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Electronic Supplementary Information

B-Site Substitution in NaCo_{1-2x}Fe_xNi_xF₃ Perovskites for Efficient Oxygen Evolution

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Figure S9. Mass activities plots for NCFNF(433) and RuO₂.



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Figure S18. Nyquist plots of the catalysts in 1 M KOH at 1.524 V (vs RHE) and the corresponding equivalent circuit.



Figure S19. Raman spectra of NCFNF(433) before (a) and after (b) the OER testing. (Raman spectrum test conditions: the wavelength of the laser is 532nm, the output power is 1.5mW, the width of the slit is 50um, and the exposure time is 5s; the powder samples were tested at room temperature)

Table S1. EDS results of $NaCo_{1-x}Fe_xNi_xF_3$.

	Atomic ratio (%)					
Catalysts	Со	Fe	Ni	Co/Fe/Ni		
NCF	1	0	0	1/0/0		
NCFNF(811)	0.81	0.12	0.07	0.81/0.12/0.07		
NCFNF(622)	0.61	0.21	0.18	0.61/0.21/0.18		
NCFNF(433)	0.43	0.27	0.29	0.43/0.27/0.29		

Table S2. ECSA results of $NaCo_{1-x}Fe_xNi_xF_3$ and RuO_2

Catalysts	ECSA for OER		
NCF	1725 cm ⁻²		
NCFNF(811)	2100 cm ⁻²		
NCFNF(622)	2475 cm ⁻²		
NCFNF(433)	2750 cm ⁻²		
NCFF(73)	650 cm ⁻²		
NCNF(73)	975 cm ⁻²		
RuO ₂	275 cm ⁻²		

Catalysts	η (mv) @10 mA cm ⁻²	Tafel slope (mVdec ⁻¹)	Stability	Ref.
NCFNF(433)	265	49	100 h	This work
K _{0.8} Na _{0.2} (MgMnFeCoNi)F ₃	314	55	10 h	1
SrCoO _{2.85-δ} F _{0.15}	420	60	20 h	2
La(CrMnFeCo ₂ Ni)O ₃	325	51.2	50 h	3
$Sr_{0.95}Ce_{0.05}Fe_{0.9}Ni_{0.1}O_{3-\delta}$	340	51	30 h	4
$Sr_2Fe_{0.8}Co_{0.2}Mo_{0.65}Ni_{0.35}O_{6-\delta}$	340	56	5 h	5
LaCoO ₃ -80nm	490	69		6
$Sr_2Co_{1.5}Fe_{0.5}O_{6-\delta}$	318	44.8	9.7 h	7
$SrNb_{0.1}Co_{0.7}Fe_{0.2}O_{3-\delta}$	390	134	10 h	8
$BaCo_{0.7}Fe_{0.2}Sn_{0.1}O_{3-\delta}$	380	69	2 h	9
$La_5Ni_4O_{13-\delta}$	390	70	1000 cv	10
La ₂ NiMnO ₆	370	58	12.5 h	11
$F\text{-}Ba_{0.5}Sr_{0.5}Co_{0.8}Fe_{0.2}O_{3\text{-}\delta}$	280	102	100 h	12
$PrBa_{0.5}Sr_{0.5}Co_{1.5}Fe_{0.5}O_{5+\delta}$	358	52	12 h	13
$La_{0\cdot 6}Sr_{0\cdot 4}Co_{0\cdot 8}Fe_{0\cdot 2}O_3@Ni_3(HITP)_2$	272	95	12 h	14
$SrCo_{0.95}P_{0.05}O_{3-\delta}$	290	52	1000 cv	15
$Ba_{0.5}Sr_{0.5}Co_{0.8}Fe_{0.2}O_{3-\delta}$	340	70	10 h	16
$Ba_2CoMo_{0.5}Nb_{0.5}O_{6-\delta}$	445	77	2 h	17

References

[S1] T. Wang, H. Chen, Z. Yang, J. Liang and S. Dai, High-Entropy Perovskite Fluorides: A New Platform for Oxygen Evolution Catalysis, J. Am. Chem. Soc., 2020, **142**, 4550-4554.

[S2] W. Wang, Y. Yang, D. Huan, L. Wang, N. Shi, Y. Xie, C. Xia, R. Peng and Y. Lu, An excellent OER electrocatalyst of cubic SrCoO_{3-δ} prepared by a simple F-doping strategy, J. Mater. Chem. A., 2019, **7**, 12538-12546.

[S3] T. X. Nguyen, Y. C. Liao, C. C. Lin, Y. H. Su and J. M. Ting, Advanced High Entropy Perovskite Oxide Electrocatalyst for Oxygen Evolution Reaction, Adv. Funct. Mater., 2021, **31**, 2101632.

[S4] S. She, Y. Zhu, X. Wu, Z. Hu, A. Shelke, W. F. Pong, Y. Chen, Y. Song, M. Liang, C. T. Chen, H. Wang, W. Zhou and Z. Shao, Realizing High and Stable Electrocatalytic Oxygen Evolution for Iron-Based Perovskites by Co-Doping-Induced Structural and Electronic Modulation, Adv. Funct. Mater., 2021, **32**, 2111091.

[S5] H. Sun, X. Xu, Z. Hu, L. H. Tjeng, J. Zhao, Q. Zhang, H.-J. Lin, C.-T. Chen, T.-S. Chan, W. Zhou and Z. Shao, Boosting the oxygen evolution reaction activity of a perovskite through introducing multi-element synergy and building an ordered structure, J. Mater. Chem. A., 2019, **7**, 9924-9932.

[S6] S. Zhou, X. Miao, X. Zhao, C. Ma, Y. Qiu, Z. Hu, J. Zhao, L. Shi and J. Zeng, Engineering electrocatalytic activity in nanosized perovskite cobaltite through surface spin-state transition, Nat. Commun., 2016, **7**, 11510.

[S7] S. R. Ede, C. N. Collins, C. D. Posada, G. George, H. Wu, W. D. Ratcliff, Y. Lin, J. Wen, S. Han and Z. Luo, Intermediate Sr₂Co_{1.5}Fe_{0.5}O_{6-δ} Tetragonal Structure between Perovskite and Brownmillerite as a Model Catalyst with Layered Oxygen Deficiency for Enhanced Electrochemical Water Oxidation, ACS. Catal., 2021, **11**, 4327-4337.

[S8] Y. Zhu, W. Zhou, Y. Zhong, Y. Bu, X. Chen, Q. Zhong, M. Liu and Z. Shao, A Perovskite Nanorod as Bifunctional Electrocatalyst for Overall Water Splitting, Adv. Energy. Mater., 2017, **7**, 1602122.

[S9] X. Xu, C. Su, W. Zhou, Y. Zhu, Y. Chen and Z. Shao, Co-doping Strategy for Developing Perovskite Oxides as Highly Efficient Electrocatalysts for Oxygen Evolution Reaction, Adv. Sci, 2016, **3**, 1500187.

[S10] S. R. Choi, J.-I. Lee, H. Park, S. W. Lee, D. Y. Kim, W. Y. An, J. H. Kim, J. Kim, H.-S. Cho and J.-Y. Park, Multiple perovskite layered lanthanum nickelate Ruddlesden-Popper systems as highly active bifunctional oxygen catalysts, Chem. Eng. J., 2021, **409**, 128226.

[S11] Y. Tong, J. Wu, P. Chen, H. Liu, W. Chu, C. Wu and Y. Xie, Vibronic Superexchange in Double Perovskite Electrocatalyst for Efficient Electrocatalytic Oxygen Evolution, J. Am. Chem. Soc., 2018, **140**, 11165-11169.

[S12] J. Xiong, H. Zhong, J. Li, X. Zhang, J. Shi, W. Cai, K. Qu, C. Zhu, Z. Yang, S. P. Beckman and H. Cheng, Engineering highly active oxygen sites in perovskite oxides for stable and efficient oxygen evolution, Appl. Catal., B, 2019, **256**, 117817.

[S13] B. Zhao, L. Zhang, D. Zhen, S. Yoo, Y. Ding, D. Chen, Y. Chen, Q. Zhang, B. Doyle, X. Xiong and M. Liu, A tailored double perovskite nanofiber catalyst enables ultrafast oxygen evolution, Nat. Commun., 2017, **8**, 14586.

[S14] Z. Li, J.-G. Li, X. Ao, H. Sun, H. Wang, M.-F. Yuen and C. Wang, Conductive metal-Organic frameworks endow high-efficient oxygen

evolution of La0·6Sr0·4Co0·8Fe0·2O3 perovskite oxide nanofibers, Electrochim. Acta, 2020, **334**, 135638, DOI: 10.1016/j.electacta.2020.135638 [S15] Y. Zhu, W. Zhou, J. Sunarso, Y. Zhong and Z. Shao, Phosphorus-Doped Perovskite Oxide as Highly Efficient Water Oxidation Electrocatalyst in Alkaline Solution, Adv. Funct. Mater., 2016, **26**, 5862-5872.

[S16] G. Chen, W. Zhou, D. Q. Guan, S. Jaka, Y. P. Zhu, X. F. Hu and W. S. Zhang, Z.P., Two orders of magnitude enhancement in oxygen evolution reactivity on amorphous Ba_{0.5}S_{r0.5}Co_{0.8}Fe_{0.2}O_{3-δ} nanofilms with tunable oxidation state, Sci. Adv., 2017, **3**, e1603206.

[S17] H. Sun, G. Chen, J. Sunarso, J. Dai, W. Zhou and Z. Shao, Molybdenum and Niobium Codoped B-Site-Ordered Double Perovskite Catalyst for Efficient Oxygen Evolution Reaction, ACS Appl. Mater. Interfaces, 2018, **10**, 16939-16942.