

Supporting information

Promotion of methane storage capacity with metal-organic frameworks of high porosity

Xin Zhang,^{a,b} Rui-Biao Lin,^b Zeid A. Alothman,^c Osamah Alduhaish,^c Taner Yildirim,^d Wei Zhou,^d Jian-Rong Li,^{*a} and Banglin Chen^{*b}

- Beijing Key Laboratory for Green Catalysis and Separation and Department of Environmental Chemical Engineering, Faculty of Environment and Life, Beijing University of Technology, Beijing 100124, P. R. China. E-mail: jrli@bjut.edu.cn.
- Department of Chemistry, University of Texas at San Antonio, One UTSA Circle, San Antonio, Texas 78249-0698, USA, E-mail: banglin.chen@utsa.edu
- Chemistry Department, College of Science, King Saud University, Riyadh 11451, Saudi Arabia.
- NIST Center for Neutron Research, Gaithersburg, Maryland 20899-6102, USA.

Additional figures and tables

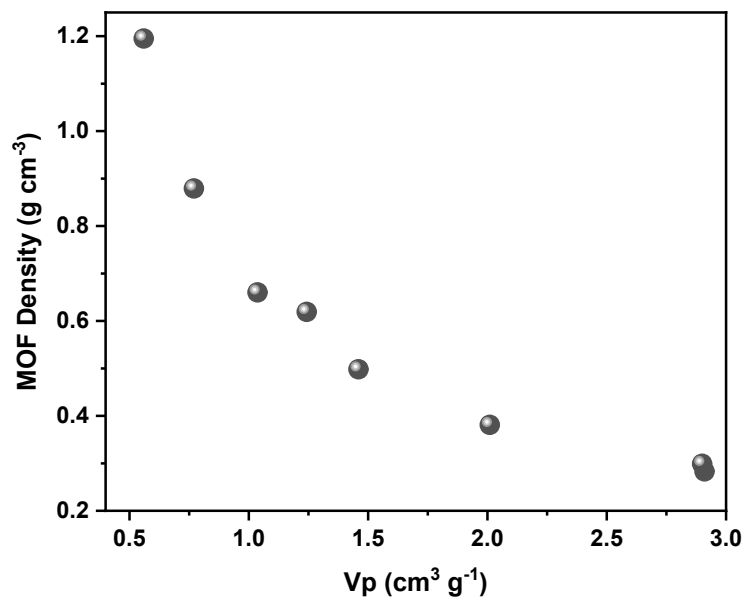


Figure S1. MOF density vs pore volume.

Table S1. Total adsorption and working capacity at 298 K and 273 K of MOFs discussed

MOF	Crystal density	Pore volume	Total adsorption 100 bar 298 K		Total adsorption 100 bar 270 K		Working capacity 5-100 bar 298 K		Working capacity 5-100 bar 270 K		Qst	Ref.
	g/cm ³		cm ³ /g	cm ³ cm ⁻³	g g ⁻¹	cm ³ cm ⁻³	g g ⁻¹	cm ³ cm ⁻³	g g ⁻¹	cm ³ cm ⁻³		
Ni-MOF-74	1.195	0.56	281	0.17	293	0.18	162	0.1	135	0.08	21.4	1
HKUST-1	0.881	0.77	277	0.23	306	0.25	207	0.17	195	0.16	17	1
ZJU-105a	0.66	1.037	258	0.28	287	0.31	203	0.22	207	0.22	14.2	2
PCN-46a	0.619	1.243	260	0.30	299	0.35	217	0.25	235	0.27	16.6	2
NU-1500-Al	0.498	1.46	237	0.34	273	0.39	202	0.29	224	0.32	13.7	3
MFU-4l-Li	0.479	1.66	248	0.37	294	0.44	220	0.30	251	0.38	13.3	4
BUT-20	0.381	2.01	224	0.42	260	0.49	197	0.37	224	0.42	12	5
NPF-200	0.389	2.17	227	0.42	275	0.51	207	0.38	248	0.46	10.7	This work
NU-1501-Fe	0.299	2.9	218	0.52	264	0.63	201	0.48	239	0.57	9.6	3
NU-1501-Al	0.283	2.91	214	0.54	262	0.66	198	0.5	238	0.60	9.7	3

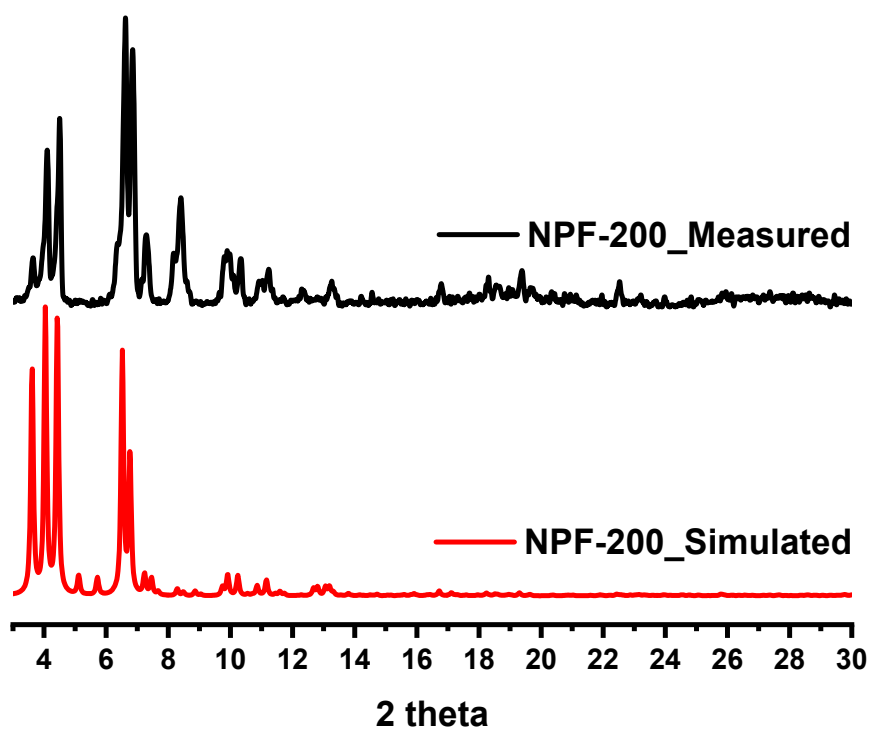


Figure S2. PXRD of NPF-200 and comparison with that simulated from single crystal structure.

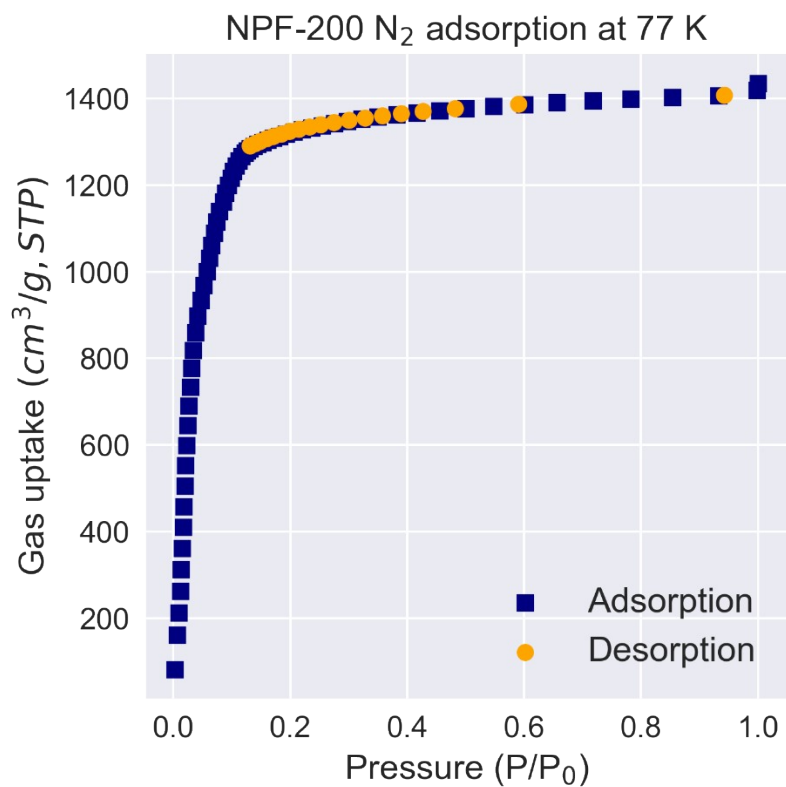


Figure S3. N_2 adsorption isotherm of NPF-200 at 77 K.

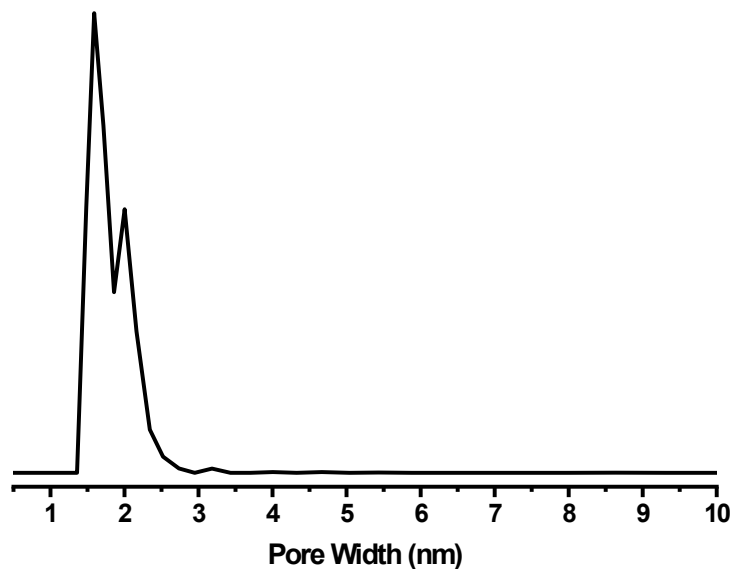


Figure S4. Pore size distribution determined from N₂ uptake at 77 K.

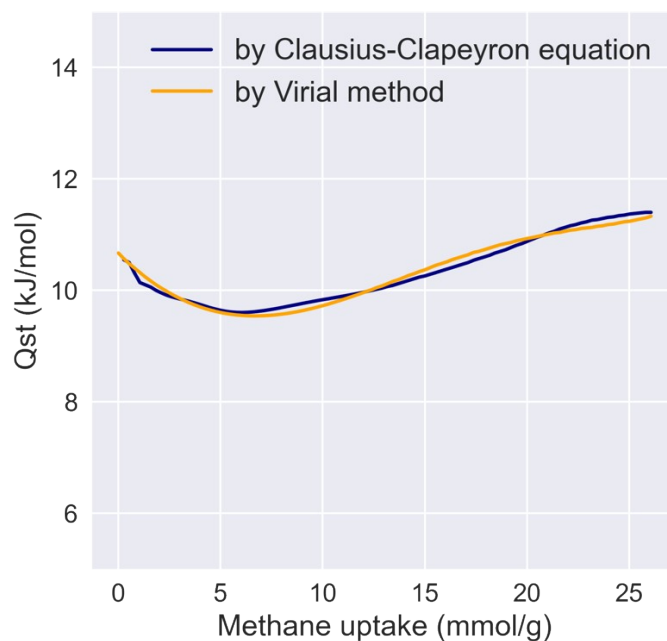


Figure S5. Qst of methane adsorption in NPF-200 calculated using virial method.

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References

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