

One-Pot Construction of CoSe Nanoparticles Anchored on Single-Atomic-Co Doped Carbon for pH-Universal Hydrogen Evolution

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Figure S1

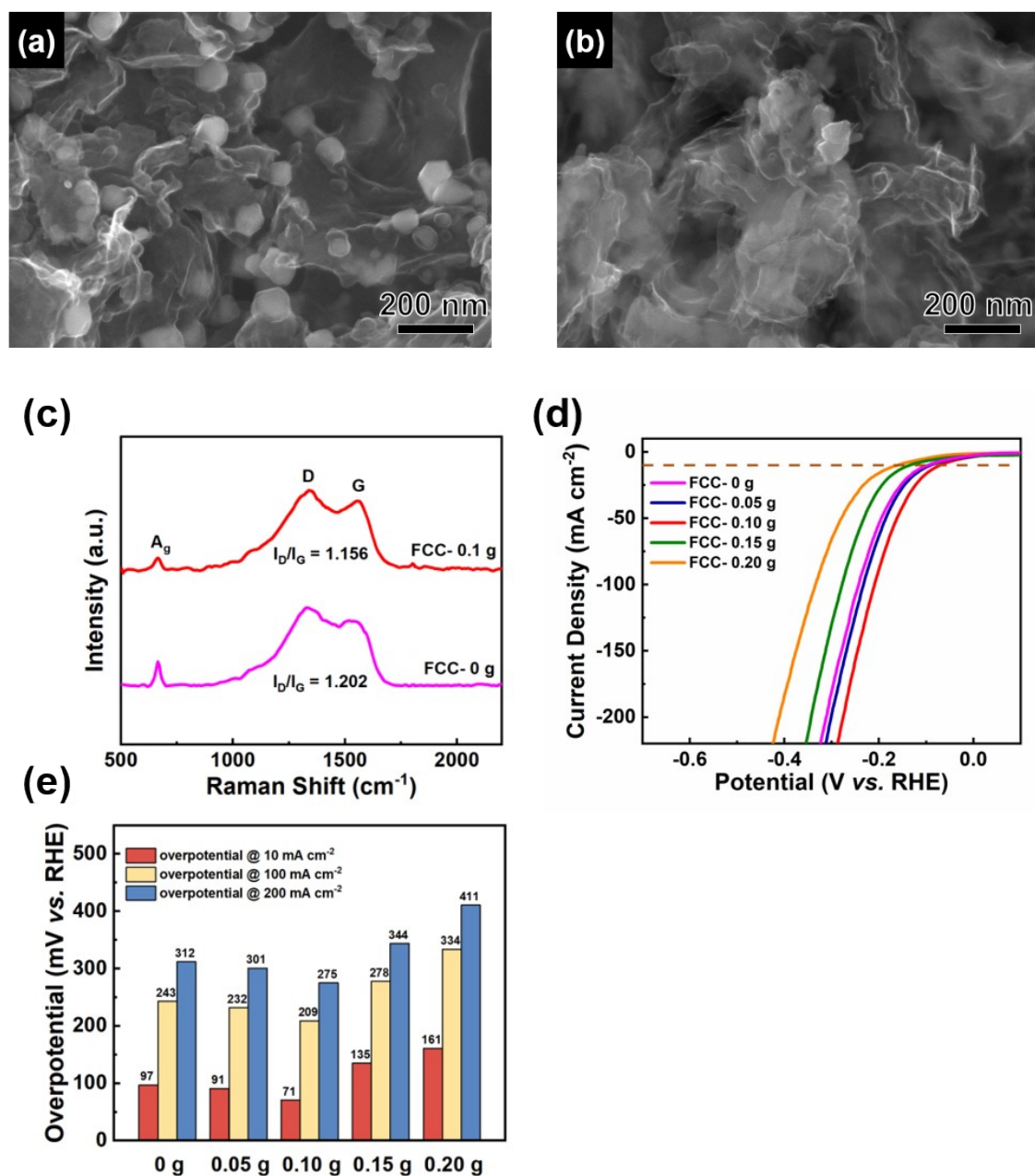


Figure S1. Synthetic optimization of CoSe/Co-N-C: amount of FCC slurry. SEM images of CoSe/Co-N-C with (a) 0.1 g slurry (optimized) and (b) 0 g slurry. (c) Raman spectra of CoSe/Co-N-C with 0 g and 0.1 g slurry. (d) HER LSV curves of CoSe/Co-N-C with 0 g-0.2 g slurry in 1.0 M KOH. (e) η_{10} , η_{100} , and η_{200} of CoSe/Co-N-C with 0 g-0.2 g slurry in 1.0 M KOH

Figure S2

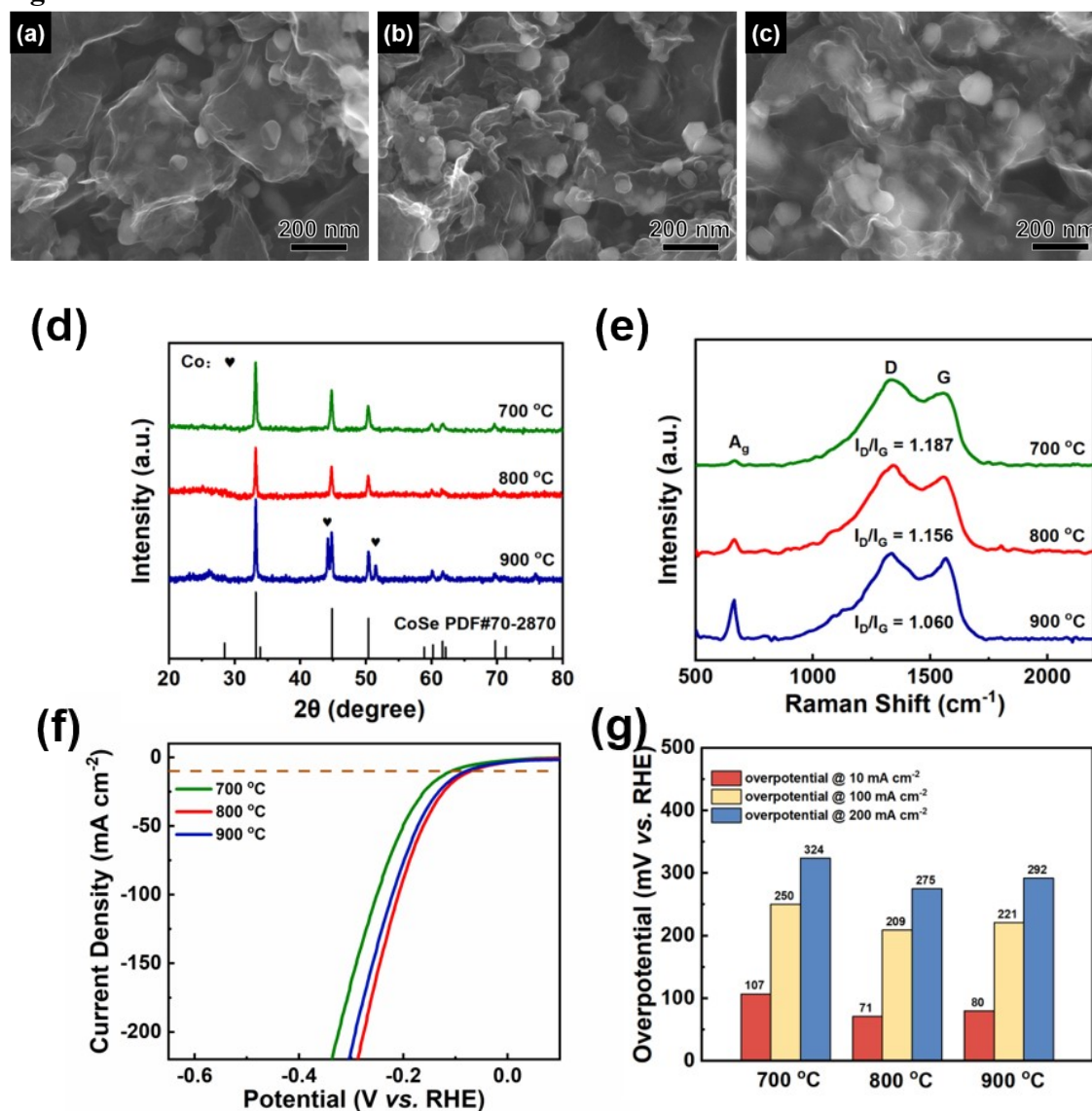


Figure S2. Synthetic optimization of CoSe/Co-N-C: pyrolysis temperature. SEM images of CoSe/Co-N-C pyrolyzed at (a) 800 °C, (b) 700 °C, and (c) 900 °C. (d) XRD spectra of CoSe/Co-N-C pyrolyzed at 700 °C, 800 °C, and 900 °C. (e) Raman spectra of CoSe/Co-N-C pyrolyzed at 700 °C, 800 °C, and 900 °C. (f) HER LSV curves of CoSe/Co-N-C pyrolyzed at 700 °C, 800 °C, and 900 °C in 1.0 M KOH. (g) η_{10} , η_{100} , and η_{200} of CoSe/Co-N-C pyrolyzed at 700 °C, 800 °C, and 900 °C in 1.0 M KOH

Figure S3

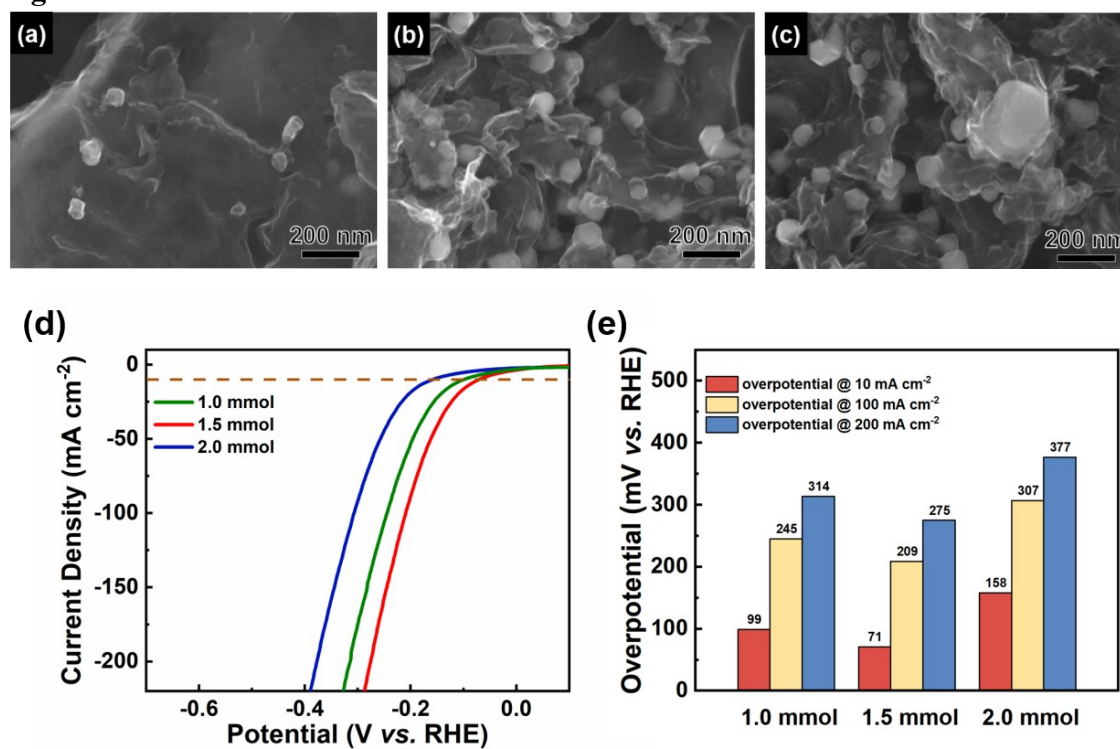


Figure S3. Synthetic optimization of CoSe/Co-N-C: total amount of CoCl₂. SEM images of CoSe/Co-N-C with (a) 1.0 mmol CoCl₂, (b) 1.5 mmol CoCl₂ (optimized), and (c) 2.0 mmol CoCl₂. (d) HER LSV curves of CoSe/Co-N-C with 1.0 mmol, 1.5 mmol, and 2.0 mmol CoCl₂ in 1.0 M KOH. (g) η_{10} , η_{100} , and η_{200} of CoSe/Co-N-C with 1.0 mmol, 1.5 mmol, and 2.0 mmol CoCl₂ in 1.0 M KOH

Figure S4

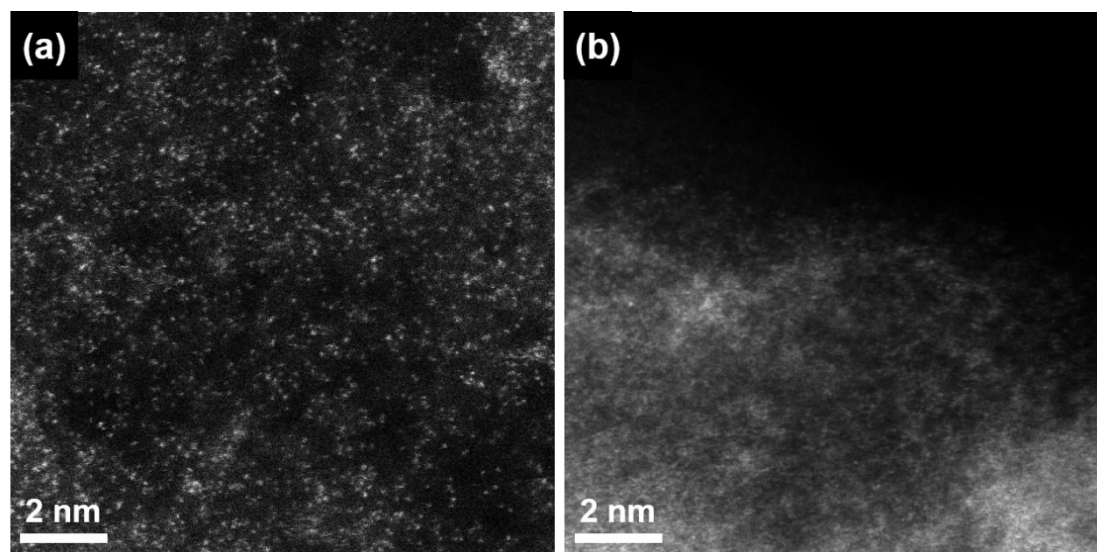
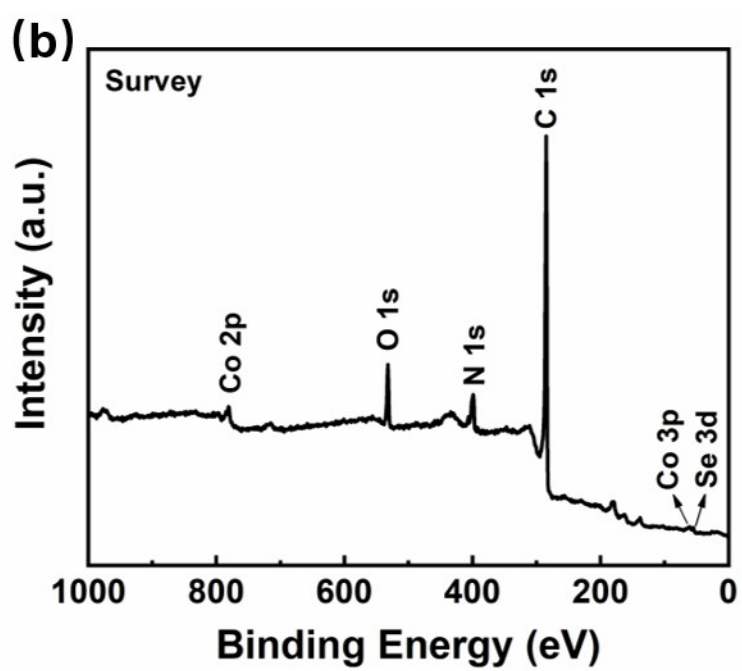
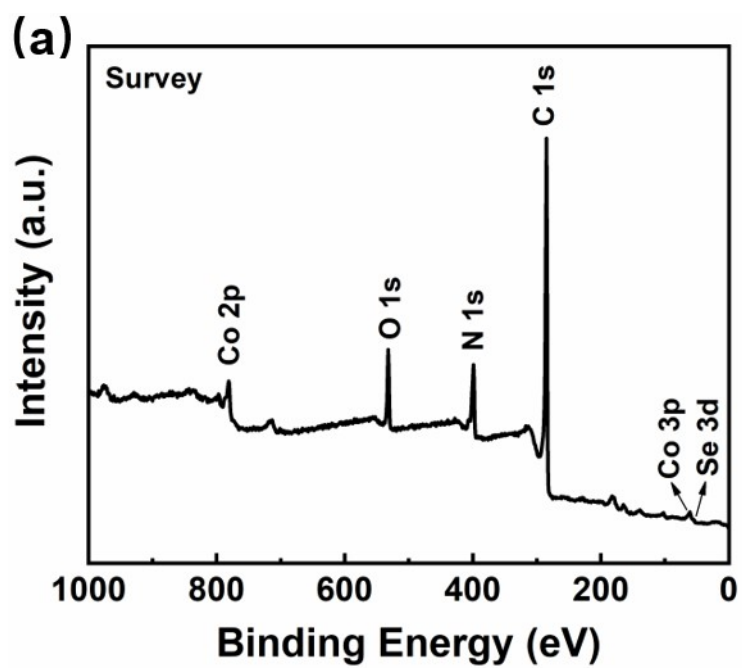


Figure S4. Aberration-corrected high-angle annular darkfield scanning transmission electron microscopy (AC HAADF-STEM) images of (a) Co-N-C and (b) CoSe/N-C.

Table S1. Porosity analysis of the representative catalysts

Catalysts	BET surface area (m² g⁻¹)	Pore size (nm)	Pore volume (cm³ g⁻¹)
CoSe/Co-N-C	177	15.14	0.67
CoSe/N-C	109	16.00	0.44
Co-N-C	308	6.32	0.49
CoSe	17	25.72	0.11

Figure S5



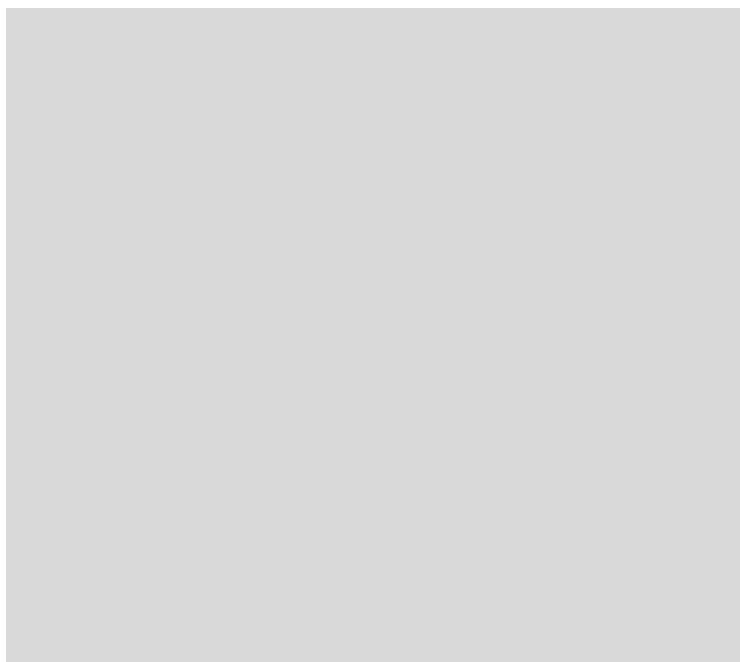


Figure S5. XPS survey spectra of (a) CoSe/Co-N-C, (b) CoSe/N-C, and (c) Co-N-C

Table S2. Binding energies of Co and Se in CoSe/Co-N-C and CoSe/N-C

XPS signal		CoSe/Co-N-C Binding Energy (Ev)	CoSe/N-C Binding Energy (eV)
Co	Co ²⁺ 2p _{1/2}	798.0	797.6
	Co ³⁺ 2p _{1/2}	796.0	795.7
	Co ²⁺ 2p _{3/2}	783.0	781.5
	Co ³⁺ 2p _{3/2}	780.6	780.1
Se	3d _{3/2}	55.6	55.9
	3d _{5/2}	54.3	54.6

Table S3. XPS elemental analysis of CoSe/Co-N-C

Name	Atomic (%)	Weight (%)
C	75.06	63.38
N	13.85	13.64
O	7.99	9.00
Co	2.31	9.59
Se	0.79	4.39

Figure S6

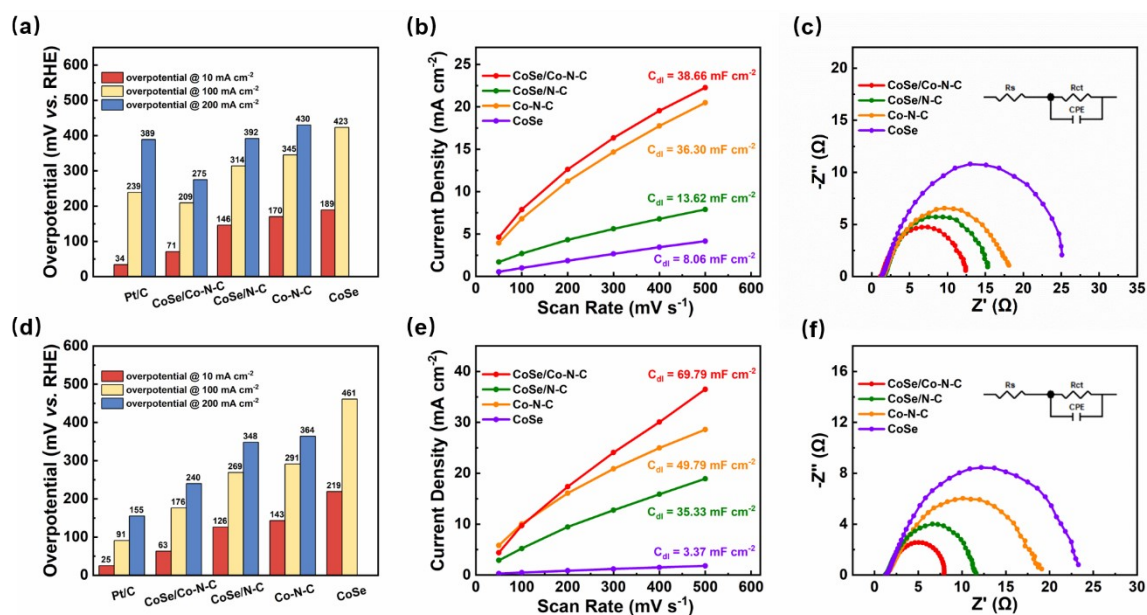


Figure S6. HER performances of CoSe/Co-N-C, CoSe/N-C, Co-N-C, CoSe, and 10% Pt/C. (a) summary of η_{10} , η_{100} , and η_{200} , (b) capacitive current density as a function of scan rates, and (c) Nyquist plots of CoSe/Co-N-C and other references in 1.0 M KOH. (d) summary of η_{10} , η_{100} , and η_{200} , (e) capacitive current density as a function of scan rates, and (f) Nyquist plots of CoSe/Co-N-C and other references in 0.5 M H₂SO₄.

Figure S7

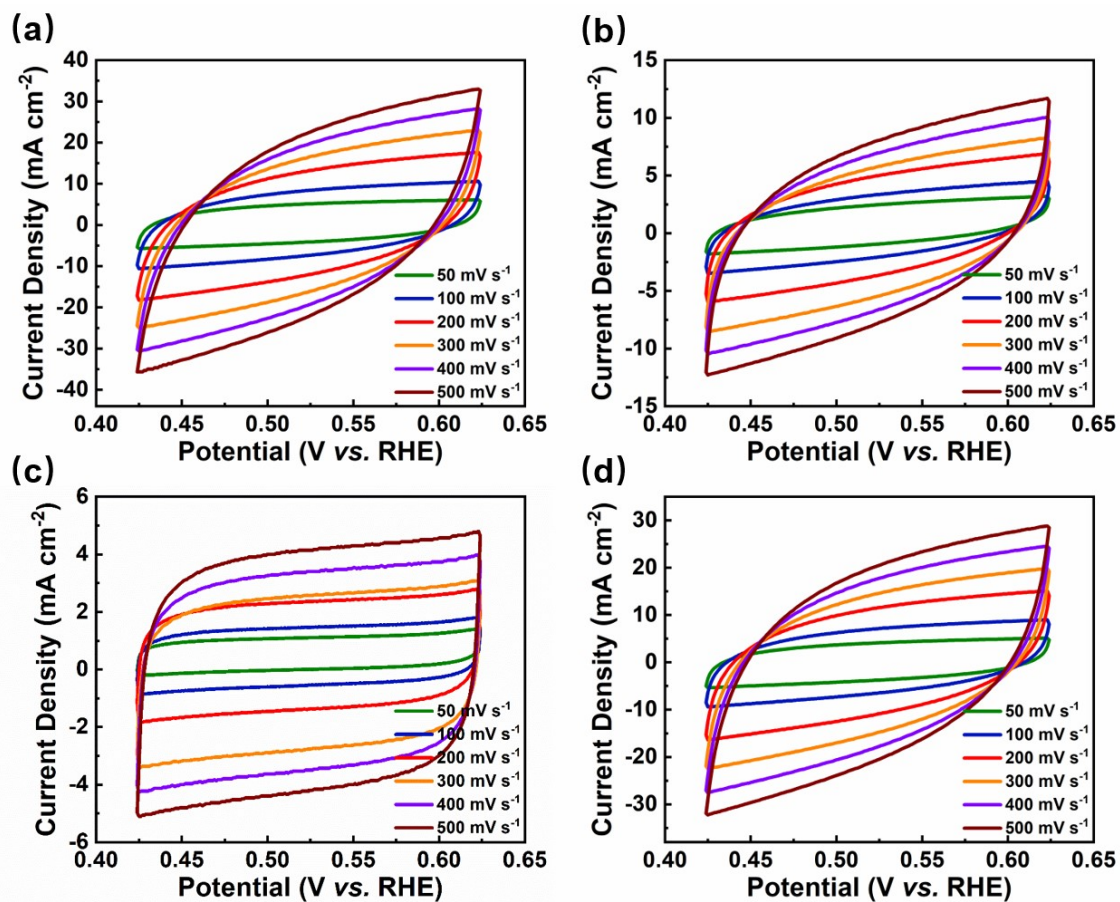


Figure S7. CV spectra of (a) CoSe/Co-N-C, (b) CoSe/N-C, (c) CoSe, and (d) Co-N-C with various scan rates (50-500 mV s⁻¹) in 1.0 M KOH.

Figure S8

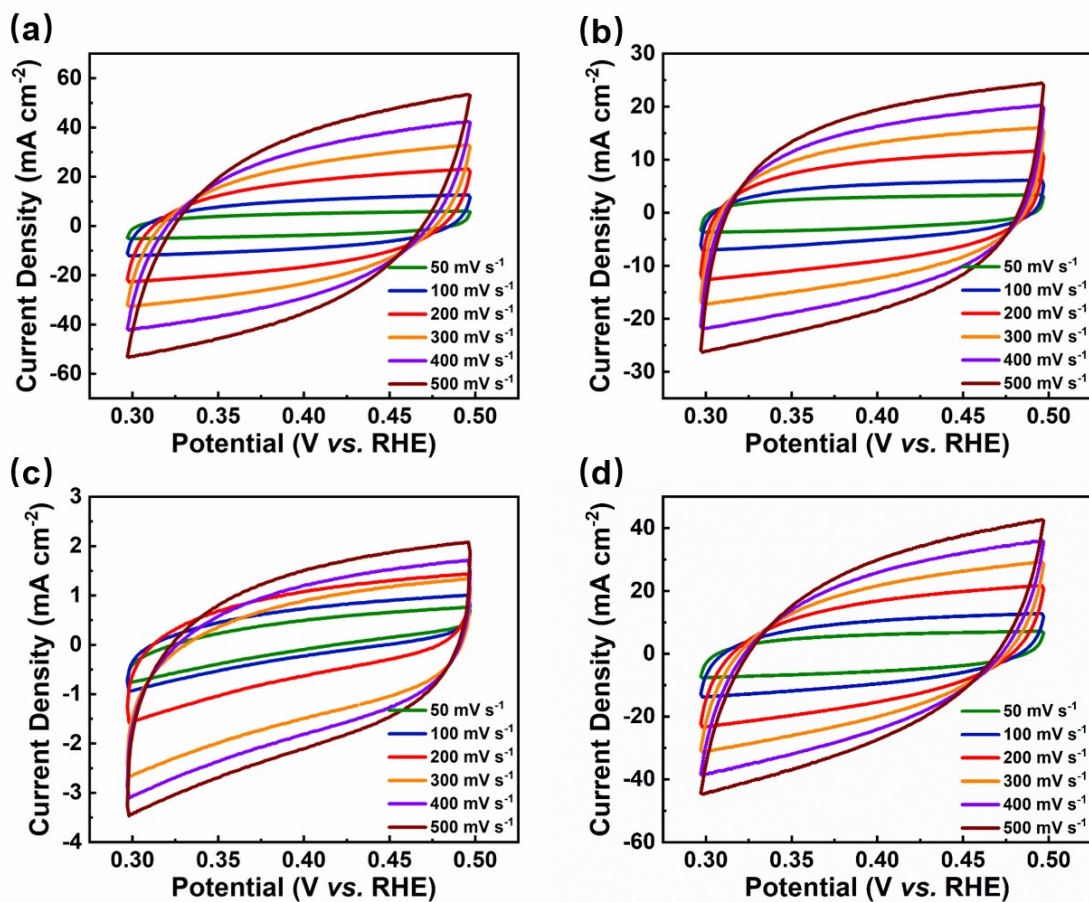


Figure S8. CV spectra of (a) CoSe/Co-N-C, (b) CoSe/N-C, (c) CoSe, and (d) Co-N-C with various scan rates (50-500 mV s^{-1}) in 0.5 M H_2SO_4 .

Figure S9

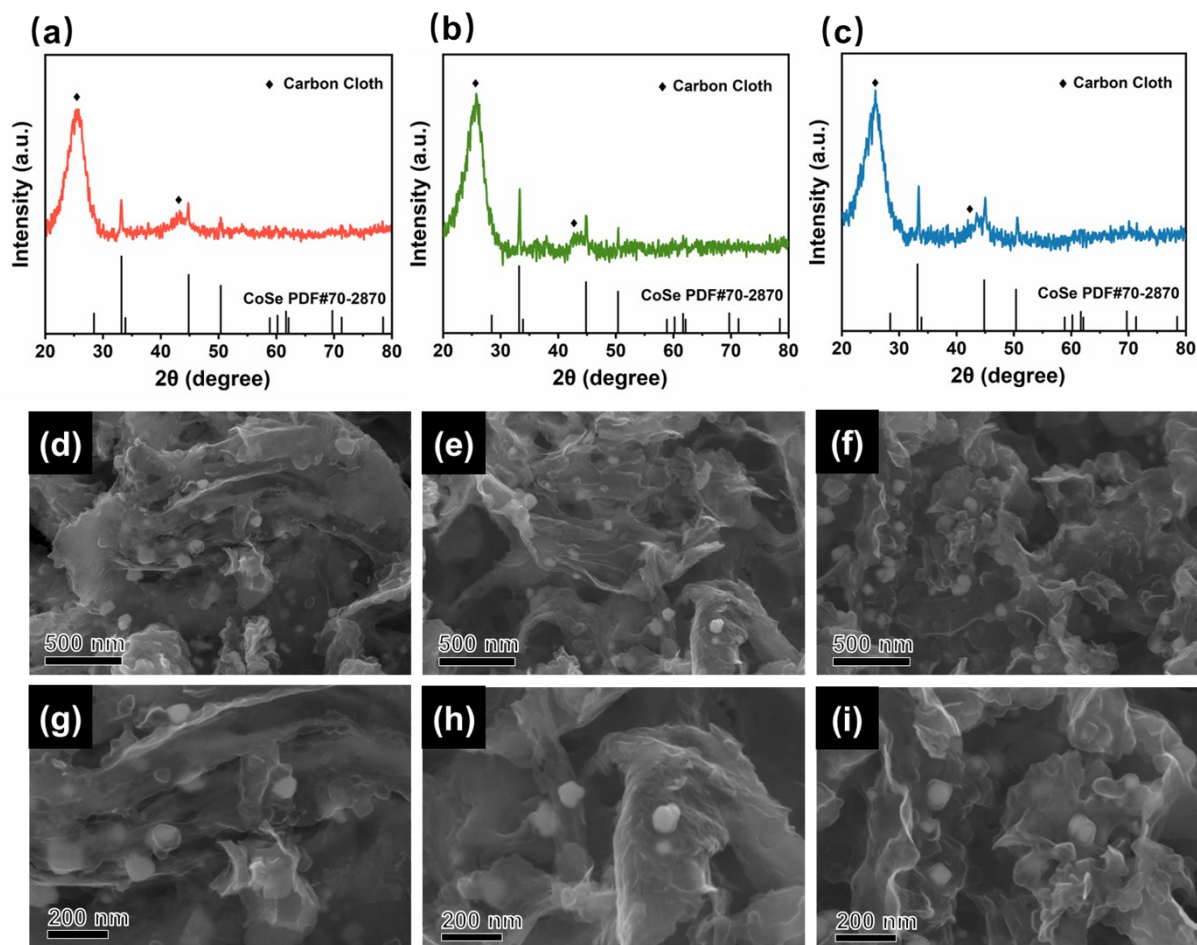


Figure S9. (a) XRD spectrum and (d, g) SEM images of CoSe/Co-N-C after 2000 CV electrocatalytic cycles in 1.0 M KOH; (b) XRD spectrum and (e, h) SEM images of CoSe/Co-N-C after 2000 CV electrocatalytic cycles in 0.5 M H₂SO₄; (c) XRD spectrum and (f, i) SEM images of CoSe/Co-N-C after 2000 CV electrocatalytic cycles in 1.0 M PBS.

Table S4. HER activity summary of CoSe/Co-N-C, CoSe/N-C, Co-N-C, and CoSe in various electrolytes.

Electrolytes		η_{10} (mV)	η_{100} (mV)	η_{200} (mV)	Tafel Slope (mV dec ⁻¹)
1.0 M KOH	CoSe/Co-N-C	71	209	275	79
	CoSe/N-C	146	314	392	139
	Co-N-C	170	345	430	147
	CoSe	189	423	N/A	180
0.5 M H ₂ SO ₄	CoSe/Co-N-C	63	176	240	65
	CoSe/N-C	126	269	348	108
	Co-N-C	143	291	364	125
	CoSe	219	461	N/A	153
1.0 M PBS	CoSe/Co-N-C	128	293	360	110

Table S5. Summary of HER performances of CoSe/Co-N-C and other TMSes electrocatalysts in the literatures.

Catalysts	Tafel slope (mV/dec)	Overpotential (mV)@10mA/cm ²	Electrolyte	Stability (h)	References
CoSe/Co-N-C	79	71	1.0 M KOH	36	This work
	65	63	0.5 M H ₂ SO ₄		
CoSe ₂ -NC	95	234	1.0 M KOH	14	1
	62	156	0.5 M H ₂ SO ₄		
NiSe ₂ @NG	37.8	248	1.0 M KOH	18	2
	79.8	201	0.5 M H ₂ SO ₄		
Ni _{0.85} Se@NC	97	135	1.0 M KOH	10	3
	85	131	0.5 M H ₂ SO ₄		
NiSe ₂ @NC	88	162	1.0 M KOH	40	4
	37	183	0.5 M H ₂ SO ₄		
MoSe ₂ -NiSe@carbon	80.6	180	1.0 M KOH	10	5
	76.3	154	0.5 M H ₂ SO ₄		
NiSe@NC	55.3	250	1.0 M KOH	12	6
	53.3	123	0.5 M H ₂ SO ₄		
Co _{0.9} Ni _{0.1} Se	58	185.7	0.5 M H ₂ SO ₄	-	7
CoS _{0.46} Se _{0.54} @C	105.43	251	1.0 M KOH	-	8
Co _{0.85} Se@NC	125	230	1.0 M KOH	10	9
CoNiSe/NC	66.5	100	1.0 M KOH	3.33	10
Co _{0.8} Mo _{0.2} Se	58.7	86.7	0.5 M H ₂ SO ₄	10	11
CoSe@NCNT/NCN	43	197	0.5 M H ₂ SO ₄	18	12

Table S6. The interaction energies (E_{int}) of CoSe/N-C and CoSe/Co-N-C, and the Bader Charge on NC and Co-N-C backbones.

Samples	E_{int} (eV)	C_{NC}/C_{Co-N-C} (e)
CoSe/N-C	-3.20	0.15
CoSe/Co-N-C	-3.48	0.34

Figure S8

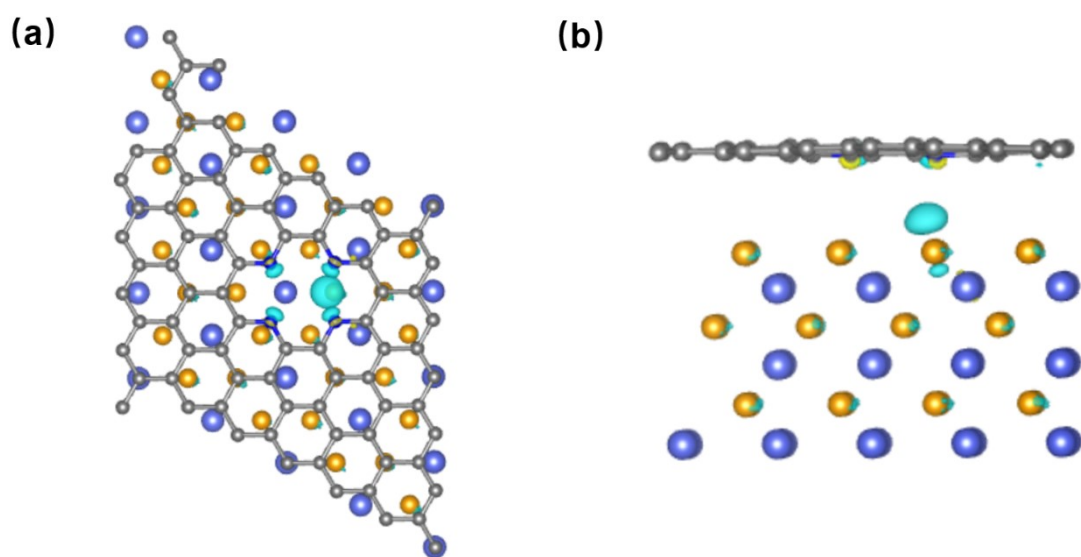


Figure S8. (a) Top view and (b) side view of the charge density in the heterostructure interface of CoSe/N-C. The charge accumulation and depletion were colored in cyan and yellow, respectively. Gray balls: C atoms; blue balls: Co atoms; yellow balls: Se atoms.

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