Electronic Supplementary Information

Ultrathin tungsten-doped hydrogenated titanium dioxide nanosheets for solardriven hydrogen evolution

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Fig. S1. Optical photos of (a) W-h-TiO₂ nanosheets and (b) h-TiO₂ nanosheets.



Fig. S2. (a) XRD pattern and (b) TEM image of W_2C nanoparticles.



Fig. S3. (a, c) AFM images and (b, d) corresponding height profiles of W-h-TiO₂ nanosheets and h-TiO₂ nanosheets.



Fig. S4. Full XPS scan spectrum of W-h-TiO₂ nanosheets.



Fig. S5. W 4d XPS spectrum of W-h-TiO₂ nanosheets.



Fig. S6. C 1s XPS spectrum of W-h-TiO₂ nanosheets.



Fig. S7. Optical bandgap of pure h-TiO₂ and W-h-TiO₂ nanosheets.



Fig. S8. Mott-Schottky plots under different frequency of (a) W-h-TiO₂ nanosheets and (b) pure h-TiO₂ nanosheets.



Fig. S9. UPS spectra of (a) h-TiO₂ nanosheets and (b) W-h-TiO₂ nanosheets.



Fig. S10. H_2 evolution rates of W-h-TiO₂ nanosheets under a constant temperature of 25 °C.



Fig. S11. H₂ evolution rates of W-h-TiO₂ nanosheets with 0.5wt% H₂PtCl₆ as cocatalyst.

Samples	molar ratio of W to Ti
W-h-TiO2-20	0.27%
W-h-TiO2-40	0.58%
W-h-TiO2-60	0.93%
W-h-TiO2-80	1.35%
W-h-TiO2-100	1.77%
W-h-TiO2-120	2.18%
W-h-TiO2-140	2.69%

Table S1. Actual molar ratio of W to Ti in W-h-TiO₂ nanosheets measured by ICP-OES.

Table S2. Time-resolved PL decay curve parameters obtained by double-exponentialfunction simulation.

samples	τ_1 (ns)	τ_2 (ns)	A ₁ (%)	A ₂ (%)	τ _{ave} (ns)
h-TiO ₂	2.71	22.61	85.92	14.08	5.52
W-h-TiO ₂	3.97	58.36	67.62	32.38	21.58

Table S3. Photocatalytic hydrogen production performance of photocatalystsreported in literatures.

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Photocatalysts	H ₂ (mmol/g/h)	Illumination	Sacrificial agent	Ref.
Ru-TiO ₂	0.85	300 W Xe lamp	3wt% Ru and 20 mL	1
		(AM 1.5G)	CH₃OH	
Pt/Ga-TiO ₂	5.122	300 W Xe lamp	0.5wt% H ₂ PtCl ₆ and	2
		(AM 1.5G)	20 mL CH₃OH	
Fe-Ni/Ag/TiO ₂	0.793	300 W Xe lamp	20 mL CH₃OH	3
		(AM 1.5G)		
S-TiO ₂	163.9	300 W Xe lamp	20 mL CH₃OH	4
		(AM 1.5G)		
N-TiO ₂ /Pt	0.57	300 W Xe lamp	20 mL CH₃OH	5
		(AM 1.5G)		
Ni/TiO ₂	3.39	300 W Xe lamp	1wt% H ₂ PtCl ₆ and 20	6
		(AM 1.5G)	mL CH₃OH	
Cu-mpTiO ₂	1.00	300 W Xe lamp	$1wt\% H_2PtCl_6$ and 20	7
		(AM 1.5G)	mL CH₃OH	
1T-WS ₂ /2H-	5.23	300 W Xe lamp	$0.5wt\% H_2PtCl_6$ and	This
WS ₂ /CdS		(AM 1.5G)	20 mL CH₃OH	work

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