

Supporting Information

Phthalocyanine-Induced Iron Active Species in Metal-Organic Framework-Derived Porous Carbon for Efficient Alkaline Zinc-Air Batteries

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General Instruments.

Scanning electron microscopy (SEM) data were obtained within a JSM-6700F field emission scanning electron microscope at 10 kV. High-resolution transmission electron microscope (HR-TEM) and energy dispersive spectroscopy (EDS) were collected on a JEOL JEM2100F microscope at a high voltage of 200 kV. Single gas adsorption/desorption were performed in the Accelerated Surface Area and Porosimetry System 2020 (ASAP 2020), and the N₂ sorption was at 77 K. The powder X-ray diffraction patterns (PXRD) were acquired with a Bruker D8 Advance using Cu K α radiation (0.154 nm). Raman spectra were obtained in a Renishaw instrument (in Via-Reflex) with the 532 nm wavelength. X-ray photo electron spectroscopy (XPS) characterizations were carried out on a high resolution electron energy analyzer (Gamma data-Scienta SES 2002) using monochromatic Al K α X-rays. The information on fraction and structure was obtained by a Rigaku Mercury CCD diffractometer applied with a graphite-monochromated Mo K α radiation (0.71073 Å) at 293 K.

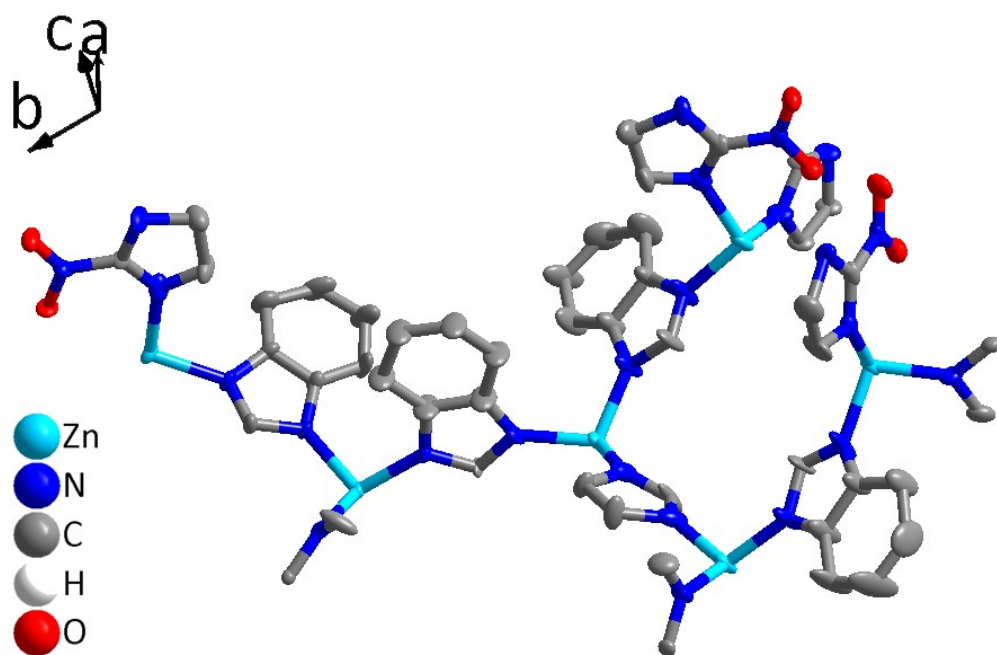


Figure S1. Asymmetric Unit (AU) of **BMM-14** without H atom.

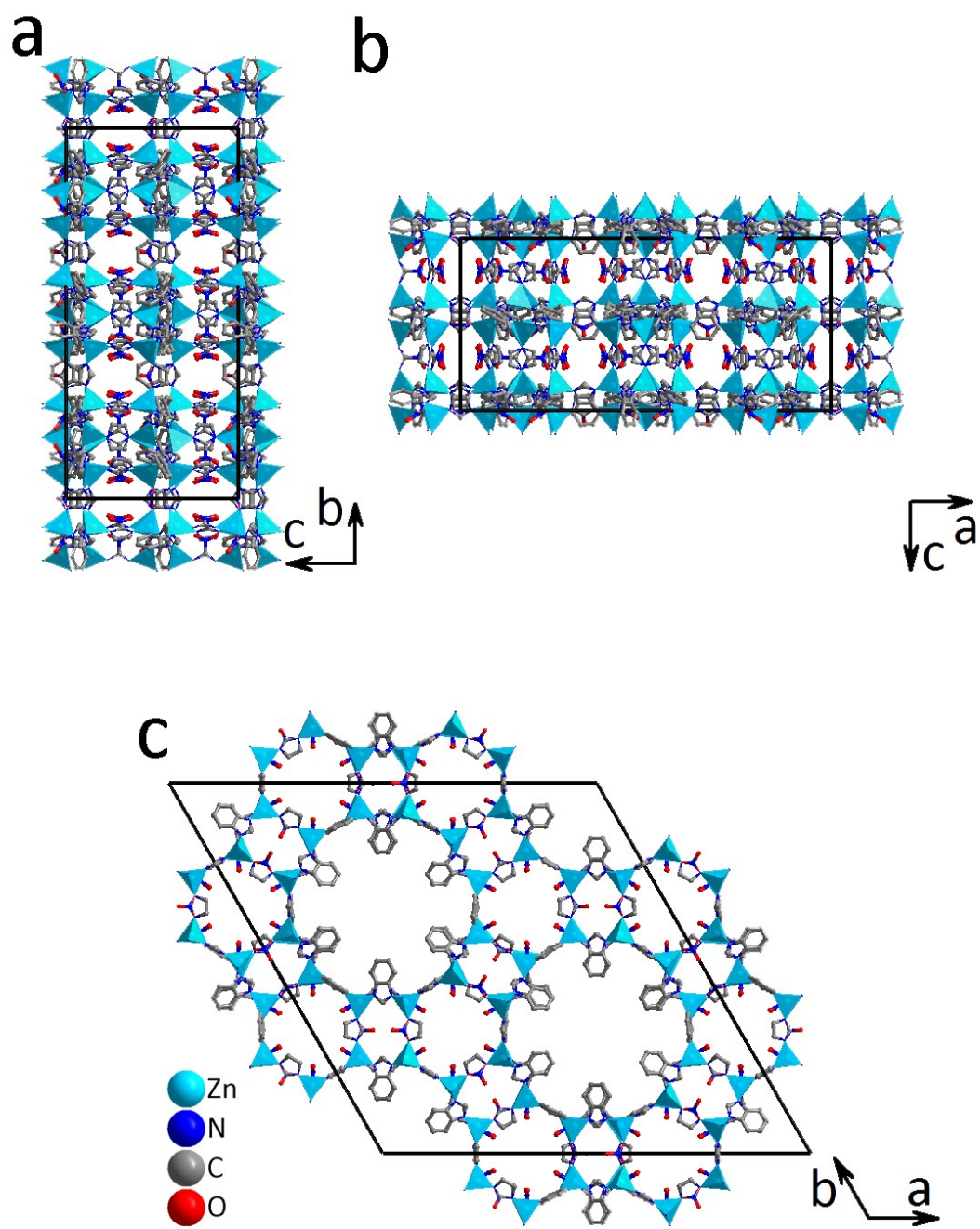


Figure S2. Structure of **BMM-14** along (a) a aix; (b) b aix and (c) c aix

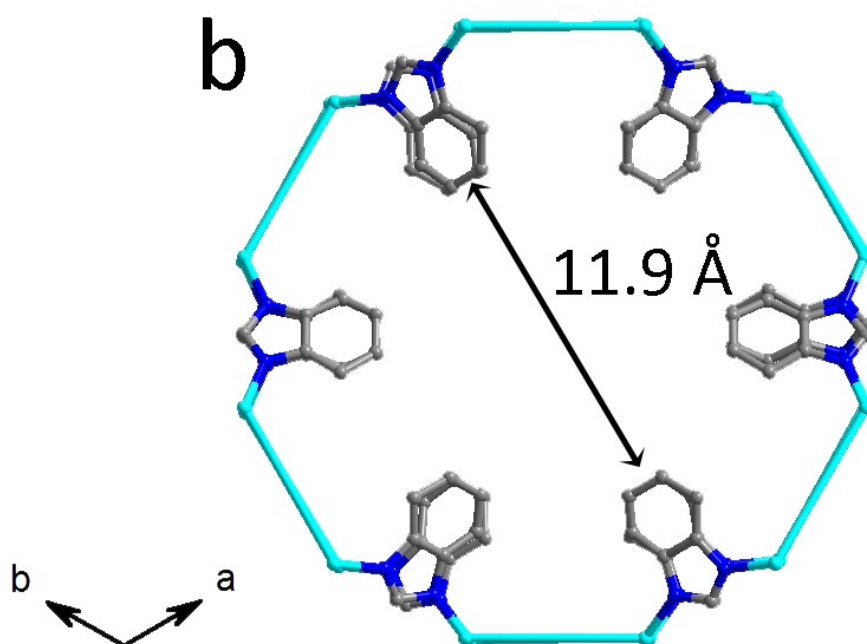
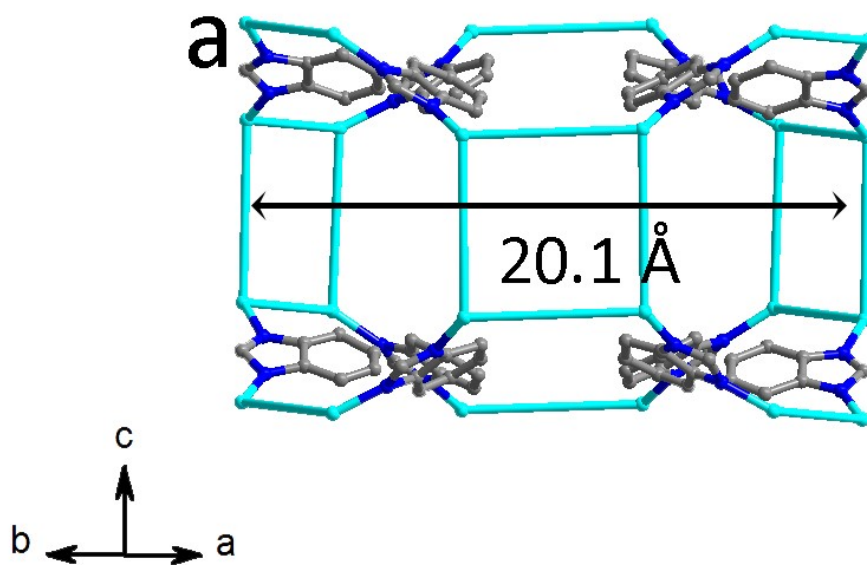


Figure S3. The influence of bIm on the channel width in the **BMM-14** structure, except for the bIm along the *ab* plane, the other ligands are replaced by stick models.

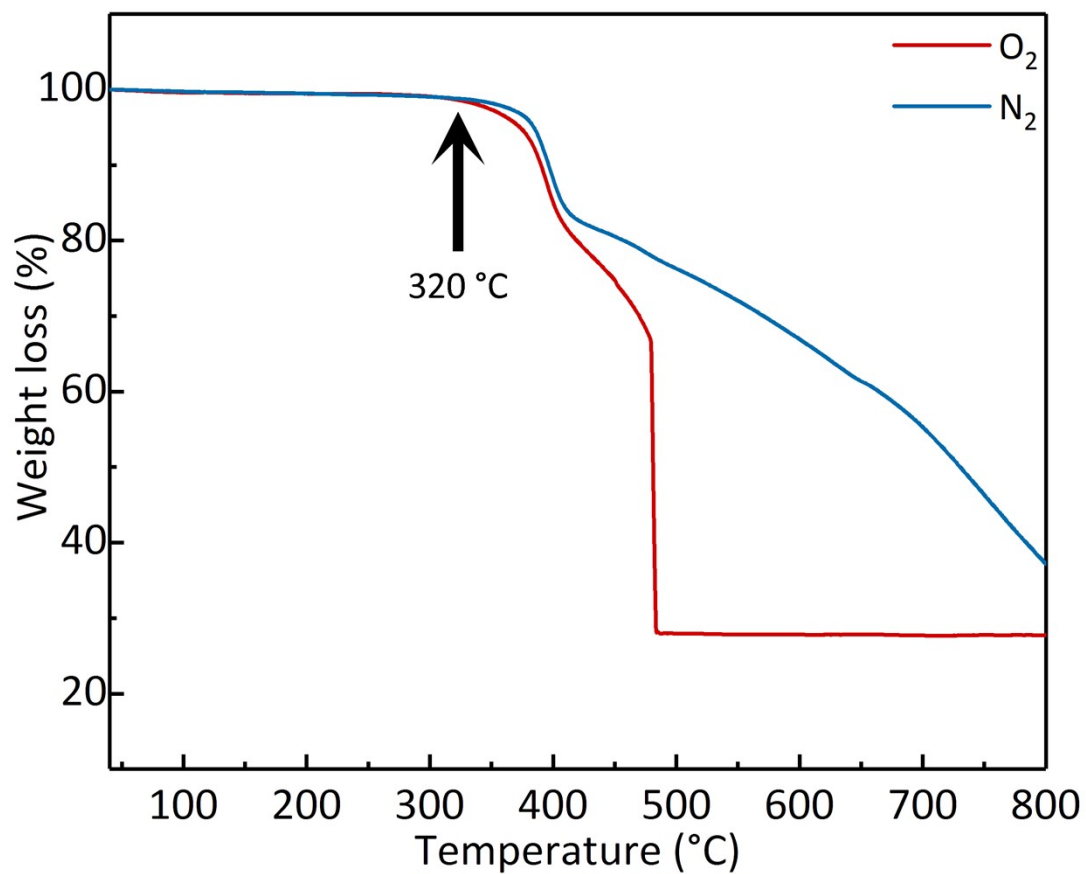


Figure S4. Thermogravimetric analysis (TGA) curves of **BMM-14** in N₂ and O₂ atmosphere.

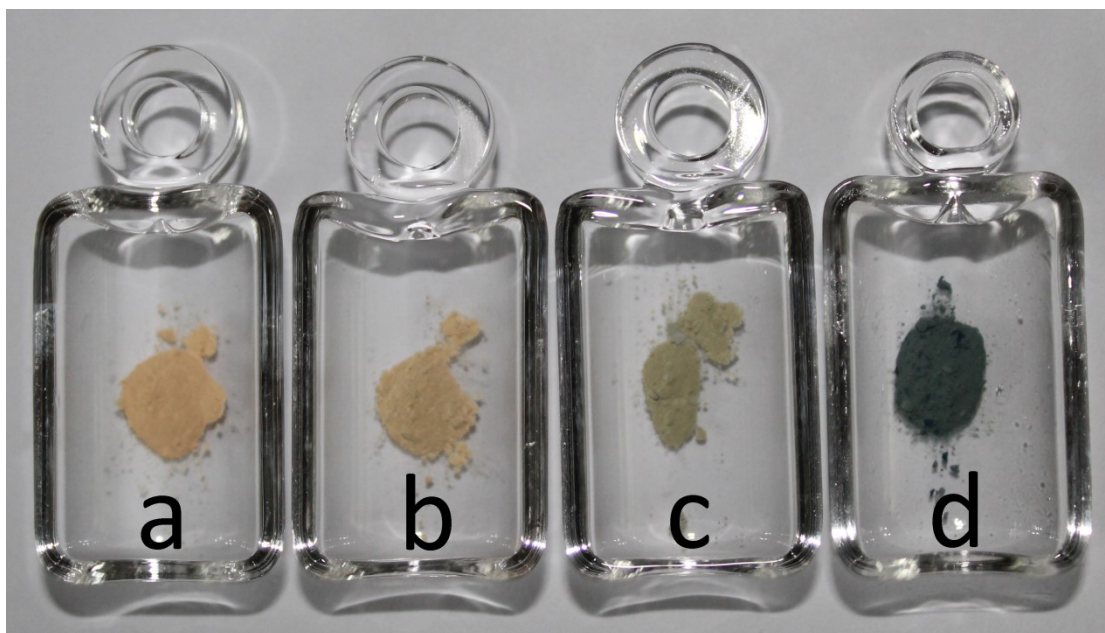


Figure S5. The color of (a) **FePc@BMM-14(0)**, (b) **FePc@BMM-14(0.1)**, (c) **FePc@BMM-14(1)** and (d) **FePc@BMM-14(10)** after dispersion.

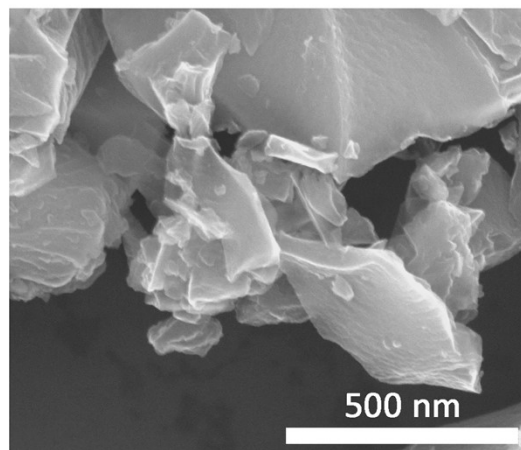
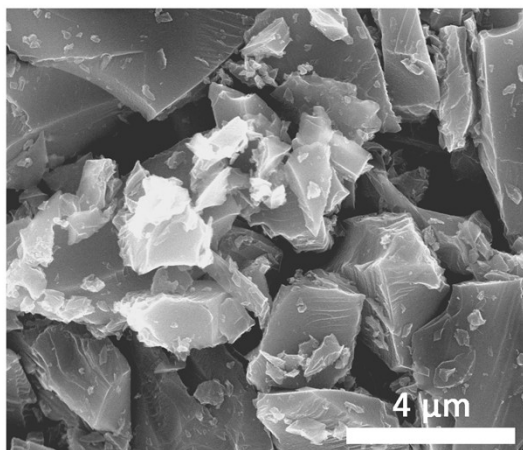


Figure S6. SEM of FePc@BMM-1(0).

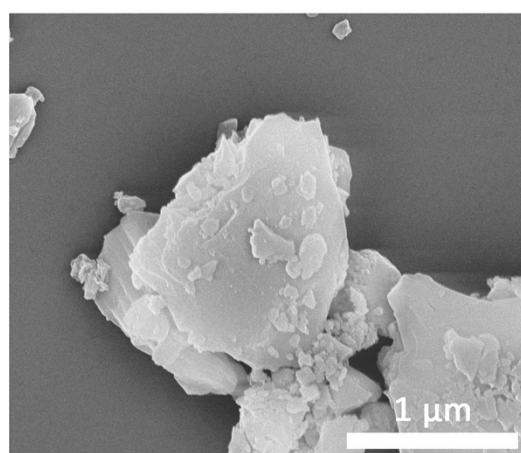
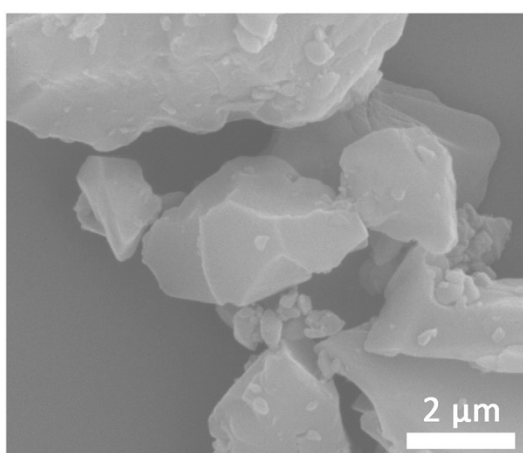


Figure S7. SEM of FePc@BMM-1(0.1).

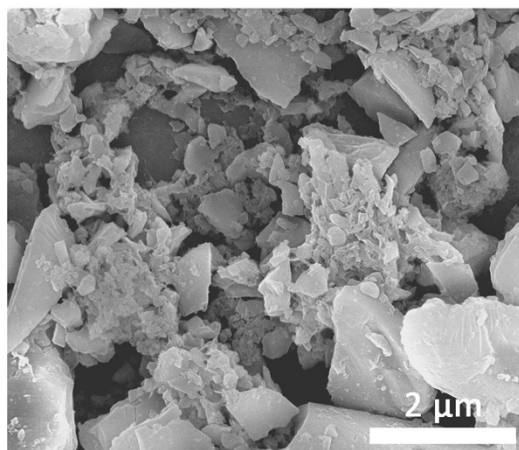
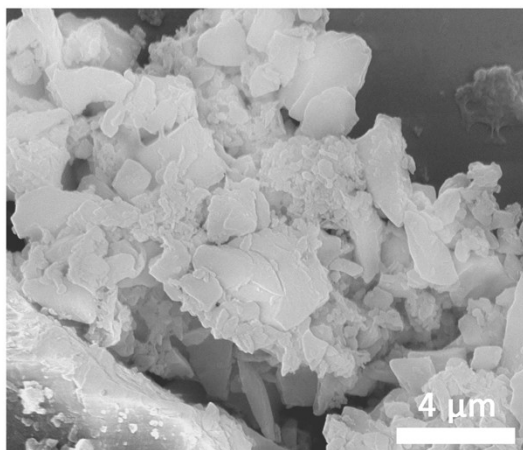


Figure S8. SEM of FePc@BMM-1(1).

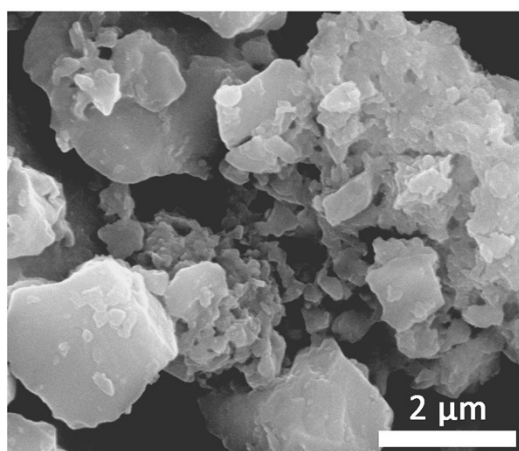
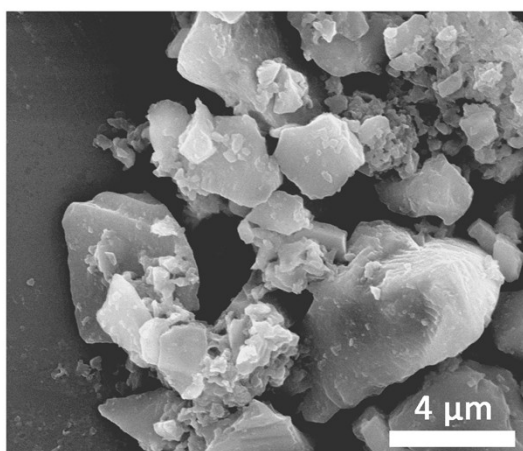


Figure S9. SEM of FePc@BMM-1(10).

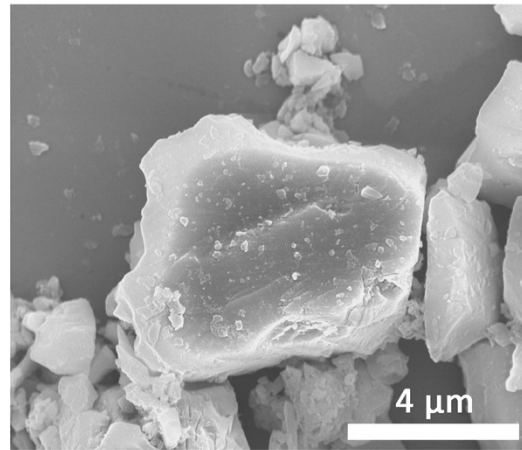
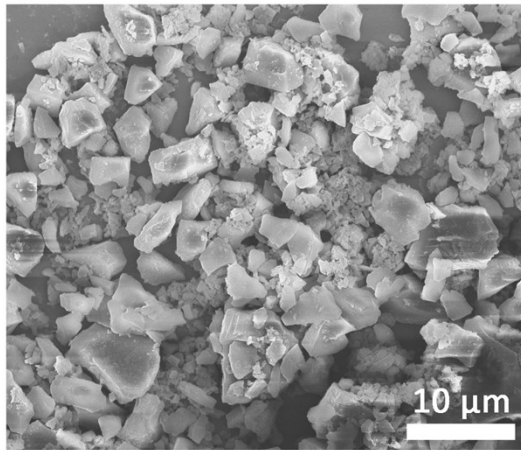


Figure S10. SEM of Fe@NC(0+900).

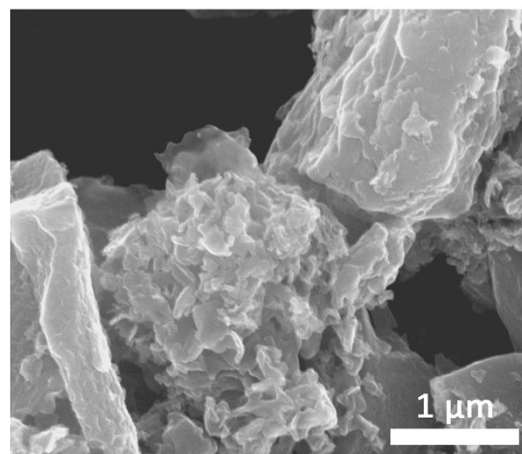
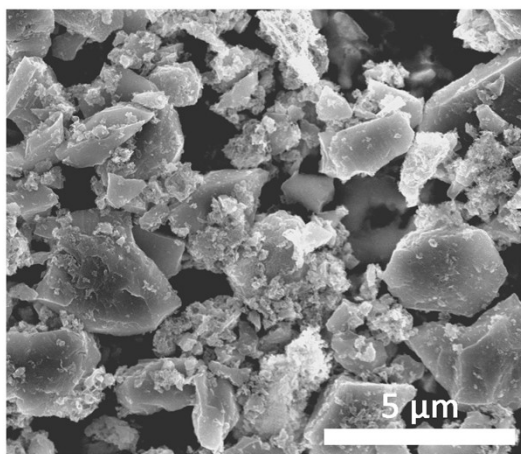


Figure S11. SEM of Fe@NC(10+700).

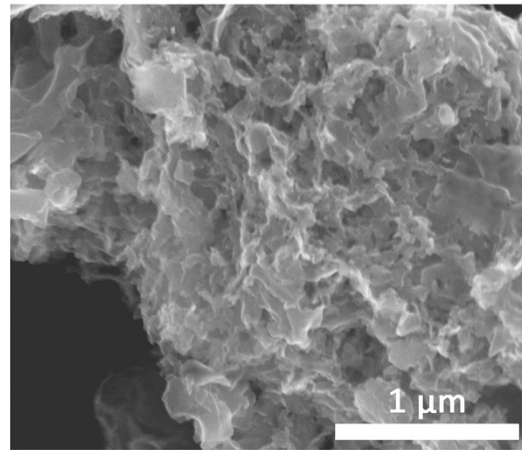
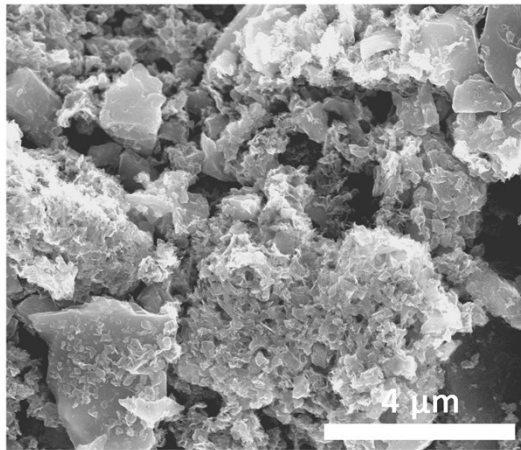


Figure S12. SEM of Fe@NC(10+800).

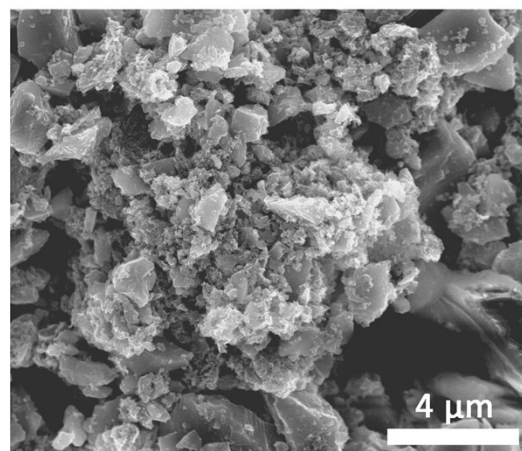
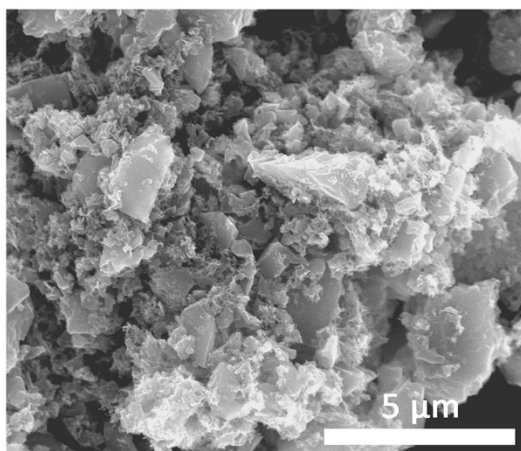


Figure S13. SEM of Fe@NC(10+900).

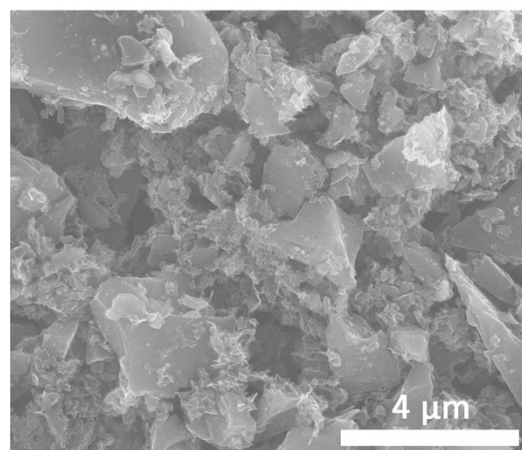
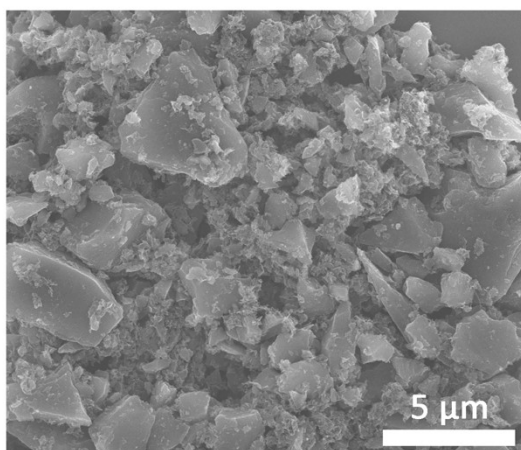


Figure S14. SEM of Fe@NC(10+1000).

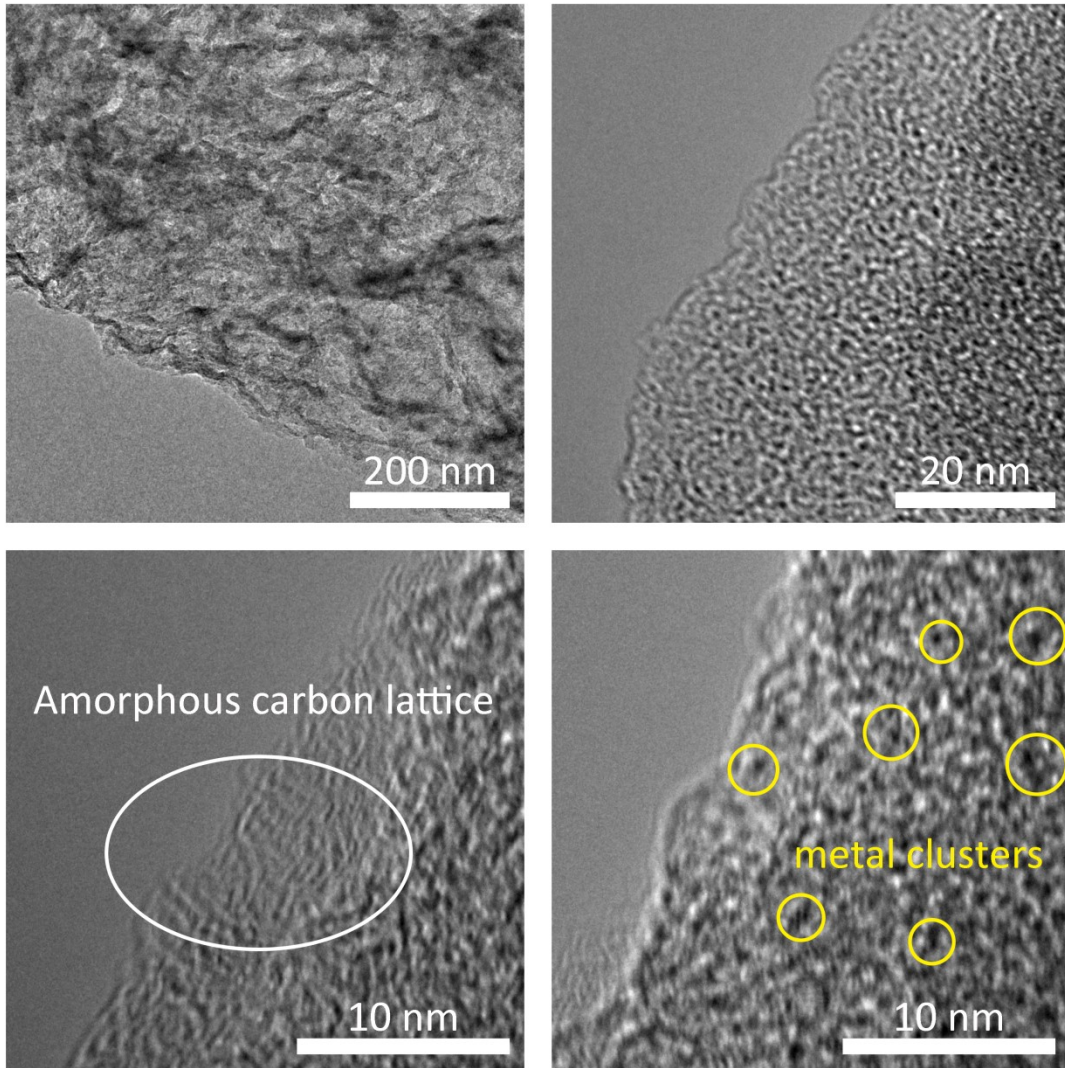


Figure S15. TEM images of **Fe@NC(10+700)**.

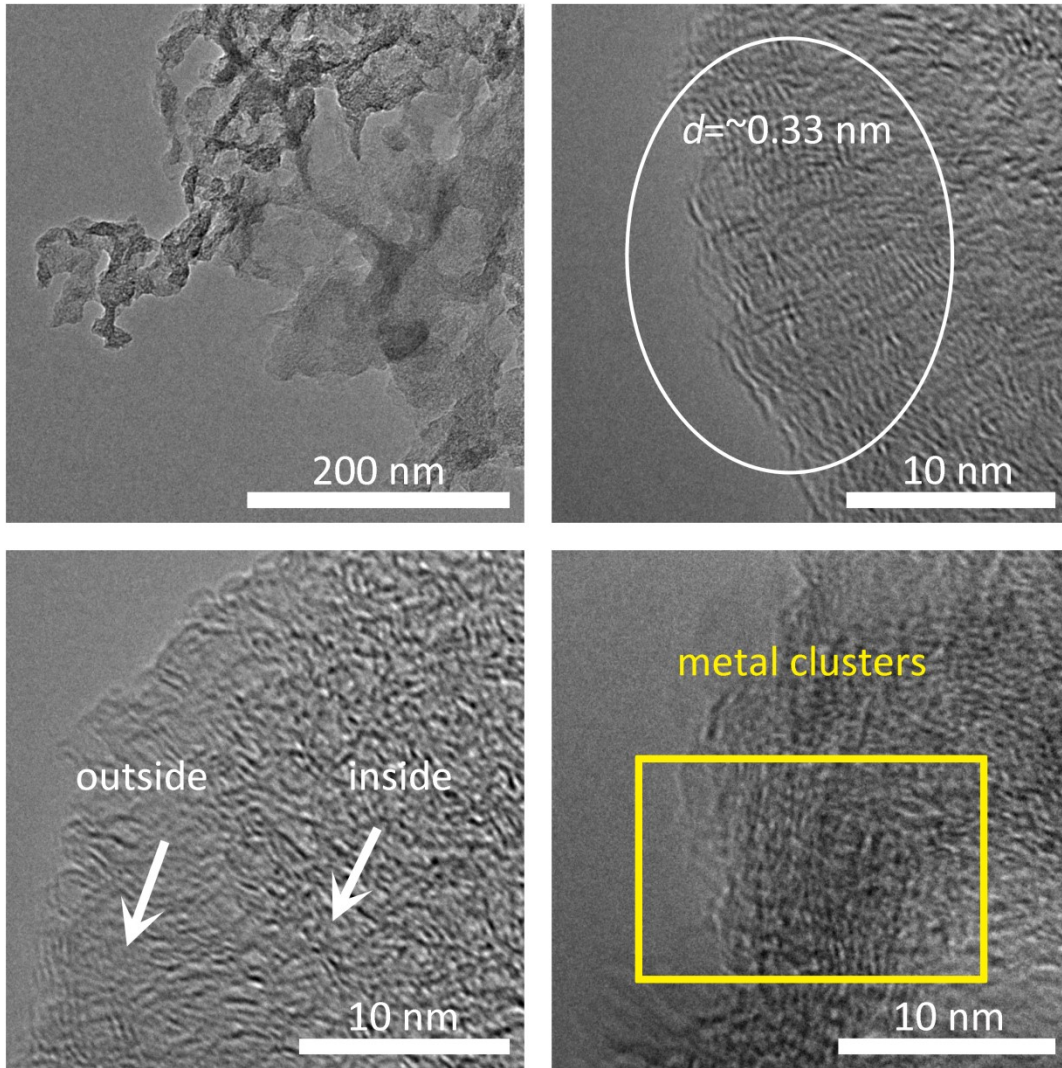


Figure S16. TEM images of **Fe@NC(10+800)**.

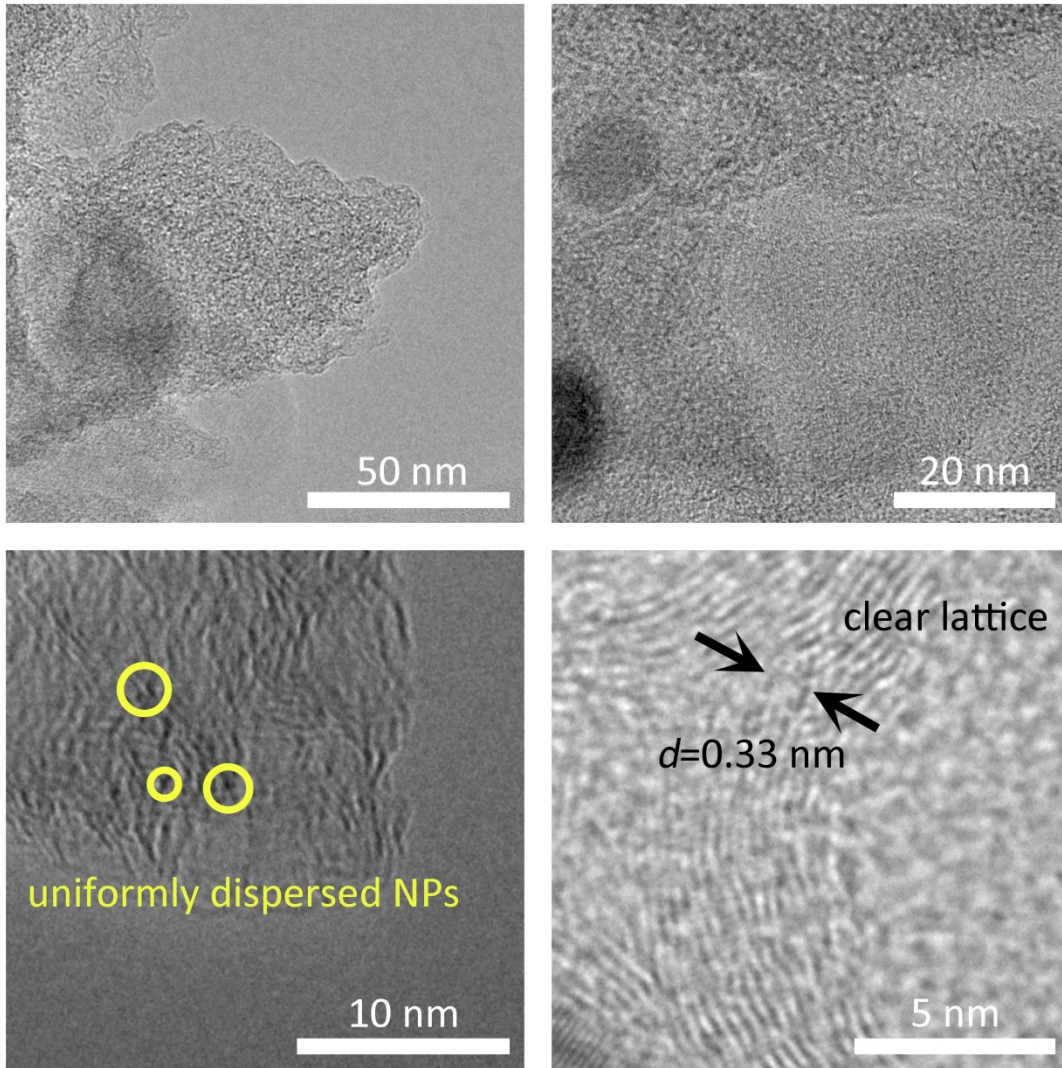


Figure S17. TEM images of **Fe@NC(10+900)**.

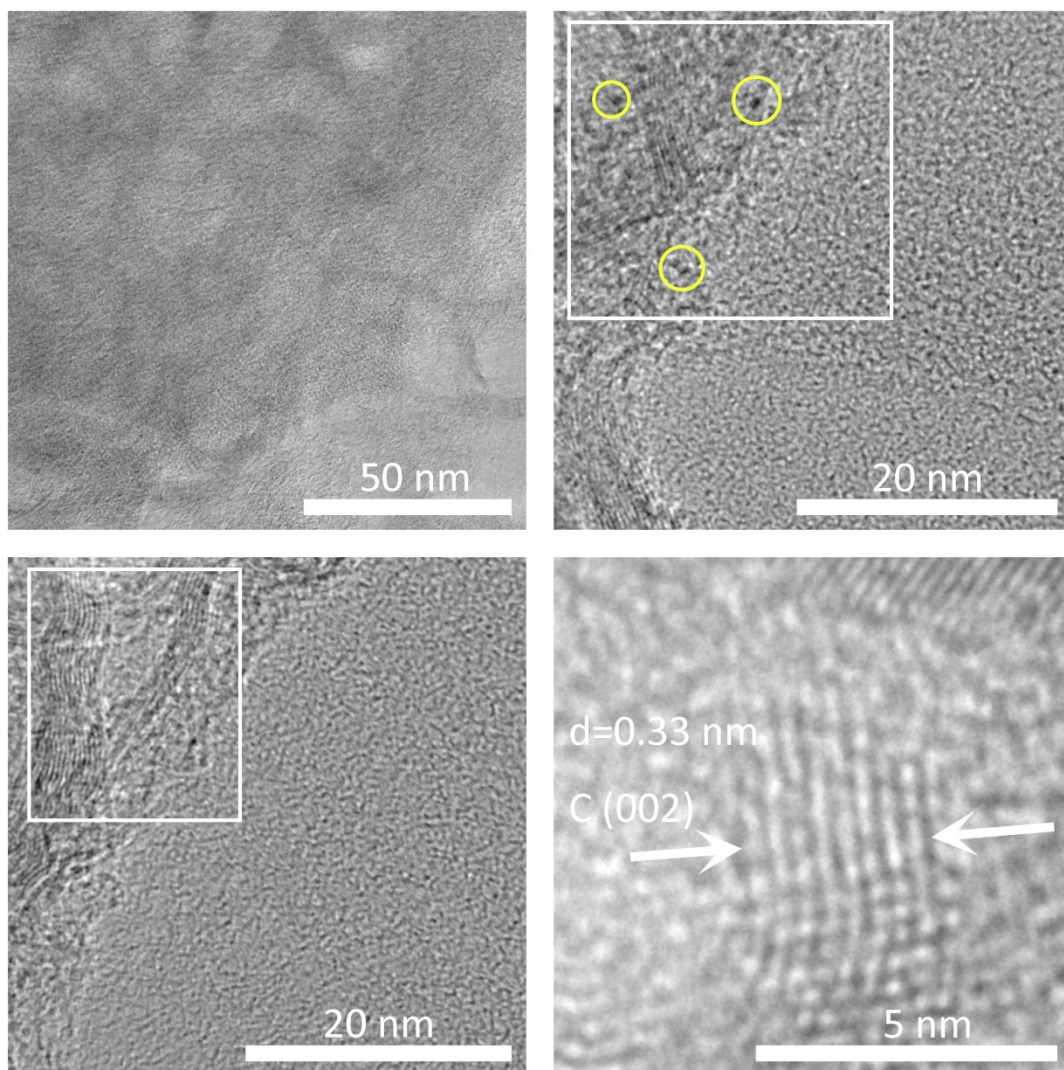


Figure S18. TEM images of **Fe@NC(10+1000)**.

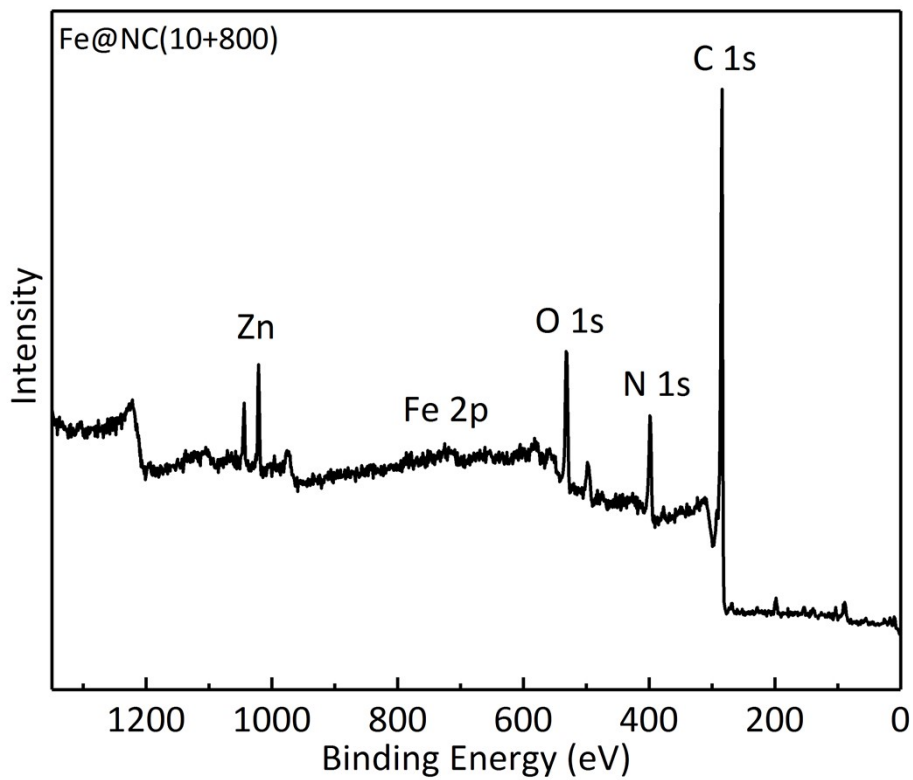
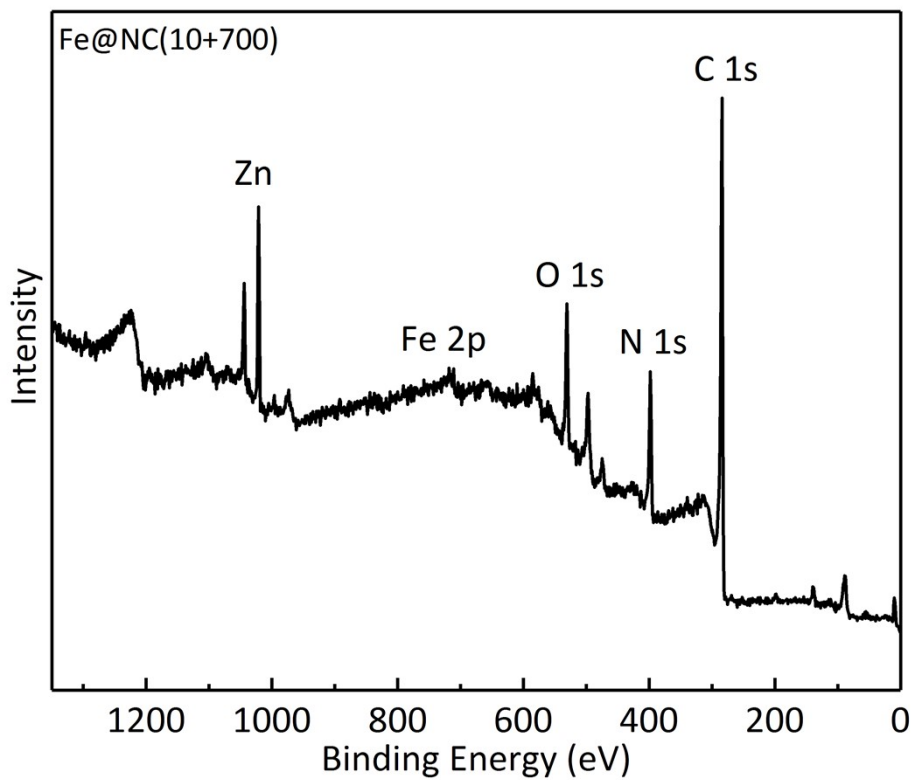


Figure S19. Full XPS survey spectra of Fe@NC(10+700) and Fe@NC(10+800).

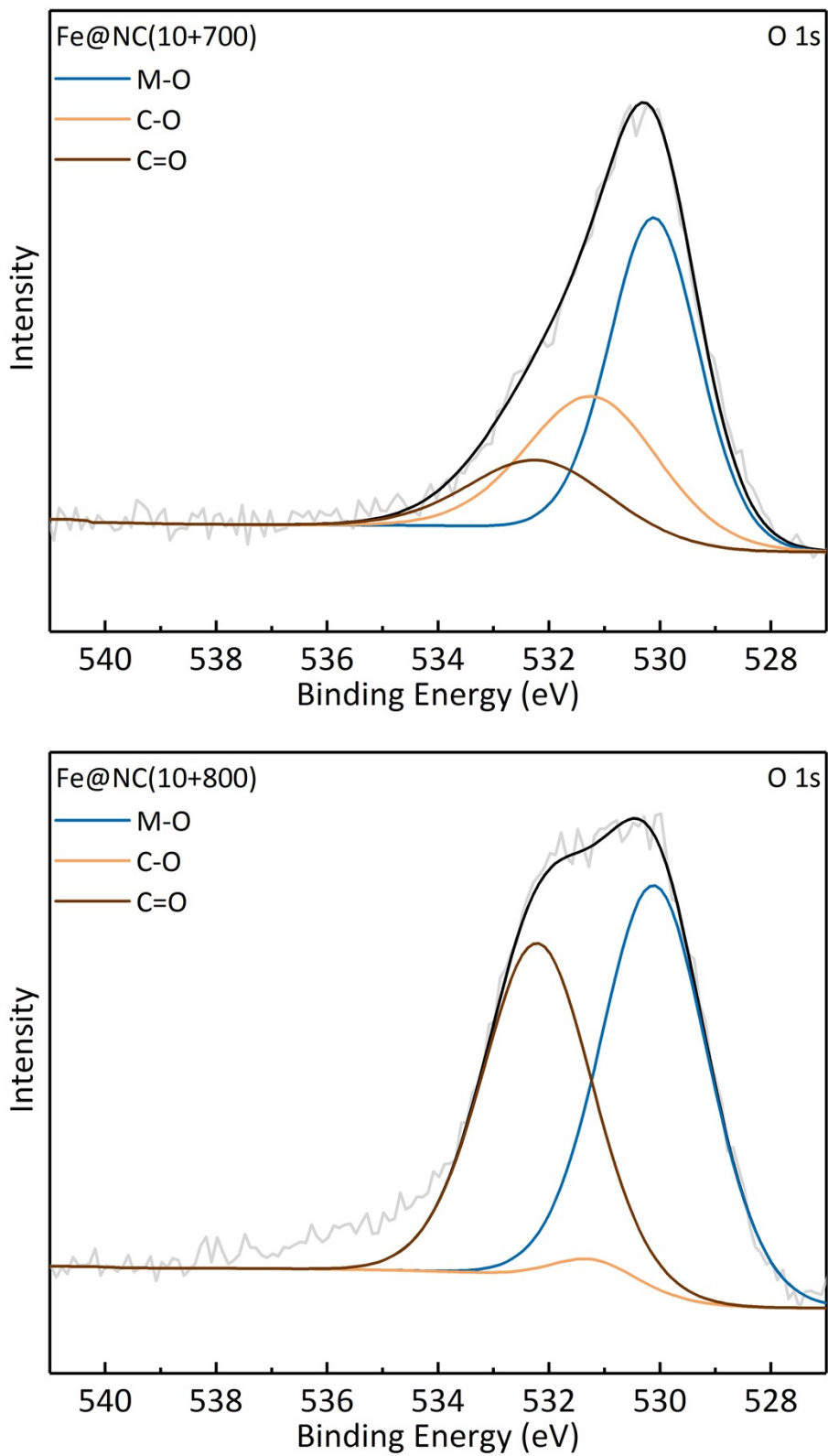


Figure S20. The deconvoluted O 1s survey spectra of **Fe@NC(10+700)** and **Fe@NC(10+800)**.

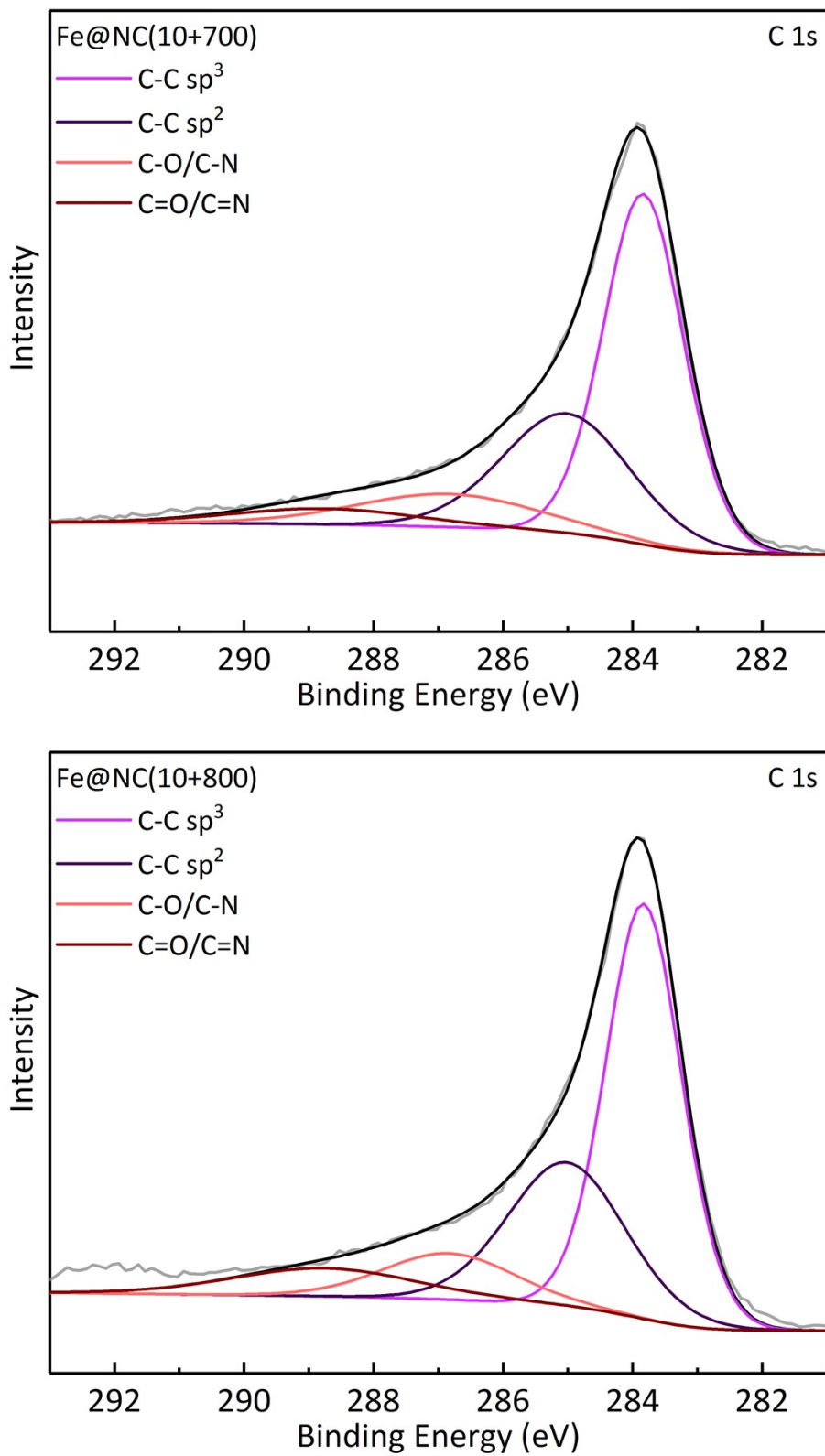


Figure S21. The deconvoluted C 1s survey spectra of **Fe@NC(10+700)** and **Fe@NC(10+800)**

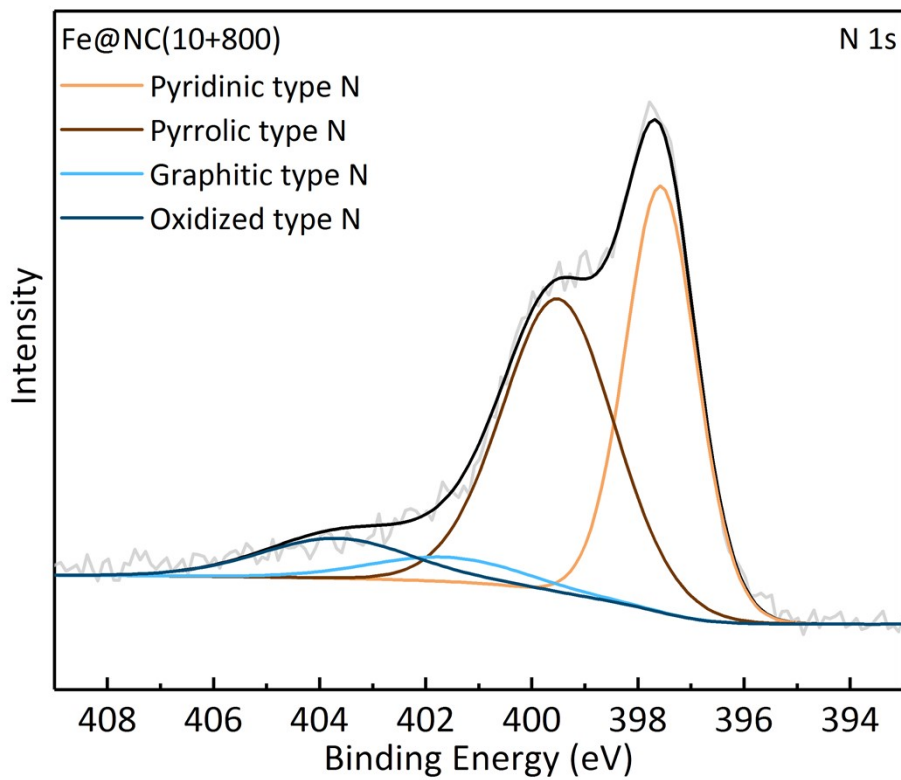
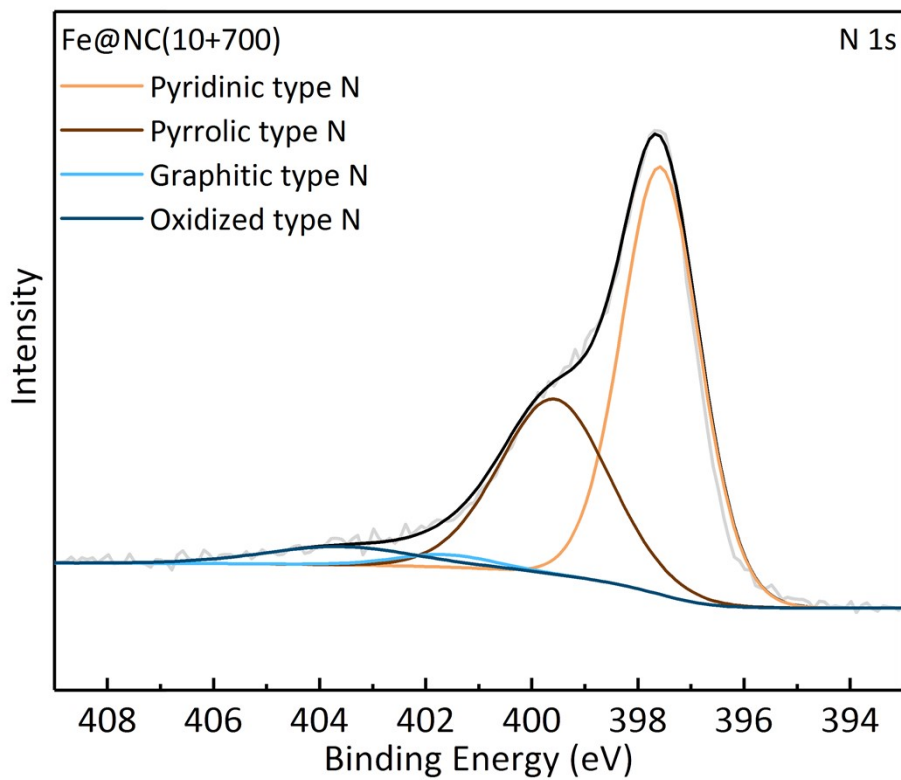


Figure S22. The deconvoluted N 1s survey spectra of **Fe@NC(10+700)** and **Fe@NC(10+800)**.

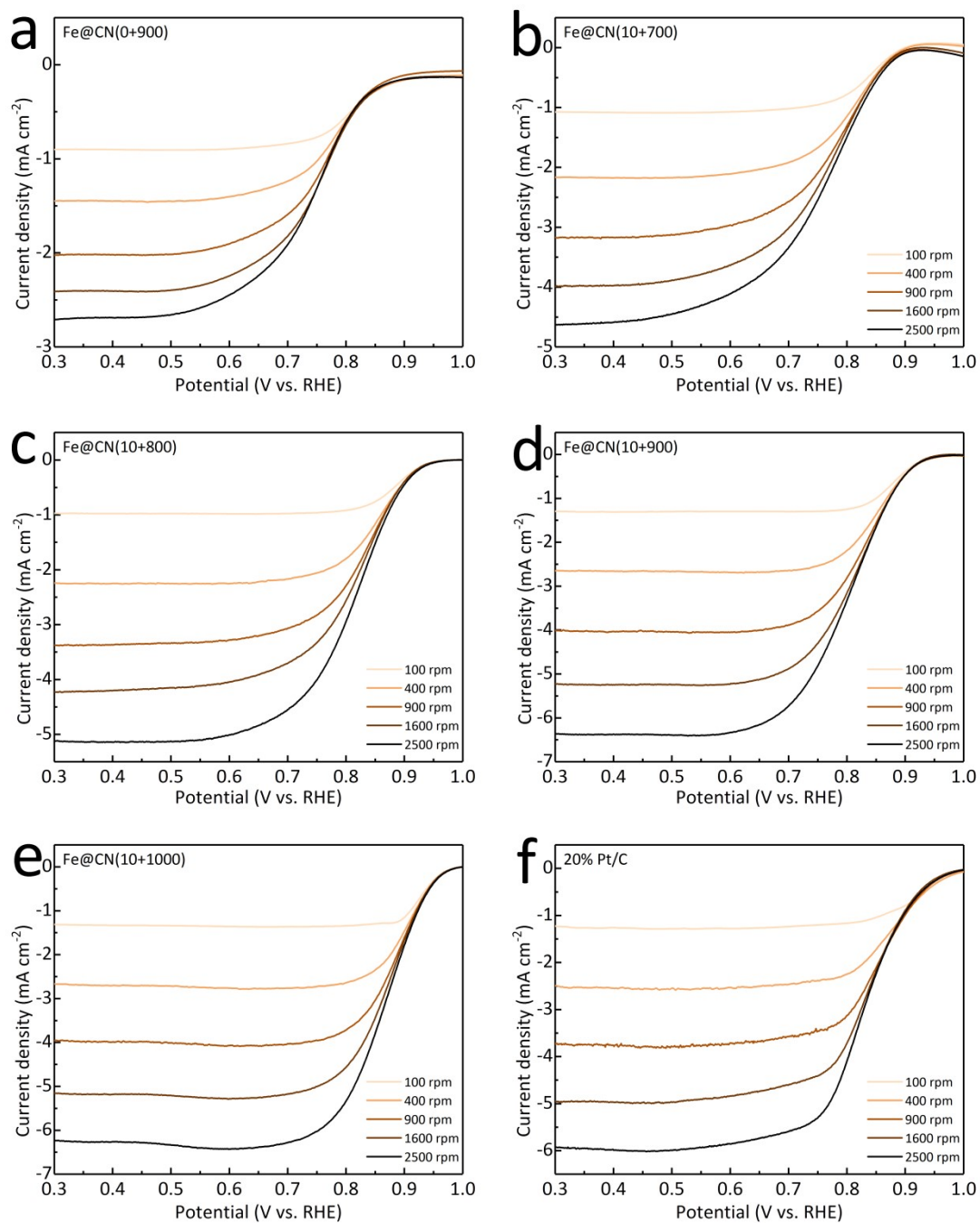


Figure S23. RRDE plots with various rotation speeds at a scan rate of 5 mV s^{-1} in O_2 -saturated 0.1 M KOH solution of (a) $\text{Fe@NC}(0+900)$, (b) $\text{Fe@NC}(10+700)$, (c) $\text{Fe@NC}(10+800)$, (d) $\text{Fe@NC}(10+900)$, (e) $\text{Fe@NC}(10+1000)$ and (f) $20\% \text{ Pt/C}$.

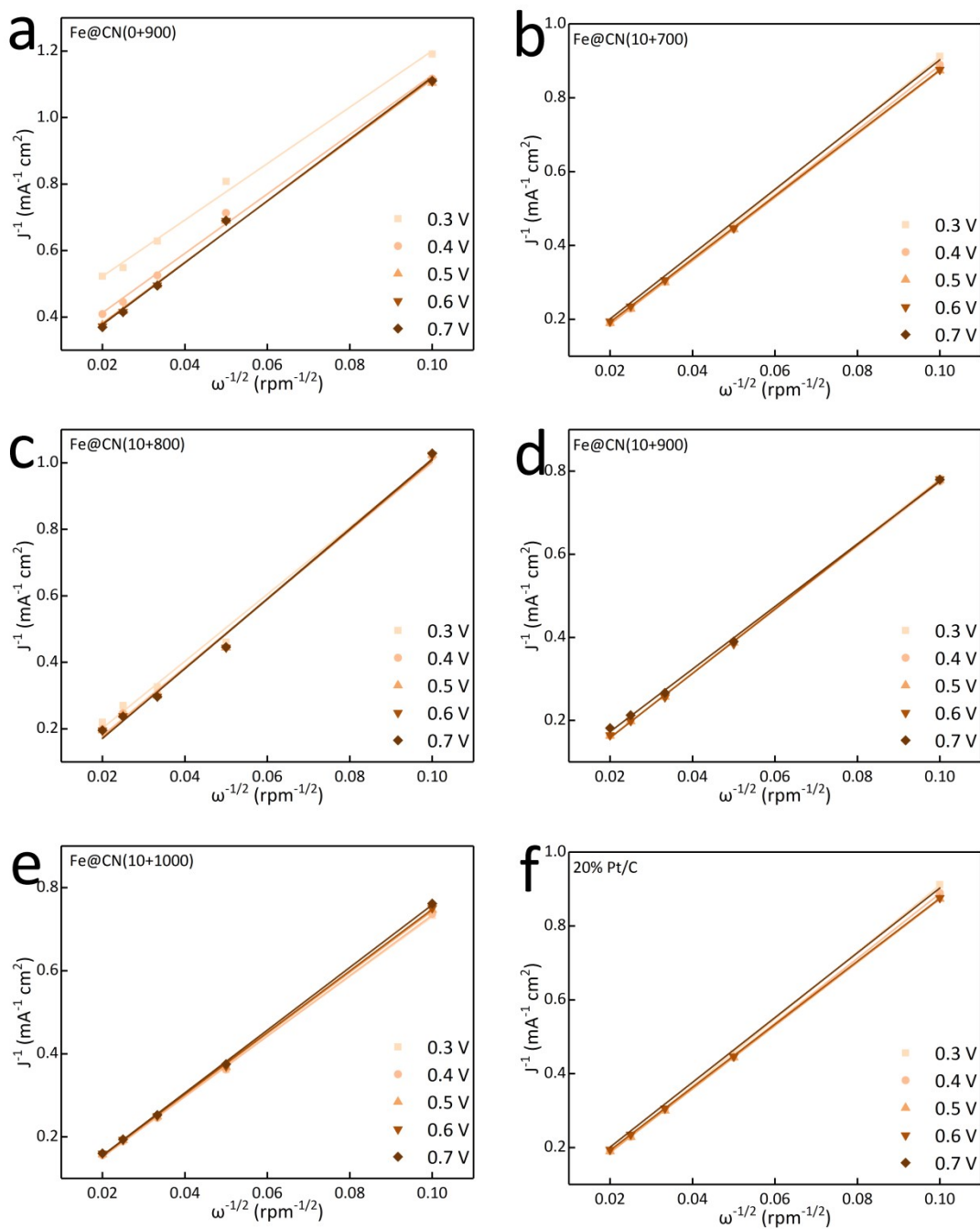


Figure S24. K-L plots at various voltages at 1600 rpm in O_2 -saturated 0.1 M KOH solution of (a) Fe@NC(0+900), (b) Fe@NC(10+700), (c) Fe@NC(10+800), (d) Fe@NC(10+900), (e) Fe@NC(10+1000) and (f) 20% Pt/C.

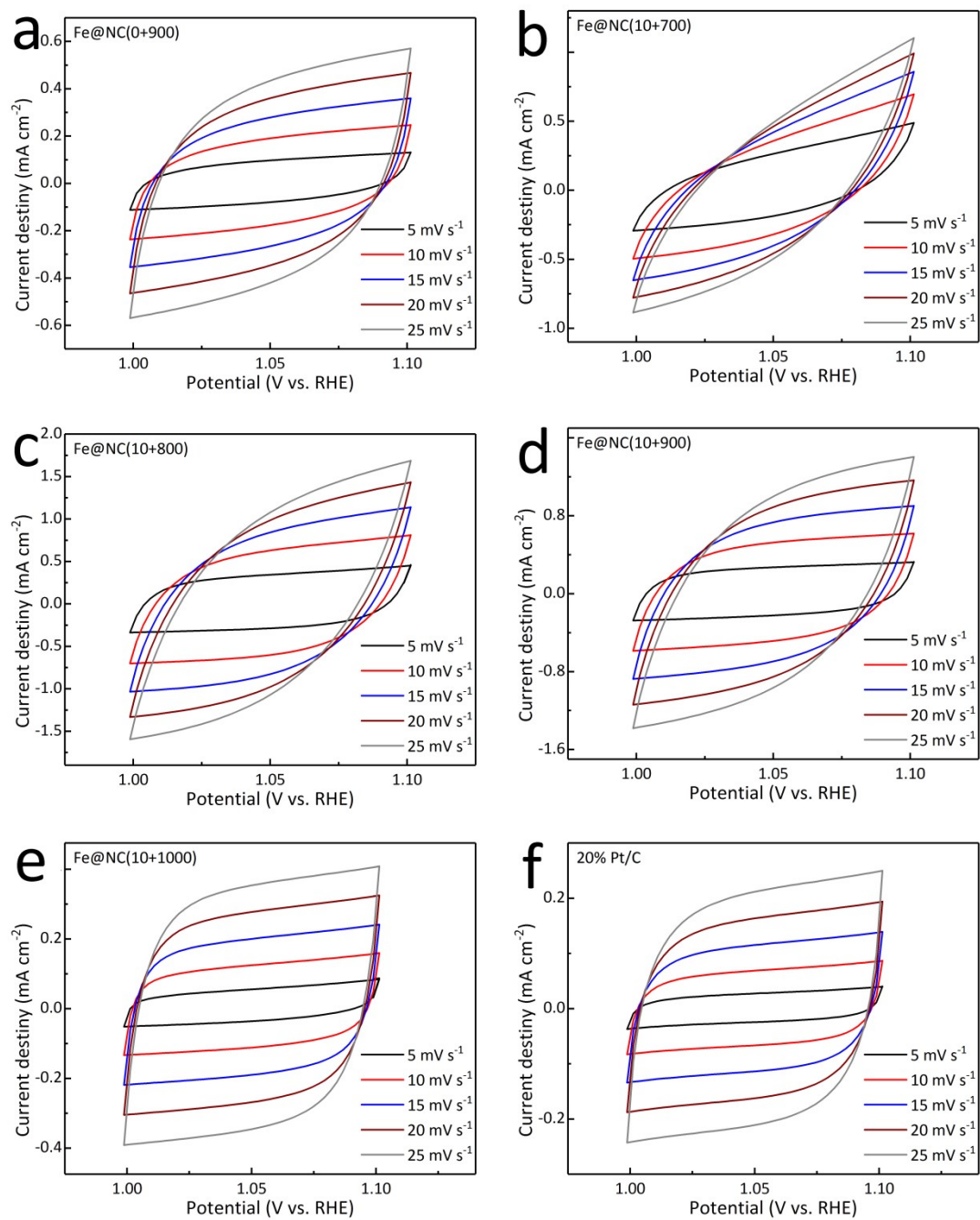


Figure S25. The C_{dl} is calculated from the cyclic voltammograms of (a) Fe@NC(0+900), (b) Fe@NC(10+700), (c) Fe@NC(10+800), (d) Fe@NC(10+900), (e) Fe@NC(10+1000) and (f) 20% Pt/C

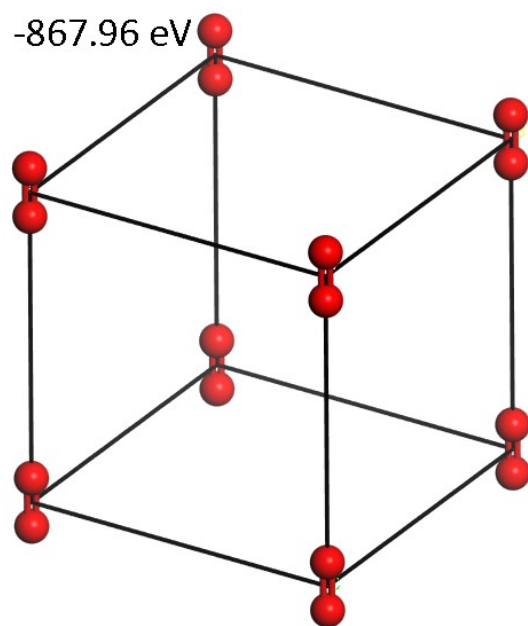


Figure S26. Geometrically optimized model for a O_2 .

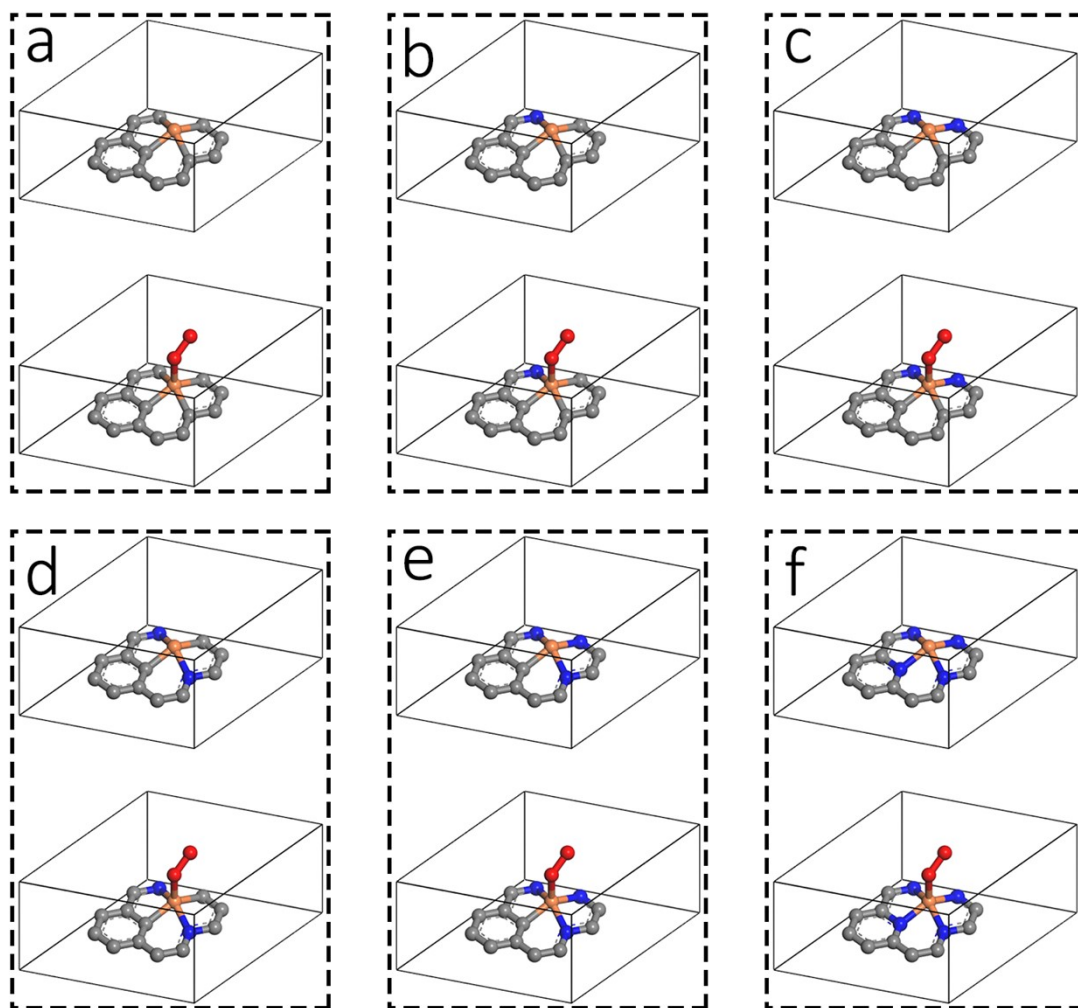


Figure S27. Optimized nanostructures of different 4-coordinate Fe active sites.

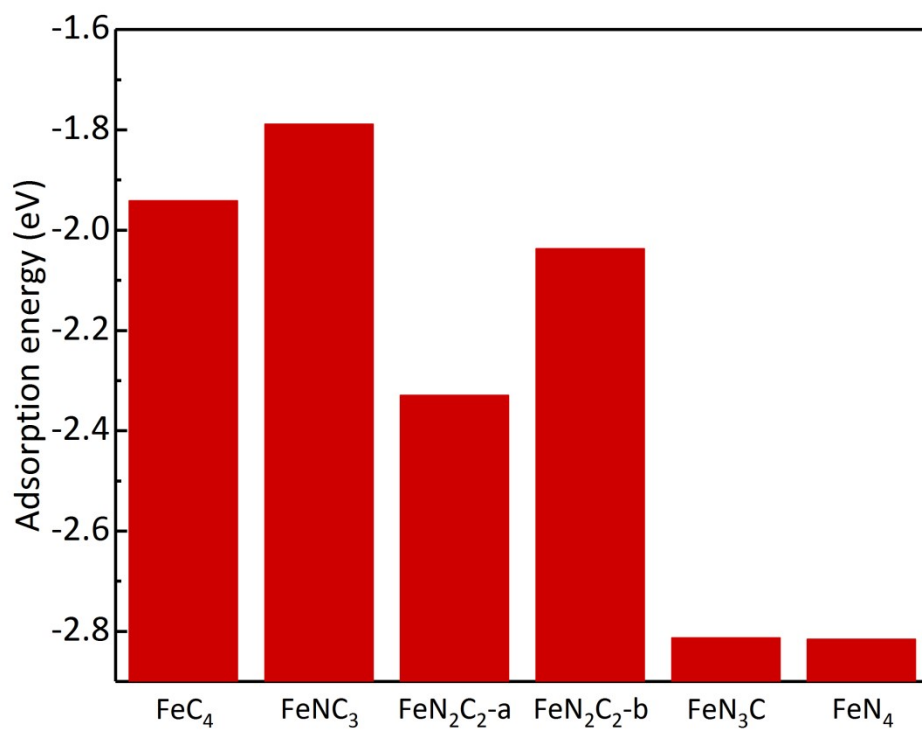


Figure S28. O₂ loaded adsorption energy of FeN_xC_y.

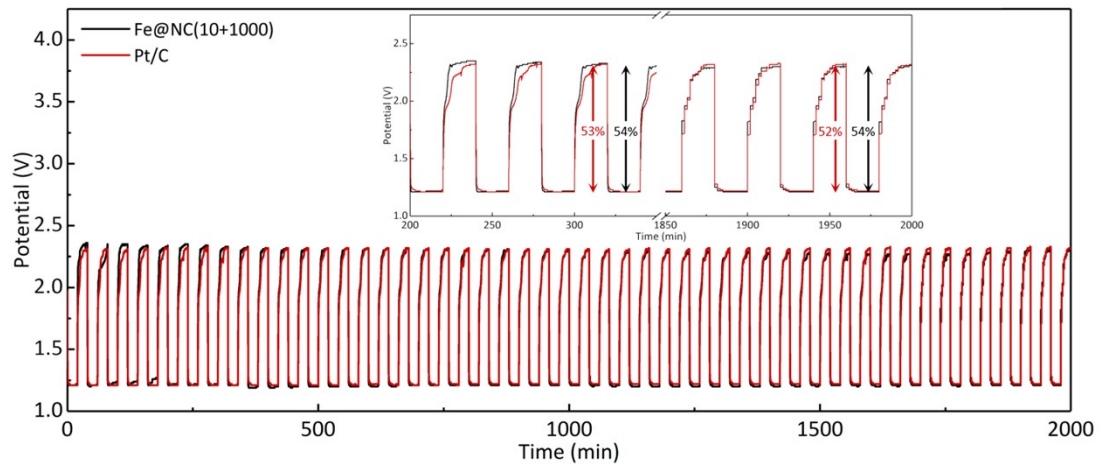


Figure S29. The long-term stability at 5 mA cm⁻² of the zinc-air battery based on **Fe@NC(10+1000)** and Pt/C.

Table S1. Summary of Crystal Data and Refinement Results for BMM-14.

Items	BMM-14
CCDC	2130792
Formula	C ₉₅ H ₇₀ N ₅₁ O ₂₁ Zn ₁₂
Mass	1523.23
crystal system	hexagonal
space group	<i>P63cm</i>
<i>a</i> (Å)	46.2077
<i>b</i> (Å)	46.2077
<i>c</i> (Å)	18.6293
α (°)	90.00
β (°)	90.00
γ (°)	120.00
<i>V</i> (Å ³)	34447.3
<i>T</i> (K)	250
<i>Z</i>	12
F(000)	9150.0
R _{int}	0.1619
R ₁ (I>2σ(I))	0.1619
wR ₂ (all reflections)	0.3658

Table S2. Summary of Gas Sorption Results for BMM-14.

	BET surface area (m ² g ⁻¹)	NLDFT micropore volume (ml g ⁻¹)	NLDFT totalpore volume (ml g ⁻¹)
Expe_BMM-14	1117.4987	0.5620	0.6991
FePc@BMM-14(0.1)	763.9485	0.3794	0.4647
FePc@BMM-14(1)	679.5249	0.3347	0.4536
FePc@BMM-14(10)	563.0729	0.2817	0.3881

Table S3. O 1s Survey Spectra of Fe@NC(x+y).

	M-O	C-O	C=O
Fe@NC(0+900)	530.1	531.2	532.2
Fe@NC(10+700)	530.1	531.2	532.2
Fe@NC(10+800)	530.1	531.4	532.2
Fe@NC(10+900)	530.1	531.3	532.2
Fe@NC(10+1000)	530.1	531.2	532.2

Table S4. C 1s Survey Spectra of Fe@NC(x+y).

	C-C sp ³	C=C sp ²	C-N	C=N
Fe@NC(0+900)	283.8	285.0	286.9	288.8
Fe@NC(10+700)	283.8	285.0	286.9	288.8
Fe@NC(10+800)	283.8	285.0	286.9	288.8
Fe@NC(10+900)	283.8	285.0	286.9	288.8
Fe@NC(10+1000)	283.8	285.0	286.9	288.8

Table S5. N 1s Survey Spectra of Fe@NC(x+y).

	Pyridinic-N	Pyrrolic-N	Graphitic-N	Oxidized-N
Fe@NC(0+900)	397.6	399.7	401.7	403.6
Fe@NC(10+700)	397.6	399.6	401.8	403.7
Fe@NC(10+800)	397.6	399.6	401.8	403.8
Fe@NC(10+900)	397.6	399.6	401.7	403.8
Fe@NC(10+1000)	397.6	399.7	401.7	403.6

Table S6. Zn 2p Survey Spectra of Fe@NC(x+y).

	Zn(II) 2p_{3/2}	Zn(II) 2p_{1/2}
Fe@NC(0+900)	1020.78	1043.83
Fe@NC(10+700)	1020.68	1043.68
Fe@NC(10+800)	1020.48	1043.58
Fe@NC(10+900)	1020.93	1044.23
Fe@NC(10+1000)	1020.98	1044.18

Table S7. Fe 2p Survey Spectra of Fe@NC(x+y).

	Fe(0)	Fe(II)	Fe(III)	Fe(0)	Fe(II)	Fe(III)
	2p_{3/2}	2p_{3/2}	2p_{3/2}	2p_{1/2}	2p_{1/2}	2p_{1/2}
Fe@NC(10+700)	706.88	709.48	712.28	720.58	722.18	724.58
Fe@NC(10+800)	706.98	709.58	712.38	720.68	722.28	724.68
Fe@NC(10+900)	706.78	709.28	711.83	720.08	721.83	724.03
Fe@NC(10+1000)	705.98	708.58	711.08	719.38	721.18	723.28

Table S8. Comparison of ORR Performance for Fe@NC(10+900/1000) with Reported Carbon-Based Electrocatalysts.

Catalyst	E_{onset} (V vs. RHE)	E_{1/2} (V vs. RHE)	J₁₆₀₀ (mA cm⁻¹)	Reference
C-CZ-4-1000	1.03	0.89	-5.9	1
Co-C@NWCs	0.94	0.83	-4.51	2
CoHNCS-0.2	0.94	0.82	-5.8	3
Co,N-CNF	0.88	0.81	-5.7	4
Fe-N-DSC	1.03	0.83	/	5
Fe-N-RFC_C240	1.00	0.91	-5.5	6
Fe-N-C HNSs	1.05	0.84	-5.8	7
Fe@NC(10+900)	0.94	0.82	-5.2	This work
Fe@NC(10+1000)	0.97	0.88	-5.2	This work

1. L. Ge, Y. Yang, L. Wang, W. Zhou, R. D. Marco, Z. G. Chen, J. Zou, Z. H. Zhu, High activity electrocatalysts from metal-organic framework-carbon nanotube templates for the oxygen reduction reaction, *Carbon* **2015**, 82, 417-424.
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