## Molecular Rotation induced giant, anisotropic negative thermal expansion in a

## hydrogen-bonded coordination framework

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## 1. Materials synthesis, characterization and measurement

**1.1. Materials.** Melamine ( $\geq$  99%, Sigma-Aldrich), Cyanuric acid (( $\geq$  98%, Sigma-Aldrich), Zinc nitrate ( $\geq$  99.5%, Sigma-Aldrich), and ammonia solution (30% in water, Chem-Supply) were used as received without further purification. Ultrapure water was used in all experiments (18.2 M $\Omega$ ·cm, Synergy UV Water Purification System, Merck Millipore). The solutions of melamine (40 mmol/L) and cyanuric acid (40 mmol/L) were prepared under 60 °C. The solutions of Zinc nitrate (100 mmol/L) and aqueous ammonia (10 wt%) were prepared under ambient environment in the laboratory.

**1.2. Preparation of CA-M-Zn crystals.** 4.5 mL cyanuric acid solution (40 mM) were mixed with 2 mL  $Zn(NO_3)_2$  solution (100 mM) in a 15 mL glass vial, followed by the addition of 3 mL melamine solution (40 mM). The vial was capped and put in a dark closet at room temperature. Transparent crystals started to form within 1 day. The addition of melamine could be in solution or solids, and the initial molar ratio of CA to M could also be random. The outcome of the crystallization is exclusive.

**1.3.** Powder X-ray diffraction. The crystals were first washed by de-ionized water, then rinsed by ethanol at room temperature for quick drying and grinded into powders. The powder X-ray diffraction (PXRD) patterns were obtained on a Rigaku Smartlab with oxford heating-cooling accessory using Cu K $\alpha$  radiation ( $\lambda$  = 1.5406 Å) working at 40 mA and 40 kV with a resolution of 0.01° (20). Cyclic heating/cooling PXRD for lattice refinement were also

performed in the same instrument by sealing crystallites into a quartz capillary, with equilibrium time of 15 minutes at each temperature.

**1.4. PXRD pattern refinement method.** The calculation of PXRD patterns from crystal structures and the comparison with the PXRD patterns were conducted in DIFFRAC.TOPAS v6. Cell parameters from in-situ PXRD patterns were obtained by sequential refinements in DIFFRAC.TOPAS v6 with a Macro written by the third author.

**1.5. Single crystal X-ray structure determination.** A single crystal of CAM-Zn was coated in Paratone N oil and mounted on a nylon loop for collection. SCXRD data for structural determination were collected on Xcalibur, Sapphire3, Gemini ultra ( $\lambda = 1.54184$  Å). And in situ SCXRD data were collected on XtaLAB Synergy, Dualflex, Pilatus 300K ( $\lambda = 0.71073$  Å). All data were processed in CrysAlis Pro<sup>1</sup>.Structure solutions were obtained using SHELXT<sup>2</sup> and refined using SHELXL<sup>3</sup> implemented within the Olex2 graphical interfacel<sup>4</sup>. All full-occupancy non-hydrogen models were refined anisotropically, while hydrogen atoms were placed in idealised positions and refined using a riding model on appropriate atoms.

Identification code	CA-M-Zn
Empirical formula	$C_{15}H_{42.34}N_{25}O_{15.66}Zn_2$
Formula weight	954.45
Temperature/K	173.00(14)
Crystal system	monoclinic
Space group	P2/c
a/Å	9.5713(2)
b/Å	6.93400(10)
c/Å	29.0393(5)
α/°	90
β/°	93.711(2)
γ/°	90
Volume/Å <sup>3</sup>	1923.22(6)
Z	2
ρ <sub>calc</sub> g/cm <sup>3</sup>	1.648
µ/mm <sup>-1</sup>	2.396
F(000)	985.0
Crystal size/mm <sup>3</sup>	0.2 × 0.2 × 0.1
Radiation	Cu Kα (λ = 1.54184)
2O range for data collection/°	10.76 to 128.346
Index ranges	-11 ≤ h ≤ 11, -7 ≤ k ≤ 8, -32 ≤ l ≤ 33
Reflections collected	18723
Independent reflections	3177 R <sub>int</sub> = 0.0275, R <sub>sigma</sub> = 0.0175
Data/restraints/parameters	3177/2/307
Goodness-of-fit on F <sup>2</sup>	1.122
Final R indexes I>=2σ (I)	R <sub>1</sub> = 0.0337, wR <sub>2</sub> = 0.1012
Final R indexes all data	R <sub>1</sub> = 0.0347, wR <sub>2</sub> = 0.1022
Largest diff. peak/hole / e Å-3	1.15/-0.30

 Table S1. Crystal data and structure refinement for CA-M-Zn.

Temperature/K	150.00(10)	200.02(10)	250.01(10)	280.01(10)	290.00(10)
Crystal system	monoclinic	monoclinic	monoclinic	monoclinic	monoclinic
Space group	P2/c	P2/c	P2/c	P2/c	P2/c
a/Å	9.5910(3)	9.6009(3)	9.6144(3)	9.6115(4)	9.6061(4)
b/Å	6.9388(2)	6.9496(2)	6.9595(2)	6.9612(3)	6.9546(3)
c/Å	29.0162(8)	29.0452(8)	29.0893(8)	29.1217(9)	29.1266(9)
α/°	90	90	90	90	90
β/°	93.547(2)	93.638(2)	93.720(3)	93.780(3)	93.692(3)
γ/°	90	90	90	90	90
Volume/ų	1927.33(10)	1934.06(10)	1942.31(10)	1944.22(13)	1941.81(13)
ρ <sub>calc</sub> g/cm³	1.673	1.667	1.660	1.658	1.660
µ/mm⁻¹	1.933	1.927	1.918	1.917	1.919
F(000)	990.0	990.0	990.0	990.0	990.0
2O range for data collection/°	4.954 to 66.918	4.946 to 66.678	4.936 to 66.786	4.932 to 66.758	4.938 to 66.832
Reflections collected	28730	29054	29162	29135	29214
Independent reflections	6386 R <sub>int</sub> = 0.0846, R <sub>sigma</sub> = 0.0598	6412 R <sub>int</sub> = 0.0825, R <sub>sigma</sub> = 0.0578	6451 R <sub>int</sub> = 0.0815, R <sub>sigma</sub> = 0.0579	6443 R <sub>int</sub> = 0.0818, R <sub>sigma</sub> = 0.0604	6443 R <sub>int</sub> = 0.0905, R <sub>sigma</sub> = 0.0639
Data/restraints/p arameters	6386/0/234	6412/0/234	6451/0/234	6443/0/234	6443/0/234
Goodness-of-fit on F <sup>2</sup>	0.860	1.096	1.091	1.115	1.113
Final R indexes I>=2σ (I)	R <sub>1</sub> = 0.0591, wR <sub>2</sub> = 0.2053	$R_1 = 0.0564,$ w $R_2 = 0.1448$	R <sub>1</sub> = 0.0551, wR <sub>2</sub> = 0.1472	R <sub>1</sub> = 0.0584, wR <sub>2</sub> = 0.1597	$R_1 = 0.0642,$ w $R_2 = 0.1808$
Final R indexes all data	$R_1 = 0.0731,$ w $R_2 = 0.2194$	$R_1 = 0.0708,$ w $R_2 = 0.1508$	$R_1 = 0.0719,$ w $R_2 = 0.1528$	$R_1 = 0.0784,$ w $R_2 = 0.1661$	$R_1 = 0.0862,$ w $R_2 = 0.1897$
Largest diff. peak/hole / e Å <sup>-3</sup>	1.43/-0.54	1.55/-0.53	1.43/-0.54	1.29/-0.51	1.48/-0.53

 Table S2.
 Variable temperature SCXRD study raw data.

Temperature/K	300.00(10)	310.00(10)	320.00(10)	325.00(10)	330.00(10)
Crystal system	monoclinic	monoclinic	monoclinic	monoclinic	monoclinic
Space group	P2/c	P2/c	P2/c	P2/c	P2/c
a/Å	9.6002(4)	9.5988(4)	9.6010(5)	9.6114(7)	9.6134(7)
b/Å	6.9504(3)	6.9453(3)	6.9455(4)	6.9454(4)	6.9448(4)
c/Å	29.1051(10)	29.0622(11)	28.9943(14)	28.8982(18)	28.8189(17)
α/°	90	90	90	90	90
β/°	93.474(3)	93.108(3)	92.476(4)	91.564(5)	90.731(5)
γ/°	90	90	90	90	90
Volume/Å <sup>3</sup>	1938.48(13)	1934.63(14)	1931.64(18)	1928.4(2)	1923.9(2)
ρ <sub>calc</sub> g/cm³	1.663	1.667	1.669	1.672	1.676
µ/mm⁻¹	1.922	1.926	1.929	1.932	1.937
F(000)	990.0	990.0	990.0	990.0	990.0
2O range for data collection/°	4.948 to 66.946	4.964 to 66.73	4.992 to 66.986	5.028 to 66.722	5.064 to 66.974
Reflections collected	29140	29045	28736	28318	28459
Independent reflections	6421 R <sub>int</sub> = 0.1002, R <sub>sigma</sub> = 0.0684	6394 R <sub>int</sub> = 0.0985, R <sub>sigma</sub> = 0.0722	6402 R <sub>int</sub> = 0.1107, R <sub>sigma</sub> = 0.0815	6379 R <sub>int</sub> = 0.1191, R <sub>sigma</sub> = 0.0910	6358 R <sub>int</sub> = 0.1227, R <sub>sigma</sub> = 0.0900
Data/restraints/p arameters	6421/0/234	6394/0/234	6402/0/234	6379/0/234	6358/0/234
Goodness-of-fit on F <sup>2</sup>	1.122	1.069	1.074	1.066	1.038
Final R indexes I>=2σ (I)	R <sub>1</sub> = 0.0684, wR <sub>2</sub> = 0.1951	R <sub>1</sub> = 0.0676, wR <sub>2</sub> = 0.1863	R <sub>1</sub> = 0.0759, wR <sub>2</sub> = 0.2053	$R_1 = 0.0858,$ w $R_2 = 0.2233$	R <sub>1</sub> = 0.0922, wR <sub>2</sub> = 0.2653
Final R indexes all data	$R_1 = 0.0917,$ w $R_2 = 0.2044$	$R_1 = 0.0946$ , w $R_2 = 0.1970$	$R_1 = 0.1064,$ w $R_2 = 0.2165$	$R_1 = 0.1224,$ w $R_2 = 0.2374$	$R_1 = 0.1287,$ w $R_2 = 0.2847$
Largest diff. peak/hole / e Å <sup>-3</sup>	1.23/-0.55	0.95/-0.72	1.14/-0.68	0.84/-0.74	1.09/-1.29

Temperature/K	335.00(10)	340.00(10)	345.00(10)	350.00(10)	355.00(10)
Crystal system	monoclinic	monoclinic	monoclinic	monoclinic	monoclinic
Space group	P2/c	P2/c	P2/c	P2/c	P2/c
a/Å	9.6173(5)	9.6204(5)	9.6162(5)	9.6232(5)	9.6271(6)
b/Å	6.9458(3)	6.9456(3)	6.9418(3)	6.9478(3)	6.9548(4)
c/Å	28.7789(12)	28.7738(10)	28.7773(10)	28.8132(10)	28.8430(12)
α/°	90	90	90	90	90
β/°	90.232(4)	90.034(4)	90.116(3)	90.030(4)	90.156(4)
γ/°	90	90	90	90	90
Volume/Å <sup>3</sup>	1922.41(15)	1922.65(15)	1920.99(15)	1926.45(15)	1931.16(18)
ρ <sub>calc</sub> g/cm <sup>3</sup>	1.677	1.677	1.678	1.674	1.670
µ/mm⁻¹	1.938	1.938	1.940	1.934	1.930
F(000)	990.0	990.0	990.0	990.0	990.0
2O range for data collection/°	5.086 to 66.866	5.092 to 66.88	5.09 to 66.936	5.09 to 66.91	5.082 to 66.948
Reflections collected	28690	28859	29022	28982	28906
Independent reflections	6374 R <sub>int</sub> = 0.1239, R <sub>sigma</sub> = 0.0910	6375 R <sub>int</sub> = 0.1194, R <sub>sigma</sub> = 0.0875	6389 R <sub>int</sub> = 0.1199, R <sub>sigma</sub> = 0.0880	6395 R <sub>int</sub> = 0.1281, R <sub>sigma</sub> = 0.0935	6403 R <sub>int</sub> = 0.1263, R <sub>sigma</sub> = 0.0954
Data/restraints/p arameters	6374/0/234	6375/0/234	6389/0/235	6395/0/234	6403/0/234
Goodness-of-fit on F <sup>2</sup>	1.070	1.050	1.028	1.037	1.004
Final R indexes I>=2σ (I)	R <sub>1</sub> = 0.0957, wR <sub>2</sub> = 0.2723	R <sub>1</sub> = 0.0914, wR <sub>2</sub> = 0.2571	R <sub>1</sub> = 0.0908, wR <sub>2</sub> = 0.2522	R <sub>1</sub> = 0.0961, wR <sub>2</sub> = 0.2693	R <sub>1</sub> = 0.0967, wR <sub>2</sub> = 0.2732
Final R indexes all data	R <sub>1</sub> = 0.1324, wR <sub>2</sub> = 0.2906	R <sub>1</sub> = 0.1285, wR <sub>2</sub> = 0.2755	$R_1 = 0.1274,$ w $R_2 = 0.2718$	R <sub>1</sub> = 0.1353, wR <sub>2</sub> = 0.2906	$R_1 = 0.1408,$ w $R_2 = 0.2981$
Largest diff. peak/hole / e Å <sup>-3</sup>	1.25/-1.47	1.24/-1.45	0.93/-1.44	1.03/-1.63	1.10/-1.65

Temperature/K	360.00(10)	365.00(10)	370.00(10)	375.00(10)
Crystal system	monoclinic	monoclinic	monoclinic	monoclinic
Space group	P2/c	P2/c	P2/c	P2/c
a/Å	9.6255(6)	9.6250(6)	9.6281(7)	9.6290(7)
b/Å	6.9592(4)	6.9587(4)	6.9664(4)	6.9689(4)
c/Å	28.8493(12)	28.8587(12)	28.8775(14)	28.8838(14)
α/°	90	90	90	90
β/°	90.396(4)	90.597(4)	90.664(5)	90.775(5)
γ/°	90	90	90	90
Volume/Å <sup>3</sup>	1932.45(18)	1932.78(18)	1936.8(2)	1938.0(2)
ρ <sub>calc</sub> g/cm³	1.668	1.668	1.665	1.664
µ/mm <sup>-1</sup>	1.928	1.928	1.924	1.923
F(000)	990.0	990.0	990.0	990.0
2⊝ range for data collection/°	5.072 to 66.792	5.064 to 66.832	5.058 to 66.794	5.054 to 66.798
Reflections collected	28971	29104	29112	29184
Independent reflections	6399 R <sub>int</sub> = 0.1245, R <sub>sigma</sub> = 0.0959	6405 R <sub>int</sub> = 0.1272, R <sub>sigma</sub> = 0.0958	6417 R <sub>int</sub> = 0.1285, R <sub>sigma</sub> = 0.0969	6429 R <sub>int</sub> = 0.1291, R <sub>sigma</sub> = 0.0994
Data/restraints/p arameters	6399/0/235	6405/0/234	6417/0/234	6429/0/234
Goodness-of-fit on F <sup>2</sup>	1.004	0.998	1.003	1.011
Final R indexes I>=2σ (I)	R <sub>1</sub> = 0.0960, wR <sub>2</sub> = 0.2717	R <sub>1</sub> = 0.0970, wR <sub>2</sub> = 0.2775	R <sub>1</sub> = 0.0972, wR <sub>2</sub> = 0.2824	R <sub>1</sub> = 0.0977, wR <sub>2</sub> = 0.2792
Final R indexes all data	$R_1 = 0.1404,$ w $R_2 = 0.2953$	$R_1 = 0.1428$ , w $R_2 = 0.3032$	$R_1 = 0.1458,$ w $R_2 = 0.3086$	$R_1 = 0.1487,$ w $R_2 = 0.3069$
Largest diff. peak/hole / e Å <sup>-3</sup>	1.02/-1.44	1.25/-1.28	1.22/-1.16	1.28/-1.13

Compound	α <sub>v</sub> (10 <sup>-6</sup> K <sup>-1</sup> )	$\Delta T$	Ref.
Oxides			
ZrW <sub>2</sub> O <sub>8</sub>	-21.9	1049.7	5
α-HfW <sub>2</sub> O <sub>8</sub>	-26.4	378	6
β-HfW <sub>2</sub> O <sub>8</sub>	-16.5	92	6
γ-ZrMo <sub>2</sub> O <sub>8</sub>	-15.0	562	7
γ-HfMo <sub>2</sub> O <sub>8</sub>	-12.0	496	8
$\alpha$ -Zr <sub>0.7</sub> Sn <sub>0.3</sub> W <sub>2</sub> O <sub>8</sub>	-40.6	100	9
$\beta Zr_{0.7}Sn_{0.3}W_2O_8$	-18.7	450	9
$\alpha - 2r_{0.95} \prod_{0.05} W_2 O_8$ $\beta - 2r_{0.95} \prod_{0.05} W_2 O_8$	-30.0	105 168	10
$Zr_{0.95} \Pi_{0.05} W_2 O_8$ $Zr_{0.96} M_{0.04} W_2 O_{8.0.02} (M = Eu^{3+}, Er^{3+} and Yb^{3+})$	-30.9	70	10
$ZrW_{1,8}V_{0,2}O_{7,9}$	-29.1	74	12
$ZrW_{1,0}V_{0,2}O_{7,0}$	-4.8	175	12
$2rV_{2}O_{7}$	-20.1	498	13
$2rV_2O_7$	-75.5	168	13
	-10.5	501	14
	-5.7	360	15
$Zr_{0}$ $z_{0}V_{1}$ $z_{0}M_{0}$ $z_{0}C$ $z_{0}$	-11.3	510	16
ScoWoQ10	-6.7	1190	17
$Dv_2W_3O_{12}$	-26.0	350	18
$Y_2W_3O_{42}$	-21.0	1358	19
$F_2 M_3 O_{12}$	-20.2	600	20
$2 \cdot 2 \cdot 3 \circ 12$ Yb <sub>2</sub> W <sub>2</sub> O <sub>12</sub>	-19.1	600	20
$L_{12}W_{2}O_{12}$	-18.5	600	20
$Y_2MQ_2Q_{12}$	-37.8	770	21
$Er_2MO_2O_{12}$	-22.7	775	22
2 = 2 = 2 = 2 Yb <sub>2</sub> Mo <sub>2</sub> O <sub>12</sub>	-18.1	775	22
$Lu_2Mo_3O_{12}$	-18.1	775	22
$Sc_2Mo_3O_{12}$	-18.9	775	22
NbPO <sub>5</sub>	-11.0	300	23
TaPO₅	-1.7	400	23
TaVO <sub>5</sub>	-8.9	580	24
TaVO₅	-21.9	200	24
NbVO <sub>5</sub>	-6.6	575	25
TaAs <sub>0.1</sub> V <sub>0.9</sub> O <sub>5</sub>	-17.5	480	23
ReO <sub>3</sub>	-1.8	80	24
$SiO_2$ ( $\beta$ -cristobalite)	-6.9	548	25
Cu <sub>2</sub> O	-7.2	231	26
Ag <sub>2</sub> O	-21	145	27
Ag <sub>2</sub> O	-27.1	173	26
AIPO <sub>4</sub> -17	-35.1	282	28
H-ZSM-5	-27	800	29
Zn <sub>2</sub> V <sub>2</sub> O <sub>7</sub>	-17.9	300	30
a-Cu <sub>2</sub> V <sub>2</sub> O <sub>7</sub>	-10.2	375	31
Fluorides	06 E	1160	20
Valif <sub>6</sub>	-20.3	1103	32
	-13.4	3/3	00 22
Fe∠rF <sub>6</sub>	-9.7	3/3	33

Table S3. Numeric data in Figure 2b.

MgZrF <sub>6</sub>	-2.4	373	34
CaHfF <sub>6</sub>	-45.3	288	32
CaNbF <sub>6</sub>	-36.5	775	35
TiZrF <sub>6</sub>	-6.1	323	36
YbZrF <sub>7</sub>	-6.0	132	37
Cyanides			
Zn(CN) <sub>2</sub>	-50.7	350	38
Cd(CN) <sub>2</sub> (double-network)	-61.2	225	38
Cd(CN) <sub>2</sub> (single-network)	-100	225	39
Zn[Ag(CN) <sub>2</sub> ] <sub>2</sub>	-13.6	275	40
AgB(CN) <sub>4</sub>	-40.0	500	41
CuB(CN) <sub>4</sub>	-9.2	300	41
YFe(CN) <sub>6</sub>	-33.7	225	42
FeCo(CN) <sub>6</sub>	-4.4	295.8	43
GaFe(CN) <sub>6</sub>	-11.9	375	44
FeFe(CN) <sub>6</sub>	-12.8	350	45
ScCo(CN) <sub>6</sub>	-19.8	575	46
TiCo(CN) <sub>6</sub>	-12.2	375	47
LaCo(CN) <sub>6</sub>	-43.9	400	48
SmCo(CN) <sub>6</sub>	-37.4	400	48
HoCo(CN) <sub>6</sub>	-30.2	400	48
LuCo(CN) <sub>6</sub>	-27.2	400	48
YCo(CN) <sub>6</sub>	-31.6	400	48
ErCo(CN) <sub>6</sub>	-27.0	275	49
CsCd[Fe(CN) <sub>6</sub> ]·0.5H <sub>2</sub> O	-26.4	200	50
$Cs_0 _7Ni[Fe(CN)_6]_0 _9 \cdot 2.9H_2O$	-1.2	200	50
Cs <sub>0.97</sub> Cu[Fe(CN) <sub>6</sub> ] <sub>0.99</sub> ·1.1H <sub>2</sub> O	-6.3	200	50
Cs <sub>0.91</sub> Ni[Fe(CN) <sub>6</sub> ] <sub>0.97</sub> ·0.4H <sub>2</sub>	-12.3	200	50
Rb <sub>0.78</sub> Fe[Fe(CN) <sub>6</sub> ] <sub>0.83</sub> ·2.8H <sub>2</sub> O	-6.3	200	50
Rb <sub>0.64</sub> Zn[Fe(CN) <sub>6</sub> ] <sub>0.88</sub> ·2.3H <sub>2</sub> O	-17.7	200	50
CdPt(CN) <sub>6</sub>	-30.1	140	51
MnPt(CN) <sub>6</sub>	-19.7	200	51
FePt(CN) <sub>6</sub>	-12.0	215	51
CoPt(CN) <sub>6</sub>	-4.8	250	51
NiPt(CN) <sub>6</sub>	-3.1	230	51
ZnPt(CN) <sub>6</sub>	-10.6	300	51
Mn <sub>3</sub> Co(CN) <sub>6</sub> ] <sub>2</sub> ·12H <sub>2</sub> O	-87.6	175	52
Fe <sub>3</sub> Co(CN) <sub>6</sub> ] <sub>2</sub> ·12H <sub>2</sub> O	-58.8	175	52
$Co_3Co(CN)_6]_2 \cdot 12H_2O$	-119.1	175	52
Ni <sub>3</sub> Co(CN) <sub>6</sub> ] <sub>2</sub> ·12H <sub>2</sub> O	-83.0	175	52
$Cu_3Co(CN)_6]_2 \cdot 12H_2O$	-58.7	175	52
Zn <sub>3</sub> Co(CN) <sub>6</sub> ] <sub>2</sub> ·12H <sub>2</sub> O	-89.1	175	52
Fe <sub>3</sub> Fe(CN) <sub>6</sub> ] <sub>2</sub> ·16H <sub>2</sub> O	-29.7	175	52
Cu <sub>3</sub> Fe(CN) <sub>6</sub> ] <sub>2</sub> ·16H <sub>2</sub> O	-59.7	175	52
$Zn_3Fe(CN)_6]_2 \cdot 16H_2O$	-118.8	175	52
MOFs			
MOF-5	-39.3	420	53
HKUST-1	-12.3	420	54
UiO-66(Zr)	-97	180	55

MIL-68(In)	-12.3	475	56
Zn <sub>8</sub> (SiO <sub>4</sub> )(m-BDC) <sub>6</sub>	-4.4	375	57
Cu-TDPAT	-19.7	400	58
$M_2C_4O_4 \cdot 2H_2O$ (M = Zn, Cd)	-13.9	250	59



Figure S1. Auxiliary H-bonds between  $NH_3$  and M(CA).







Figure S4. N<sub>2</sub> Adsorption/desorption curves. Degas temperature: 105 °C.



**Figure S5**. PXRD patterns of crystals obtained at different conditions. L-L means all reactants were in solutions state at the commencement of reaction; following is the molar ratio of CA:M. S-L means that the M added in the reaction system was particles (without dissolved) and CA was in solution state (the amount of M do not affect the outcome).



Figure S6. PXRD patterns at varied temperatures.



Figure S7. unit cell parameters refined from in-situ PXRD patterns: (a) a and b axis,

(b) *c* axis, (c) beta angle, (d) volume change.



**Figure S8**. the length change of (a) *a* axis, (b) *b* axis, (c) axis, (d)  $\beta$  angle upon cyclic heating/cooling.

Figure S9. illustration of  $\theta_2$  and  $\,\omega_2^{}$  rotations of CA02 with respect to Zn.



**Figure S10**. (a) Length change of coordination subunits CA-Zn-CA-Zn-CA with respect to temperature (measured between the two farthest oxygen atoms) and (b) the corresponding dimension change.



Figure S11. (a) Average length of Zn-N(CA) bonds, (b) average length of triple H-bonds

between M-CA01, (c) the corresponding dimension change.

## Reference

- 1. Diffraction, R. O. CrysAlis Pro, 40\_64.53a; Rigaku Oxford Diffraction: 2019.
- 2. Sheldrick, G. M., SHELXT Integrated space-group and crystal-structure determination. Acta Crystallographica a-Foundation and Advances 2015, 71, 3-8.
- 3. Sheldrick, G. M., A short history of SHELX. Acta Crystallographica Section A 2008, 64, 112-122.
- Dolomanov, O. V.; Bourhis, L. J.; Gildea, R. J.; Howard, J. A. K.; Puschmann, H., OLEX2: a complete structure solution, refinement and analysis program. Journal of Applied Crystallography 2009, 42, 339-341.
- 5. T. Mary, J. Evans, T. Vogt, A. Sleight, Science 272 (1996) 90-92.
- 6. Y. Yamamura, N. Nakajima, T. Tsuji, Phys. Rev. B 64 (2001) 184109.
- 7. C. Lind, A.P. Wilkinson, Z. Hu, S. Short, J.D. Jorgensen, Chem. Mater. 10 (1998) 2335–2337.
- 8. C. Lind, Negative thermal expansion materials related to cubic zirconium tungstate, Georgia Institute of Technology, 2001.
- 9. C. De Meyer, F. Bouree, J.S. Evans, K. De Buysser, E. Bruneel, I. Van Driessche, S. Hoste, J. Mater. Chem. 14 (2004) 2988–2994.
- K. De Buysser, I. Van Driessche, B.V. Putte, J. Schaubroeck, S. Hoste, J. Solid State Chem. 180 (2007) 2310–2315.
- 11. H.H. Li, J.S. Han, H. Ma, L. Huang, X.H. Zhao, J. Solid State Chem. 180 (2007) 852-857.
- 12. X. Chen, F. Guo, X. Deng, J. Tao, H. Ma, X. Zhao, J. Alloys Compd. 537 (2012) 227–231.
- 13. R. Withers, J. Evans, J. Hanson, A. Sleight, J. Solid State Chem. France 137 (1998) 161–167.
- G. Wallez, P.E. Raison, N. Dacheux, N. Clavier, D. Bykov, L. Delevoye, K. Popa, D. Bregiroux, A.N. Fitch, R.J. Konings, Inorg. Chem. 51 (2012) 4314–4322.
- 15. K. White, P.L. Lee, P.J. Chupas, K. Chapman, E.A. Payzant, A.C. Jupe, W. Bassett, C.S. Zha, A.P. Wilkinson, Chem. Mater. 20 (2008) 3728–3734.
- 16. M. Zhang, Y. Mao, J. Guo, W. Zhou, M. Chao, N. Zhang, M. Yang, X. Kong, X. Kong, E. Liang, RSC Adv. 7 (2017) 3934–3940.
- 17. T. Mary, A. Sleight, J. Mater. Res. 14 (1999) 912–915.
- 18. W. Cao, Q. Li, K. Lin, Z. Liu, J. Deng, J. Chen, X. Xing, RSC Adv. 6 (2016) 96275–96280.
- 19. P. Forster, A. Sleight, Int. J. Inorg. Mater. 1 (1999) 123-127.
- 20. S. Sumithra, A. Tyagi, A. Umarji, Mater. Sci. Eng., B 116 (2005) 14–18.
- 21. B. Marinkovic, P. Jardim, R. De Avillez, F. Rizzo, Solid State Sci. 7 (2005) 1377-1383.
- 22. S. Sumithra, A. Umarji, Solid State Sci. 8 (2006) 1453–1458.
- 23. T.G. Amos, Negative thermal expansion in AOMO<sub>4</sub> compounds, Oregon State University, 2000.
- 24. X. Wang, Q. Huang, J. Deng, R. Yu, J. Chen, X. Xing, Inorg. Chem. 50 (2011) 2685–2690.
- 25. J. Wang, J. Deng, R. Yu, J. Chen, X. Xing, Dalton Trans. 40 (2011) 3394–3397.
- 26. W. Tiano, M. Dapiaggi, G. Artioli, J. Appl. Crystallogr. 36 (2003) 1461–1463.
- 27. B.J. Kennedy, Y. Kubota, K. Kato, Solid State Commun. 136 (2005) 177–180.
- 28. M.P. Attfield, A.W. Sleight, Chem. Mater. 10 (1998) 2013–2019.
- 29. B. Marinkovic, P. Jardim, A. Saavedra, L. Lau, C. Baehtz, R. De Avillez, F. Rizzo, Micropor. Mesopor. Mat. 71 (2004) 117–124.

- 30. M. Rotermel, T. Krasnenko, Crystallogra. Rep. 62 (2017) 703–709.
- N. Shi, A. Sanson, A. Venier, L. Fan, C. Sun, X. Xing, J. Chen, Chem. Comm. 56 (2020) 10666– 10669.
- J.C. Hancock, K.W. Chapman, G.J. Halder, C.R. Morelock, B.S. Kaplan, L.C. Gallington, A. Bongiorno, C. Han, S. Zhou, A.P. Wilkinson, Chem. Mater. 27 (2015) 3912–3918.
- 33. L. Hu, J. Chen, J. Xu, N. Wang, F. Han, Y. Ren, Z. Pan, Y. Rong, R. Huang, J. Deng, J. Am. Chem. Soc. 138 (2016) 14530–14533.
- 34. J. Xu, L. Hu, Y. Song, F. Han, Y. Qiao, J. Deng, J. Chen, X. Xing, J. Am. Ceram. Soc. 100 (2017) 5385–5388.
- 35. B.R. Hester, J.C. Hancock, S.H. Lapidus, A.P. Wilkinson, Chem. Mater. 29 (2017) 823-831.
- 36. C. Yang, Y. Zhang, J. Bai, B. Qu, P. Tong, M. Wang, J. Lin, R. Zhang, H. Tong, Y. Wu, J. Mater. Chem. C 6 (2018) 5148–5152.
- J.O. Ticknor, B.R. Hester, J.W. Adkins, W. Xu, A.A. Yakovenko, A.P. Wilkinson, Chem. Mater. 30 (2018) 3071–3077.
- 38. A.L. Goodwin, C.J. Kepert, Phys. Rev. B 71 (2005) 140301.
- 39. A.E. Phillips, A.L. Goodwin, G.J. Halder, P.D. Southon, C.J. Kepert, Angew. Chem. Int. Ed. 47 (2008) 1396–1399.
- 40. A.L. Goodwin, B.J. Kennedy, C.J. Kepert, J. Am. Chem. Soc. 131 (2009) 6334-6335.
- 41. Q. Gao, J. Wang, A. Sanson, Q. Sun, E. Liang, X. Xing, J. Chen, J. Am. Chem. Soc. 142 (2020) 6935–6939.
- 42. Q. Gao, J. Chen, Q. Sun, D. Chang, Q. Huang, H. Wu, A. Sanson, R. Milazzo, H. Zhu, Q. Li, Angew. Chem. Int. Ed. 56 (2017) 9023–9028.
- 43. S. Margadonna, K. Prassides, A.N. Fitch, J. Am. Chem. Soc. 126 (2004) 15390- 15391.
- 44. Q. Gao, N. Shi, Q. Sun, A. Sanson, R. Milazzo, A. Carnera, H. Zhu, S.H. Lapidus, Y. Ren, Q. Huang, Inorg. Chem. 57 (2018) 10918–10924.
- 45. N. Shi, Q. Gao, A. Sanson, Q. Li, L. Fan, Y. Ren, L. Olivi, J. Chen, X. Xing, Dalton Trans. 48 (2019) 3658–3663.
- Q. Gao, Y. Sun, N. Shi, R. Milazzo, S. Pollastri, L. Olivi, Q. Huang, H. Liu, A. Sanson, Q. Sun, Scr. Mater. 187 (2020) 119–124.
- 47. Y. Li, Q. Gao, D. Chang, P. Sun, J. Liu, Y. Jia, E. Liang, Q. Sun, J. Phys, Condens. Mat. 32 (2020) 455703.
- 48. S.G. Duyker, V.K. Peterson, G.J. Kearley, A.J. Ramirez-Cuesta, C.J. Kepert, Angew. Chem. Int. Ed. 52 (2013) 5266–5270.
- 49. T. Pretsch, K.W. Chapman, G.J. Halder, C.J. Kepert, Chem. Comm. (2006) 1857–1859.
- 50. T. Matsuda, J. Kim, K. Ohoyama, Y. Moritomo, Phys. Rev. B 79 (2009) 172302.
- 51. K.W. Chapman, P.J. Chupas, C.J. Kepert, J. Am. Chem. Soc. 128 (2006) 7009-7014.
- 52. S. Adak, L.L. Daemen, M. Hartl, D. Williams, J. Summerhill, H. Nakotte, J. Solid State Chem. 184 (2011) 2854–2861.
- 53. N. Lock, Y. Wu, M. Christensen, L.J. Cameron, V.K. Peterson, A.J. Bridgeman, C.J. Kepert, B.B. Iversen, J. Phys. Chem. C 114 (2010) 16181–16186.

- 54. Y. Wu, A. Kobayashi, G.J. Halder, V.K. Peterson, K.W. Chapman, N. Lock, P.D. Southon, C.J. Kepert, Angew. Chem. Int. Ed. 47 (2008) 8929–8932.
- 55. M.J. Cliffe, J.A. Hill, C.A. Murray, F.X. Coudert, A.L. Goodwin, Phys. Chem. Chem. Phys. 17 (2015) 11586–11592.
- 56. Z. Liu, Q. Li, H. Zhu, K. Lin, J. Deng, J. Chen, X. Xing, Chem. Comm. 54 (2018) 5712–5715.
- 57. Z. Liu, X. Jiang, C. Wang, C. Liu, Z. Lin, J. Deng, J. Chen, X. Xing, Inorg. Chem. Front. 6 (2019) 1675–1679.
- 58. M. Asgari, I. Kochetygov, H. Abedini, W.L. Queen, Nano Res. (2020) 1-7.
- 59. Z. Liu, R. Ma, J. Deng, J. Chen, X. Xing, Chem. Mater. 32 (2020) 2893–2898.