

Supplementary Information

Cesium Manganese Halide Perovskite-Analogue Nanocrystals with Highly Efficient Energy Conversion for Flexible Multifunctional Fibers

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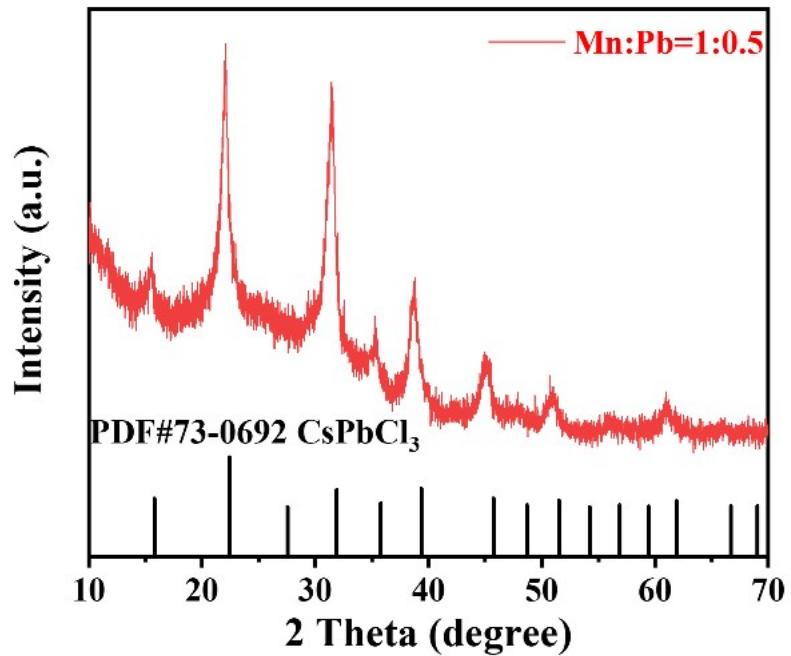


Fig. S1 XRD pattern of Pb²⁺-incorporated Mn-based PA NCs with Mn:Pb=1:0.5.

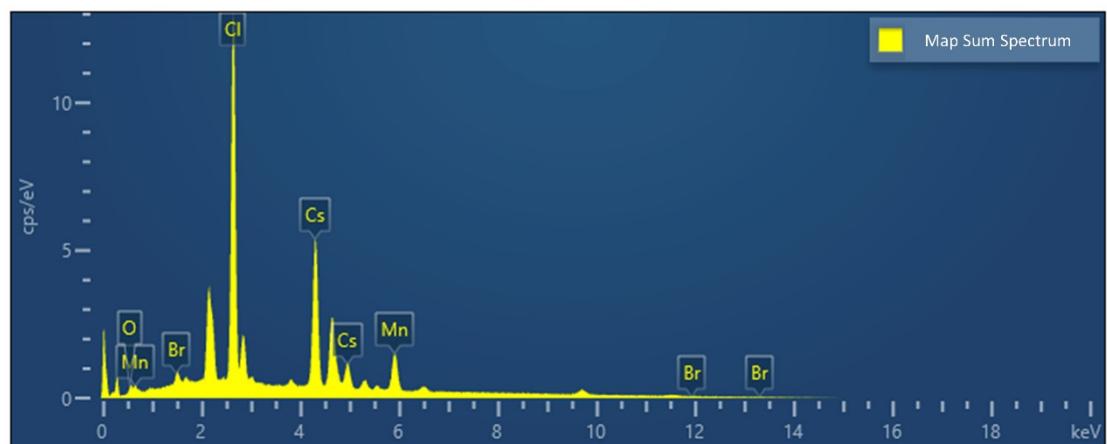


Fig. S2. EDS spectrum recording from Cs₂Mn(Cl/Br)₄·2H₂O PA NCs, showing the existence of Cs, Mn, Cl, Br and O elemental signals.

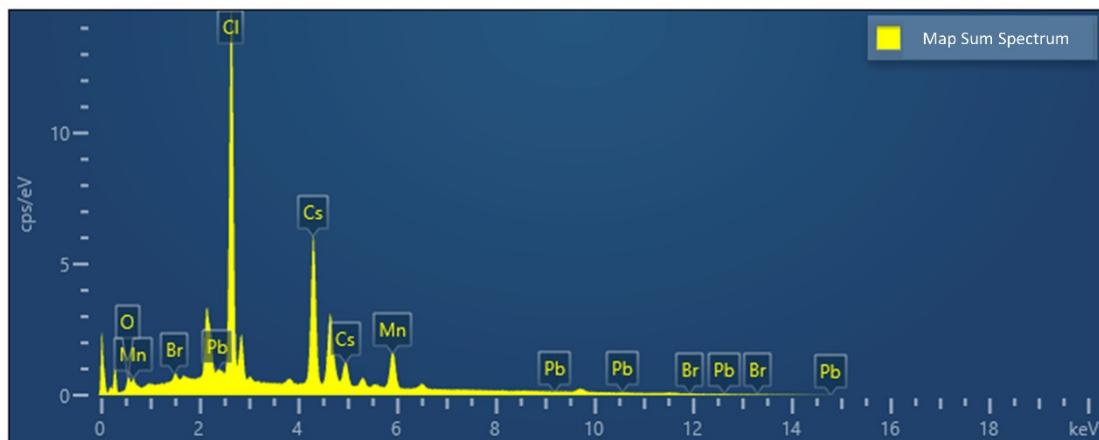


Fig. S3. EDS spectrum recording from Pb^{2+} doped $\text{Cs}_2\text{Mn}(\text{Cl}/\text{Br})_4 \cdot 2\text{H}_2\text{O}$ PA NCs ($\text{Mn}:\text{Pb} = 1:0.05$), showing the existence of Cs, Pb, Mn, Cl, Br and O elemental signals.

Table. S1 Ratio of elements in the products ($\text{Mn}:\text{Pb} = 1:0.05$) and Pristine detected by EDS.

Sample	EDS (atomic ratio %)						
	(Mn:Pb)	Cs	Mn	Pb	Cl	Br	O
Pristine		30.82	15.30	/	51.23	1.35	1.30
1:0.05		30.14	14.73	0.24	50.43	0.84	3.62

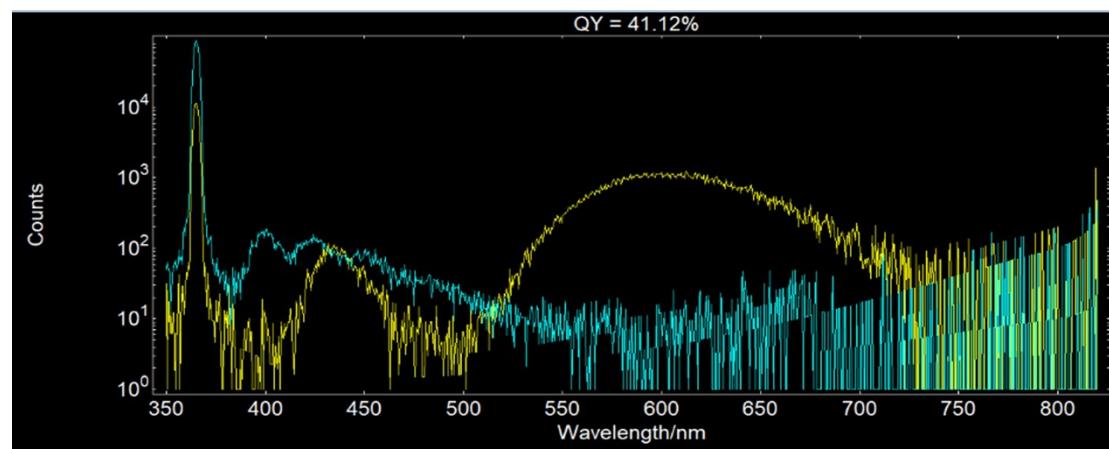


Fig. S4. Absolute PLQY of Pb^{2+} doped $\text{Cs}_2\text{Mn}(\text{Cl}/\text{Br})_4 \cdot 2\text{H}_2\text{O}$ PA NCs ($\text{Mn}:\text{Pb} = 1:0.40$) at 598 nm.

Table. S2 Summary of Fitting Parameters for TRPL Profiles of PA NCs with Different Mn–Pb Ratios Using a Double-exponential Function for Mn²⁺ emission.

Sample (Mn:Pb)	τ_1 (ns)	τ_2 (ns)	τ_{ave} (ns)	PLQY of Mn ²⁺
1:0.05	0.36	4.82	3.26	2.84%
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Sample (Mn:Pb)	τ_1 (μ s)	τ_2 (μ s)	τ_{ave} (μ s)	PLQY of Mn ²⁺
1:0.10	363.32	1617.25	1491.24	7.31%
1:0.15	488.49	1473.99	1367.47	26.14%
1:0.20	713.39	1465.63	1327.80	33.53%
1:0.40	717.27	1291.52	1081.71	41.12%

Table. S3 Manganese-based perovskites.

formula	method	temperature (°C)	peak wavelength (nm)	PLQY (%)	refs
CsMnCl ₃	hot-injection	140	660	0.7	1
CsMnBr ₃	hot-injection	170	643	54	2
Cs ₃ MnBr ₅	hot-injection	200	520	48	3
CsMn _{1-x} Pb _x (Cl/Br) ₃ ·2H ₂ O (x<50%)	supersaturation recrystallization	room temperature	≈600	41.8	4
Cs ₂ Mn _{1-x} Pb _x (Cl/Br) ₄ ·2H ₂ O	hot-injection	150	≈598	41.12	this work

References

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