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Supporting information

Efficient solution-processable non-doped hybridized local and chargetransfer (HLCT) emitter for a simplified organic light-emitting diode

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Synthesis and characterization



Scheme 1 Synthesis of CPBzFC. Reagents and conditions: i) (4-(carbazol-N-yl)phenyl)boronic acid,

Pd(PPh₃)₄, 10% K₂CO₃ (aq), THF, reflux; ii) bispinacolatodiboron, KOAc, Pd(dppf)₂Cl₂, toluene, reflux; iii) 1,8-dibromooctane, *tert*-BuNH₄Br, KOH, DFM, H₂O; iv) 3,6-di-*tert*-butylcarbazole, KOH, DMF; v) 3,6-di-*tert*-butylcarbazole, CuI, K₃PO₄, (±)-*trans*-1,2-diaminocyclohexane, toluene, reflux; vi) Pd(PPh₃)₄, 10% K₂CO₃ (aq), THF, reflux.

4-(4-(Carbazol-9-yl)phenyl)-7-bromobenzothiadiazole (2)

4,7-Dibromobenzothiadiazole (3) (2.0 g, 7.0 mmol) and (4-(carbazol-9-yl)phenyl)boronic acid (4) (1.0 g, 3.5 mmol) were dissolved in dried THF (75 mL) under nitrogen atmosphere. 10% K₂CO₃ (aq) (10 ml) was added, stirred the mixture was at ambient temperature. Tetrakris(triphenylphosphene)palladium(0) $Pd(PPh_3)_4$ (0.2 g, 0.17 mmol) was then added to reaction flask and degassed for 10 minutes. The reaction mixture was heated to reflux for 16 hours. After completion, allowed to room temperature, water was added to reaction mixture and extracted by dichloromethane (3 x 50 mL). The organic layer was washed with brine (3 x 20 mL), dried (Na₂SO₄) and concentrated *in vacuo*. The product was purified by column chromatography on silica gel eluting with dichloromethane:hexane (1:2) to give yellow solids (1.1 g, 71%). ¹H NMR (600 MHz, CDCl₃) δ 8.20 – 8.15 (4H, m, Ar-*H*), 8.00 (1H, d, *J* = 7.5 Hz, Ar-*H*), 7.76 (2H, d, *J* = 8.4 Hz, Ar-*H*), 7.71 (1H, d, *J* = 7.5 Hz, Ar-*H*), 7.55 (2H, d, *J* = 8.2 Hz, Ar-*H*), 7.44 (2H, t, *J* = 7.4 Hz, Ar-*H*), 7.31 (2H, t, *J* = 7.4 Hz, Ar-*H*); ¹³C NMR (151 MHz, CDCl₃) δ 154.0, 153.1, 140.7, 138.2, 135.5, 133.0, 132.3, 130.6, 128.4, 127.2, 126.1, 123.6, 120.4, 120.2, 113.7, 109.9. HRMS MALDI-TOF (*m/z*) calcd (M⁺) for C₂₄H₁₄BrN₃S: 456.3610; found: 457.2473 (M+H⁺).

9,9'-((2,7-Bibromo-9H-fluorene-9,9-diyl)bis(octane-8,1-diyl))bis(3,6-di-tert-butyl-9H-carbazole) (6) 2,7-Dibromofluorene (4) (2.00 g, 6.17 mmol) and *t*-butyl ammonium bromide (0.198 g, 0.617 mmol) were weighted into round bottom flask, and then solution of KOH (3.461, 61.17 mmol) in water (3.5 mL) was added, the mixtures were stirred at room temperature for 30 min, after that 1,8-dibromooctane (3.4 mL, 18.51 mmol) was added and stirred at 70 °C for 4 h. After cooling down the reaction to the room temperature, the mixtures were poured in water, extracted with dichloromethane (DCM, 3 x 20 mL) and brine solution (50 mL), dried over Na₂SO₄ and concentrated on a rotary evaporator. Finally, the residue was purified by column chromatography eluting with hexane to give pale-yellow solids (compound **5**). ¹H NMR (600 MHz, CDCl₃): δ = 7.43 (s, 2H), 7.52 (d, 2H, *J* = 7.92 Hz), 7.46 (d, 2H, *J* = 8.04 Hz), 3.34 (t, 4H, *J* = 6.72, 6.72 Hz), 1.91 (t, 4H, *J* = 7.74, 8.10 Hz), 1.79-1.75 (m,4H), 1.33-1.28 (m, 4H), 1.12-1.06 (m, 12H), 0.58 (brs, 4H). HRMS APCI (*m*/*z*) calcd (M⁺) for C₂₉H₃₈Br₄: 705.9666; found 706.9927.

3,6-Di-*tert*-butylcarbazole (8.90 g, 31.897 mmol) and KOH (5.72 g, 0.10 mol) were dissolved in DMF (80 mL) and then compound **5** (8.00 g, 12.759 mmol) was added and stirred at room temperature for 4.5 h. The mixtures were poured in water, extracted with dichloromethane (DCM, 3 x 10 mL) and brine solution (20 mL), dried over Na₂SO₄ and evaporated by a rotary evaporator. Finally, the residue was purified by column chromatography eluting with DCM/hexane (1:5) (v/v) to give white solids (9.92 g, 80%). ¹H NMR (600 MHz, CDCl₃): $\delta = 8.08$ (s, 4H), 7.48-7.45 (m, 6H), 7.41 (d, 4H, *J* = 7.5 Hz), 7.24 (s, 4H), 4.16 (t, 4H, *J* = 6.96, 6.90 Hz), 1.89-1.86 (m, 4H), 1.76-1.74 (m, 4H), 1.45 (s, 36H), 1.29-1.25 (m, 5H), 1.12 (m, 4H), 1.01 (s, 8H), 0.55 (brs, 4H). HRMS APCI (*m/z*) calcd (M⁺) for C₆₉H₈₆Br₂N₂: 1102.5564; found 1103.5642.

EML	$V_{on}(V)$	L_{max} (cd m ⁻²)	EQE_{max} (%)	CE_{max} (cd A ⁻¹)	Ref
	3.5	~2000	4.5%	12.0	1
C ₄ H ₉ -{ C ₂ H ₅					
2EHO-TPA-CNPE					
X X X	3.2	4050	3.3	2.61	2
8 got					
- 0 2					
ASTO -					
СРРІ СРРІ					
C ₂ H ₅					-
∕_c⁴H ^a	4.0	~500	2.36	2.23	3
C4H9-C2H5					
2EHO-CNPE					

Table S1. Electroluminescent data of the previously reported solution-processed non-doped HLCT

 OLEDs.

* * * * * * * * * * * * * * * * * * *	3.2	30300	6.74	14.38	This work
· · · · · · · · · · · · · · · · · · ·	5.7	1103	2.20	2.74	-
C P P C P C P C P C P C P C P C P C P C	5.1	5521	2.88	3.71	4

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Fig. S1 Repeated CV scan.

Fig. S2 Copies of ¹H NMR (600 MHz, CDCl₃) and ¹³C-NMR (151 MHz, CDCl₃) and HRMS mass spectra













