

SUPPORTING INFORMATION

Interface engineering of hierarchical P-doped NiSe/2H-MoSe₂ nanorod arrays for efficient hydrogen evolution

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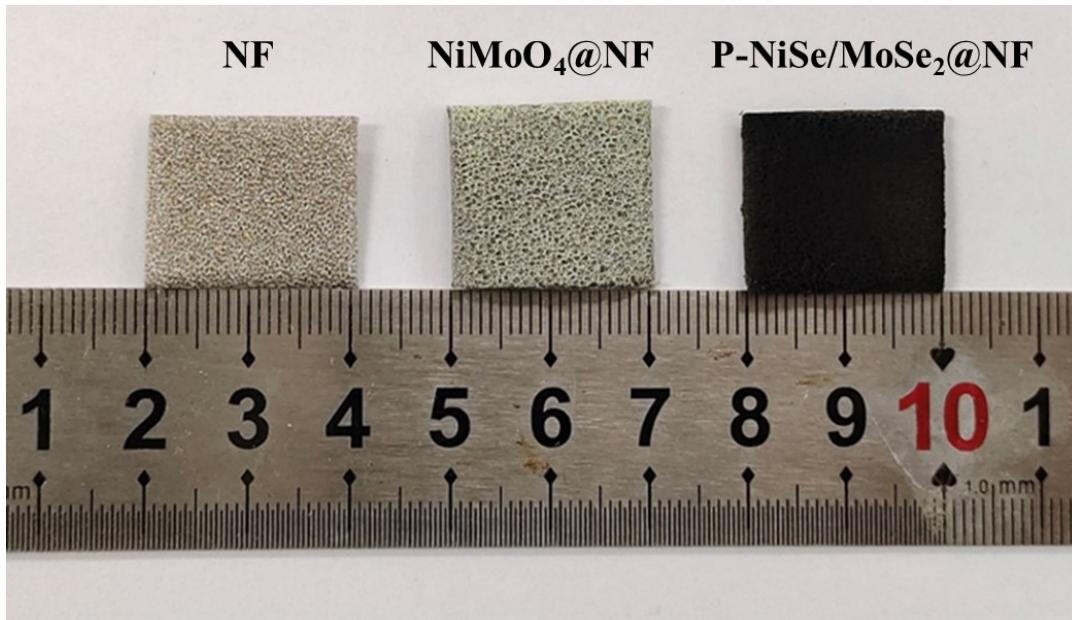


Fig. S1. Digital image of NF, NiMoO₄@NF, and P-NiSe/MoSe₂@NF catalysts.

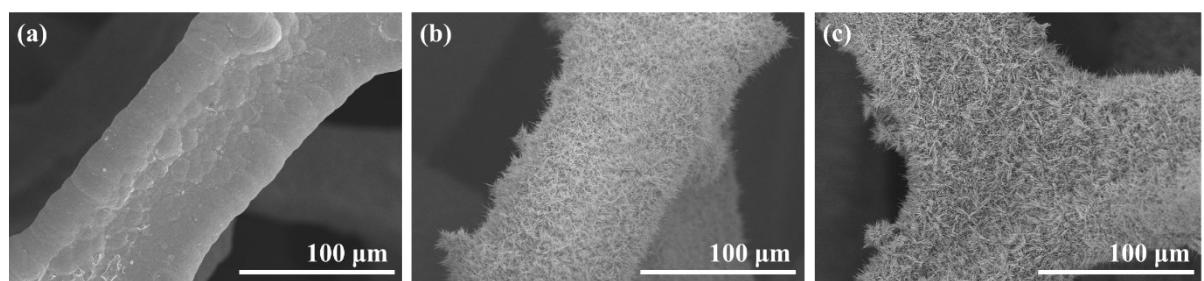


Fig. S2. SEM images of (a) NF, (b) NiMoO₄@NF, and (c) P-NiSe/MoSe₂@NF.

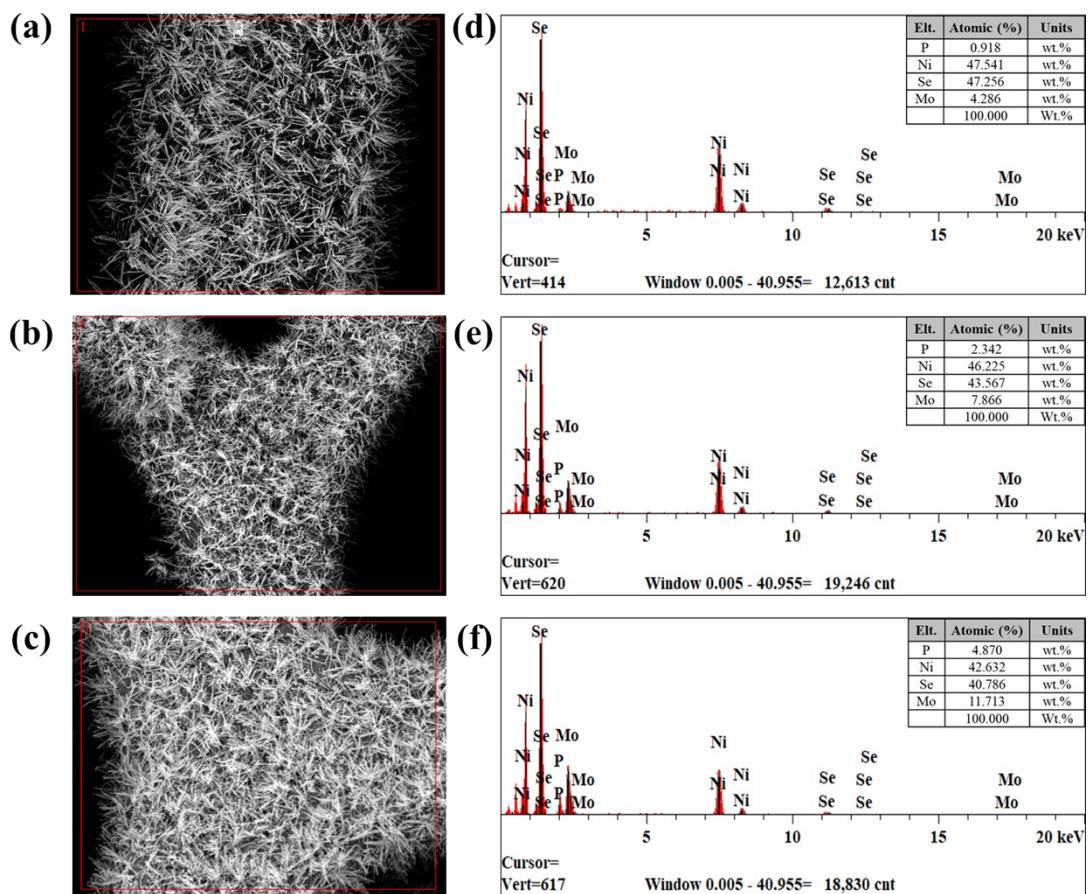


Fig. S3. SEM images and EDS analysis of (a, d) P-NiSe/MoSe₂@NF-1, (b, e)P-NiSe/MoSe₂@NF-2 and (c, f)P-NiSe/MoSe₂@NF-3.

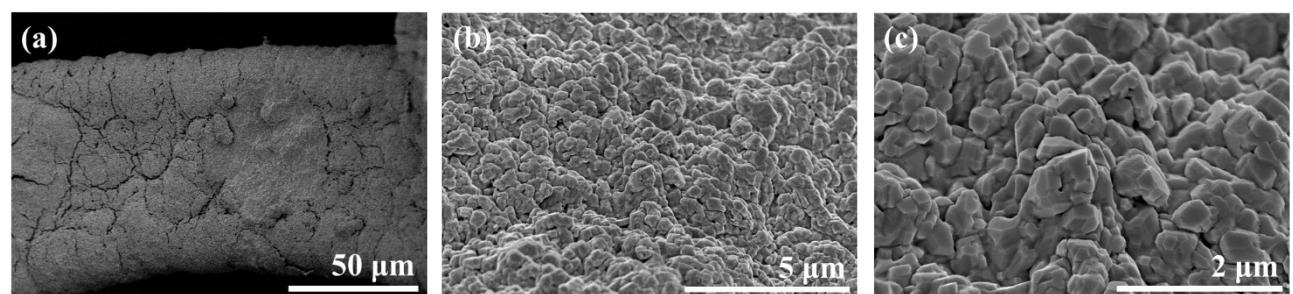


Fig. S4. SEM images of $\text{Ni}_x\text{Se}_y@\text{NF}$ (a-c).

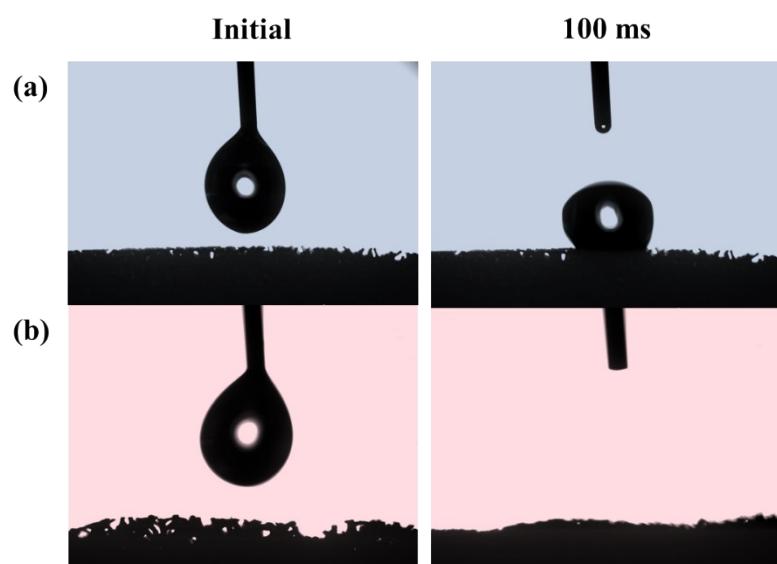


Fig. S5. Wettability test of (a) NF and (b) P-NiSe/MoSe₂@NF electrode.

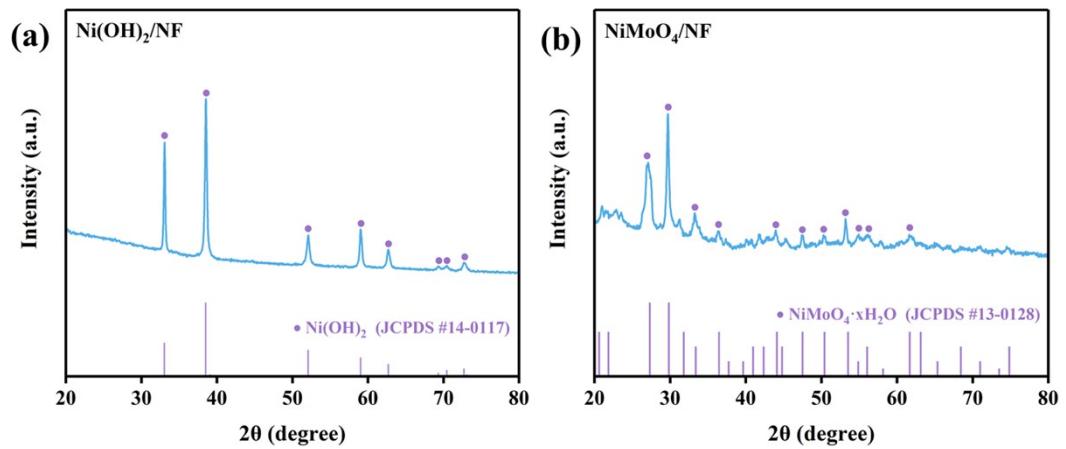


Fig. S6. The typical XRD pattern of (a) $\text{Ni}(\text{OH})_2@\text{NF}$ and (b) $\text{NiMoO}_4@\text{NF}$ precursor.

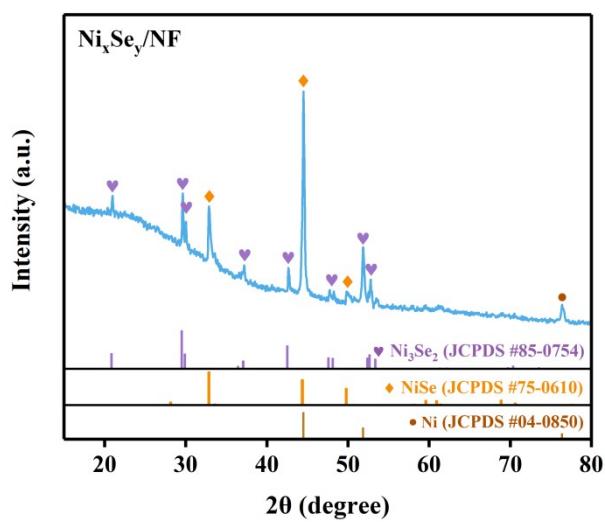


Fig. S7. XRD pattern of $\text{Ni}_x\text{Se}_y@\text{NF}$.

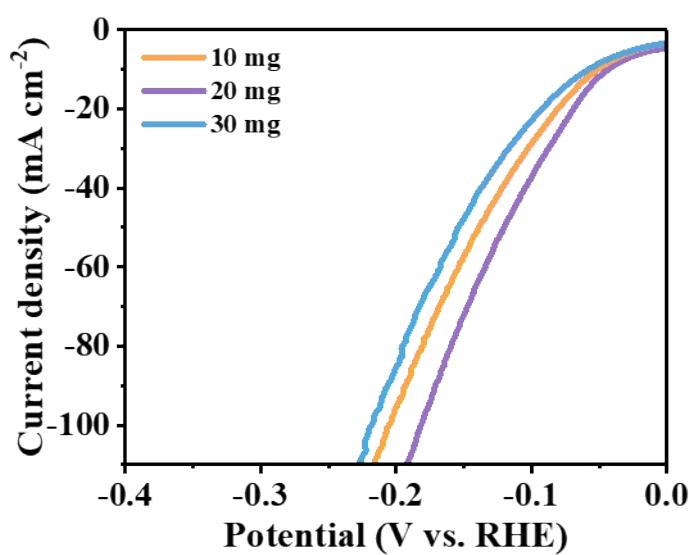


Fig. S8. LSV curves of samples with different doping amounts.

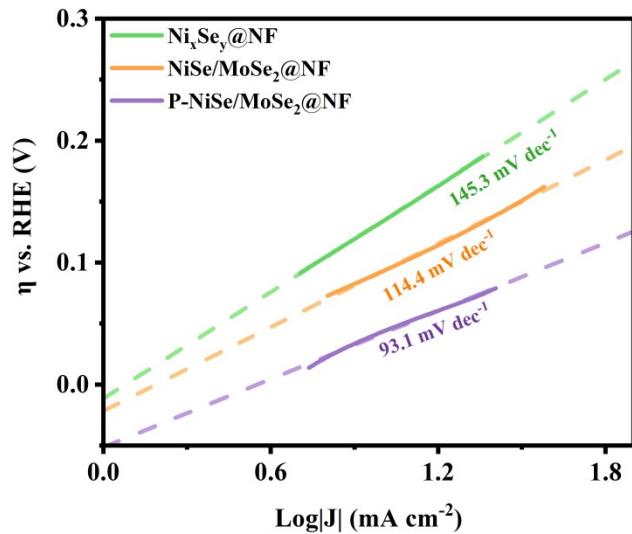


Fig. S9. Tafel plots of $\text{Ni}_x\text{Se}_y@\text{NF}$, $\text{NiSe}/\text{MoSe}_2@\text{NF}$, and $\text{P-NiSe}/\text{MoSe}_2@\text{NF}$ catalysts to calculate the exchange current density.

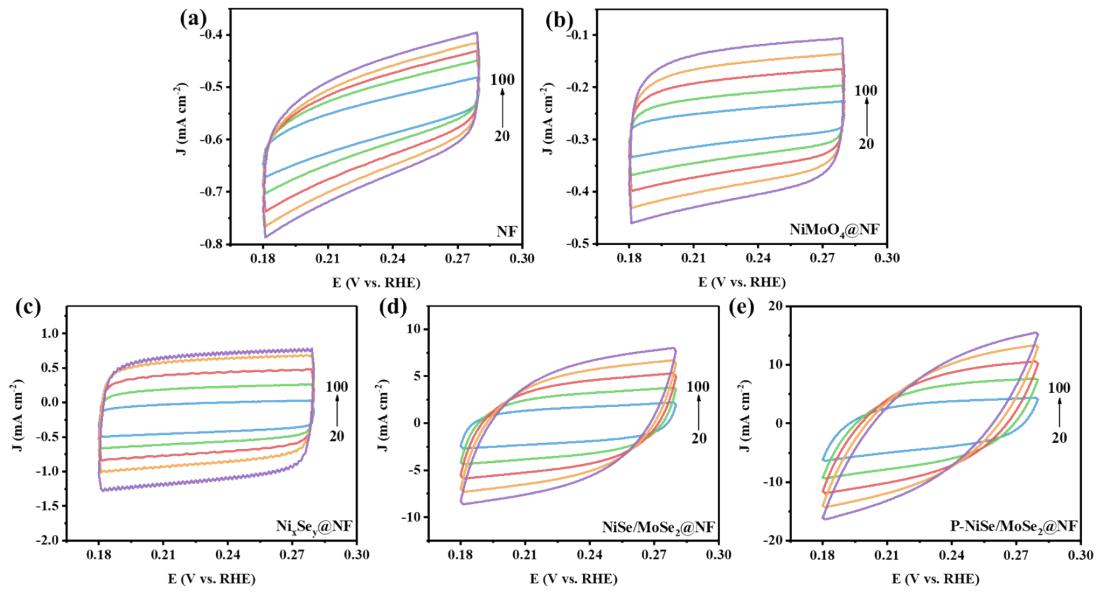


Fig. S10. CV curves within a non-faradaic reaction region of $0.18 \sim 0.28$ V versus RHE at different scan rates for (a) NF (b) $\text{NiMoO}_4@\text{NF}$, (c) $\text{Ni}_x\text{Se}_y@\text{NF}$, (d) $\text{NiSe}/\text{MoSe}_2@\text{NF}$, and (e) P- $\text{NiSe}/\text{MoSe}_2@\text{NF}$.

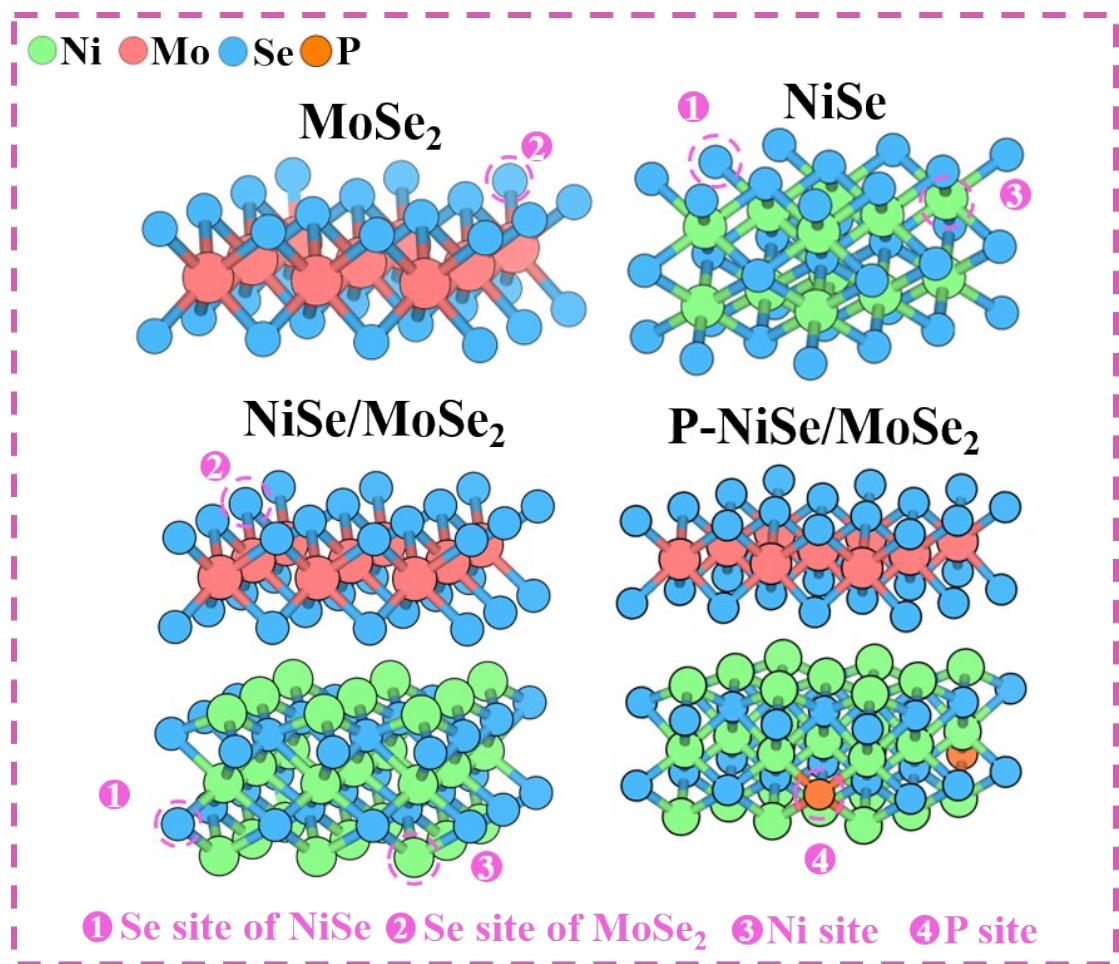


Fig. S11. Optimized geometric models of MoSe₂, NiSe, NiSe/MoSe₂ and P-NiSe/MoSe₂ with possible H* adsorption sites.

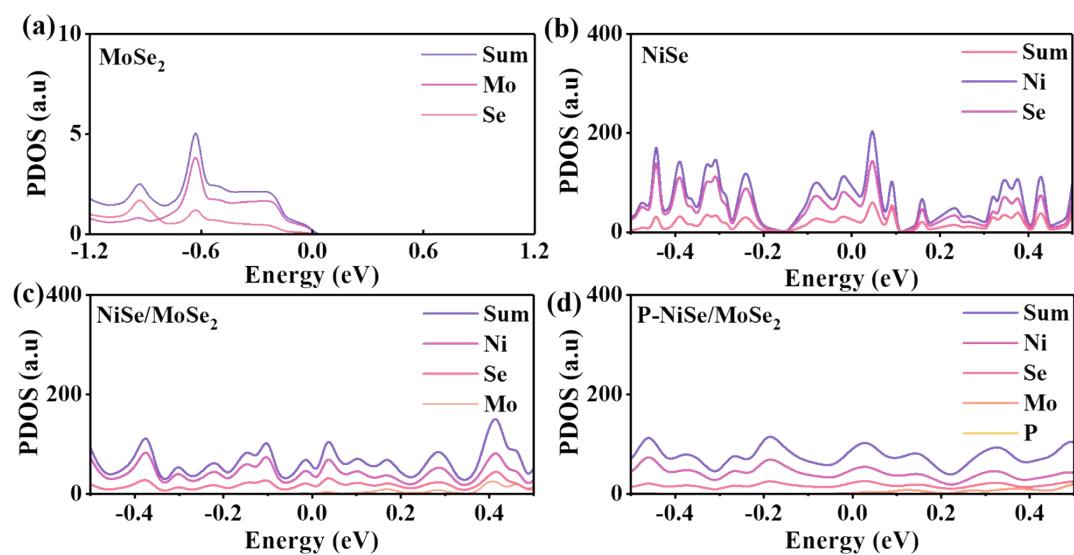


Fig. S12. Projected electronic density of states (PDOS) of (a) MoSe₂, (b) NiSe, (c) NiSe/MoSe₂ and (d) P-NiSe/MoSe₂.

Table S1. Amount of NaH₂PO₂ used and phosphorus doping for different samples.

| Sample | Dosage of NaH₂PO₂ (mg) | P atomic ratio (%) |
|----------------|---|---------------------------|
| P-NiSe/MoSe2-1 | 10 mg | 0.918 |
| P-NiSe/MoSe2-2 | 20 mg | 2.342 |
| P-NiSe/MoSe2-3 | 30 mg | 4.870 |

Table S2. Comparison of HER catalytic performance of recently reported catalysts.

| Catalyst | Electrolyte | η_{10} vs. RHE (mV) | η_{100} vs. RHE (mV) | Ref |
|--|--------------------------------------|-----------------------------|------------------------------------|------------------|
| P-NiSe/MoSe₂@NF | 1.0 M KOH | 43 | 182 | This work |
| NiMoSe/NF-2 | 1.0 M KOH | 91 | 223 | 1 |
| NiMoO ₄ @MoSe ₂ /Ni _x Se _y /NF | 1.0 M KOH | 69 | / | 2 |
| P-NiSe ₂ /MoSe ₂ @CC | 1.0 M KOH | 175 | / | 3 |
| | 0.5 M H ₂ SO ₄ | 93 | / | |
| 1T-MoSe ₂ /NiSe | 1.0 M KOH | 120 | 200 mV @ 50 mA cm ⁻² | 4 |
| MoSe ₂ @Ni _{0.85} Se | 1.0 M KOH | 117 | 204 | 5 |
| | 0.5 M H ₂ SO ₄ | 154 | / | |
| MoSe ₂ -NiSe@C | 1.0 M KOH | 180 | / | 6 |
| MoSe ₂ /NiSe ₂ @CFP | 0.5 M H ₂ SO ₄ | 193 | 249 | 7 |
| NiMoSe | 0.5 H ₂ SO ₄ | 197 | / | 8 |
| SC-NiSe/MoSe ₂ @CFP | 1.0 M KOH | 71 | 129 | 9 |
| Mo, S-codoped NiSe/NF | 1.0 M KOH | 88 | / | 10 |
| NiMoSe@Ni _{0.33} Mo _{0.67} O ₄ | 1.0 M KOH | 82.1 | / | 11 |
| MoSe ₂ /NiSe-2 | 1.0 M KOH | 87 | / | 12 |

Table S3. The calculated adsorption energies (eV) of the hydrogen atom.

| | Ni site | Se site | P site |
|---|---------|---------|--------|
| MoSe ₂ | -- | 2.10 | -- |
| NiSe | 0.23 | 0.22 | -- |
| NiSe/MoSe ₂ -NiSe | 0.88 | 0.17 | -- |
| NiSe/MoSe ₂ -MoSe ₂ | -- | 1.60 | -- |
| P-NiSe/MoSe ₂ | -- | -- | 0.16 |

Reference

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