

Synthesis of push-pull-activated ynl ethers and their evaluation in the bioorthogonal hydroamination reaction

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Massachusetts, 02115***Supporting Information****Table of Contents**

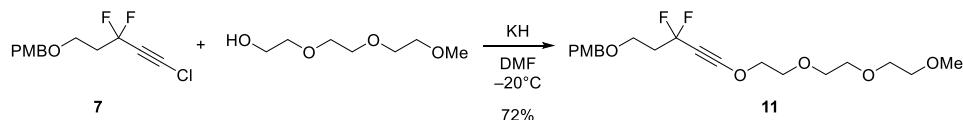
<u>General Chemical Procedures</u>	S2
<u>General Chemical Materials</u>	S2
<u>General Chemical Instrumentation</u>	S2
<u>Synthetic Procedures and Characterizations</u>	S3
<u>Stability Study</u>	S8
<u>Kinetic Study</u>	S14
<u>Computational Details</u>	S16
<u>References</u>	S23
<u>Copies of NMR Spectra</u>	S24

General Chemical Procedures. All reactions were conducted in flame-dried round-bottom flasks under a positive pressure of nitrogen unless otherwise stated. Gas-tight syringes with stainless steel needles or cannulae were used to transfer air- and moisture-sensitive liquids. Flash column chromatography was performed using granular silica gel (60-Å pore size, 40–63 µm, Silicycle). Analytical thin layer chromatography (TLC) was performed using glass plates pre-coated with 0.25 mm silica gel impregnated with a fluorescent indicator (254 nm, Silicycle). TLC plates were visualized by exposure to short wave ultraviolet light (254 nm) and/or an aqueous solution of potassium permanganate (KMnO_4). Organic solutions were concentrated at 20 °C on rotary evaporators capable of achieving a minimum pressure of ~2 torr unless otherwise stated. Room temperature is defined as 21.5 ± 2.5 °C. Reaction heating was performed using a UCON fluid heating bath.

General Chemical Materials. All solvents were purchased from Fisher Scientific or Sigma–Aldrich. Unless otherwise stated chemical reagents were purchased from Fisher Scientific, Sigma–Aldrich, Alfa Aesar, Oakwood Chemical, Acros Organics, Combi-Blocks, or TCI America. CMA refers to a solution of 80:18:2 v/v/v chloroform:methanol:ammonium hydroxide (28–30% ammonia solution). Chloroform used in CMA solutions and as co-elutents in silica gel column chromatography were stabilized with 0.75% v/v ethanol. Chloroform used in all hydroamination reactions were stabilized with pentene.

General Chemical Instrumentation. Proton nuclear magnetic resonance (^1H NMR) spectra, recorded with a 500 MHz Avance III Spectrometer with multi-nuclear Smart probe, are reported in parts per million on the δ scale, and are referenced from the residual protium in the NMR solvent (CDCl_3 : δ 7.24, CD_3OD : δ 3.31, D_2O : δ 4.79, CD_3CN : δ 1.94). Data are reported as follows: chemical shift [multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, dd = doublet of doublet, dt = doublet of triplet, dq = doublet of quartet, td = triplet of doublet, tt = triplet of triplet, ddd = doublet of doublet of doublet), coupling constant(s) in Hertz, integration, assignment]. Carbon-13 nuclear magnetic resonance (^{13}C NMR) spectra, recorded with a 500 MHz Avance III Spectrometer with multi-nuclear Smart probe, are referenced from the carbon resonances of the solvent (CDCl_3 : δ 77.23, CD_3OD : δ 49.15). Fluorine-19 nuclear magnetic resonance (^{19}F NMR) is calibrated from the fluorine resonances of the benzotrifluoride (CDCl_3 : δ -62.76, CD_3OD : δ -64.24, CD_3CN : δ -63.22). Data are reported as follows: chemical shift (assignment). Infrared data (IR) were obtained with a Cary 630 Fourier transform infrared spectrometer equipped with a diamond ATR objective and are reported as follows: frequency of absorption (cm^{-1}), intensity of absorption (s = strong, m = medium, w = weak, br = broad). High resolution mass spectra (HRMS) were recorded on a Q Exactive™ Plus Hybrid Quadrupole-Orbitrap™ Mass Spectrometer using an electrospray ionization (ESI), atmospheric pressure ionization (API), or electron ionization (EI) source. Automated C_{18} reverse phase chromatography was performed using a Isolera One (Biotage) purification system. High performance liquid chromatography (HPLC) purification was performed using an Agilent 1260 Infinity system.

Synthetic Procedures and Characterizations



14,14-difluoro-18-(4-methoxyphenyl)-2,5,8,11,17-pentaoxaoctadec-12-yne (11):

Chloroalkyne **7** (28.7 mg, 105 μmol , 1 equiv) and potassium hydride (8.42 mg, 210 μmol , 2.00 equiv) were sequentially added to a solution of triethylene glycol monomethyl ether (17.8 μL , 110 μmol , 1.05 equiv) in *N,N*-dimethylformamide (1.00 mL) at -20°C . After 1 h, the reaction was quenched with saturated aqueous ammonium chloride solution (1 mL) at -20°C then diluted with diethyl ether (15 mL) and washed with saturated aqueous ammonium chloride solution (10 mL). The aqueous layer was extracted with diethyl ether (15 mL). The combined organic layers were washed with brine (20 mL), dried over anhydrous sodium sulfate, filtered, and concentrated under reduced pressure. The resulting crude mixture was purified by flash column chromatography on silica gel (eluent: 20% acetone in hexanes) to provide alkyne **11** (30.3 mg, 72%) as a colorless oil.

^1H NMR (500 MHz, CDCl_3 , 25 $^\circ\text{C}$) δ 7.24 (d, $J = 8.5$ Hz, 1H), 6.86 (d, $J = 8.6$ Hz, 1H), 4.43 (s, 2H), 4.23–4.19 (m, 2H), 3.78 (s, 3H), 3.76–3.72 (m, 2H), 3.69–3.60 (m, 8H), 3.55–3.52 (m, 2H), 3.36 (s, 3H), 2.34 (tt, $J = 7.1$ Hz, 2H).

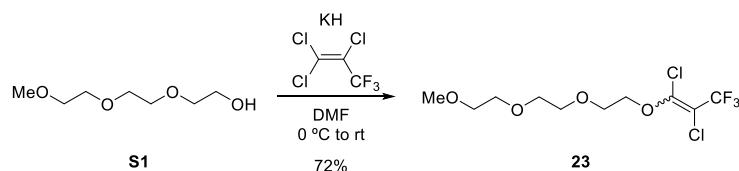
^{13}C NMR (126 MHz, CD_3OD , 25 $^\circ\text{C}$) δ 161.0, 131.6, 130.7, 116.4 (t, $J = 230.1$ Hz), 114.9, 96.7 (t, $J = 7.1$ Hz), 80.15, 73.8, 73.1, 71.8, 71.7, 71.6, 69.7, 65.6 (t, $J = 4.6$ Hz), 59.2, 55.8, 41.4 (t, $J = 27.5$ Hz), 34.2 (t, $J = 42.8$ Hz).

^{19}F NMR (471 MHz, CDCl_3 , 25 $^\circ\text{C}$) δ –76.4.

FTIR (thin film) cm^{-1} : 2873 (b), 2274 (m), 1513 (m), 1245 (m), 1088 (s), 1029 (s), 948 (m), 820 (m).

HRMS (ESI) (m/z) calc'd for $\text{C}_{20}\text{H}_{29}\text{F}_2\text{O}_6$ [$\text{M}+\text{H}]^+$: 403.1927, found: 403.1920.

TLC (20% acetone in hexanes) R_f : 0.46 (KMnO₄).



12,13-dichloro-14,14,14-trifluoro-2,5,8,11-tetraoxatetradec-12-ene (23):

1,1,2-Trichloro-3,3,3-trifluoropropene (857 μL , 7.00 mmol, 2.00 equiv) and potassium hydride (421 mg, 10.5 mmol, 3.00 equiv) were sequentially added to a solution of triethylene glycol monomethyl ether (563 μL , 3.50 mmol, 1 equiv) in *N,N*-dimethylformamide (14 mL) at 0 $^\circ\text{C}$ and allowed to warm to room temperature. After 7.5 h, the reaction was quenched with water (4 mL) at 0 $^\circ\text{C}$ then diluted with diethyl ether (100 mL) and washed with water (40 mL). The aqueous layer was extracted with diethyl ether (3×25 mL). The combined organic layers were washed with brine (50 mL), dried over anhydrous sodium sulfate, filtered, and concentrated under reduced pressure. The resulting crude residue was purified by flash column chromatography on silica gel (eluent: 5% acetone in hexanes) to provide enol ether **23** (821 mg, 72%, *E:Z* 2.15:1) as a colorless oil.

¹H NMR (500 MHz, CDCl₃, 25 °C) [E] δ 4.32–4.27 (m, 2H), 3.79–3.75 (m, 2H), 3.69–3.61 (m, 6H), 3.55–3.51 (m, 2H), 3.36 (s, 3H). [Z] δ 4.25–4.21 (m, 2H), 3.76–3.72 (m, 2H), 3.65–3.62 (m, 6H), 3.55–3.51 (m, 2H), 3.36 (s, 3H).

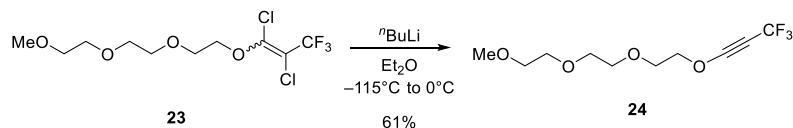
¹³C NMR (126 MHz, CDCl₃, 25 °C) [E] δ 146.0 (q, *J* = 2.5 Hz), 121.3 (q, *J* = 271.0 Hz), 102.2 (q, *J* = 39.4 Hz), 72.1, 71.3, 71.2, 70.8, 70.8, 69.3, 59.2. [Z] 150.6 (q, *J* = 2.6 Hz), 121.0 (q, *J* = 272.7 Hz), 106.5 (q, *J* = 38.3 Hz), 72.9, 72.1, 71.0, 70.8, 70.8, 69.1, 59.2.

¹⁹F NMR (471 MHz, CDCl₃, 25 °C) [E] δ -59.9. [Z] δ -60.8.

FTIR (thin film) cm^{-1} : [E] 2877 (b), 1625 (m), 1297 (w), 1207 (w), 1122 (s), 1025 (m), 719 (w). [Z] 2877 (b), 1628 (m), 1301 (s), 1189 (m), 1125 (s), 1036 (m), 693 (m).

HRMS (ESI) (*m/z*) calc'd for C₁₀H₁₉Cl₂F₃NO₄ [M+NH₄]⁺: 344.0638, 346.0608, found: 344.0634, 346.0603.

TLC (20% acetone in hexanes) [E] Rf: 0.44 (KMnO₄). [Z] Rf: 0.46 (KMnO₄).



14,14,14-trifluoro-2,5,8,11-tetraoxatetradec-12-yne (24):

n-Butyllithium (160 μ L, 399 μ mol, 1.30 equiv, 2.5 M in hexanes) was added dropwise to a solution of enol ether **22** (100 mg, 307 μ mol, 1 equiv) in diethyl ether (10 mL) at -115°C . The reaction flask was removed from the cooling bath then allowed to warm to 0°C over 20 min. The reaction was then quenched with saturated aqueous ammonium chloride solution (1 mL) then diluted with dichloromethane (60 mL) and washed with saturated aqueous ammonium chloride solution (20 mL). The aqueous layer was extracted with dichloromethane (3×20 mL). The combined organic layers were washed with brine (40 mL), dried over anhydrous sodium sulfate, filtered, and concentrated under reduced pressure. The resulting crude mixture was purified by flash column chromatography on silica gel (eluent: gradient, 20–40% ethyl acetate in hexanes) to provide alkyne **24** (47.7 mg, 61%) as a colorless oil.

¹H NMR (500 MHz, CD₃OD, 25 °C) δ 4.46–4.43 (m, 2H), 3.83–3.80 (m, 2H), 3.68–3.63 (m, 6H), 3.57–3.53 (m, 2H), 3.36 (s, 3H).

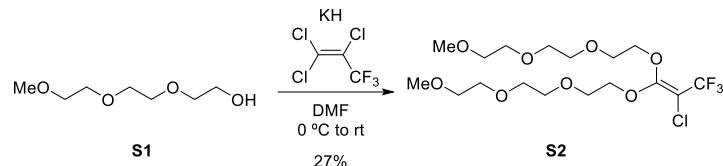
¹³C NMR (126 MHz, CD₃OD, 25 °C) δ 117.8 (q, *J* = 253.0 Hz), 96.0 (q, *J* = 6.3 Hz), 81.5, 73.1, 71.9, 71.7, 71.6, 69.7, 59.2, 28.9 (q, *J* = 53.8 Hz).

¹⁹F NMR (471 MHz, CDCl₃, 25 °C) δ -46.0.

FTIR (thin film) cm^{-1} : 2878 (b), 2281 (m), 1115 (s), 1077 (s), 939.3 (w), 828 (w).

HRMS (ESI) (*m/z*) calc'd for C₁₀H₁₅F₃NaO₄ [M+Na]⁺: 279.0815, found: 279.0810.

TLC (40% ethyl acetate in hexanes) R_f : 0.25 ($KMnO_4$).



12-(1-chloro-2,2,2-trifluoroethylidene)-2,5,8,11,13,16,19,22-octaoxatricosane (S2):

1,1,2-Trichloro-3,3,3-trifluoropropene (30.5 μ L, 249 μ mol, 1 equiv) and potassium hydride (60.0 mg, 1.49 mmol, 6.00 equiv) were sequentially added to a solution of triethylene glycol monomethyl ether (80.5 μ L, 498 μ mol, 2.00 equiv) in *N,N*-dimethylformamide (2 mL) at 0 °C and allowed to warm to room temperature. After 7.5 h, the reaction was quenched with water (1 mL) at 0 °C then diluted with diethyl ether (20 mL) and washed with water (10 mL). The aqueous layer was extracted with diethyl ether (3 \times 10 mL). The combined organic layers were washed with brine (20 mL), dried over anhydrous sodium sulfate, filtered, and concentrated under reduced pressure. The resulting crude residue was purified by flash column chromatography on silica gel (eluent: 20% acetone in hexanes) to provide the enol ether S2 (62.0 mg, 27%) as a colorless oil.

¹H NMR (500 MHz, CDCl₃, 25 °C) δ 4.23–4.19 (m, 2H), 4.13–4.08 (m, 2H), 3.75–3.71 (m, 2H), 3.71–3.67 (m, 2H), 3.67–3.59 (m, 12H), 3.54–3.50 (m, 4H), 3.36 (s, 3H), 3.35 (s, 3H).

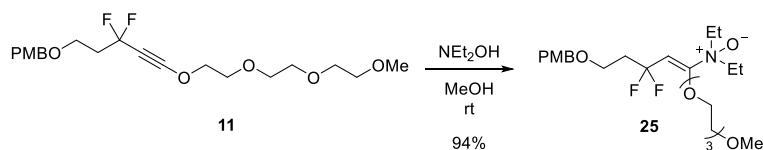
¹³C NMR (126 MHz, CD₃OD, 25 °C) δ 161.9 (q, *J* = 2.2 Hz), 124.2 (q, *J* = 268.1 Hz), 89.0 (q, *J* = 111.0 Hz), 73.1, 73.1, 71.9, 71.8, 71.7, 71.7, 71.6, 71.6, 71.5, 70.9, 70.8, 59.2, 59.2.

¹⁹F NMR (471 MHz, CDCl₃, 25 °C) δ -60.4.

FTIR (thin film) cm^{-1} : 2874 (b), 1655 (m), 1454 (w), 1316 (m), 1103 (s), 1062 (s), 1025 (2), 980 (m), 850 (w).

HRMS (ESI) (*m/z*) calc'd for C₁₇H₃₀ClF₃NaO₈ [M+Na]⁺: 477.1474, 479.1444; found: 477.1465, 479.1430.

TLC (40% acetone in hexanes) R_f : 0.45 ($KMnO_4$).



(Z)-N,N-diethyl-14,14-difluoro-18-(4-methoxyphenyl)-2,5,8,11,17-pentaoxaoctadec-12-en-12-amine oxide (25):

N,N-Diethylhydroxylamine (5.60 μ L, 55.0 μ mol, 1.30 equiv) was added to a solution of alkyne **11** (17.0 mg, 42.3 μ mol, 1 equiv) in methanol (500 μ L) at room temperature. After 1 h, the reaction mixture was diluted with dichloromethane (5 mL) and immediately purified by flash chromatography on silica gel (eluent: gradient, 3 \rightarrow 30% CMA in chloroform) to provide enamine *N*-oxide **25** (19.4 mg, 94%) as a colorless oil.

¹H NMR (500 MHz, CD₃OD, 25 °C) δ 7.27 (d, *J* = 8.6 Hz, 2H), 6.89 (d, *J* = 8.6 Hz, 2H), 6.44 (t, *J* = 15.1, 1H), 4.44 (s, 2H), 4.39–4.37 (m, 2H), 3.83–3.76 (m, 2H), 3.78 (s, 3H), 3.73–3.71 (m, 2H), 3.66–3.59 (m, 8H), 3.54–3.51 (m, 2H), 3.35 (s, 3H), 3.28–3.23 (m, 2H), 2.38 (tt, *J* = 15.6, 6.5 Hz, 2H), 1.19 (t, *J* = 7.1 Hz, 6H).

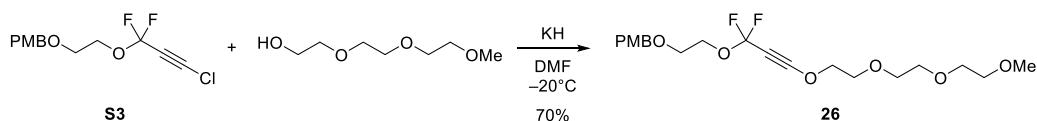
¹³C NMR (126 MHz, CD₃OD, 25 °C) δ 161.0, 159.7 (t, *J* = 4.6 Hz), 131.5, 130.7, 122.0 (t, *J* = 238.3 Hz), 114.9, 103.9 (t, *J* = 26.9 Hz), 78.2 (t, *J* = 6.2 Hz), 73.9, 73.2, 71.7, 71.6, 71.6, 71.5, 70.7, 64.8 (t, *J* = 5.3 Hz), 59.2, 55.8, 40.1 (t, *J* = 26.2 Hz), 8.7.

¹⁹F NMR (471 MHz, CD₃OD, 25 °C) δ -87.2.

FTIR (thin film) cm^{-1} : 2874 (b), 1696 (m), 1513 (m), 1245 (m), 1100 (s), 820 (m).

HRMS (ESI) (*m/z*) calc'd for C₂₄H₄₀F₂NO₇ [M+H]⁺: 492.2767, found: 492.2761.

TLC (25% CMA in chloroform) R_f : 0.30 ($KMnO_4$).



14,14-difluoro-19-(4-methoxyphenyl)-2,5,8,11,15,18-hexaoxanonadec-12-yne (26):

Chloroalkyne **S3** (40.1 mg, 138 μmol , 1 equiv) and potassium hydride (11.1 mg, 276 μmol , 2.00 equiv) were sequentially added to a solution of triethylene glycol monomethyl ether (23.3 μL , 145 μmol , 1.05 equiv) in *N,N*-dimethylformamide (1.30 mL) at -20°C . After 1 h, the reaction was quenched with saturated aqueous ammonium chloride solution (1 mL) at -20°C then diluted with diethyl ether (15 mL) and washed with saturated aqueous ammonium chloride solution (10 mL). The aqueous layer was extracted with diethyl ether (15 mL). The combined organic layers were washed with brine (20 mL), dried over anhydrous sodium sulfate, filtered, and concentrated under reduced pressure. The resulting crude mixture was purified by flash column chromatography on silica gel (eluent: 20% acetone in hexanes) to provide alkyne **26** (40.2 mg, 70%) as a colorless oil.

¹H NMR (500 MHz, CD₃OD, 25 °C) δ 7.27 (d, *J* = 6.7 Hz, 2H), 6.90 (d, *J* = 8.6 Hz, 2H), 4.49 (s, 2H), 4.35–4.32 (m, 2H), 4.01–3.99 (m, 2H), 3.79 (s, 3H), 3.79–3.76 (m, 2H), 3.66–3.60 (m, 8H), 3.55–3.52 (m, 2H), 3.31 (s, 3H).

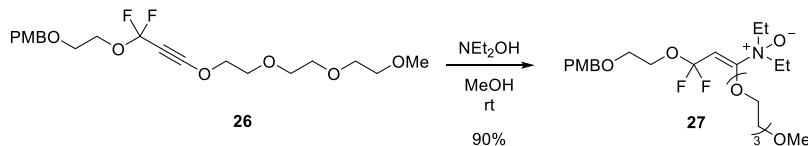
¹³C NMR (126 MHz, CD₃OD, 25 °C) δ 161.0, 131.5, 130.7, 117.8 (t, *J* = 239.0 Hz), 114.9, 94.4 (t, *J* = 6.1 Hz), 80.6, 73.9, 73.1, 71.8, 71.7, 71.6, 69.7, 69.1, 66.1 (t, *J* = 3.0 Hz), 59.2, 55.8, 31.8 (t, *J* = 56.2 Hz).

¹⁹F NMR (471 MHz, CDCl₃, 25 °C) δ -51.0.

FTIR (thin film) cm^{-1} : 2874 (b), 2277 (m), 1513 (m), 1163 (m), 1096 (s), 1003 (s), 816 (m).

HRMS (ESI) (*m/z*) calc'd for C₂₀H₂₉F₂O₇ [M+H]⁺: 419.1876, found: 419.1868.

TLC (20% acetone in hexanes) Rf: 0.45 (KMnO₄).



(Z)-N,N-diethyl-14,14-difluoro-19-(4-methoxyphenyl)-2,5,8,11,15,18-hexaoxanonadec-12-en-12-amine oxide (27):

N,N-Diethylhydroxylamine (7.70 μ L, 79.6 μ mol, 1.30 equiv) was added to a solution of alkyne **26** (25.6 mg, 61.2 μ mol, 1 equiv) in methanol (500 μ L) at room temperature. After 1 h, the reaction mixture was diluted with dichloromethane (5 mL) and immediately purified by flash chromatography on silica gel (eluent: gradient, 3 \rightarrow 30% CMA in chloroform) to provide enamine *N*-oxide **27** (28.0 mg, 90%) as a colorless oil.

^1H NMR (500 MHz, CD₃OD, 25 °C) δ 7.26 (d, J = 8.7 Hz, 2H), 6.90 (d, J = 8.7 Hz, 2H), 6.38 (t, J = 8.2, 1H), 4.47 (s, 2H), 4.42–4.40 (m, 2H), 4.11–4.09 (m, 2H), 3.79 (s, 3H), 3.82–3.75 (m, 2H), 3.71–3.69 (m, 2H), 3.60–3.57 (m, 4H), 3.56–3.53 (m, 2H), 3.53–3.51 (m, 2H), 3.49–3.47 (m, 2H), 3.35 (s, 3H), 3.28–3.23 (m, 2H), 1.2 (t, J = 7.1 Hz, 6H).

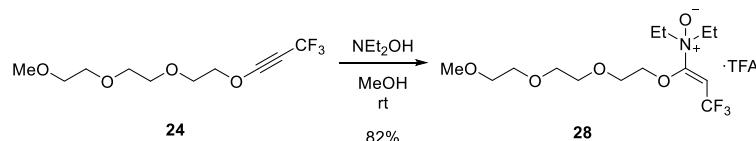
^{13}C NMR (126 MHz, CD₃OD, 25 °C) δ 161.0, 160.8 (t, J = 7.5 Hz), 131.6, 130.7, 123.5 (t, J = 253.0 Hz), 115.0, 100.5 (t, J = 41.3 Hz), 77.7 (t, J = 3.5 Hz), 73.8, 73.2, 71.7, 71.5, 71.5, 70.5, 69.1, 64.7 (t, J = 5.5 Hz), 64.1, 59.2, 55.9, 8.6.

^{19}F NMR (471 MHz, CD₃OD, 25 °C) δ -61.8.

FTIR (thin film) cm⁻¹: 2874 (b), 1703 (w) 1513 (w), 1345 (m), 1245 (m), 1103 (s), 824 (w).

HRMS (ESI) (*m/z*) calc'd for C₂₄H₄₀F₂NO₈ [M+H]⁺: 508.2717, found: 508.2709.

TLC (20% CMA in chloroform) R_f: 0.15 (KMnO₄).



(Z)-*N,N*-diethyl-14,14,14-trifluoro-2,5,8,11-tetraoxatetradec-12-en-12-amine oxide trifluoroacetate (28):

N,N-Diethylhydroxylamine (9.90 μ L, 96.3 μ mol, 1.35 equiv) was added to a solution of alkyne **24** (18.3 mg, 71.3 μ mol, 1 equiv) in methanol (700 μ L) at room temperature. After 1 h, the reaction mixture was diluted with water (150 μ L), and the resulting solution was purified by automated C₁₈ reverse phase column chromatography (30 g C₁₈ silica gel, 25 μ m spherical particles, eluent: 20% MeCN/H₂O + 0.1% TFA (1 CV), gradient 20 \rightarrow 80% MeCN/H₂O + 0.1% TFA (12 CV)). Fractions containing the desired compound were combined, and the solvent was removed under reduced pressure at 0 °C with a rotary evaporator to provide enamine *N*-oxide **28** (45.8 mg, 28:TFA of 1:3.8, 82%) as a white solid.

^1H NMR (500 MHz, CD₃OD, 25 °C) δ 6.39 (q, J = 8.2 Hz, 1H), 4.60–4.58 (m, 2H), 4.28 (dq, J = 14.1, 7.1 Hz, 2H), 3.82 (dq, J = 14.1, 7.1 Hz, 2H), 3.81–3.79 (m, 2H), 3.68–3.66 (m, 2H), 3.64–3.59 (m, 4H), 3.55–3.52 (m, 2H), 3.36 (s, 3H), 1.36 (t, J = 7.1 Hz, 6H).

^{13}C NMR (126 MHz, CD₃OD, 25 °C) δ 156.6 (q, J = 6.1 Hz), 123.7 (q, J = 268.3 Hz), 99.3 (q, J = 38.6 Hz), 80.9 (q, J = 4.1 Hz), 73.2, 71.5, 71.5, 71.4, 70.0, 64.4, 59.2, 8.4 [TFA δ 160.9 (q, J = 40.0 Hz), 117.2 (q, J = 285.0 Hz)].

^{19}F NMR (471 MHz, CD₃OD, 25 °C) δ -56.3 [TFA δ -77.5].

FTIR (thin film) cm⁻¹: 2881 (b), 1782 (w), 1707 (w), 1357 (w), 1170 (s), 1133 (s), 947 (w).

HRMS (ESI) (*m/z*) calc'd for C₁₄H₂₇F₃NO₅ [M+H]⁺: 346.1836, found: 346.1832.

TLC (30% CMA in chloroform) R_f: 0.20 (KMnO₄).

Stability Study

Stability experiments for alkynes **11**, **24**, and **26** were carried out in 50% CD₃CN/PBS (pH 7.0) with or without glutathione (2 mM) at room temperature. Reactions were monitored via ¹⁹F NMR spectroscopy using α,α,α -benzotrifluoride as an internal standard.

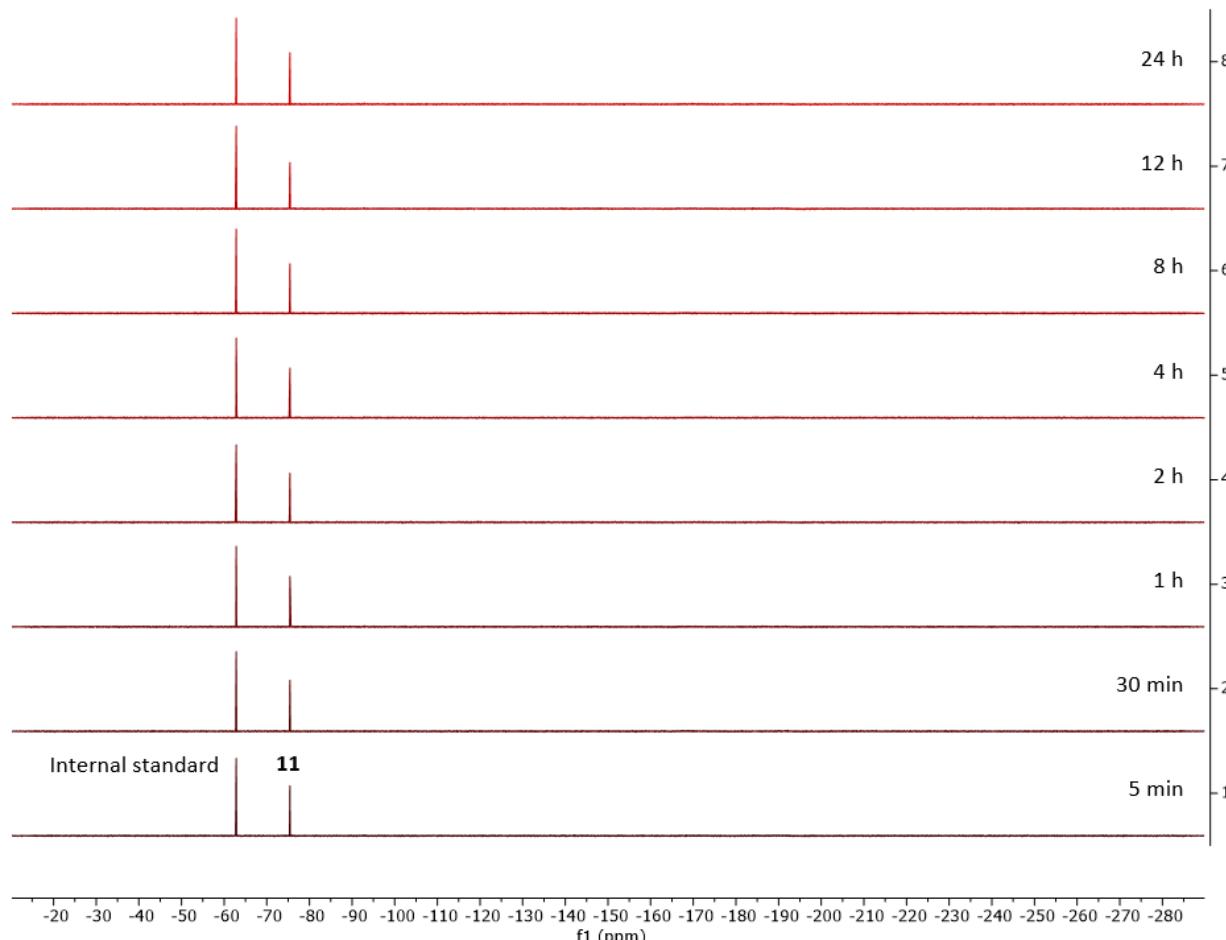


Figure S1. Alkyne **11** (500 μ M) is stable in 50% CD₃CN/PBS (pH 7.0) and exhibits no apparent degradation over 24 h.

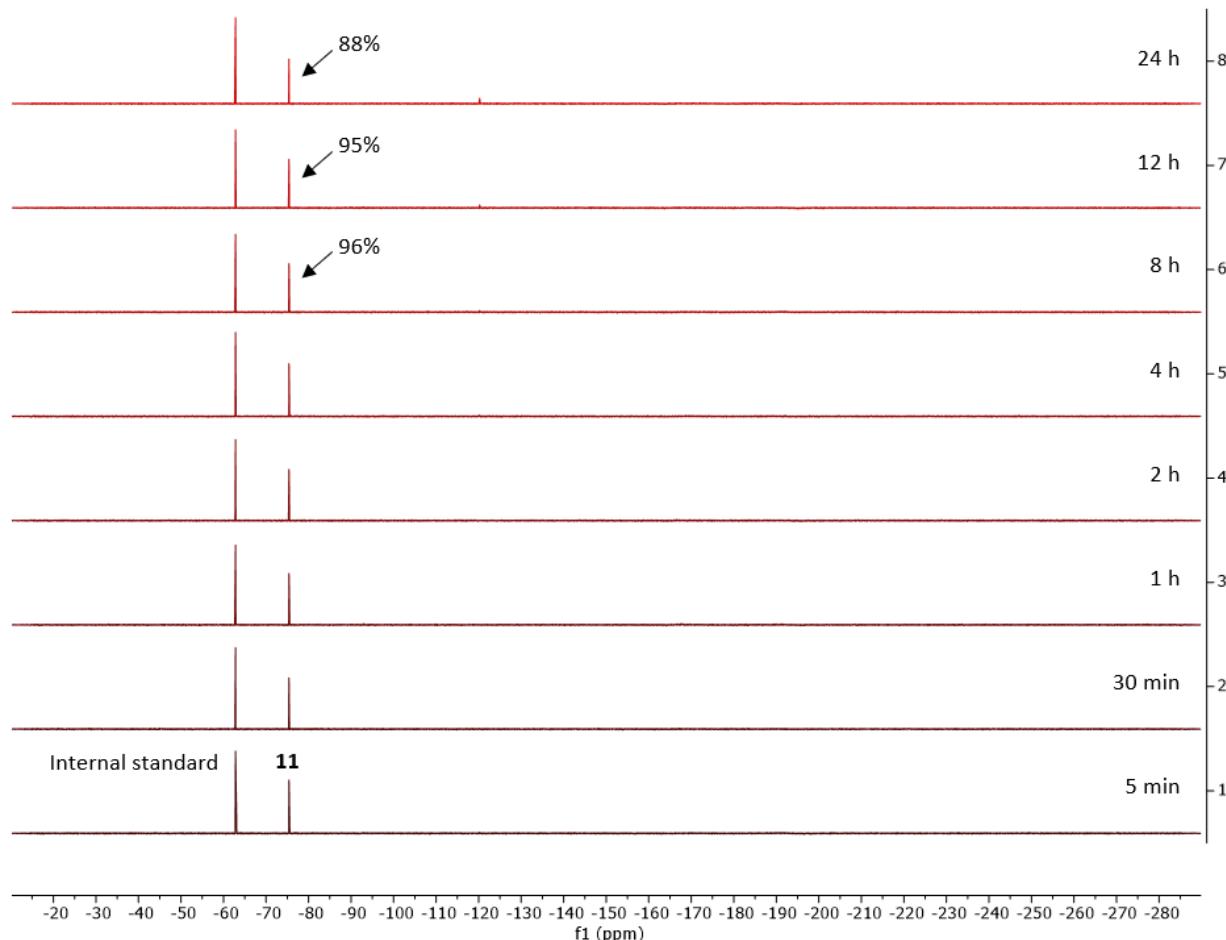


Figure S2. Alkyne **11** (500 μ M) has a half-life of 125 h in 50% CD₃CN/PBS (pH 7.0) in the presence of glutathione (2 mM) and exhibits only 5% degradation over 12 h.

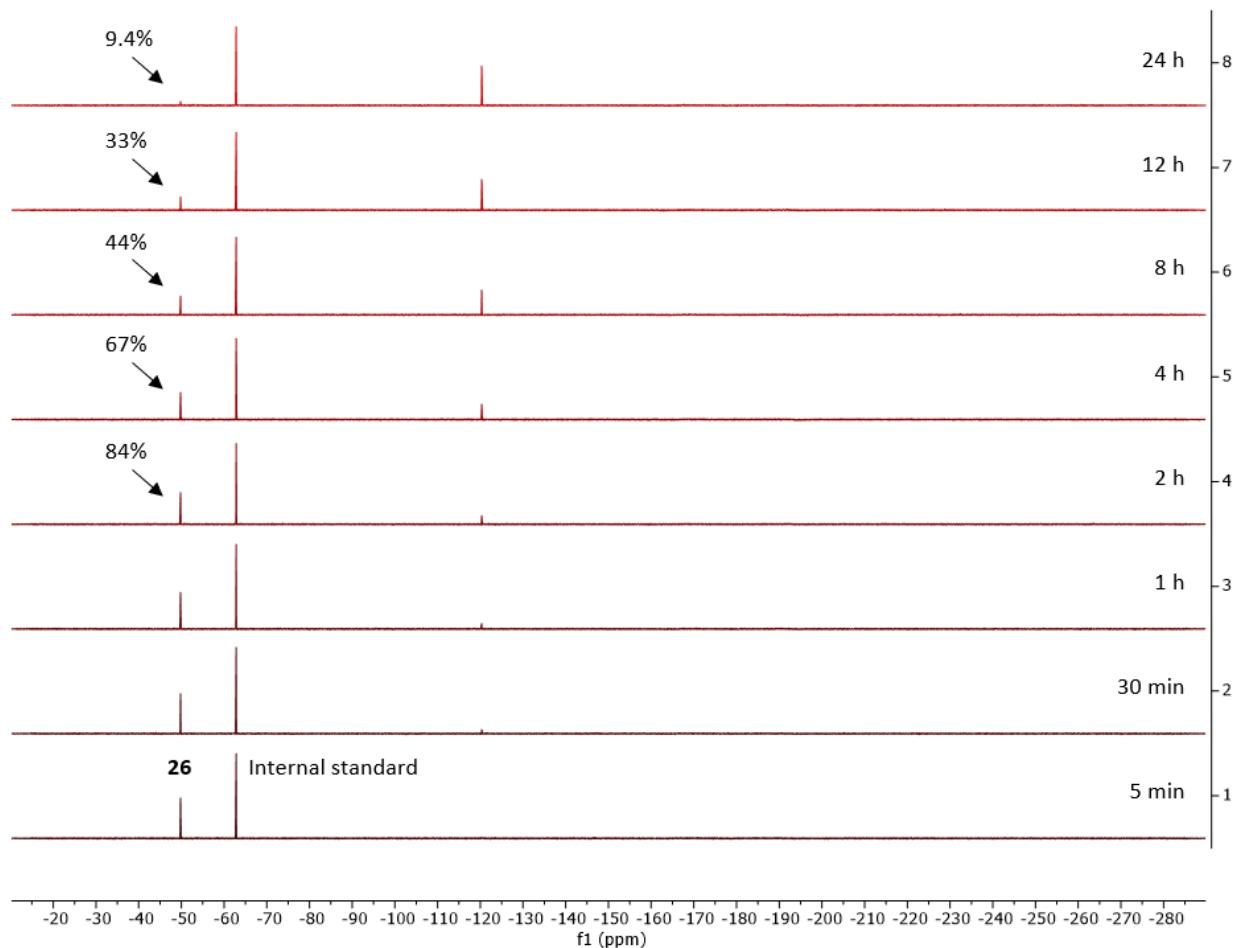


Figure S3. Alkyne **26** (500 μM) has a half-life of 7.1 h in 50% $\text{CD}_3\text{CN}/\text{PBS}$ (pH 7.0).

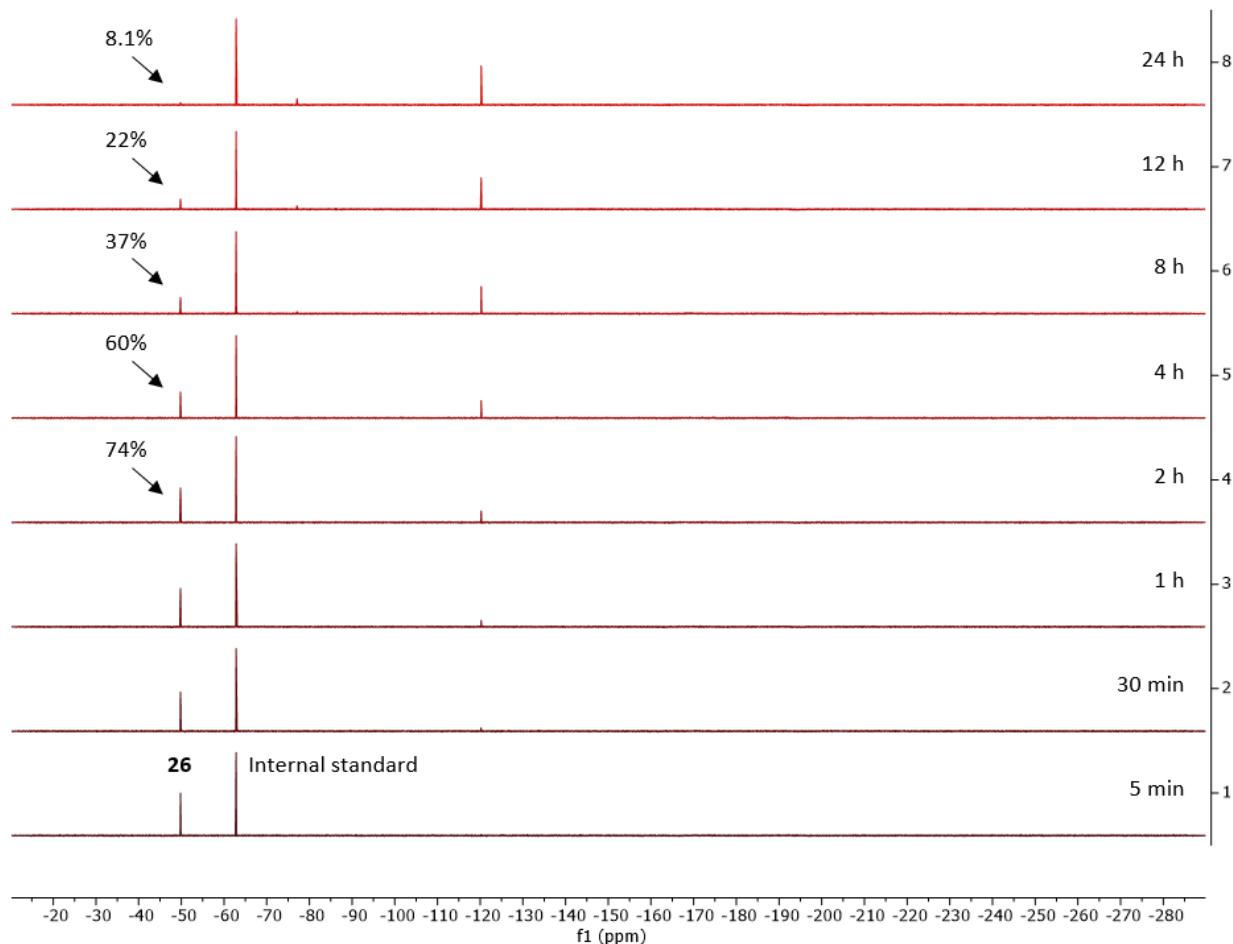


Figure S4. Alkyne **26** (500 μ M) has a half-life of 6.5 h in 50% CD₃CN/PBS (pH 7.0) in the presence of glutathione (2 mM).

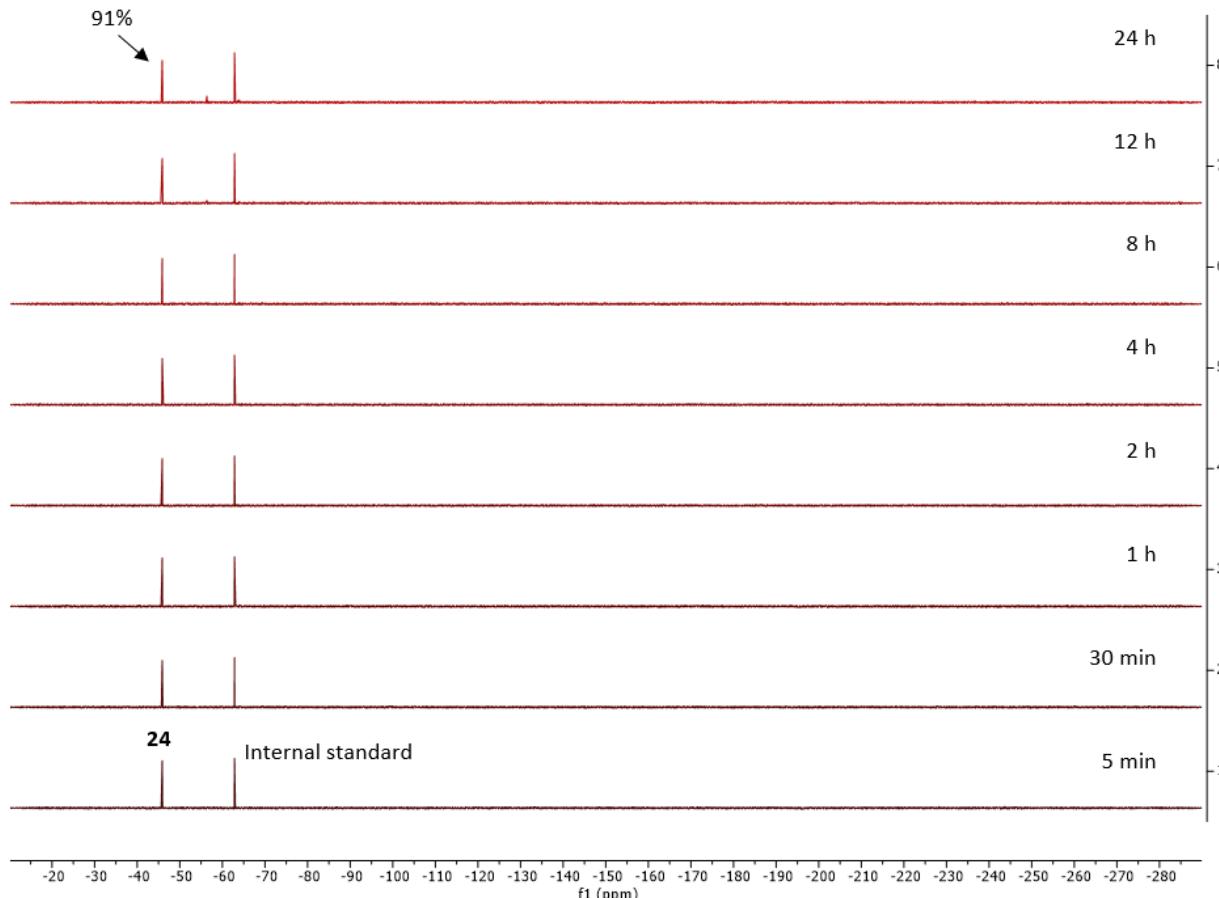


Figure S5. Alkyne **24** (500 μM) has a half-life of 4.6 days in 50% $\text{CD}_3\text{CN}/\text{PBS}$ (pH 7.0) and exhibits no apparent degradation over 24 h.

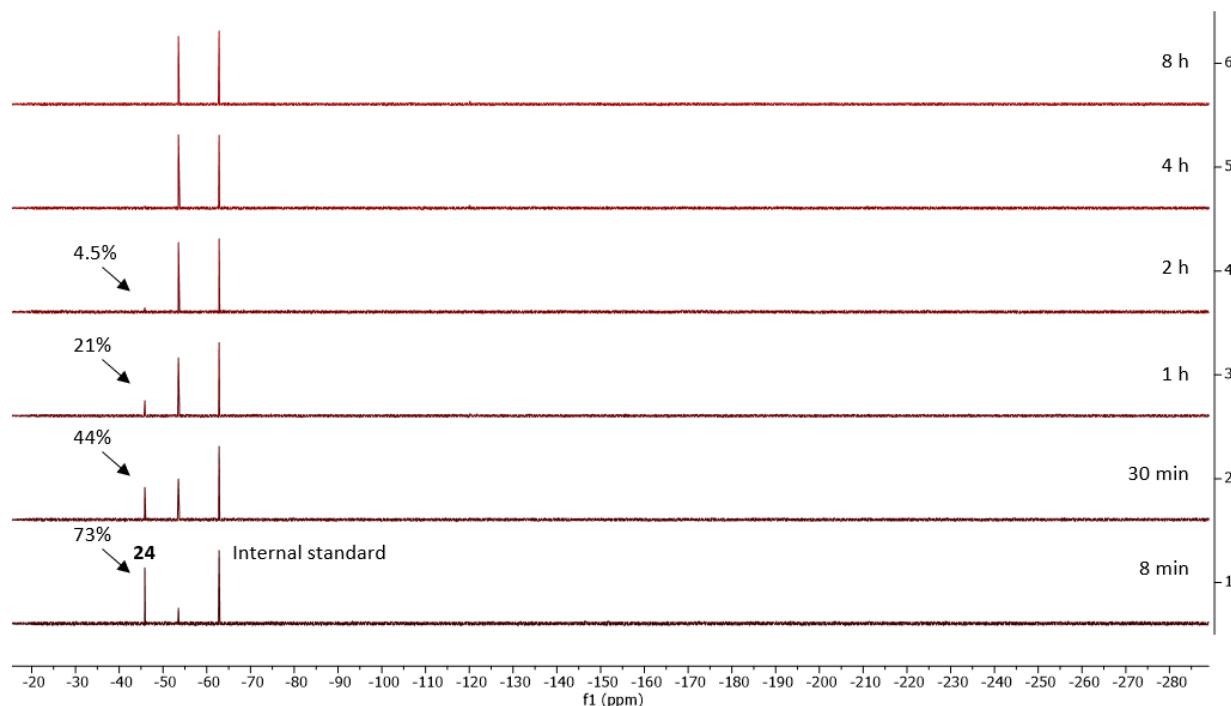


Figure S6. Alkyne **24** (500 μ M) has a half-life of 24 min in 50% CD₃CN/PBS (pH 7.0) in the presence of glutathione (2 mM).

Kinetics Study

All kinetics experiments were carried out at room temperature in methanol or CD₃CN. Reactions in CD₃CN were monitored via ¹⁹F NMR spectroscopy using α,α,α -benzotrifluoride (5 mM) as an internal standard. Reactions in methanol were monitored by ¹⁹F NMR spectroscopy using α,α,α -benzotrifluoride (5 mM) as an internal standard. A capillary NMR tube (diameter 3 mm) filled with D₂O was inserted into each NMR tube (diameter 5 mm) to lock the magnetic field. Second-order kinetics were performed by combining alkynes **11**, **24**, and **26** with *N,N*-diethylhydroxylamine in a 1:1 molar ratio to achieve a starting concentration of 5 mM for each compound. The reported errors for rate constants represent the standard deviation of the mean for triplicate experiments.

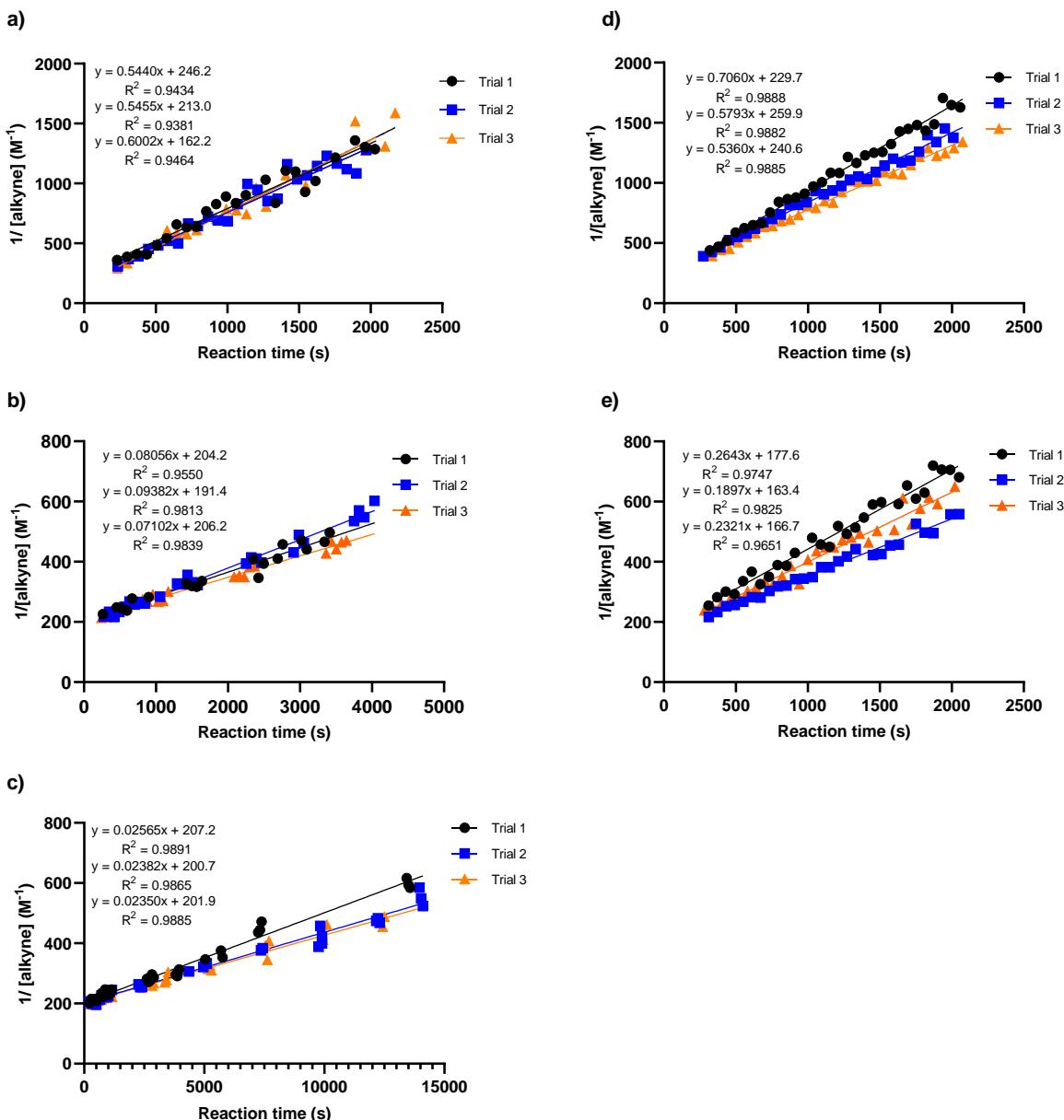


Figure S7. Reaction plots used to calculate second-order rate constants between alkynes **11**, **24**, and **26** (5 mM) and *N,N*-diethylhydroxylamine (5 mM). a) Alkyne **24** in methanol; b) Alkyne **26** in methanol; c)

Alkyne **11** in methanol; d) Alkyne **26** in CD₃CN; e) Alkyne **11** in CD₃CN. Each panel shows results performed in triplicate.

Computational Details

All calculations were conducted with Gaussian 09 software.¹ Geometry optimization of all species was performed using the M06-2X functional² with the 6-31G(d,p) basis set. Frequency analysis was carried out to ensure the stationary point was either a minimum or a transition state. Intrinsic reaction coordinates were computed for all transition states. Single-point calculations were carried out using the M06-2X functional with the 6-311G(2d,p) basis set. Hybridization was analyzed using natural bond orbital (NBO)³⁻⁴ analysis implemented in Gaussian.

- Cartesian coordinates of optimized structures (Å)

NMe₂OH

N	0.00000000	0.02490000	-0.41697400
C	-1.19858000	-0.64238700	0.06738400
H	-1.22931400	-1.65128600	-0.35177800
H	-1.22122900	-0.70365200	1.16601500
C	1.19858300	-0.64238200	0.06738400
H	1.22932400	-1.65127900	-0.35178200
H	1.22123100	-0.70365100	1.16601500
O	-0.00000300	1.31390800	0.19839700
H	-0.00000300	1.91418700	-0.55581600
H	2.07548200	-0.09062800	-0.27480500
H	-2.07548200	-0.09063900	-0.27480900

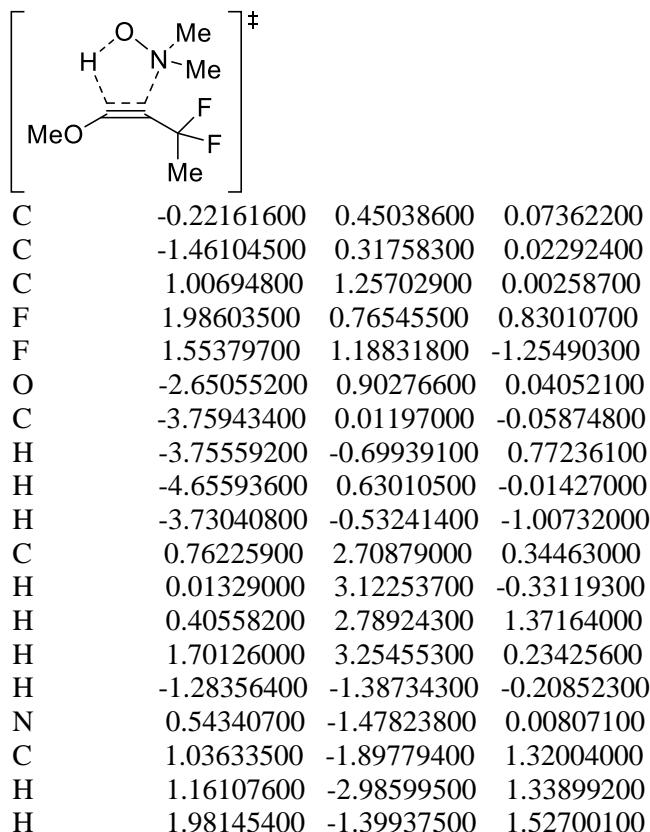
Chemical structure of NMe₂OMe:

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C	-1.25869100	0.40602000	0.01334500
C	1.36785600	-0.07542600	0.00111700
F	1.72755700	-0.78447900	1.11142900
F	1.70023000	-0.86049200	-1.06576600
O	-2.52975200	0.67671200	0.01808100
C	-3.37589700	-0.48488500	-0.01784100
H	-4.39710500	-0.10921000	-0.00610900
H	-3.18834600	-1.05370000	-0.93119700
H	-3.18813500	-1.10941700	0.85830800
C	2.18348300	1.19580300	-0.05287100
H	1.94187100	1.75015900	-0.96013100
H	1.96386100	1.81181300	0.81954900
H	3.24097200	0.92533500	-0.05648000

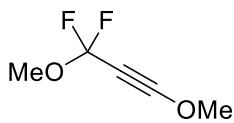
Chemical structure of the transition state (TS) of the reaction:

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C	0.37424400	0.64446200	-0.00360000

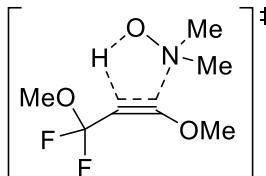
C	-2.12613100	-0.28994800	-0.04282000
F	-2.81458400	0.49766800	-0.92881300
F	-2.29772300	-1.57592500	-0.49858700
O	1.19235400	1.69148400	0.01989200
C	0.47111800	2.92567400	-0.01249600
H	-0.16157700	3.01581600	0.87515600
H	1.22045500	3.71616400	-0.02499600
H	-0.15319500	2.97356800	-0.90859600
C	-2.79074600	-0.18039400	1.31236100
H	-2.28919000	-0.83711400	2.02393100
H	-2.73373300	0.84934500	1.66696700
H	-3.83634400	-0.47841900	1.21277000
H	0.27554400	-1.55286200	-0.06935000
N	1.91293800	-0.69611100	0.01577300
C	2.66860200	-0.61627700	1.25911200
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H	1.96345100	-0.71062000	2.08580200
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H	3.22353200	0.39963900	-1.17139000
H	3.47644300	-1.37612600	-1.23539100
H	2.06767000	-0.63809100	-2.04861500
O	1.21730000	-1.88581700	-0.02137500



H	0.29382400	-1.60206700	2.06273100
C	1.49511100	-1.70001600	-1.07788000
H	2.39124600	-1.11088600	-0.88170600
H	1.74478500	-2.76496500	-1.15140400
H	1.03363200	-1.36238600	-2.00506300
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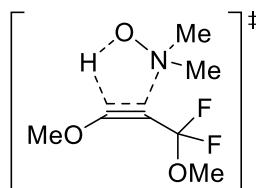

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O	-1.70333400	-0.91221900	-0.05690800
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H	-3.38928000	-0.06022400	-0.92795800
H	-3.57904200	-1.64642500	-0.12969900
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F	-1.22001400	1.04140900	-1.02701300
O	2.89548900	-0.77109900	0.04717300
C	3.79695200	0.34321400	-0.06571100
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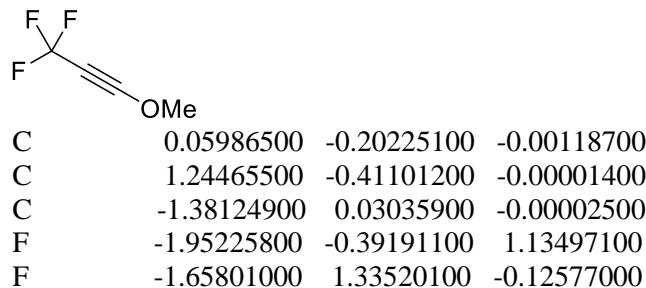

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C	3.67242200	-1.18650900	0.39678200
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H	3.85556500	-2.18669100	0.78609500
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F	2.43798100	1.20020300	0.53674400
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O	-1.22234700	-1.97507100	-0.07613400

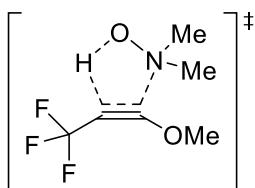


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C	-2.61891800	-2.11324600	-0.15910000
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H	1.97958400	-0.42189300	1.98940800
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H	3.53218400	-0.22293400	-1.38336200
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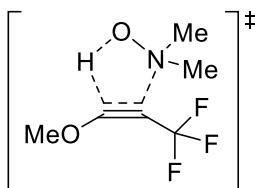


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F	-1.95225800	-0.39191100	1.13497100
F	-1.65801000	1.33520100	-0.12577000

O	2.51760800	-0.64622000	0.00081800
C	3.34075800	0.53546500	0.00012500
H	4.36815800	0.17825700	0.00249100
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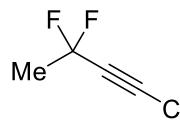


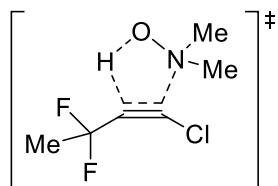
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H	-2.01223600	-0.68216300	2.06965200
H	-3.19358100	0.35848400	1.23080200
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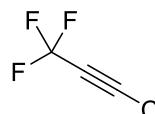
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F	-0.78895200	1.77151500	-0.88710400
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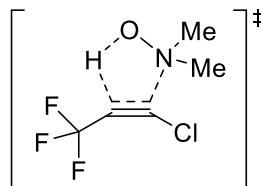
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H	-3.06961800	-1.37612000	-1.32185500
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H	-1.45992800	-1.10764400	-2.04740400
O	-1.25841400	-2.11451700	0.28409100
F	1.28356400	2.02957100	-0.32180400

			
C	-0.11215400	-0.05054700	-0.00022200
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C	2.22246786	-0.12714374	-0.04357807
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N	-1.80811160	-0.74389330	0.02426074
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H	-3.30315863	-1.56296015	-1.22178530
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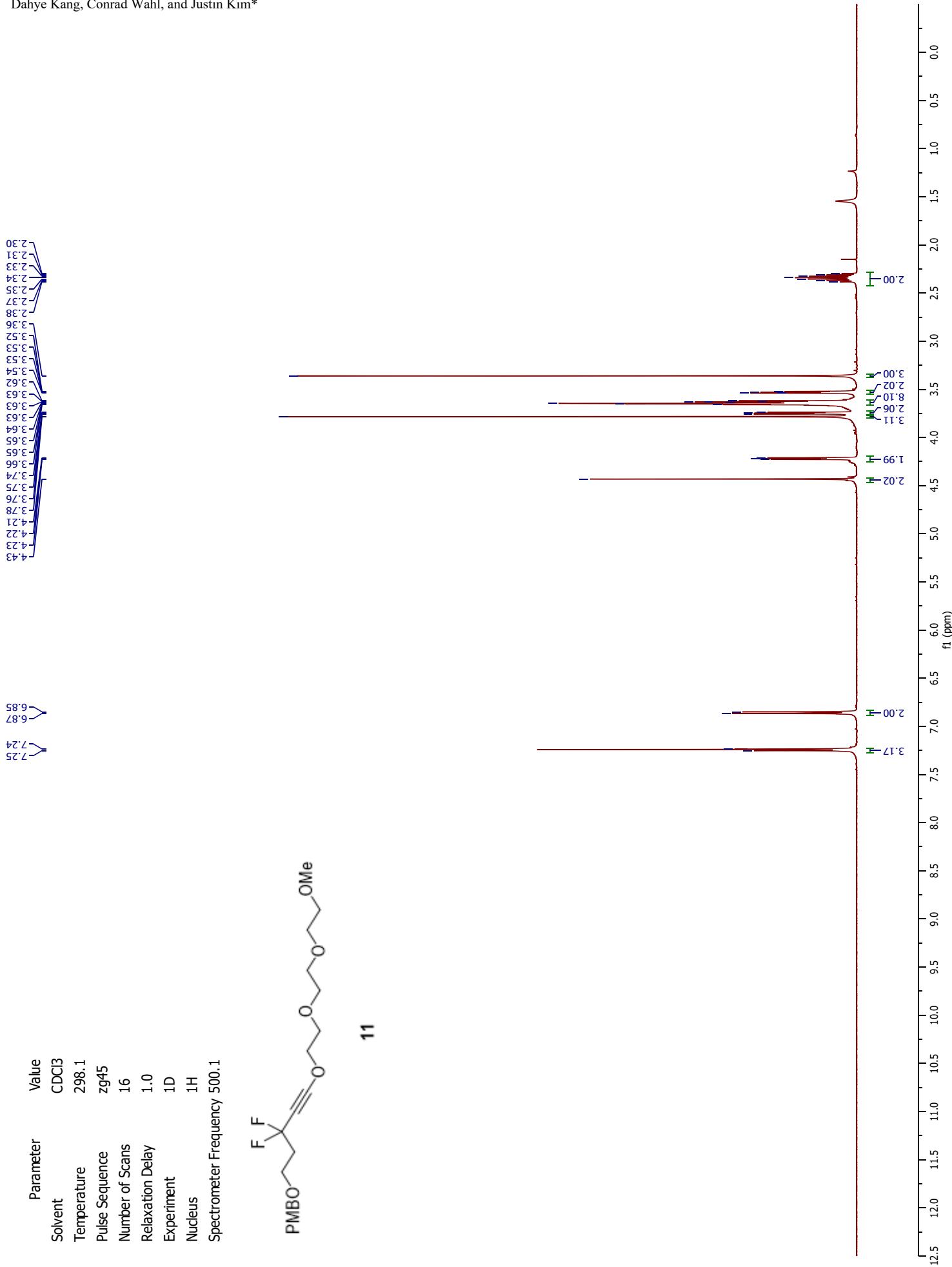
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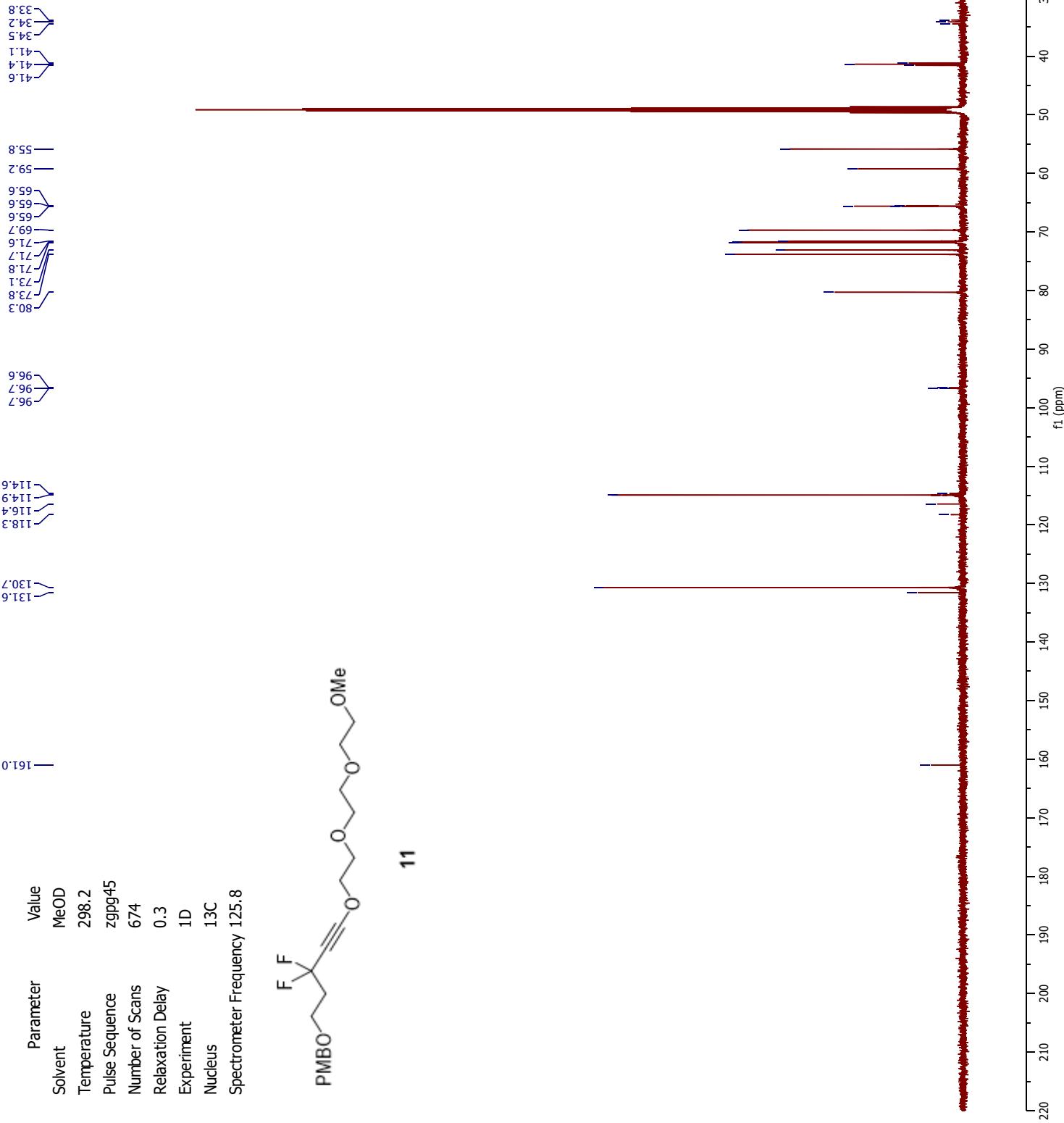

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Cl -2.93537600 0.00008600 -0.00009300

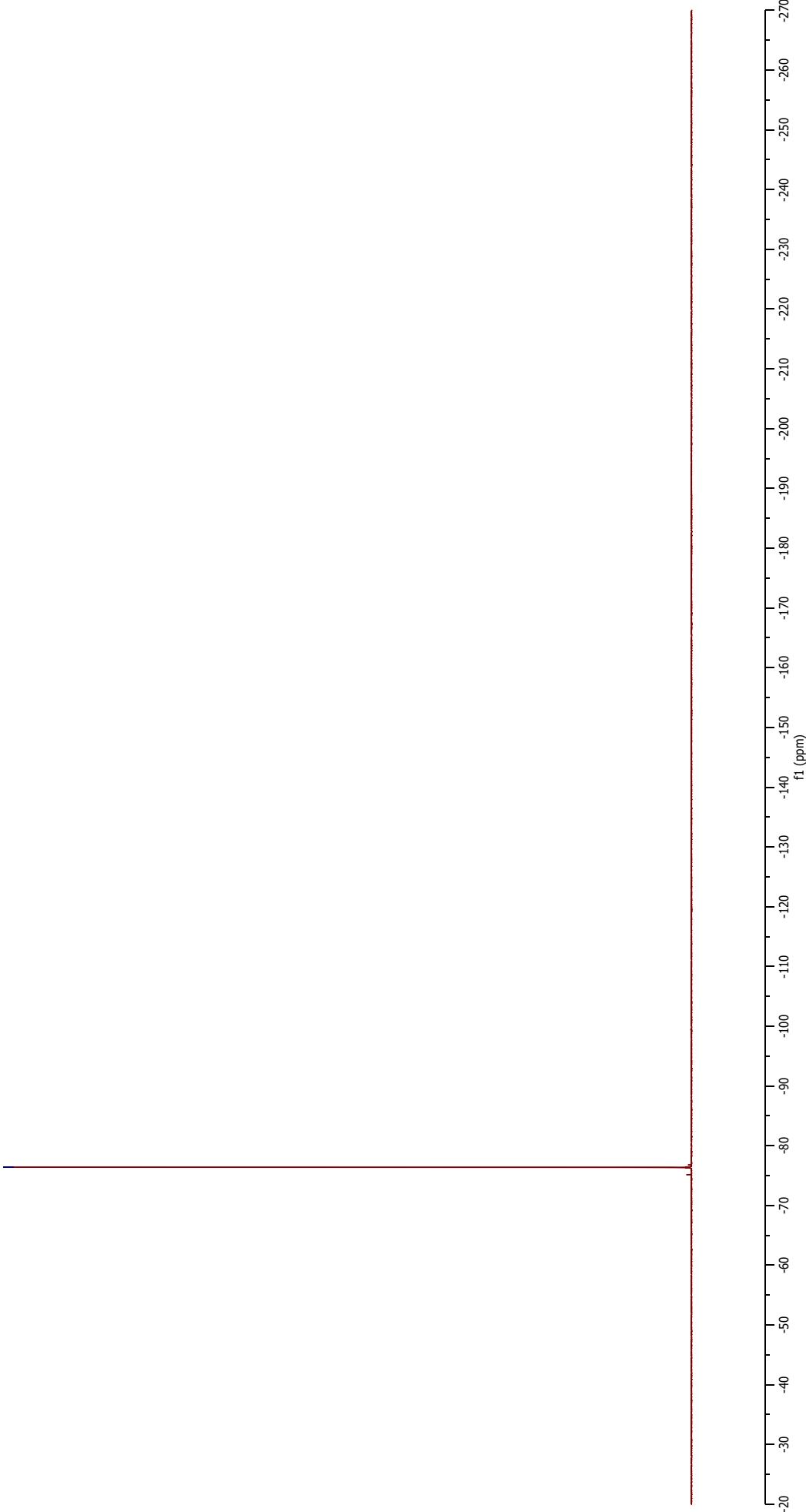
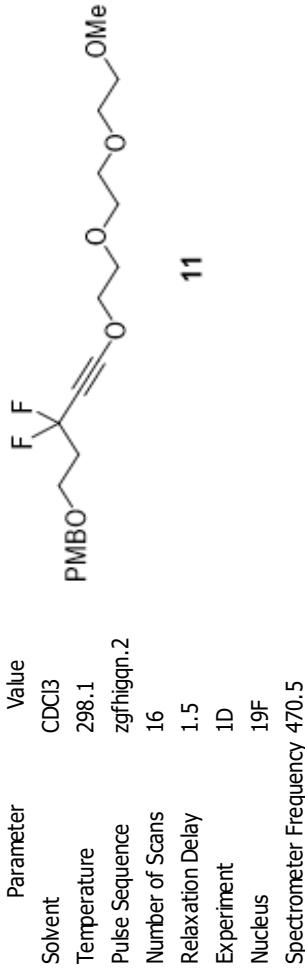

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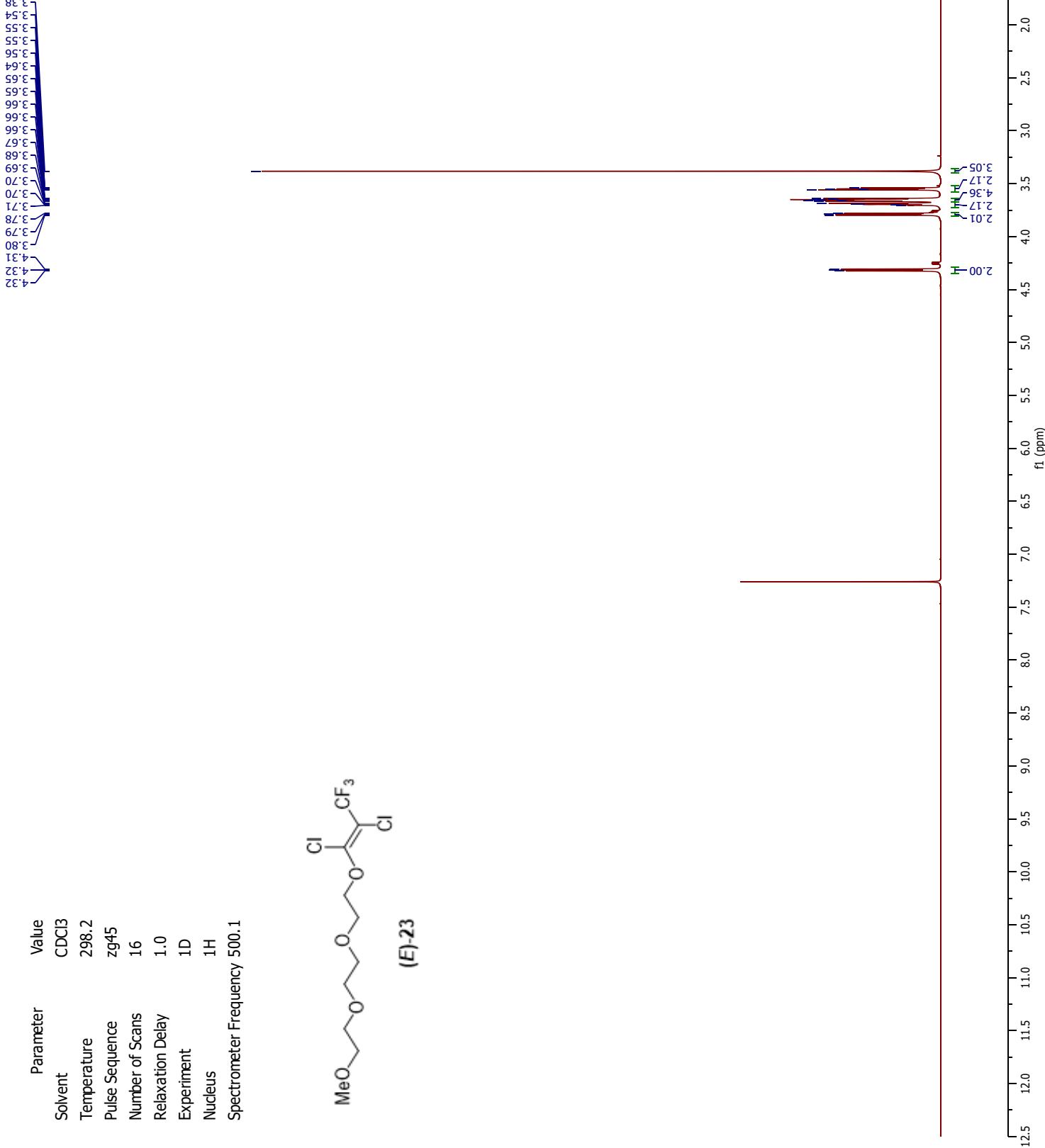
References

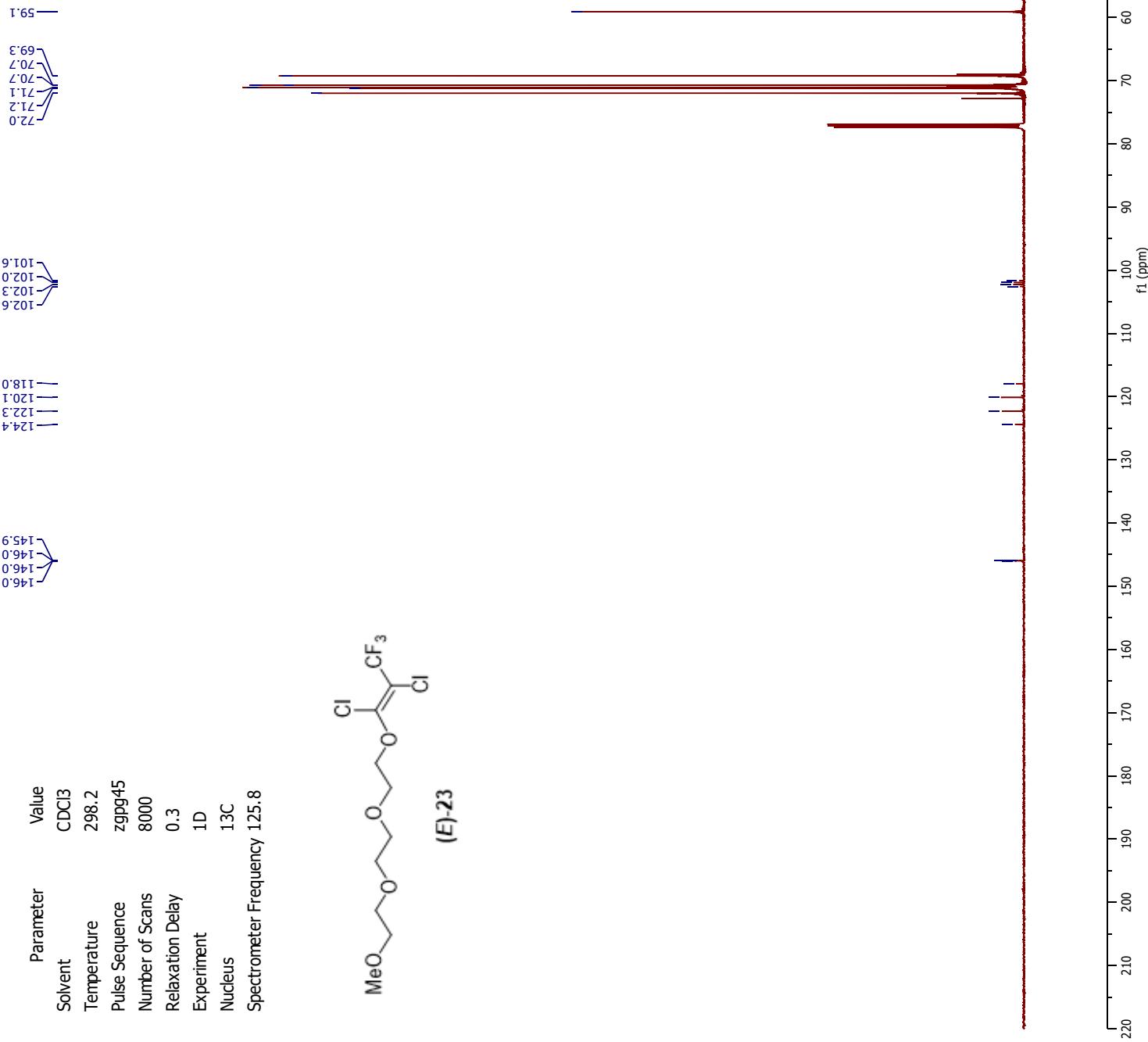
- (1) Gaussian 16, Revision C.01, Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Petersson, G. A.; Nakatsuji, H.; Li, X.; Caricato, M.; Marenich, A. V.; Bloino, J.; Janesko, B. G.; Gomperts, R.; Mennucci, B.; Hratchian, H. P.; Ortiz, J. V.; Izmaylov, A. F.; Sonnenberg, J. L.; Williams; Ding, F.; Lipparini, F.; Egidi, F.; Goings, J.; Peng, B.; Petrone, A.; Henderson, T.; Ranasinghe, D.; Zakrzewski, V. G.; Gao, J.; Rega, N.; Zheng, G.; Liang, W.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Throssell, K.; Montgomery Jr., J. A.; Peralta, J. E.; Ogliaro, F.; Bearpark, M. J.; Heyd, J. J.; Brothers, E. N.; Kudin, K. N.; Staroverov, V. N.; Keith, T. A.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A. P.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Millam, J. M.; Klene, M.; Adamo, C.; Cammi, R.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Farkas, O.; Foresman, J. B.; Fox, D. J. Gaussian, Inc., Wallingford CT, **2019**.
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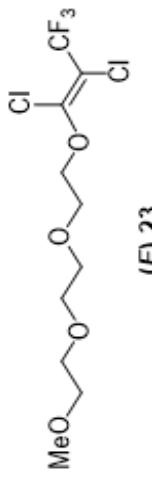






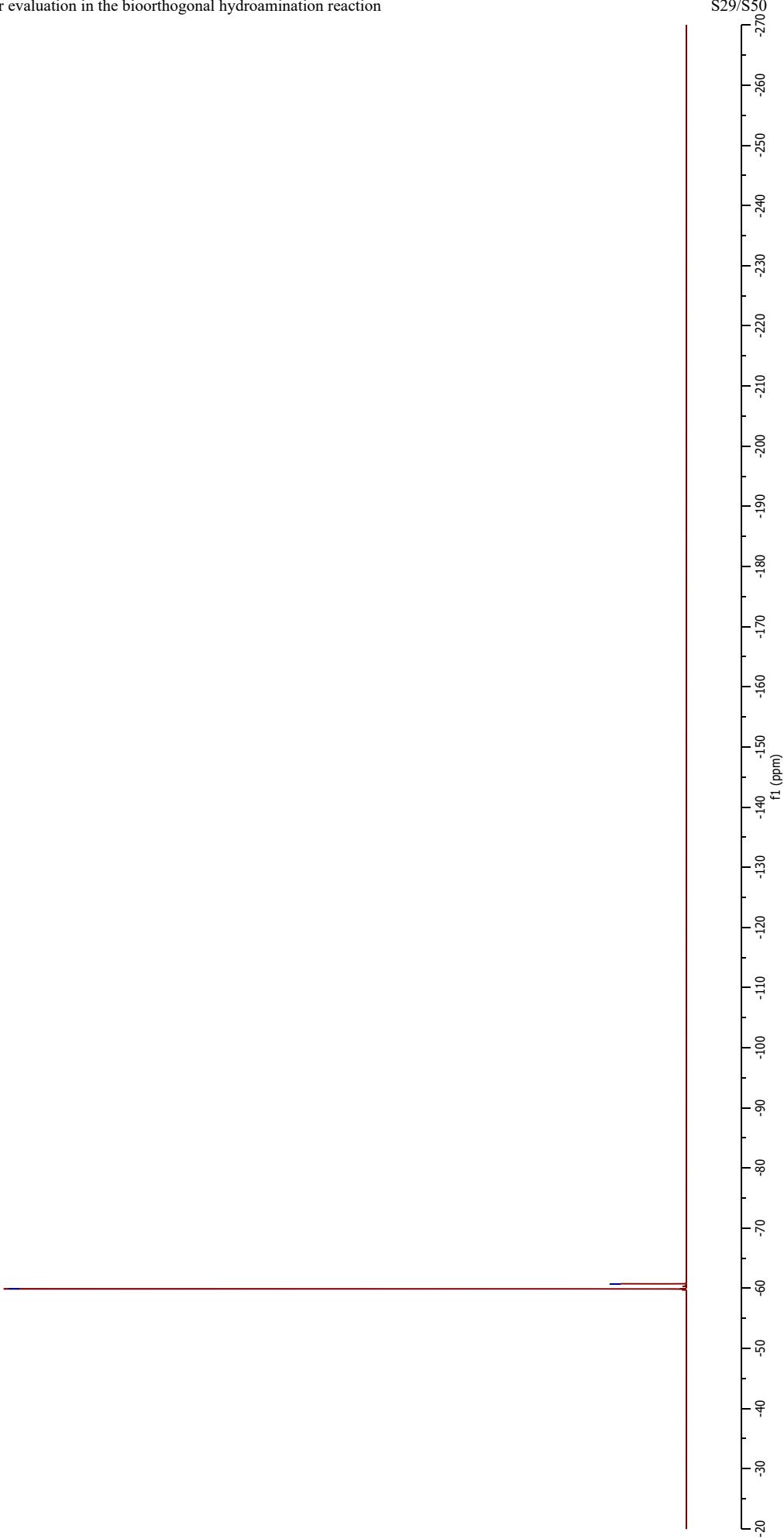


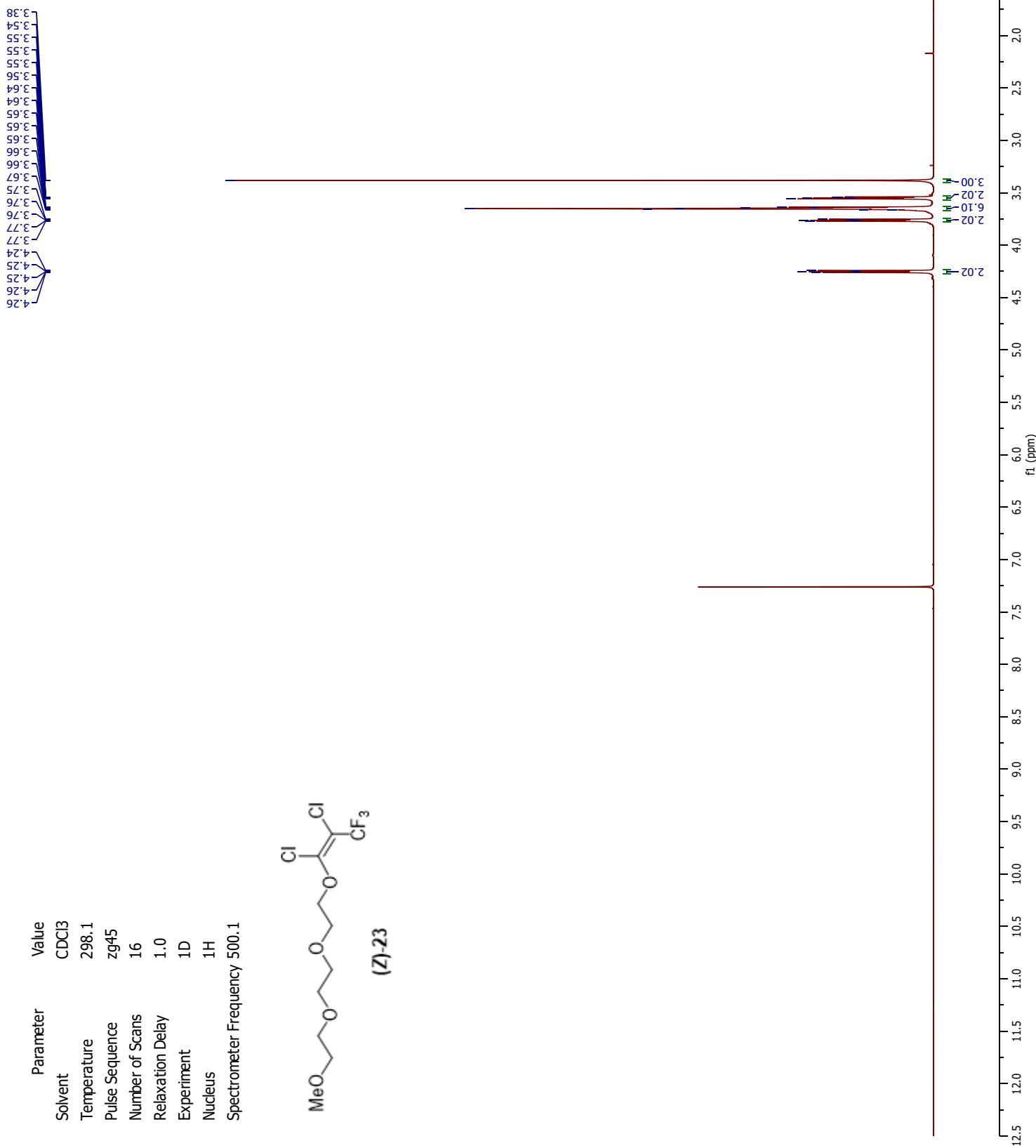


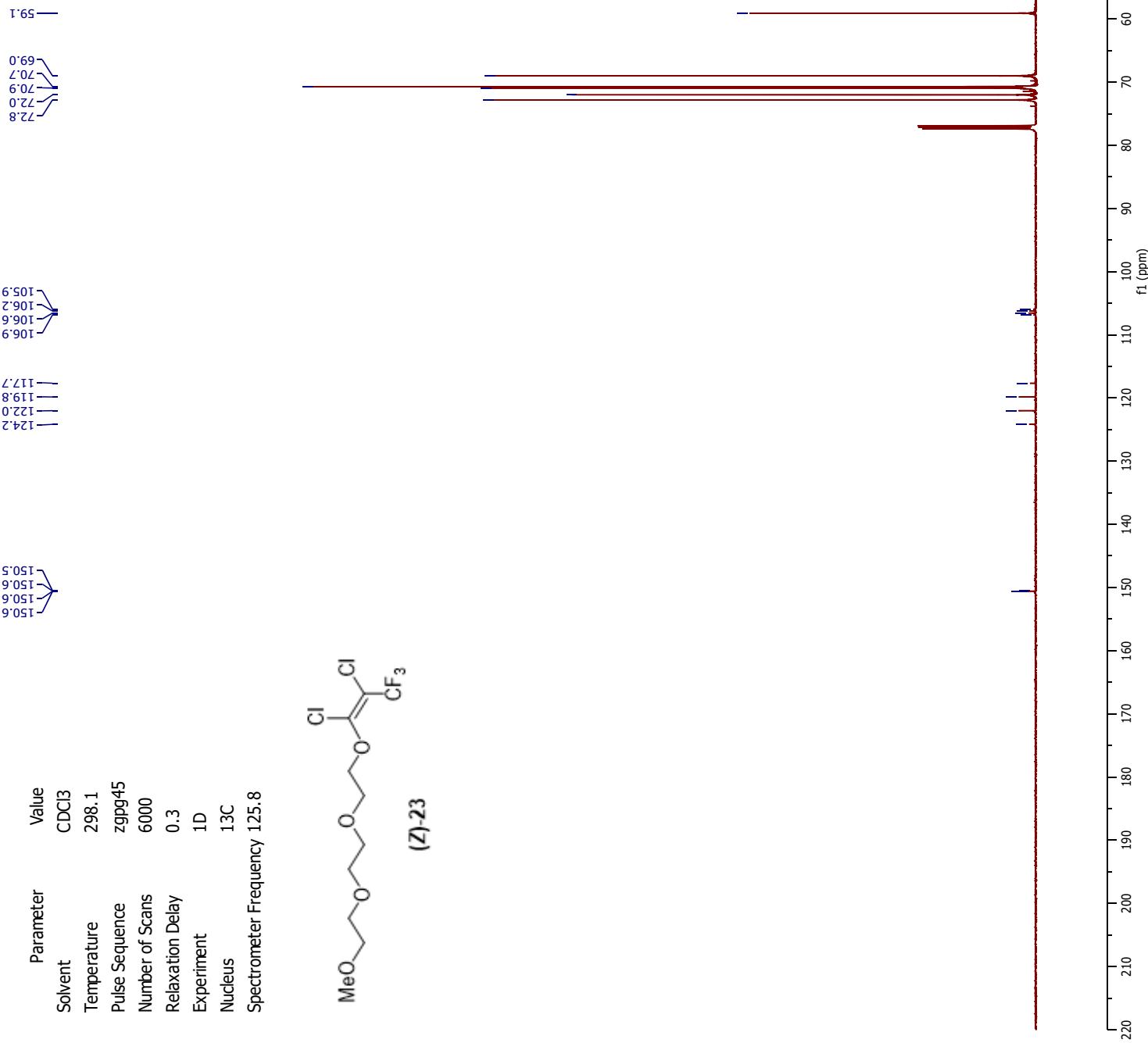


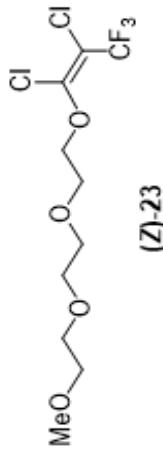
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Relaxation Delay	1.5
Experiment	1D
Nucleus	¹⁹ F
Spectrometer Frequency	470.5

8.00
6.95

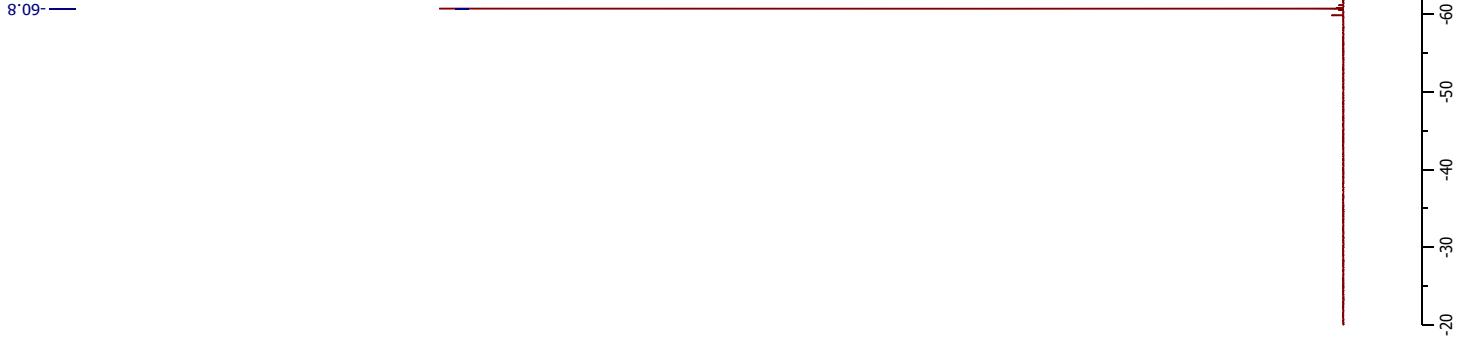


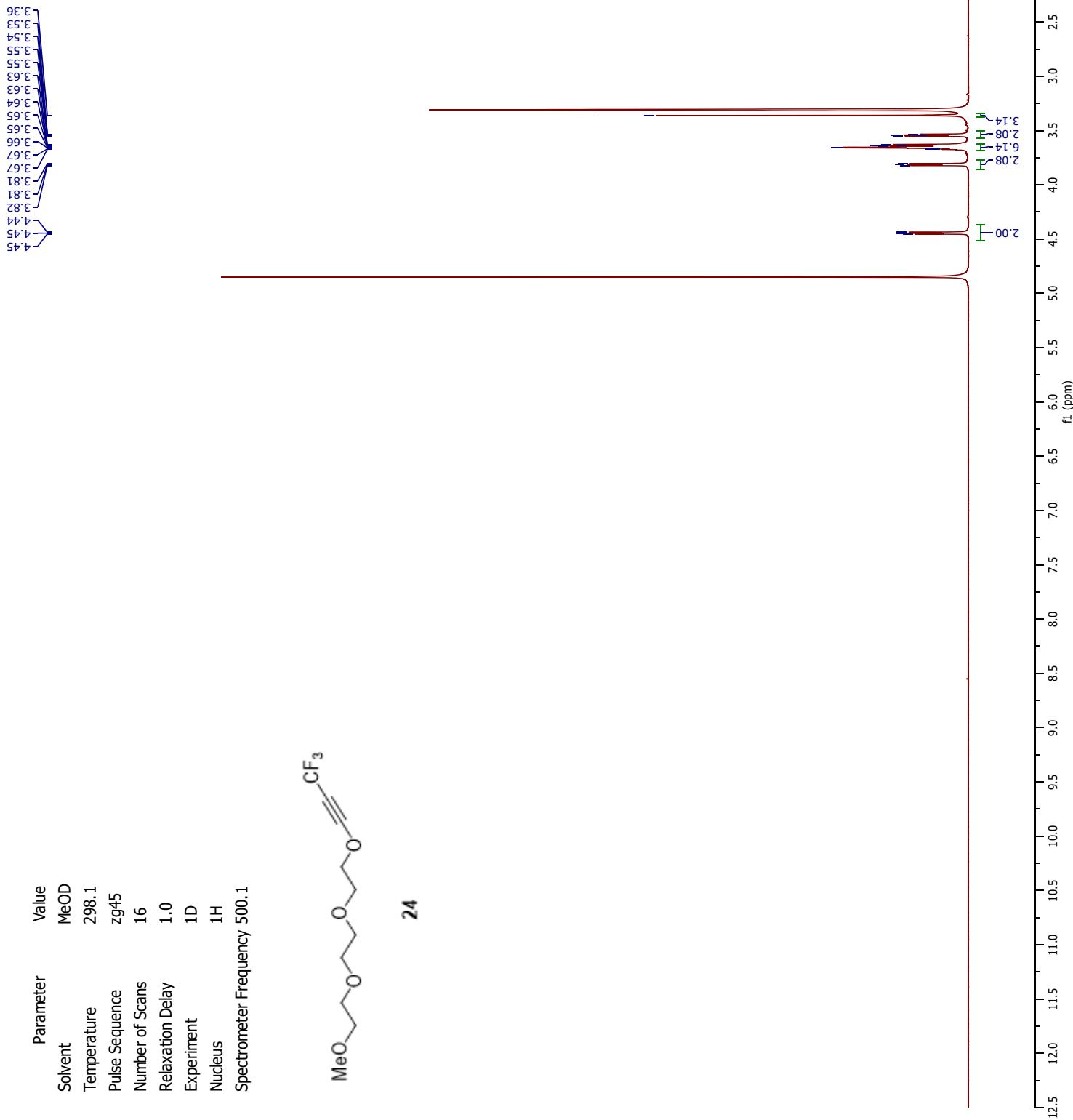






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Solvent	CDCl ₃
Temperature	298.2
Pulse Sequence	zgfhqg
Number of Scans	32
Relaxation Delay	1.5
Experiment	1D
Nucleus	19F
Spectrometer Frequency	470.5





28.3
28.7
29.2
29.6

59.2

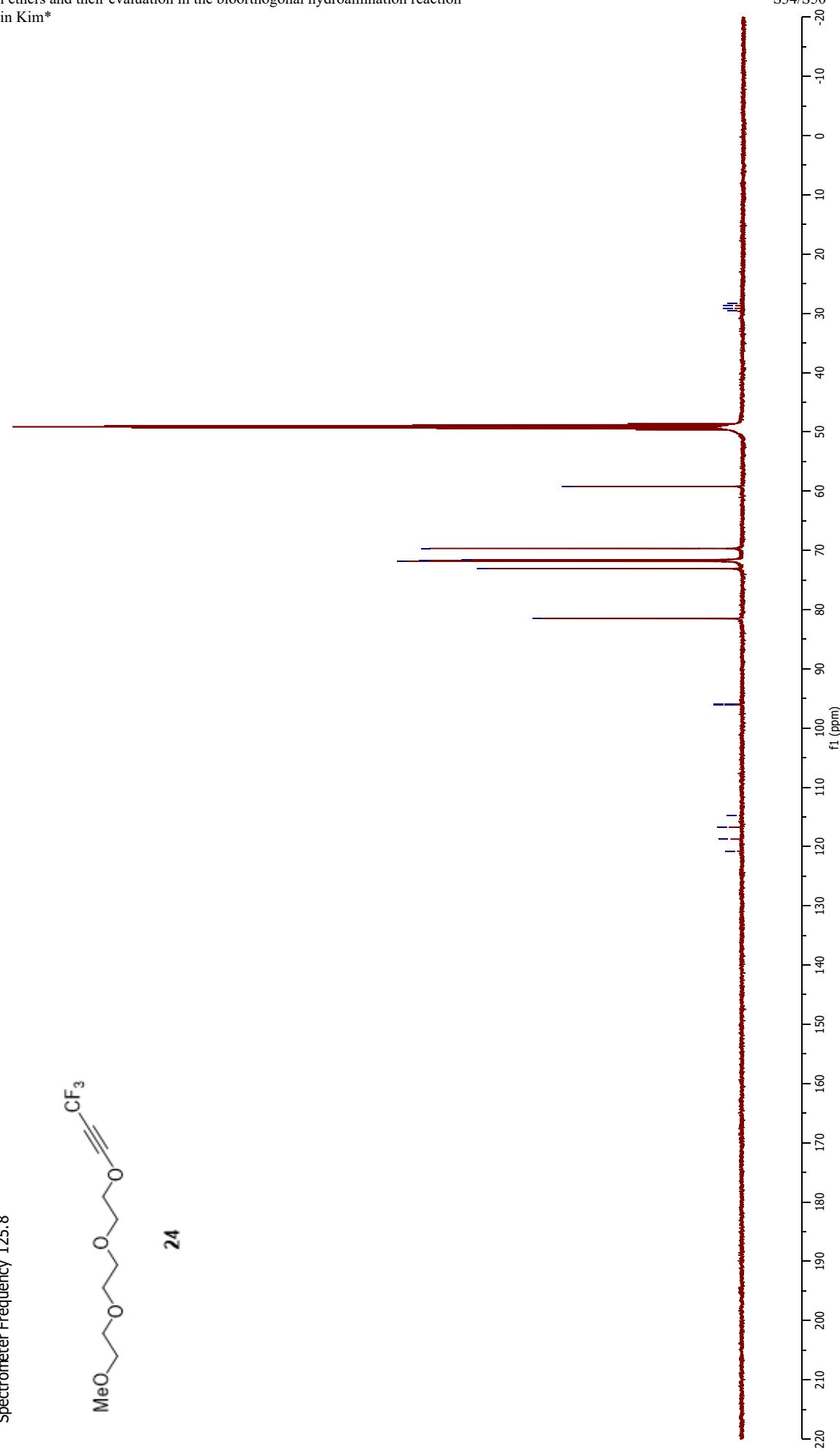
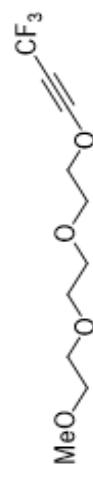
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71.6
71.7
73.1

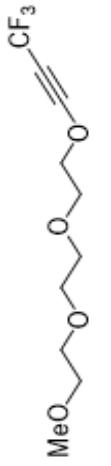
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95.9
96.0
96.1

114.7
116.7
118.8
120.8

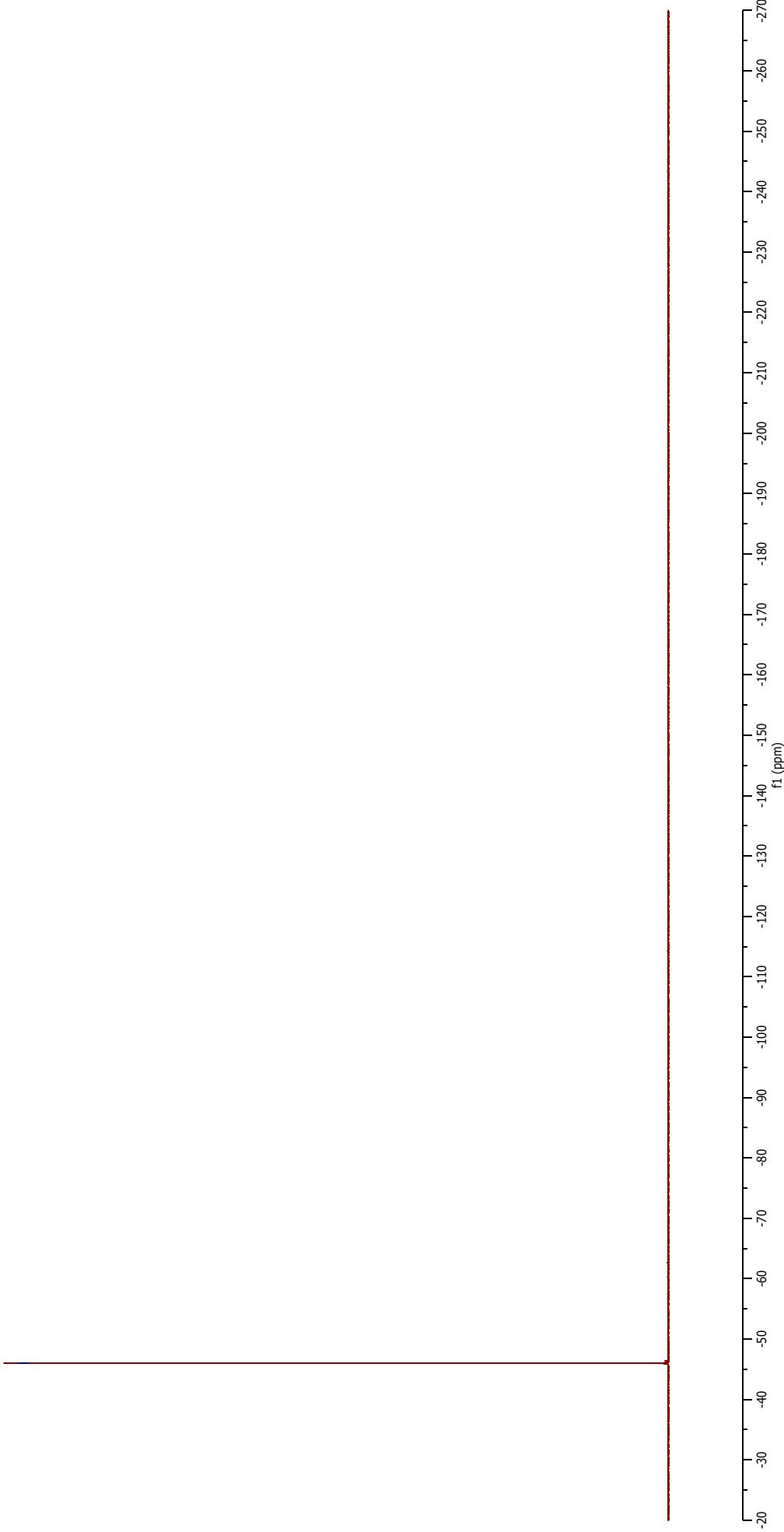
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Experiment	1D
Nucleus	¹³ C
Spectrometer Frequency	125.8

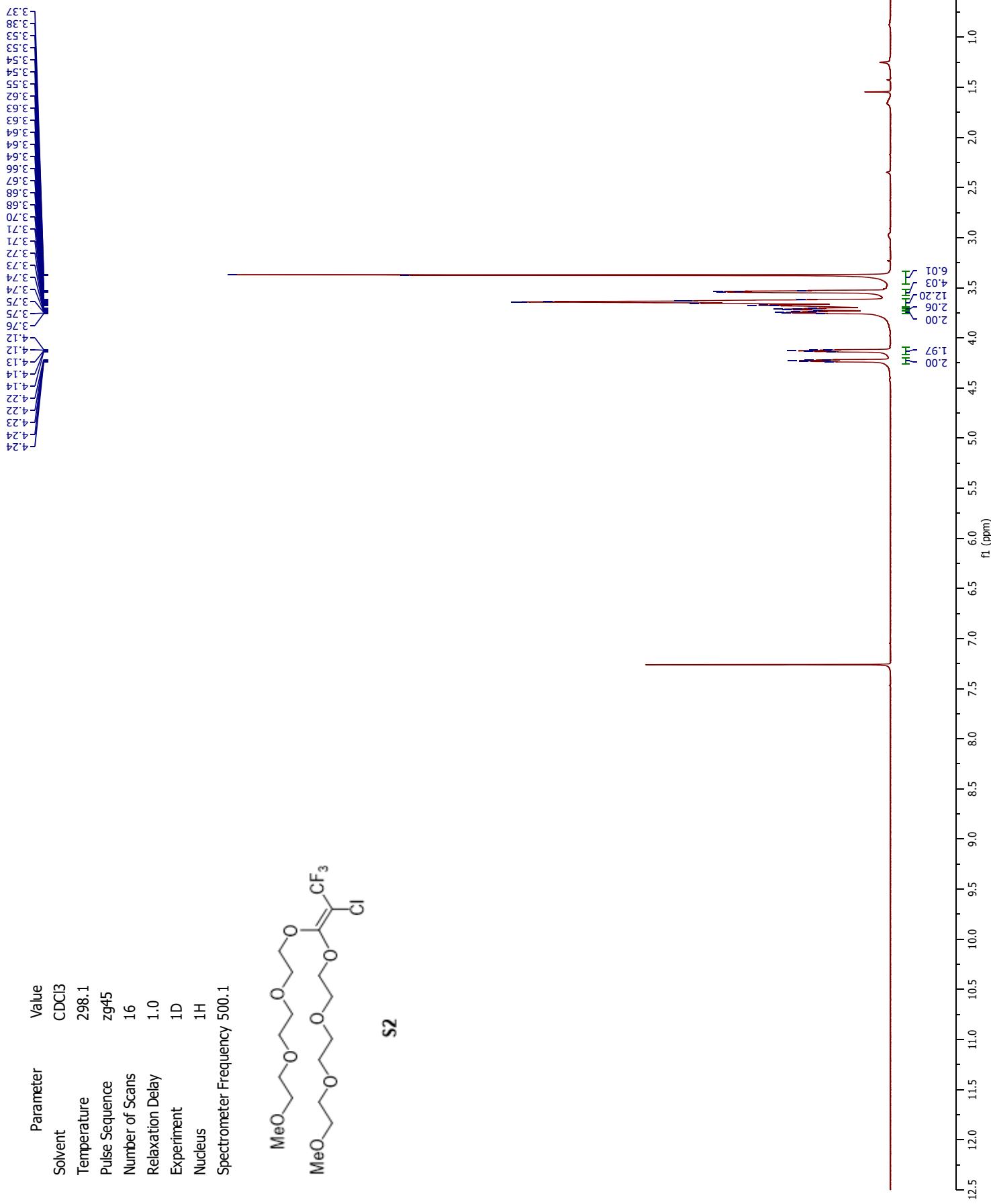


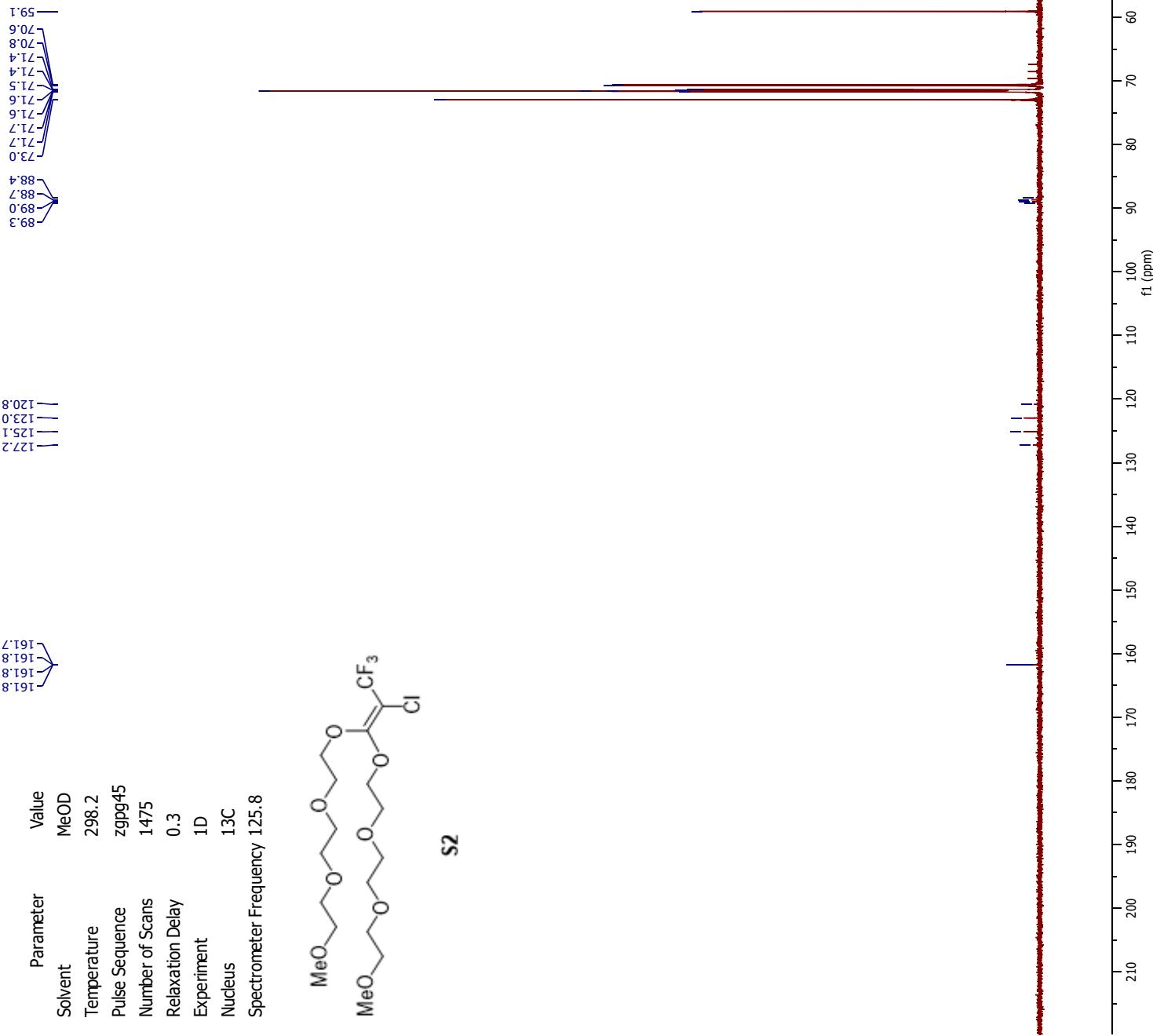


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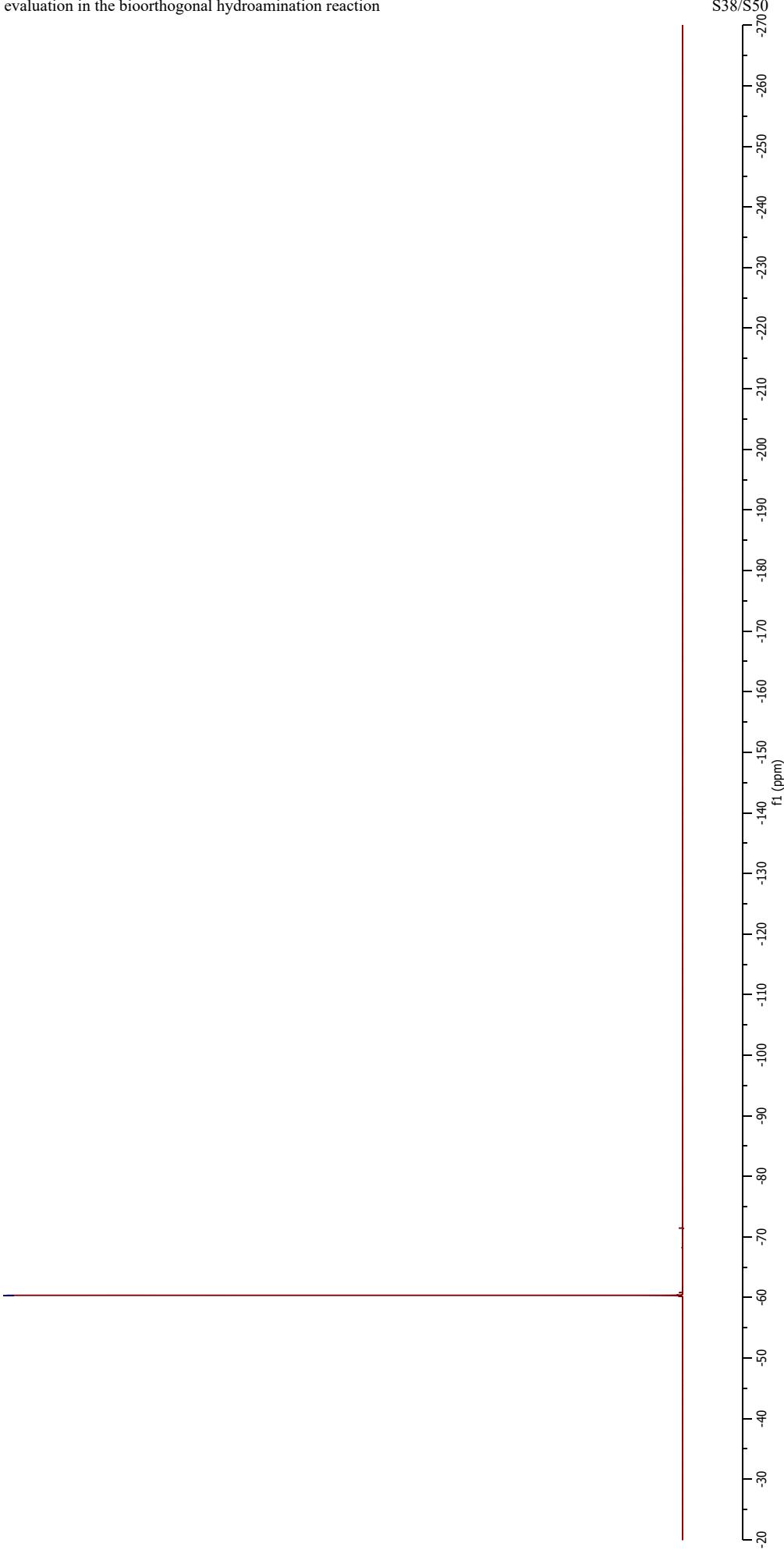
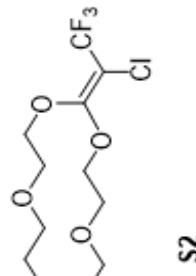
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Number of Scans	8
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Experiment	1D
Nucleus	19F
Spectrometer Frequency	470.5

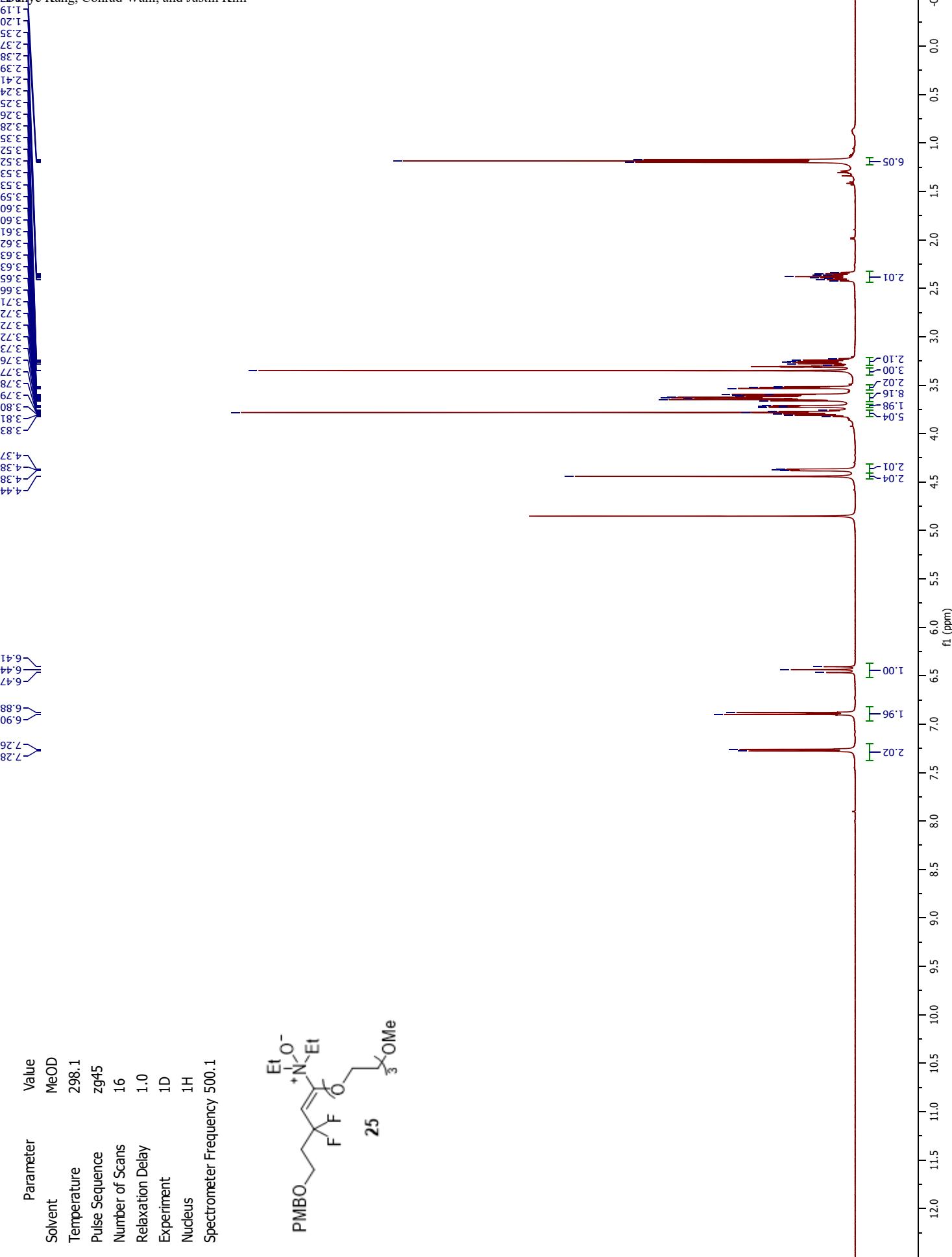


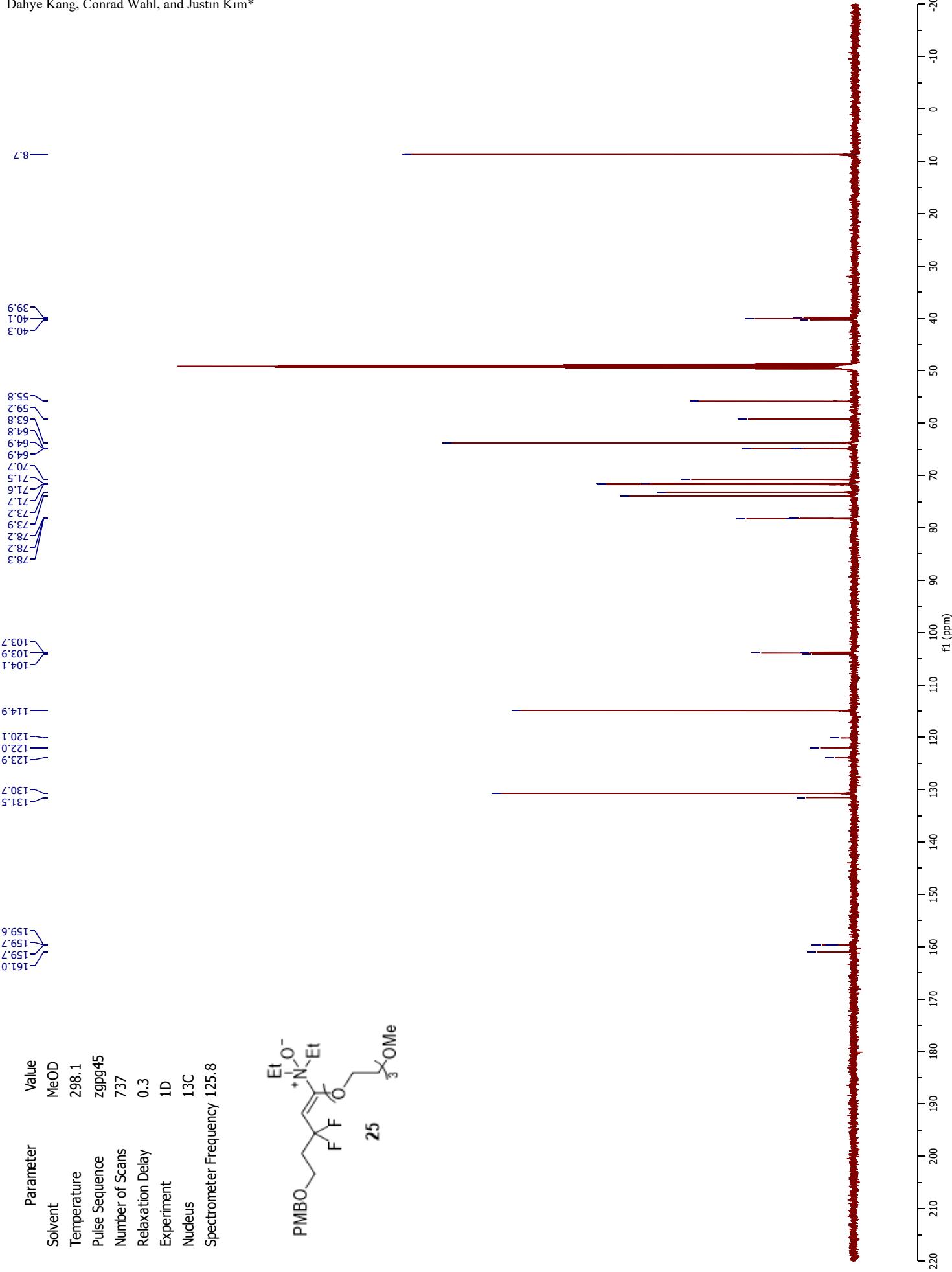


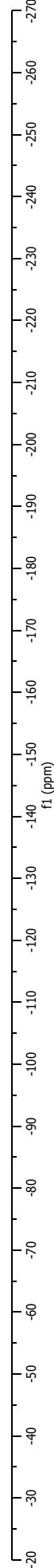
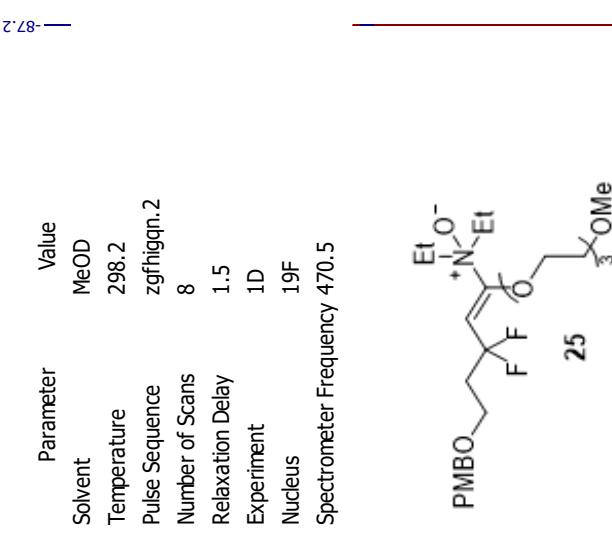


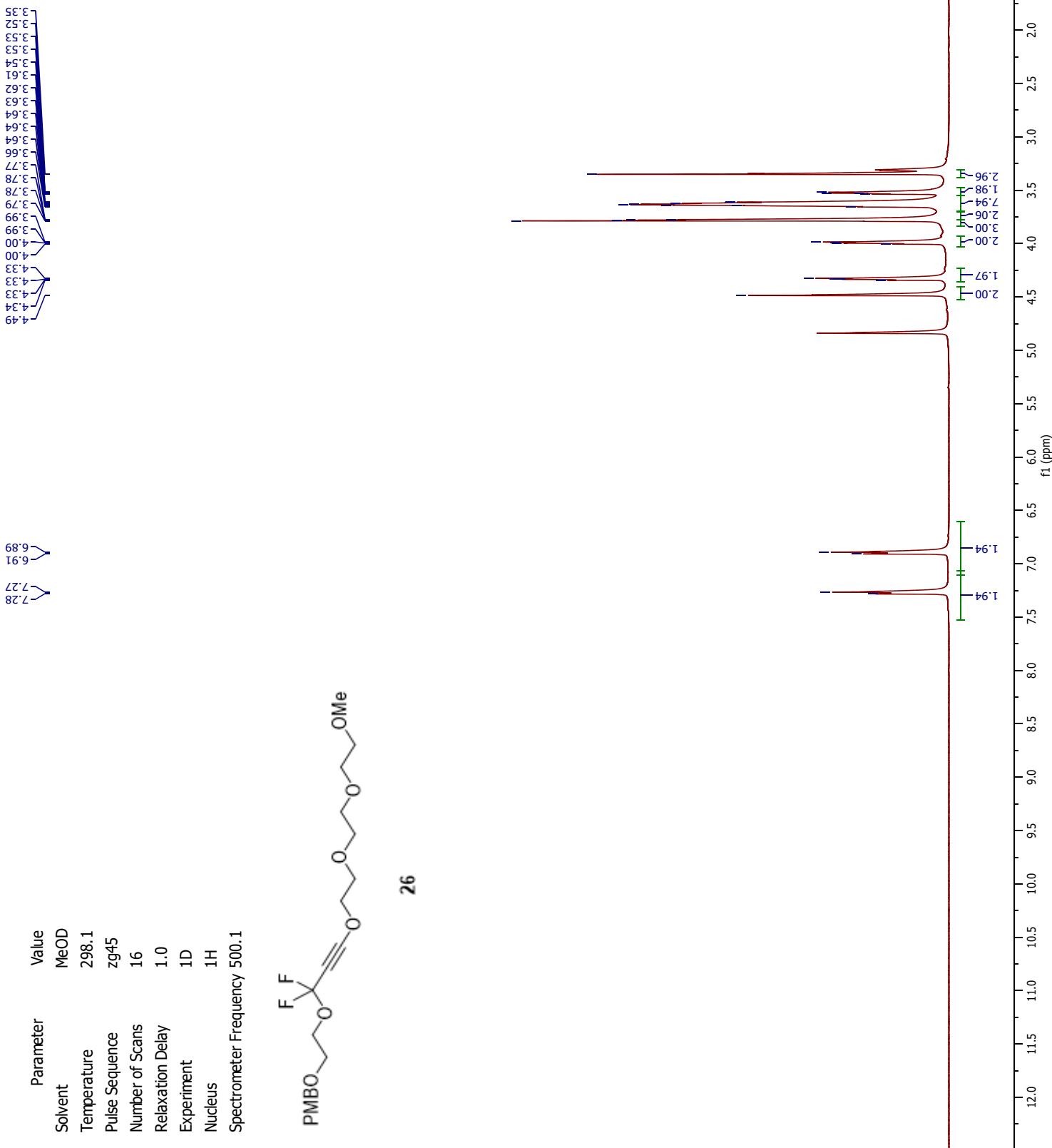
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Number of Scans	32
Relaxation Delay	1.5
Experiment	1D
Nucleus	¹⁹ F
Spectrometer Frequency	470.5











32.2
31.8
31.3

55.8
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66.1
66.1
69.1
69.7
71.5
71.7
71.8
73.1
73.9
80.6

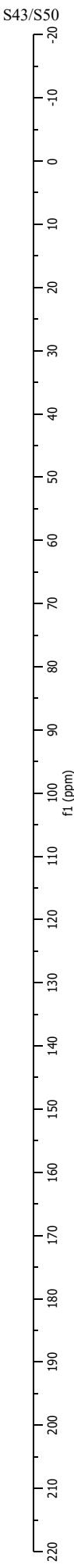
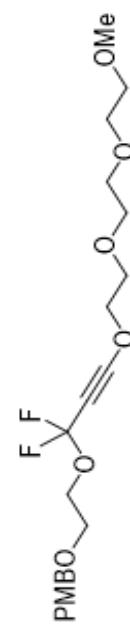
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94.5

114.9
115.9
117.8
117.8
117.8

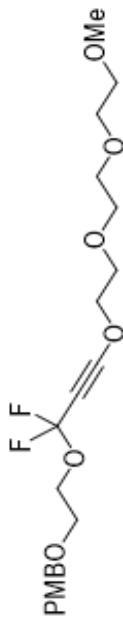
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131.5

161.0

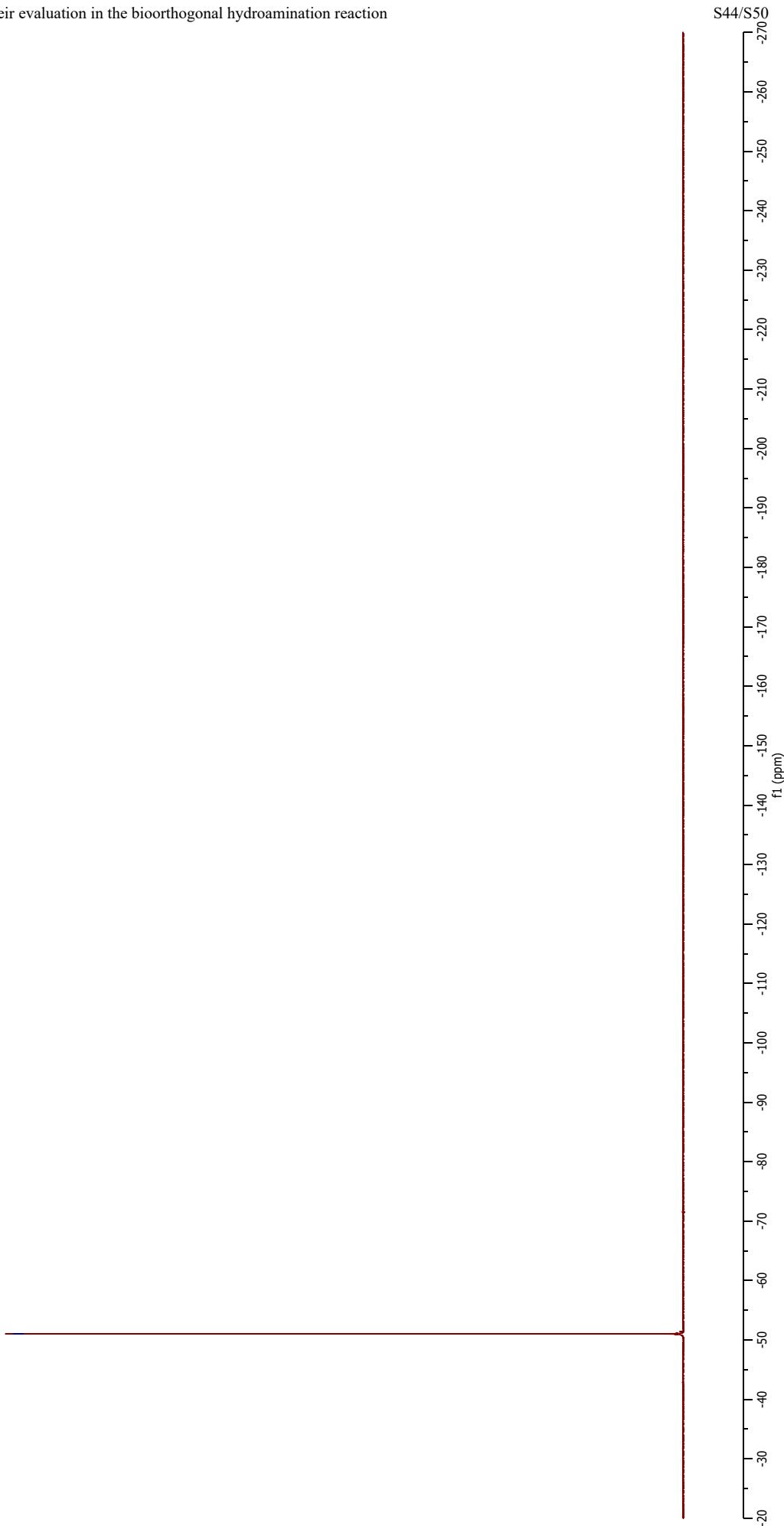
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Relaxation Delay	0.3
Experiment	1D
Nucleus	¹³ C
Spectrometer Frequency	125.8

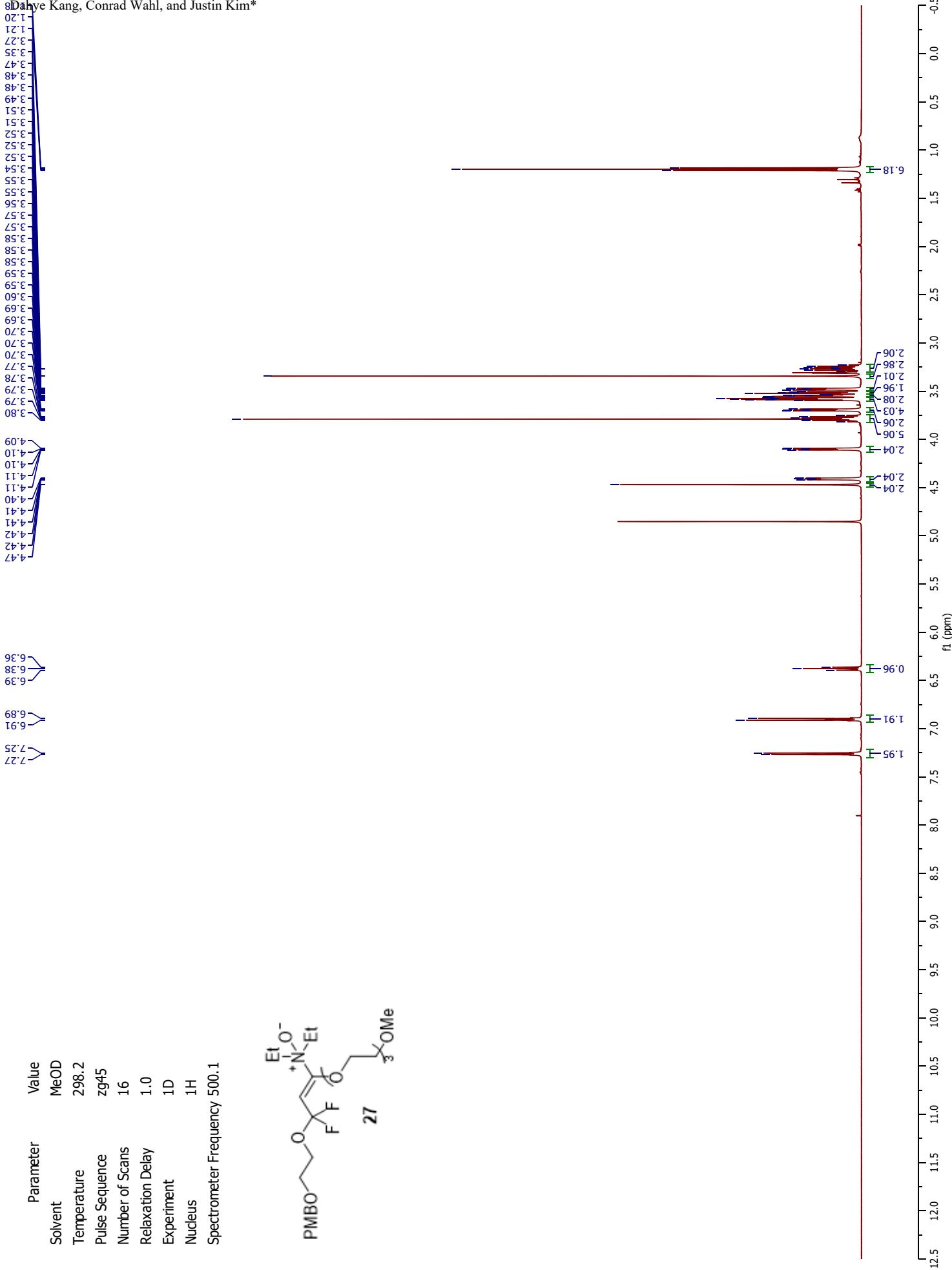


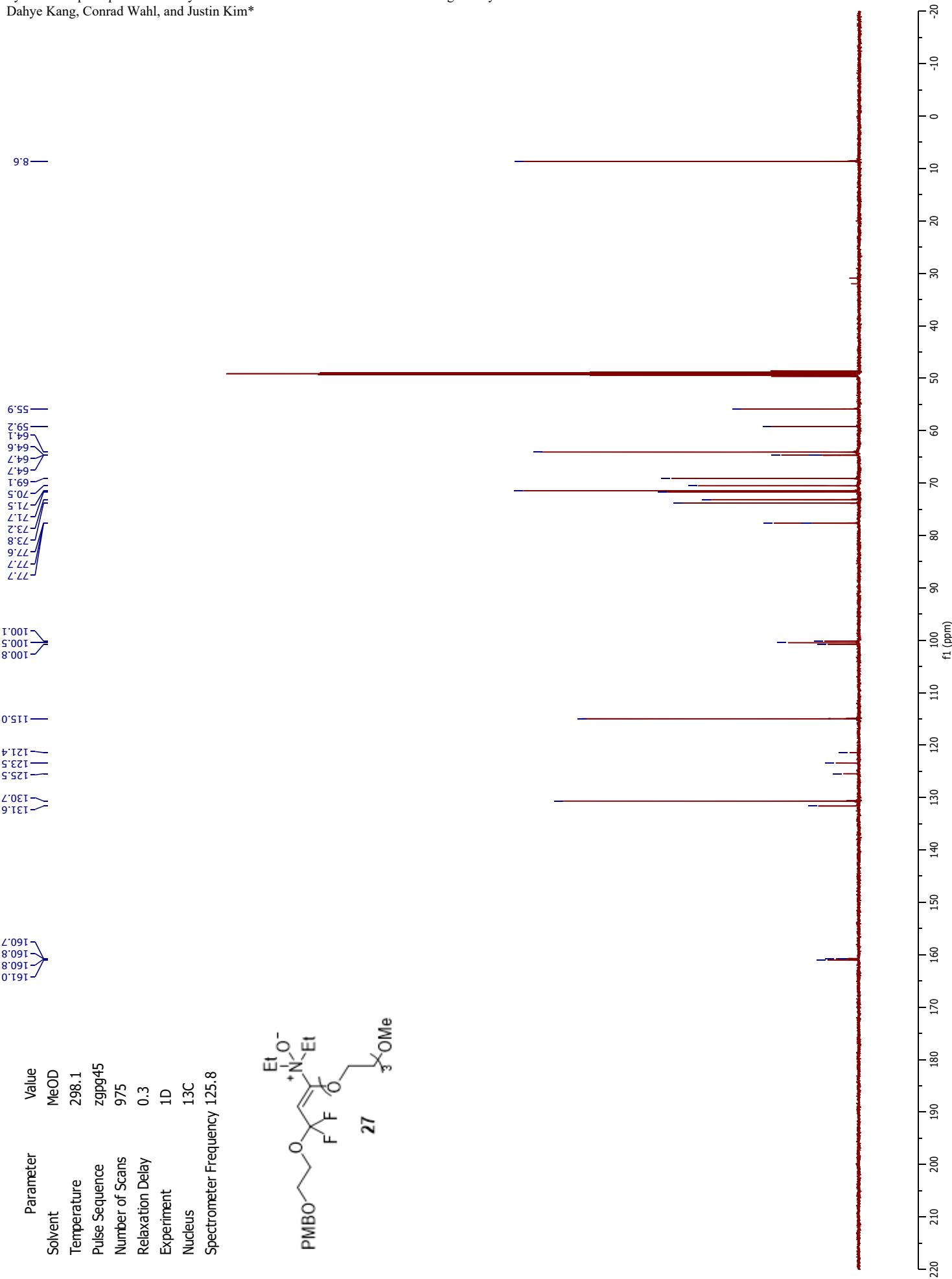
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Experiment	1D
Nucleus	¹⁹ F
Spectrometer Frequency	470.5



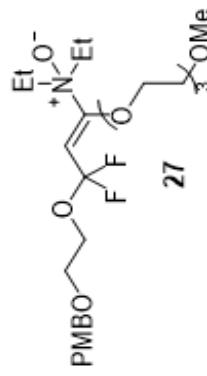
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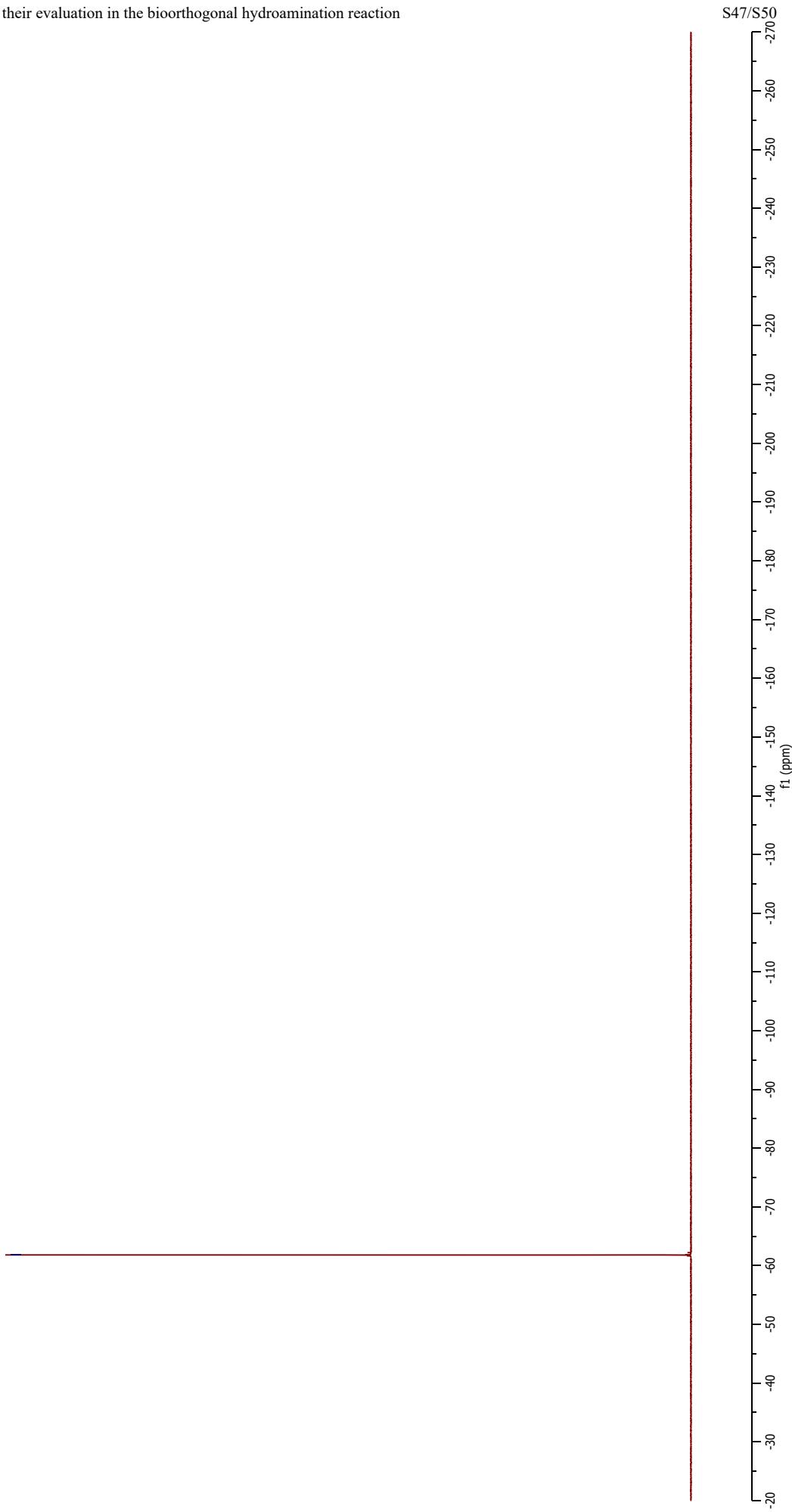


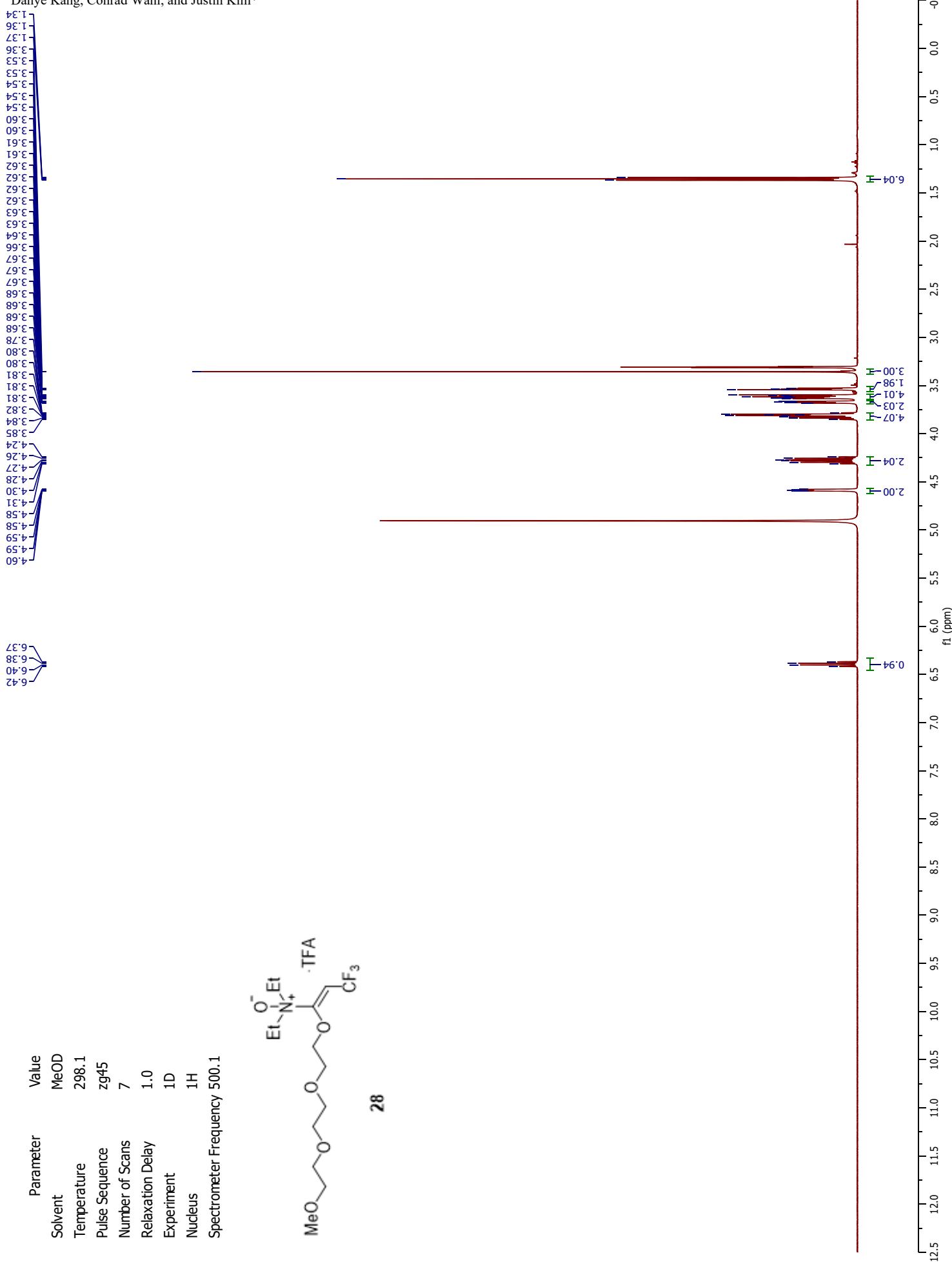


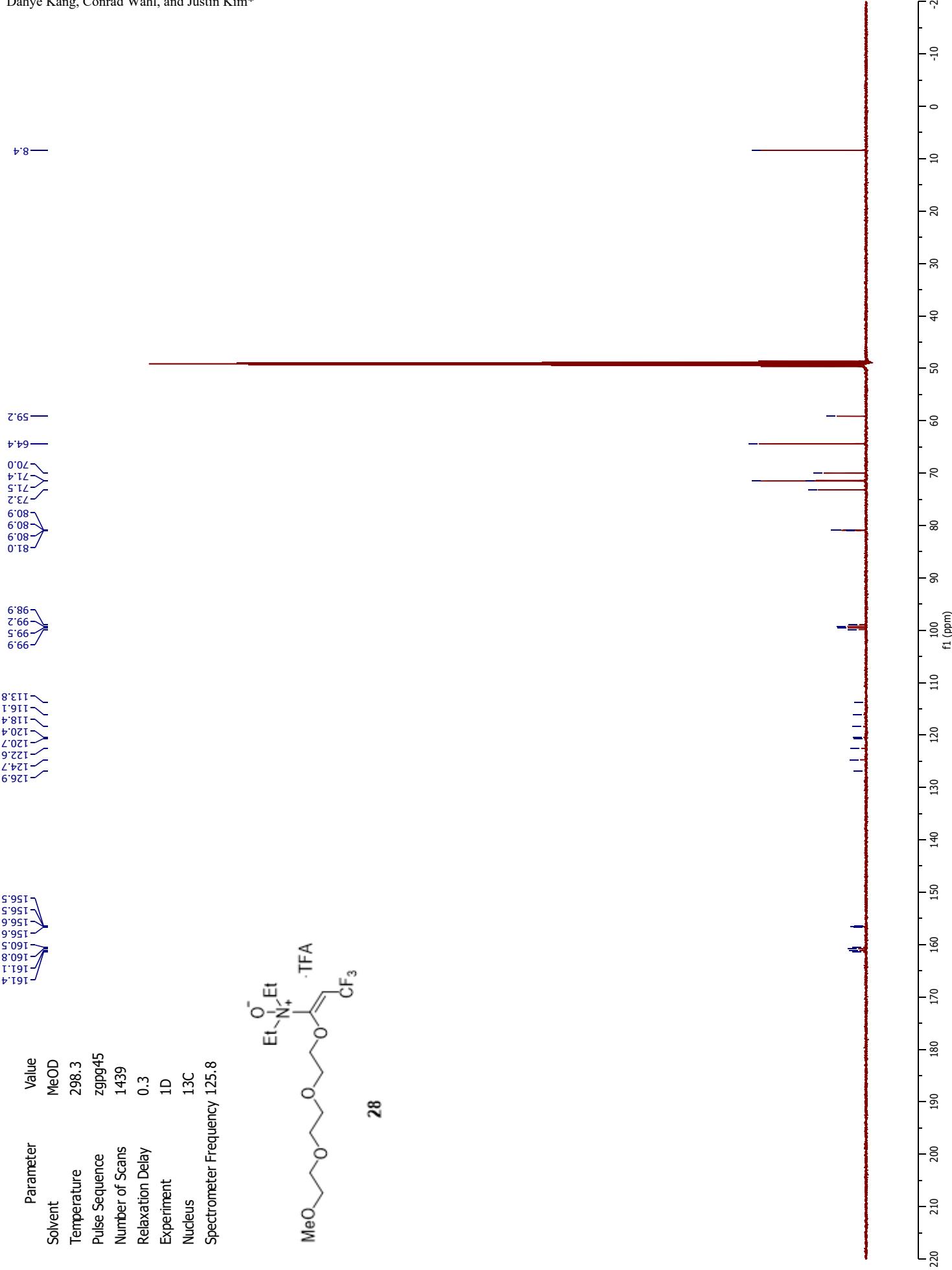
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Solvent	MeOD
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Experiment	1D
Nucleus	19F
Spectrometer Frequency	470.5

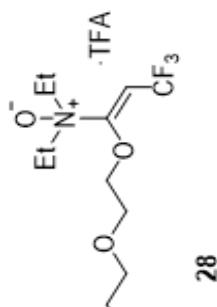


—61.8









Parameter	Value
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Experiment	1D
Nucleus	19F
Spectrometer Frequency	470.5

-77.5

-56.3

