# **Supporting Information**

### Pd-Catalyzed Three-Component Decarboxylative Coupling Reactions Among Alkylidene Pyrazolones, Allyl Carbonates and Active Methylene Compounds

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#### **1. General Information**

Proton (<sup>1</sup>H) and carbon (<sup>13</sup>C) NMR spectra were recorded on 400 MHz instrument (400 MHz for <sup>1</sup>H NMR, 100 MHz for <sup>13</sup>C NMR) and calibrated using tetramethylsilane (TMS) as internal reference. High resolution mass spectra (HRMS) were recorded under electrospray ionization (ESI) conditions. The melting point of compounds was determined by a melting point instrument. Flash column chromatography was performed on silica gel (0.035-0.070 mm) by using compressed air. Thin layer chromatography (TLC) was carried out on 0.25 mm SDS silica gel coated glass plates (60F254). Eluted plates were visualized using a 254 nm UV lamp. Unless otherwise indicated, all reagents were commercially available and used without further purification. All solvents were distilled from the appropriate drying agents immediately before using. Alkylidene pyrazolones **1a–1g** were synthesized according to the reported procedures.<sup>1</sup> Allyl carbonates **2a–2h** were prepared according to literature procedures.<sup>2</sup>

### 2. Optimization of Coupling Reaction Conditions



solated yield.

Under the reaction conditions of  $Pd_2(dba)_3$  CHCl<sub>3</sub> and ligand (±)-L5 in THF at room temperature, we examined the different loading ratios of 1a/2a/3a for their effects on the decarboxylative coupling reaction of alkylidene pyrazolone 1a, allyl carbonate 2a and diethyl malonate 3a as depicted in Table S1.

# **3.** Control Reactions for Preliminary Investigations on Coupling Reaction Mechanism

**Table S2**. Screening of allyl carbonates<sup>a</sup>

$Ph$ $OR^{1}$ O $N$ $N$ $+$ $O$	+ $CO_2Et$	$\begin{array}{c} CHCl_3  EtO_2C \\ L5  Ph \\ O \\ O \\ N^{-N} \\ N \\ \end{array}$	
	3a	Ph <b>4aaa</b>	
Entry	$\mathbb{R}^1$	Time	$\text{Yield}^{b}(\%)$
1	$4-NO_2C_6H_4$	12 h	55
2	$4-MeC_6H_4$	8 h	73
3	4-MeOC <sub>6</sub> H <sub>4</sub>	0.5 h	83
4	Me	10 min	90

<sup>*a*</sup> Reactions were carried out with **1a** (0.1 mmol), **2** (0.15 mmol), **3a** (0.2 mmol), Pd<sub>2</sub>(dba)<sub>3</sub> CHCl<sub>3</sub> (0.005 mmol), ligand ( $\pm$ )-L5 (0.02 mmol) in THF (1.5 mL) at room temperature. <sup>*b*</sup> Isolated yield.

Under the reaction conditions of  $Pd_2(dba)_3$  CHCl<sub>3</sub> and ligand (±)-L5 in THF at room temperature, we explored in situ formed base  $R^1O^-$  for their effects on the decarboxylative coupling reaction of alkylidene pyrazolone 1a, allyl carbonates 2 and diethyl malonate 3a as depicted in Table S2.

### 4. X-Ray Crystallographic Analysis of Products 4aaa



Figure 1.X-ray single crystal structure of 4aaa (with thermal ellipsoids shown at the 50% probability level)

Identification code	4aaa
Empirical formula	$C_{27}H_{30}N_2O_5$
Formula weight	462.53
Temperature/K	113.15
Crystal system	monoclinic
Space group	I2/a
a/Å	22.1744(9)
b/Å	7.6768(3)
c/Å	29.6328(13)

$\alpha/^{\circ}$	90
β/°	98.428(4)
$\gamma/^{\circ}$	90
Volume/Å <sup>3</sup>	4989.9(4)
Z	8
$\rho_{calc}g/cm^3$	1.231
µ/mm <sup>-1</sup>	0.085
F(000)	1968.0
Crystal size/mm <sup>3</sup>	0.22  imes 0.2  imes 0.16
Radiation	MoKα ( $\lambda$ = 0.71073)
$2\Theta$ range for data collection/°	4.954 to 56.562
Index ranges	$-29 \le h \le 29, -9 \le k \le 10, -39 \le l \le 38$
Reflections collected	25550
Independent reflections	6133 [ $R_{int} = 0.0727, R_{sigma} = 0.0548$ ]
Data/restraints/parameters	6133/0/311
Goodness-of-fit on F <sup>2</sup>	1.024
Final R indexes [I>= $2\sigma$ (I)]	$R_1=0.0586,wR_2=0.1347$
Final R indexes [all data]	$R_1=0.0878,wR_2=0.1531$
Largest diff. peak/hole / e Å <sup>-3</sup>	0.80/-0.33



## 5. <sup>1</sup>H-<sup>13</sup>C HSQC and DEPT-135 of 4aea



### 6. NMR Spectra of Products of 4-6

























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### **7.** References

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