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## **Supporting information for:**

Sulfur vacancies modulated nickel-doped  $Co_4S_3$  hollow nanocubes/nitrogen-doped  $V_2CT_x$  MXene nanosheets composite for optimizing hydrogen evolution reaction

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## Characterization

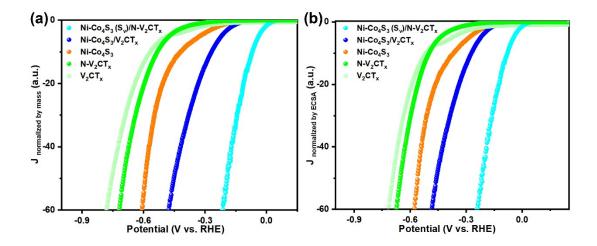
X-ray diffraction (XRD) measurements were implemented by a powder X-ray diffraction system (Rigaku, TTR-III). The X-ray photoelectron spectroscopy (XPS) measurements were implemented by a Thermo ESCALAB 250Xi spectrometer with monochromatic Al Kα radiation (hγ=1486.6 eV). The structure was investigated by scanning electron microscope (FE-SEM) (Hitachi, SU8000) and a transmission electron microscopy (TEM) (JEOL, JEM-2010, 200 kV). Nitrogen adsorption-desorption experiments were carried out at 77.35 K by means of an Autosorb-1 (Quantachrome Instruments) analyzer.

## **Electrochemical measurements**

The measurements on the electrochemical performances were performed using a CHI 660E electrochemical workstation in a standard three-electrode mode. All electrochemical performance tests were performed in 0.5 M H<sub>2</sub>SO<sub>4</sub> solution. The working electrode was prepared by passing the slurry containing 4 mg Ni-Co<sub>4</sub>S<sub>3</sub> (Sv)/N-V<sub>2</sub>CT<sub>x</sub> (85 wt.%) with acetyleneblack (10 wt.%) and PTFE (5 wt.%) in 10 mL absolute ethanol. After that, the electrode was made by heating at 60 °C for 40 min. The as-synthesized Ni-Co<sub>4</sub>S<sub>3</sub> (Sv)/N-V<sub>2</sub>CT<sub>x</sub> slurry was coated onto the glass carbon electrode by applying suitable compressive force. The working electrode, reference electrode, and counter electrode are self-supporting electrode carrying electrocatalysts, saturated calomel electrode and graphite rods, respectively. The HER potentials used in this work are corrected with a reversible hydrogen electrode.

Table S1 The various porous parameters of the samples

Samples	Specific Surface Area	Pore Diameter
$V_2CT_x$	$6.65 \text{ m}^2/\text{g}$	3.39 nm
$Ni-Co_4S_3 (S_v)/N-V_2CT_x$	$12.58 \text{ m}^2/\text{g}$	2.20 nm



**Figure S1** LSV curves of catalysts after normalizing the current by loading mass (a) electrochemical active surface area (b).

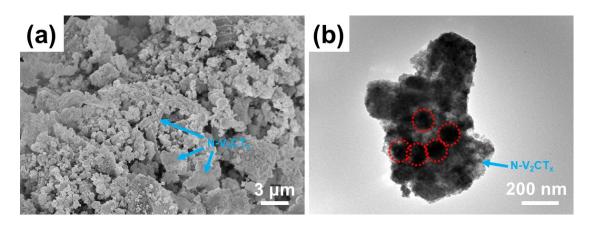


Figure S2 SEM (a) and TEM (b) images of Ni-Co<sub>4</sub>S<sub>3</sub> (S $_{v}$ )/N-V<sub>2</sub>CT $_{x}$  after test.