

Supporting Information

KNa_{0.78}Eu_{0.27}In_{3.80}B₁₂S₁₂: A Novel Hexanary Thioborate Featuring B₁₂S₁₂ Cluster and Diverse InS_x (x = 4, 5, 6) Units

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Table S1. EDS analysis for **1**

Single crystal	#1	#2	#3
In	20.92	20.83	21.66
Eu	1.74	1.56	1.69
In/Eu	12.02	13.35	13.82

Table S2. The Rietveld powder XRD pattern refinement result.

	<i>R</i> (obs)	<i>wR</i> (obs)	<i>R</i> (all)	<i>wR</i> (all)	<i>R</i> _p	<i>wR</i> _p
value	4.35	5.25	4.36	5.26	3.19	4.27

Table S3. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters (U_{eq}^a , $\text{\AA}^2 \times 10^3$) for **1**.

Atom	Wyck. site	x	y	z	$U_{\text{eq}}/\text{\AA}^2$	SOF
K(1)	$4i$	3869.2(18)	0	8308.4(18)	47.1(6)	1
Na(1A)	$4i$	5487(5)	5000	4678(5)	42(2)	0.564
Na(1B)	$2c$	5000	5000	5000	42(2)	0.436
Eu(1)	$4f$	2500	2500	5000	35.6(7)	0.268
In(1)	$4e$	7500	2500	10000	20.7(3)	0.798
In(2)	$4i$	7563.8(4)	0	7464.1(4)	20.15(17)	1
In(3)	$8j$	5353.4(3)	2688.6(4)	7381.2(3)	23.33(15)	1
B(1)	$4i$	4953(6)	5000	8835(6)	16.4(19)	1
B(2)	$4i$	6151(6)	0	5349(6)	13.8(18)	1
B(3)	$4i$	5467(6)	0	6158(6)	12.9(18)	1
B(4)	$8j$	4456(4)	863(6)	5675(4)	13.9(13)	1
B(5)	$4i$	6027(6)	5000	9712(6)	14.6(19)	1
B(6)	$8j$	5315(4)	3622(6)	9558(4)	14.5(13)	1
B(7)	$8j$	4486(4)	1407(6)	4527(4)	12.7(12)	1
B(8)	$8j$	5853(4)	4142(6)	10716(4)	15.8(13)	1
S(1)	$5i$	5938.5(13)	0	7449.7(13)	16.6(4)	1
S(2)	$4i$	7430.2(13)	0	5754.8(13)	17.8(4)	1
S(3)	$8j$	3902.2(9)	1744.9(15)	6482.4(10)	19.1(3)	1
S(4)	$4i$	4820.5(15)	5000	7541.4(13)	21.6(5)	1
S(5)	$8j$	8233.1(9)	1837.5(14)	8501.7(10)	18.2(3)	1
S(6)	$8j$	6095.5(10)	2866.1(15)	6079.8(10)	21.2(3)	1
S(7)	$8j$	5689.3(10)	2126.2(14)	9094.0(10)	20.0(3)	1
S(8)	$4i$	7143.8(15)	5000	9412.6(16)	36.3(6)	1

^a U_{eq} is defined as one third of the trace of the orthogonalized U_{ij} tensor.

Table S4. Selected bond distances (Å) for **1**.^a

Bond	Dist.	Bond	Dist.
Eu(1)–S(2)#8	2.8478(8)	B(1)–B(5)	1.799(12)
Eu(1)–S(2)#9	2.8478(8)	B(1)–B(6)#7	1.786(8)
Eu(1)–S(3)	2.7360(14)	B(1)–B(6)	1.786(8)
Eu(1)–S(3)#10	2.7360(14)	B(1)–B(8)#12	1.779(9)
Eu(1)–S(6)#5	2.9966(14)	B(1)–B(8)#13	1.779(9)
Eu(1)–S(6)#3	2.9966(14)	B(2)–B(3)	1.769(12)
K(1)–S(1)	3.687(3)	B(2)–B(4)#5	1.786(9)
K(1)–S(3)#1	3.243(2)	B(2)–B(4)#9	1.786(9)
K(1)–S(3)	3.243(2)	B(2)–B(7)#9	1.797(8)
K(1)–S(5)#2	3.4754(17)	B(2)–B(7)#5	1.797(8)
K(1)–S(5)#3	3.4754(17)	B(3)–B(4)#1	1.768(9)
K(1)–S(7)	3.508(2)	B(3)–B(4)	1.768(9)
K(1)–S(7)#1	3.508(2)	B(3)–B(7)#9	1.789(8)
K(1)–S(8)#2	3.425(3)	B(3)–B(7)#5	1.789(8)
Na(1A)–S(4)#4	3.155(7)	B(4)–B(4)#1	1.803(13)
Na(1A)–S(6)#5	3.261(6)	B(4)–B(7)#5	1.806(8)
Na(1A)–S(6)#7	3.007(5)	B(4)–B(7)	1.783(8)
Na(1A)–S(6)	3.007(5)	B(5)–B(6)	1.782(8)
Na(1A)–S(6)#4	3.261(6)	B(5)–B(6)#7	1.782(8)
In(1)–S(5)	2.7917(14)	B(5)–B(8)#7	1.796(9)
In(1)–S(5)#11	2.7917(14)	B(5)–B(8)	1.796(9)
In(1)–S(7)	2.7634(15)	B(6)–B(6)#12	1.798(12)
In(1)–S(7)#11	2.7634(15)	B(6)–B(8)	1.764(8)
In(1)–S(8)#11	2.7588(8)	B(6)–B(8)#12	1.805(8)
In(1)–S(8)	2.7588(8)	B(7)–B(7)#5	1.802(12)

In(2)–S(1)	2.474(2)	B(8)–B(8)#7	1.793(14)
In(2)–S(2)	2.4528(19)	B(1)–S(4)	1.849(9)
In(2)–S(5)#1	2.4946(15)	B(1)–S(4)#7	1.849(9)
In(2)–S(5)	2.4947(15)	B(2)–S(2)	1.886(9)
In(3)–S(1)	2.9409(8)	B(3)–S(1)	1.838(9)
In(3)–S(3)	2.4675(15)	B(4)–S(3)	1.863(6)
In(3)–S(4)#7	2.5778(9)	B(5)–S(8)	1.866(9)
In(3)–S(4)	2.5778(9)	B(6)–S(7)	1.852(6)
In(3)–S(6)	2.4596(14)	B(7)–S(6)#5	1.868(6)
In(3)–S(7)	2.4915(14)	B(8)–S(5)#11	1.863(6)

Symmetry codes: #1 $+x, -y, +z$; #2 $-1/2+x, -1/2+y, +z$; #3 $-1/2+x, 1/2-y, +z$; #4 $1-x, 1-y, 1-z$; #5 $1-x, +y, 1-z$; #6 $1/2+x, 1/2+y, +z$; #7 $+x, 1-y, +z$; #8 $-1/2+x, 1/2+y, +z$; #9 $1-x, -y, 1-z$; #10 $1/2-x, 1/2-y, 1-z$; #11 $3/2-x, 1/2-y, 2-z$; #12 $1-x, +y, 2-z$; #13 $1-x, 1-y, 2-z$.

Table S5. The incomplete occupations of Eu and In in literature.

Compounds	Occupancy	Ref.
$\text{Eu}_{1.836}\text{Ta}_{15}\text{O}_{32}$	0.306 for Eu	1
$\text{Eu}_{0.45}\text{NbO}_{3.175}$	0.45 for Eu	2
$\text{Na}_5\text{CaIn}_{1.67}\text{Sb}_4$	0.83 for In	3
$\text{La}_3\text{In}_{0.34}\text{GeS}_7$	0.341 for In	4

Table S6. The In–S distances in literature.

Compounds	$d(\text{In-S})/\text{\AA}$	Ref.
NaIn_3S_5	2.504–2.929	5
$\text{Ba}_{18}\text{F}_{18}\text{In}_8\text{S}_{21}$	2.383–2.941	6
$\text{Fe}_{1.5}\text{Pb}_{5.5}\text{In}_{10}\text{S}_{22}$	2.479–2.952	7
$\text{Ca}_{0.76}\text{In}_{2.84}\text{S}_5$	2.479–2.916	8
$\text{La}_4\text{Ag}_2\text{In}_4\text{S}_{13}$	2.401–2.932	9

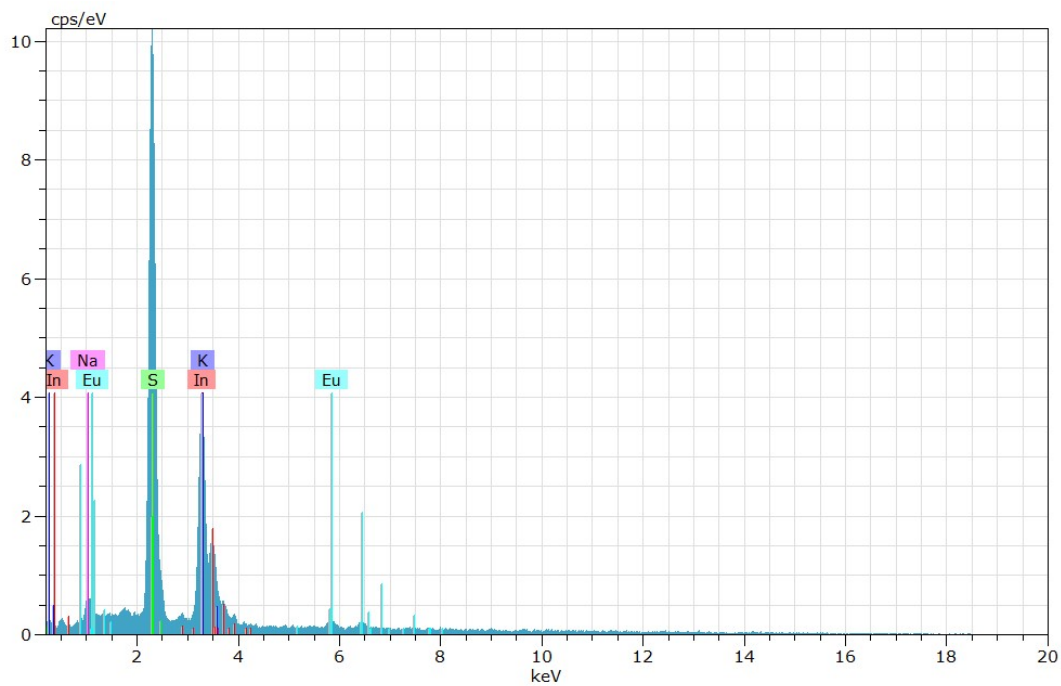


Fig. S1 EDS analysis for **1**.

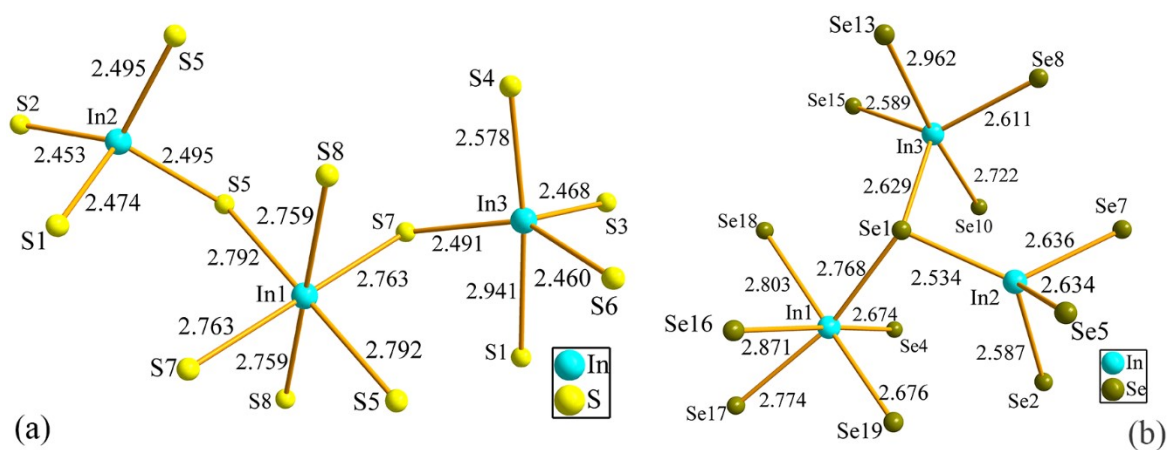


Fig. S2 Diverse InQ_x coordination styles in **1** (a) and $\text{K}_5\text{In}_3\text{P}_6\text{Se}_{19}$ (b).

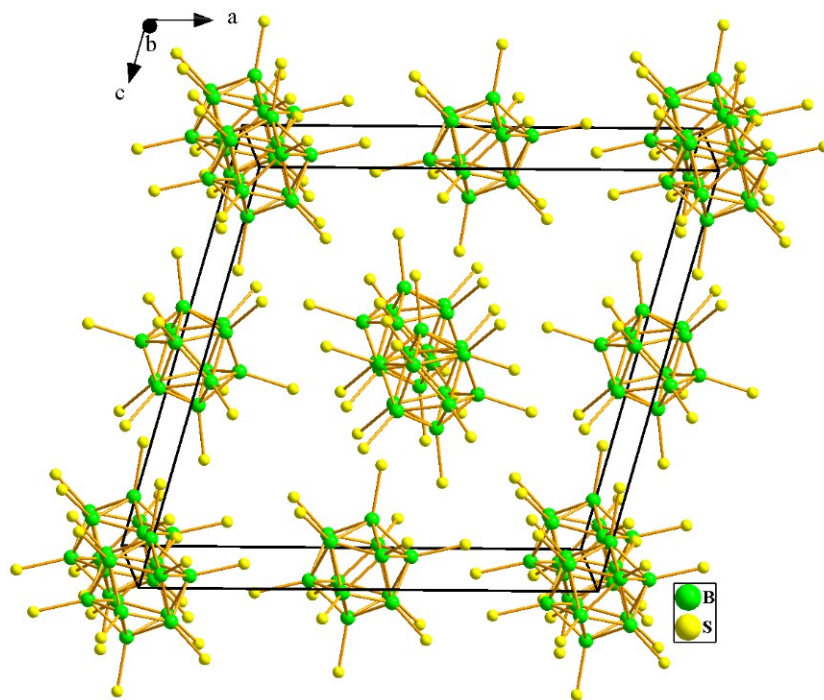


Fig. S3 The unit cell of $\text{KNa}_{0.78}\text{Eu}_{0.27}\text{In}_{3.80}\text{B}_{12}\text{S}_{12}$. For clarity, only those B and S atoms in the above.

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