

Electronic Supplementary Information

Metal-Organic-Framework Embellished through Ion Etching Method for Highly Enhanced Electrochemical Oxygen Evolution

Qiuxiang Mou,^a Zhenhang Xu,^b Wei Zuo,^b Tianyu Shi,^b Erlei Li,^a Gongzhen Cheng,^b Xinghai Liu,^a
Huaming Zheng,^{*c} Houbin Li,^{*a} and Pingping Zhao^{*a}

^a *Research Center of Graphic Communication, Printing and Packaging, Wuhan University, Wuhan, 430072, China.*

^b *College of Chemistry and Molecular Sciences, Wuhan University, Wuhan, 430072, China.*

^c *School of Material Sciences & Engineering, Wuhan Institute of Technology, Wuhan, 430073 Hubei, P.R. China.*

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1. Supplementary Figures

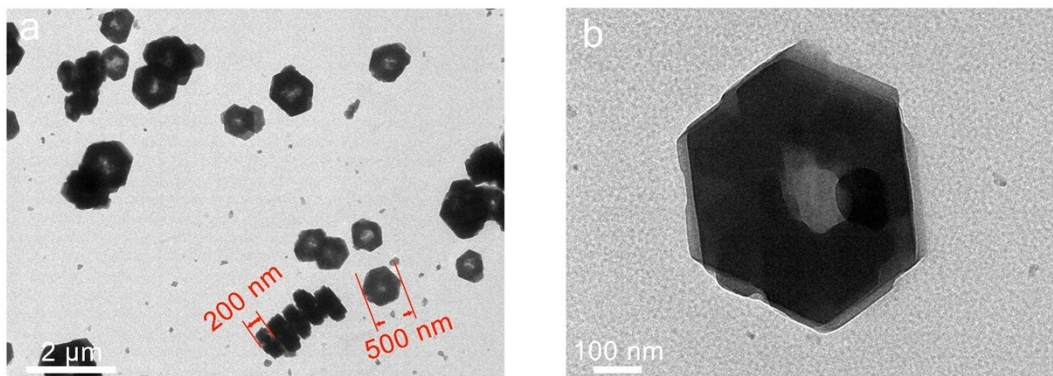


Fig. S1. TEM images of Ni MOF.

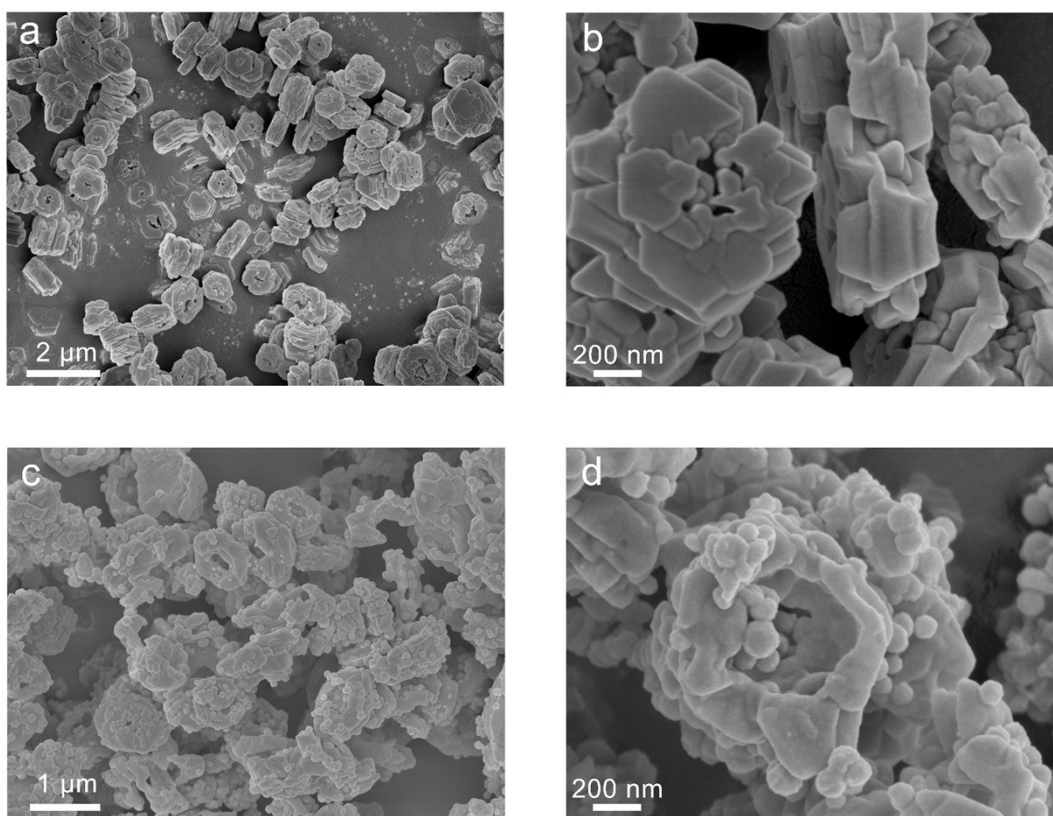


Fig. S2. SEM images of (a) (b) Ni MOF and (c) (d) Ni-MOF-Fe-2.

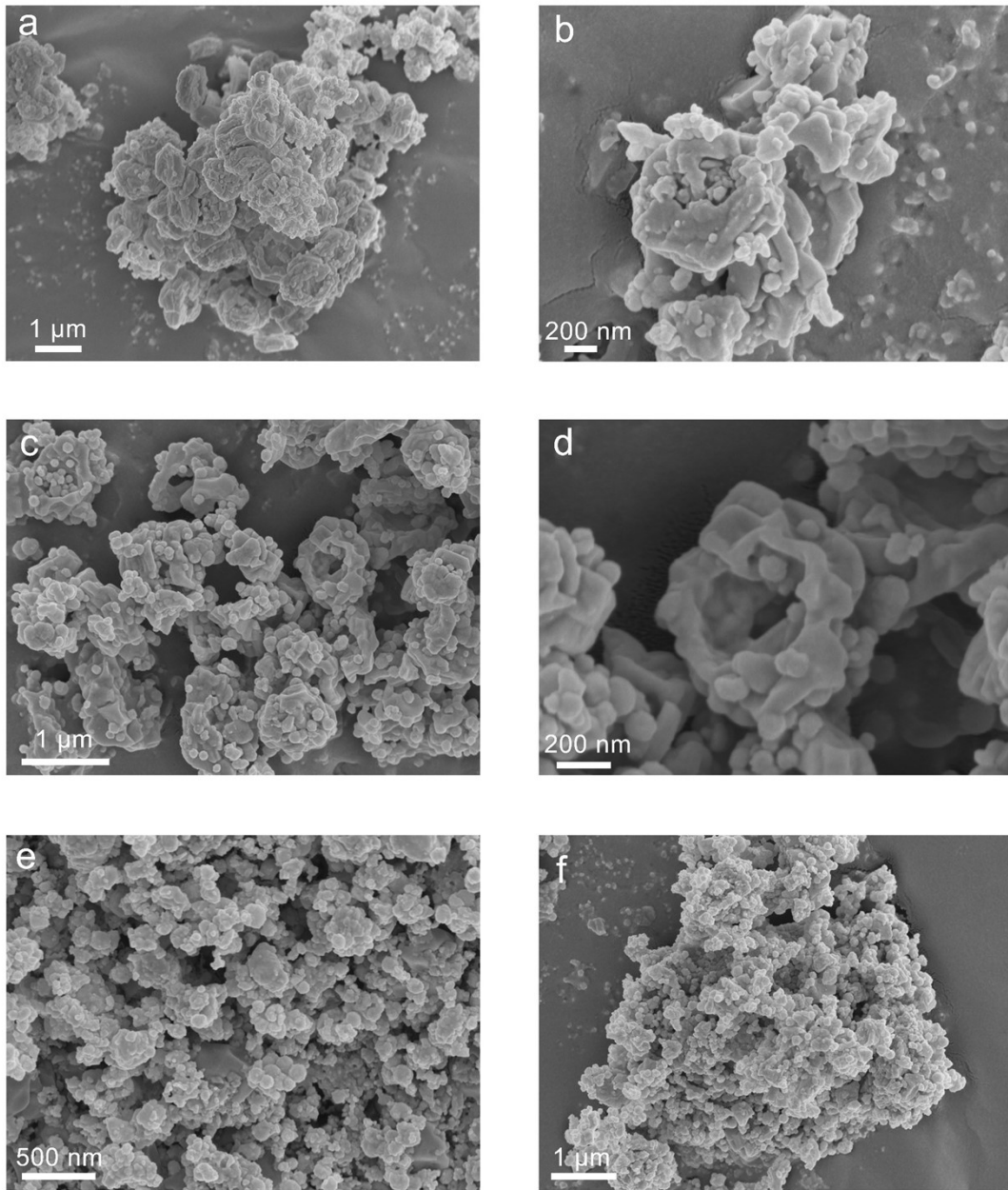


Fig. S3. SEM images of (a) (b) Ni-MOF-Fe-1, (c) (d) Ni-MOF-Fe-2 and (e) (f) Ni-MOF-Fe-3.

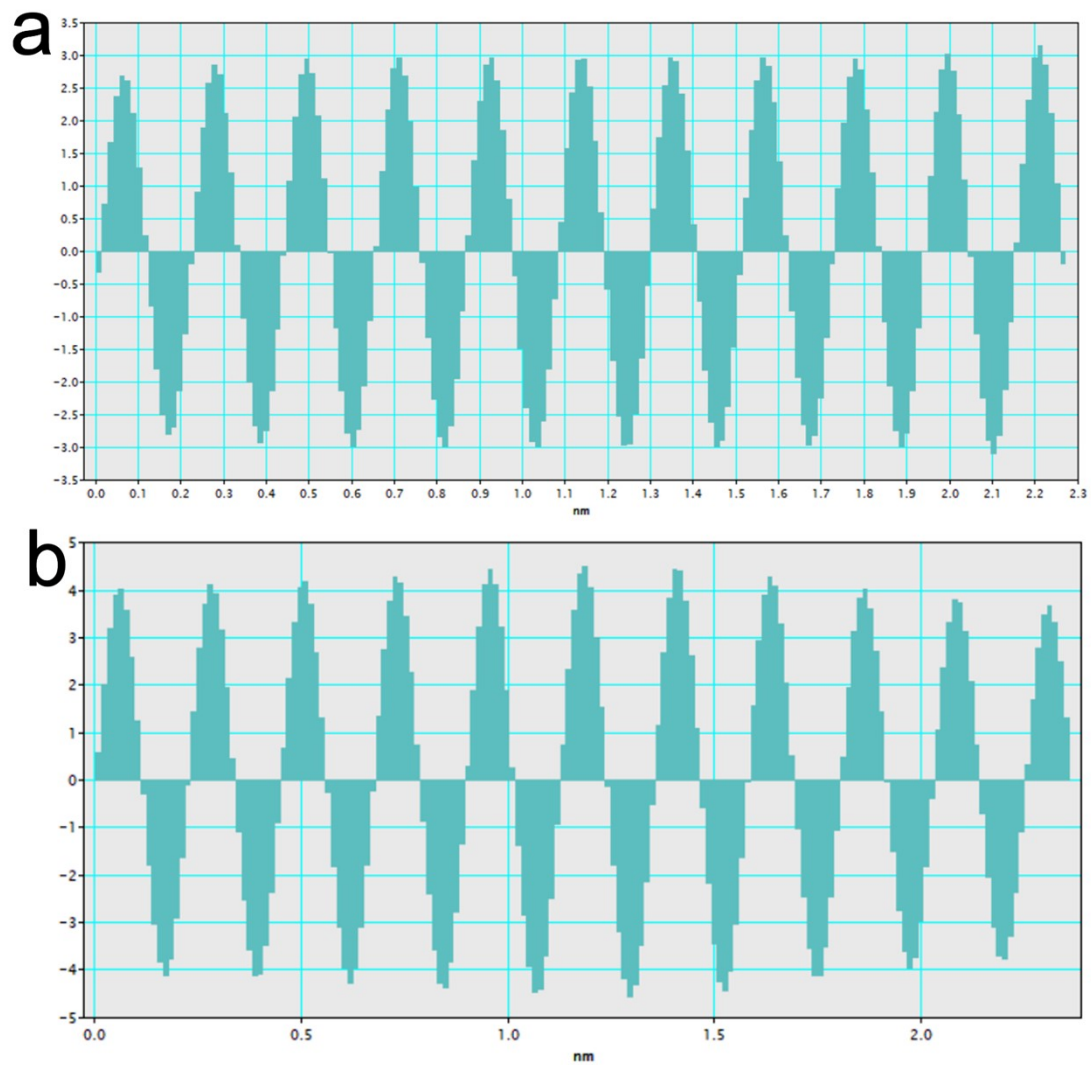


Fig. S4. Contrast intensity profile of the (a) (-121) plane and (b) (-120) plane.

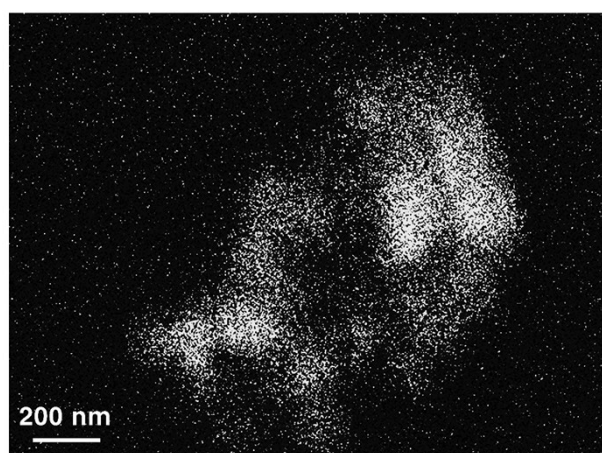


Fig. S5. HAADF-STEM image of Ni-MOF-Fe-2.

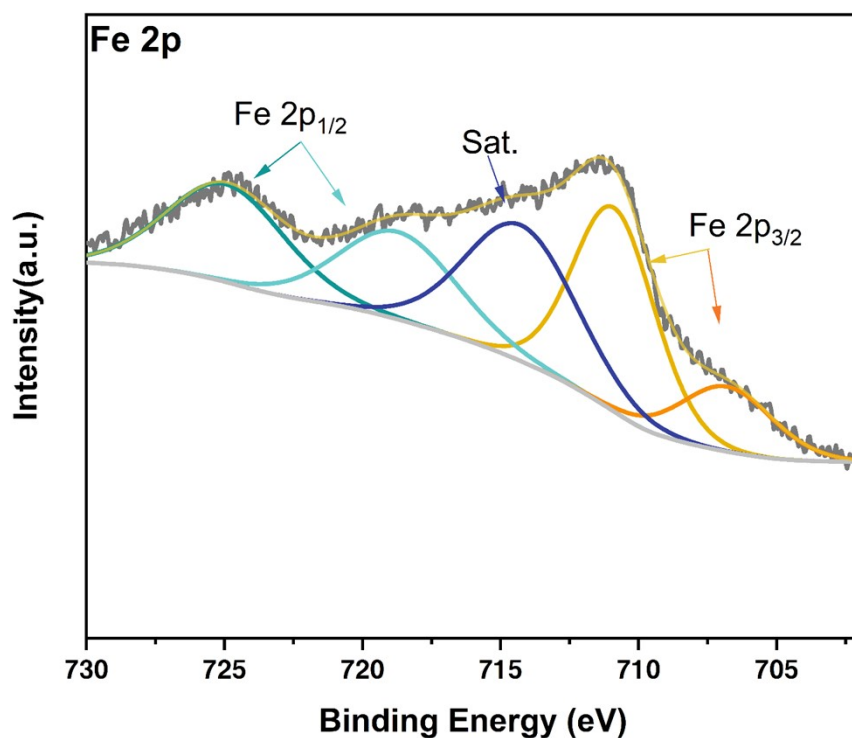


Fig. S6. High-resolution XPS spectrum of Fe 2p in Ni-MOF-Fe-2.

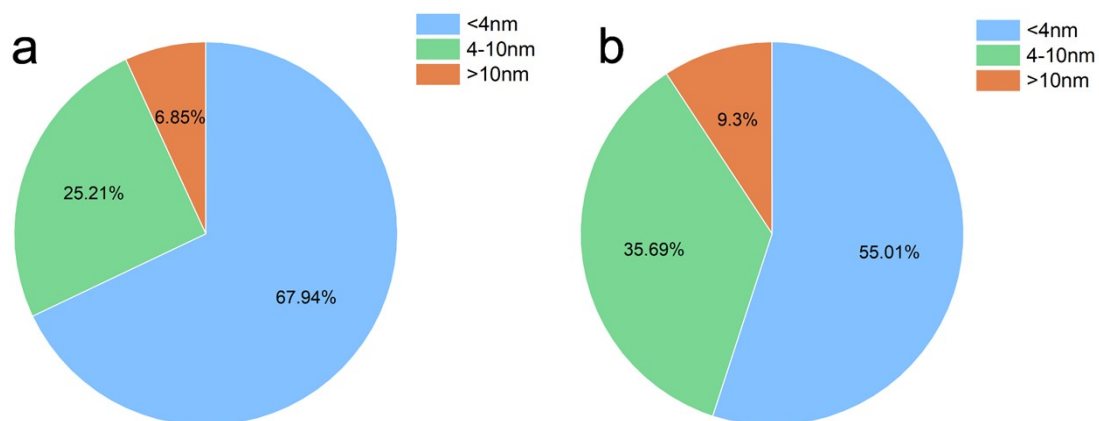


Fig. S7. Pie charts of pore width in (a) Ni MOF and (b) Ni-MOF-Fe-2, where the pore width proportion of Ni-MOF-Fe-2 showed a wider distribution than that of Ni MOF, and more big pores generated in Ni-MOF-Fe-2, intuitively indicating that the structure of MOF etched by iron ions has been effectively changed.

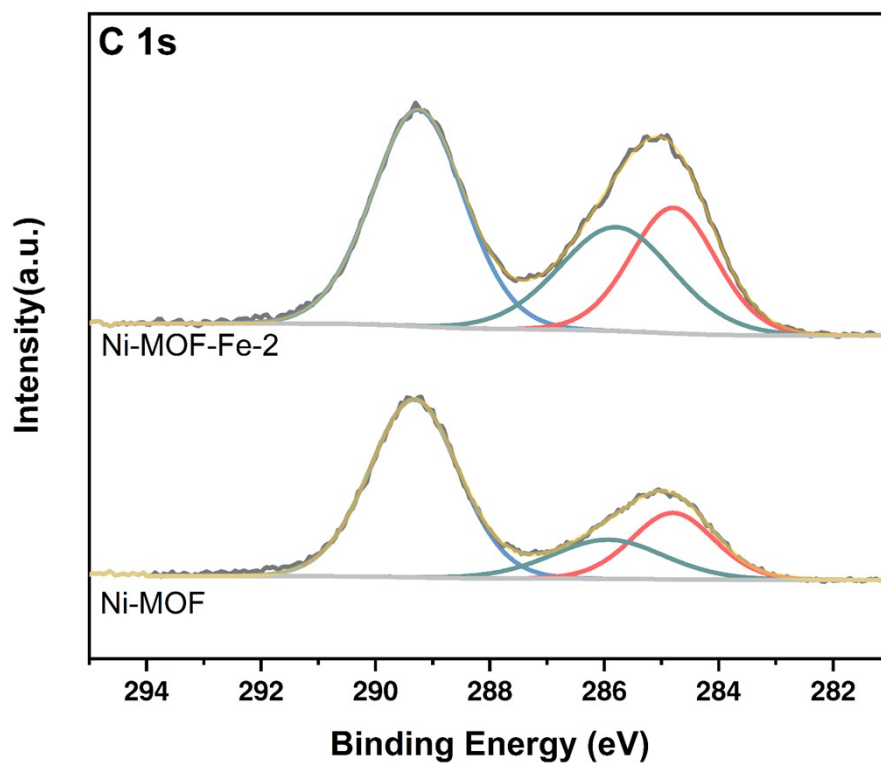


Fig. S8. High-resolution XPS spectra of C 1s in Ni MOF and Ni-MOF-Fe-2.

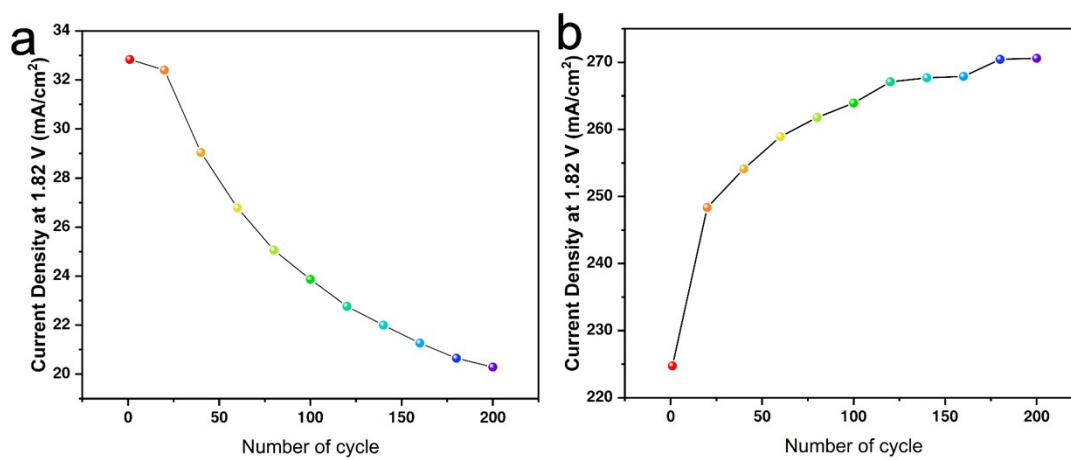


Fig. S9. OER activity at 1.66 V (vs. RHE) for (a) Ni MOF and (b) Ni-MOF-Fe-2.

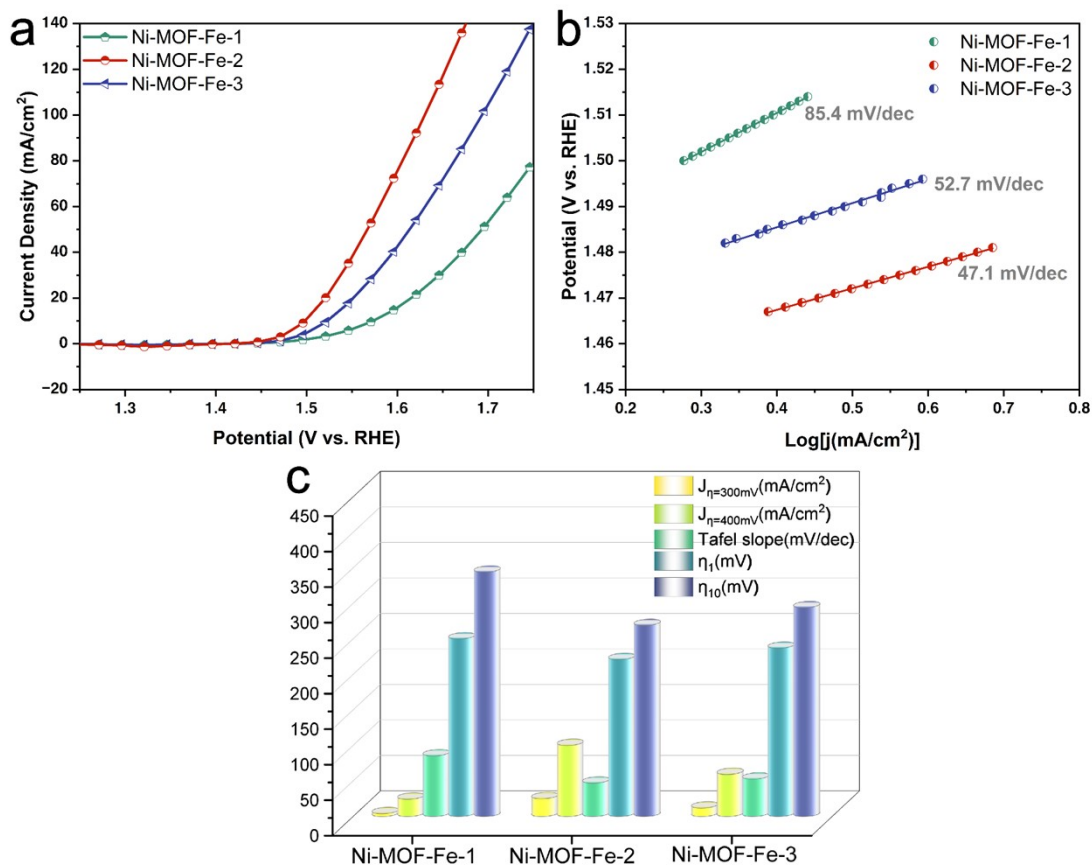


Fig. S10. OER activity of the Fe etching catalysts. (a) LSV OER polarization curves of the catalysts without iR -correction. (b) Corresponding Tafel slopes in (a). (c) Comparison of the overpotentials at 1 mA/cm² (η_1) and 10 mA/cm² (η_{10}), the current densities under an overpotential of 300 mV and 400 mV, and the Tafel slopes obtained from the catalysts.

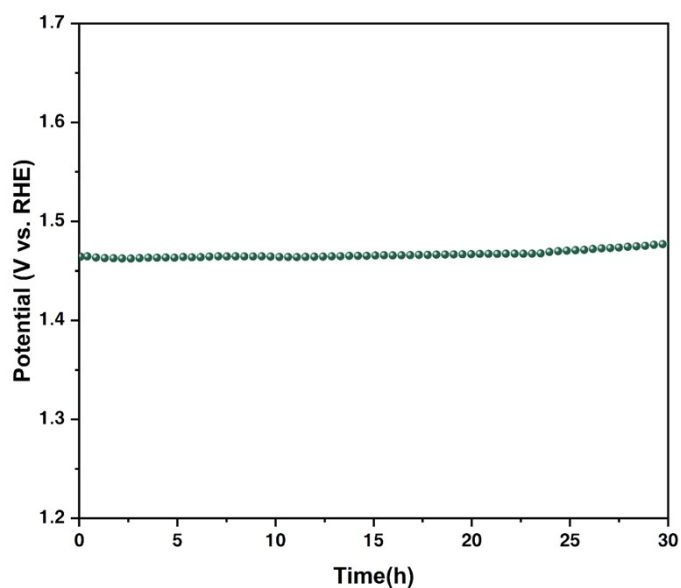


Fig. S11. Chronopotentiometric curve of Ni-MOF-Fe-2 at 10 mA/cm².

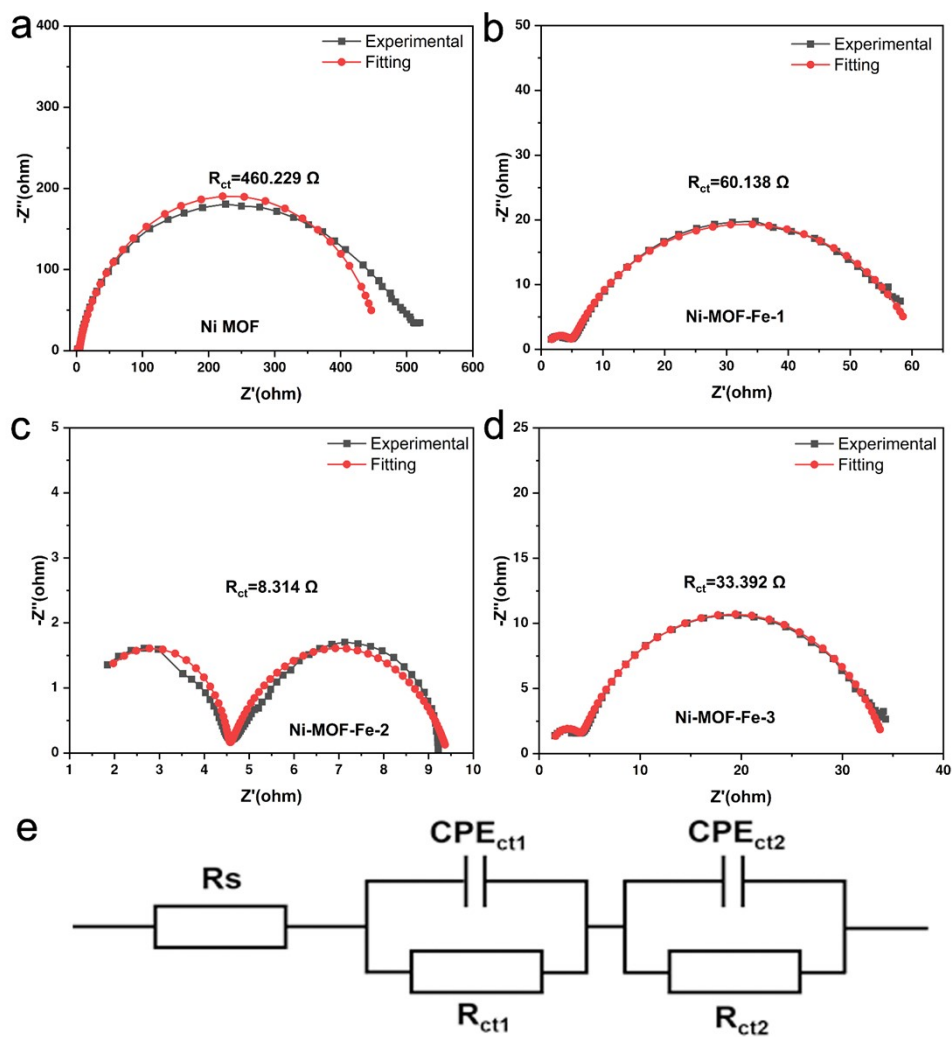


Fig. S12. Experimental data and fitting results of (a) Ni MOF, (b) Ni-MOF-Fe-1, (c) Ni-MOF-Fe-2 and (d) Ni-MOF-Fe-3. (e) The equivalent circuit for modeling the measured electrochemical response in this manuscript.

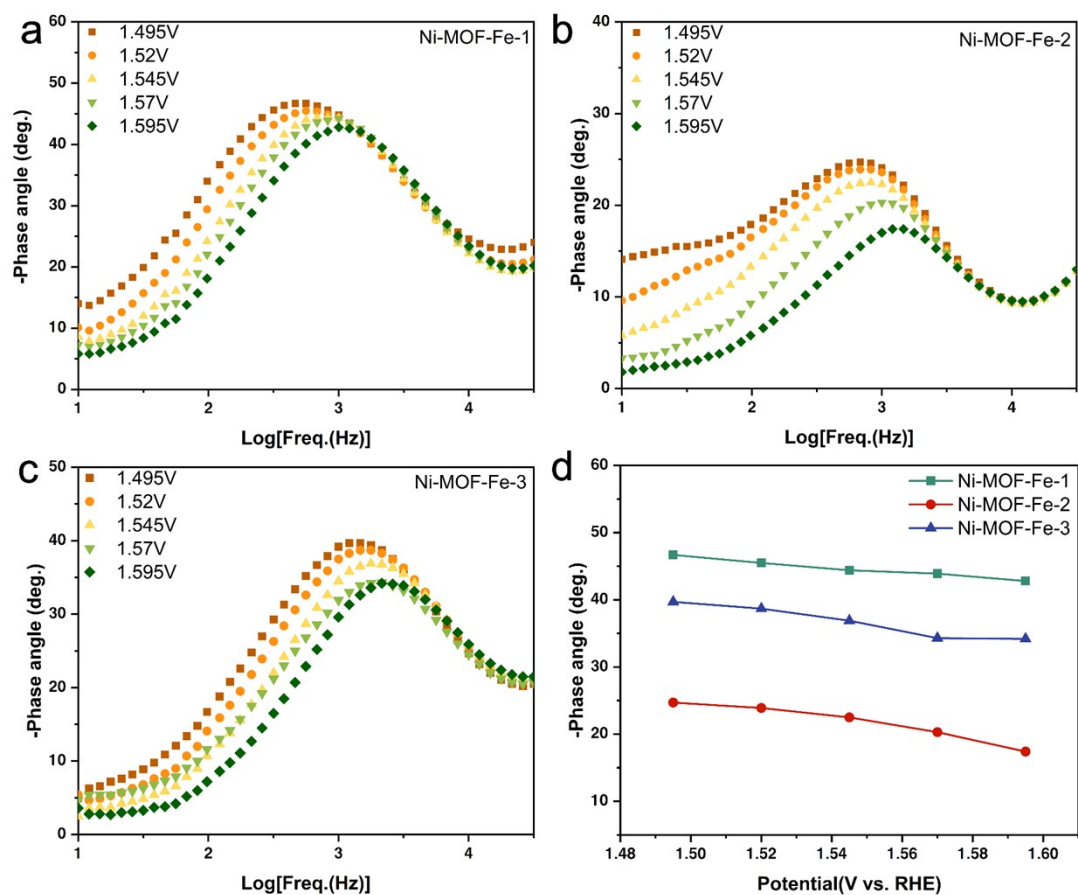


Fig. S13. Bode plots of (a) Ni-MOF-Fe-1, (b) Ni-MOF-Fe-2 and (c) Ni-MOF-Fe-3 obtained under different potentials. (d) Potential dependence of the phase angle of the samples.

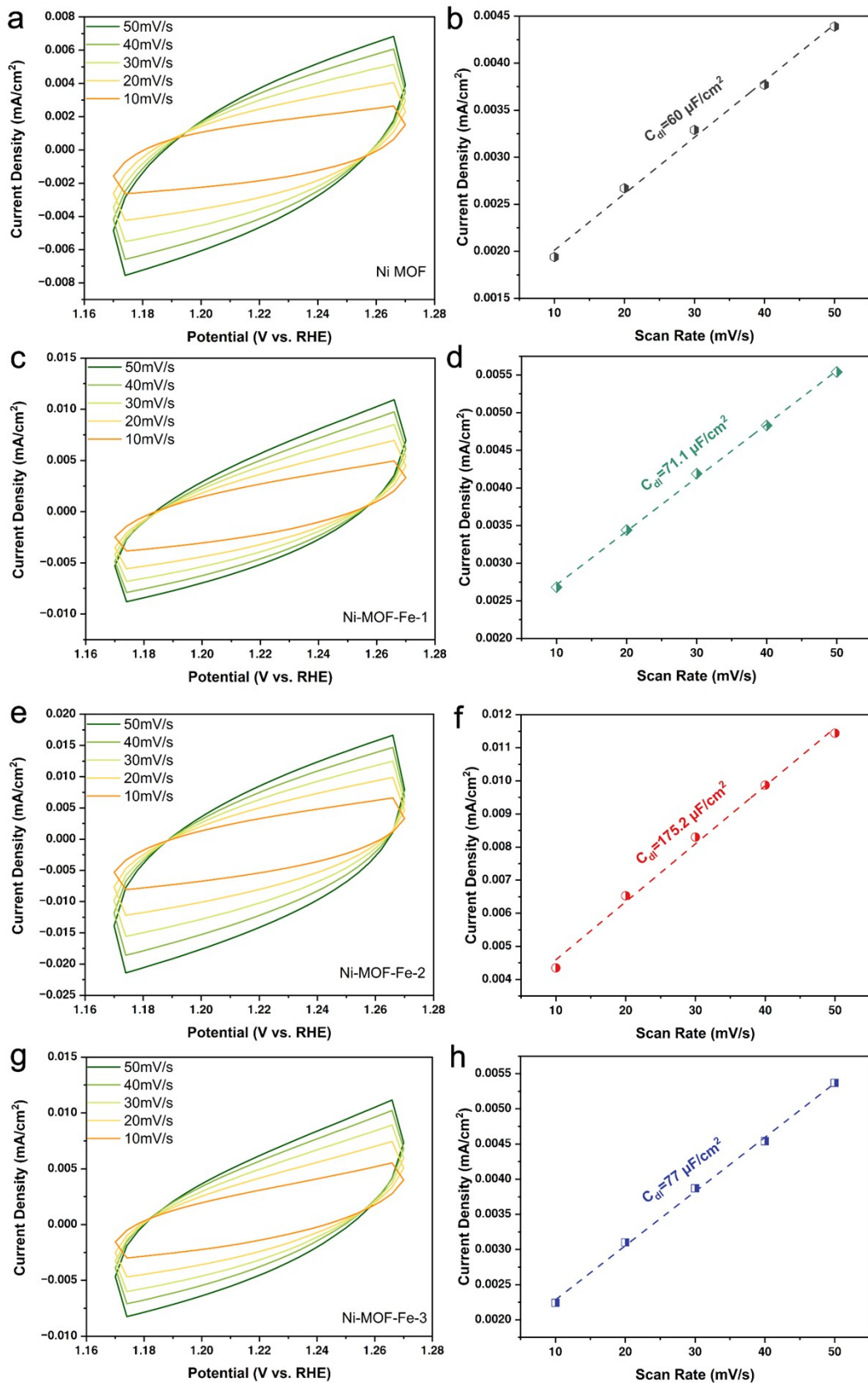


Fig. S14. CV curves and current density differences at 1.222 V vs. RHE against the scan rate for the estimation of C_{dl} of the samples. (a) (b) Ni-MOF, (c) (d) Ni-MOF-Fe-1, (e) (f) Ni-MOF-Fe-2. (g) (h) Ni-MOF-Fe-3.

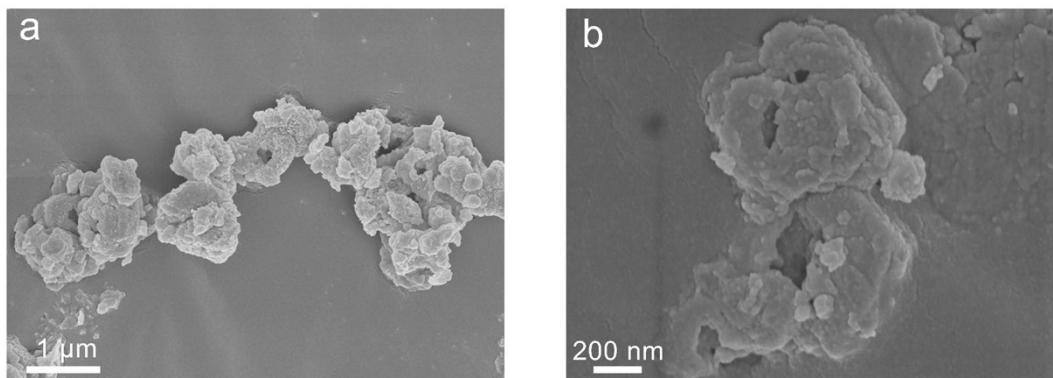


Fig. S15. SEM images of the post-OER Ni-MOF-Fe-2.

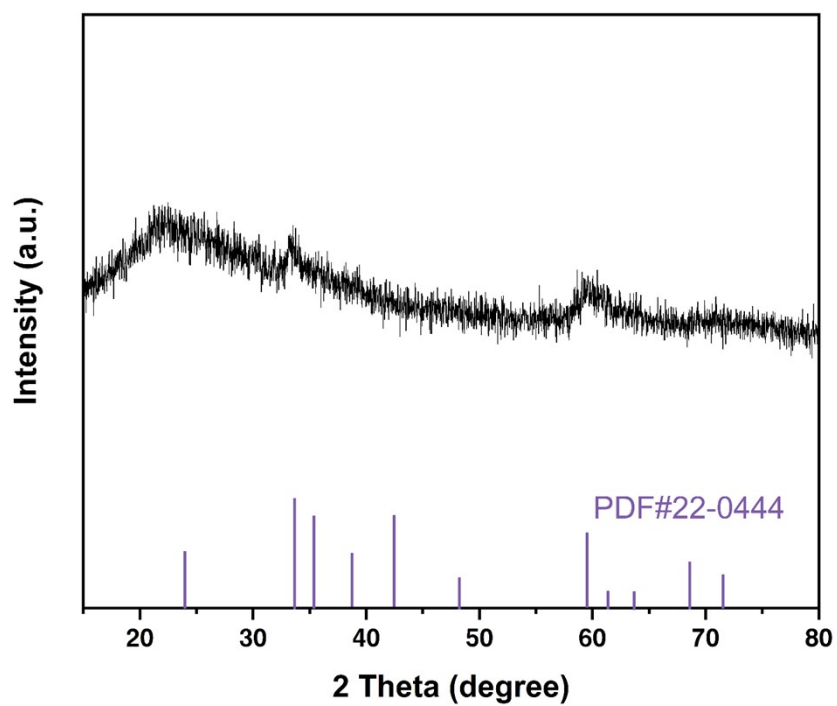


Fig. S16. XRD pattern of the post-OER Ni-MOF-Fe-2.

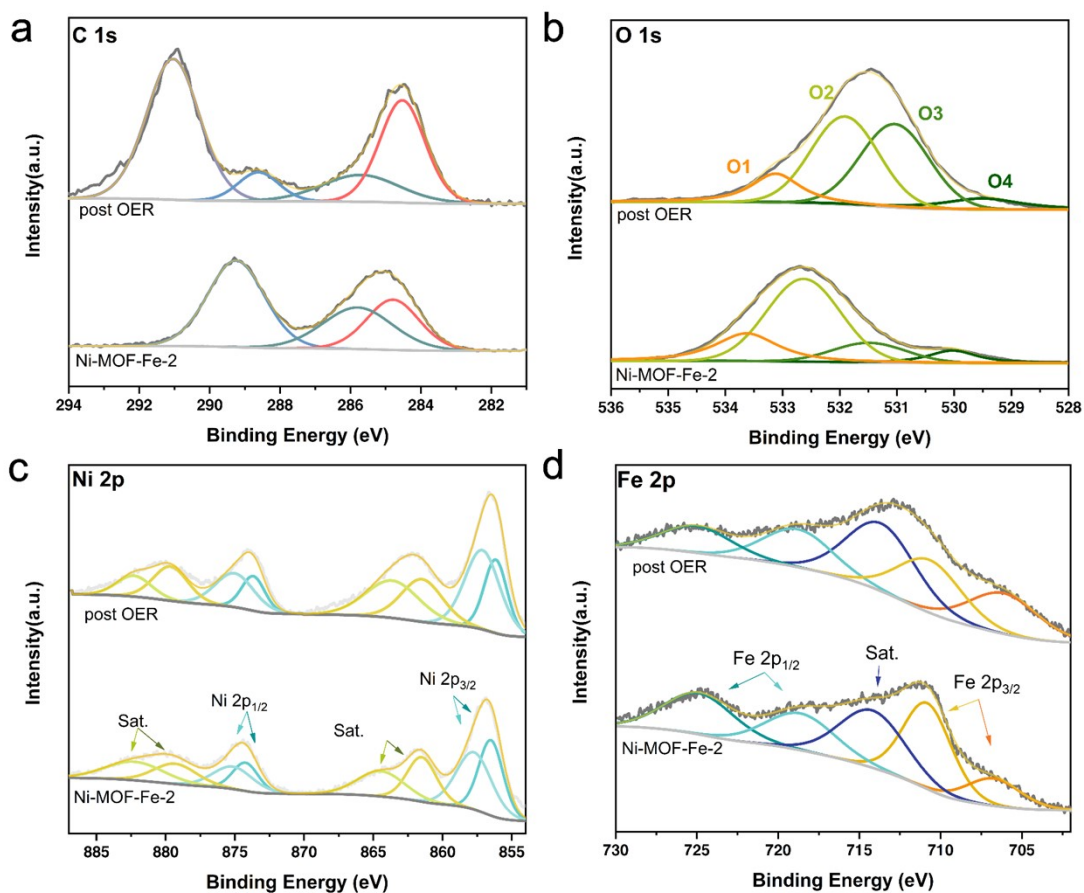


Fig. S17. High-resolution XPS spectra of (a) C 1s, (b) O 1s (c) Ni 2p and (d) Fe 2p of Ni-MOF-Fe-2 before and after OER process.

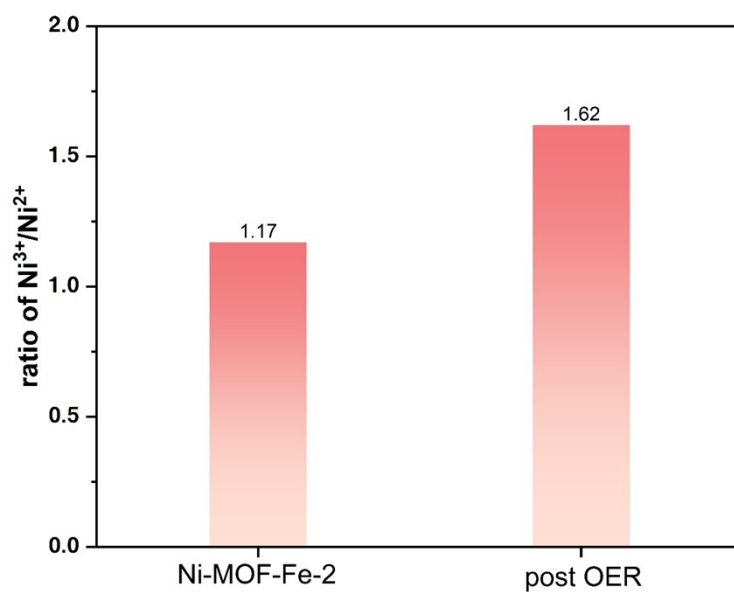


Fig. S18. Ratio of Ni³⁺/Ni²⁺ in Ni-MOF-Fe-2 before and after OER process.

2. Supplementary Tables

Table S1. EDX results of Ni-MOF-Fe-2.

Element	Weight(%)	Atomic(%)	Net Error(%)
O K	37.01	67.82	0.94
Fe K	28.46	14.94	1.05
Ni K	34.53	17.24	0.17

Table S2. Comparison of OER performance of catalysts in this work with reported electrocatalysts.

Catalyst	η (mV) at 10 mA cm ⁻²	Tafel Slope (mV dec ⁻¹)	Electrolyte	Loading mass (μ g cm ⁻²)	Reference
Ni-MOF-Fe-2	269	47.1	1 M KOH	300	This work
Ni-MOF-Fe-1	344	85.4	1 M KOH	300	
Ni-MOF-Fe-3	294	52.7	1 M KOH	300	
Ni SAs@S/N-CMF	285	50.8	1 M KOH	750	1
BCN/ZrO ₂	301	75	1 M KOH	141	2
Ru@NiV-LDH	272	60	1 M KOH	100	3
FeNi/(FeNi) ₉ S ₈ /N	283	64.03	1 M KOH	420	4
CoFe-LDH/GP	252	61	1 M KOH	6500	5
3Co-LaMOH O _V @NC	330	129	0.1 M KOH	500	6
CoNi-SAs/NC	340	58.7	1 M KOH	1400	7
CO-Fe ₂ O ₃	439	99	0.1 M KOH	784	8
NCMC	290	73	1 M KOH	280	9
Co/MoC@N-C	290	90	1 M KOH	455	10
Ni ₃ S ₂	295	52	1 M KOH	600	11
NiFeZr MOFs-0.12	288	66	1 M KOH	250	12
CoP NFs	323	49.6	1 M KOH	265	13
Ni-MOF@CNT	370	138.2	1 M KOH	1000	14
CoP/NCNHP	310	70	1 M KOH	400	15
NiCu-MOF	290	169	1 M KOH	2000	16
Co ₃ S ₄ @MoS ₂	280	43	1 M KOH	283	17
NiFe-HXR	289	43	1 M KOH	150	18
Fe ₂ /Co ₁ GNCL	350	70	1 M KOH	255	19
NiFe MOF OM NFH	270	123	1 M KOH	400	20
W ₂ N/WC	320	94.5	1 M KOH	200	21
Mo-Co ₃ O ₄ /CNTs	280	63	1 M KOH	250	22
CoFeBiP	273	77.3	1 M KOH	300	23
VC-MOF-Fe	339	78	1 M KOH	500	24
NiPc-NiFe _{0.09}	300	55	1 M KOH	285	25
Fe-Co ₃ O ₄	262	43	1 M KOH	250	26

ZIF-9(III)/Co LDH-15	297	65	1 M KOH	189	27
Co ₂ P/CoNPC	326	72.6	1 M KOH	390	28
MOF3_A	424	--	0.5 M KOH	125	29
Co ₂ FeO ₄ /NCNTs	420	96.7	0.1 M KOH	200	30
aMOF-NC	249	39.5	1 M KOH	250	31
Co _{1.5} Fe _{0.5} P	278	57	1 M KOH	250	32
FeCo(OH) _x -30	295	59	1 M KOH	255	33
O-CoMoS	272	45	1 M KOH	1000	34
(Co,0.3Ni-HMT)	330	66	1 M KOH	182	35

Table S3. EIS parameters of the catalysts.

catalysts	R _s (Ω)	R _{ct1} (Ω)	CPE _{ct1} (mF)	R _{ct2} (Ω)	CPE _{ct2} (mF)	R _{ct} (Ω)
Ni MOF	1.237	2.829	1.119	457.4	0.88477	460.23
Ni-MOF-Fe-1	0.95117	3.748	0.98628	56.39	0.76604	60.14
Ni-MOF-Fe-2	1.111	3.436	0.95588	4.878	0.74285	8.31
Ni-MOF-Fe-3	1.166	2.612	1.097	30.84	0.77259	33.39

Note: R_{ct} is equal to R_{ct1} plus to R_{ct2}.

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