

Supporting Information

For

Modulation of Chain-Breaking Antioxidant Activity of Phenolic Organochalcogens at Various pH and Co-antioxidants

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Figure S1. The R_{inh} profile of antioxidant **8** using NAC as co-antioxidant at pH 1-7 (HCl/NaOH regulated).

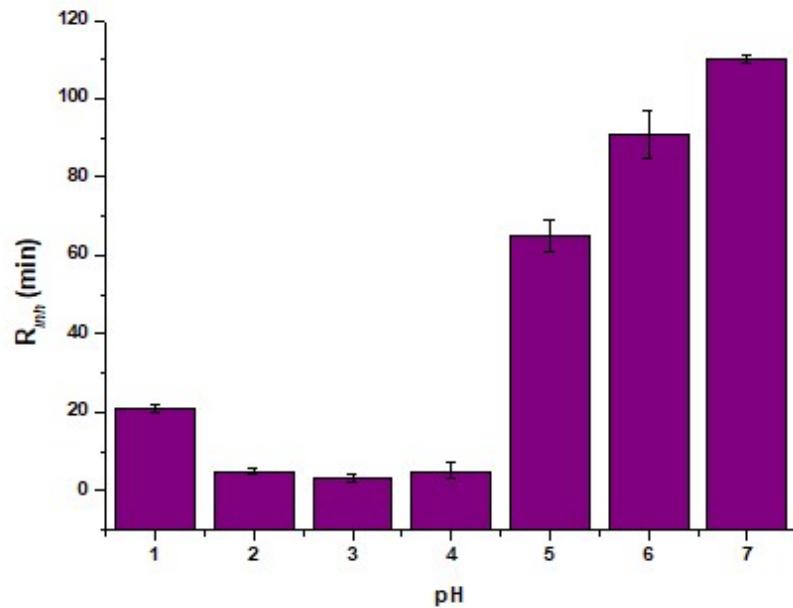


Figure S2. The T_{inh} profile of antioxidant **8** using NAC as co-antioxidant at pH 1-7 (HCl/NaOH regulated).

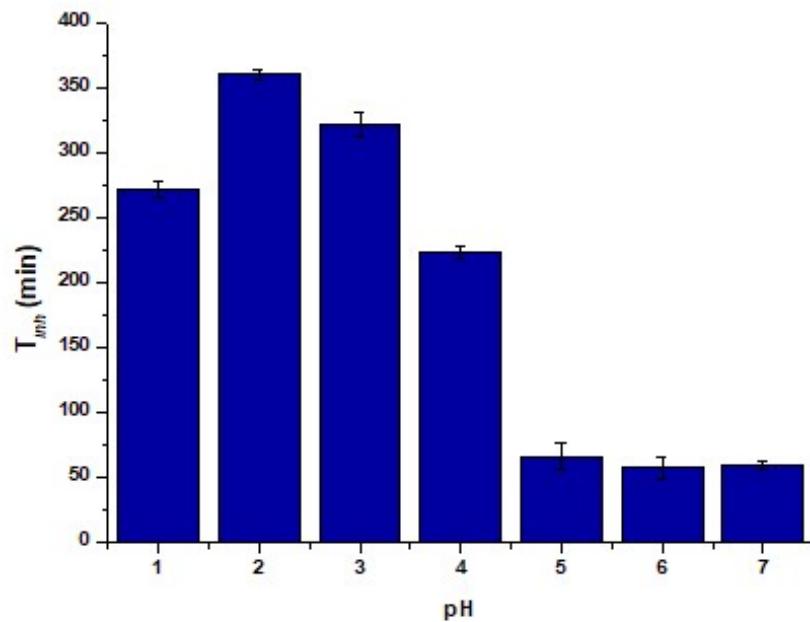


Figure S3. The R_{inh} profile of antioxidant **9** using NAC as co-antioxidant at pH 1-7 (HCl/NaOH regulated)

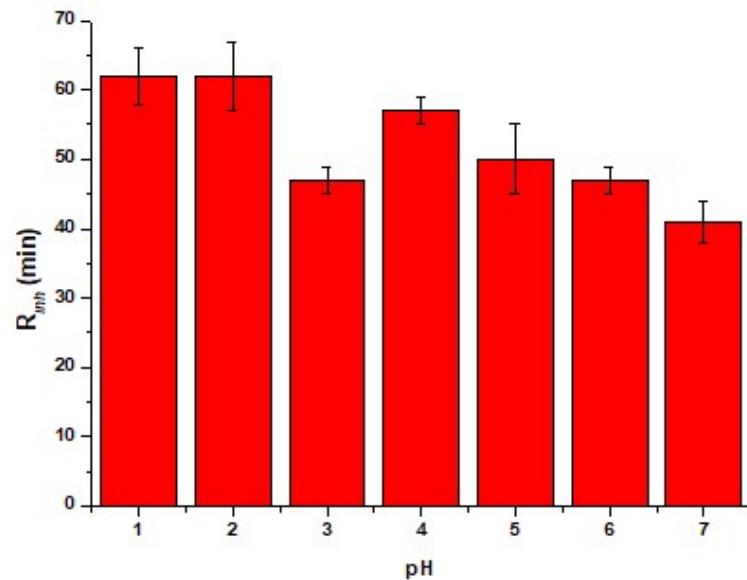


Figure S4. The T_{inh} profile of antioxidant **9** using NAC as co-antioxidant at pH 1-7 (HCl/NaOH regulated).

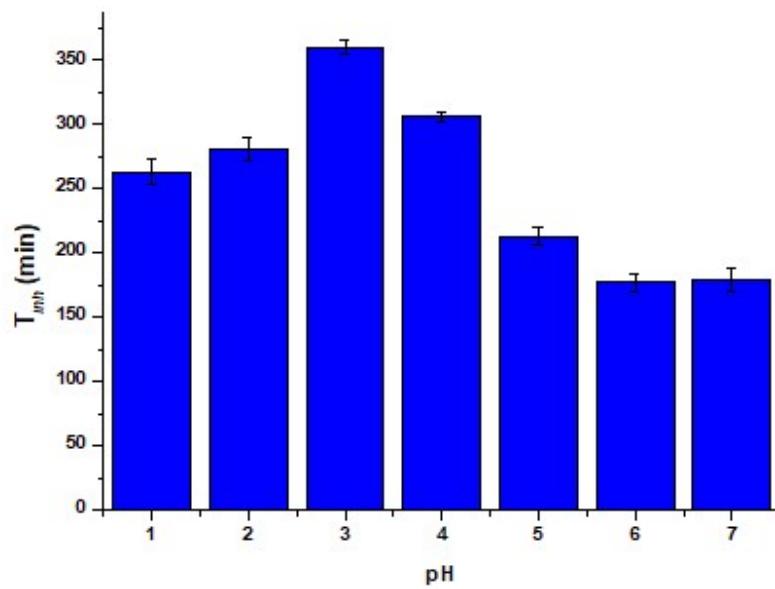


Figure S5. Control experiment without antioxidants (**8** and **9**) using NAC

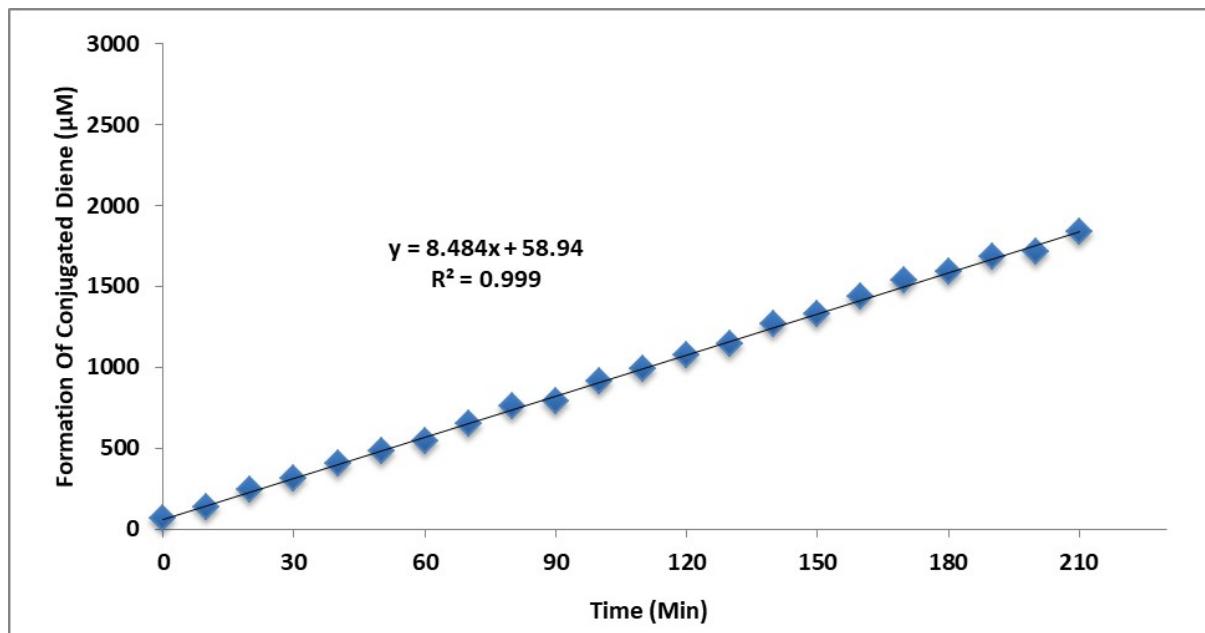


Figure S6. Control experiment without antioxidants (**8** and **9**) using GSH

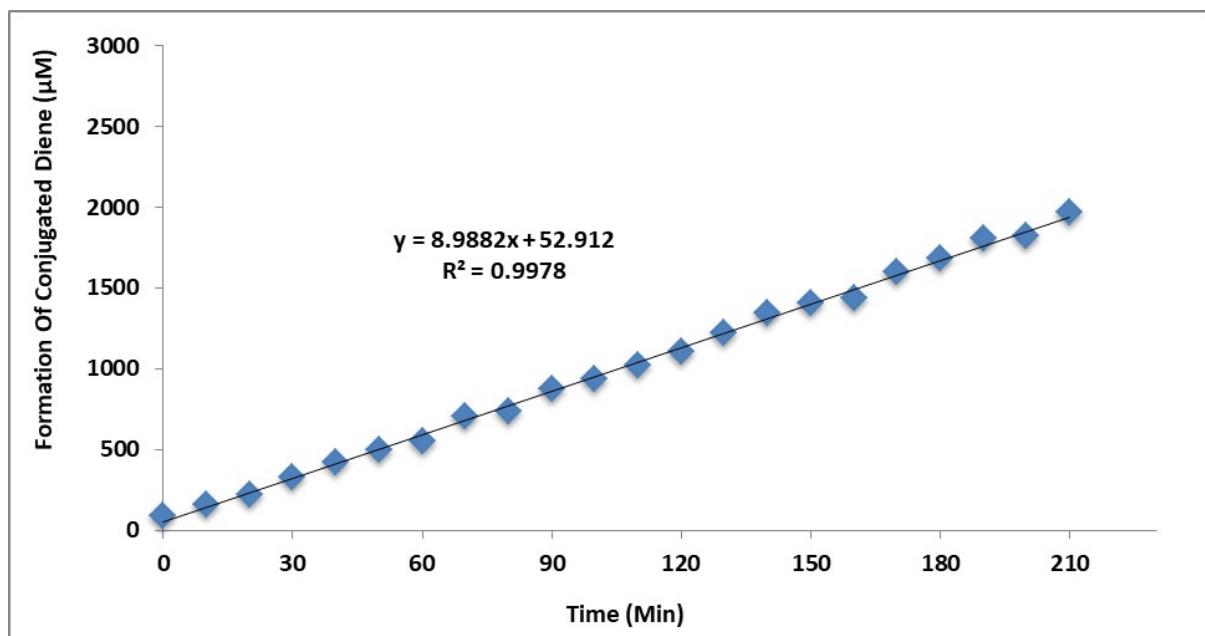


Figure S7. Control experiment without antioxidants (**8** and **9**) using DTT

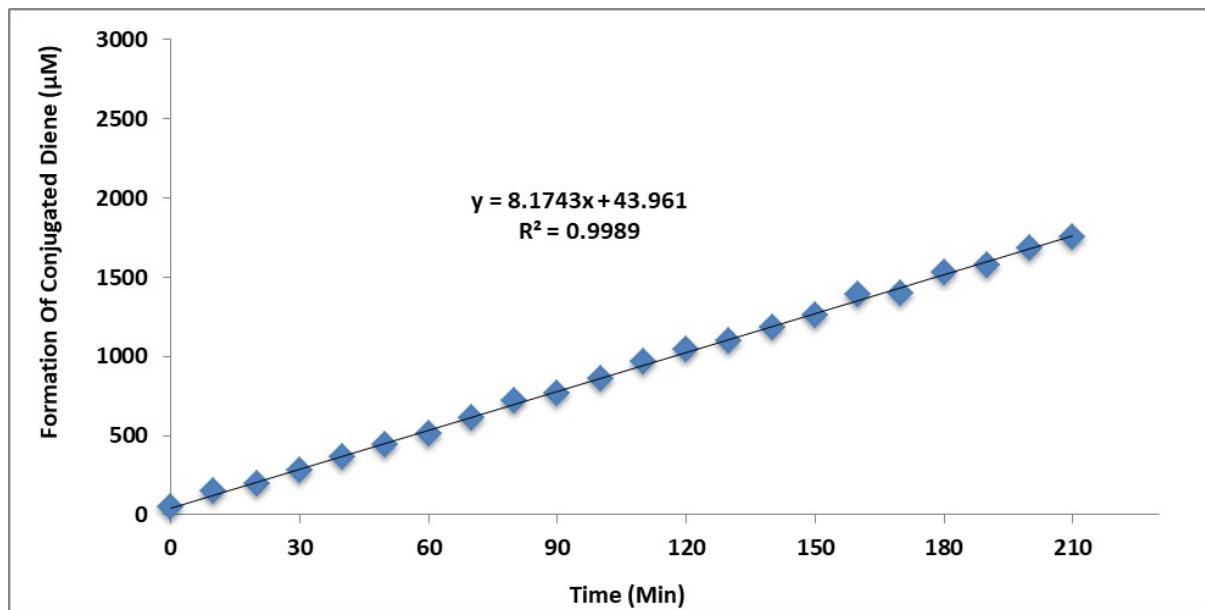


Figure S8. Control experiment without antioxidants (**8** and **9**) using AscOH

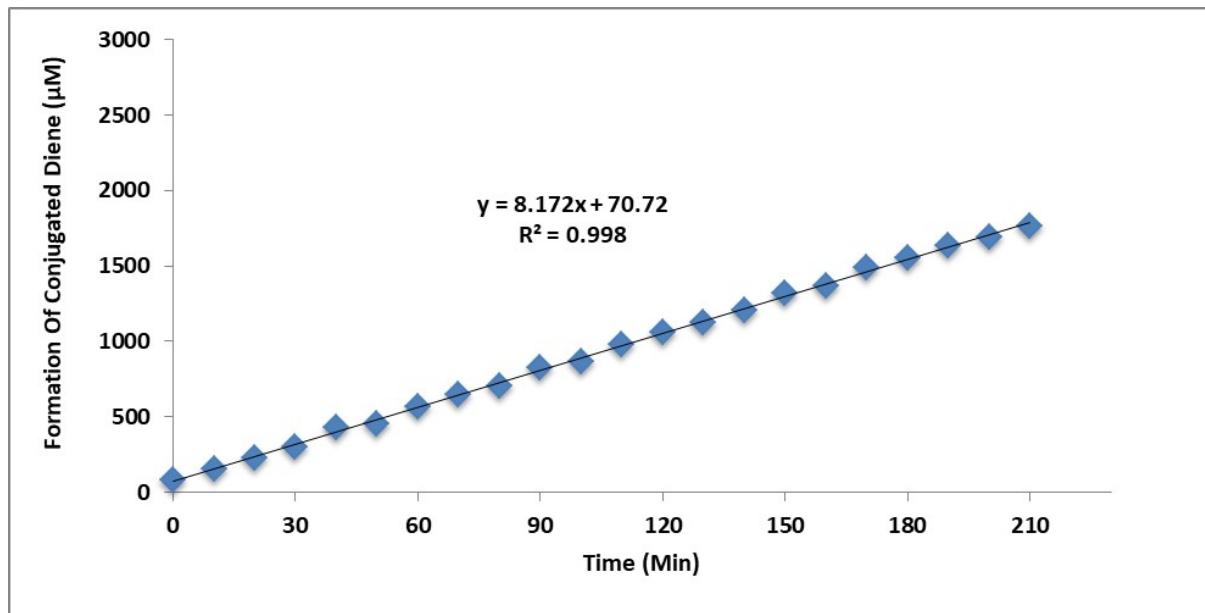


Figure S9. Control experiment without antioxidants (**8** and **9**) using sodium ascorbate

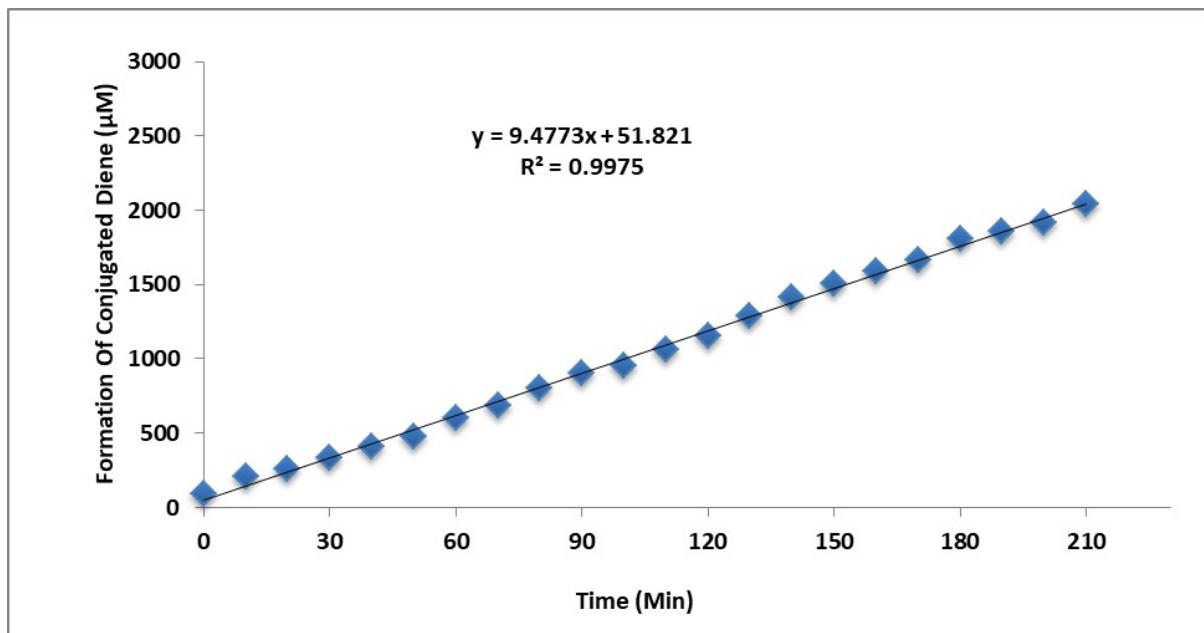


Figure S10. Control experiment without antioxidants (**8** and **9**) using $K_4[Fe(CN)_6] \cdot 3H_2O$

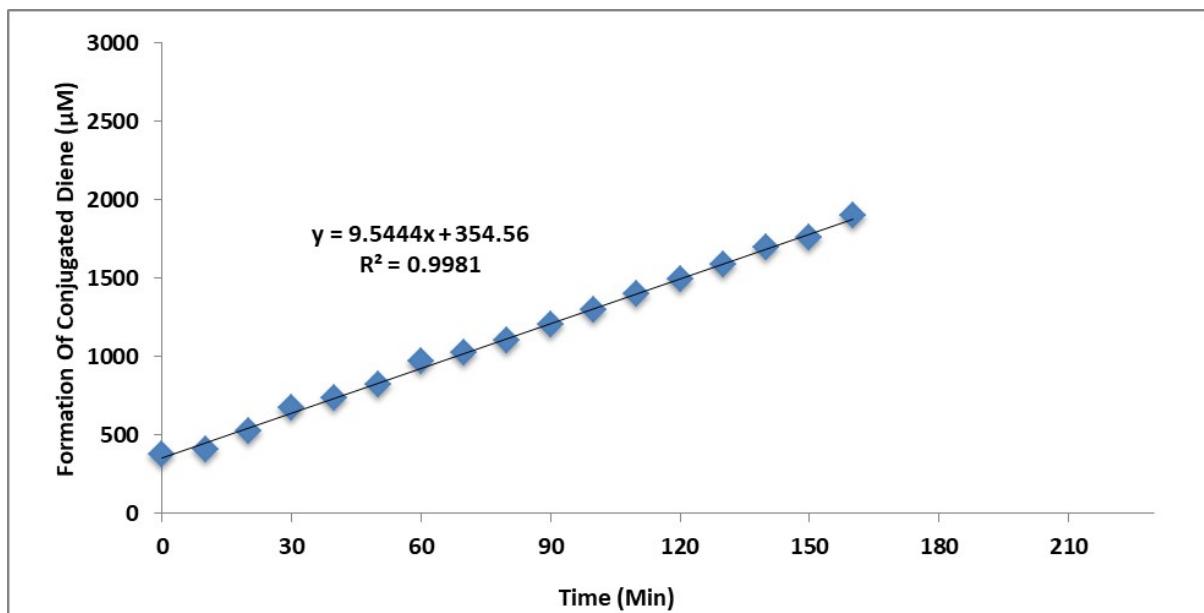


Figure S11. Control experiment without antioxidants (**8** and **9**) using $K_4[Fe(CN)_6].3H_2O$ + PTC

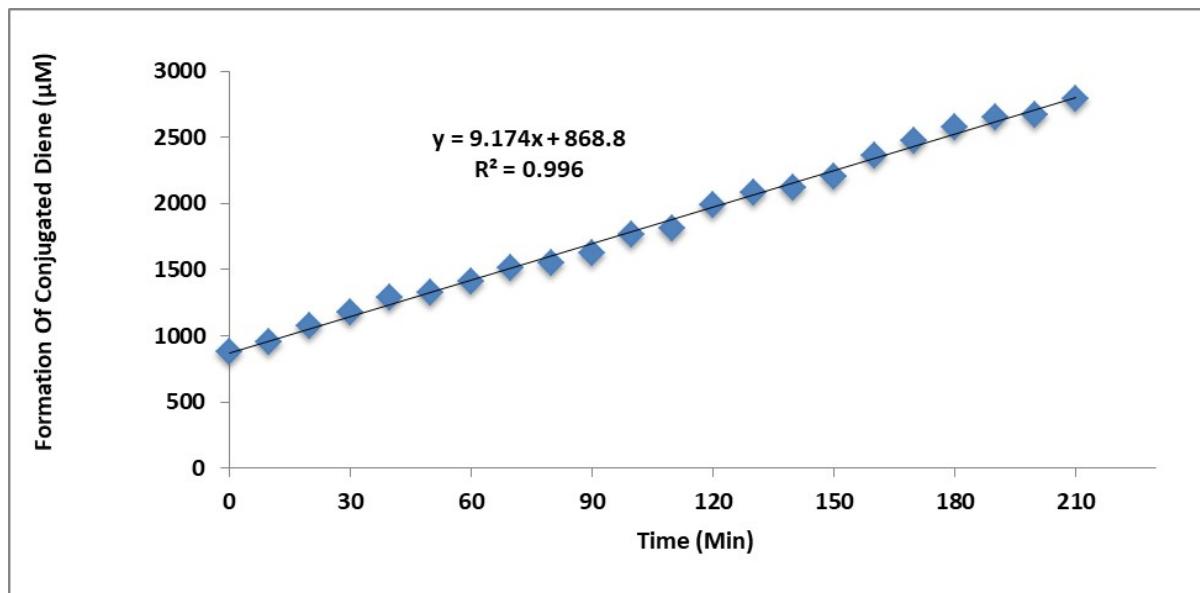


Figure S12. Control experiment without antioxidants (**8** and **9**) using $Na_2S_2O_5$

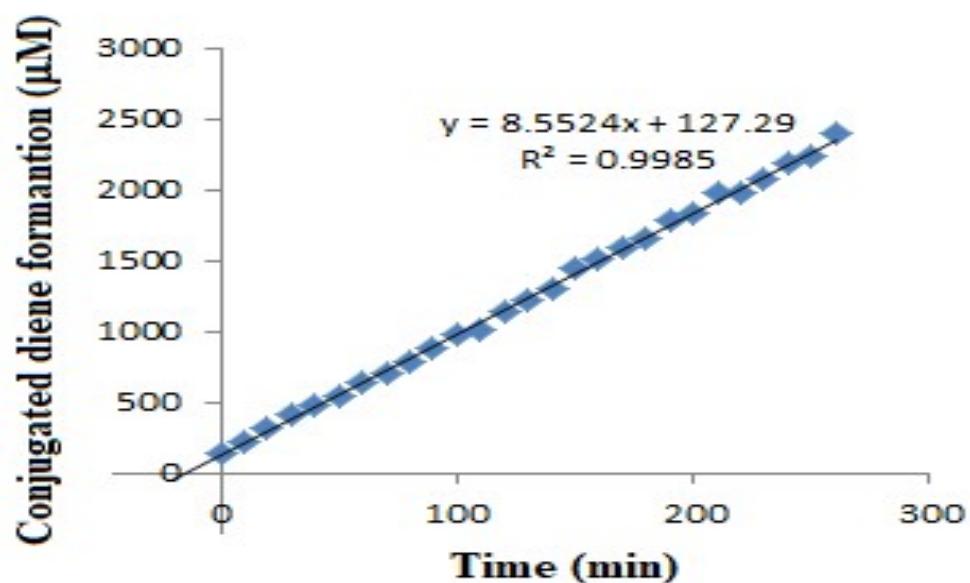


Figure S13. Control experiment without antioxidants (**8** and **9**) using $\text{Na}_2\text{S}_2\text{O}_5 + \text{PTC}$

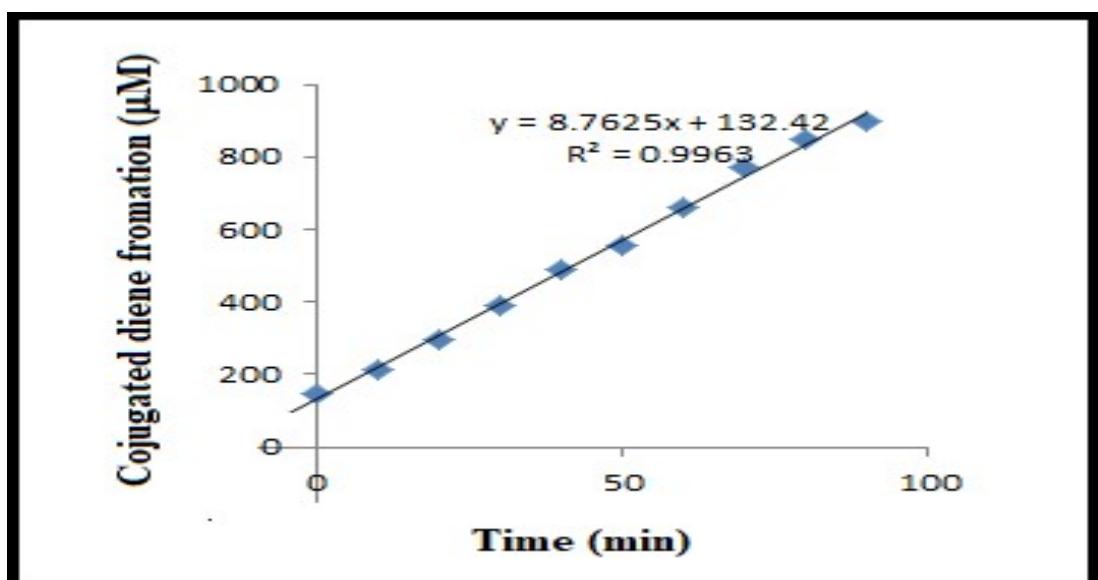


Figure S14. Control experiment without antioxidants (**8** and **9**) using TCEP

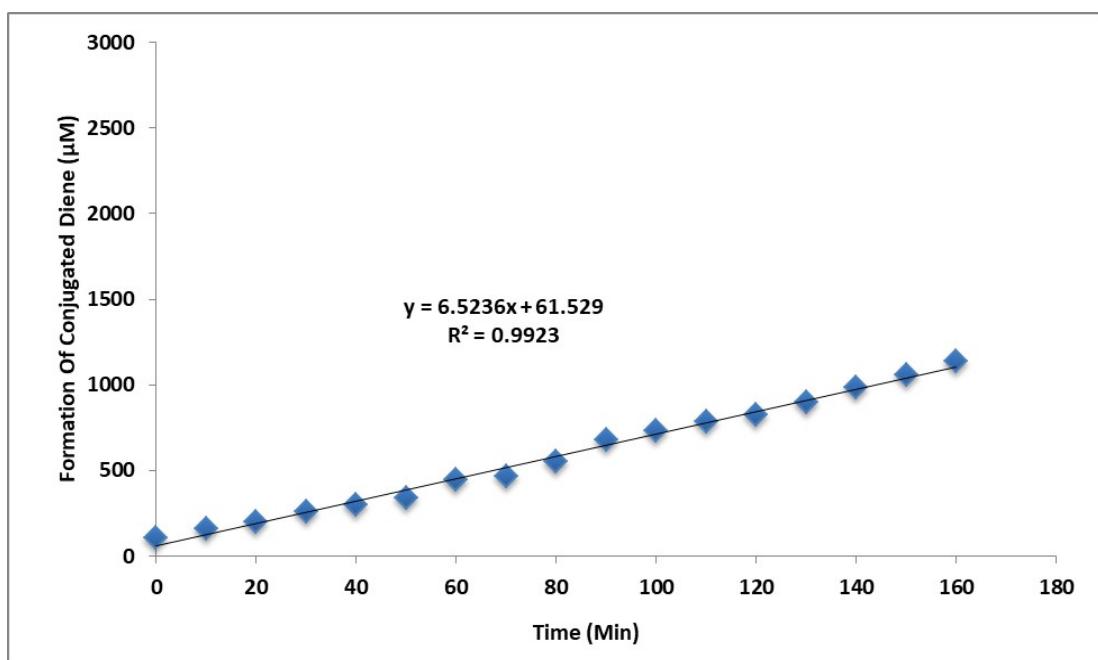


Figure S15. Experiment with antioxidant **8** using GSH

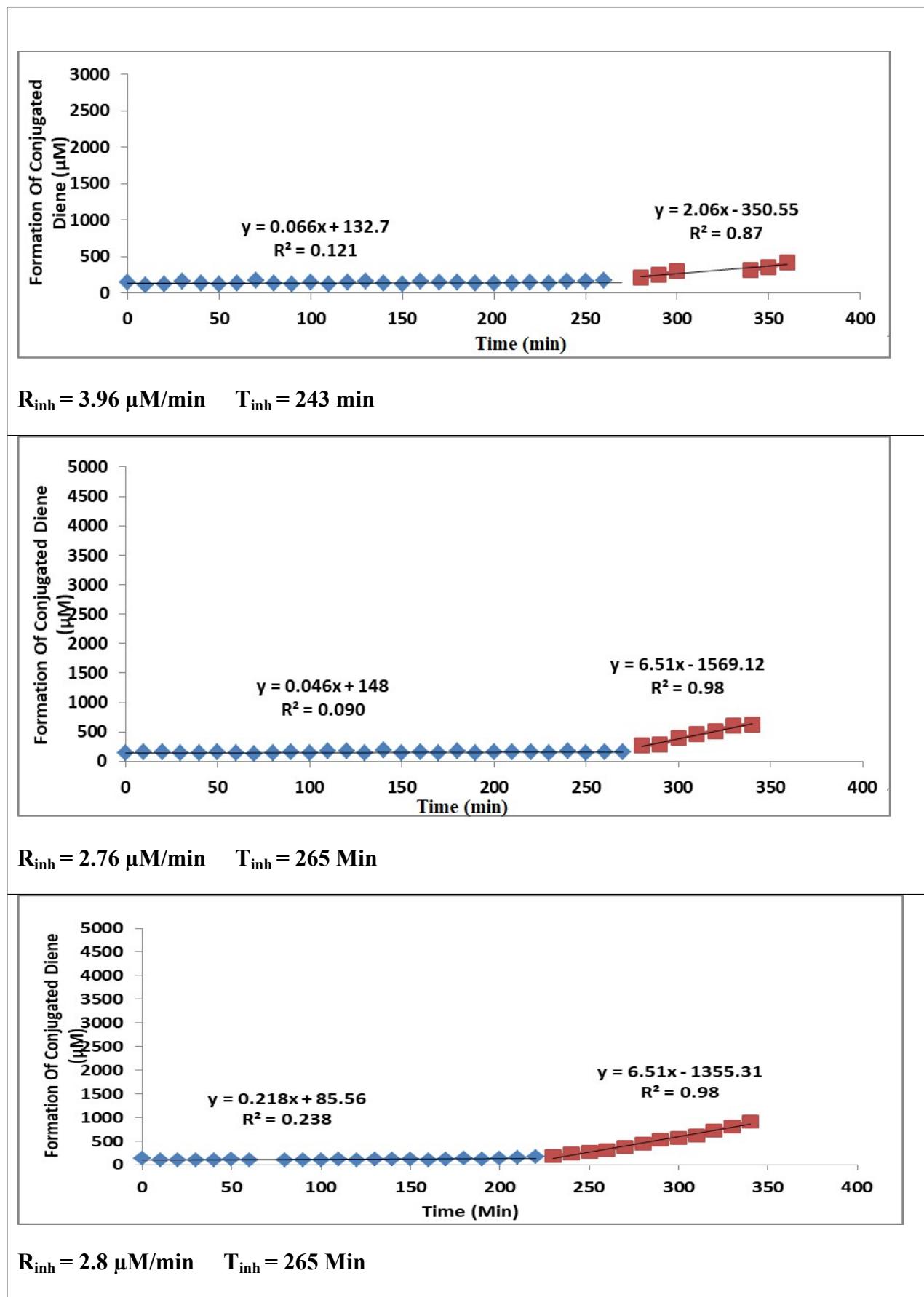


Figure S16. Experiment with antioxidant **8** using DTT

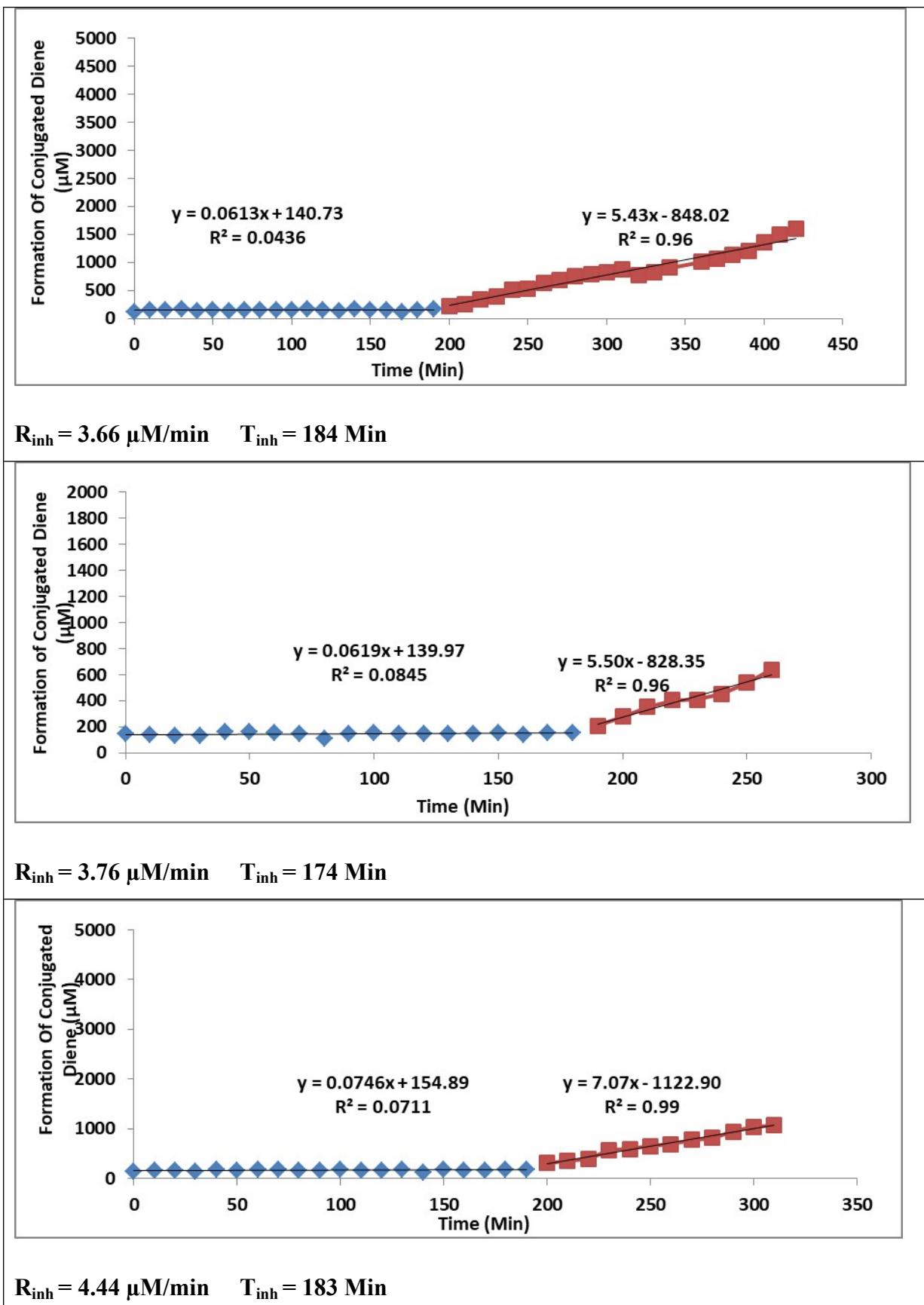
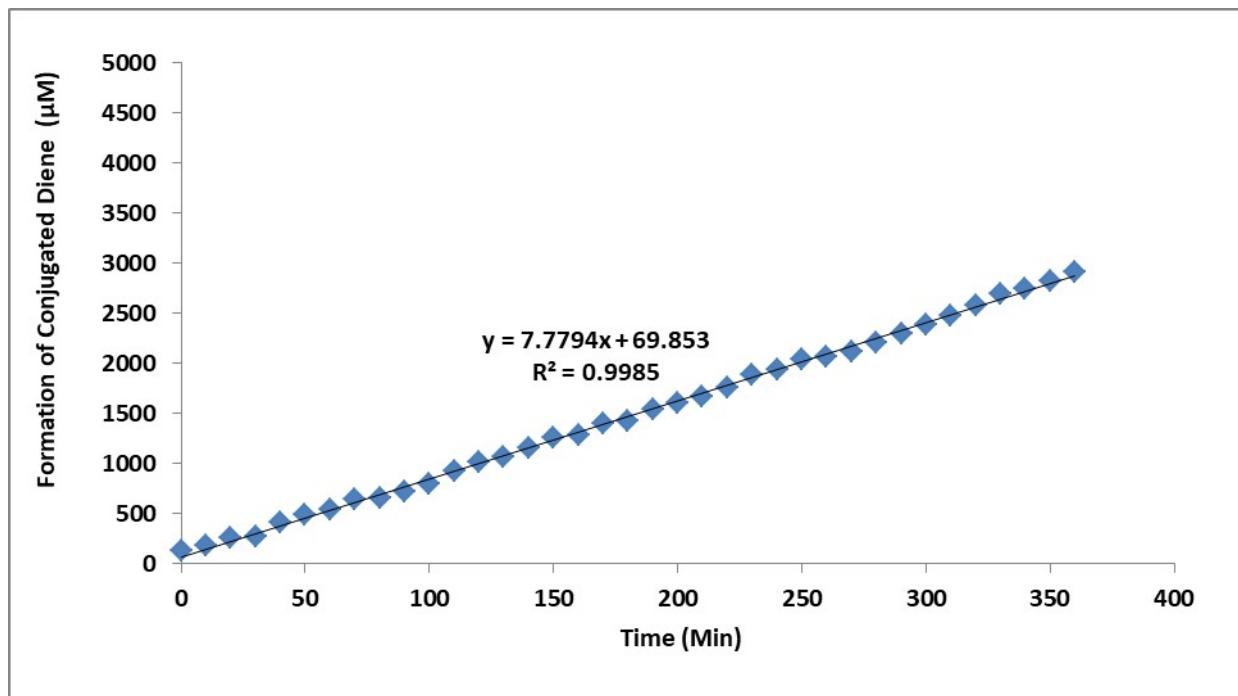
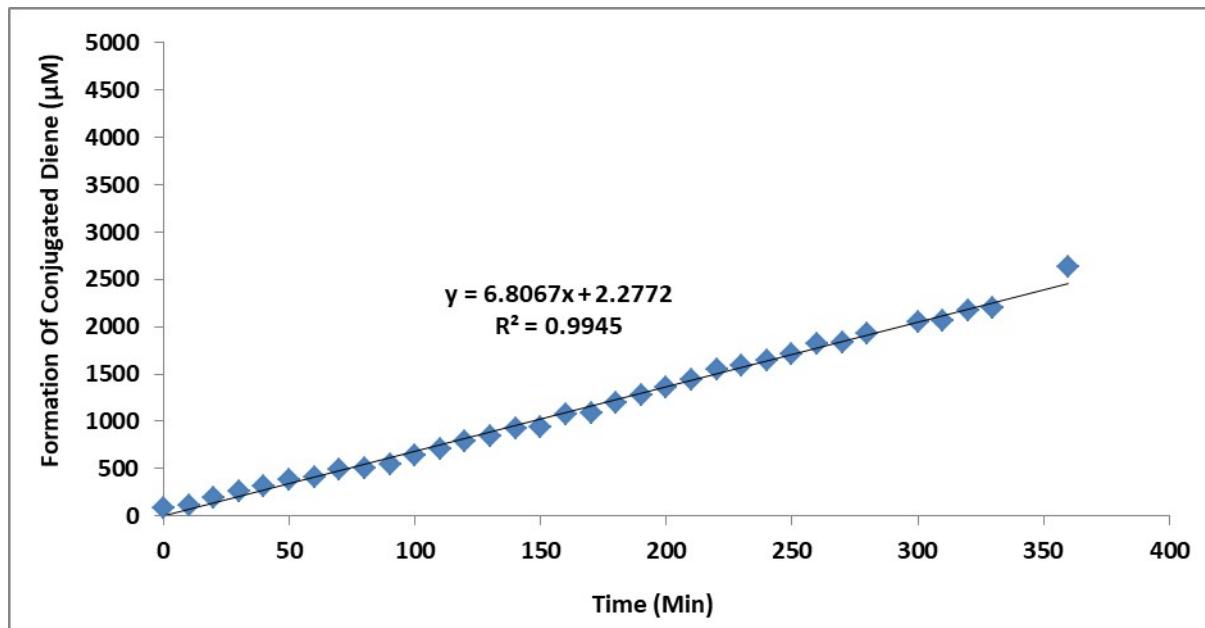


Figure S17. Experiment with antioxidant **8** using AscOH



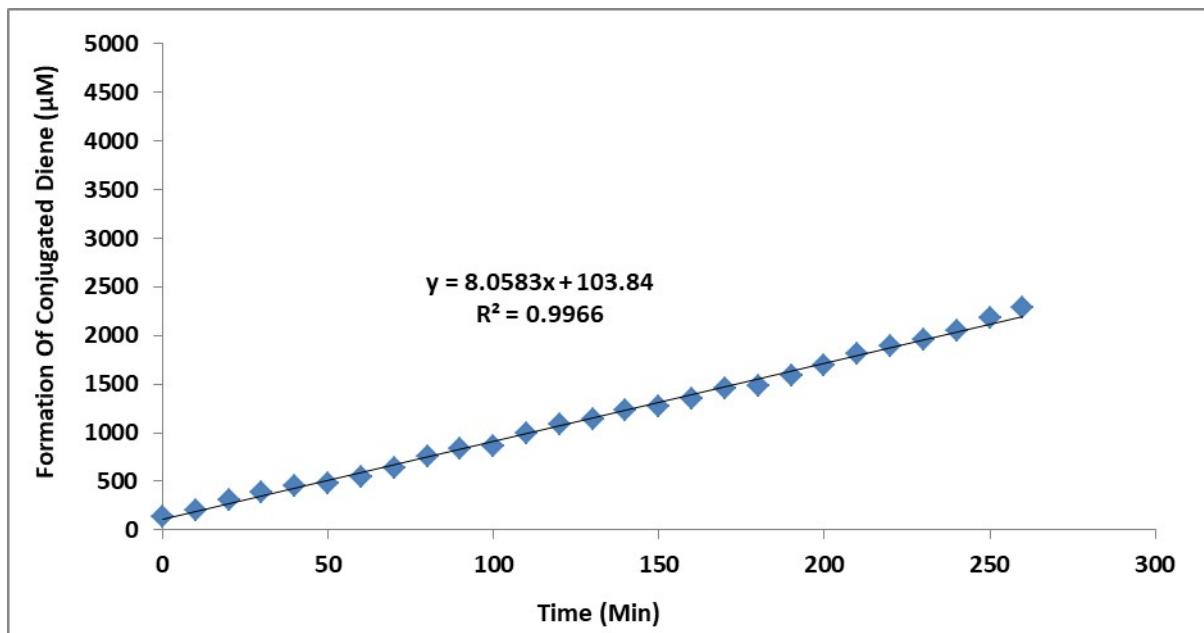
$$R_{\text{inh}} = 466.74 \mu\text{M}/\text{min}$$

Figure S18. Experiment with antioxidant **8** using sodium ascorbate



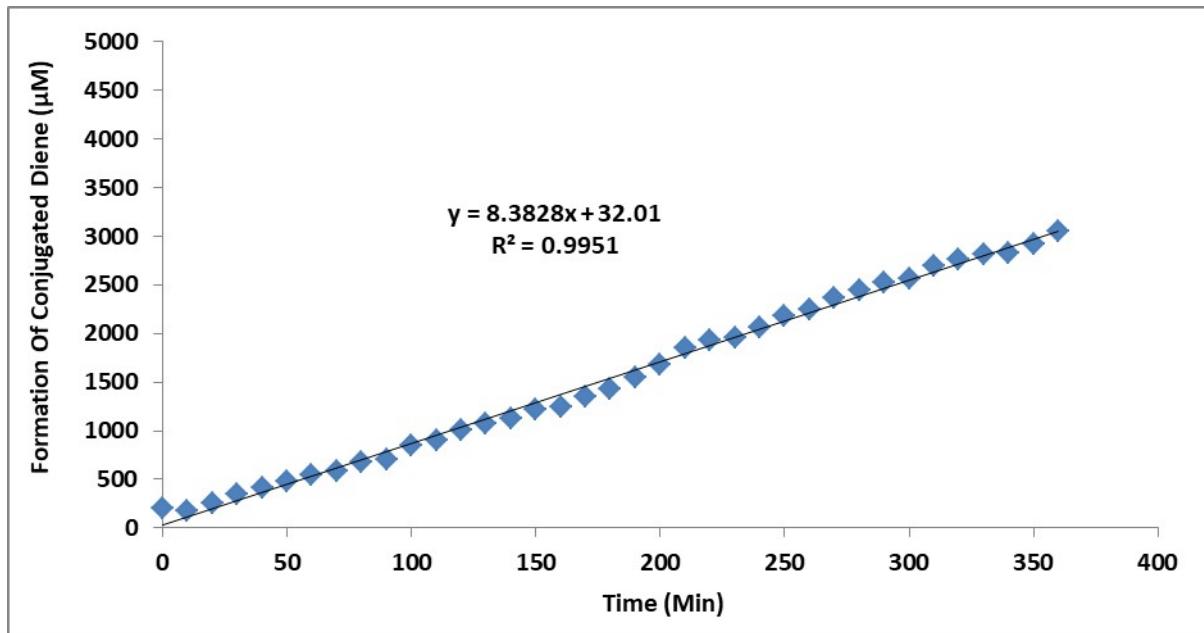
$$R_{\text{inh}} = 408 \mu\text{M}/\text{min}$$

Figure S19. Experiment with antioxidant **8** using $K_4[Fe(CN)_6].3H_2O$



$$R_{inh} = 484 \mu M/min$$

Figure S20. Experiment with antioxidant **8** using $K_4[Fe(CN)_6].3H_2O + PTC$



$$R_{inh} = 503 \mu M/min$$

Figure S21. Experiment with antioxidant **8** using PCET

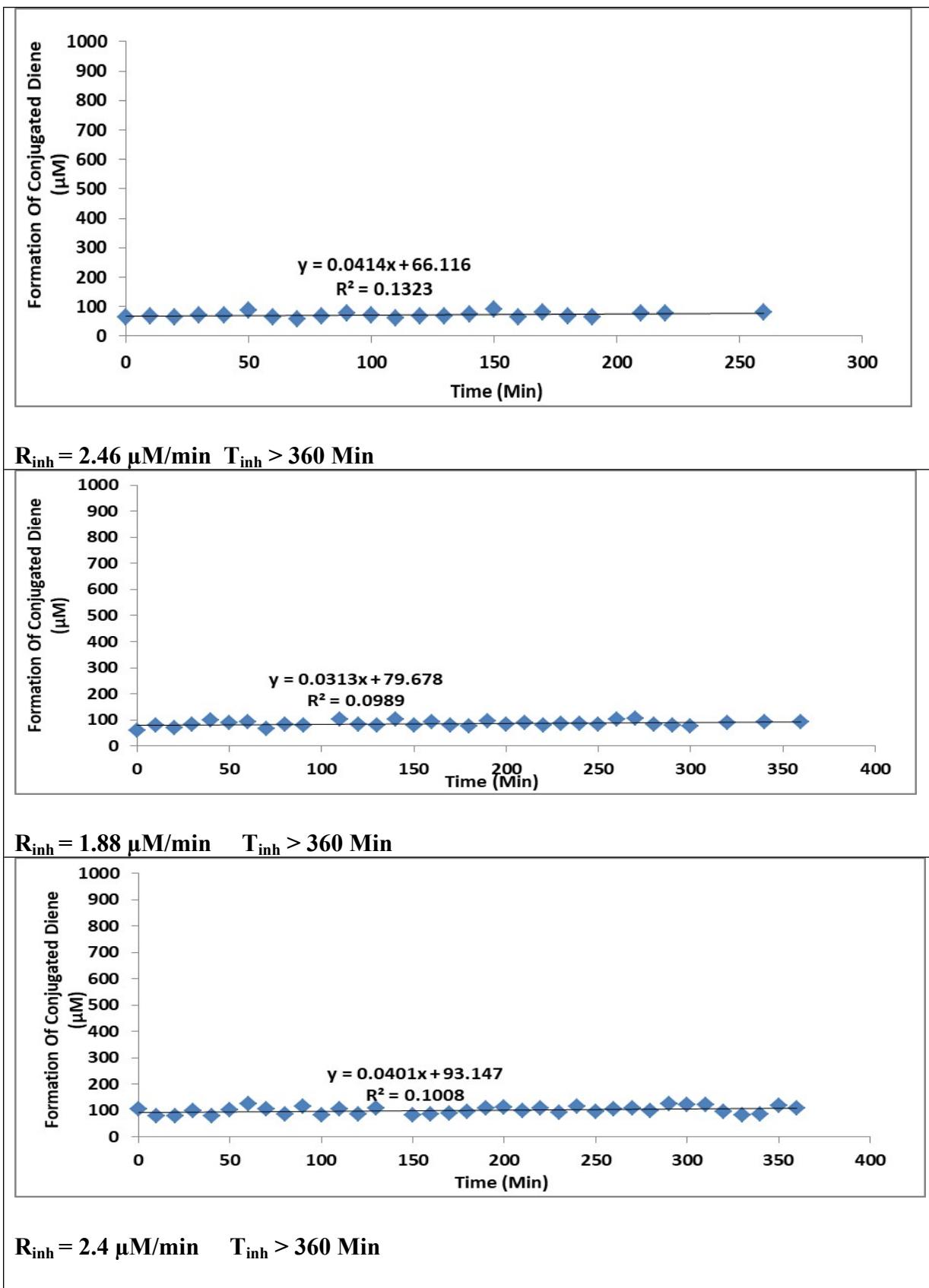
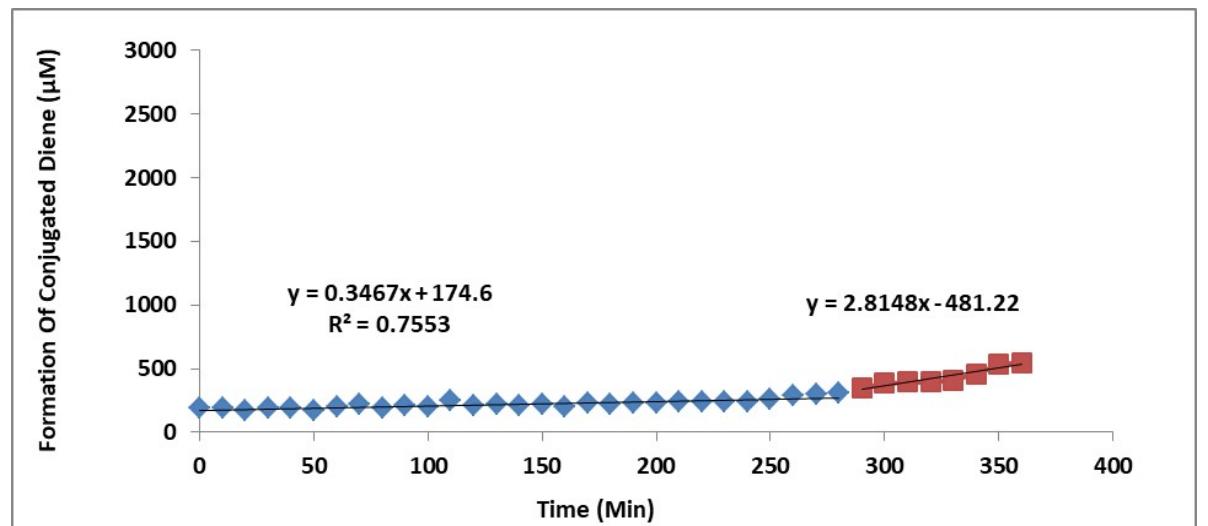
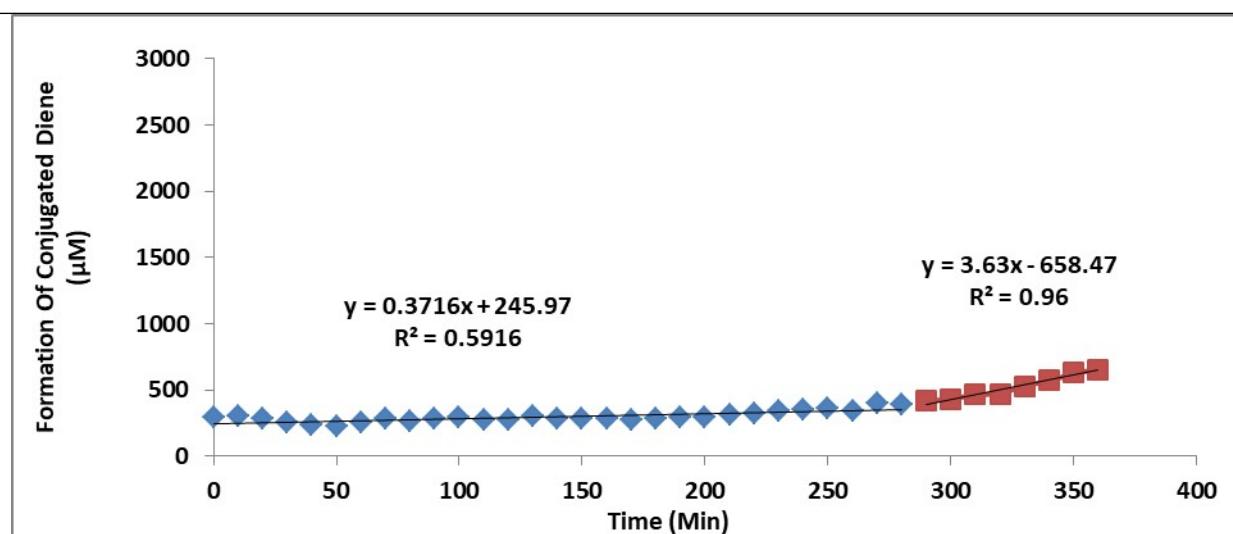


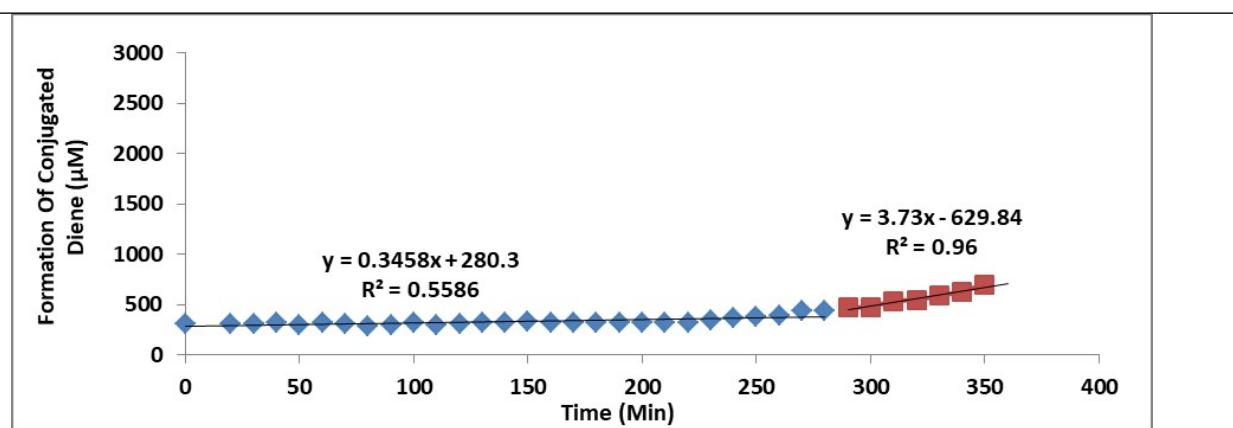
Figure S22. Experiment with antioxidant **8** using NAC pH = 1



$$R_{inh} = 20.8 \mu\text{M}/\text{Min} \quad T_{inh} = 266 \text{ Min}$$

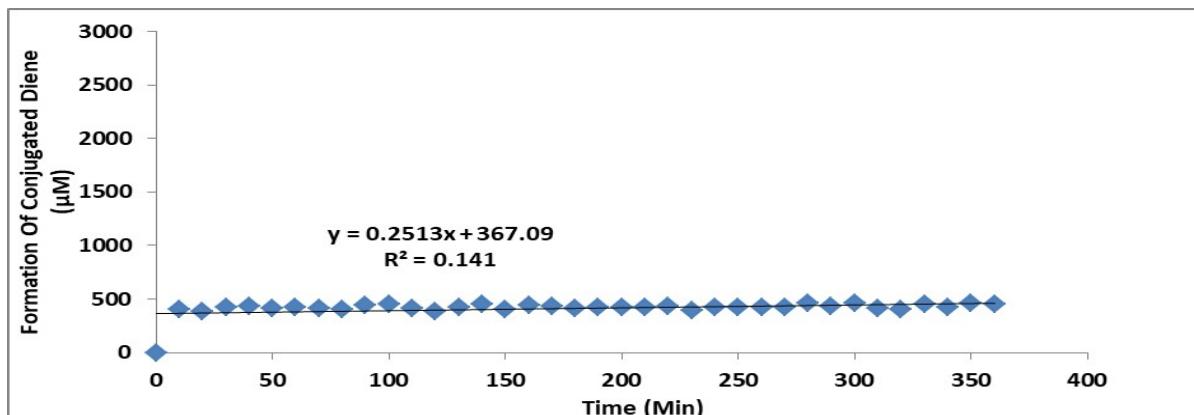


$$R_{inh} = 22.26 \mu\text{M}/\text{Min} \quad T_{inh} = 278 \text{ Min}$$

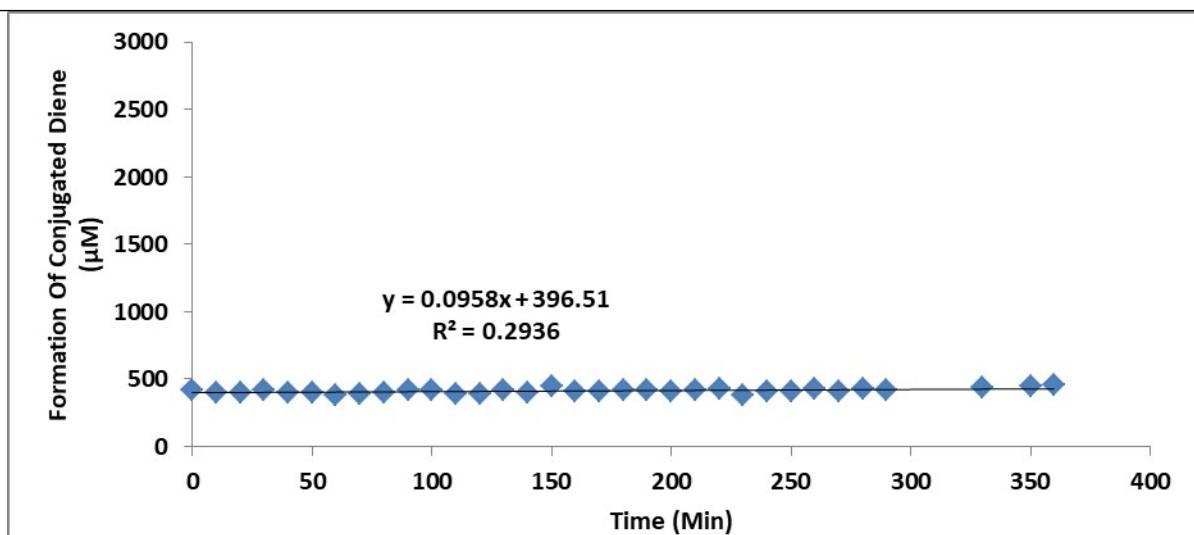


$$R_{inh} = 20.7 \mu\text{M}/\text{Min} \quad T_{inh} = 269 \text{ Min}$$

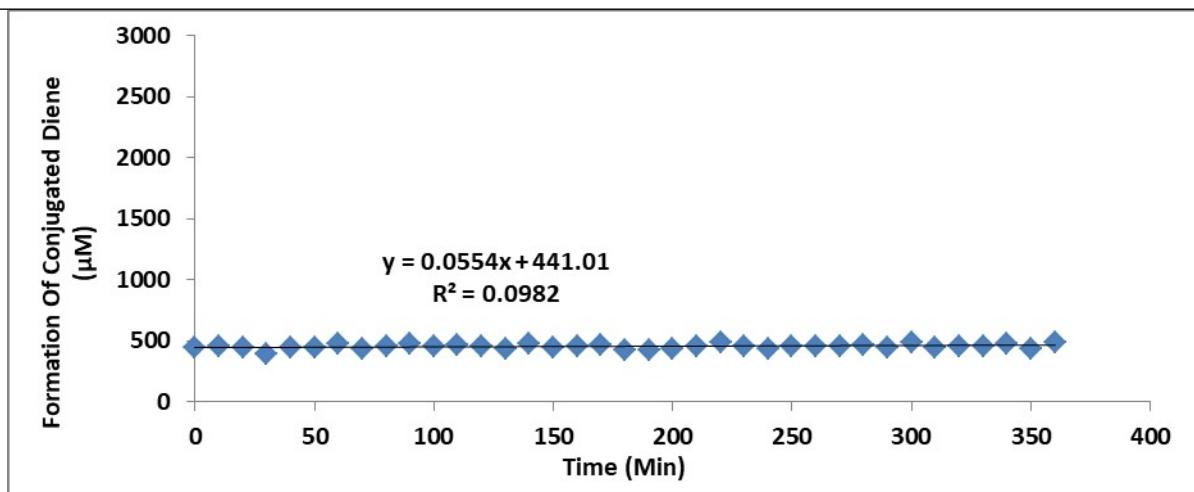
Figure S23. Experiment with antioxidant **8** using NAC pH = 2



$$R_{inh} = 5.06 \mu\text{M}/\text{Min} \quad T_{inh} > 360 \text{ Min}$$

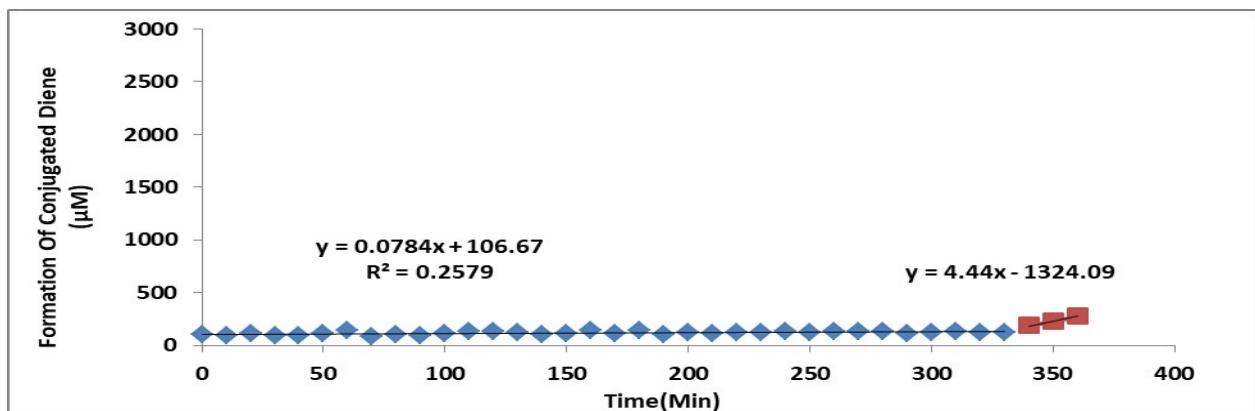


$$R_{inh} = 5.7 \mu\text{M}/\text{Min} \quad T_{inh} > 360 \text{ Min}$$

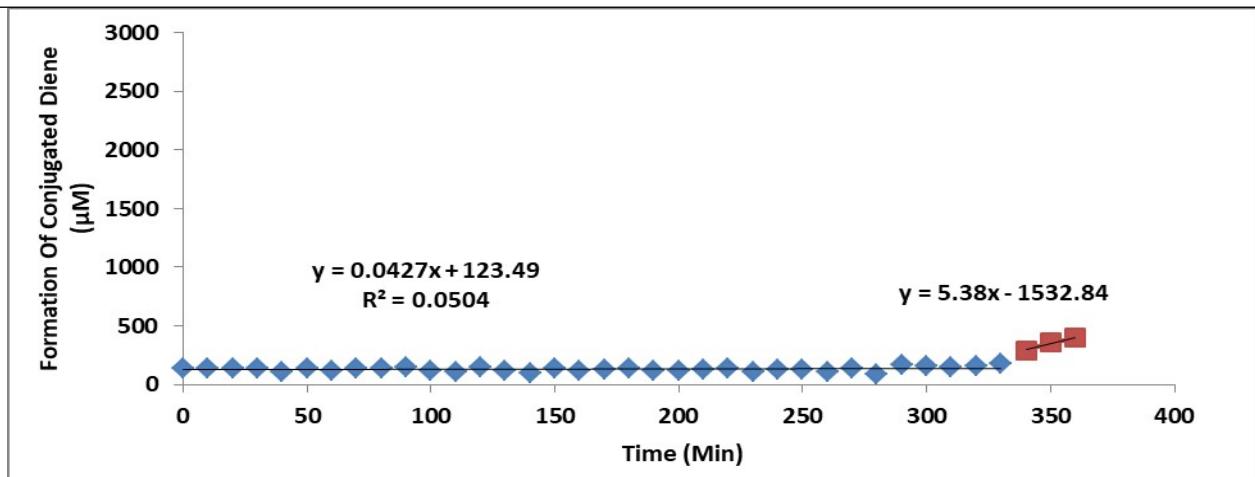


$$R_{inh} = 3.3 \mu\text{M}/\text{Min} \quad T_{inh} > 360 \text{ Min}$$

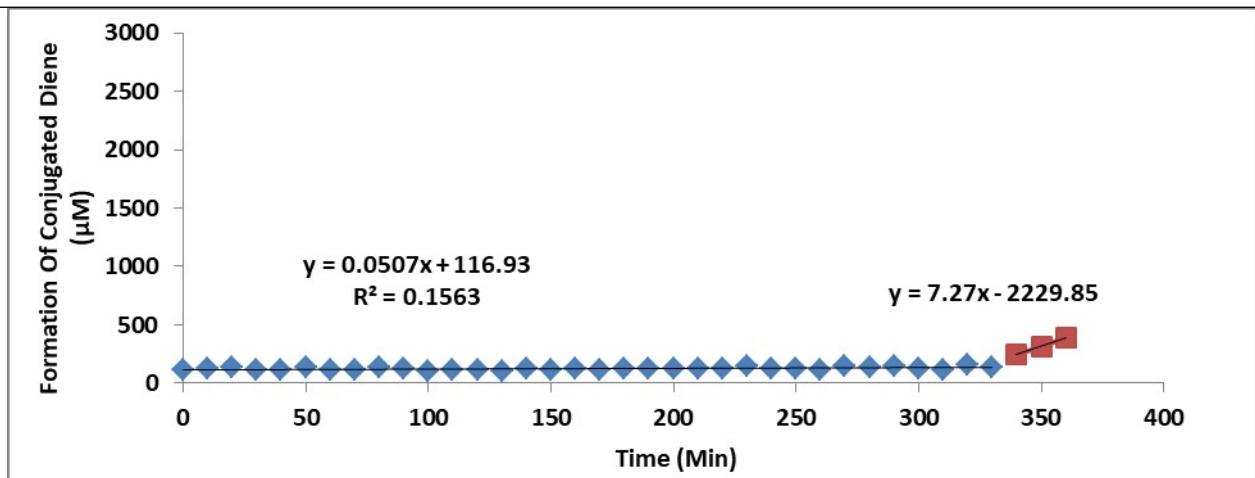
Figure S24. Experiment with antioxidant **8** using NAC pH = 3



$$R_{\text{inh}}=4.7 \mu\text{M}/\text{Min} \quad T_{\text{inh}}=328 \text{ Min}$$



$$R_{\text{inh}}=2.52 \mu\text{M}/\text{Min} \quad T_{\text{inh}}=311 \text{ Min}$$



$$R_{\text{inh}}=3.0 \mu\text{M}/\text{Min} \quad T_{\text{inh}}=325 \text{ Min}$$

Figure S25. Experiment with antioxidant **8** using NAC pH = 4

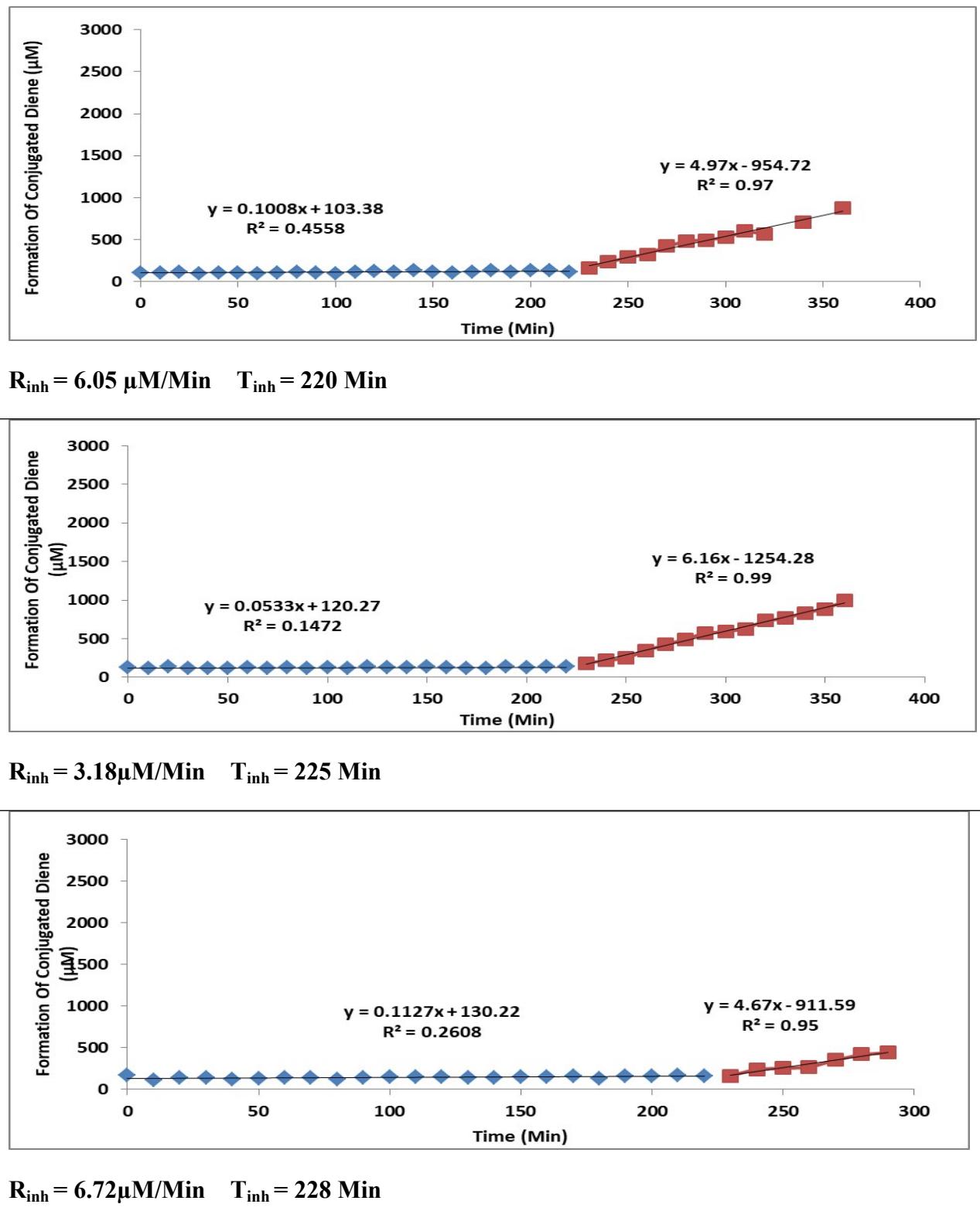


Figure S26. Experiment with antioxidant **8** using NAC pH = 5

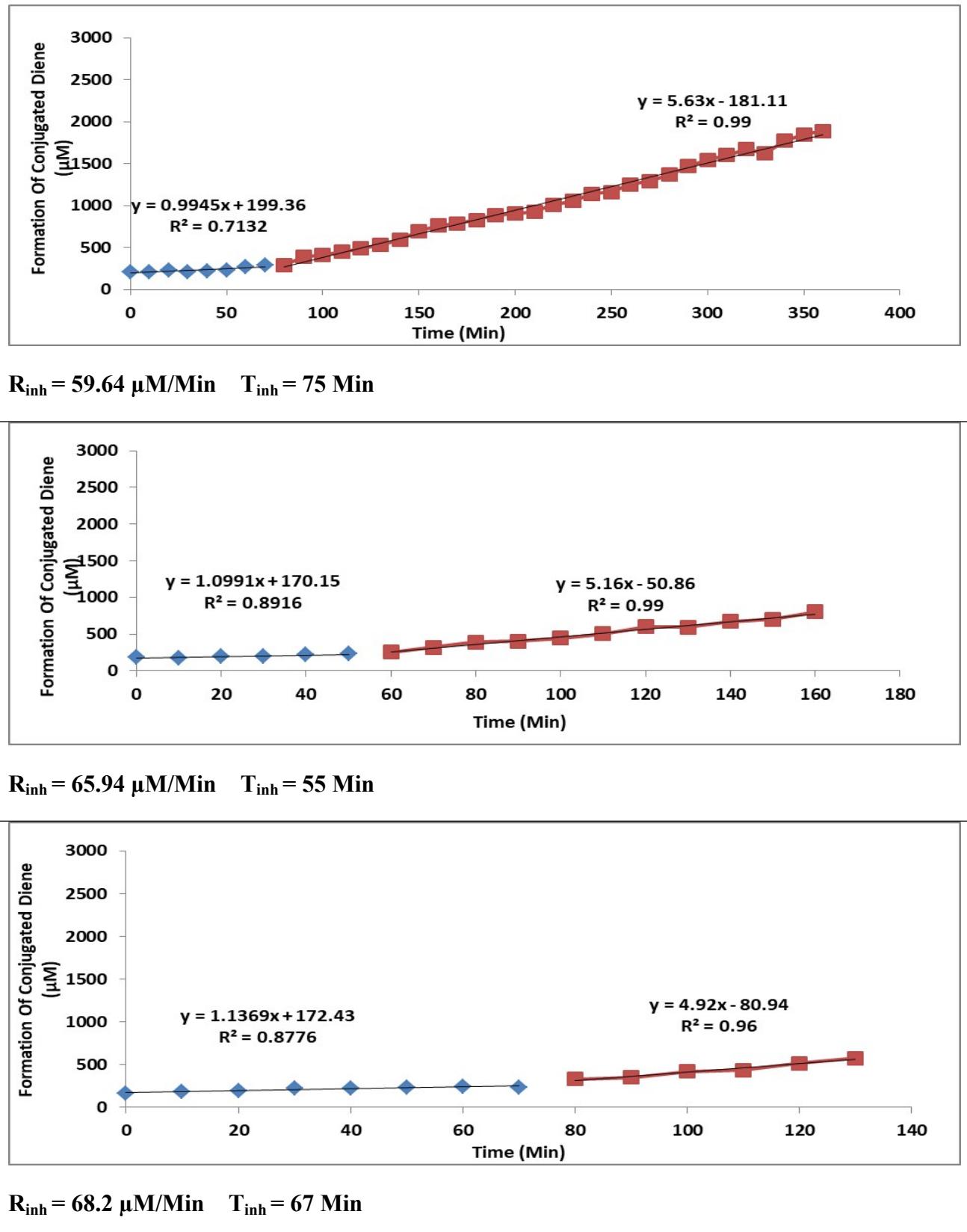


Figure S27. Experiment with antioxidant **8** using NAC pH = 6

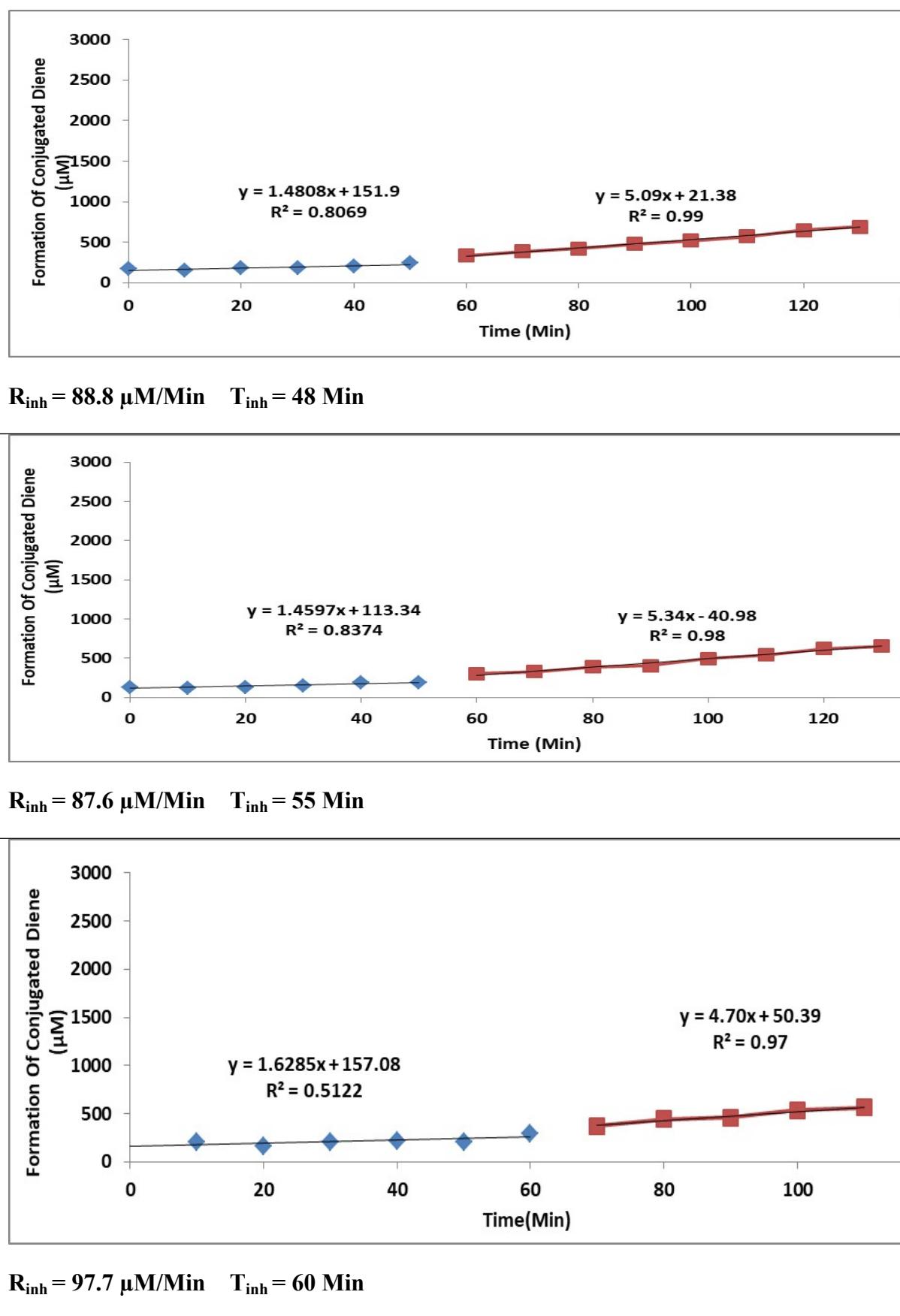
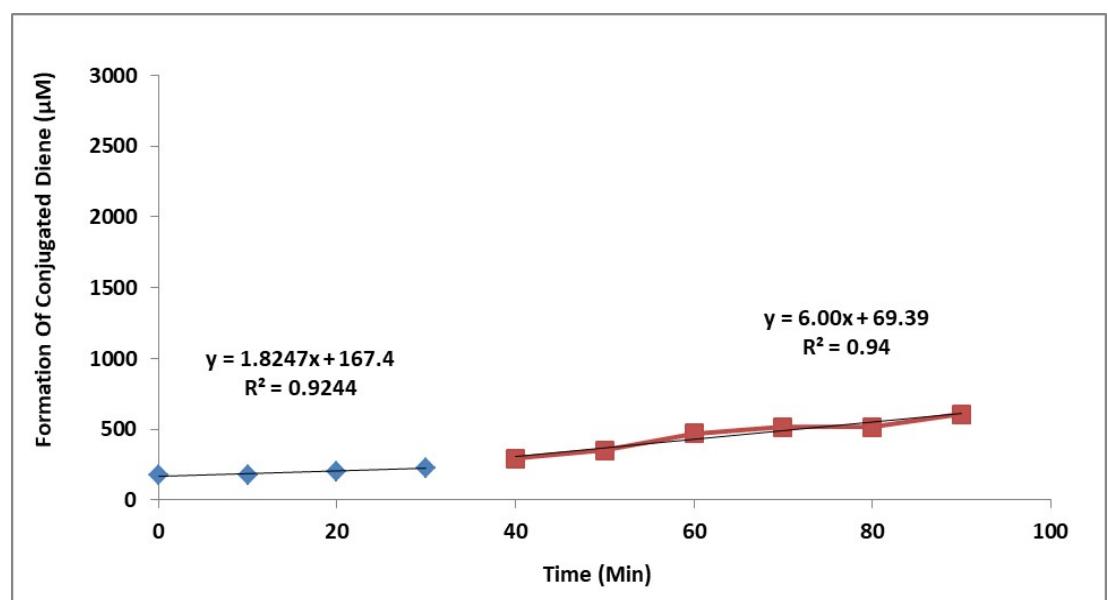
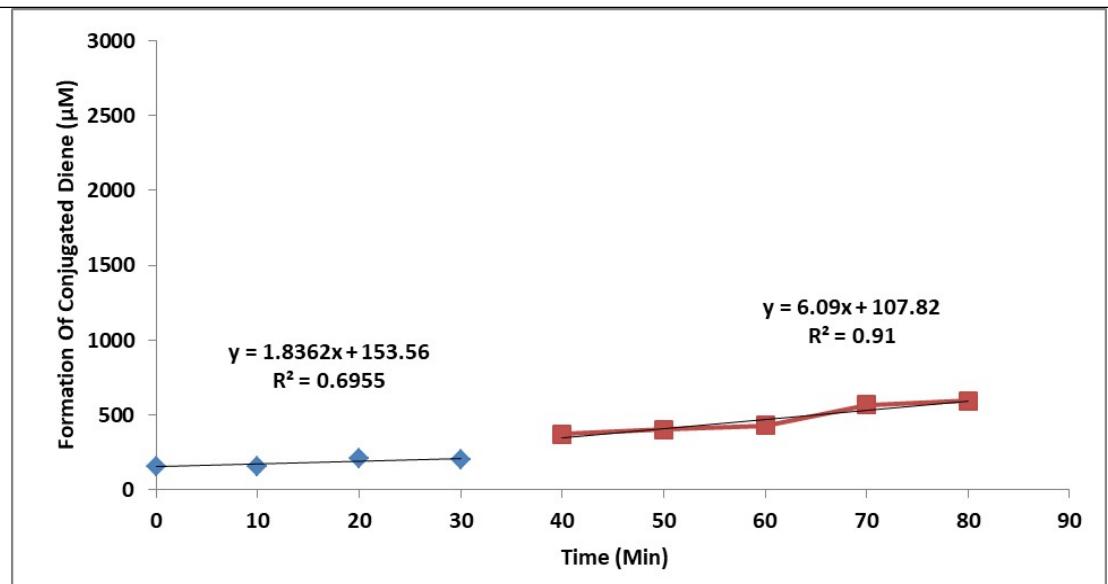


Figure S28. Experiment with antioxidant **8** using NAC pH = 7

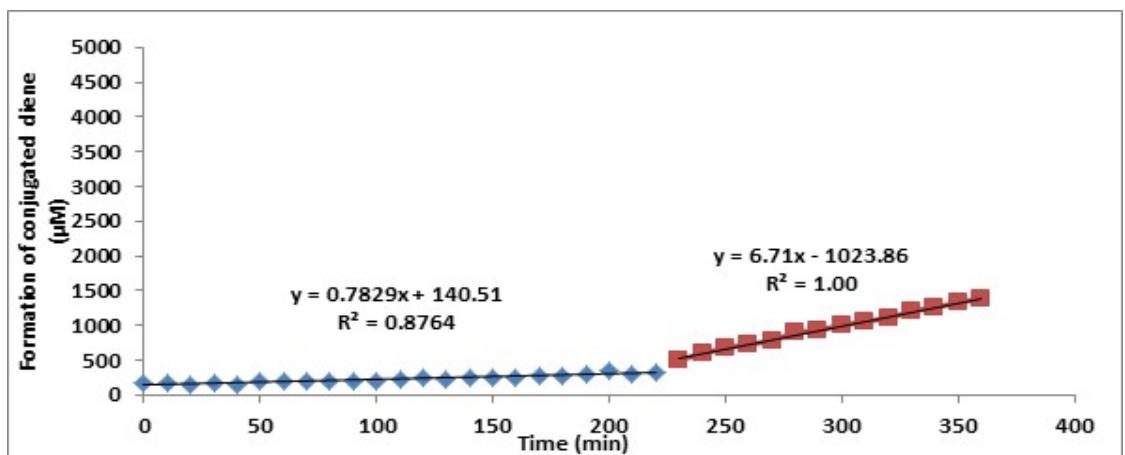


$$R_{\text{inh}} = 109.5 \mu\text{M}/\text{Min} \quad T_{\text{inh}} = 57 \text{ Min}$$

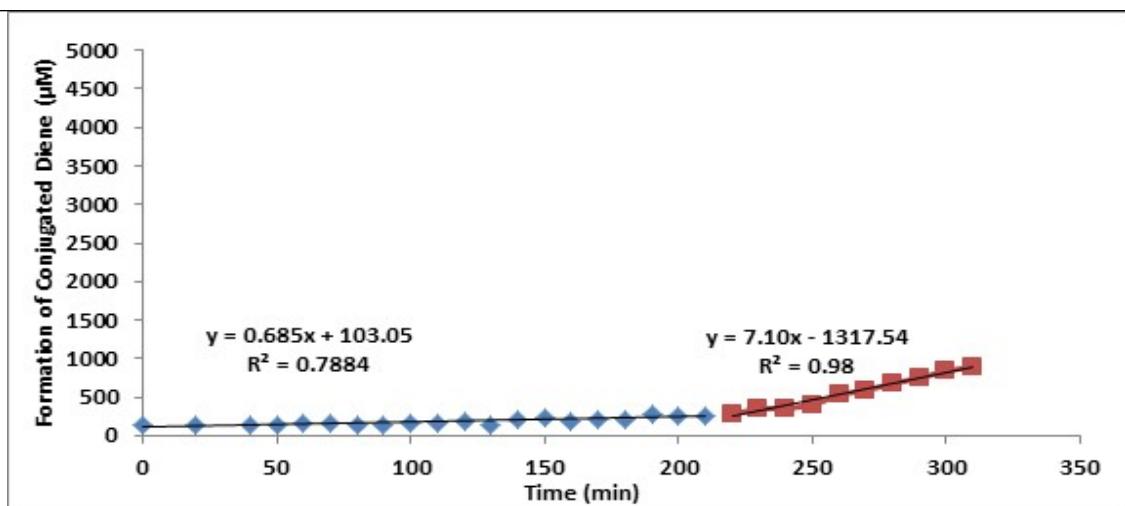


$$R_{\text{inh}} = 110.2 \mu\text{M}/\text{Min} \quad T_{\text{inh}} = 61 \text{ Min}$$

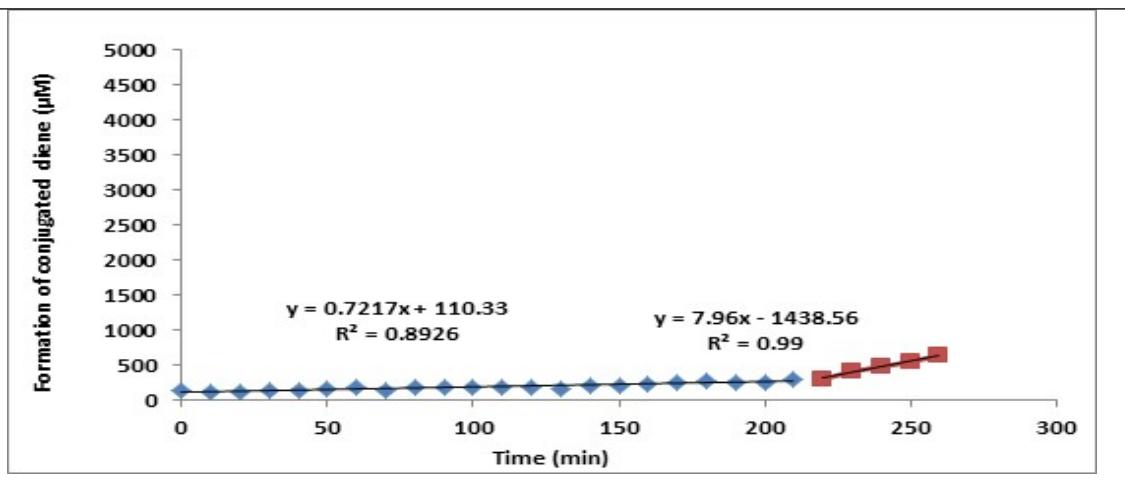
Figure S29. Experiment with antioxidant **9** using Ascorbic Acid



$$R_{\text{inh}} = 46.92 \mu\text{M}/\text{min} \quad T_{\text{inh}} = 210 \text{ Min}$$

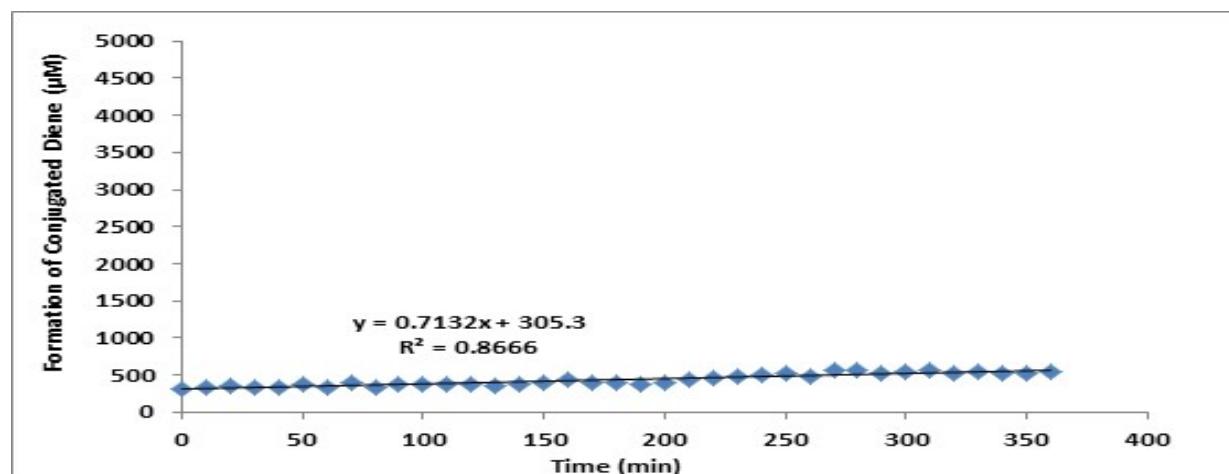


$$R_{\text{inh}} = 41.4 \mu\text{M}/\text{min} \quad T_{\text{inh}} = 208 \text{ Min}$$

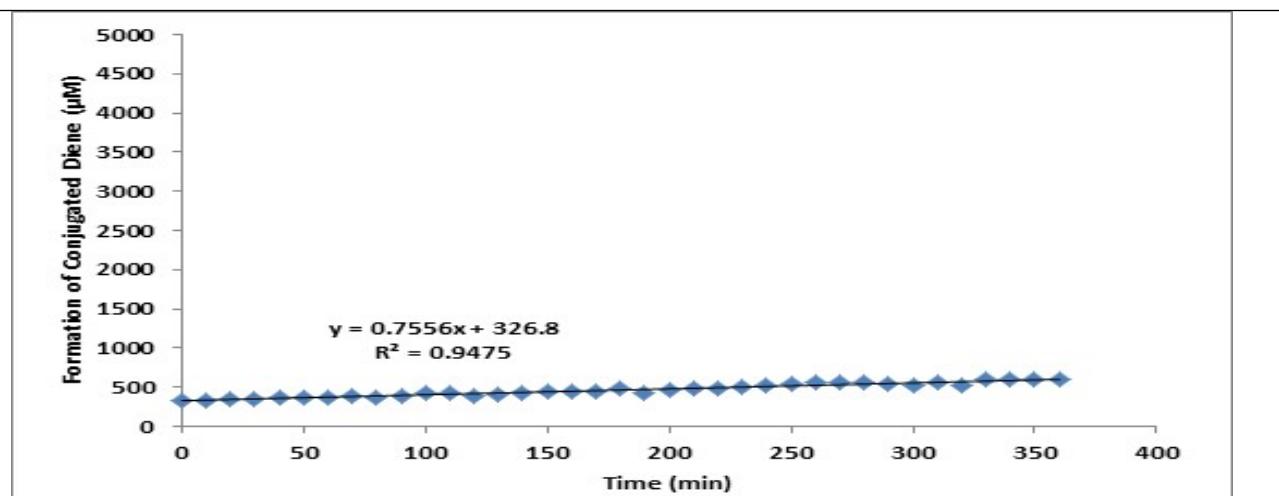


$$R_{\text{inh}} = 43.3 \mu\text{M}/\text{min} \quad T_{\text{inh}} = 212 \text{ Min}$$

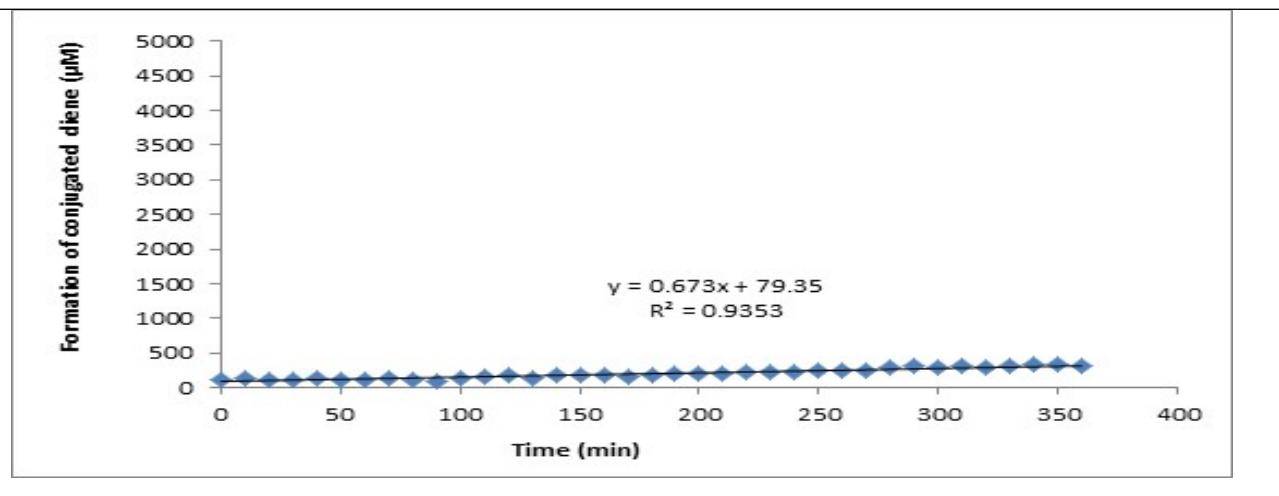
Figure S30. Experiment with antioxidant **9** using DTT



$$R_{\text{inh}} = 42.8 \text{ } \mu\text{M/min} \quad T_{\text{inh}} > 360 \text{ Min}$$

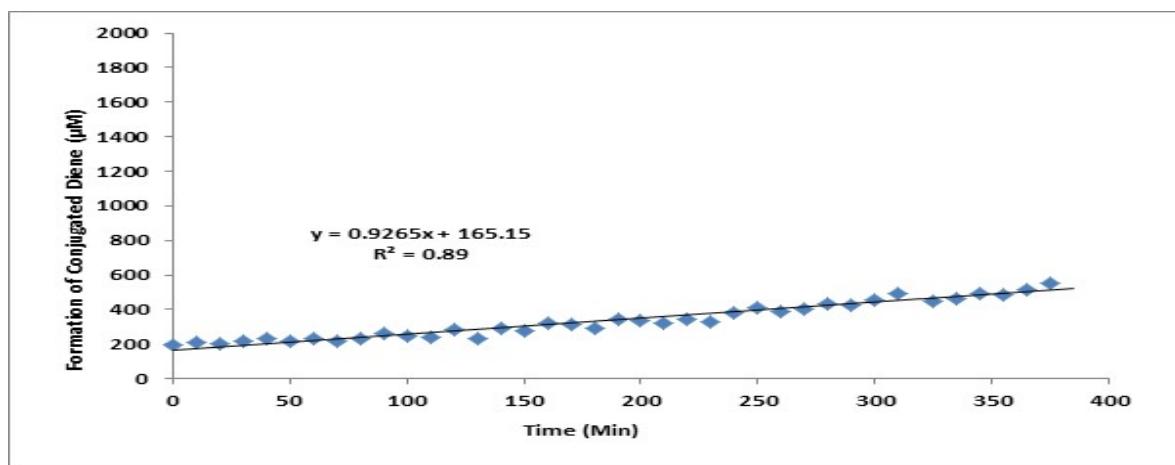


$$R_{\text{inh}} = 45.3 \text{ } \mu\text{M/min} \quad T_{\text{inh}} > 360 \text{ Min}$$

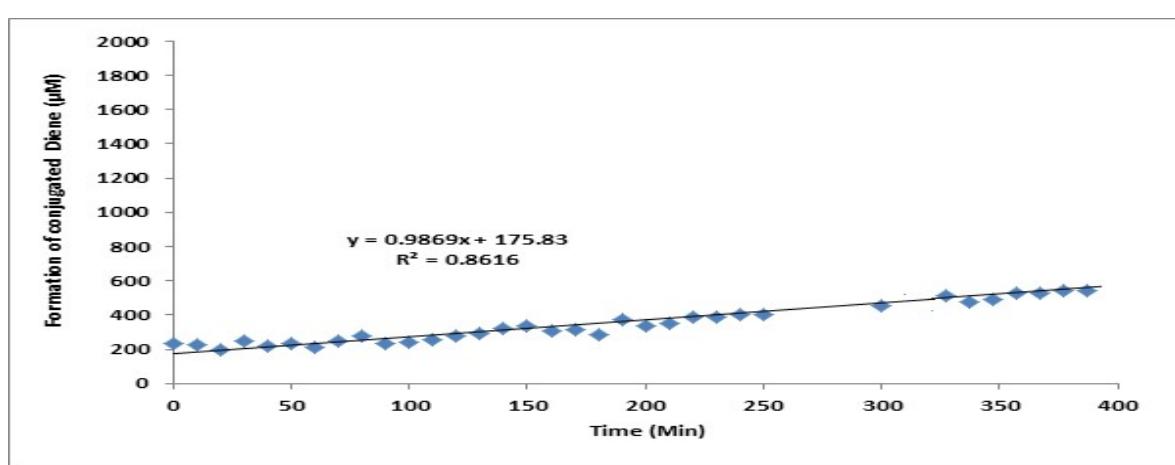


$$R_{\text{inh}} = 40.4 \text{ } \mu\text{M/min} \quad T_{\text{inh}} > 360 \text{ Min}$$

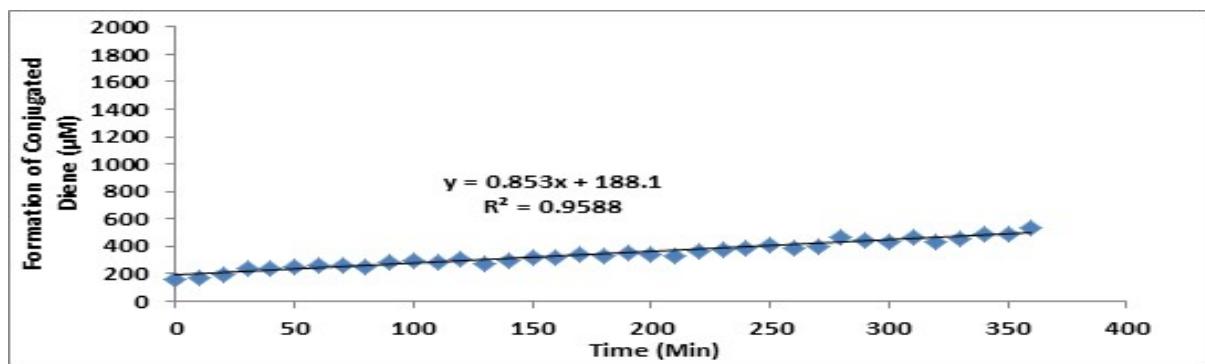
Figure S31. Experiment with antioxidant **9** using NAC



$$R_{\text{inh}} = 55.56 \mu\text{M}/\text{min} \quad T_{\text{inh}} > 360 \text{ Min}$$

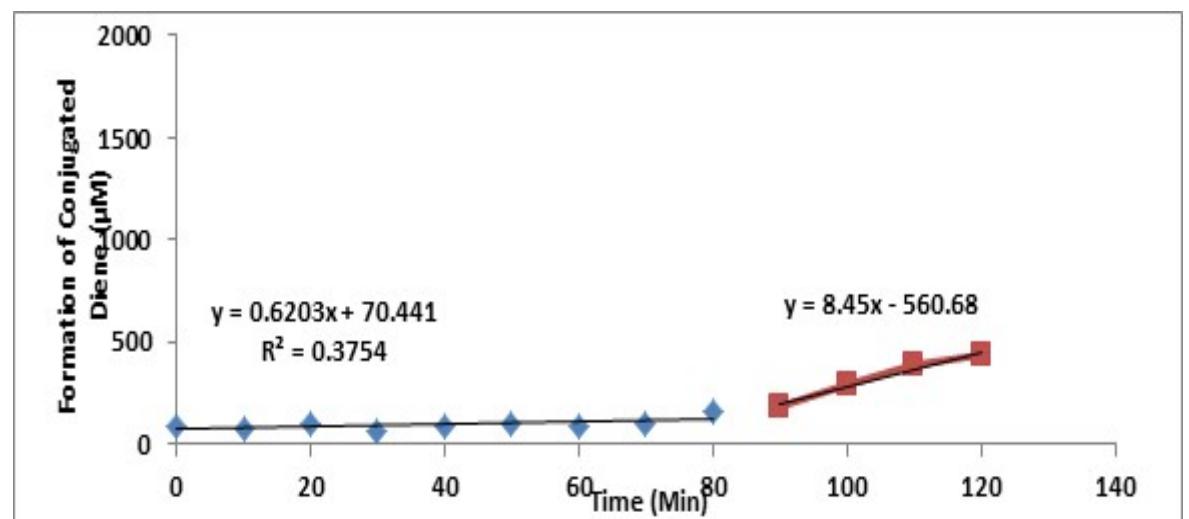


$$R_{\text{inh}} = 59.16 \mu\text{M}/\text{min} \quad T_{\text{inh}} > 360 \text{ Min}$$

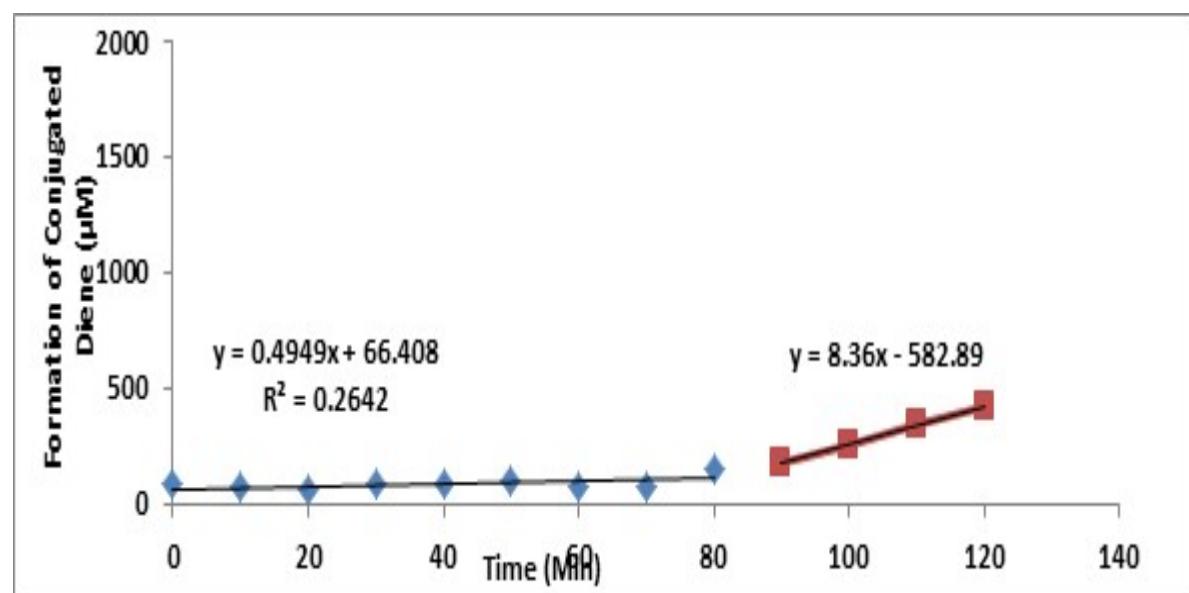


$$R_{\text{inh}} = 51.2 \mu\text{M}/\text{min} \quad T_{\text{inh}} > 360 \text{ Min}$$

Figure S32. Experiment with antioxidant **9** using $K_4Fe(CN)_6 \cdot 4H_2O$

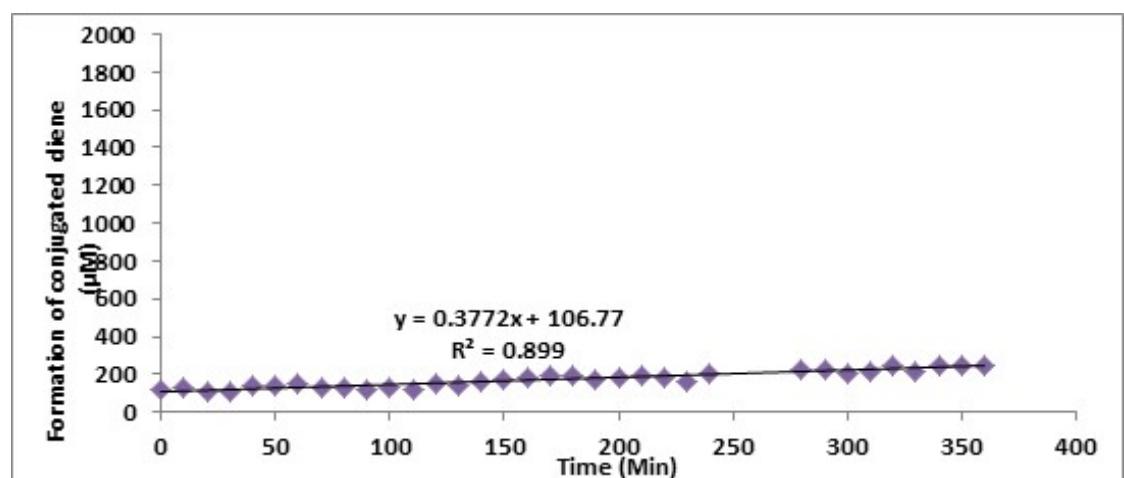


$$R_{inh} = 37.2 \mu M/min \quad T_{inh} = 80.602 \text{ Min}$$

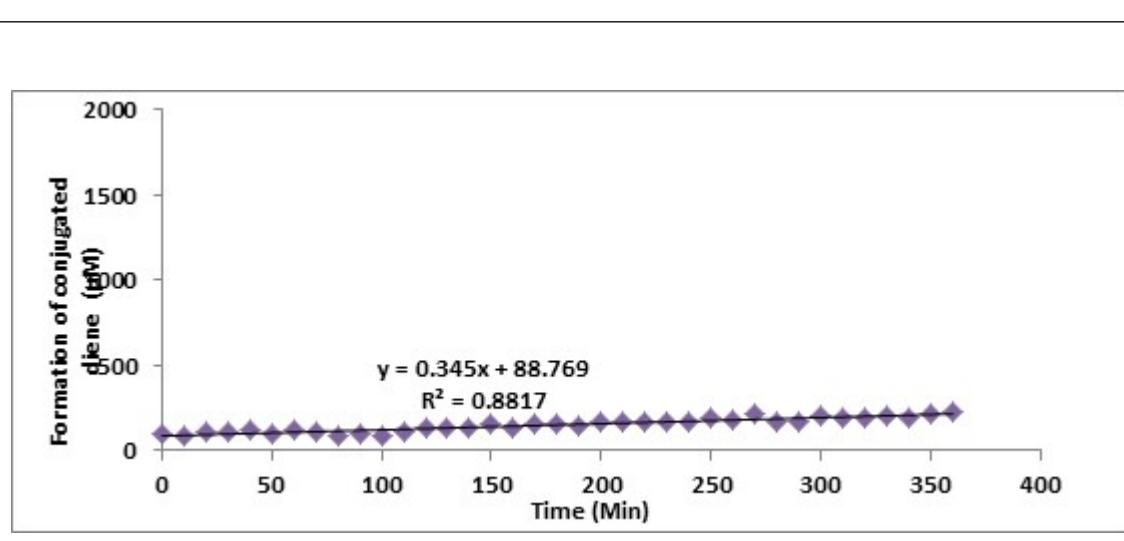


$$R_{inh} = 29.64 \mu M/min \quad T_{inh} = 82.54 \text{ Min}$$

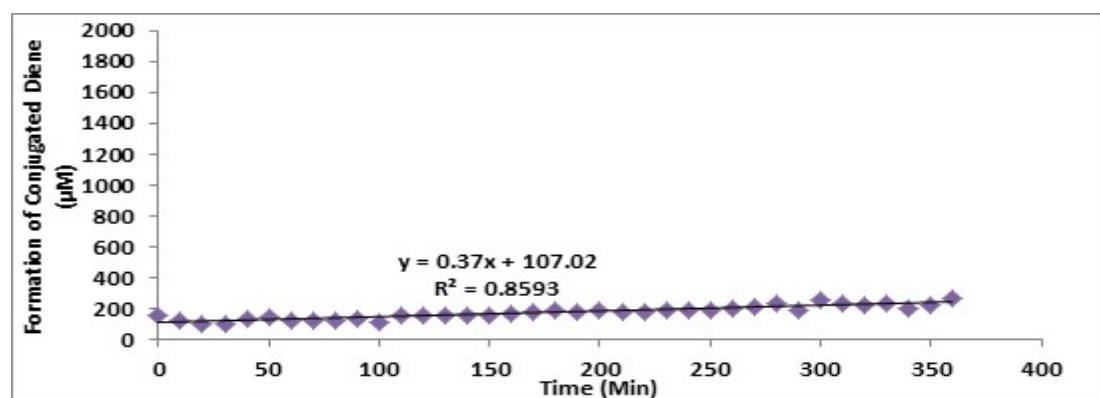
Figure S33. Experiment with antioxidant **9** using PTCP



$$R_{\text{inh}} = 22.62 \mu\text{M}/\text{min} \quad T_{\text{inh}} > 360 \text{ Min}$$

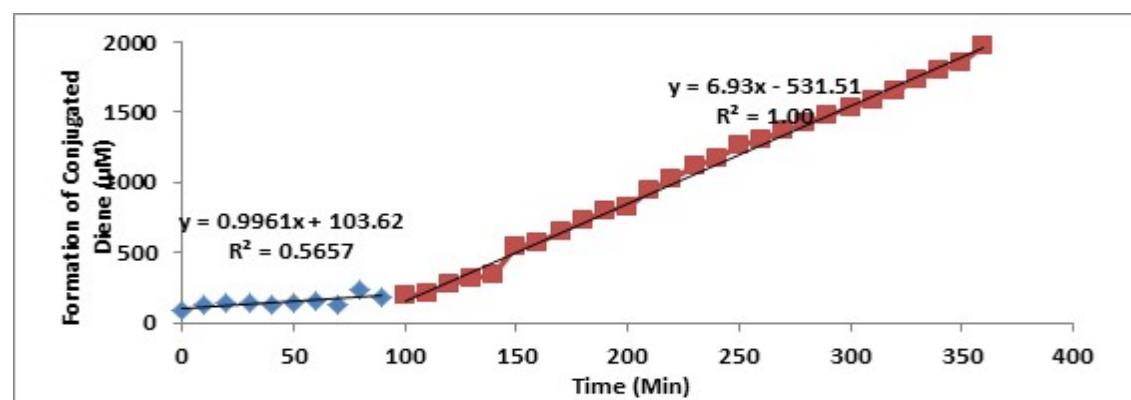


$$R_{\text{inh}} = 20.7 \mu\text{M}/\text{min} \quad T_{\text{inh}} > 360 \text{ Min}$$

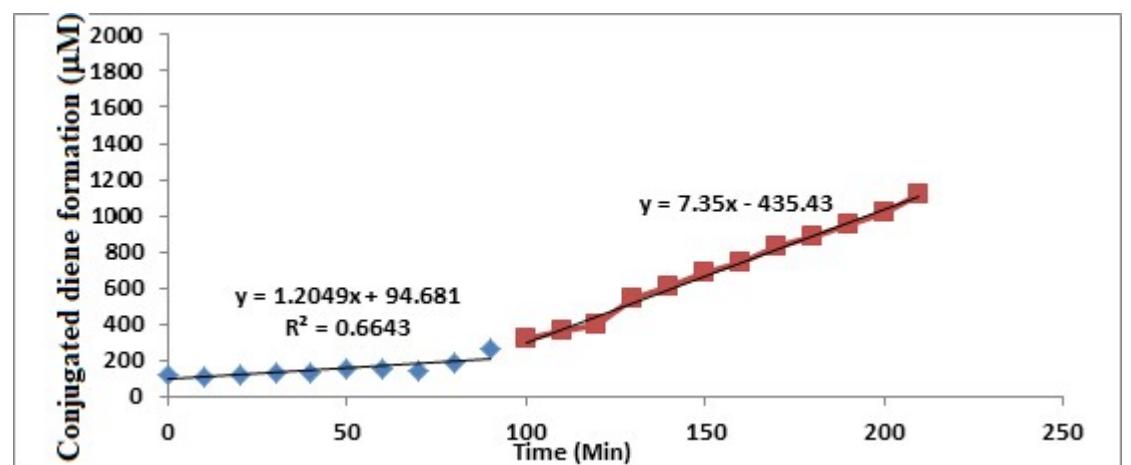


$$R_{\text{inh}} = 22.2 \mu\text{M}/\text{min} \quad T_{\text{inh}} > 360 \text{ Min}$$

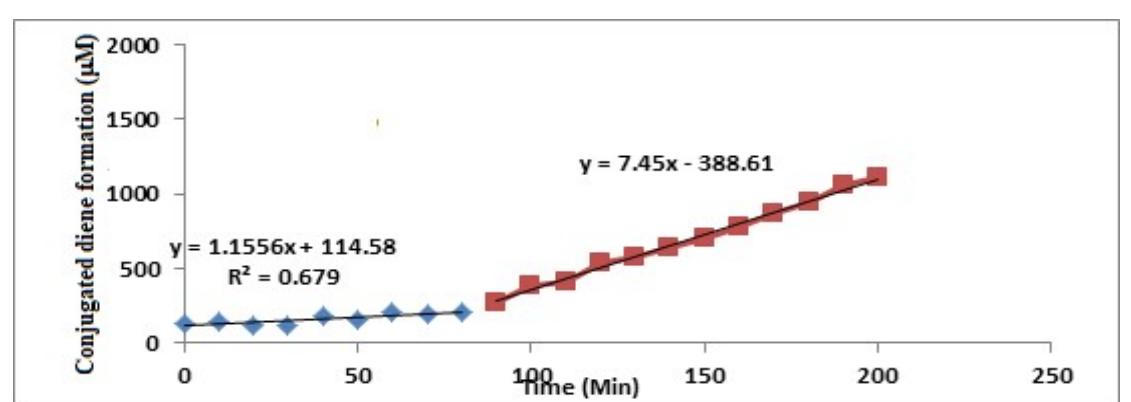
Figure S34. Experiment with antioxidant **9** using Sodium Ascorbate



$$R_{\text{inh}} = 59.76 \mu\text{M}/\text{min} \quad T_{\text{inh}} = 106.4 \text{ Min}$$

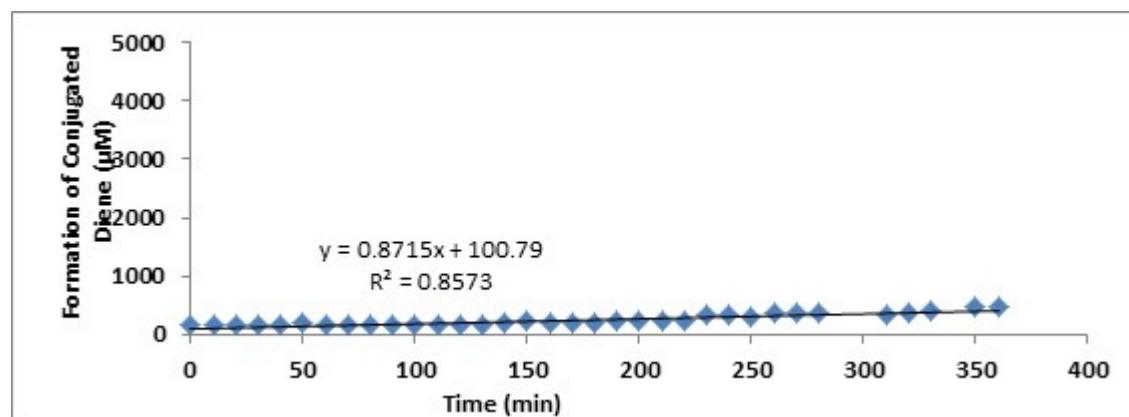


$$R_{\text{inh}} = 72.24 \mu\text{M}/\text{min} \quad T_{\text{inh}} = 86.2 \text{ Min}$$

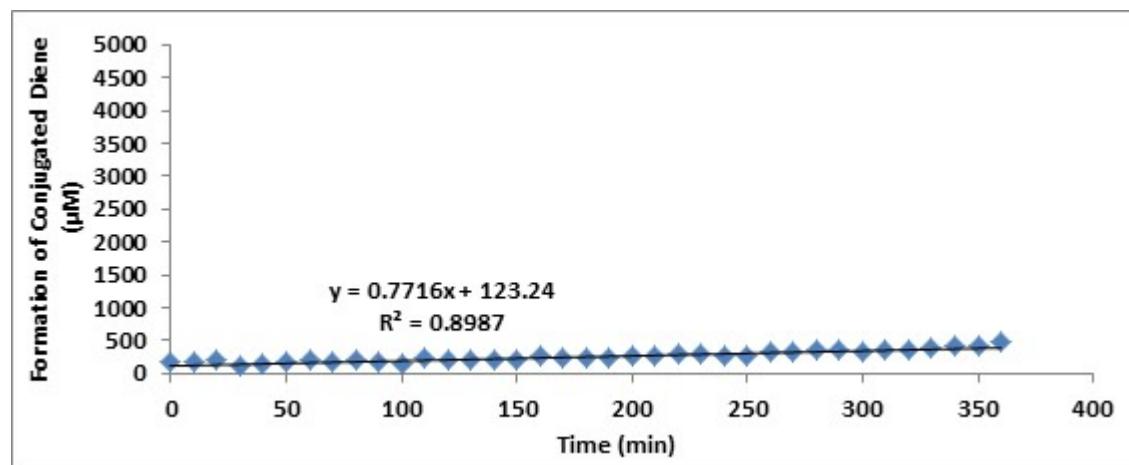


$$R_{\text{inh}} = 69.3 \mu\text{M}/\text{min} \quad T_{\text{inh}} = 85.34 \text{ Min}$$

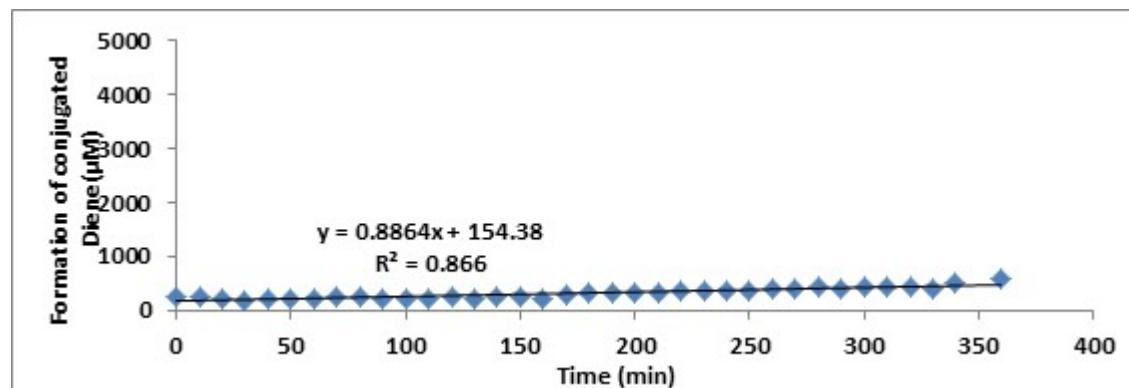
Figure S35. Experiment with antioxidant **9** using $\text{K}_4[\text{Fe}(\text{CN})_6] \cdot 3\text{H}_2\text{O}$ + PTC



$$R_{\text{inh}} = 52.26 \mu\text{M}/\text{min} \quad T_{\text{inh}} > 360 \text{ Min}$$

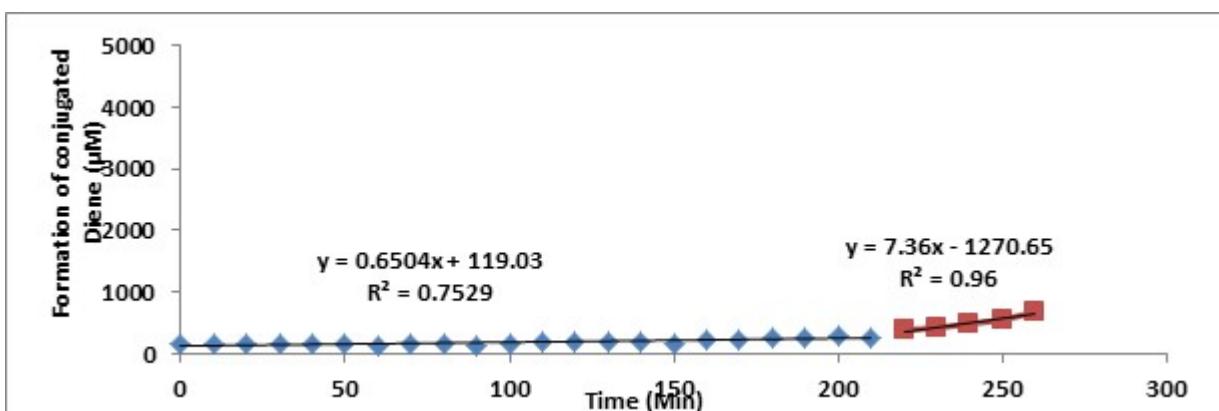


$$R_{\text{inh}} = 46.26 \mu\text{M}/\text{min} \quad T_{\text{inh}} > 360 \text{ Min}$$

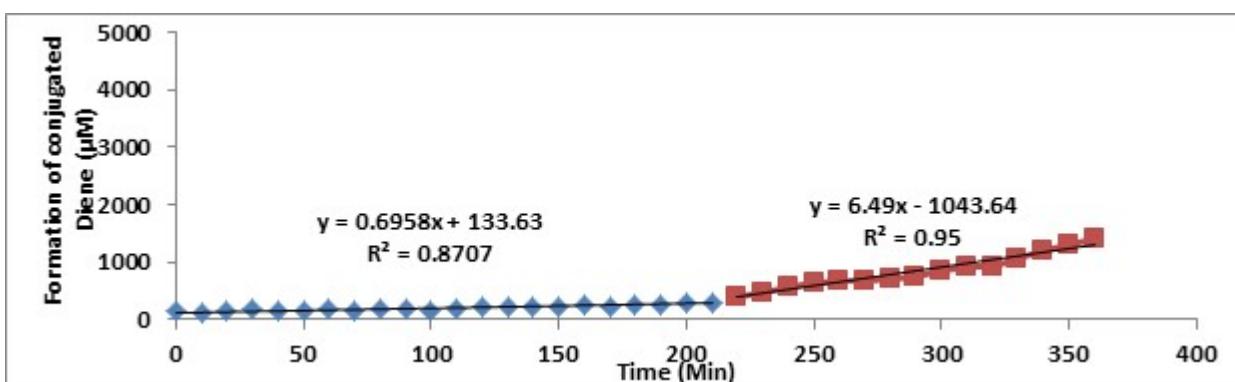


$$R_{\text{inh}} = 53.16 \mu\text{M}/\text{min} \quad T_{\text{inh}} > 360 \text{ Min}$$

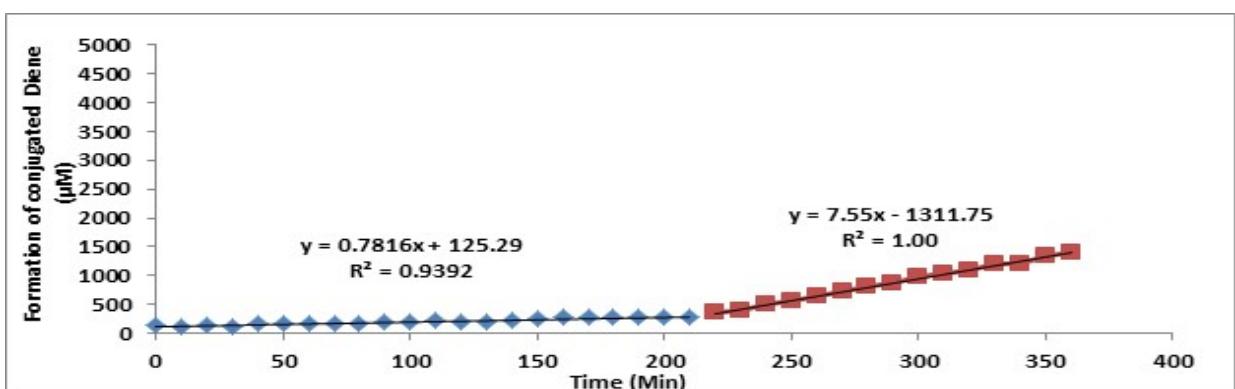
Figure S36. Experiment with antioxidant **9** using GSH



$$R_{inh} = 39 \mu\text{M}/\text{min} \quad T_{inh} = 207.10 \text{ Min}$$



$$R_{inh} = 41.7 \mu\text{M}/\text{min} \quad T_{inh} = 203.14 \text{ Min}$$



$$R_{inh} = 46.86 \mu\text{M}/\text{min} \quad T_{inh} = 212.28 \text{ Min}$$

Figure S37. Experiment with antioxidant **9** using NAC pH 1

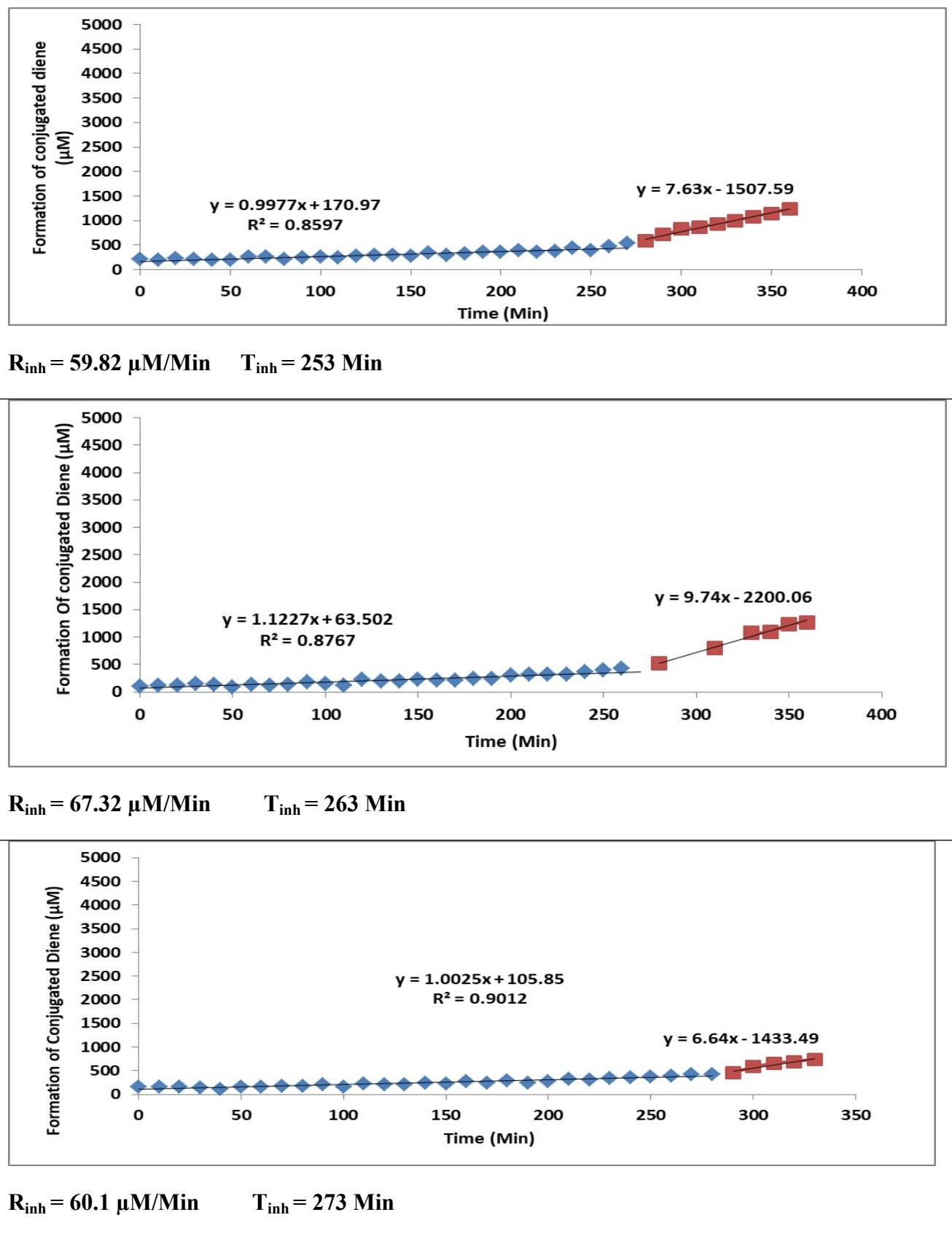


Figure S38. Experiment with antioxidant **9** using NAC pH 2

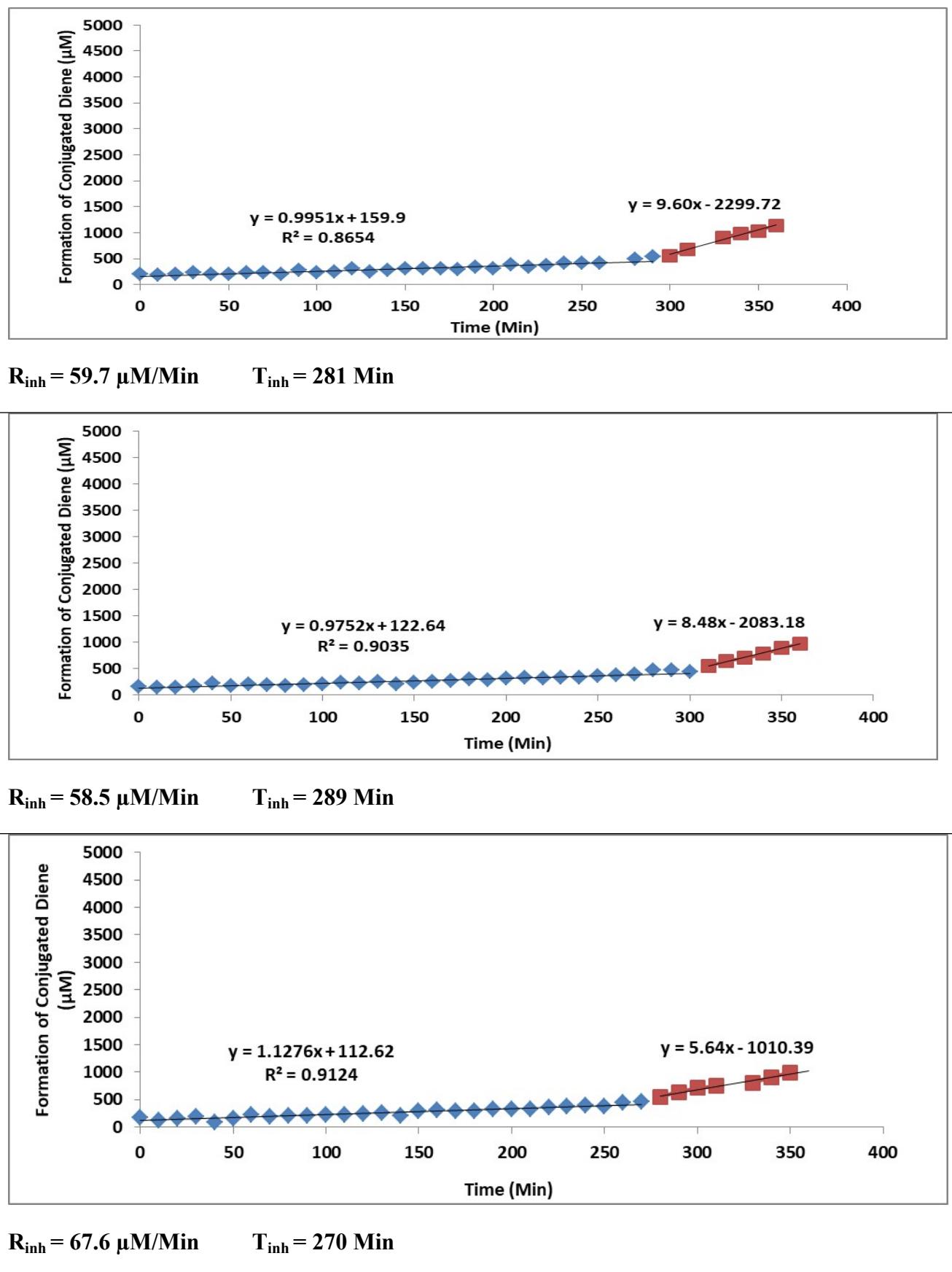
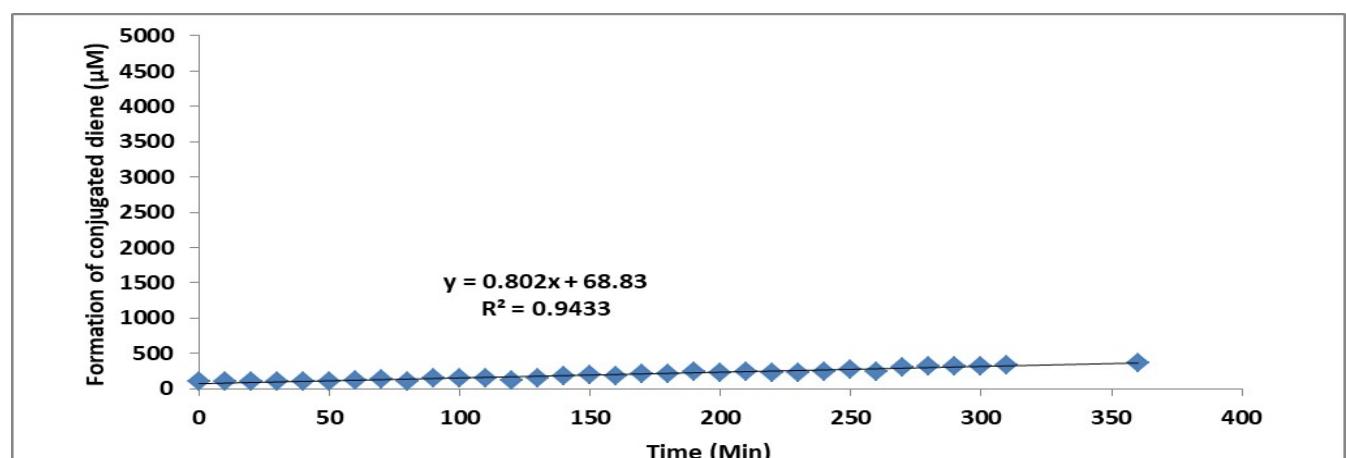
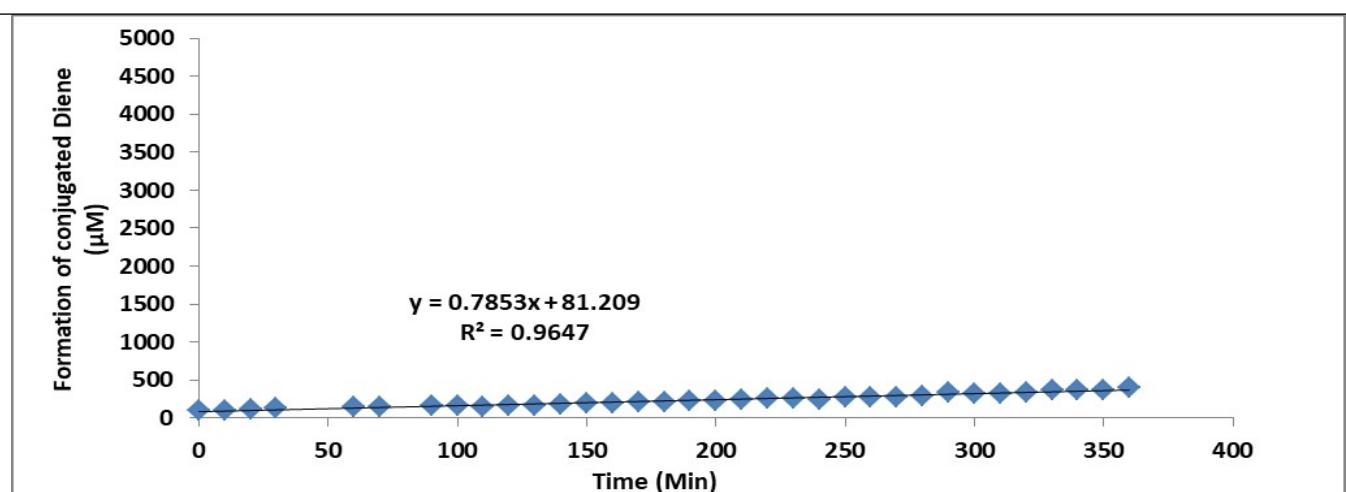


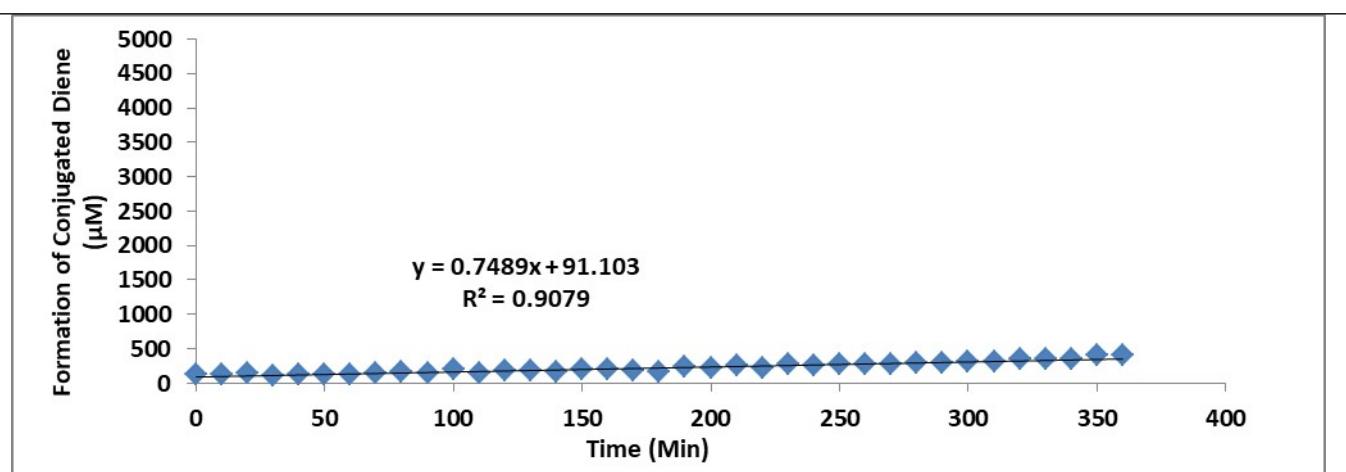
Figure S39. Experiment with antioxidant **9** using NAC pH 3



$$R_{\text{inh}} = 48.12 \mu\text{M}/\text{Min} \quad T_{\text{inh}} > 360 \text{ Min}$$



$$R_{\text{inh}} = 47.1 \mu\text{M}/\text{Min} \quad T_{\text{inh}} > 360 \text{ Min}$$



$$R_{\text{inh}} = 44.88 \mu\text{M}/\text{Min} \quad T_{\text{inh}} > 360 \text{ Min}$$

Figure S40. Experiment with antioxidant **9** using NAC pH 4

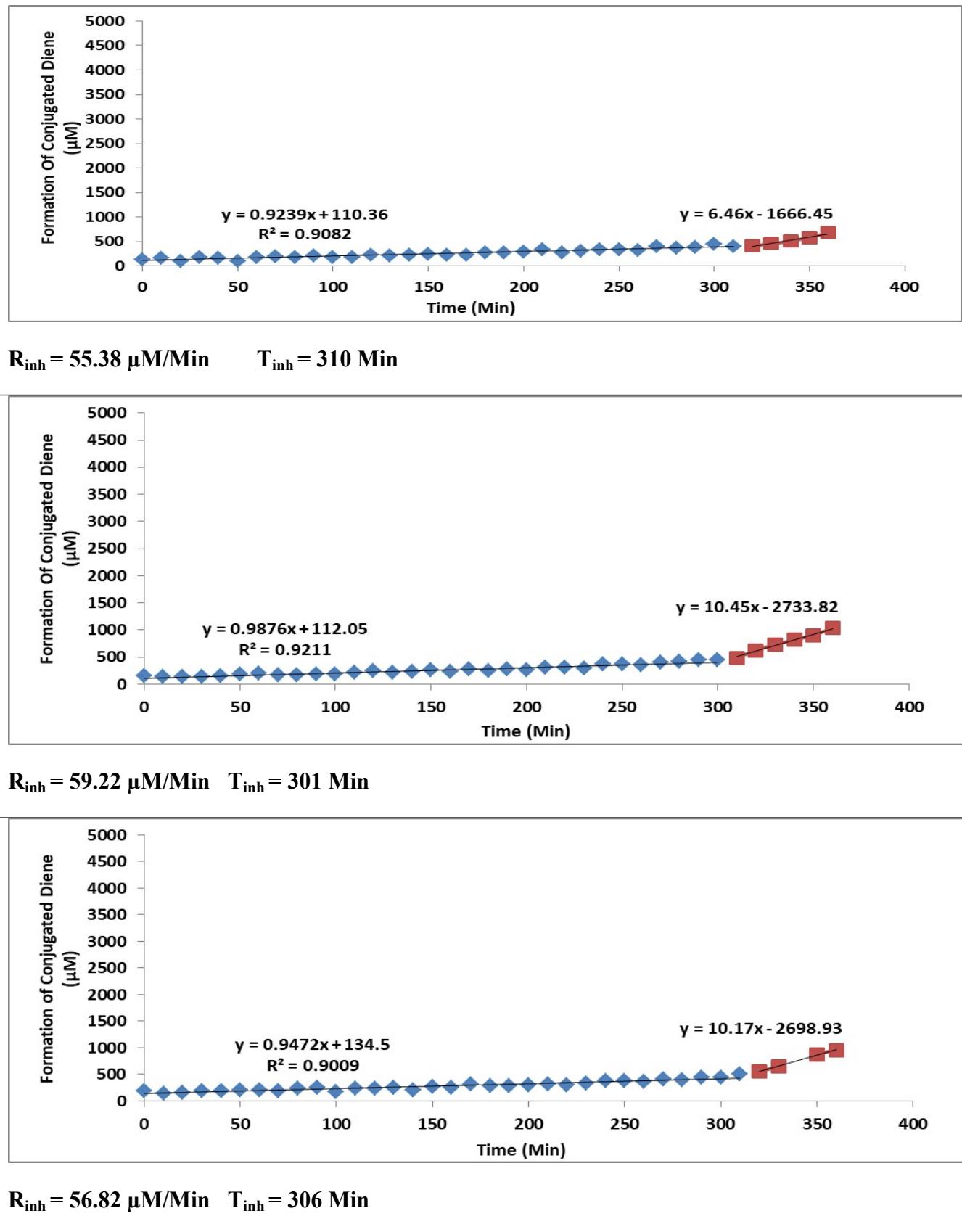


Figure S41. Experiment with antioxidant **9** using NAC pH 5

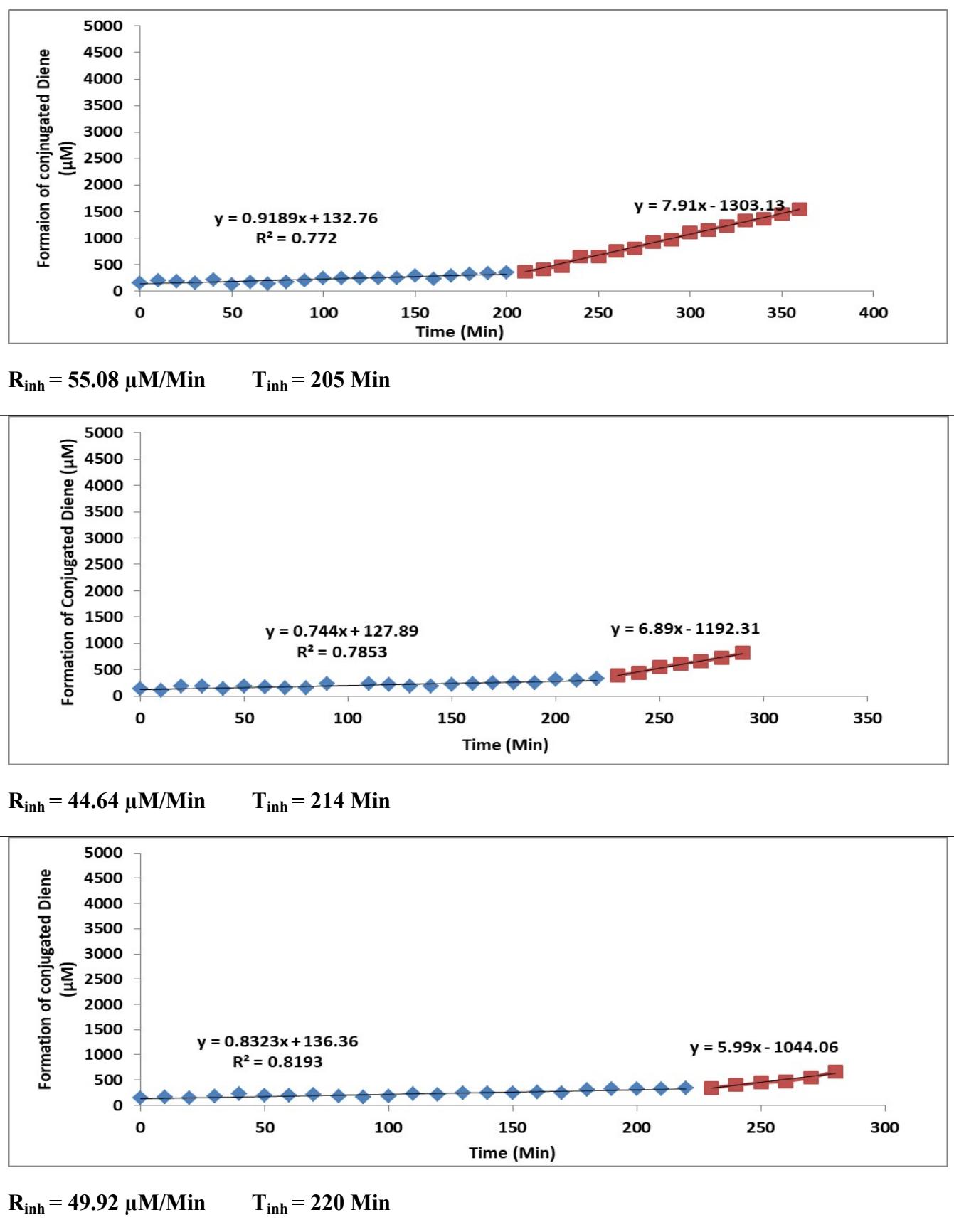


Figure S42. Experiment with antioxidant **9** using NAC pH 6

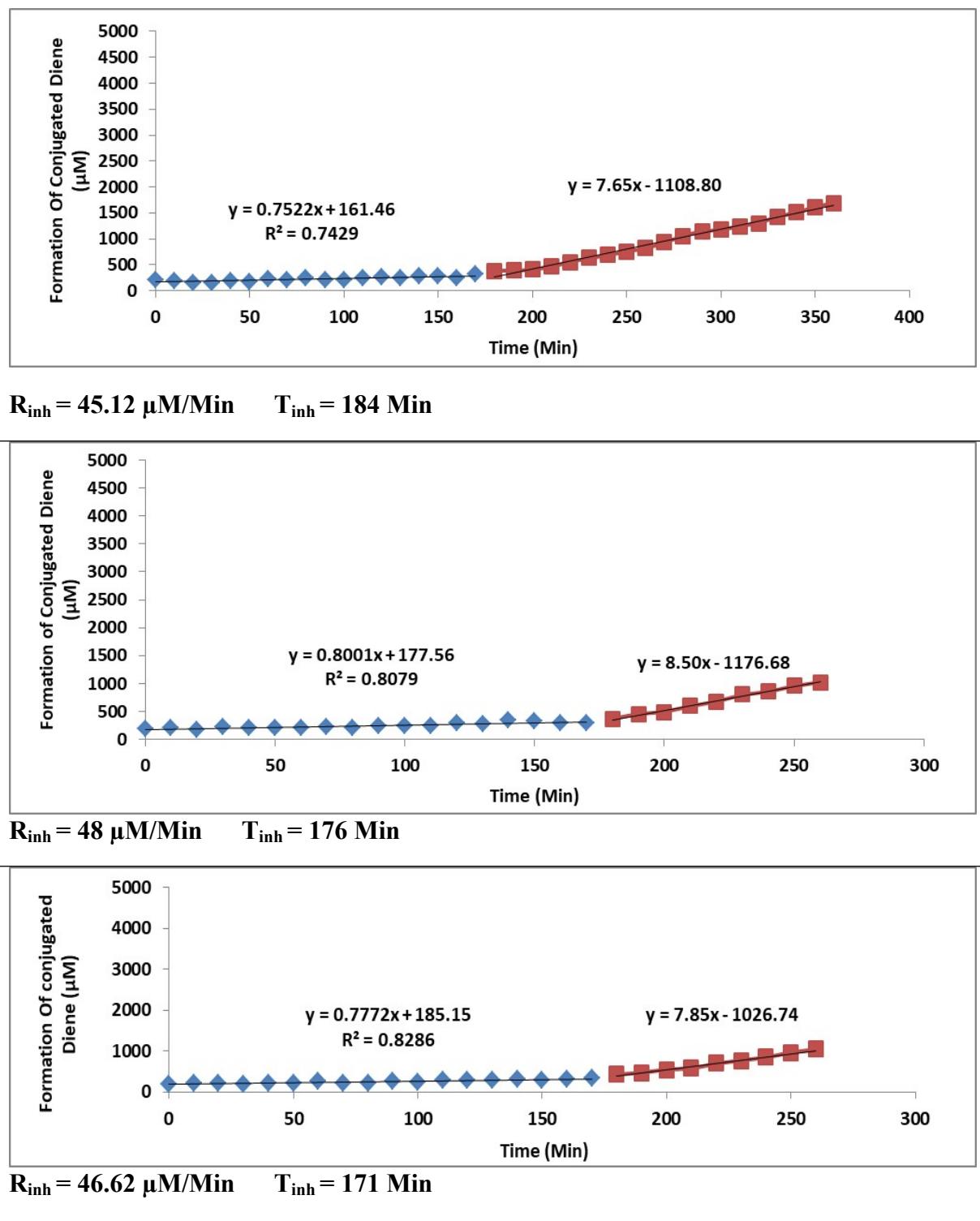
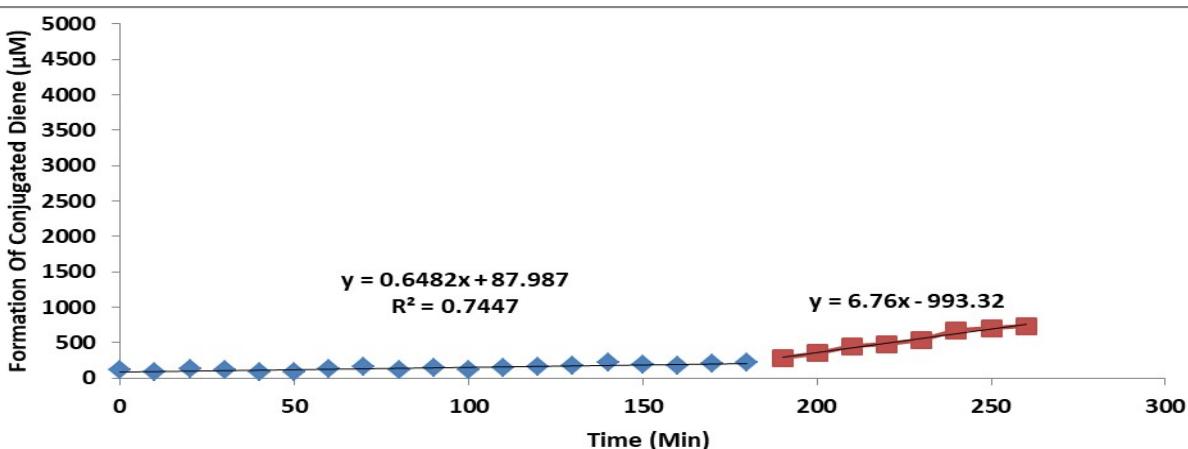
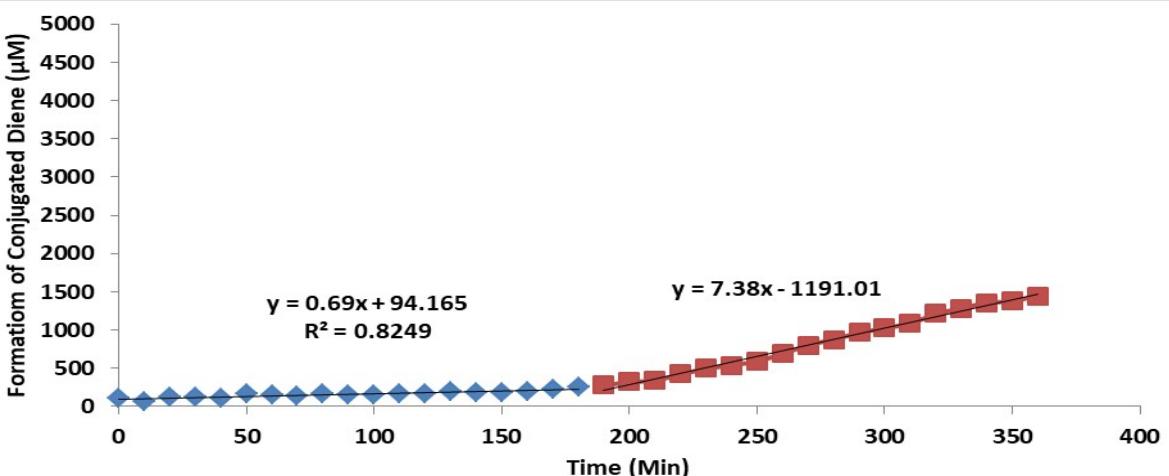


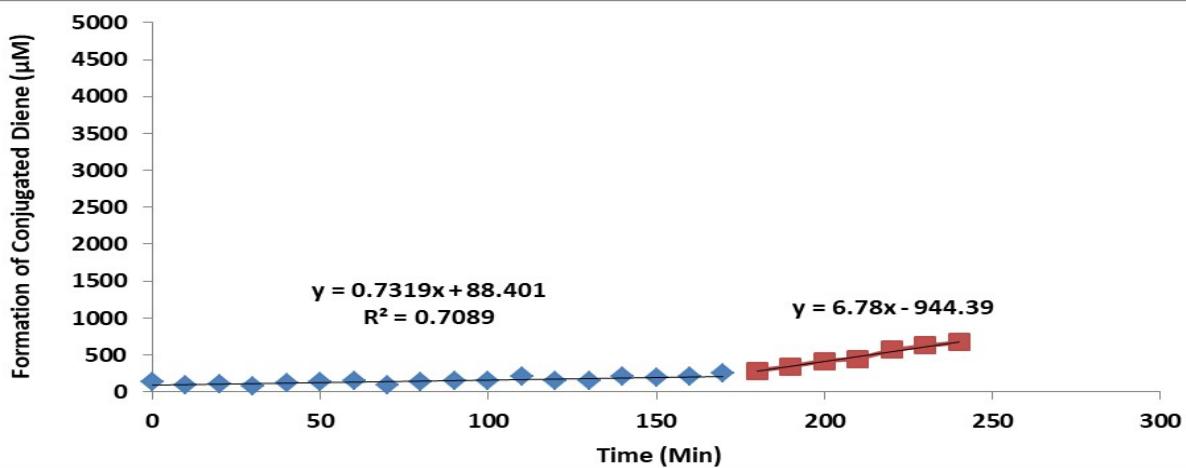
Figure S43. Experiment with antioxidant **9** using NAC pH 7



$$R_{\text{inh}} = 38.9 \mu\text{M}/\text{Min} \quad T_{\text{inh}} = 177 \text{ Min}$$



$$R_{\text{inh}} = 41.4 \mu\text{M}/\text{Min} \quad T_{\text{inh}} = 193 \text{ Min}$$



$$R_{\text{inh}} = 43.9 \mu\text{M}/\text{Min} \quad T_{\text{inh}} = 171 \text{ Min}$$

Scheme S1. Proposed mechanism for the quenching of LOO^\bullet by antioxidant **9** in the presence of aqueous co-antioxidants.

