

Supporting Information

Size exclusion propyne/propylene separation in a ultramicroporous yet hydrophobic metal-organic framework

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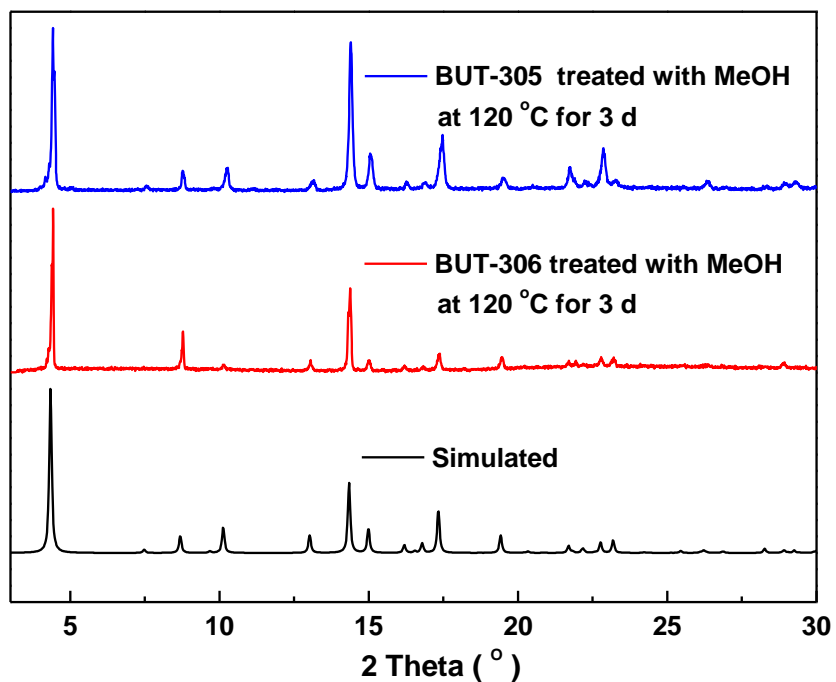


Fig. S1 The PXRD patterns of **BUT-305** and **-306** treated with MeOH at 120 °C for 3 days.

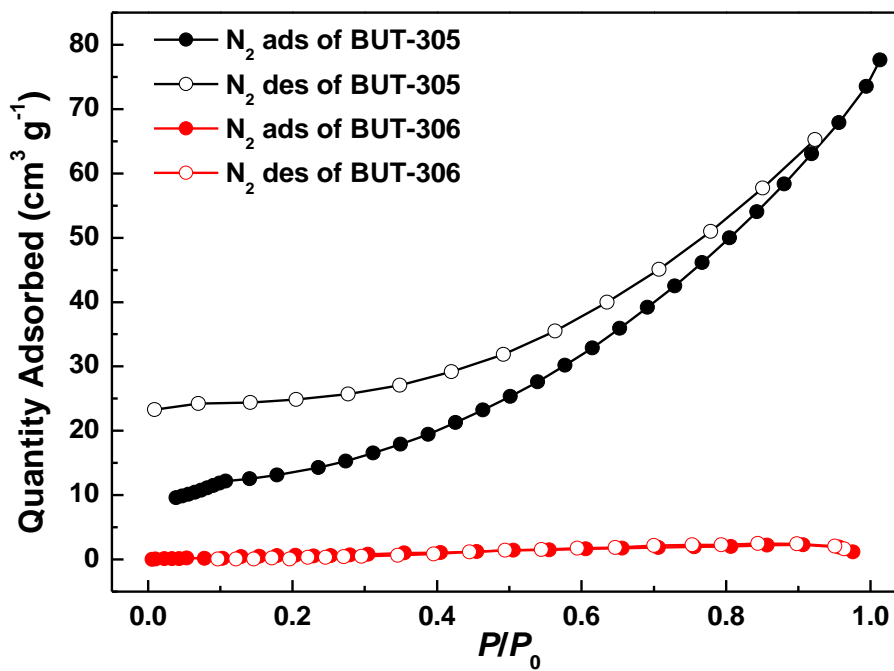


Fig. S2 N_2 adsorption isotherms of **BUT-305** and **-306** at 77 K.

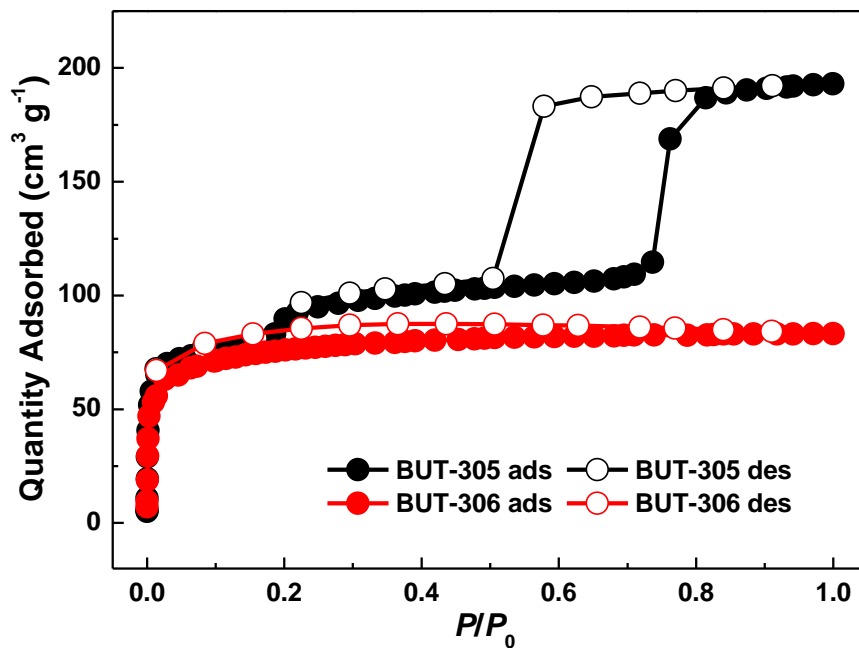


Fig. S3 The CO₂ adsorption isotherms of **BUT-305** and **-306** at 195 K.

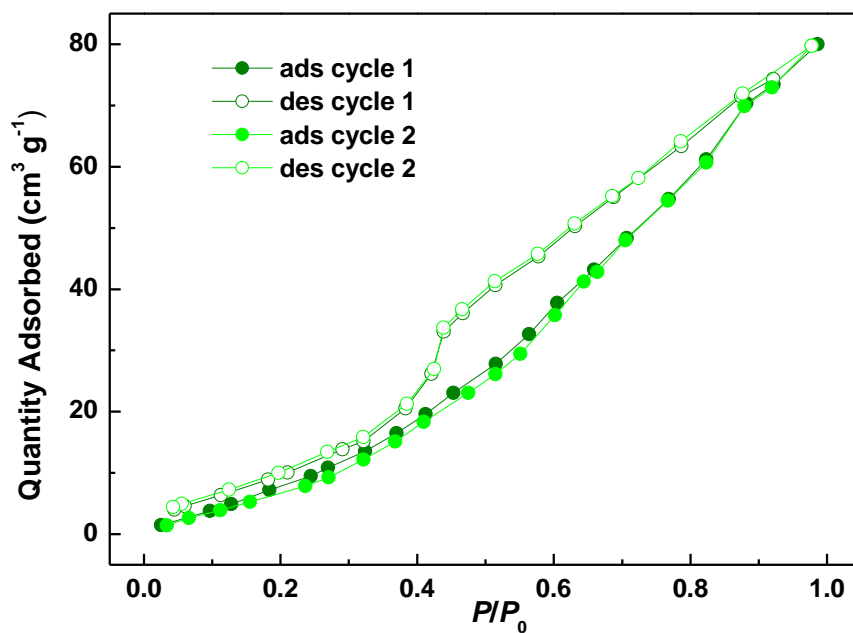


Fig. S4 Two repeated measurements of H₂O adsorption isotherms of **BUT-306** at 298 K.

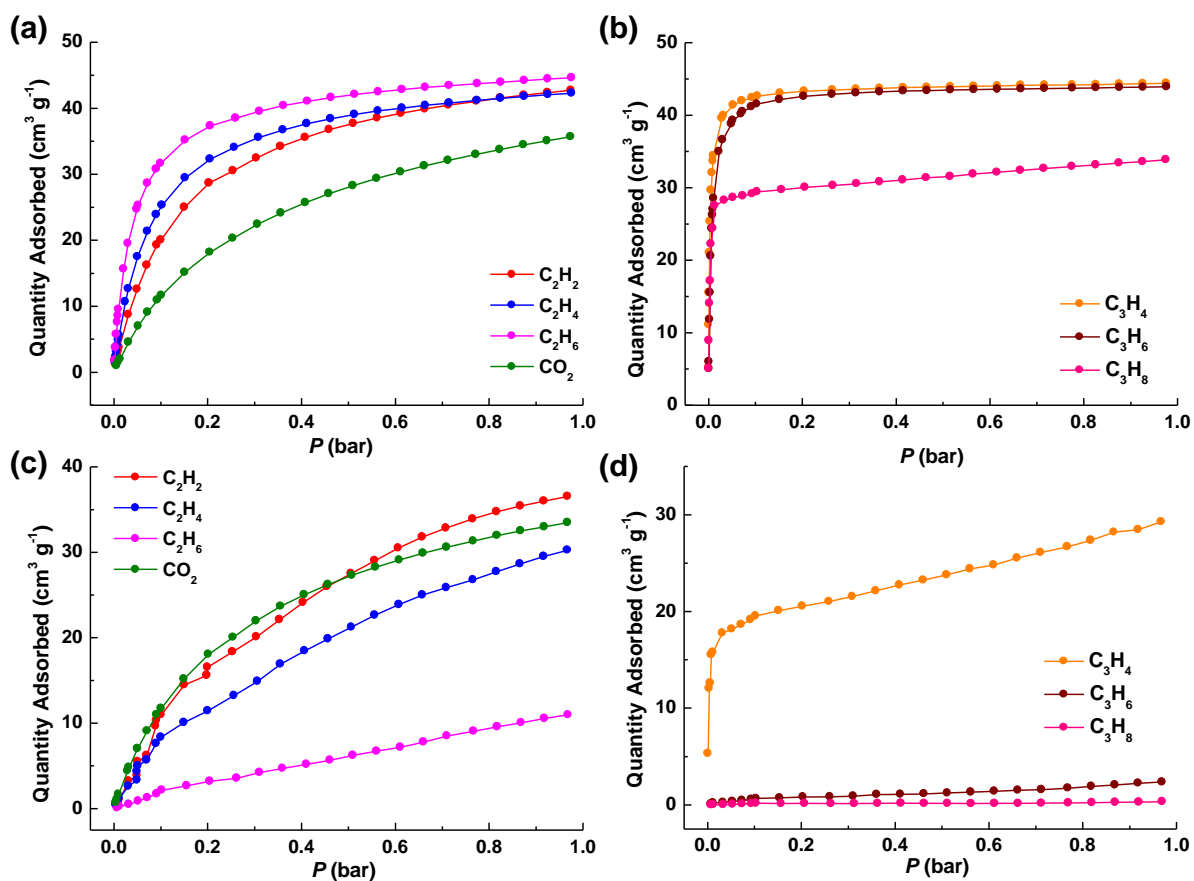


Fig. S5 Single-component gas adsorption isotherms (CO_2 , C_2H_2 , C_2H_4 , C_2H_6 , C_3H_4 , C_3H_6 , and C_3H_8) of (a, b) **BUT-305** and (c, d) **BUT-306** recorded at 298 K.

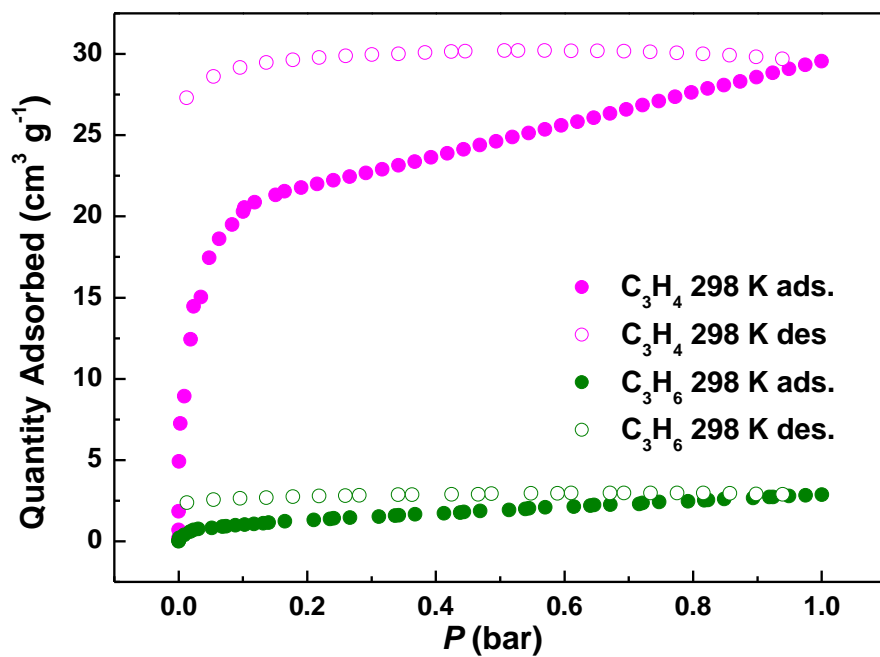


Fig. S6 C_3H_4 , C_3H_6 adsorption and desorption isotherms of **BUT-306** measured at 298 K.

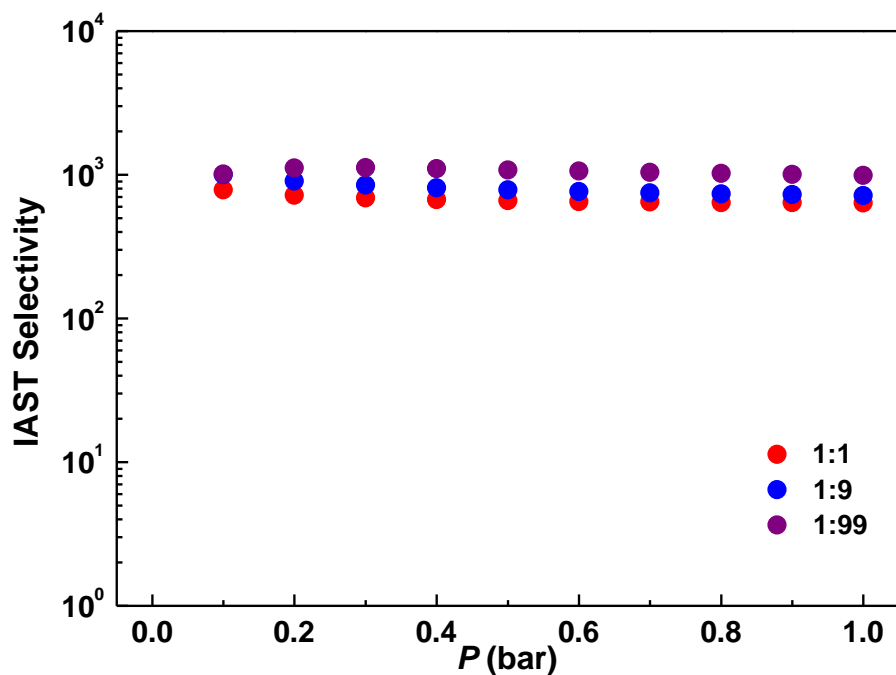


Fig. S7 IAST selectivities of **BUT-306** for 1:1, 1:9 and 1:99 (v/v) C_3H_4/C_3H_6 gas mixtures at 298 K calculated from single-component gas adsorption isotherms.

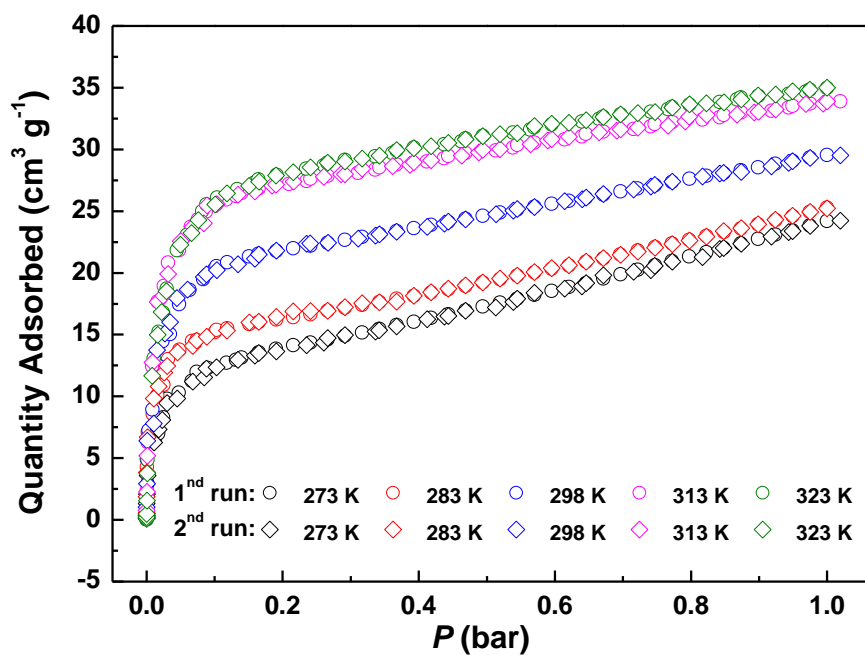


Fig. S8 The C_3H_4 adsorption isotherms of **BUT-306** recorded at 273, 283, 298, 313, and 323 K, respectively. The measurements were repeated to verify the data.

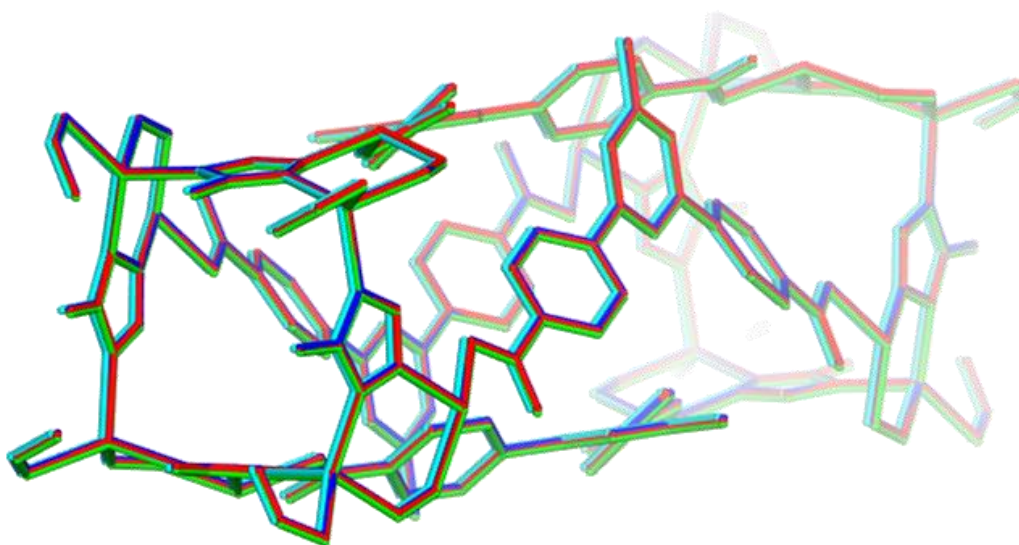


Fig. S9 Overlapping of the framework structures **BUT-306** recorded at 273 K (blue), 298 K (red), and 313 K (green) and that of C₃H₄-loaded **BUT-306** recorded at 200 K (turquoise).

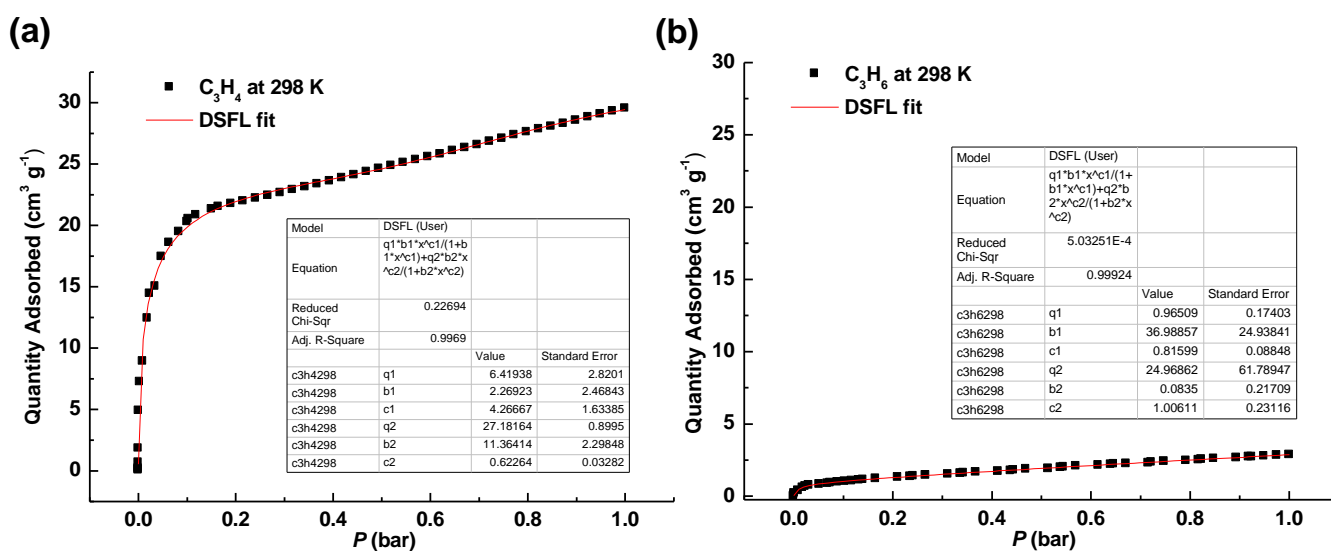


Fig. S10 C₃H₄ and C₃H₆ adsorption isotherms of **BUT-306** measured at 298 K and its double-site Langmuir-Freundlich fits.

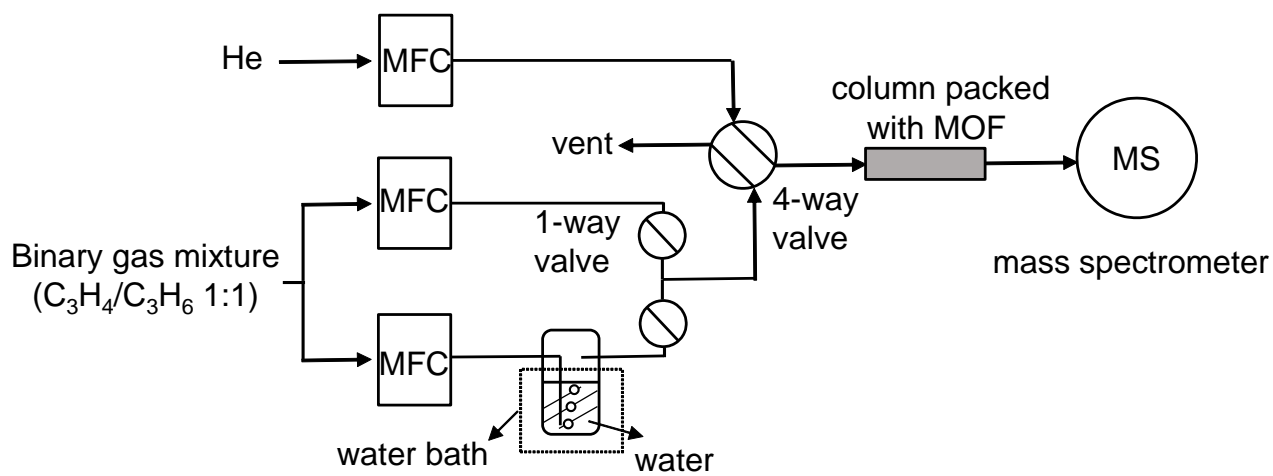


Fig. S11 Schematic diagram of column breakthrough experiment device.

Table S1 Crystal data and structure refinement parameters for **BUT-306** at 273, 298 and 313 K, and C₃H₄-loaded **BUT-306** at 200 K.

	BUT-306	BUT-306	BUT-306	C ₃ H ₄ -loaded BUT-306
Temp. (K)	273	298	313	200
Formula	C _{12.5} H ₁₀ N ₄ O ₂ Zn	C _{12.5} H ₁₀ N ₄ O ₂ Zn	C _{12.5} H ₁₀ N ₄ O ₂ Zn	C _{12.73} H _{10.2} N ₄ O ₂ Zn
Crystal system	tetragonal	tetragonal	tetragonal	tetragonal
Space group	<i>I4/m</i>	<i>I4/m</i>	<i>I4/m</i>	<i>I4/m</i>
<i>a</i> (Å)	12.3290(2)	12.3420(2)	12.3425(2)	12.3020(10)
<i>c</i> (Å)	40.6790(6)	40.6826(9)	40.6799(5)	40.7025(5)
<i>α</i> (°)	90	90	90	90
<i>V</i> (Å ³)	6183.4(2)	6197.0(2)	6197.1(2)	6159.9(13)
<i>Z</i>	16	16	16	16
<i>D_c</i> (g cm ⁻³)	1.348	1.345	1.345	1.365
<i>F</i> (000)	2544.0	2544.0	2544.0	2569.0
Reflections collected	11595	11499	11532	11249
Unique reflections	2364	2343	2326	3109
GOF	1.031	1.031	1.020	1.060
<i>R</i> ₁ ^a [<i>I</i> > 2σ(<i>I</i>)]	0.0424	0.0588	0.0425	0.0371
w <i>R</i> ₂ ^b (all data)	0.1243	0.1192	0.1263	0.1048
CCDC deposition number	2168051	2168052	2168053	2168054

^a $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$. ^b $wR_2 = |\sum w(|F_o|^2 - |F_c|^2)| / \sum |w(F_o)^2|^{1/2}$, where $w = 1/[\sigma^2(F_o^2) + (aP)^2 + bP]$. $P = (F_o^2 + 2F_c^2)/3$

Table S2 Pawley refinement parameters for the PXRD pattern of **BUT-305**.

Formula	Zn ₂ (ATZ) ₂ (TPDC)
crystal system	tetragonal
space group	<i>I4/m</i>
<i>a</i> / Å	12.38096(0.00647)
<i>c</i> / Å	40.65355(0.02070)
<i>R</i> _{wp} / %	8.62
<i>R</i> _p / %	6.05
<i>U</i>	0.34090(0.18549)
<i>V</i>	-0.14615(0.07521)
<i>W</i>	0.02189(0.01621)
<i>NA</i>	0.28626(0.01743)
<i>NB</i>	0.01061(0.00102)
Zero Point / °	0.37865(0.00167)
Shift #1	-0.48957(0.00097)
Shift #2	0.10628(0.03413)
<i>P1</i>	-0.20988(0.00230)
<i>P2</i>	-0.11320(0.00109)
<i>P3</i>	0.21332(0.00475)
<i>P4</i>	0.18014(0.00218)

Table S3 Comparison of C₃H₄ and C₃H₆ uptake (mmol/g, mmol/cm³ and cm³/cm³) from gas adsorption isotherms at various pressures, and the C₃H₄/C₃H₆ uptake ratio at 0.01/0.01 bar and 0.01/0.99 bar for various reported MOFs at 298 K.

MOFs	C ₃ H ₄ adsorption			C ₃ H ₆ adsorption			C ₃ H ₄ /C ₃ H ₆		Ref.
	mmol g ⁻¹			mmol g ⁻¹			adsorption ratio		
	0.01 bar	0.1 bar	0.99 bar	0.01 bar	0.1 bar	0.99 bar	0.99/0.99	0.01/0.99	
UTSA-200	2.99	3.3	3.58	0.02	0.33	1.2	2.98	2.49	
SIFSIX-3-Ni	2.56	2.73	2.85	0.15	2.42	2.72	1.05	0.98	
ZU-62	2.19	3.01	3.64	0.17	2.26	2.67	1.36	0.82	
SIFSIX-2-Cu-i	1.96	3.22	3.77	0.15	2.19	2.63	1.43	0.74	
SIFSIX-1-Cu	2.57	6.79	8.63	0.3	4.92	5.88	1.47	0.44	
ELM-12	1.82	2.54	2.77	0.19	1.11	1.43	1.94	1.27	
ZJUT-1	0.4	1.07	2.24	0.06	0.54	0.84	2.67	0.47	Ref. [32] in maintext
Mg-MOF-74	2.71	7.32	9.4	2.03	6.07	6.49	1.33	0.42	
Co-MOF-74	2.85	5.8	7.47	1.27	5.64	5.95	1.26	0.48	
Ni-MOF-74	1.8	4.68	5.51	1.53	4.5	4.78	1.15	0.38	
Fe-MOF-74	2.18	6.67	7.94	1.67	6.25	6.63	1.20	0.33	
Cu-BTC	1.47	8.17	10.47	1.36	7.9	8.33	1.26	0.18	
Fe-BTT	1.74	6.87	12.41	1.52	8.03	8.74	1.42	0.2	

Cr-BTT	1.08	4.42	7.28	0.76	5.04	5.85	1.24	0.18	
MIL-100(Cr)	1.52	4.98	14.51	0.63	4.53	6.25	2.32	0.24	
MIL-100(Fe)	1.34	4.74	17.14	0.52	4.52	6.96	2.46	0.19	
UIO-66	1.58	5.29	10.23	0.36	2.7	3.33	3.07	0.47	
ZIF-8	0.13	1.44	6.27	0.08	3.15	4.07	1.54	0.03	
UTSA-74-Zn	0.33	2.13	7.43	0.09	2.47	5.07	1.45	0.06	
UTSA-100	1.87	4	5.35	1.34	2.68	3.04	1.76	0.62	
Co-gallate	1.61	2.4	3.21	0.09	0.5	1.49	2.15	1.08	
Mg-gallate	1.65	2.7	3.75	0.08	1.0	1.50	2.50	1.10	Ref. [51] in maintext
Ni-gallate	1.1	1.9	2.65	0.18	0.4	0.9	2.94	1.22	
NKMOF-1	1.78	2.10	2.90	0.65	0.75	1.50	1.93	1.19	Ref. [52] in maintext
BUT-306	0.44	0.89	1.29	0.02	0.04	0.12	10.75	3.67	this work
