

## Supporting Information

### Understanding the Synergistic Influence of Propylammonium Bromide Additive and Erbium Doped CsPbI<sub>2</sub>Br for Highly Stable Inorganic Perovskite Solar Cells

*Jyoti V. Patil,<sup>a, b</sup> Sawanta S. Mali,<sup>b</sup> Shivaji B. Sadale,<sup>c</sup> Chang Kook Hong<sup>a, b\*</sup>*

*<sup>a</sup>Optoelectronic Convergence Research Center, School of Chemical Engineering, Chonnam National University, Gwangju, South Korea 61186.*

*<sup>b</sup>Polymer Energy Materials Laboratory, School of Chemical Engineering, Chonnam National University, Gwangju, South Korea 61186.*

*<sup>c</sup>Department of Technology, Shivaji University, Kolhapur, 416004, Maharashtra, India*

**\*Corresponding author:** hongck@chonnam.ac.kr

**S1: Details of different chemicals:**

The chemical materials including cesium iodide (CsI) (Sigma, 99.999 %), lead bromide (PbBr<sub>2</sub>) (Sigma, ≥ 98%), lead iodide (PbI<sub>2</sub>) (Sigma, 99%), erbium (III) chloride (ErCl<sub>3</sub>) (Sigma, 99.99%), *n*-Propylammonium bromide (PABr) (Sigma, 99.99%), dimethyl sulfoxide (DMSO), (N, N-dimethylformamide (DMF), poly(3-hexylthiophene-2,5-diyl) (P3HT) (Sigma-Aldrich, average Mw 50,000-100,000) were purchased from Sigma-Aldrich. The hot-air gun (BOSCH, GHG 630 DCE Hot Air Gun -0601 94C 740) was used as a hot-air source.

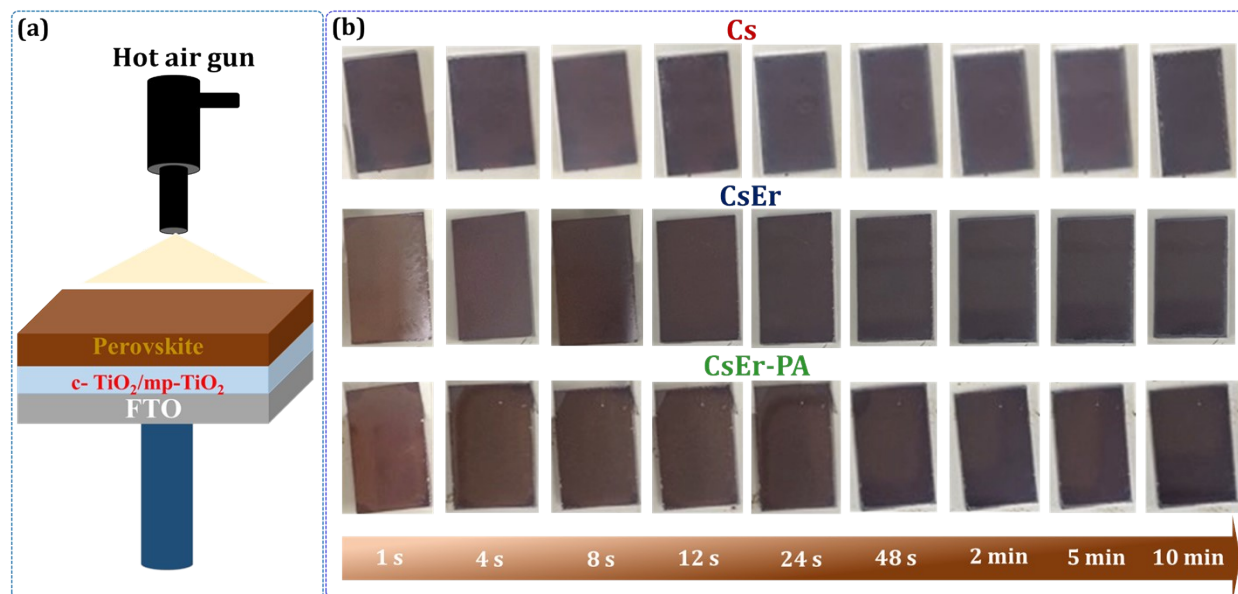
**S2: Details of different characterization of perovskite material and IPVSC-based devices:**

The surface morphological images were recorded by a scanning electron microscope (SEM; S-4700, Hitachi). The X-ray diffraction (XRD) measurements were carried out using a D/MAX Ultima III-XRD spectrometer (PAN analysis, Japan) with Cu K $\alpha$  line of 1.5410 Å. Optical absorption measurements were carried out on a UV-vis spectrophotometer (Varian, CARY, 300 Conc.) in the 500-900 nm wavelength range. Energy-dispersive X-ray (EDX) elemental mapping was carried out in the scanning transmission electron microscopy (STEM) mode. The contrast against the background of HAADF-STEM image has been carried out background selection: Rois mode at detector angle: 14.6 degree. The elemental information was studied using an X-ray photoelectron spectroscopy (XPS) (VG Multilab 2000-Thermo Scientific, USA, K-Alpha) with a multi-channel detector, which can endure high photonic energies from 0.1 to 3 keV. The ultraviolet-photoelectron spectroscopy (UPS) measurements were carried out using Thermo VG Scientific Sigma Probe, UK, UV Monochromatic He I (21.22 eV) gas, (Bias -5.0 V, Aperture size 8 mm adjustable) for UPS base UHV pressure 1 x 10<sup>-9</sup> mB without surface treatment. The spectral response was taken by a spectral IPCE measurements system (K3100, Mc-Science), which was equipped with a monochromator, a K240×E 300 lamp source connected with K401 OLS XE300W lamp power supply and a K102 signal amplifier. Prior to the use of the light intensity were calibrated using a Si-photodiode (Model: S1337-1010BQ) and In GaAs photodiode (model: G12180-050A) for 300-1100 nm and 1100-1400 nm calibration respectively. Measurements were taken in EQE mode. Time-resolved photoluminescence (TRPL) decay transients were measured at 650 nm using excitation with a 470 nm light pulse at a frequency of 5MHz from the Spectrophotometer F-7000.

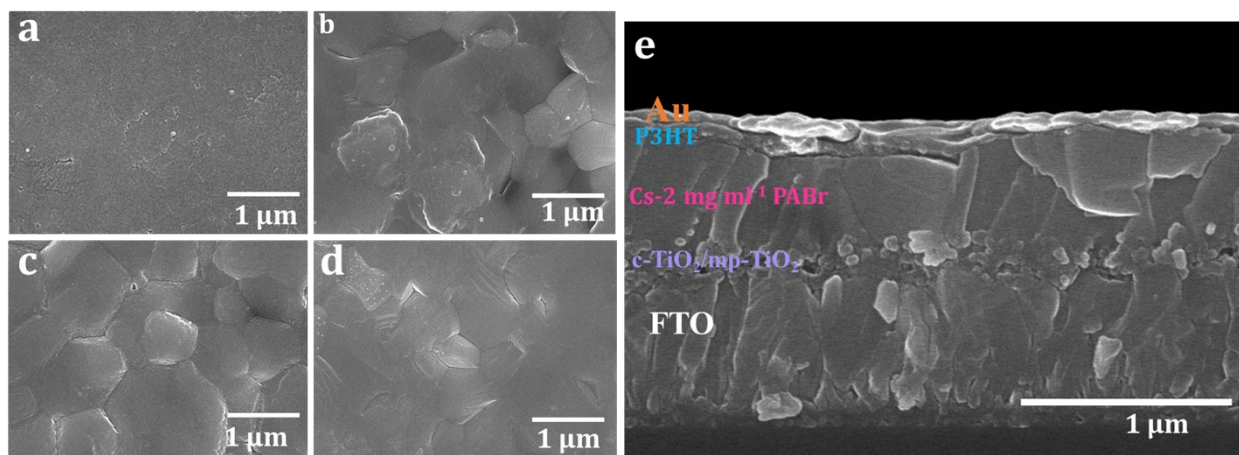
The IPVSCs were illuminated using a solar simulator at AM 1.5 G for 10 s, where the light intensity was adjusted with an NREL-calibrated Si solar cell with a KG-5 filter to 1 sun intensity

(100 mW cm<sup>-2</sup>). The photovoltaic performance of IPVSC devices were measured by metal shadow masks with active small area of 0.3 x 0.3 = 0.09 cm<sup>2</sup>. The exact illumination to the active area was fixed via attaching thin metal shadow mask from back side throughout measurements. The *J-V* curves were measured along with the reverse scan direction from 1.5 V to - 0.05 V or the forward scan direction from -0.05 V to 1.5 V. Using the Mott–Gurney law, the trap density of the electron-only device with an architecture of FTO/mp-TiO<sub>2</sub>/perovskite/PCBM/Ag was measured with the space charge limited current (SCLC) model from 0 to 7 V under dark conditions. The Nyquist plots were obtained via conducting electrochemical impedance spectroscopy (EIS) analysis with the help of Iviumstat (Ivium Technologies B.V., Eindhoven, and the Netherlands) at an open-circuit potential at frequencies ranging from 10<sup>-1</sup> to 10<sup>5</sup> Hz with AC amplitude of 10 mV. For equivalent circuit analysis, Z-view 2.8d was used. The steady-state efficiencies were obtained by tracking the maximum power point. The IPVSC devices were illuminated using a solar simulator at AM 1.5 G for 10 s, where the light intensity was adjusted with an NREL-calibrated Si solar cell with a KG-5 filter to 1 sun intensity (100 mW cm<sup>-2</sup>). The device stability was tested in an air at 65 °C without encapsulation and after measurement devices were stored in ambient conditions. The *J-V* curves were periodically measured under AM1.5 G simulated sunlight at room temperature. For thermal stability, perovskite devices were kept on the hot plate at 65 °C and taken out during measurements. We monitored devices temperature ~ 20 °C by continuous air-steam flow and relative humidity was ~ 30 % for continuous illumination testing.

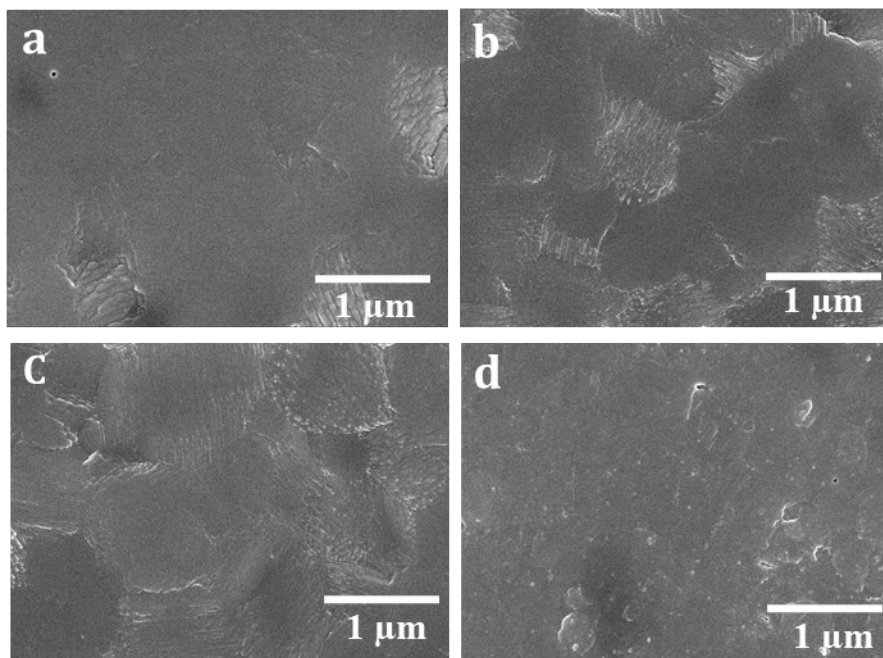
**Figure S1:** (a) The Cs, CsEr and CsEr-PA perovskite thin films deposited by the hot air flow method and (b) time dependent photographs of the Cs, CsEr and CsEr-PA-based perovskite thin films.



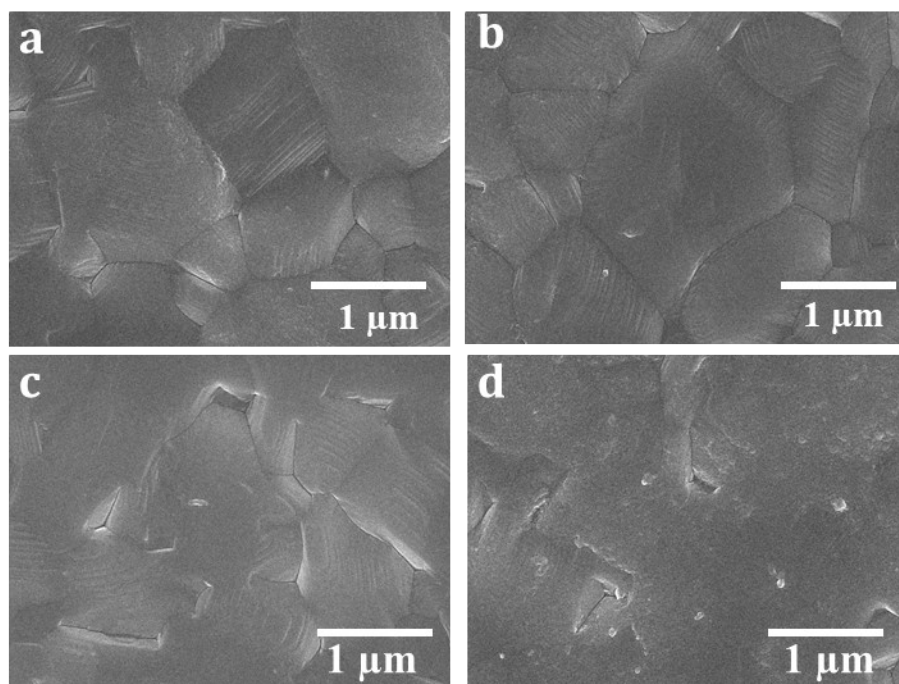
**Figure S2:** SEM images: (a) CsPbI<sub>2</sub>Br, (b) Cs-1 mg ml<sup>-1</sup> PABr, (c) Cs-2 mg ml<sup>-1</sup> PABr, (d) Cs-3 mg ml<sup>-1</sup> PABr and (e) Cross section of Cs-2 mg ml<sup>-1</sup> PABr-based IPVSC.



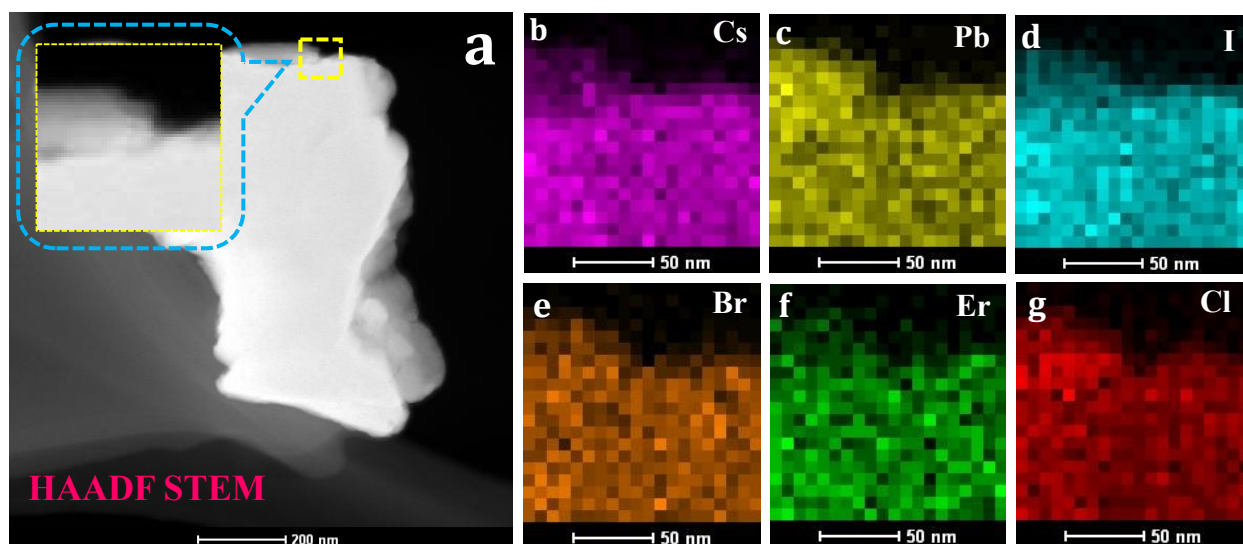
**Figure S3: Scanning electron microscopic images:** (a)  $\text{CsPb}_{0.99}\text{Er}_{0.01}\text{I}_2\text{BrCl}_{0.03}$ , (b)  $\text{CsPb}_{0.98}\text{Er}_{0.02}\text{I}_2\text{BrCl}_{0.06}$ , (c)  $\text{CsPb}_{0.97}\text{Er}_{0.03}\text{I}_2\text{BrCl}_{0.09}$ , (d)  $\text{CsPb}_{0.96}\text{Er}_{0.04}\text{I}_2\text{BrCl}_{0.12}$  perovskite films.



**Figure S4: Scanning electron microscopic images:** (a) CsEr-1mg ml<sup>-1</sup> PABr, (b) CsEr-2mg ml<sup>-1</sup> PABr, (c) CsEr-3mg ml<sup>-1</sup> PABr and (d) CsEr-4mg ml<sup>-1</sup> PABr-based perovskite thin films.

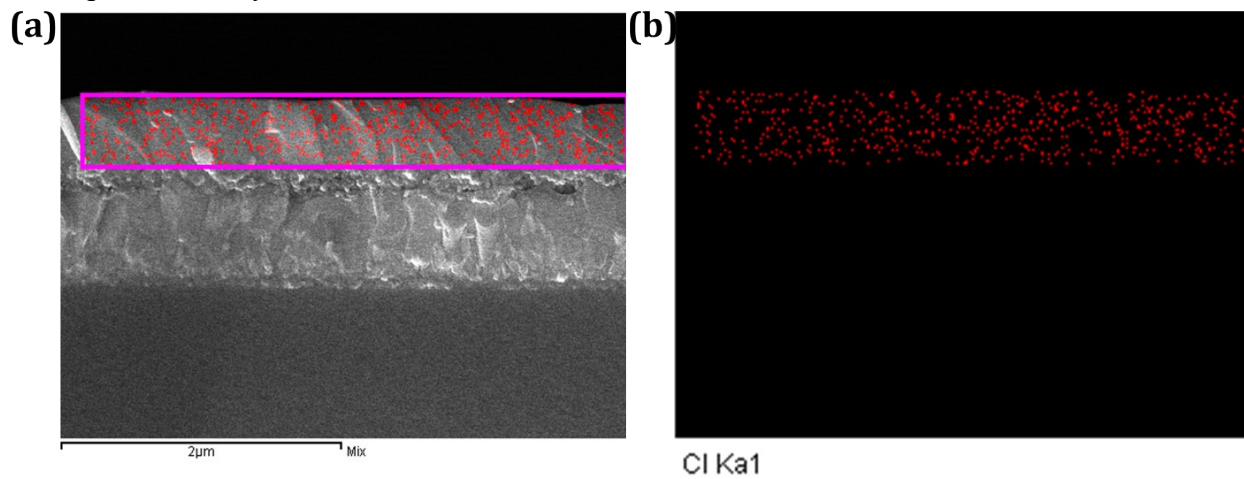


**Figure S5:** Elemental mapping: (a) HAADF STEM image, (b) cesium, (c) lead, (d) iodide, (e) bromide, (f) erbium and (g) chloride.

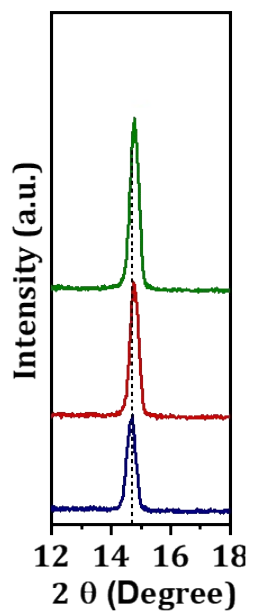




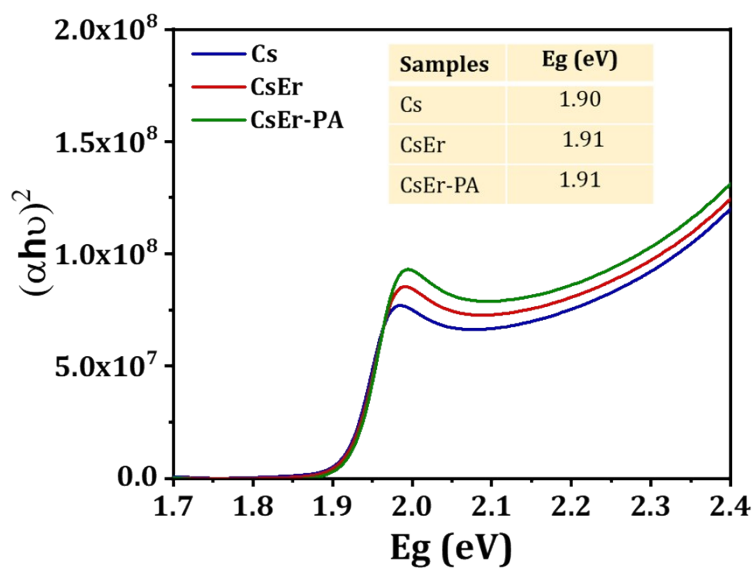
**Figure S6:** (a) Cross-sectional SEM image and (b) elemental mapping of  $\text{ErCl}_3$  doped  $\text{CsPbI}_2\text{Br}$ -based perovskite layer.



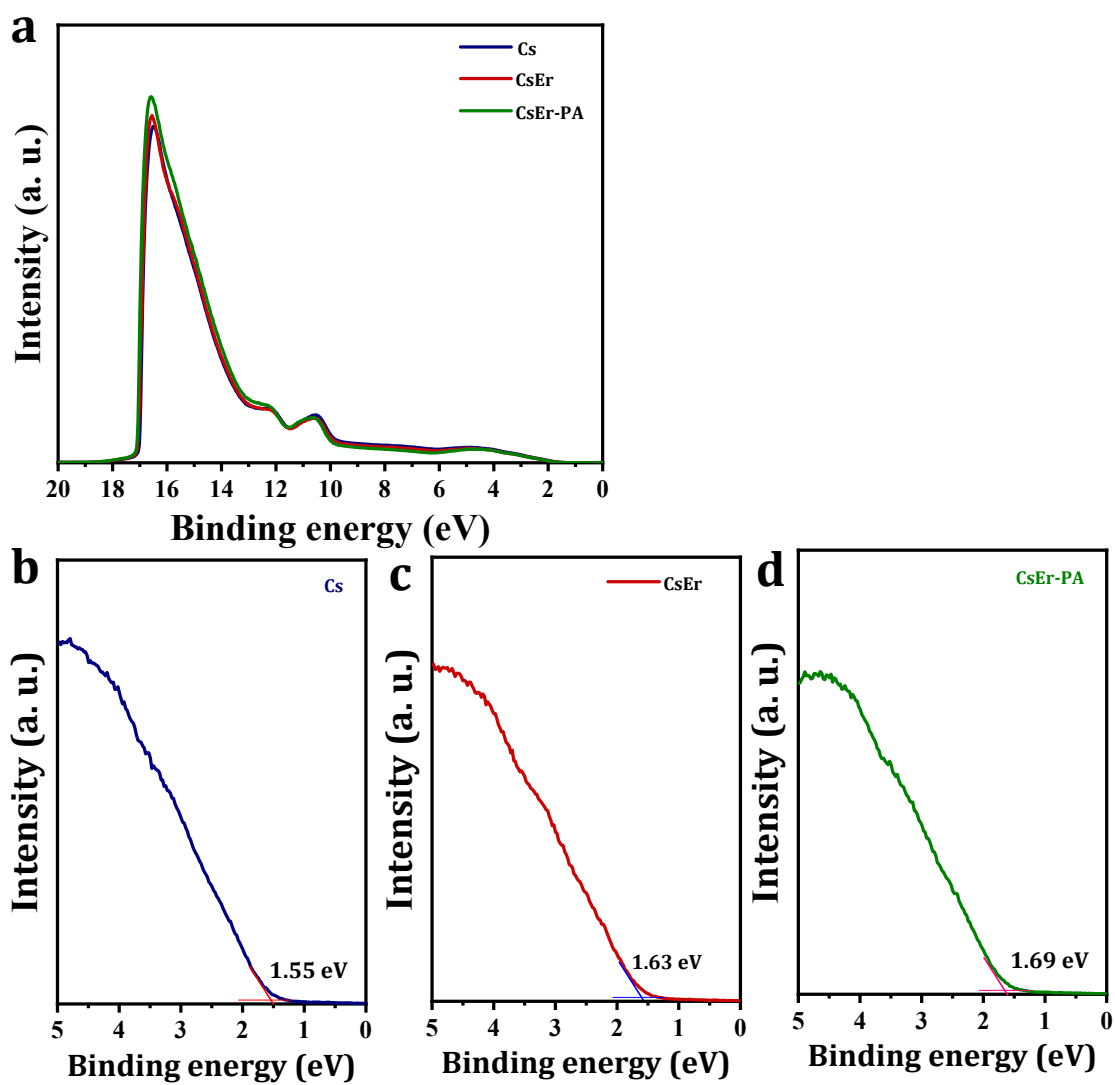
**Figure S7:** Magnified X-ray diffraction (XRD) patterns ( $2\theta=12-18$ ) of Cs, CsEr and CsEr-PA perovskite films.



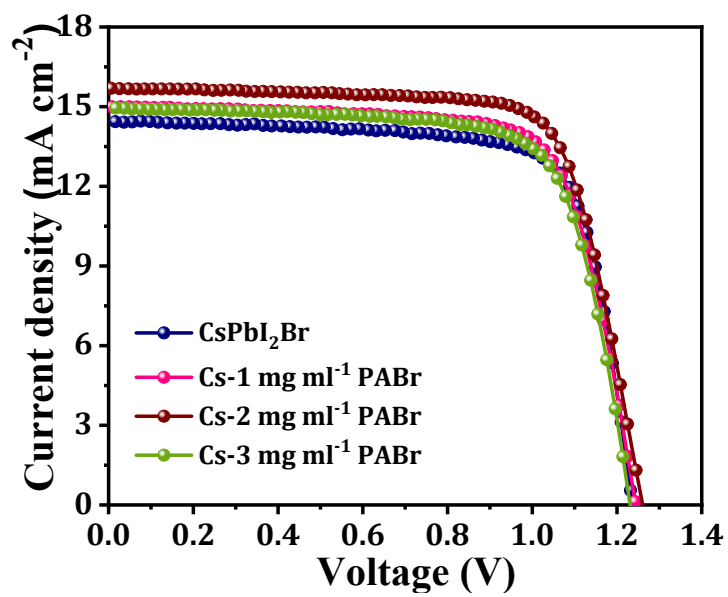
**Figure S8:** Tauc plots of the Cs, CsEr and CsEr-PA perovskite thin films calculated from the UV-Vis absorption spectra.



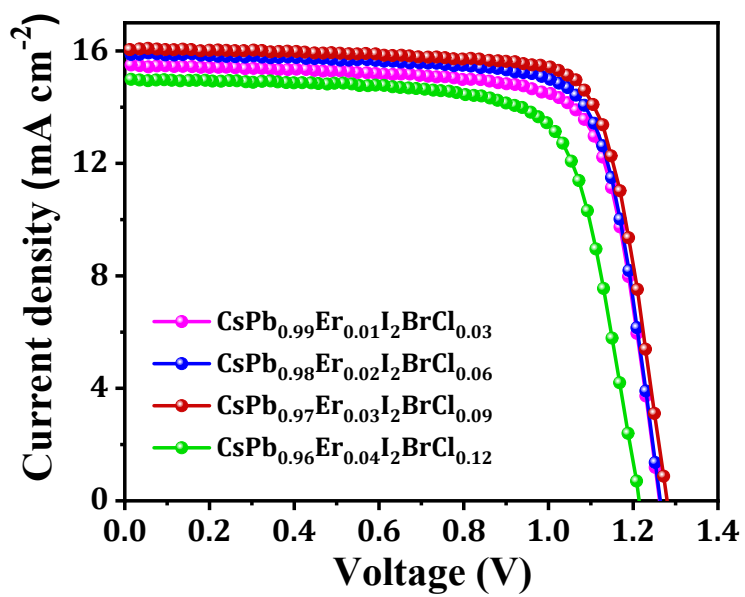
**Figure S9:** (a) UPS survey spectra of perovskite films, valence band region of the (b) Cs, (c) CsEr and (d) CsE-PA perovskites on the mp-TiO<sub>2</sub>/FTO substrate.



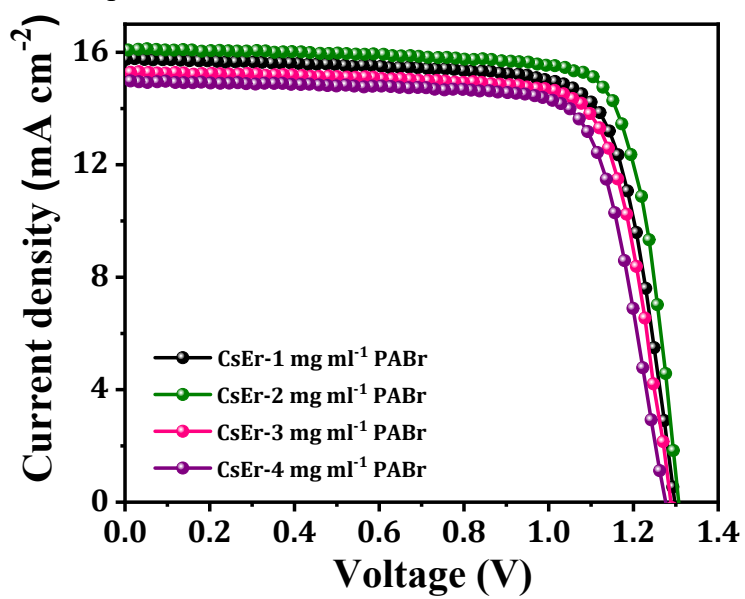
**Figure S10:**  $J$ - $V$  curve of the different amounts of the PABr (1 - 3 mg ml<sup>-1</sup>) additive added CsPbI<sub>2</sub>Br perovskites-based IPVSC devices.



**Figure S11:**  $J$ - $V$  curves of the  $\text{CsPb}_{0.99}\text{Er}_{0.01}\text{I}_2\text{BrCl}_{0.03}$ ,  $\text{CsPb}_{0.98}\text{Er}_{0.02}\text{I}_2\text{BrCl}_{0.06}$ ,  $\text{CsPb}_{0.97}\text{Er}_{0.03}\text{I}_2\text{BrCl}_{0.09}$ ,  $\text{CsPb}_{0.96}\text{Er}_{0.04}\text{I}_2\text{BrCl}_{0.12}$  perovskites-based IPVSC devices.



**Figure S12:**  $J$ - $V$  curves of the CsEr-1 mg ml<sup>-1</sup> PABr, CsEr-2 mg ml<sup>-1</sup> PABr, CsEr-3 mg ml<sup>-1</sup> PABr and CsEr-4 mg ml<sup>-1</sup> PABr perovskites-based IPVSC devices.



**Table S1:** XPS core-level peak positions of the Cs, CsEr and CsEr-PA-based perovskite thin films.

Core level	CsPbI <sub>2</sub> Br			CsEr			CsEr-PA		
	Start BE	Peak BE	End BE	Start BE	Peak BE	End BE	Start BE	Peak BE	End BE
Cs3d5	728.03	<b>724.46</b>	719.97	731.23	<b>724.56</b>	719.31	728.45	<b>724.64</b>	720.24
Cs3d3	741.89	<b>738.41</b>	734.89	744.01	<b>738.5</b>	732.94	742.32	<b>738.61</b>	735.33
Pb4f7	139.99	<b>137.81</b>	134.39	139.99	<b>137.81</b>	134.39	140.64	<b>138.06</b>	134.45
Pb4f5	147.24	<b>142.8</b>	139.99	147.24	<b>142.8</b>	139.99	147.45	<b>143.04</b>	140.64
I3d5	623.2	<b>618.88</b>	613.58	623.07	<b>618.95</b>	613.99	624.52	<b>619.05</b>	613.64
I3d3	633.95	<b>630.35</b>	625.8	634.55	<b>630.42</b>	626.06	635.16	<b>630.53</b>	626.03
Br3d	71.16	<b>68.03</b>	63.8	71.57	<b>68.35</b>	64.77	71.43	<b>68.37</b>	64.81
Er3d	-	-	-	174.19	<b>169.45</b>	167.15	174.93	<b>169.86</b>	166.76



**Table S2:** Different parameters calculated from the XRD of the Cs, CsEr and CsEr-PA-based perovskite thin films.

Perovskites	Main peak position (°)	FWHM (main peak)	$d_{(100)}$ spacing (nm)	Lattice constant (nm)
Cs	14.63	0.180	0.6049	0.6048
CsEr	14.66	0.324	0.6034	0.6033
CsEr-PA	14.67	0.330	0.6032	0.6031

**Table S3:** TRPL measurements of the Cs, CsEr and CsEr-PA-based perovskite thin films deposited on the mp-TiO<sub>2</sub>/FTO substrates.

Perovskites films	$\tau_1$ (ns)	$\tau_2$ (ns)	$\tau_3$ (ns)	$A_1^*$	$A_2^*$	$A_3^*$	$\langle\tau_{avg}\rangle$ (ns)
Cs	17	3.2	0.901	6.5	3.4	0.98	4.19
CsEr	41	7.6	1.6	7.5	2.3	0.149	11
CsEr-PA	103	11	2	5.7	3.6	0.1266	30

**Table S4:** Calculated parameters for energy level of the Cs, CsEr and CsEr-PA-based perovskite films from UPS spectra.

Perovskites	$E_{\text{cutoff}}$ (eV)	WF (eV)	$E_{\text{onset}}$ (eV)	VBM (eV)	$\Delta E_g$ (eV)	CBM (eV)
Cs	17.03	4.19	1.30	5.49	1.90	3.59
CsEr	17.07	4.15	1.31	5.46	1.91	3.55
CsEr-PA	17.11	4.11	1.33	5.44	1.91	3.53

$WF = 21.22 - E_{\text{cutoff}}$ ,  $VBM = WF + E_{\text{onset}}$ ,  $CBM = VBM - E_g$

**Table S5:**  $J-V$  analysis of the CsPbI<sub>2</sub>Br, Cs-1 mg ml<sup>-1</sup> PABr, Cs-2 mg ml<sup>-1</sup> PABr and Cs-3 mg ml<sup>-1</sup> PABr perovskites-based IPVSC devices.

Perovskites	$V_{\text{OC}}$ (V)	$J_{\text{SC}}$ (mA cm <sup>-2</sup> )	FF (%)	PCE (%)
CsPbI <sub>2</sub> Br	1.235	14.45	74.18	13.20
Cs-1 mg ml <sup>-1</sup> PABr	1.241	15.08	74.24	13.89
Cs-2 mg ml <sup>-1</sup> PABr	1.261	15.69	74.42	14.72
Cs-3 mg ml <sup>-1</sup> PABr	1.228	14.97	73.49	13.50

**Table S6:**  $J$ - $V$  analysis of the CsPb<sub>0.99</sub>Er<sub>0.01</sub>I<sub>2</sub>BrCl<sub>0.03</sub>, CsPb<sub>0.98</sub>Er<sub>0.02</sub>I<sub>2</sub>BrCl<sub>0.06</sub>, CsPb<sub>0.97</sub>Er<sub>0.03</sub>I<sub>2</sub>BrCl<sub>0.09</sub> and CsPb<sub>0.96</sub>Er<sub>0.04</sub>I<sub>2</sub>BrCl<sub>0.12</sub> perovskites-based IPVSC devices.

Perovskites	V <sub>OC</sub> (V)	J <sub>SC</sub> (mA cm <sup>-2</sup> )	FF (%)	PCE (%)
CsPb <sub>0.99</sub> Er <sub>0.01</sub> I <sub>2</sub> BrCl <sub>0.03</sub>	1.260	15.49	75.58	14.75
CsPb <sub>0.98</sub> Er <sub>0.02</sub> I <sub>2</sub> BrCl <sub>0.06</sub>	1.262	15.89	76.00	15.24
CsPb <sub>0.97</sub> Er <sub>0.03</sub> I <sub>2</sub> BrCl <sub>0.09</sub>	1.280	16.01	76.79	15.73
CsPb <sub>0.96</sub> Er <sub>0.04</sub> I <sub>2</sub> BrCl <sub>0.12</sub>	1.212	15.00	72	13.08

**Table S7:**  $J$ - $V$  analysis of the CsEr-1 mg ml<sup>-1</sup> PABr, CsEr-2 mg ml<sup>-1</sup> PABr, CsEr-3 mg ml<sup>-1</sup> PABr and CsEr-4 mg ml<sup>-1</sup> PABr perovskites-based IPVSC devices.

Perovskites	V <sub>OC</sub> (V)	J <sub>SC</sub> (mA cm <sup>-2</sup> )	FF (%)	PCE (%)
CsEr-1 mg ml <sup>-1</sup> PABr	1.293	15.89	76.89	15.79
CsEr-2 mg ml <sup>-1</sup> PABr	1.304	16.16	79.44	16.74
CsEr-3 mg ml <sup>-1</sup> PABr	1.284	15.29	75.52	14.82
CsEr-4 mg ml <sup>-1</sup> PABr	1.274	14.96	74.65	14.22

**Table S8:** EIS fitting parameters of the Cs, CsEr and CsEr-PA-based IPVSC devices.

Perovskites	R <sub>S</sub> (Ω)	R <sub>rec</sub> (Ω)
Cs	184.67	5337
CsEr	84.99	6544
CsEr-PA	45.59	7857

**Table S9:** Photovoltaic parameters of the Cs, CsEr and CsEr-PA-based IPVSCs.

Sr. no	CsPbI <sub>2</sub> Br				CsEr				CsEr-PA			
	J <sub>sc</sub> (mAcm <sup>-2</sup> )	V <sub>oc</sub> (V)	FF (%)	PCE (%)	J <sub>sc</sub> (mAcm <sup>-2</sup> )	V <sub>oc</sub> (V)	FF (%)	PCE (%)	J <sub>sc</sub> (mAcm <sup>-2</sup> )	V <sub>oc</sub> (V)	FF (%)	PCE (%)
1	1.105	12.10	67.50	<b>9.02</b>	1.160	14.55	72.89	<b>12.30</b>	1.198	14.77	76	<b>13.44</b>
2	1.130	12.86	68	<b>9.88</b>	1.163	14.58	72.95	<b>12.36</b>	1.205	14.88	76.35	<b>13.68</b>
3	1.135	12.95	68.52	<b>10.07</b>	1.166	14.60	73	<b>12.42</b>	1.210	14.93	76.42	<b>13.80</b>
4	1.143	13.0	68.85	<b>10.23</b>	1.176	14.70	73.25	<b>12.69</b>	1.215	15.05	76.58	<b>14</b>
5	1.154	13.10	68.91	<b>10.41</b>	1.188	14.85	73.76	<b>13.01</b>	1.219	15.10	76.70	<b>14.11</b>
6	1.161	13.29	69	<b>10.64</b>	1.198	15	74	<b>13.29</b>	1.230	15.25	76.86	<b>14.41</b>
7	1.172	13.48	69.35	<b>10.95</b>	1.200	15.02	74.45	<b>13.41</b>	1.245	15.35	77	<b>14.71</b>
8	1.185	13.95	69.49	<b>11.44</b>	1.210	15.15	75	<b>13.74</b>	1.253	15.40	77.35	<b>14.92</b>
9	1.191	14.00	70	<b>11.67</b>	1.221	15.23	75.88	<b>14.11</b>	1.265	15.45	77.40	<b>15.12</b>
10	1.199	14.05	70.85	<b>11.93</b>	1.237	15.52	76.22	<b>14.63</b>	1.277	15.56	77.56	<b>15.41</b>
11	1.209	14.09	71.20	<b>12.12</b>	1.245	15.66	76.41	<b>14.89</b>	1.284	15.71	78	<b>15.83</b>
12	1.222	14.15	72	<b>12.44</b>	1.258	15.75	76.72	<b>15.35</b>	1.286	15.88	78.45	<b>16.02</b>
13	1.228	14.36	73	<b>12.87</b>	1.268	15.81	76.84	<b>15.20</b>	1.292	16	78.56	<b>16.23</b>
14	1.231	14.41	74	<b>13.12</b>	1.276	15.95	76.69	<b>15.60</b>	1.300	16.10	78.71	<b>16.47</b>
15	1.235	14.45	74.18	<b>13.2</b>	1.280	16.01	76.79	<b>15.73</b>	1.304	16.16	79.44	<b>16.74</b>

❖ **TRPL measurement details**

The obtained TRPL lifetime parameters were well fitted with a tri-exponential decay function from equation S1 [S1].

$$I(t) = I_0 + A_1 \exp\left(-\frac{t-t_0}{\tau_1}\right) + A_2 \exp\left(-\frac{t-t_0}{\tau_2}\right) + A_3 \exp\left(-\frac{t-t_0}{\tau_3}\right) \quad \text{--- (S1)}$$

Where,  $\tau_1$ ,  $\tau_2$  and  $\tau_3$  are first, second and third-order decay time,  $A_1$ ,  $A_2$  and  $A_3$  are respective weight factors of each decay channel. Here,  $\tau_1$  stands for the fast decay lifetime and  $\tau_2$  and  $\tau_3$  stand for the slow decay function. The average lifetimes  $\langle \tau_{avg} \rangle$  of Cs, CsEr and CsEr-PA inorganic perovskite thin films were calculated from equation S2 [S1].

$$\langle \tau_{avg} \rangle = \frac{\sum_n A_n \tau_n^2}{\sum_m A_m \tau_m^2} \quad \text{--- (S2)}$$

❖  **$V_{OC}$  as a function of illumination intensity**

The ideality factor ( $\eta$ ) yielded by the slope of the fitted data, which was calculated by equations S3 and S4 [S2].

$$V_{OC} = nkT \ln(I)/q + A \quad \text{----- (S3)}$$

Where,  $k$  = Boltzmann constant,  $T$  = the temperature in Kelvin,  $q$  = the elementary charge and  $A$  is a constant according to the Shockley–Read–Hall (SRH) recombination mechanism [S3].

Equation S3 can be simplified for  $\eta$  as,

$$\eta = \text{slope} \times \frac{q}{kT} \quad \text{----- (S4)}$$

❖ **Charge Transport Studies (Trap density)**

The trap density ( $N_t$ ) of the Cs, CsEr and CsEr-PA perovskite devices can be calculated according to equation S5 via trap-filled limit voltage ( $V_{TFL}$ ) [S4, S5].

$$N_t = \frac{2\varepsilon_0\varepsilon V_{TFL}}{eL^2} \quad \text{----- (S5)}$$

(Where  $\varepsilon_0$  = permittivity of vacuum,  $\varepsilon$  = relative dielectric constant,  $e$  = elementary charge, and  $L$  = perovskite film thickness).

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- [S5] J. Zhu, B. He, Z. Gong, Y. Ding, W. Zhang, X. Li, Z. Zong, H. Chen and Q. Tang, Grain enlargement and defect passivation with melamine additives for high efficiency and stable  $\text{CsPbBr}_3$  perovskite solar cells, *ChemSusChem*, 2020, **13**, 1834–1843.