (at%)	Total
Ru-WO <sub>3</sub> -5%	97.36%	2.74%	100%
Ru-WO <sub>3</sub> -10%	94.76%	5.24%	100%
Ru-WO <sub>3</sub> -20%	90.61%	9.39%	100%
Ru-WO <sub>3</sub> -30%	89.24%	10.76%	100%

**Table S2.** Comparison of exchange current density ( $j_0$ ) between this work and reported Ru-based HOR catalysts

<b>Catalyst</b>	<b><math>j_0</math> (mA cm<sup>-2</sup>)</b>	<b>Electrolyte</b>	<b>Ref</b>
Ru-WO <sub>3</sub>	12.81	0.1 M KOH	This work
Ru Colloidosomes	2.86	0.1 M KOH	1
Ru <sub>c</sub> /NHCS	1.74	0.1 M KOH	2
Ru-Cr <sub>1</sub> (OH)-1.1	5.8	0.1 M KOH	3
O-RuNi@C-400	1.56	0.1 M KOH	4
HEA NSWs	6.42	0.1 M KOH	5
Ru/Ni-NiO@C	4.44	0.1 M KOH	6
Ru/Meso C	9.23	0.1 M KOH	7
Pt-Ru	2.98	0.1 M KOH	8
Ru modified Pt	5.52	0.1 M KOH	9
RuO <sub>2</sub> -Pt/C	4.77	0.1 M KOH	10

Ru <sub>7</sub> Ni <sub>3</sub> /C	1.8	0.1 M KOH	11
Ru-Ni diatomic sites	2.69	0.1 M KOH	12
RuP@NOC	2.64	0.1 M KOH	13
Ru@Pt <sub>2</sub> MLE	1.78	0.1 M KOH	14
RuRh-Co	1.91	0.1 M KOH	15

**Table S3.** Values of  $\Delta G_{H^*}$ ,  $E_{H^*}-E^*$ ,  $\Delta EZPE$  and  $T\Delta S$  for H\* adsorption.

Catalysts	$\Delta G_{H^*}$ (eV)	$E_{H^*}-E^*$ (eV)	$\Delta EZPE$ (eV)	$T\Delta S$ (eV)
Ru-WO <sub>3</sub>	-0.015	-0.297	0.077	-0.205
Pt (111)	-0.110	-0.342	0.047	-0.205
Ru (0001)	-0.296	-0.523	0.022	-0.205

**Table S4.** Values of  $\Delta G_{OH^*}$ ,  $E_{OH^*}-E^*$ ,  $\Delta EZPE$  and  $T\Delta S$  for H\* adsorption.

Catalysts	$\Delta G_{OH^*}$ (eV)	$E_{OH^*}-E^*$ (eV)	$\Delta EZPE$ (eV)	$T\Delta S$ (eV)
Ru-WO <sub>3</sub>	-0.582	-1.03	-0.020	-0.470
Pt (111)	1.450	1.038	-0.058	-0.470
Ru (0001)	0.230	-0.181	-0.058	-0.470

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