## 1,1,2-Trifunctionalization of Terminal Alkynes by Imine/Borane Frustrated Lewis Pairs

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#### **General information**

All syntheses involving air and moisture sensitive compounds were carried out using standard Schlenk-type glassware in glove box under argon atmosphere. Super-dry solvents were used and stored under argon atmosphere in glove box. NMR spectra were recorded on Brucker Avance III 500 MHz (<sup>1</sup>H: 500 MHz, <sup>13</sup>C: 126 MHz, <sup>19</sup>F: 470 MHz, <sup>11</sup>B: 160 MHz) at room temperature if not specified otherwise. <sup>1</sup>H NMR and <sup>13</sup>C NMR: chemical shifts are given relative to TMS and referenced to the solvent signal. <sup>19</sup>F NMR: chemical shifts are given relative to CFCl<sub>3</sub> ( $\delta = 0$ , external reference), <sup>11</sup>B NMR: chemical shifts are given relative to BF<sub>3</sub>·Et<sub>2</sub>O ( $\delta = 0$ , external reference). HRMS was recorded on Waters Xevo G2-XS QTOF. IR was recorded on Shimadzu Affinity-1. Melting points were obtained with Inesa WRS-2 instrument.

Single crystal x-ray diffraction was done by Bruker D8 Venture microfocus source diffractometer. Using Olex2<sup>[1]</sup>, the structure was solved with the SHELXT<sup>[2]</sup> structure solution program using Intrinsic Phasing and refined with the SHELXL<sup>[3]</sup> refinement package using Least Squares minimization.

<sup>[1]</sup>Dolomanov, O.V., Bourhis, L.J., Gildea, R.J, Howard, J.A.K. & Puschmann, H. (2009), J. Appl. Cryst. 42, 339-341.

<sup>[2]</sup> Sheldrick, G.M. (2015). Acta Cryst. A71, 3-8.

<sup>[3]</sup> Sheldrick, G.M. (2015). Acta Cryst. C71, 3-8.

CCDC 2059106-2059108, 2062051-2062052, 2093061 contain the supplementary crystallographic data. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data\_request/cif, or by emailing data\_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: + 44 1223 336033.

**Note:** Single crystal x-ray diffraction of compounds **4a**, **4c**, **4g**, **4i** was measured at 100 K, and the diffraction of compounds **4d**, **4f** was measured at 200 K. It is because the crystals of compounds **4d**, **4f** cracked at temperature of 100 K. To obtain high quality diffraction data, the diffraction signals were recorded at temperature of 200 K.

#### Synthesis and characterization of compound 4a

Phenyl acetylene **1a** (30.6 mg, 0.3 mmol) was added to a solution of tris(pentafluorophenyl)borane **2a** (153.6 mg, 0.3 mmol) and (*E*)-N-tert-butyl-1-phenylmethanimine **3a** (48.4 mg, 0.3 mmol) in toluene (2 mL). The reaction mixture was heated at 80 °C for 12 hours. After completion of reaction, all volatiles were removed *in vacuo* and the residue was purified by flash chromatography through silica gel column (petroleum ether/ethyl acetate =10:1). After drying under vacuum Compound **4a** (230.2 mg, 99% yield) was obtained as a white solid. Crystals suitable for the X-ray crystal structure analysis were obtained by slow diffusion of pentane to a solution of compound **4a** in dichloromethane at room temperature.

**M.p.**: 189.9-191.7 °C. **HRMS** (ESI, *m/z*) calcd for C<sub>37</sub>H<sub>20</sub>BF<sub>15</sub>N<sup>-</sup>: 774.1499 [M-H]<sup>-</sup>, found: 774.1488.

**IR** (KBr): 3286, 1647, 1527, 1444, 1278, 1190, 1083, 970, 794, 696 cm<sup>-1</sup>.

<sup>1</sup>**H** NMR (500 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  7.51 (m, 1H), 7.04 (m, 1H), 6.87-6.82 (m, 3H), 6.78-6.72 (m, 3H), 6.66 (m, 1H), 6.64 (m, 1H), 6.53 (m, 1H), 5.61 (d, *J* = 8.0 Hz, 1H), 0.80 (s, 9H).

<sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, C<sub>6</sub>D<sub>6</sub>): δ 148.8, 139.8, 136.2, 130.3, 130.0, 129.35, 129.28, 128.3, 127.6, 125.2, 74.5, 62.2, 28.4, [C<sub>6</sub>F<sub>5</sub> not listed].

<sup>11</sup>B{<sup>1</sup>H} NMR (160 MHz,  $C_6D_6$ ):  $\delta = 0.2 (v_{1/2} \sim 380 \text{ Hz}).$ 

<sup>19</sup>**F NMR** (470 MHz, C<sub>6</sub>D<sub>6</sub>): δ -126.4 (m, 1F), -127.2 (m, 1F), -127.5 (m, 1F), -140.0 (m, 1F), -140.1 (m, 1F), -144.0 (m, 1F), -154.1 (t, J = 20.8 Hz, 1F), -156.4 (t, J = 21.2 Hz, 1F), -156.7 (t, J = 20.8 Hz, 1F), -162.1 (m, 1F), -162.5 (m, 1F), -162.7 (m, 1F), -162.9 (m, 1F), -164.4 (m, 1F), -164.5 (m, 1F).

<sup>1</sup>**H**, <sup>1</sup>**H**-**GCOSY** (500 MHz / 500 MHz,  $C_6D_6$ ):  $\delta^{-1}H / \delta^{-1}H = 6.66/5.61$  (NH/CH).

<sup>1</sup>**H**, <sup>13</sup>**C-GHSQC** (500 MHz / 126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>13</sup>C = 5.61/74.5 (CHN), 0.80/28.4 (*t*-Bu).

<sup>1</sup>**H**, <sup>13</sup>**C-GHMBC** (500 MHz / 126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>13</sup>C = 0.80/62.2 (*t*-Bu).











**X-ray crystal structure analysis of compound 4a:** A suitable crystal was selected and measured on a Bruker D8 VENTURE diffractometer. The crystal was kept at 100.15 K during data collection. Using Olex2, the structure was solved with the olex2.solve structure solution program using Charge Flipping and refined with the SHELXL refinement package using Least Squares minimization.



Figure S1. Structure of compound 4a at 50% probability level

Empirical formula	$C_{37}H_{21}BF_{15}N$
Formula weight	775.36
Temperature/K	100.15
Crystal system	triclinic
Space group	P-1
a/Å	9.5209(7)
b/Å	10.6117(10)
c/Å	16.6483(16)
$\alpha/^{\circ}$	81.458(3)
β/°	75.451(3)
γ/°	82.822(3)
Volume/Å <sup>3</sup>	1603.3(2)
Ζ	2
$\rho_{calc}g/cm^3$	1.606
$\mu/\text{mm}^{-1}$	0.154
F(000)	780.0
Crystal size/mm <sup>3</sup>	$0.41 \times 0.38 \times 0.15$
Radiation	MoKa ( $\lambda = 0.71073$ )
$2\Theta$ range for data collection/°	4.438 to 55.046
Index ranges	$-12 \le h \le 12, -13 \le k \le 13, -21 \le 1 \le 21$
Reflections collected	31062
Independent reflections	7378 [ $R_{int} = 0.0383, R_{sigma} = 0.0346$ ]
Data/restraints/parameters	7378/0/494
Goodness-of-fit on F <sup>2</sup>	1.032
Final R indexes [I>= $2\sigma$ (I)]	$R_1 = 0.0394, wR_2 = 0.0869$
Final R indexes [all data]	$R_1 = 0.0613, wR_2 = 0.0961$
Largest diff. peak/hole / e Å <sup>-3</sup>	0.36/-0.27

Table S1 Crystal data and structure refinement for compound 4a

#### Synthesis and characterization of compound 4b

1-Ethynyl-4-fluorobenzene **1b** (36.0 mg, 0.3 mmol) was added to a solution of tris(pentafluorophenyl)borane **2a** (153.6 mg, 0.3 mmol) and (*E*)-N-tert-butyl-1-phenylmethanimine **3a** (48.4 mg, 0.3 mmol) in toluene (2 mL). The reaction mixture was heated at 80 °C for 12 hours. After completion of reaction, all volatiles were removed *in vacuo* and the residue was purified by flash chromatography through silica gel column (petroleum ether/ethyl acetate =10:1). After drying under vacuum Compound **4b** (235.8 mg, 98% yield) was obtained as an off white solid.

**M.p.**: 213.3-214.6 °C. **HRMS** (ESI, *m/z*) calcd for C<sub>37</sub>H<sub>19</sub>BF<sub>16</sub>N<sup>-</sup>: 792.1355 [M-H]<sup>-</sup>, found: 792.1379.

**IR** (KBr): 3286, 2981, 1639, 1527, 1444, 1276, 1186, 1109, 964, 839, 692, 536 cm<sup>-1</sup>.

<sup>1</sup>**H NMR** (500 MHz, C<sub>6</sub>D<sub>6</sub>) 7.45 (m, 1H), 6.03 (m, 1H), 6.86 (m, 1H), 6.76 (m, 1H), 6.67-6.62 (m, 2H), 6.61 (m, 1H), 6.48 (m, 1H), 6.40 (m, 2H), 5.50 (d, *J* = 7.5 Hz, 1H), 0.78 (s, 9H).

<sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  162.4 (d, J = 248.2 Hz), 147.8, 139.6, 132.0 (d, J = 3.5 Hz), 130.2, 130.1, 129.5 (d, J = 8.0 Hz), 129.4 (d, J = 7.0 Hz), 125.0, 115.4 (d, J = 21.4 Hz), 74.4, 62.2, 28.3, [C<sub>6</sub>F<sub>5</sub> not listed].

<sup>11</sup>B{<sup>1</sup>H} NMR (160 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta = 0.2$  (v<sub>1/2</sub> ~ 390 Hz).

<sup>19</sup>**F NMR** (470 MHz, C<sub>6</sub>D<sub>6</sub>): δ -113.0 (m, 1F), -126.3 (m, 1F), -127.1 (m, 1F), -127.9 (m, 1F), -140.0 – -140.2 (m, 2F), -144.1 (m, 1F), -153.9 (t, J = 20.9 Hz, 1F), -156.0 (t, J = 21.2 Hz, 1F), -156.5 (t, J = 20.8 Hz, 1F), -162.02 (m, 1F), -162.3 (m, 1F), -162.5 (m, 1F), -162.7 (m, 1F), -164.2 (m, 1F), -164.4 (m, 1F).

<sup>1</sup>**H**, <sup>1</sup>**H**-**GCOSY** (500 MHz / 500 MHz,  $C_6D_6$ ):  $\delta^{-1}H / \delta^{-1}H = 6.61/5.50$  (NH/CH).

<sup>1</sup>**H**, <sup>13</sup>**C-GHSQC** (500 MHz / 126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>13</sup>C = 5.50/74.4 (CHN), 0.78/28.3 (*t*-Bu).

<sup>1</sup>**H**, <sup>13</sup>**C-GHMBC** (500 MHz / 126 MHz,  $C_6D_6$ ):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>13</sup>C = 0.78/62.2 CH<sub>3</sub> (*t*-Bu).



**S**11









S13



#### Synthesis and characterization of compound 4c

1-Chloro-4-ethynylbenzene **1c** (41.0 mg, 0.3 mmol) was added to a solution of tris(pentafluorophenyl)borane **2a** (153.6 mg, 0.3 mmol) and (*E*)-N-tert-butyl-1-phenylmethanimine **3a** (48.4 mg, 0.3 mmol) in toluene (2 mL). The reaction mixture was heated at 80 °C for 12 hours. After completion of reaction, all volatiles were removed *in vacuo* and the residue was purified by flash chromatography through silica gel column (petroleum ether/ethyl acetate =10:1). After drying under vacuum Compound **4c** (240.6 mg, 99% yield) was obtained as a white solid. Crystals suitable for the X-ray crystal structure analysis were obtained by slow diffusion of pentane to a solution of compound **4c** in dichloromethane at room temperature.

**M.p.**: 191.3-192.1 °C. **HRMS** (ESI, m/z) calcd for C<sub>37</sub>H<sub>19</sub>BClF<sub>15</sub>N<sup>-</sup>: 808.1060 [M-H]<sup>-</sup>, found: 808.1057.

**IR** (KBr): 3288, 2980, 1647, 1525, 1462, 1278, 1083, 987, 698 cm<sup>-1</sup>.

<sup>1</sup>**H NMR** (500 MHz, C<sub>6</sub>D<sub>6</sub>)  $\delta$  7.42 (m, 1H), 7.01 (m, 1H), 6.83 (m, 1H), 6.73 (d, *J* = 8.4 Hz, 2H), 6.73 (m, 1H), 6.61 (d, *J* = 8.4 Hz, 2H), 6.60 (m, 1H), 6.47 (m, 1H), 5.50 (d, *J* = 7.0 Hz, 1H), 0.76 (s, 9H).

<sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  147.5, 139.5, 134.5, 134.2, 130.1, 129.5, 129.4, 129.0, 128.7, 125.0, 74.2 (d, *J* = 7.0 Hz), 62.2, 28.3, [C<sub>6</sub>F<sub>5</sub> not listed].

<sup>11</sup>B{<sup>1</sup>H} NMR (160 MHz,  $C_6D_6$ ):  $\delta = 0.1$  (v1/2 ~ 390Hz).

<sup>19</sup>**F NMR** (470 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  -126.3 (m, 1F), -127.1 (m, 1F), -127.9 (m, 1F), -140.0 – -140.2 (m, 2F), -144.1 (m, 1F), -153.7 (t, *J* = 21.0 Hz, 1F), -155.6 (t, *J* = 21.2 Hz, 1F), -156.3 (t, *J* = 20.8 Hz, 1F), -161.9 (m, 2F), -162.3 (m, 1F), -162.6 (m, 1F), -164.1 (m, 1F), -164.4 (m, 1F).

<sup>1</sup>**H**, <sup>1</sup>**H**-**GCOSY** (500 MHz / 500 MHz,  $C_6D_6$ ):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>1</sup>H = 6.60 /5.50 (NH/CH).

<sup>1</sup>**H**, <sup>13</sup>**C-GHSQC** (500 MHz / 126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>13</sup>C = 5.50/74.2 (CHN), 0.76/28.3 (*t*-Bu).

<sup>1</sup>**H**, <sup>13</sup>**C-GHMBC** (500 MHz / 126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>13</sup>C = 0.76/62.2 (*t*-Bu).











**S**18



**X-ray crystal structure analysis of compound 4c:** A suitable crystal was selected and measured on a Bruker D8 VENTURE diffractometer. The crystal was kept at 100.15 K during data collection. Using Olex2, the structure was solved with the olex2.solve structure solution program using Charge Flipping and refined with the SHELXL refinement package using Least Squares minimization.



Figure S2. Structure of compound 4c at 50% probability level

Empirical formula	$C_{37}H_{20}BClF_{15}N$
Formula weight	809.80
Temperature/K	100.15
Crystal system	monoclinic
Space group	$P2_1/c$
a/Å	10.9258(5)
b/Å	14.4818(7)
c/Å	23.2722(11)
a/°	90
β/°	103.041(2)
$\gamma/^{\circ}$	90
Volume/Å <sup>3</sup>	3587.3(3)
Z	4
$ ho_{calc}g/cm^3$	1.499
$\mu/\text{mm}^{-1}$	0.213
F(000)	1624.0
Crystal size/mm <sup>3</sup>	$0.35 \times 0.28 \times 0.24$
Radiation	MoKα ( $\lambda = 0.71073$ )
$2\Theta$ range for data collection/°	4.564 to 55.062
Index ranges	$-14 \le h \le 14, -18 \le k \le 18, -30 \le 1 \le 30$
Reflections collected	67863
Independent reflections	8262 [ $R_{int} = 0.0784, R_{sigma} = 0.0430$ ]
Data/restraints/parameters	8262/0/503
Goodness-of-fit on F <sup>2</sup>	1.061
Final R indexes [I>= $2\sigma$ (I)]	$R_1 = 0.0423, wR_2 = 0.0972$
Final R indexes [all data]	$R_1 = 0.0680, wR_2 = 0.1073$
Largest diff. peak/hole / e Å <sup>-3</sup>	0.38/-0.35

 Table S2 Crystal data and structure refinement for compound

#### Synthesis and characterization of compound 4d

1-Bromo-4-ethynylbenzene **1d** (54.3 mg, 0.3 mmol) was added to a solution of tris(pentafluorophenyl)borane **2a** (153.6 mg, 0.3 mmol) and (*E*)-N-tert-butyl-1-phenylmethanimine **3a** (48.4 mg, 0.3 mmol) in toluene (2 mL). The reaction mixture was heated at 80 °C for 12 hours. After completion of reaction, all volatiles were removed *in vacuo* and the residue was purified by flash chromatography through silica gel column (petroleum ether/ethyl acetate =10:1). After drying under vacuum Compound **4d** (253.0 mg, 97% yield) was obtained as a white solid. Crystals suitable for the X-ray crystal structure analysis were obtained by slow diffusion of pentane to a solution of compound **4d** in dichloromethane at room temperature.

**M.p.**: 198.3-199.5 °C. **HRMS** (ESI, m/z) calcd for C<sub>37</sub>H<sub>19</sub>BBrF<sub>15</sub>N<sup>-</sup>: 852.0554 [M-H]<sup>-</sup>, found: 852.0566.

**IR** (KBr): 3290, 2993, 1649, 1517, 1448, 1282, 1186, 1097, 987, 781, 702, 542 cm<sup>-1</sup>.

<sup>1</sup>**H** NMR (500 MHz, C<sub>6</sub>D<sub>6</sub>)  $\delta$  7.41 (m, 1H), 7.00 (m, 1H), 6.88 (d, *J* = 8.4 Hz, 2H), 6.82 (m, 1H), 6.71 (m, 1H), 6.59 (dd, *J* = 28.3, 7.5 Hz, 1H), 6.54 (d, *J* = 8.4 Hz, 2H), 6.45 (m, 1H), 5.48 (d, *J* = 7.5 Hz, 1H), 0.75 (s, 9H).

<sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  147.5, 139.5, 134.9, 131.6, 130.1, 129.5, 129.4, 129.3, 125.0, 122.5, 74.1 (d, *J* = 7.1 Hz), 62.2, 28.3, [C<sub>6</sub>F<sub>5</sub> not listed].

<sup>11</sup>B{<sup>1</sup>H} NMR (160 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta = -0.1$  (v<sub>1/2</sub> ~ 330Hz).

<sup>19</sup>**F NMR** (470 MHz,  $C_6D_6$ ):  $\delta$  -126.3 (m, 1F), -127.1 (m, 1F), -127.9 (m, 1F), -140.0 – -140.2 (m, 2F), -144.1 (m, 1F), -153.5 (t, J = 21.1 Hz, 1F), -155.4 (t, J = 21.3 Hz, 1F), -156.1 (t, J = 20.8 Hz, 1F), -161.8 (m, 1F), -161.9 (m, 1F), -162.2 (m, 1F), -162.5 (m, 1F), -164.0 (m, 1F), -164.3 (m, 1F).

<sup>1</sup>**H**, <sup>1</sup>**H**-**GCOSY** (500 MHz / 500 MHz,  $C_6D_6$ ):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>1</sup>H = 6.59 /5.48 (NH/CH).

<sup>1</sup>**H**, <sup>13</sup>**C-GHSQC** (500 MHz / 126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>13</sup>C = 5.48/74.1 (CHN), 0.75/28.3 (*t*-Bu).

<sup>1</sup>**H**, <sup>13</sup>**C-GHMBC** (500 MHz / 126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>13</sup>C = 0.75/62.2 (*t*-Bu).



S23







S25



**X-ray crystal structure analysis of compound 4d:** A suitable crystal was selected and measured on a Bruker D8 VENTURE diffractometer. The crystal was kept at 200.00 K during data collection. Using Olex2, the structure was solved with the olex2.solve structure solution program using Charge Flipping and refined with the SHELXL refinement package using Least Squares minimization.



Figure S3. Structure of compound 4d at 50% probability level

Empirical formula	$C_{37}H_{20}BBrF_{15}N$
Formula weight	854.26
Temperature/K	200.00
Crystal system	monoclinic
Space group	$P2_1/c$
a/Å	10.6971(5)
b/Å	29.5285(12)
c/Å	10.9499(5)
a/°	90
β/°	93.930(2)
$\gamma/^{\circ}$	90
Volume/Å <sup>3</sup>	3450.6(3)
Ζ	4
$\rho_{calc}g/cm^3$	1.644
$\mu/\text{mm}^{-1}$	1.303
F(000)	1696.0
Crystal size/mm <sup>3</sup>	$0.32 \times 0.25 \times 0.23$
Radiation	MoKα ( $\lambda$ = 0.71073)
20 range for data collection/°	4.638 to 55.278
Index ranges	$-13 \le h \le 13, -32 \le k \le 38, -14 \le l \le 14$
Reflections collected	25199
Independent reflections	7989 [ $R_{int} = 0.0430, R_{sigma} = 0.0511$ ]
Data/restraints/parameters	7989/0/504
Goodness-of-fit on F <sup>2</sup>	1.013
Final R indexes [I>= $2\sigma$ (I)]	$R_1 = 0.0438, wR_2 = 0.0884$
Final R indexes [all data]	$R_1 = 0.0797, wR_2 = 0.1010$
Largest diff. peak/hole / e Å <sup>-3</sup>	0.66/-0.95

 Table S3 Crystal data and structure refinement for compound 4d

#### Synthesis and characterization of compound 4e

1-Ethynyl-4-(trifluoromethyl)benzene **1e** (51.0 mg, 0.3 mmol) was added to a solution of tris(pentafluorophenyl)borane **2a** (153.6 mg, 0.3 mmol) and (*E*)-N-tert-butyl-1-phenylmethanimine **3a** (48.4 mg, 0.3 mmol) in toluene (2 mL). The reaction mixture was heated at 80 °C for 12 hours. After completion of reaction, all volatiles were removed *in vacuo* and the residue was purified by flash chromatography through silica gel column (petroleum ether/ethyl acetate =10:1). After drying under vacuum Compound **4e** (249.7 mg, 98% yield) was obtained as a white solid.

**M.p.**: 209.6-210.8 °C. **HRMS** (ESI, *m/z*) calcd for C<sub>38</sub>H<sub>19</sub>BF<sub>18</sub>N<sup>-</sup>: 842.1323 [M-H]<sup>-</sup>, found: 842.1371.

**IR** (KBr): 3292, 3001, 1643, 1527, 1450, 1327, 1134, 1066, 989, 837, 707 cm<sup>-1</sup>.

<sup>1</sup>**H** NMR (500 MHz,  $C_6D_6$ )  $\delta$  7.43 (m, 1H), 7.01 (m, 1H), 6.99 (d, J = 8.2 Hz, 2H), 6.81 (m, 1H), 6.77 (d, J = 8.2 Hz, 2H), 6.68 (m, 1H), 6.62 (dd, J = 28.4, 7.6 Hz, 1H), 6.42 (m, 1H), 5.52 (d, J = 7.8 Hz, 1H), 0.76 (s, 9H).

<sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, C<sub>6</sub>D<sub>6</sub>): δ 147.2, 139.8, 139.2, 130.2, 130.0, 129.53, 129.52, 128.2, 125.34, 125.32 (q, *J* = 3.7 Hz), 124.9, 74.1 (d, *J* = 6.8 Hz), 62.4, 28.3, [C<sub>6</sub>F<sub>5</sub> not listed].

<sup>11</sup>B{<sup>1</sup>H} NMR (160 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta = 0.1$  (v<sub>1/2</sub> ~ 370Hz).

<sup>19</sup>**F NMR** (470 MHz, C<sub>6</sub>D<sub>6</sub>): δ -62.8 (s, 3F), -126.3 (m, 1F), -127.2 (m, 1F), -128.1 (m, 1F), -140.0 – -140.2 (m, 2F), -144.1 (m, 1F), -153.5 (t, J = 21.1 Hz, 1F), -155.3 (t, J = 21.3 Hz, 1F), -156.1 (t, J = 20.8 Hz, 1F), -161.7 (m, 1F), -161.8 (m, 1F), -162.2 (m, 1F), -162.5 (m, 1F), -163.9 (m, 1F), -164.4 (m, 1F).

<sup>1</sup>**H**, <sup>1</sup>**H**-**GCOSY** (500 MHz / 500 MHz,  $C_6D_6$ ):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>1</sup>H = 6.62 /5.52 (NH/CH).

<sup>1</sup>**H**, <sup>13</sup>**C-GHSQC** (500 MHz / 126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>13</sup>C = 5.52/74.1 (CHN), 0.76/28.3 (*t*-Bu).

<sup>1</sup>**H**, <sup>13</sup>**C-GHMBC** (500 MHz / 126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>13</sup>C = 0.76/62.4 (*t*-Bu).



**S**30





S32





#### Synthesis and characterization of compound 4f

1-Ethynyl-4-methylbenzene **1f** (34.8 mg, 0.3 mmol) was added to a solution of tris(pentafluorophenyl)borane **2a** (153.6 mg, 0.3 mmol) and (*E*)-N-tert-butyl-1-phenylmethanimine **3a** (48.4 mg, 0.3 mmol) in toluene (2 mL). The reaction mixture was heated at 80 °C for 24 hours. After completion of reaction, all volatiles were removed *in vacuo* and the residue was purified by flash chromatography through silica gel column (petroleum ether/ethyl acetate =10:1). After drying under vacuum Compound **4f** (222.7 mg, 94% yield) was obtained as a white solid. Crystals suitable for the X-ray crystal structure analysis were obtained by slow diffusion of pentane to a solution of compound **4d** in dichloromethane at room temperature.

**M.p.**: 188.6-189.9 °C. **HRMS** (ESI, *m/z*) calcd for C<sub>38</sub>H<sub>22</sub>BF<sub>15</sub>N<sup>-</sup>: 788.1606 [M-H]<sup>-</sup>, found: 788.1604.

**IR** (KBr): 3290, 2922, 1639, 1525, 1462, 1278, 1190, 1097, 987, 775, 698 cm<sup>-1</sup>.

<sup>1</sup>**H NMR** (500 MHz, C<sub>6</sub>D<sub>6</sub>) δ 7.52 (m, 1H), 7.03 (m, 1H), 6.83 (m, 1H), 6.78 (d, *J* = 7.5 Hz, 2H), 6.75 (m, 1H), 6.67 (m, 1H), 6.61 (d, *J* = 7.5 Hz, 2H), 6.57 (m, 1H), 5.61 (d, *J* = 7.5 Hz, 1H), 1.71 (s, 3H), 0.78 (s, 9H).

<sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  148.8, 140.0, 137.7, 133.3, 130.3, 130.0, 129.3, 129.2, 129.1, 127.5, 125.3, 74.5 (d, *J* = 7.0 Hz), 62.1, 28.4, 20.8, [C<sub>6</sub>F<sub>5</sub> not listed].

<sup>11</sup>B{<sup>1</sup>H} NMR (160 MHz,  $C_6D_6$ ):  $\delta = 0.1$  (v<sub>1/2</sub> ~ 310Hz).

<sup>19</sup>**F NMR** (470 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  -126.3 (m, 1F), -127.0 (m, 1F), -127.5 (m, 1F), -140.0 – -140.3 (m, 2F), -144.1 (m, 1F), -153.9 (t, *J* = 21.1 Hz, 1F), -156.2 (t, *J* = 21.3 Hz, 1F), -156.5 (t, *J* = 20.8 Hz, 1F), -162.1 (m, 1F), -162.3 (m, 1F), -162.5 (m, 1F), -162.6 (m, 1F), -164.4 (m, 1F), -164.5 (m, 1F).

<sup>1</sup>**H**, <sup>1</sup>**H**-**GCOSY** (500 MHz / 500 MHz,  $C_6D_6$ ):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>1</sup>H = 6.67 /5.61 (NH/CH).

<sup>1</sup>**H**, <sup>13</sup>**C-GHSQC** (500 MHz / 126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>13</sup>C = 5.61/74.5 (CHN), 1.71/20.8 (Me<sup>Ph</sup>), 0.78/28.4 (*t*-Bu).

<sup>1</sup>H, <sup>13</sup>C-GHMBC (500 MHz / 126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>13</sup>C = 0.78/62.1 (*t*-Bu).



 $^{13}C$  NMR (126 MHz, C<sub>6</sub>D<sub>6</sub>) of compound 4f







S37



S38

**X-ray crystal structure analysis of compound 4f:** A suitable crystal was selected and measured on a Bruker D8 VENTURE diffractometer. The crystal was kept at 200.15 K during data collection. Using Olex2, the structure was solved with the olex2.solve structure solution program using Charge Flipping and refined with the SHELXL refinement package using Least Squares minimization.



Figure S4. Structure of compound 4f at 50% probability level

Empirical formula	$C_{38}H_{23}BF_{15}N$
Formula weight	789.38
Temperature/K	200.15
Crystal system	monoclinic
Space group	$P2_1/c$
a/Å	10.9186(4)
b/Å	14.4717(5)
c/Å	23.7175(9)
$\alpha/^{\circ}$	90
β/°	102.9300(10)
$\gamma/^{\circ}$	90
Volume/Å <sup>3</sup>	3652.6(2)
Z	4
$\rho_{calc}g/cm^3$	1.435
$\mu/\text{mm}^{-1}$	0.137
F(000)	1592.0
Crystal size/mm <sup>3</sup>	$0.31 \times 0.28 \times 0.22$
Radiation	MoKα ( $\lambda = 0.71073$ )
$2\Theta$ range for data collection/°	4.51 to 55.08
Index ranges	$-14 \le h \le 14, -18 \le k \le 18, -23 \le l \le 30$
Reflections collected	35922
Independent reflections	8377 [ $R_{int} = 0.0385, R_{sigma} = 0.0418$ ]
Data/restraints/parameters	8377/0/504
Goodness-of-fit on F <sup>2</sup>	1.031
Final R indexes [I>= $2\sigma$ (I)]	$R_1 = 0.0559, wR_2 = 0.1544$
Final R indexes [all data]	$R_1 = 0.0927, wR_2 = 0.1772$
Largest diff. peak/hole / e Å <sup>-3</sup>	0.27/-0.28

Table S4 Crystal data and structure refinement for compound 4f

#### Synthesis and characterization of compound 4g

1-(*tert*-Butyl)-4-ethynylbenzene **1g** (47.4 mg, 0.3 mmol) was added to a solution of tris(pentafluorophenyl)borane **2a** (153.6 mg, 0.3 mmol) and (*E*)-N-tert-butyl-1-phenylmethanimine **3a** (48.4 mg, 0.3 mmol) in toluene (2 mL). The reaction mixture was heated at 80 °C for 24 hours. After completion of reaction, all volatiles were removed *in vacuo* and the residue was purified by flash chromatography through silica gel column (petroleum ether/ethyl acetate =10:1). After drying under vacuum Compound **4g** (231.8 mg, 93% yield) was obtained as a white solid. Crystals suitable for the X-ray crystal structure analysis were obtained by slow diffusion of pentane to a solution of compound **4g** in dichloromethane at room temperature.

**M.p.**: 146.1-147.8 °C. **HRMS** (ESI, *m/z*) calcd for C<sub>41</sub>H<sub>28</sub>BF<sub>15</sub>N<sup>-</sup>: 830.2075 [M-H]<sup>-</sup>, found: 830.2091.

**IR** (KBr): 3284, 2974, 1649, 1527, 1448, 1288, 1184, 1093, 989, 790, 600, 559 cm<sup>-1</sup>.

<sup>1</sup>**H** NMR (500 MHz,  $C_6D_6$ )  $\delta$  7.58 (m, 1H), 7.06 (m, 1H), 6.92 (d, J = 8.0 Hz, 2H), 6.86 (d, J = 8.0 Hz, 2H), 6.84 (m, 1H), 6.74 (m, 1H), 6.66 (dd, J = 28.0, 7.0 Hz, 1H), 6.55 (m, 1H), 5.61 (d, J = 7.0 Hz, 1H), 0.90 (s, 9H), 0.79 (s, 9H).

<sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  150.9, 148.7, 140.1, 133.3, 130.5, 130.0, 129.31, 129.28, 127.6, 125.4, 125.3, 74.7 (d, *J* = 7.0 Hz), 62.1, 34.3, 31.0, 28.4, [C<sub>6</sub>F<sub>5</sub> not listed].

<sup>11</sup>B{<sup>1</sup>H} NMR (160 MHz,  $C_6D_6$ ):  $\delta = 0.1$  (v1/2 ~ 350Hz).

<sup>19</sup>**F NMR** (470 MHz,  $C_6D_6$ ):  $\delta$  -126.3 (m, 1F), -127.0 (m, 1F), -127.5 (m, 1F), -139.9 (m, 1F), -140.0 (m, 1F), -144.3 (m, 1F), -153.9 (t, *J* = 21.1 Hz, 1F), -156.2 (t, *J* = 21.3 Hz, 1F), -156.5 (t, *J* = 20.8 Hz, 1F), -162.0 (m, 1F), -162.3 (m, 1F), -162.4 (m, 1F), -162.6 (m, 1F), -164.4 (m, 1F), -164.5 (m, 1F).

<sup>1</sup>**H**, <sup>1</sup>**H**-**GCOSY** (500 MHz / 500 MHz,  $C_6D_6$ ):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>1</sup>H = 6.66 /5.61 (NH/CH).

<sup>1</sup>**H**, <sup>13</sup>**C-GHSQC** (500 MHz / 126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>13</sup>C = 5.61/74.7 (CHN), 0.90/31.0 (*t*-Bu<sup>Ph</sup>), 0.79/28.4 (*t*-Bu<sup>N</sup>).

<sup>1</sup>**H**, <sup>13</sup>**C-GHMBC** (500 MHz / 126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>13</sup>C = 0.90/34.3 (*t*-Bu<sup>Ph</sup>), 0.79/62.1 (*t*-Bu<sup>N</sup>).



60 155 150 145 140 135 130 125 120 115 110 105 100 95 90 85 80 75 70 65 60 55 50 45 40 35 30 2 <sup>13</sup>C NMR (126 MHz, C<sub>6</sub>D<sub>6</sub>) of compound 4g



<sup>11</sup>B NMR (160 MHz, C<sub>6</sub>D<sub>6</sub>) of compound 4g





S44



**X-ray crystal structure analysis of compound 4g:** A suitable crystal was selected and measured on a Bruker D8 VENTURE diffractometer. The crystal was kept at 100.00 K during data collection. Using Olex2, the structure was solved with the olex2.solve structure solution program using Charge Flipping and refined with the SHELXL refinement package using Least Squares minimization.



Figure S5. Structure of compound 4g at 50% probability level

Empirical formula	$C_{41}H_{29}BF_{15}N$
Formula weight	831.46
Temperature/K	100.00
Crystal system	triclinic
Space group	P-1
a/Å	14.4842(6)
b/Å	14.7009(6)
c/Å	21.3117(9)
$\alpha/^{\circ}$	75.764(2)
β/°	80.113(2)
γ/°	75.132(2)
Volume/Å <sup>3</sup>	4222.6(3)
Ζ	4
$\rho_{calc}g/cm^3$	1.308
$\mu/\text{mm}^{-1}$	1.079
F(000)	1688.0
Crystal size/mm <sup>3</sup>	$0.31 \times 0.25 \times 0.19$
Radiation	$CuK\alpha$ ( $\lambda = 1.54178$ )
$2\Theta$ range for data collection/°	4.306 to 136.742
Index ranges	$-17 \le h \le 17, -17 \le k \le 16, -25 \le l \le 25$
Reflections collected	70657
Independent reflections	15489 [ $R_{int} = 0.0333$ , $R_{sigma} = 0.0243$ ]
Data/restraints/parameters	15489/0/1065
Goodness-of-fit on F <sup>2</sup>	1.020
Final R indexes [I>= $2\sigma$ (I)]	$R_1 = 0.0351, wR_2 = 0.0869$
Final R indexes [all data]	$R_1 = 0.0373, wR_2 = 0.0882$
Largest diff. peak/hole / e Å <sup>-3</sup>	0.39/-0.42

Table S5 Crystal data and structure refinement for compound 4g

#### Synthesis and characterization of compound 4h

Phenyl acetylene **1a** (30.6 mg, 0.3 mmol) was added to a solution of tris(pentafluorophenyl)borane **2a** (153.6 mg, 0.3 mmol) and (*E*)-N-phenyl-1-phenylmethanimine **3b** (54.3 mg, 0.3 mmol) in toluene (2 mL). The reaction mixture was heated at 80 °C for 24 hours. After completion of reaction, all volatiles were removed *in vacuo* and the residue was purified by flash chromatography through silica gel column (petroleum ether/ethyl acetate =10:1). After drying under vacuum Compound **4h** (228.6 mg, 96% yield) was obtained as a white solid.

**M.p.**: 228.1-229.6 °C. **HRMS** (ESI, *m/z*) calcd for C<sub>39</sub>H<sub>16</sub>BF<sub>15</sub>N<sup>-</sup>: 794.1136 [M-H]<sup>-</sup>, found: 796.1168.

**IR** (KBr): 3267, 1648, 1523, 1452, 1282, 1095, 977, 794, 702 cm<sup>-1</sup>.

<sup>1</sup>**H** NMR (500 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  8.15 (m, 1H), 7.10-7.04 (m, 4H), 6.88-6.82 (m, 4H), 6.80 (m, 1H), 6.75 (m, 1H), 6.73-6.68 (m, 2H), 6.60-6.54 (m, 3H), 5.81 (d, *J* = 7.5 Hz, 1H).

<sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, C<sub>6</sub>D<sub>6</sub>): δ 146.7, 138.5, 135.9, 135.6, 130.1, 129.7, 129.5, 128.59, 128.56, 128.54, 128.4, 128.1, 121.6, 78.9, [C<sub>6</sub>F<sub>5</sub> not listed].

<sup>11</sup>B{<sup>1</sup>H} NMR (160 MHz,  $C_6D_6$ ):  $\delta = 2.1$  (v1/2 ~ 420 Hz).

<sup>19</sup>**F NMR** (470 MHz, C<sub>6</sub>D<sub>6</sub>): δ -127.9 (br, 1F), -129.8 (br, 1F), -131.8 (br, 1F), -139.0 (m, 1F), -139.8 (br, 1F), -140.6 (m, 1F), -155.9 (t, J = 20.8 Hz, 1F), -157.1 (t, J = 20.8 Hz, 1F), -157.3 (t, J = 21.4 Hz, 1F), -162.5 (br, 1F), -163.0 (br, 1F), -163.5 (m, 1F), -163.8 (m, 1F), -164.0 (br, 1F), -164.6 (br, 1F).

<sup>1</sup>**H**, <sup>1</sup>**H**-**GCOSY** (500 MHz / 500 MHz,  $C_6D_6$ ):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>1</sup>H = 8.15/5.81 (NH/CH).

<sup>1</sup>**H**, <sup>13</sup>**C-GHSQC** (500 MHz / 126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>13</sup>C = 5.81/78.9 (CHN).



 $^{13}C$  NMR (126 MHz,  $C_6D_6)$  of compound 4h



<sup>11</sup>B NMR (160 MHz, C<sub>6</sub>D<sub>6</sub>) of compound 4h







S51

#### Synthesis and characterization of compound 4i

Styrene (31.2 mg, 0.3 mmol) was added to a solution of bis(pentafluorophenyl)borane (103.8 mg, 0.3 mmol) in toluene (2 mL) and the resulting mixture was stirred at room temperature for 30 mins. Then (*E*)-N-tert-butyl-1-phenylmethanimine **3a** (48.4 mg, 0.3 mmol) and phenyl acetylene **1a** (30.6 mg, 0.3 mmol) were added. The reaction mixture was heated at 80 °C for 12 hours. After completion of reaction, all volatiles were removed *in vacuo* and the residue was purified by flash chromatography through silica gel column (petroleum ether/ethyl acetate =10:1). After drying under vacuum Compound **4i** (130.5 mg, 61% yield) was obtained as a white solid. Crystals suitable for the X-ray crystal structure analysis were obtained by slow diffusion of pentane to a solution of compound **4i** in dichloromethane at room temperature.

**HRMS** (ESI, m/z) calcd for C<sub>39</sub>H<sub>29</sub>BF<sub>10</sub>N<sup>-</sup>: 712.2233 [M-H]<sup>-</sup>, found: 712.2243.

**IR** (KBr): 3302, 2924, 1643, 1581, 1453, 1280, 1186, 1091, 1082, 954, 700 cm<sup>-1</sup>.

<sup>1</sup>**H** NMR (500 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  7.05-6.83 (m, 15H), 6.45 (dd, *J* = 30.5, 6.6 Hz, 1H), 5.26 (d, *J* = 6.8 Hz, 1H), 2.61-2.26 (m, 4H), 0.75 (s, 9H).

<sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, C<sub>6</sub>D<sub>6</sub>): δ 142.9, 142.1, 141.2, 137.8, 129.3, 129.0, 128.7, 128.4, 128.3, 127.1, 126.1, 74.6, 61.3, 36.9, 32.9, 28.4, [C<sub>6</sub>F<sub>5</sub> not listed].

<sup>11</sup>**B**{<sup>1</sup>**H**} **NMR** (160 MHz,  $C_6D_6$ ):  $\delta = 0.9 (v_{1/2} \sim 380 \text{ Hz}).$ 

<sup>19</sup>**F NMR** (470 MHz, C<sub>6</sub>D<sub>6</sub>): δ -124.5 (m, 1F), -125.1 (m, 1F), -129.8 (m, 1F), -139.2 (m, 1F), -154.6 (t, *J* = 21.2 Hz, 1F), -157.3 (t, *J* = 21.2 Hz, 1F), -162.1 – -162.3 (m, 2F), -162.7 (m, 1F), -163.5 (m, 1F).

<sup>1</sup>**H**, <sup>1</sup>**H**-**GCOSY** (500 MHz / 500 MHz,  $C_6D_6$ ):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>1</sup>H = 6.45/5.26 (NH/CH).

<sup>1</sup>**H**, <sup>13</sup>**C-GHSQC** (500 MHz / 126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>13</sup>C = 5.26/74.6 (CHN), 2.61-2.26/36.9, 32.9 (CH<sub>2</sub>CH<sub>2</sub>), 0.75/28.4 (*t*-Bu).

<sup>1</sup>**H**, <sup>13</sup>**C-GHMBC** (500 MHz / 126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  <sup>1</sup>H /  $\delta$  <sup>13</sup>C = 0.75/61.3 (*t*-Bu).



55 150 145 140 135 130 125 120 115 110 105 100 95 90 85 80 75 70 65 60 55 50 45 40 35 30 25 2 <sup>13</sup>C NMR (126 MHz, C<sub>6</sub>D<sub>6</sub>) of compound 4i



 $^{11}B$  NMR (160 MHz, C<sub>6</sub>D<sub>6</sub>) of compound 4i



S54



S55



**X-ray crystal structure analysis of compound 4i:** A suitable crystal was selected and measured on a Bruker D8 VENTURE diffractometer. The crystal was kept at 100.00 K during data collection. Using Olex2, the structure was solved with the olex2.solve structure solution program using Charge Flipping and refined with the SHELXL refinement package using Least Squares minimization.



Figure S6. Structure of compound 4i at 50% probability level

Empirical formula	C <sub>39</sub> H <sub>30</sub> BF <sub>10</sub> N
Formula weight	713.45
Temperature/K	100.00
Crystal system	triclinic
Space group	P-1
a/Å	9.8553(3)
b/Å	12.1888(3)
c/Å	14.5385(5)
α/°	110.891(2)
β/°	91.771(2)
$\gamma/^{\circ}$	95.1840(10)
Volume/Å <sup>3</sup>	1621.20(9)
Z	2
$ ho_{calc}g/cm^3$	1.462
$\mu/\text{mm}^{-1}$	1.076
F(000)	732.0
Crystal size/mm <sup>3</sup>	$0.41 \times 0.39 \times 0.32$
Radiation	$CuK\alpha$ ( $\lambda = 1.54178$ )
$2\Theta$ range for data collection/°	6.522 to 133.174
Index ranges	$-11 \le h \le 11, -14 \le k \le 14, -17 \le l \le 17$
Reflections collected	24199
Independent reflections	5735 [ $R_{int} = 0.0331$ , $R_{sigma} = 0.0293$ ]
Data/restraints/parameters	5735/0/467
Goodness-of-fit on F <sup>2</sup>	1.043
Final R indexes [I>= $2\sigma$ (I)]	$R_1 = 0.0348, wR_2 = 0.0859$
Final R indexes [all data]	$R_1 = 0.0383, wR_2 = 0.0879$
Largest diff. peak/hole / e Å <sup>-3</sup>	0.27/-0.23

Table S6 Crystal data and structure refinement for compound 4i

#### **Deuterium Labelling experiment:**

#### synthesis and characterization of compound 4i-D

Styrene (31.2 mg, 0.3 mmol) was added to a solution of bis(pentafluorophenyl)borane (103.8 mg, 0.3 mmol) in toluene (2 mL) and the resulting mixture was stirred at room temperature for 30 mins. Then (*E*)-N-tert-butyl-1-phenylmethanimine **3a** (48.4 mg, 0.3 mmol) and terminal D-phenyl acetylene **1a-D** (30.9 mg, 0.3 mmol) were added. The reaction mixture was heated at 80 °C for 12 hours. After completion of reaction, all volatiles were removed *in vacuo* and the residue was purified by flash chromatography through silica gel column (petroleum ether/ethyl acetate =10:1). After drying under vacuum Compound **4i-D** (134.9 mg, 63% yield) was obtained as a white solid.

**HRMS** (ESI, *m/z*) calcd for C<sub>39</sub>H<sub>30</sub>DBF<sub>10</sub>N<sup>+</sup>: 715.2447 [M+H]<sup>+</sup>, found: 715.2440.

<sup>1</sup>**H NMR** (500 MHz, C<sub>6</sub>D<sub>6</sub>): δ7.05-6.83 (m, 15H), 5.26 (s, 1H), 2.61-2.26 (m, 4H), 0.75 (s, 9H).





.5 7.4 7.3 7.2 7.1 7.0 6.9 6.8 6.7 6.6 6.5 6.4 6.3 6.2 6.1 6.0 5.9 5.8 5.7 5.6 5.5 5.4 5.3 5.2 5.1 <sup>1</sup>H NMR (500 MHz, C<sub>6</sub>D<sub>6</sub>) of compound 4i and 4i-D

# Trials with internal alkynes, other imines and synthetic application of compound 4a

Trials with internal alkynes:

other imines:

Ph 
$$\longrightarrow$$
 H + B(C<sub>6</sub>F<sub>5</sub>)<sub>3</sub> +  $\overset{Me}{\bigvee}_{Ph} \xrightarrow{80 \circ C}$  N.D.  
Ph  $\longrightarrow$  H + B(C<sub>6</sub>F<sub>5</sub>)<sub>3</sub> +  $\overset{tBu}{\bigvee}_{Hu} \xrightarrow{N}_{tBu} \xrightarrow{80 \circ C}$  N.D.

Trials for synthetic application of compound **4a**:

refs: Org. Lett., 2011, 13, 62-65; Org. Lett., 2015, 17, 3682–3685.



The compound **4a** (139.5 mg, 0.18 mmol) was dissolved in THF (10 mL) under argon atmosphere in a Schlenk tube. To this solution  $Pd(PPh_3)_4$  (20 mg, 0.02 mmol, 10%), PhI (0.1 mL) and NaOH (3 M in water, 3 mL) were added. The two-phase mixture was heated to 70 °C for 20 h. After the reaction was cooled to room temperature, water was

added to the mixture. The organic phase was separated and the aqueous phase was extracted with EA. All the organic phases were combined, dried over Na<sub>2</sub>SO<sub>4</sub> and filtrated. The filtrate was dried and the residue was purified by chromatography (SiO<sub>2</sub>) with PE:EA = 60:1 as the eluent. The product was obtained as a colorless liquid (68.2 mg, 88%). <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>)  $\delta$  7.44 – 7.40 (m, 2H), 7.37 – 7.33 (m, 2H), 7.31 – 7.26 (m, 1H), 7.20 – 7.12 (m, 4H), 6.93 – 6.88 (m, 2H), 4.80 (s, 1H), 1.27 (br s, 1H), 1.24 (s, 9H); <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>)  $\delta$  154.91, 143.33, 139.61, 128.82, 128.14, 128.12, 127.93, 127.65, 127.33, 112.34, 63.45, 52.22, 29.97, [C6F5 not listed]; <sup>19</sup>F NMR (565 MHz, CDCl<sub>3</sub>)  $\delta$  -139.37 (m, 2F), -157.35 (t, *J* = 20.8 Hz, 1F), -163.37 – -163.51 (m, 2F); HRMS (ESI, m/z) calcd for C<sub>25</sub>H<sub>23</sub>F<sub>5</sub>N<sup>+</sup>: 432.1745 [M+H]<sup>+</sup>, found: 432.1758.





<sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>) of compound 4a-deB

