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Haber-independent, asymmetric synthesis of the marine alkaloid epi-leptosphaerin from a chitin-derived chiral pool synthon

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SUPPORTING INFORMATION

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Experimental Procedures

General

All reactions were performed under an atmosphere of dry nitrogen using oven dried glassware unless otherwise stated. Commercially available starting materials and reagents were used as received unless otherwise noted. Anhydrous solvents were used as supplied. Methanol, pyridine and N,N-dimethylformamide were dried using an LC Technology solvent purification system. Thin layer chromatography (TLC) was performed using F254 0.2 mm silica plates, followed by visualisation with UV irradiation at 254 nm, and staining with ethanolic vanillin solution. Flash column chromatography was performed using 63 -100 µm silica gel. Infrared (IR) spectra were recorded with an FT-IR spectrometer using a diamond ATR sampling accessory. Absorption maxima are expressed in wavenumbers (cm⁻¹). Melting points were collected using a Kofler hotstage apparatus and have not been corrected. NMR spectra were recorded at 298 K in D₂O, (CD₃)₂CO, CDCl₃ or pyridine – d_5 solution using a spectrometer operating at 400 MHz for ¹H nuclei and 100 MHz for 13 C nuclei. Chemical shifts are reported in ppm on the δ scale and were measured relative to the protium solvent that the sample was measured in: D₂O (δ 4.79 ppm) peak (¹H NMR) (CD₃)₂CO (δ 2.05 ppm) peak (¹H NMR) or δ 29.8 and 206.3 ppm (¹³C NMR), CDCl₃ (δ 7.26 ppm) peak (1 H NMR) or δ 77.2 ppm (13 C NMR) and pyridine – d_{5} (δ 8.74, 7.58 and 7.22 ppm) peak (¹H NMR) or δ 150.4, 135.9 and 123.9 ppm (¹³C NMR). ¹H NMR shift values are reported as chemical shift δ , relative integral, multiplicity (s, singlet, d, doublet, t, triplet, q, quartet, m, multiplet, b, broad), coupling constant (J, reported in Hertz [Hz]) and assignment. ¹³C NMR shift values are reported as chemical shift δ and assignment. Assignments were made with the aid of COSY, HSQC (edited), and HMBC experiments. High resolution mass spectra were obtained using electrospray ionisation on a microTOF QII mass spectrometer. Samples were dissolved in an appropriate solvent (DMSO, DCM, MeOH or MeCN) and diluted to a nominal concentration of 3 µg mL⁻¹ using either MeOH or MeCN, prior to direct infusion into the instrument. X-ray diffraction measurements of single crystals were performed on a Rigaku Oxford Diffraction XtaLAB-Synergy-S single-crystal diffractometer with a PILATUS 200K hybrid pixel array detector using Cu K α radiation ($\lambda = 1.54184$ Å). The data were processed with the

SHELX2018-3 and Olex2 software packages.¹⁻³ All non-hydrogen atoms were refined anisotropically. Hydrogen atoms were inserted at calculated positions or located directly and refined with a riding model or without restrictions. Mercury 2020.3.1⁴ was used to visualize the molecular structure. Crystal growth for X-ray crystallographic analysis purposes was achieved using slow vapour diffusion.

N-(5-((R)-1,2-dihydroxyethyl)-5-methoxy-2-oxo-2,5-dihydrofuran-3-yl)acetamide (5)

4,5,6,7-Tetrachloro-2',4',5',7'-tetraiodofluorescein (Rose Bengal) (11 mg, 0.0108 mmol, 1 mol %) was added to a solution of Di-HAF⁵ (200 mg, 1.08 mmol) in methanol (80 mL). The solution was purged with oxygen for 20 min. and cooled to -78 °C. The mixture was irradiated with a desk lamp under an oxygen atmosphere for 2 h, then warmed to room temperature. The solvent was evaporated *in vacuo* and the crude material purified by flash column chromatography eluting with dichloromethane-methanol (9.5:0.5) to give the *title compound* as a colourless oil. (40.4 mg, 0.175 mmol, 16%, *dr* 1:0.39); HRMS (ESI) *m/z* [M + Na]⁺, calcd for C₉H₁₃NO₆Na 254.0635, found 254.0628; v_{max}/cm⁻¹ (neat) 3283, 3040, 2944, 2844, 1767, 1694, 1657, 1533, 1371, 1313, 1153, 1037; δ_H (400 MHz, (CD₃)₂CO) 9.15 (1 H, br s, NH), 7.08 (0.4 H, s, ArH), 7.04 (1 H, s, Ar*H), 4.35 (1 H, m, OH), 4.22 (0.4 H, s, OH), 3.91 – 3.87 (1 H, m, C*H), 3.81 – 3.78 (1.4 H, m, 0.5 x 1 C*H₂, 1 x CH), 3.70 – 3.67 (0.4 H, m, 0.5 x 1 CH₂), 3.59 – 3.51 (1.4 H, m, 0.5 x 1 C*H₂, 0.5 x 1 C+H₂), 3.22 (1.2 H, s, OMe), 3.21 (3 H, s, OMe*), 2.19 (3 H, s, C*H₃), 2.18 (1.2 H, s, Me)); δ_C 100 MHz, (CD₃)₂CO 170.3 (C*), 170.2 (C), 167.7 (C*), 131.0 (C), 130.5 (C*), 124.1 (C*H), 122.9 (CH), 112.1 (C), 111.6 (C*), 75.5 (CH), 74.9 (C*H), 63.2 (CH₂), 62.9 (C*H₂), 51.5 (OMe), 51.3 (OMe*), 23.4 (Me + Me*), 1 x C not observed.

^{*} Denotes the major diastereomer.

N-(5-(2,2,3,3,8,8,9,9-Octamethyl-4,7-dioxa-3,8-disiladecan-5-yl)furan-3-yl)acetamide (6)

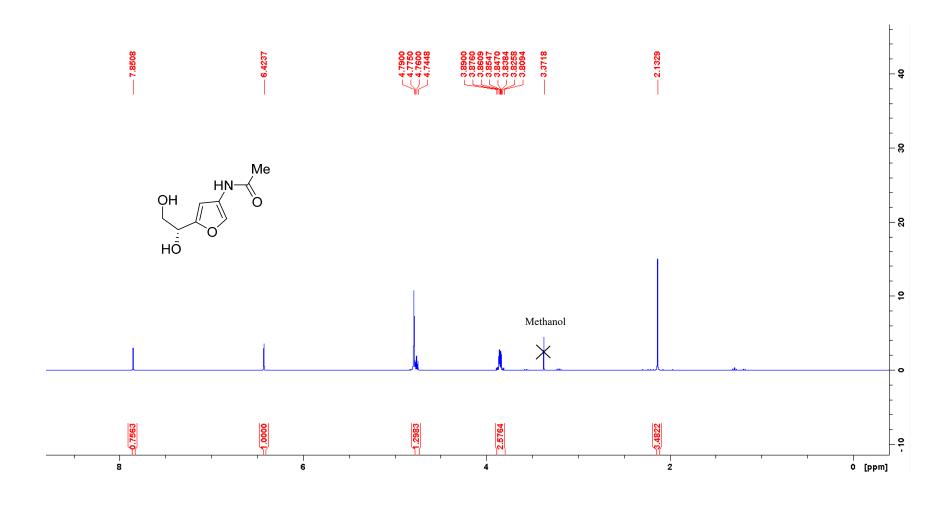
To a solution of Di-HAF (200 mg, 1.08 mmol) in dry DMF (8 mL) was added imidazole (485 mg, 7.13 mmol, 6.6 eq.) followed by *tert*-butyldimethylsilyl chloride (488 mg, 3.24 mmol, 3 eq.). The mixture was stirred at room temperature. for 2 h. The reaction mixture was diluted with ethyl acetate (70 mL) and washed with water (3 x 30 mL). The organic layer was dried (Na₂SO₄), filtered, concentrated *in vacuo* and the crude residue purified by flash chromatography on silica gel eluting with light petroleum/acetone (10:1) to give the *title compound* as a pale orange oil (272 mg, 0.657 mmol, 61%); $[\alpha]_D^{24}$ + 16.9 (*c* 1.0, MeOH); HRMS (ESI) *m/z* [M + Na]⁺, calcd for C₂₀H₃₉NO₄Si₂Na 436.2310, found 436.2308; v_{max}/cm⁻¹ (neat) 3283, 2954, 2930, 2886, 2858, 1661, 1573, 1472, 1464, 1362, 1252, 1125, 1093, 1006, 968, 831, 775, 666, 556; δ_H (400 MHz, (CD₃)₂CO) 9.11 (1 H, br s, NH), 7.92 (1 H, d, *J* 1.0, CH), 6.29 (1 H, s, CH), 4.70 (1 H, t, *J* 6.0, CH), 3.81 (1 H, dd, *J* 5.4, 4.7, CH of CH₂), 3.75 (1 H, dd, *J* 6.8, 3.4, CH of CH₂), 2.02 (3 H, s, Ac), 0.88 (9 H, s, 'Bu), 0.87 (9 H, s, 'Bu), 0.09 (3 H, s, SiMe), 0.05 (3 H, s, SiMe), 0.03 (3 H, s, SiMe), δ_C (100 MHz, (CD₃)₂CO) 167.8 (C), 154.4 (C), 131.3 (CH), 126.6 (C), 102.7 (CH), 70.9 (CH), 67.6 (CH₂), 26.3 (C(Me)₃), 26.2 (C(Me)₃), 23.0 (Ac), 18.9 (C), 18.8 (C), -3.2 (C), -4.6 (SiMe), -4.8 (SiMe), -5.2 (SiMe), -5.3 (SiMe), 3 x C not observed

epi-Leptosphaerin

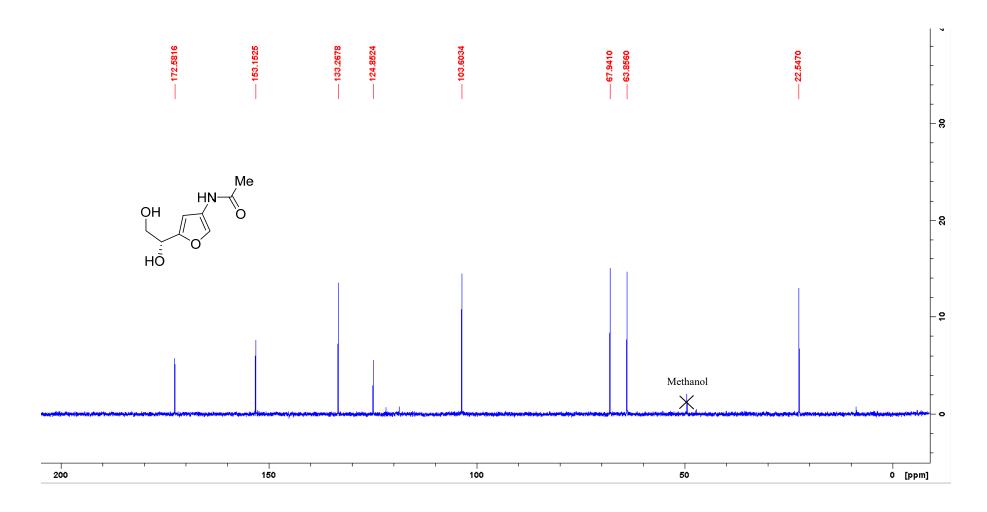
4,5,6,7-Tetrachloro-2',4',5',7'-tetraiodofluorescein (Rose Bengal) (5 mg, 0.00484 mmol, 1 mol %) was added to a solution of **6** (200 mg, 0.484 mmol) in methanol (80 mL). The solution was purged

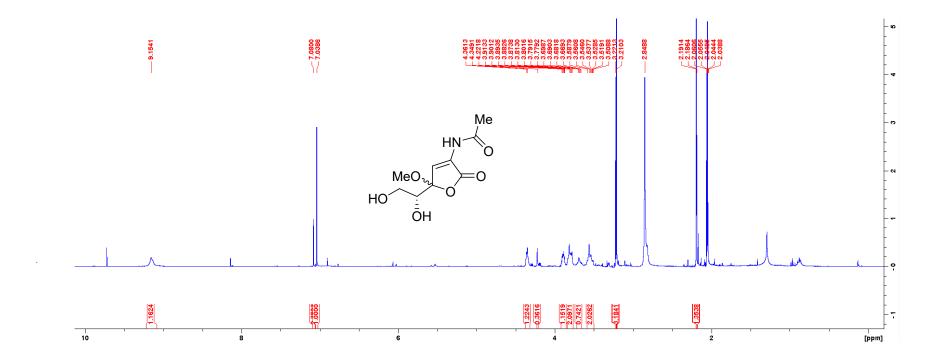
with oxygen for 30 minutes and cooled to -78 °C. The mixture was irradiated with a lamp under an oxygen atmosphere for 2 hours before it was allowed to warm to room temperature. The solvent was evaporated *in vacuo* and the remaining oil was dissolved in dichloromethane (16 mL). An oxalic acid solution (0.127M, 18.5 mL) was added and the mixture was stirred for 50 min. at room temperature. The aqueous phase was extracted with dichloromethane (3 x 40 mL) and the combined organic phases were dried (Na₂SO₄), filtered, and concentrated *in vacuo*. The catalyst was removed from the crude mixture using flash chromatography on silica with an eluent of light petroleum/acetone (3:1), giving the crude hydroxybutenolide 7 (~176 mg) as an orange oil that was used immediately in the next step.

The crude hydroxybutenolide (~176 mg, 0.38 mmol) was dissolved in methanol (9 mL) and the solution was cooled to 0 °C. Cerium trichloride heptahydrate (71.2 mg, 0.191 mmol, 0.5 mol eq.) and sodium borohydride (57.8 mg, 1.528 mmol, 4 mol eq.) were added and the resulting mixture was stirred for 1.5 h at 0 °C. Concentrated HCl was added dropwise until the solution turned a pale orange and had a pH of ~2, after which the mixture was stirred for a further 1 h at 0 °C. The mixture was allowed to warm to room temperature and then the solvent was evaporated *in vacuo*. The crude residue was purified using flash chromatography on silica gel eluting with dichloromethanemethanol (20:1) gave the *title compound* as a colourless solid (15 mg, 0.0746 mmol, 15.4% from 6); NMR data see Table S1; $[\alpha]_D^{24} - 32.9$ (c 0.07, H₂O) [lit⁶ $[\alpha]_D^{25} + 6.5$ (c 0.07, H₂O)]; mp 175.1 - 175.6 °C (lit⁶ mp not provided]; HRMS (ESI) m/z [M + Na]⁺, calcd for C₈H₁₁NO₅Na 224.0529, found 224.0529; v_{max}/cm^{-1} (neat) 3436, 3352, 3299, 3137, 1741, 1692, 1653, 1532, 1404, 1377, 1341, 1314, 1252, 1201, 1133, 1102, 1073, 1030, 918, 866, 833, 776;

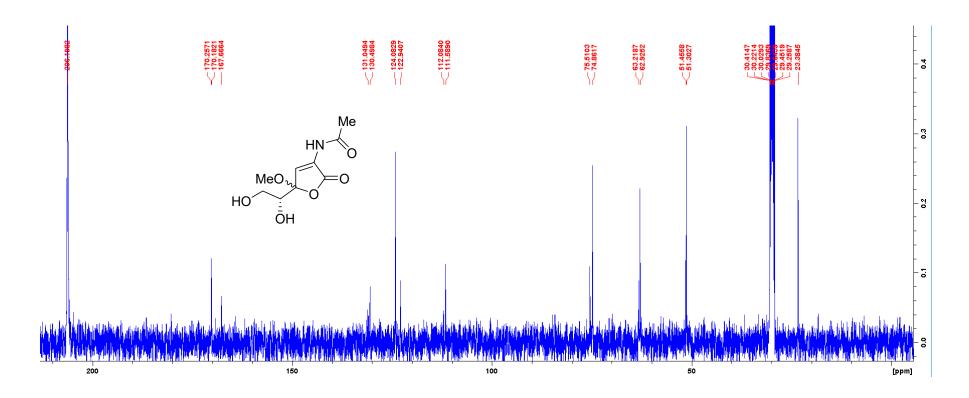


 ^{13}C NMR spectrum of Di-HAF (100 MHz, D₂O)

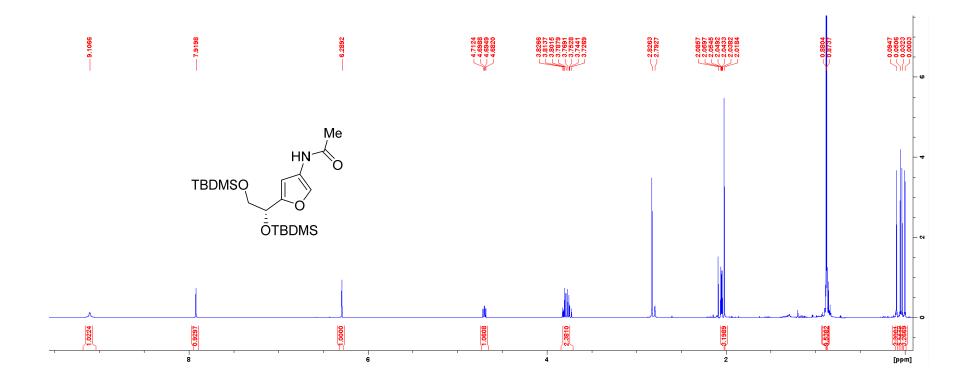




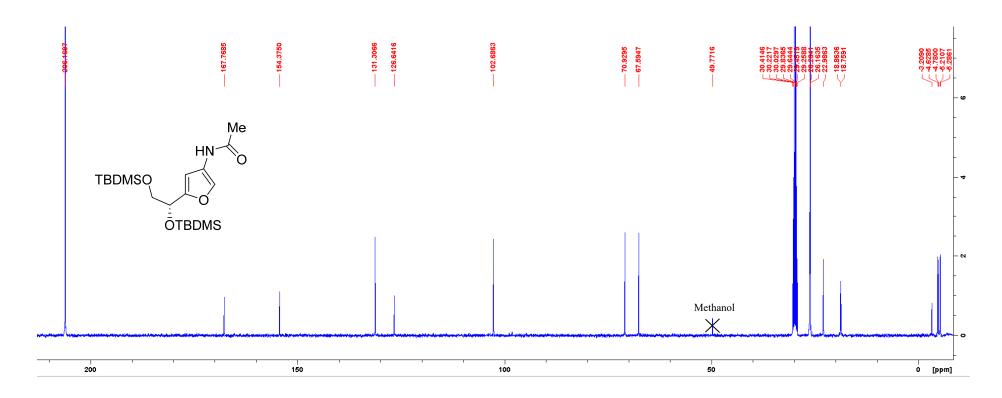
¹³C NMR spectrum of **5** (100 MHz, (CD₃)₂CO)



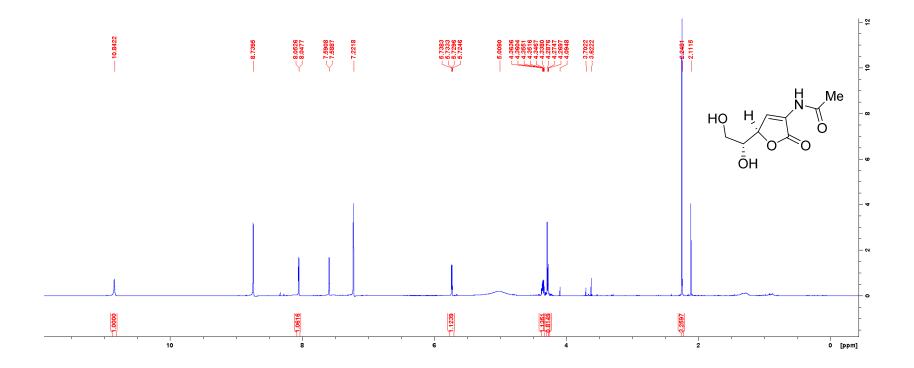
¹H NMR spectrum of **6** (400 MHz, (CD₃)₂CO)



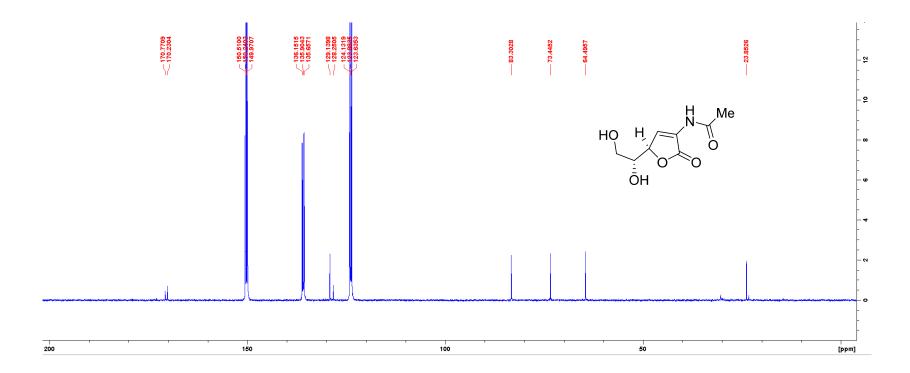
¹³C NMR spectrum of **6** (100 MHz, (CD₃)₂CO)



¹H NMR spectrum of *epi*-leptosphaerin (400 MHz, pyridine-*d*₅)



¹³C NMR spectrum of *epi*-leptosphaerin (100 MHz, pyridine-*d*₅)



¹ H NMR (pyridine- <i>ds</i>) ppm				¹³ C NMR (pyridine- <i>ds</i>) ppm			
Isolated ⁶	Synthesised (this work)	Assignment	Difference	Isolated ⁶	Synthesised (this work)	Assignment	Difference
10.86 (1 H, s, NH)	10.84 (1 H, s, NH)	NH	0.02	170.8	170.8	C-1	0
8.05 (1H, d, J 2.0, CH)	8.05 (1 H, d, J 2.0, CH)	H-3	0	170.2	170.2	C-7	0
5.73 (1H, dd, <i>J</i> 3.3, 2.0, CH)	5.73 (1 H, dd, <i>J</i> 2.0, 1.5 CH)	H-4	0	129.1	129.1	C-3	0
4.35 (1H, bs, CH)	4.35 (1 H, m, CH)	H-5	0	128.3	128.3	C-2	0
4.29 (2H, m, CH ₂)	4.29 (1 H, s 1 x CH of CH ₂)	1 x H of H ₂ -6	0	83.3	83.3	C-4	0
2.25 (3 H, s, Me)	4.27 (1 H, d, <i>J</i> 2.0, 1 x CH of CH ₂)	1 x H of H ₂ -6	0.02	73.4	73.4	C-5	0
	2.25 (3 H, s, Me)	H ₃ -8	0	64.5	64.5	C-6	0
				23.8	23.9	C-8	-0.1

Compound	<i>epi</i> -leptosphaerin
CCDC number	2177368
Empirical formula	C ₈ H ₁₁ NO ₅
Formula weight	201.18
Temperature/K	100.0(3)
Crystal system	orthorhombic
Space group	P2 ₁ 2 ₁ 2 ₁
a/Å	6.0853(2)
b/Å	6.9127(2)
c/Å	21.3235(4)
α/°	90
β/°	90
<u>y</u> /°	90
Volume/Å ³	896.99(4)
Z	4
pcalcg/cm ³	1.490
μ/mm ⁻¹	1.079
F(000)	424.0
Crystal size/mm ³	$0.12 \times 0.1 \times 0.1$
Radiation	Cu K α ($\lambda = 1.54184$)
2Θ range for data collection/°	8.294 to 135.466
Index ranges	$-7 \le h \le 7, -8 \le k \le 8, -25 \le l \le 9$
Reflections collected	5090
Independent reflections	$[1620 \ [R_{int} = 0.0343, R_{sigma} = 0.0339]]$
Data/restraints/parameters	1620/0/137
Goodness-of-fit on F ²	1.058
Final R indexes [I>=2σ (I)]	$R_1 = 0.0294, wR_2 = 0.0731$
Final R indexes [all data]	$R_1 = 0.0307, wR_2 = 0.0741$
Largest diff. peak/hole / e Å ⁻³	0.17/-0.19
Flack parameter	0.08(14)

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