

## Supporting Information

# Rational Design of Cobaltocenium-containing Polythioether Type Metallo-polyelectrolyte as HCl Corrosion Inhibitor for Mild Steel

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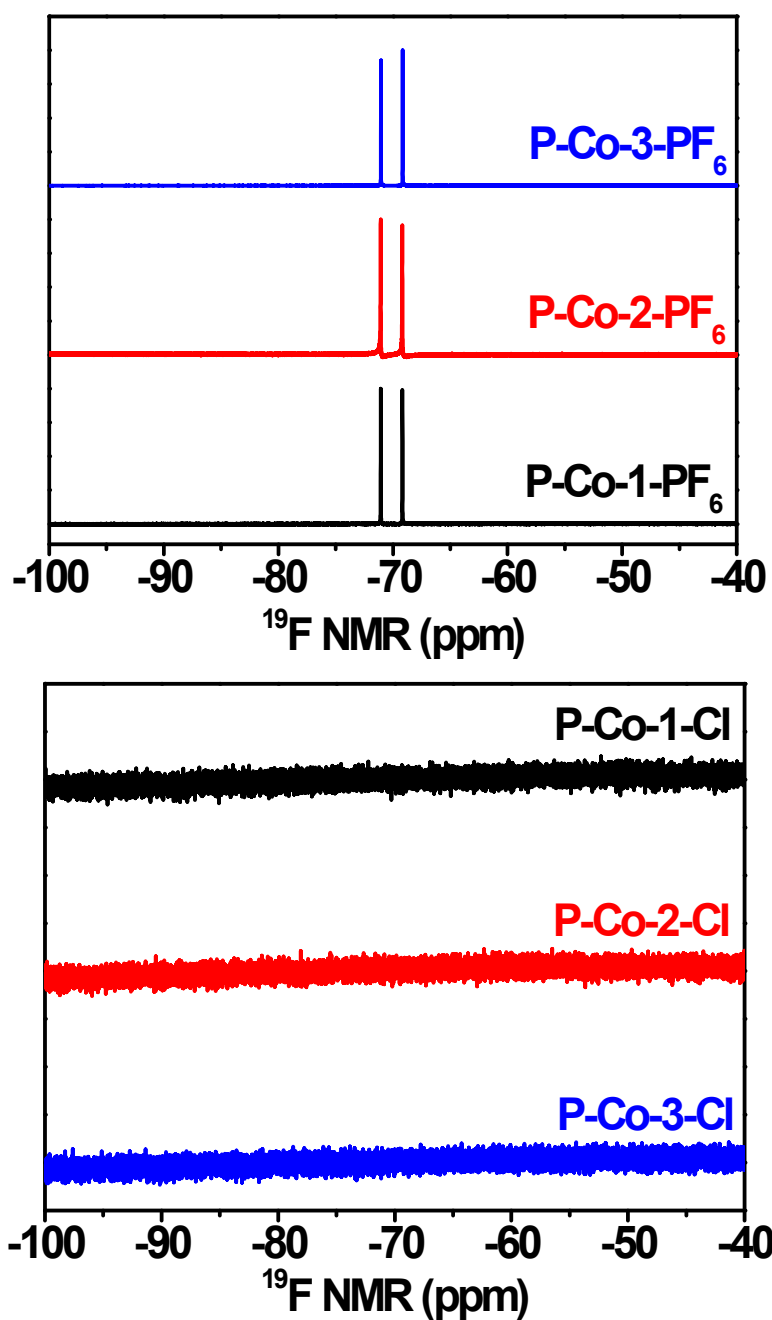
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**Figure S1.**  $^{19}\text{F}$  NMR spectra of resulted inhibitors before (top) and after (bottom) counterion exchange.



**Figure S2.** Photos of the resulted inhibitors: P-Co-1-PF<sub>6</sub>, P-Co-2-PF<sub>6</sub>, P-Co-3-PF<sub>6</sub> (from left to right).



Figure S3. Photos of the resulted inhibitor in  $D_2O$ : P-Co-1- $PF_6$ , P-Co-2- $PF_6$ , P-Co-3- $PF_6$  (from left to right).

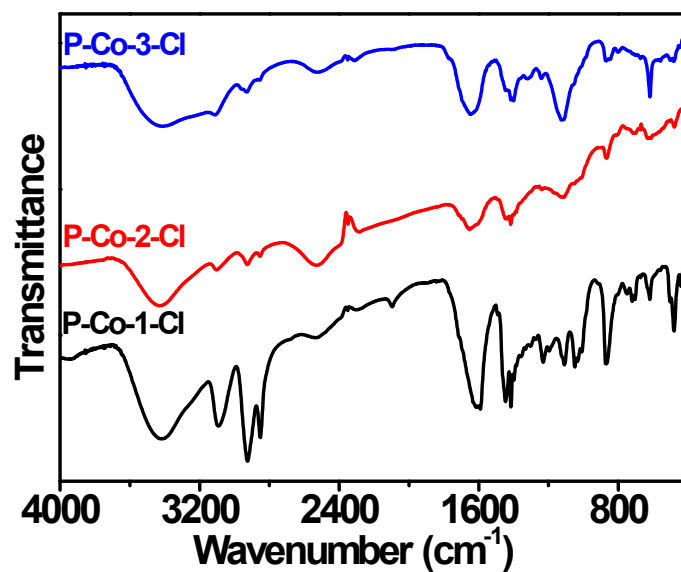
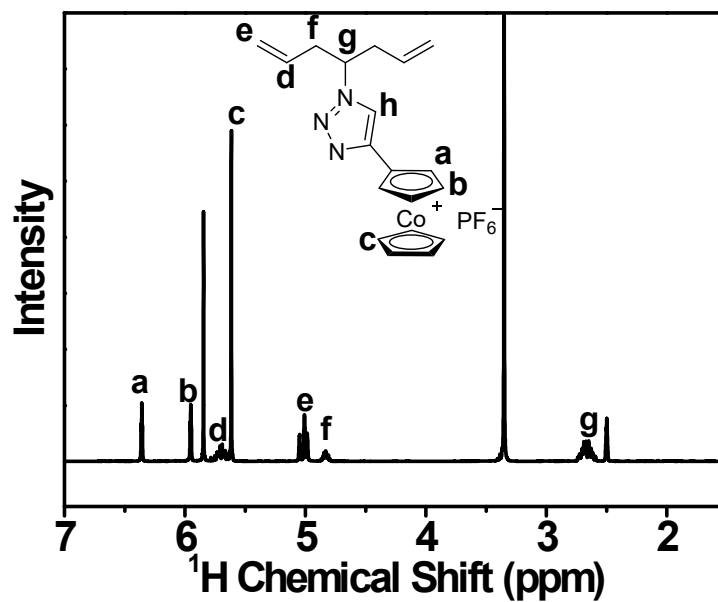
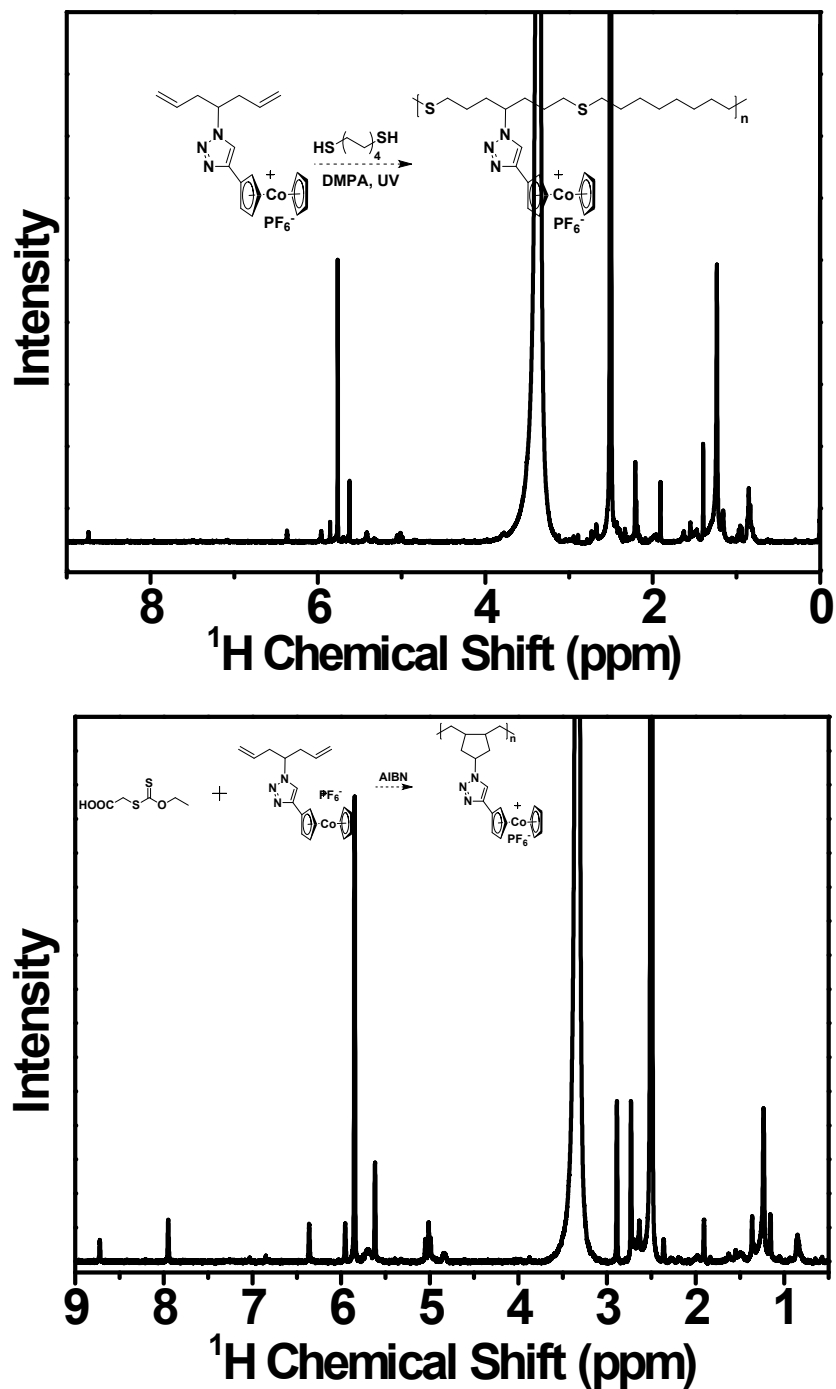


Figure S4. FT-IR spectra of resulted inhibitors.





**Figure S5.**  $^1\text{H}$  NMR spectra in  $d_6$ -DMSO for cobaltocenium-based diene monomer (top), crude after photo-induced thiol-ene polymerization (middle), crude after RAFT/MADIX polymerization (bottom).

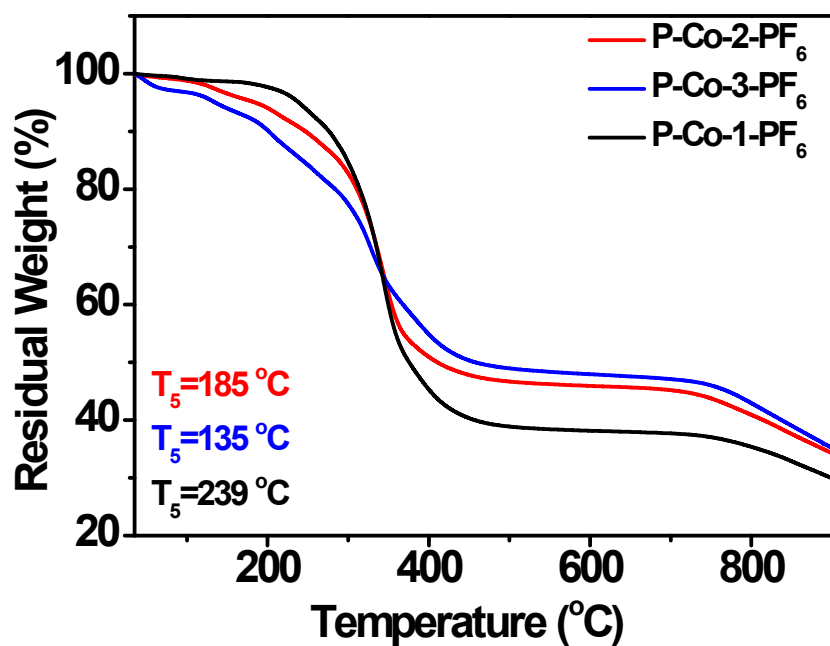


Figure S6. TGA curves of three cobaltocenium-containing polythioethers.

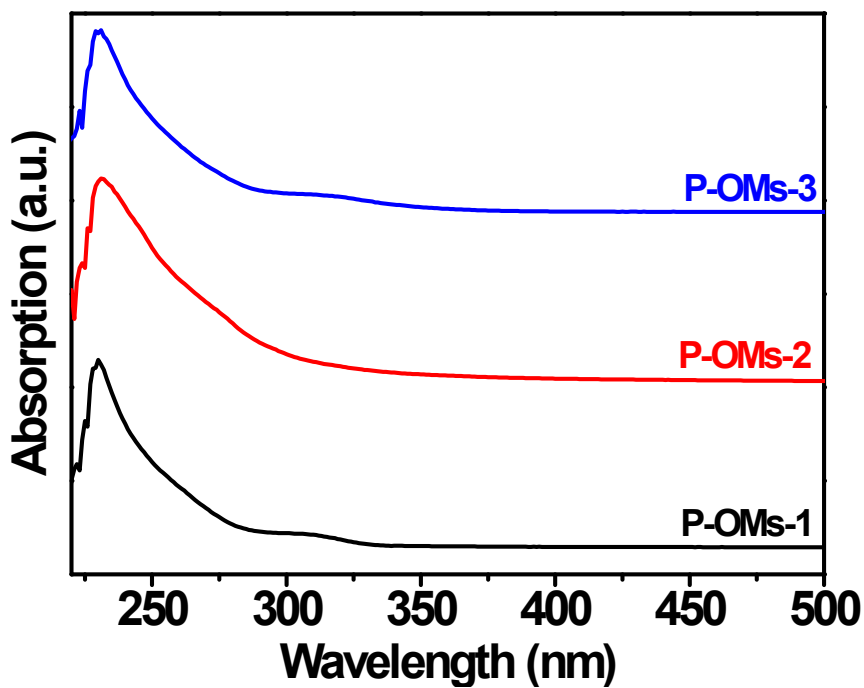
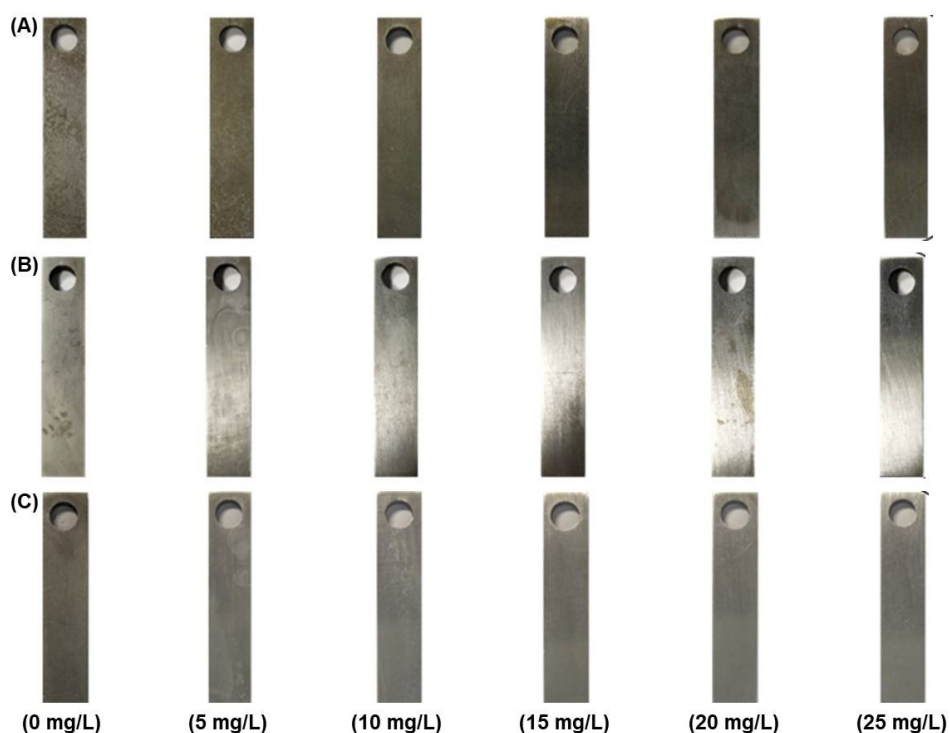
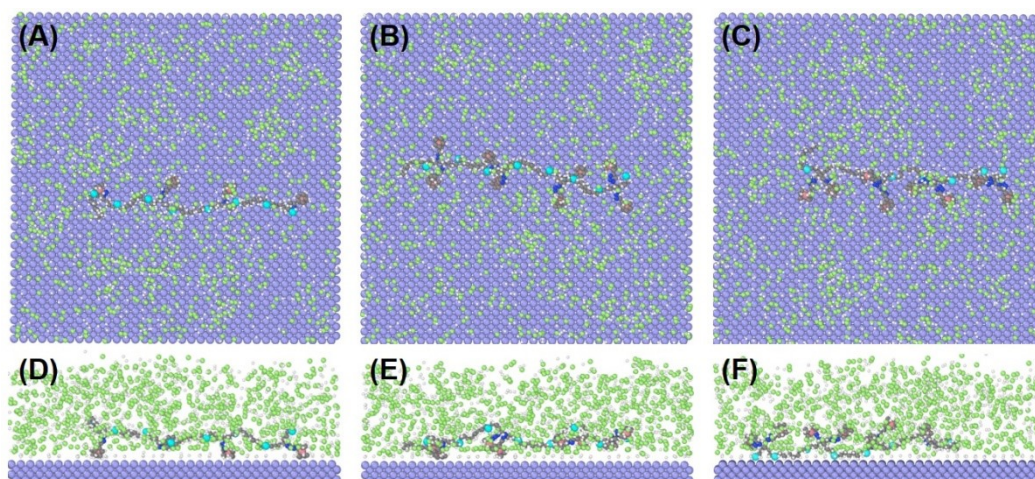


Figure S7. UV-visible spectra of precursor polythioethers.



**Figure S8.** Optical images of steel immersed in corrosion systems with corrosion inhibitors and without inhibitors at different concentration of inhibitors for 2 h. (A: P-Co-1-Cl; B: P-Co-2-Cl; C: P-Co-3-Cl)



**Figure S9.** The optimized configuration for the inhibitors P-Co-1-Cl, P-Co-2-Cl, P-Co-3-Cl with four repeat units adsorbed on Fe (110) in the presence of H<sub>2</sub>O obtained from molecular dynamic simulation: (A)-(C) top view; (D)-(F) side view. (● Cl, ○ H<sup>+</sup>, ● S, ● N, ● Co).

**Table S1.** The molecular weight of methylsulfonate-containing polythioether.

Samples	$M_n$ (g/mol)	$\bar{D}$
<b>P-OMs-1</b>	6900	1.63
<b>P-OMs-2</b>	8900	1.60
<b>P-OMs-3</b>	6600	1.46

**Table S2.** Electrochemical parameters from the potentiodynamic polarization measurements for mild steel in 5 wt% HCl with different concentration of **P-Co-2-Cl** at 25 °C.

$C$ (mg/L)	$E_{corr}$ (mV)	$I_{corr}$ ( $\mu\text{A}\cdot\text{cm}^{-2}$ )	$-b_c$ ( $\text{mV}\cdot\text{dec}^{-1}$ )	$b_a$ ( $\text{mV}\cdot\text{dec}^{-1}$ )	$\eta$ (%)
0	-460.5	82.73	82.00	94.80	/
5	-486.2	20.36	44.15	78.16	75.39
10	-494.9	17.75	95.46	59.35	78.55
15	-505.4	16.17	72.17	50.07	80.46
20	-525.0	14.36	103.55	34.58	82.64
25	-519.9	11.86	69.01	51.71	85.67

**Table S3.** Electrochemical parameters from the potentiodynamic polarization measurements for mild steel in 5 wt% HCl with different concentration of **P-Co-3-Cl** at 25 °C.

$C$ (mg/L)	$E_{corr}$ (mV)	$I_{corr}$ ( $\mu\text{A}\cdot\text{cm}^{-2}$ )	$-b_c$ ( $\text{mV}\cdot\text{dec}^{-1}$ )	$b_a$ ( $\text{mV}\cdot\text{dec}^{-1}$ )	$\eta$ (%)
0	-460.5	82.73	82.00	94.80	/
5	-503.8	19.43	79.90	46.92	76.52
10	-503.1	16.84	52.64	67.43	79.65
15	-519.3	14.19	64.40	54.64	82.85
20	-536.8	10.88	62.90	55.77	86.85
25	-543.8	9.65	143.32	27.22	88.34

**Table S4.** Electrochemical parameters from the EIS measurements for mild steel in 5 wt% HCl with different concentration of **P-Co-1-Cl** at 25 °C.

$C$ (mg/L)	$R_s$ ( $\Omega\cdot\text{cm}^2$ )	$Y_0$	$n$	$C_{dl}$ ( $\mu\text{F}\cdot\text{cm}^{-2}$ )	$R_{ct}$ ( $\Omega\cdot\text{cm}^2$ )	$f_{max}$	$\eta$ (%)
0	0.6141	$2.037\times 10^{-5}$	0.8927	32.65	121.3	12.4	/
5	0.6363	$5.909\times 10^{-5}$	1	59.09	499.8	7.97	75.7
10	0.3050	$4.460\times 10^{-5}$	1	44.60	547.9	8.94	77.9
15	0.3259	$8.457\times 10^{-5}$	1	84.57	665.9	6.35	81.8
20	0.2476	$4.037\times 10^{-5}$	1	40.37	800.9	7.97	84.9
25	0.2513	$1.778\times 10^{-5}$	1	17.78	557.1	5.66	78.2

**Table S5.** Electrochemical parameters from the EIS measurements for mild steel in 5 wt% HCl with different concentration of **P-Co-2-Cl** at 25 °C.

C (mg/L)	$R_s$ ( $\Omega \cdot \text{cm}^2$ )	$Y_0$ ( $\text{W}^{-1}\text{cm}^{-2}$ )	n	$C_{dl}$ ( $\mu\text{F} \cdot \text{cm}^{-2}$ )	$R_{ct}$ ( $\Omega \cdot \text{cm}^2$ )	$f_{max}$	$\eta(\%)$
0	0.6141	$2.037 \times 10^{-5}$	0.8927	32.65	121.3	12.4	/
5	0.1167	$4.009 \times 10^{-5}$	1	40.09	461.7	5.66	73.7
10	0.1557	$7.431 \times 10^{-4}$	1	74.31	600.3	4.97	79.8
15	0.1141	$6.449 \times 10^{-5}$	1	64.49	655.0	3.94	81.5
20	0.1558	$6.122 \times 10^{-4}$	1	61.22	806.0	3.94	85.0
25	0.1507	$4.999 \times 10^{-4}$	1	49.99	918.9	4.97	86.8

**Table S6.** Electrochemical parameters from the EIS measurements for mild steel in 5 wt% HCl with different concentration of **P-Co-3-Cl** at 25 °C.

C (mg/L)	$R_s$ ( $\Omega \cdot \text{cm}^2$ )	$Y_0$	n	$C_{dl}$ ( $\mu\text{F} \cdot \text{cm}^{-2}$ )	$R_{ct}$ ( $\Omega \cdot \text{cm}^2$ )	$f_{max}$	$\eta(\%)$
0	0.6141	$2.037 \times 10^{-5}$	0.8927	32.65	121.3	12.4	/
5	0.1991	$1.443 \times 10^{-5}$	1	14.43	820.8	7.97	85.2
10	0.1935	$1.847 \times 10^{-5}$	1	18.47	1028	6.35	88.2
15	0.1657	$1.742 \times 10^{-5}$	1	17.42	1429	5.66	91.5
20	0.1981	$1.310 \times 10^{-5}$	1	13.10	961.2	6.35	87.4
25	0.1793	$1.668 \times 10^{-5}$	1	16.68	918.4	6.35	86.8

**Table S7.** The elemental composition for mild steel.

Element	C	Si	Mn	P	S	Cr	Zn	Mo	Al	Cu	Ti	Fe
Content (%)	0.08	0.18	0.45	0.03	0.03	0.12	0.00 03	0.00 1	0.01	0.25	0.25	98.6 0