# **Supporting Information for**

# **Thiophene/Selenophene-Based Windmill-Shaped Triple**

# [6]Helicenes

Wan Xu, Jichang Wei, Kun Tang, Zhiying Ma, Jianwu Shi, Chunli Li\* and Hua Wang\*

Engineering Research Center for Nanomaterials, Henan University, Kaifeng, 475004, China

\*Corresponding author: Chunli Li (Email: chunli79@126.com); Hua Wang (Email:

hwang@henu.edu.cn).

# Table of Contents

1. Experimental Section	S3
General procedures and materials.	S3
Synthesis of <b>3</b>	S4
Synthesis of (TMS) <sub>2</sub> - <b>DTS</b>	S4
Synthesis of <b>4b</b>	S4
Synthesis of <b>5b</b>	S5
Synthesis of <b>6a</b> .	S5
Synthesis of <b>6b</b>	S6
Synthesis of <b>6c</b>	S6
Synthesis of 6d.	S6
Synthesis of <i>rac</i> -1a	S7
Synthesis of <i>rac</i> -1b	S7
Synthesis of <i>rac</i> -1c	S8
Synthesis of <i>rac</i> -1d	S8
2. NMR and HRMS Spectra	S8
NMR and HRMS spectra of <b>3</b> .	S8
NMR and HRMS spectra of (TMS) <sub>2</sub> -DTS	S10
NMR and HRMS spectra of 4b.	S11

NMR and HRMS spectra of <b>5b</b>	S13
NMR and HRMS spectra of <b>6a</b> .	S14
NMR and HRMS spectra of <b>6b</b>	S16
NMR and HRMS spectra of <b>6c</b> .	S17
NMR and HRMS spectra of 6d.	S18
NMR and HRMS spectra of <i>rac</i> -1a.	S19
NMR and HRMS spectra of <i>rac</i> -1b.	S20
NMR and HRMS spectra of <i>rac</i> -1c	S22
NMR and HRMS spectra of <i>rac</i> -1d.	S23
3. Quantum Calculation of <i>rac</i> -1(a)–(d)	S25
UV-vis spectra prediction and transition nature <i>rac</i> -1(a)–(d)	S25
4. Resolution of <i>rac</i> -1c	S27
HPLC trace of (+)-1c, (-)-1c, and <i>rac</i> -1c	S27
Racemization of (-)-1c	S27
5. X-ray Crystallographic Data	S30
X-ray crystallographic data of <i>rac</i> -1a	S30
X-ray crystallographic data of <i>rac</i> -1b	
X-ray crystallographic data of <i>rac</i> -1c	
X-ray crystallographic data of <i>rac</i> -1d	S36
6. References	S41

#### 1. Experimental Section

#### **General Procedures and Materials**

Ether and tetrahydrofuran (THF) for use on vacuum line were freshly distilled from sodium/benzophenone prior to use. *t*-BuLi (pentane) and *n*-BuLi (hexane) were obtained from Energy Chemical, and their concentrations were determined by titration with *N*-pivaloyl-o-toluidine.<sup>1</sup> Column chromatography was carried out on silica gel (300–400 mesh). Analytical thin-layer chromatography was performed on glass plates of silica gel GF-254 with detection by UV. Standard techniques for synthesis under inert atmosphere, using gasbag and Schlenk glassware equipped with an 8 mm PTFE vacuum stopcock, were employed. All starting materials and reagents were commercially available.

<sup>1</sup>H NMR and <sup>13</sup>C NMR spectra were recorded on 300 or 400 or 600 MHz NMR instruments using CDCl<sub>3</sub> ( $\delta$  H (7.26 ppm) and CDCl<sub>3</sub>  $\delta$  C (77.00 ppm))as solvent. IR spectra were obtained using an FT-IR instrument. HRMS analysis was carried out on a mass spectrometer equipped with EI<sup>+</sup>, DART-FTICR and DART-Positive. Melting point determination was taken on a Melt-Temp apparatus and was uncorrected. Ultraviolet-visible (UV-vis) absorption spectra were recorded on PE Lambda 950 equipment. The steady-state photoluminescence (PL) spectra and the fluorescence measurements for all samples were recorded on a JY HORIBA FluoroLog-3. CD spectrum was recorded on Aviv Biomedical Inc Model 420SF. Cyclic voltammetry (CV) behaviors were recorded on IM6eX equipment with three electrode cell in the solution of 0.1 M tetrabutylammonium hexafluorophosphate (Bu<sub>4</sub>NPF<sub>6</sub>) dissolved in CH<sub>2</sub>Cl<sub>2</sub>. Platinum electrode (0.6 cm<sup>2</sup>), Pt wire and Ag/AgCl electrode were used as the working electrode, counter electrode and reference electrode, respectively. The X-ray crystallographic analyses were performed using crystals of compounds rac-1(a)-(d) with sizes of  $0.48 \times 0.23 \times 0.22$ ,  $0.17 \times 0.09 \times 0.04$ ,  $0.1611 \times 0.1222 \times 0.0669$  and 0.1525 $\times$  0.0577  $\times$  0.0404 mm<sup>3</sup>, respectively. The intensity data were collected with the  $\omega$  scan mode (296, 298, 293, 293 K) on a diffractometer with CCD detector using Cu K $\alpha$  radiation ( $\lambda = 1.54178$ , 1.54184 Å). The data were corrected for Lorentz and polarization effects, and absorption corrections were performed using SADABS program.<sup>2</sup> The crystal structures were solved using the SHELXTL program and refined using full-matrix least-squares.<sup>3</sup> Further details were in the deposited CIF files. Slow evaporation of solutions of rac-1(a)-(d) in CHCl<sub>3</sub>/CH<sub>3</sub>OH (3/1, 2/1, 3/2, 1/1, v/v) were employed for growing single crystals. The fluorescence quantum yields ( $\Phi_F$ ) of *rac*-1(**a**)–(**d**) were characterized in dichloromethane with quinine sulfate in 0.1 N H<sub>2</sub>SO<sub>4</sub> as the control. The predicted absorption spectra were computed within the TD-DFT/PCM approaches based on B3LYP/6-31G\* level. HOMO and LUMO distributions and isomerization barriers of *rac*-1(**a**)–(**d**) (i.e., the TMS replaced by hydrogen atom) were at the B3LYP/6-31G\* level.

#### Synthesis of 2,7-Bis(trimethylsilyl)-[1,2]diselenino[3,4-b:6,5-b']dithiophene (3)

*n*-BuLi (2.38 M in hexane, 6.12 mL, 14.56 mmol, 2.05 equiv) was added dropwise into a solution of **2** (3.10 g, 6.62 mmol) in Et<sub>2</sub>O (100 mL) at -78 °C, and then the reaction mixture was kept for 2 h. Dry Se (metals basis) (2.61 g, 33.1 mmol, 5.0 equiv) was added at -78 °C; the reaction mixture was kept at -78 °C for 2 h and -55 °C for 2 h, and then warmed up slowly to ambient temperature overnight. The reaction mixture was quenched with methanol at -78 °C, extracted with dichloromethane (3 × 25 mL), and then washed with water (3 × 30 mL). After drying over anhydrous MgSO<sub>4</sub>, the solvent was removed under vacuum. The residue was purified by column chromatography on silica gel with petrol ether (60–90 °C) as eluent to yield **3** (1.02 g, 33%) as a red solid. Mp: 137–138 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.30 (s, 2H), 0.35 (s, 18H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  142.3, 142.2, 132.3, 119.2, -0.1. HRMS (DART-Positive) *m/z* [M + H]<sup>+</sup> calcd for C<sub>14</sub>H<sub>21</sub>S<sub>2</sub>Se<sub>2</sub>Si<sub>2</sub> 468.8948; found 468.8950. IR (KBr): 3050, 2947, 1252, 986, 830 cm<sup>-1</sup>.

## Synthesis of 2,5-Bis(trimethylsilanyl)dithieno[2,3-b:3',2'-d]selenophene((TMS)<sub>2</sub>-DTS)

Compound **3** (300 mg, 0.64 mmol) and Cu nanopowder (200 mg, 3.22 mmol, 5.0 equiv) were heated to 145 °C without use of any solvent for 0.5 h. The residue was purified by column chromatography on silica gel with petrol ether (60–90 °C) as eluent to yield (TMS)<sub>2</sub>-**DTS** (120.7 mg, 50%) as a yellow liquid. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.51 (s, 2H), 0.36 (s, 18H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  145.6, 141.6, 141.2, 126.4, 0.0. HRMS (DART-Positive) m/z [M + H]<sup>+</sup> calcd for C<sub>14</sub>H<sub>21</sub>SeS<sub>2</sub>Si<sub>2</sub> 388.9783; found 388.9785. IR (KBr): 3061, 2978, 1251, 963, 845 cm<sup>-1</sup>.

#### Synthesis of 5-Bromo-2-(trimethylsilanyl)dithieno[2,3-b:3',2'-d]selenophene (4b)

 $(TMS)_2$ -**DTS** (605 mg, 1.56 mmol) was dissolved in CHCl<sub>3</sub> (20 mL), and NBS (333 mg, 1.87 mmol, 1.2 equiv) was dissolved in 16 mL of HOAc/CHCl<sub>3</sub> (v/v = 1/8) and added dropwise under good stirring at 0 °C in an ice-water bath. After stirring for 2 h, the reaction mixture was quenched with H<sub>2</sub>O, extracted with dichloromethane (3 × 10 mL), and then washed with saturated NaHCO<sub>3</sub> (20

mL) and water (2 × 20 mL). The organic layer was dried over anhydrous MgSO<sub>4</sub>. After the solvent was removed under vacuum, the residue was washed with 3 × 5 mL hexane to yield **4b** (326.2 mg, 53%) as a yellow liquid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.44 (s, 1H), 7.41 (s, 1H), 0.37 (s, 9H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  146.3, 142.0, 140.7, 139.0, 134.7, 126.2, 123.5, 113.2, -0.1. HRMS (EI<sup>+</sup>) *m/z* [M<sup>+</sup>] calcd for C<sub>11</sub>H<sub>11</sub>BrS<sub>2</sub><sup>74</sup>SeSi 387.8480; found 387.8478. IR (KBr): 3084, 2963, 1259, 910, 842 cm<sup>-1</sup>.

#### Synthesis of 5-(trimethylsilyl)dithieno[2,3-b:3',2'-d]selenophene-2-carbaldehyde (5b)

*n*-BuLi (2.45 M in hexane, 0.98 mL, 2.40 mmol, 1.05 equiv) was added dropwise to **4b** (900 mg, 2.28 mmol) in THF (30 mL) at -78 °C. After 2 h at -78 °C, DMF (0.36 mL, 4.57 mmol, 2.0 equiv) was added at -78 °C and kept 0.5 h, and then the reaction mixture was warmed up slowly to ambient temperature overnight. The reaction mixture was quenched with H<sub>2</sub>O and extracted with dichloromethane (3 × 15 mL). The organic layer was washed with saturated NaHCO<sub>3</sub> (20 mL) and water (2 × 20 mL) and then dried over anhydrous MgSO<sub>4</sub>. The residue was purified by column chromatography on silica gel with petrol ether (60–90 °C) Hexane/CH<sub>2</sub>Cl<sub>2</sub> (v/v = 2/1) as eluent to yield **5b** (243 mg, 31%) as a brown solid. Mp: 124–125 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  9.89 (s, 1H), 8.01 (s, 1H), 7.54 (s, 1H), 0.39 (s, 9H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  182.3, 147.7, 147.5, 147.2, 141.9, 141.6, 140.0, 128.7, 126.1, –0.1. HRMS (DART-FTICR) *m/z* [M + H]<sup>+</sup> calcd for C<sub>12</sub>H<sub>13</sub>OS<sub>2</sub>SeSi 344.9337; found 344.9337. IR (KBr): 3091, 2938, 1656, 993, 849 cm<sup>-1</sup>.

#### 1,3,5-tris(2-(5-(trimethylsilyl)dithieno[2,3-b:3',2'-d]thiophene-2-yl)vinyl)benzene (6a)

*n*-BuLi (2.5 M in hexane, 0.29 mL, 0.74 mmol, 3.5 equiv) was added dropwise into the phosphonium salt (241 mg, 0.21 mmol, 1.0 equiv) in THF at -78 °C for 2 h, then a solution of 5- (trimethylsilanyl)dithieno[2,3-*b*:3',2'-*d*]thiophene-2-carbaldehyde **5a** (200 mg, 0.67 mmol, 3.2 equiv) in dry THF (5 mL) was added at -78 °C. After keeping stirring for 2 h, the reaction mixture was warmed up slowly to ambient temperature overnight. The reaction was quenched with water at -78 °C, extracted with dichloromethane (3 × 15 mL), washed with H<sub>2</sub>O (3 × 25 mL), and then dried over anhydrous MgSO<sub>4</sub>. After the solvent was removed in vacuum, the residue was purified by column chromatography on silica gel with petroleum ether (60–90 °C) Hexane/CH<sub>2</sub>Cl<sub>2</sub>(v/v = 1/1) as eluent to give 1,3,5-tris(2-(5-(trimethylsilyl)dithieno[2,3-*b*:3',2'-*d*]thiophene-2-yl)vinyl)-benzene **6a** (100.3 mg, 50%) as a yellow solid. Mp: 157–159 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$ 

7.45 (s, 6H), 7.32 (t, J = 8.0 Hz, 6H), 6.91 (d, J = 16.0 Hz, 3H), 0.40 (s, 27H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  145.2, 144.6, 143.7, 140.6, 138.4, 137.8, 137.5, 127.4, 124.9, 123.5, 123.0, 118.1, -0.1. HRMS (MALDI MS EI<sup>+</sup>) m/z [M<sup>+</sup>] calcd for C<sub>45</sub>H<sub>42</sub>S<sub>9</sub>Si<sub>3</sub> 954.0081; found 954.0075. IR (KBr) 3017, 2955, 1259, 941, 833 cm<sup>-1</sup>.

#### 1,3,5-tris(2-(5-(trimethylsilyl)dithieno[2,3-b:3',2'-d]selenophene-2-yl)vinyl)benzene (6b)

Compound **6b** was synthesized according to the procedure for making **6a**. From the reaction on the 200 mg scale of **5b**, 27.9 mg (14%) of **6b** was obtained as yellow solid. Mp: >300 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) (for a mixture of *cis* and *trans* isomers of **6b**)  $\delta$  7.51 (d, *J* = 8.0 Hz), 7.49 (d, *J* = 6.0 Hz), 7.45 (d, *J* = 1.6 Hz), 7.39 (d, *J* = 7.2 Hz), 7.33(s), 7.27(s), 6.97 (s), 6.93 (t, *J* = 8.0 Hz), 6.83 (d, *J* = 8.0 Hz), 6.63 (d, *J* = 12.0 Hz), 0.39 (s), 0.38 (s), 0.33 (s); the ratio of integral areas of the peaks are 1 : 1 : 0.5 : 1.5 : 0.5 : 0.5 : 0.5 : 1 : 0.3 : 0.3 : 7 : 3 : 1. HRMS (AP-MALDI- Positive) *m/z* [M<sup>+</sup>] calcd for C<sub>45</sub>H<sub>42</sub>S<sub>6</sub><sup>80</sup>Se<sub>3</sub>Si<sub>3</sub> 1097.8409; found 1097.8435. IR (KBr) 3029, 2954, 1259, 941, 834 cm<sup>-1</sup>.

#### 1,3,5-tris(2-(5-(trimethylsilyl)diseleno[2,3-b:3',2'-d]thiophene-2-yl)vinyl)benzene (6c)

Compound **6c** was synthesized according to the procedure for making **6a**. From the reaction on the 200 mg scale of **5c**, 21.7 mg (11%) of **6c** was obtained as yellow solid. Mp: 258–260 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) (for a mixture of *cis* and *trans* isomers of **6c**)  $\delta$  7.74 (s), 7.71 (d, *J* = 4.4 Hz), 7.56 (d, *J* = 6.0 Hz), 7.49 (d, *J* = 3.6 Hz), 7.43 (d, *J* = 11.2 Hz), 7.37 (d, *J* = 7.2 Hz), 6.93 (d, *J* = 11.6 Hz), 6.83 (d, *J* = 6.0 Hz), 6.79 (d, *J* = 6.0 Hz), 6.66 (d, *J* = 10.8 Hz), 0.39 (s), 0.38 (s), 0.35 (s); the ratio of integral areas of the peaks are 1 : 0.5 : 1 : 1 : 1 : 0.15 : 0.85 : 0.85 : 0.15 : 9 : 3 : 3. HRMS (AP-MALDI-Positive) *m/z* [M<sup>+</sup>] calcd for C<sub>45</sub>H<sub>42</sub>S<sub>3</sub><sup>80</sup>Se<sub>6</sub>Si<sub>3</sub> 1241.6742; found 1241.6768. IR (KBr) 3015, 2950, 1245, 908, 833 cm<sup>-1</sup>.

#### 1,3,5-tris(2-(5-(trimethylsilyl)diseleno[2,3-b:3',2'-d]selenophene-2-yl)vinyl)benzene (6d)

Compound **6d** was synthesized according to the procedure for making **6a**. From the reaction on the 200 mg scale of **5d**, 19.7 mg (10%) of **6d** was obtained as yellow solid. Mp: 280–282 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) (for a mixture of *cis* and *trans* isomers of **6d**)  $\delta$  7.76 (s), 7.74 (d, *J* = 4.0 Hz), 7.59 (d, *J* = 6.0 Hz), 7.53 (d, *J* = 11.6 Hz), 7.48 (s), 7.43 (d, *J* = 10.4 Hz), 7.38 (d, *J* = 4.4 Hz), 7.33(s), 6.93 (d, *J* = 11.6 Hz), 6.83 (d, *J* = 4.8 Hz), 6.79 (d, *J* = 4.4 Hz), 6.65 (d, *J* = 11.6 Hz), 0.39 (s), 0.37 (s), 0.35 (s); the ratio of integral areas of the peaks are 1 : 1 : 1 : 1 : 1 : 1 : 0.7 : 0.3 : 0.3 :

0.7 : 0.7 : 0.3 : 9 : 6 : 3. HRMS (AP-MALDI-Positive) *m*/*z* [M<sup>+</sup>] calcd for C<sub>45</sub>H<sub>42</sub><sup>80</sup>Se<sub>9</sub>Si<sub>3</sub> 1385.5076; found 1385.5099. IR (KBr) 3010, 2950, 1245, 911, 832 cm<sup>-1</sup>.

# 1,3,5-tris(trimethylsilyl)benzo[1,2-b:3,4-b']-tris(dithieno[2,3-b:3',2'-d]thiophene)benzene (rac-1a)

Iodine (22 mg, 0.088 mmol, 3.0 equiv) and 2-methyloxirane (0.1 mL) were added into solution of **6a** (28 mg, 0.029 mmol) in dry toluene (25 mL). The reaction solution was irradiated with a 450 W unfiltered Hg medium pressure lamp. The reaction was monitored by thin layer chromatography every hour, and irradiation was stopped when the **6a** was consumed. The reaction was quenched with saturated Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> (10 mL). The reaction mixture was extracted with dichloromethane (3 × 10 mL) and washed with H<sub>2</sub>O (3 × 10 mL) and then dried over anhydrous MgSO<sub>4</sub>. After the solvent was removed in vacuum, the crude product was purified by PTLC with petroleum ether (60–90 °C) Hexane/CH<sub>2</sub>Cl<sub>2</sub> (v/v = 2/1) as developer to yield *rac*-**1a** (10.0 mg, 35%) as a light yellow solid. Mp: 258–260 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.47 (d, *J* = 8.8 Hz, 3H), 7.77 (d, *J* = 8.8 Hz, 3H), 6.53 (s, 3H), 0.10 (s, 27H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  143.1, 143.0, 142.0, 141.6, 141.6, 135.1, 130.2, 127.6, 127.5, 127.2, 126.6, 119.2, -0.3. HRMS (MALDI MS EI<sup>+</sup>) *m/z* [M<sup>+</sup>] calcd for C<sub>45</sub>H<sub>36</sub>S<sub>9</sub>S<sub>13</sub> 947.9611, found 947.9606. IR (KBr) 3074, 2953, 2920, 1253, 979, 838 cm<sup>-1</sup>.

# 1,3,5-tris(trimethylsilyl)benzo[1,2-b:3,4-b']-tris(dithieno[2,3-b:3',2'-d]selenophene)benzene (rac-1b)

*Rac*-1**b** was synthesized according to the procedure for making *rac*-1**a**. From the reaction on the 32.5 mg scale of **6b**, 6.5 mg (20%) of *rac*-1**b** was obtained as yellow solid. Mp: > 300 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.35 (d, *J* = 8.8 Hz, 3H), 7.72 (d, *J* = 8.4 Hz, 3H), 6.51 (s, 3H), 0.02 (s, 27H). <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$  144.2, 143.5, 142.4, 140.5, 140.2, 136.7, 131.8, 128.3, 127.9, 127.6, 126.9, 118.8, -0.4. HRMS (AP-MALDI-Positive) *m/z* [M<sup>+</sup>] calcd for C<sub>45</sub>H<sub>36</sub>S<sub>6</sub>Se<sub>3</sub>Si<sub>3</sub> 1091.7049; found 1091.7936. IR (KBr) 3061, 2963, 1265, 970, 834 cm<sup>-1</sup>.

# 1,3,5-tris(trimethylsilyl)benzo[1,2-b:3,4-b']-tris(diseleno[2,3-b:3',2'-d]thiophene)benzene (rac-1c)

*Rac*-1c was synthesized according to the procedure for making *rac*-1a. From the reaction on the 35.5 mg scale of 6c, 3.5 mg (10%) of *rac*-1c was obtained as yellow solid. Mp: 290–292 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.27 (d, *J* = 8.8 Hz, 3H), 7.69 (d, *J* = 8.8 Hz, 3H), 6.68 (s, 3H), -0.05

(s, 27H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>)  $\delta$  147.2, 145.4, 144.7, 142.8, 139.3, 139.2, 132.7, 129.4, 128.8, 128.7, 126.6, 122.0, -0.2. HRMS (DART-Positive) m/z [M<sup>+</sup>] calcd for C<sub>45</sub>H<sub>36</sub>S<sub>3</sub>Se<sub>6</sub>Si<sub>3</sub> 1235.6278; found 1235.6322. IR (KBr) 3080, 2957, 1251, 944, 833 cm<sup>-1</sup>.

1,3,5-tris(trimethylsilyl)benzo[1,2-b:3,4-b']-tris(diseleno[2,3-b:3',2'-d]selenophene)benzene (rac-1d)

*Rac*-1d was synthesized according to the procedure for making *rac*-1a. From the reaction on the 37.3 mg scale of 6d, 3.0 mg (8%) of *rac*-1d was obtained as yellow solid. Mp: >300 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.15 (d, *J* = 8.4 Hz, 3H), 7.65 (d, *J* = 8.4 Hz, 3H), 6.71 (s, 3H), -0.16 (s, 27H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>)  $\delta$  148.4, 146.9, 145.8, 141.2, 141.2, 138.6, 133.9, 130.5, 128.9, 128.4, 126.5, 121.6, -0.4. HRMS (DART-Positive) *m/z* [M<sup>+</sup>] calcd for C<sub>45</sub>H<sub>36</sub>Se<sub>9</sub>Si<sub>3</sub> 1379.4612; found 1379.4645. IR (KBr) 3070, 2950, 1251, 935, 833 cm<sup>-1</sup>.

## 2. NMR and HRMS Spectra





Fig. S1. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) spectrum of 3



210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10

Fig. S2. <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) spectra of 3



Fig. S3. HRMS spectra of 3

NMR and HRMS spectra of ((TMS)<sub>2</sub>-DTS)



Fig. S4. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) spectrum of ((TMS)<sub>2</sub>-DTS)



Fig. S5. <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) spectra of ((TMS)<sub>2</sub>-DTS)



Fig. S6. HRMS spectra of ((TMS)<sub>2</sub>-DTS)





Fig. S7. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) spectrum of 4b





Fig. S8. <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) spectrum of 4b

lChi	National Center for Organic Mass Spectrometry in Shanghai Shanghai Institute of Organic Chemistry Chinese Academic of Sciences High Resolution MS Data Report					
Instrume	nt: Waters Mic	cromass	GCT Pre	emier	Ionisation	Mode: EI <sup>+</sup>
Electron	Energy: 70eV	Ca	ard Seria	l Numb	er: GCT-P-	Г19-ОЅ-866
Operator	:: Li	D	ate: 2019	9/11/18		
Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula
387.8478	387.8480	-0.2	-0.5	8.0	5546026.5	C11H11SiS274SeBr
	387.8473	0.5	1.3	9.0	5546026.5	C11H7O4S74SeBr
	387.8472	0.6	1.5	19.0	5546025.5	C18HSiS2Br
	387.8466	1.2	3.1	18.5	5546026.5	C17HNO74SeBr
	387.8498	-2.0	-5.2	13.5	5546026.5	C13H5NO2Si74SeBr

Fig. S9. HRMS data of 4b

## NMR and HRMS spectra of 5b



**Fig. S11.** <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) spectra of **5b** 



Fig. S12. HRMS spectra of 5b





Fig. S13. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) spectra of 6a



200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10

## Fig. S14. <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) spectra of 6a



Fig. S15. HRMS data of 6a

#### NMR and HRMS spectra of 6b







Fig. S17. HRMS spectrum of 6b

#### NMR and HRMS spectra of 6c







Fig. S19. HRMS spectrum of 6c

#### NMR and HRMS spectra of 6d





National Center for Organic Mass Spectrometry in Shanghai Shanghai Institute of Organic Chemistry Chinese Academic of Sciences High Resolution MS DATA REPORT Instrument: Thermo Scientific Q Exactive HF Orbitrap-FTMS Card Serial Number: H-W190998 Operator : QHL Operation Mode: AP- MALDI Positive Ion Mode Date: 2019/11/1 Elemental composition search on mass 1385.5099 m/z= 1380.5099-1390.5099 m/z Theo. Mass Delta RDB Composition equiv. (ppm) 1385.5099 1385.5076 28.0 C45 H42 Se9 Si3 1.67 XW-3Se-w-s #683 RT: 4.13 AV: 1 NL: 3.51E3 T: FTMS + p NSI Full ms [400.0000-1500.0000] 1377.5145 100-90-1379.5177 1375.5197 80-1380.5176 70 Relative Abundance 1374.5222 1383.5118 60 1373.5194 50-1385.5099 1372.5198 40-1370.5212 30 20-10 0 1370 1375 1380 1385 m/z

Fig. S21. HRMS spectrum of 6d

## NMR and HRMS spectra of rac-1a



210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 **Fig. S23.** <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) spectra of *rac*-1a Shanghai Mass Spectrometry Center Shanghai Institute of Organic Chemistry Chinese Academic of Sciences High Resolution MS DATA REPORT



Instrument: IonSpec 4.7 Tesla FTMSCard Serial Number: WI132017Operator: HuaQinDate: 2013/10/16Operation Mode: MALDI/DHB

#### **Elemental Composition Search Report:**

#### Target Mass:

Target m/z =  $947.9608 \pm 0.004$ Charge = +1

#### Possible Elements:

Element:	Exact Mass:	Min:	Max:
С	12.000000	0	100
н	1.007825	0	100
Si	27.976927	0	4
S	31.972071	0	10

Additional Search Restrictions:

DBE Limit Mode = Both Integer and Half-Integer Minimum DBE = 0

#### Search Results:

Number of Hits = 5

m/z	Delta m/z	DBE	Formula
947.96057	0.00023	28.0	C45H36Si3S9+1
947.96193	-0.00113	55.0	C <sub>62</sub> H <sub>16</sub> SiS <sub>5</sub> <sup>+1</sup>
947.95856	0.00224	60.0	C <sub>65</sub> H <sub>12</sub> SiS <sub>4</sub> <sup>+1</sup>
947.96394	-0.00314	23.0	C42H40Si3S10+1
947.95720	0.00360	33.0	$C_{48}H_{32}Si_3S_8^{+1}$

Fig. S24. HRMS data of rac-1a

## NMR and HRMS spectra of rac-1b



Fig. S25. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) spectrum of rac-1b









Fig. S27. HRMS spectra of rac-1b

## NMR and HRMS spectra of rac-1c



Fig. S28. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) spectrum of *rac*-1c



200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10

Fig. S29. <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) spectra of *rac*-1c



Fig. S30. HRMS spectra of rac-1c

## NMR and HRMS spectra of rac-1d



Fig. S31. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) spectrum of rac-1d









National Center for Organic Mass Spectrometry in Shanghai Shanghai Institute of Organic Chemistry Chinese Academic of Sciences High Resolution MS DATA REPORT

Fig. S33. HRMS spectra of rac-1d

## 3. Quantum Calculation of *rac*-1(a)–(d)



## UV-vis spectra prediction and transition nature of *rac*-1(a)–(d)

**Fig. S34.** The simulated UV-vis spectra of compounds *rac*-1(a)–(d) in dichloroethane solvent at the PCM-TD-B3LYP/6-31G\* level of theory.

**Table S1** Selected computational absorption energies (nm), oscillator strength (f), and transitionnature in dichloromethane solvent for compounds rac-1(a)-(d) at the TD-B3LYP/6-31G\* level oftheory.

		states	λ	f	Transition Contributions
					HOMO-1 $\rightarrow$ LUMO (9%)
		<b>C</b> . C	262 0 (2 42)	0.6268	HOMO-1 $\rightarrow$ LUMO+1 (39%)
		$S_0 \rightarrow S_3$	303.0 (3.42)	0.0208	HOMO $\rightarrow$ LUMO (39%)
waa 1a	Dand I				$HOMO \rightarrow LUMO+1(9\%)$
rac-1a	Danu-I				HOMO−1 → LUMO (39%)
		5 . 5	363.0 (3.42)	0.6270	HOMO-1 $\rightarrow$ LUMO+1 (9%)
		$S_0 \rightarrow S_4$			HOMO $\rightarrow$ LUMO (9%)
					$HOMO \rightarrow LUMO+1(39\%)$
		$S_0 \rightarrow S_3$	366.7 (3.38)	0.6282	HOMO−1 → LUMO (31%)
					HOMO−1 → LUMO+1 (17%)
					HOMO $\rightarrow$ LUMO (16%)
uga <b>1b</b> Da	Dand I				$HOMO \rightarrow LUMO+1(31\%)$
740-10	Danu-I				HOMO−1 → LUMO (16%)
		5.5	2667(2.28)	0.6279	HOMO−1 → LUMO+1 (31%)
		$S_0 \rightarrow S_4$	300.7 (3.38)	0.0278	HOMO $\rightarrow$ LUMO (31%)
					$HOMO \rightarrow LUMO+1(17\%)$

			201.2 (2.25)	0.5000	HOMO−1 → LUMO (9%) HOMO−1 → LUMO+1 (40%)
		$S_0 \rightarrow S_3$	381.2 (3.25)	0.5000	HOMO $\rightarrow$ LUMO (36%)
waa <b>1</b> 0	Dand I	T			$HOMO \rightarrow LUMO+1(11\%)$
740-10	ac-ic Band-i			HOMO−1 → LUMO (37%)	
		$S_0 \rightarrow S_4$	380.9 (3.26)	0.5101	HOMO−1 → LUMO+1 (11%)
					HOMO $\rightarrow$ LUMO (9%)
					$HOMO \rightarrow LUMO+1(39\%)$
				0.4095	HOMO−1 → LUMO (46%)
<i>rac-</i> 1d	D 11	$S_0 \rightarrow S_3$	383.3 (3.24)	0.4985	$HOMO \rightarrow LUMO+1(48\%)$
	Dand-1		202.0 (2.24)	0 5061	HOMO−1 → LUMO (49%)
		$S_0 \rightarrow S_4$	383.0 (3.24)	0.3061	$HOMO \rightarrow LUMO+1(45\%)$



**Fig. S35.** Visualizations of HOMO and LUMO distributions (B3LYP/6-31G\*) for *rac*-1(a)–(d).

Compound	$\lambda_{onset}$ (nm)	$\Phi_{ m F}$	Time (ns)	E <sub>ox</sub> (V)	E <sub>g</sub> <sup>opt</sup> (eV)	HOMO (eV) experimental	HOMO (eV) theory	LUMO (eV) experimental	LUMO (eV)
							meery		- meery
rac-1	437	0.029	3.35	1.18	2.84	-5.54	-5.37	-2.70	-1.59
rac- <b>2</b>	441	0.020	3.10	1.16	2.81	-5.50	-5.32	-2.69	-1.56
rac-3	446	0.011	2.47	1.14	2.78	-5.47	-5.30	-2.69	-1.66
rac-4	449	0.008	2.11	1.10	2.76	-5.44	-5.25	-2.68	-1.63

 Table S2 Optical and electrochemical data of *rac*-1(a)–(d).

Measured in anhydrous CH<sub>2</sub>Cl<sub>2</sub>/Bu<sub>4</sub>F<sub>6</sub>NP (0.1 M), [C] =  $1 \times 10^{-3}$  mol L<sup>-1</sup>, vs Fc/Fc<sup>+</sup>.  $E_{\text{HOMO}} = -[E_{\text{ox}}^{\text{Onset}} - E_{(\text{Fc/Fc}^+)} + 4.8]$ eV.  $E_{\text{LUMO}} = E_{\text{g (opt)}} + E_{\text{HOMO}}$ .

## 4. Resolution of *rac*-1c

The resolution of the *rac*-1c was carried out by chiral HPLC with methanol/dichloromethane/diethylamine (80/20/0.1, v/v/v) as eluent. The two enantiomers were obtained on a semipreparative scaled chiral column (ID). From the 15 mg scale of *rac*-1c, 0.6 mg (*ee* > 97%) of (+)-1c and 1.6 mg (*ee* > 99%) of (-)-1c were efficiently obtained.

## HPLC trace of (+)-1c, (-)-1c, and rac-1c



**Fig. S36.** HPLC trace of *rac*-1c (top), (+)-1c (bottom left, ee > 97%) and (-)-1c (bottom right, ee > 99%) at room temperature. Conditions: Eluent: methanol/dichloromethane/diethylamine (80/20/0.1, v/v/v), Flow Rate: 1.0 mL/min, Column: CHIRALPAK-ID.

## Racemization of (-)-1c

Racemization of (–)-1c was carried out in  $CH_2Cl_2$  by heating at different temperatures. The process was monitored from time to time by chiral HPLC (CHIRALPAK-ID) with methanol/dichloromethane/diethylamine (80/20/0.1, v/v/v) as eluent. The half-life of racemization of (–)-1c is proposed as below:

$$(+) -\mathbf{1c} \rightarrow (-) -\mathbf{1c}$$
$$t = 0 \qquad C_0 \qquad 0$$
$$t = t \qquad C_t \qquad C_0 - C_t$$

The racemization of (-)-1c could be taken as first-order reaction, so

$$\ln C_0 / C_t = kt$$

Here,  $C_0$  is the concentration of (-)-1c before heating, and  $C_t$  is the concentration of (-)-1c after heating for time of *t*.

Because

$$ee = \frac{C_t - (C_0 - C_t)}{C_t + (C_0 + C_t)} = \frac{2C_t - C_0}{C_0} = 2\frac{C_t}{C_0} - 1$$
$$\Rightarrow \frac{C_t}{C_0} = \frac{ee + 1}{2}$$

So, we can obtain the formula as below:

$$\frac{C_0}{\ln C_t} = \ln \frac{2}{ee+1} = kt$$
$$\Rightarrow \ln \frac{ee+1}{2} = -kt$$





Fig. S37. Time-dependent enantiomeric excess value decay profiles at (a) 70 °C, (b) 75 °C,
(c) 80 °C, (d) 85 °C and (e) 90 °C, respectively.



Fig. S38. The half-life of (-)-1c at different temperatures.

 Table S3. The half-life of (-)-1c at different temperatures.

T/ºC	70	75	80	85	90
<i>t</i> <sub>1/2</sub> /h	7.61	5.19	4.00	3.00	2.34

# 5. X-ray Crystallographic Data

# X-ray crystallographic data of *rac*-1a

Table S4. Crystal data and structure refinement for rac-1a

Identification code	rac-1a
Empirical formula	$C_{45}H_{36}OS_9Si_3$
Formula weight	965.55
Temperature	296(2) K
Wavelength	0.71073 Å
Crystal system, space group	Monoclinic, $P2_1/c$
Unit cell dimensions	$a = 25.663(6) \text{ Å} \qquad \alpha = 90^{\circ}$
	$b = 18.397(5) \text{ Å}$ $\beta = 104.320(5)^{\circ}$
	$c = 23.432(6) \text{ Å} \qquad \gamma = 90^{\circ}$
Volume	10719(5) Å <sup>3</sup>
Z	8
Density (Calculated)	1.197 Mg/m <sup>3</sup>
Absorption coefficient	$0.469 \text{ mm}^{-1}$
F(000)	4000
Crystal size	$0.48 \times 0.23 \times 0.22 \text{ mm}^3$
Theta range for data collection	1.98 to 25.00°
Limiting indices	-26<=h<=30, -21<=k<=21, -24<=l<=27
Reflections collected / unique	$54813 / 18767 [R_{int} = 0.0435]$
Completeness to theta $= 25.00$	99.4 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.9038 and 0.8060
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	18767 / 84 / 1063
Goodness-of-fit on F <sup>2</sup>	1.129
Final R indices [I>2sigma(I)]	$R_1 = 0.0791, wR_2 = 0.1948$
R indices (all data)	$R_1 = 0.1233, wR_2 = 0.2077$
Largest diff. peak and hole	1.291 and -0.564 e. Å <sup>-3</sup>



Fig. S39. Molecular structure for *rac*-1a, Carbon, selenium, sulfur, and silicon atoms are depicted with thermal ellipsoids set at 30% probability level, and all hydrogen atoms are omitted for clarity.

# X-ray crystallographic data of *rac*-1b

Identific Empiric Formul	cation code cal formula a weight	<i>rac</i> -1b C <sub>45</sub> H <sub>36</sub> S <sub>6</sub> Se <sub>3</sub> Si <sub>3</sub>	
Empiric	al formula a weight	$C_{45}H_{36}S_6Se_3Si_3$	
Formul	a weight		
1 Official		1090.25	
Temper	ature	298.01 K	
Crystal	system, space group	Monoclinic, $P2_1/c$	
Unit cel	l dimensions	a = 25.663(3) Å	$\alpha = 90^{\circ}$
		b = 18.4115(18) Å	$\beta = 90^{\circ}$
		c = 23.619(2) Å	$\gamma = 90^{\circ}$
Volume	;	10803.4(18) Å <sup>3</sup>	
Ζ		8	
Density	(Calculated)	1.341 Mg/m <sup>3</sup>	
Absorp	tion coefficient	5.502 mm <sup>-1</sup>	
F(000)		4368	
Crystal	size	$0.17 \times 0.09 \times 0.04$ r	nm <sup>3</sup>
Radiatio	on	$CuK\alpha$ ( $\lambda = 1.54178$ )	)
Theta ra	ange for data collection	5.974 to 134.144°	
Index ra	anges	$-30 \le h \le 30, -21 \le 1$	$k \le 21, -28 \le l \le 23$
Reflect	ons collected	88921	
Indeper	dent reflections	19069 [ $R_{int} = 0.1462$	2, $R_{sigma} = 0.0953$ ]
Data / r	estraints / parameters	19069 / 353 / 1108	
Goodne	ss-of-fit on F <sup>2</sup>	1.097	
Final R	indices [I>2sigma(I)]	$R_1 = 0.0714, WR_2 =$	0.1780
R indic	es (all data)	$R_1 = 0.1219, WR_2 =$	0.2100
Largest	diff. peak and hole	0.58 and -0.72e. Å-3	3

Table S5. Crystal data and structure refinement for rac-1b



Fig. S40. Molecular structure for *rac*-1b, Carbon, selenium, sulfur, and silicon atoms are depicted with thermal ellipsoids set at 30% probability level, and all hydrogen atoms are omitted for clarity.

# X-ray crystallographic data of *rac*-1c

Identification code	rac-1c
Empirical formula	$C_{45}H_{36}S_3Se_6Si_3$
Formula weight	1230.96
Temperature	293(2) K
Crystal system, space group	cubic, Pa-3
Unit cell dimensions	$a = 21.30032(10) \text{ Å}  \alpha = 90^{\circ}$
	$b = 21.30032(10) \text{ Å}  \beta = 90^{\circ}$
	$c = 21.30032(10) \text{ Å}  \gamma = 90^{\circ}$
Volume	9664.03(13) Å <sup>3</sup>
Z	7.99992
Density (Calculated)	1.692 Mg/m <sup>3</sup>
Absorption coefficient	7.524 mm <sup>-1</sup>
F(000)	4800.0
Crystal size	$0.1611 \times 0.1222 \times 0.0669 \ mm^3$
Radiation	$CuK\alpha$ ( $\lambda = 1.54184$ )
Theta range for data collection	7.188 to 133.822°
Index ranges	$-19 \le h \le 25, -25 \le k \le 24, -19 \le l \le 25$
Reflections collected	23768
Independent reflections	2879 [ $R_{int} = 0.0392$ , $R_{sigma} = 0.0202$ ]
Data / restraints / parameters	2879 / 0 / 175
Goodness-of-fit on F <sup>2</sup>	1.049
Final R indices [I>2sigma(I)]	$R_1 = 0.0370, wR_2 = 0.0985$
R indices (all data)	$R_1 = 0.0454, wR_2 = 0.1034$
Largest diff. peak and hole	0.63 and -0.26e. Å <sup>-3</sup>

 Table S6. Crystal data and structure refinement for rac-1c



**Fig. S41.** Molecular structure for *rac***-1c**, Carbon, selenium, sulfur, and silicon atoms are depicted with thermal ellipsoids set at 30% probability level, and all hydrogen atoms are omitted for clarity.

# X-ray crystallographic data of *rac*-1d

Identification code	rac-1d
Empirical formula	$C_{91}H_{74}Cl_2Se_{18}Si_6$
Formula weight	2828.22
Temperature	293(2) K
Crystal system, space group	Monoclinic, $P2_1/n$
Unit cell dimensions	$a = 15.2593(17) \text{ Å} \alpha = 90^{\circ}$
	$b = 19.9001(3) \text{ Å}$ $\beta = 95.0760(10)^{\circ}$
	$c = 16.53532(18) \text{ Å } \gamma = 90^{\circ}$
Volume	5001.44(10) Å <sup>3</sup>
Z	2
Density (Calculated)	1.878 Mg/m <sup>3</sup>
Absorption coefficient	9.099 mm <sup>-1</sup>
F(000)	2700.0
Crystal size	$0.1525 \times 0.0577 \times 0.0404 \ mm^3$
Radiation	$CuK\alpha$ ( $\lambda = 1.54184$ )
Theta range for data collection	6.966 to 134.16°
Index ranges	$\text{-18} \le h \le 18,  \text{-21} \le k \le 23,  \text{-12} \le l \le 19$
Reflections collected	21782
Independent reflections	$8906 [R_{int} = 0.0297, R_{sigma} = 0.0386]$
Data / restraints / parameters	8906 / 106 / 586
Goodness-of-fit on F <sup>2</sup>	1.041
Final R indices [I>2sigma(I)]	$R_1 = 0.0455, wR_2 = 0.1246$
R indices (all data)	$R_1 = 0.0596, wR_2 = 0.1354$
Largest diff. peak and hole	0.83 and -0.52e. $Å^{-3}$

 Table S7. Crystal data and structure refinement for rac-1d



Fig. S42. Molecular structure for *rac*-1d, Carbon, selenium, sulfur, and silicon atoms are depicted with thermal ellipsoids set at 30% probability level, and all hydrogen atoms are omitted for clarity.



Fig. S43. Multiple interactions in the crystal packings of *rac*-1a.



Fig. S44. Multiple interactions in the crystal packings of *rac*-1b.



Fig. S45. Multiple interactions in the crystal packings of *rac*-1c.



Fig. S46. Multiple interactions in the crystal packings of *rac*-1d.

Compound	Intermolecular interactions (Å)				
	S6…S17: 3.39, S10…S13: 3.49, S11…S13: 3.60, C13…S16: 3.40,				
1	C54…S15: 3.44, C83…S6: 3.38, S18…H46A: 2.78, C11…H78A:				
rac-1a	2.85, C52…H71B: 2.81, C66…H89C: 2.79, C79…H36A: 2.85,				
	C83…H46A: 2.63				
	S2…S3: 3.43, S8…S12: 3.45, C27…S10: 3.32, C67…S7: 3.49,				
rac-1b	Se6…S8: 3.47, Se4…Se6: 3.63, H13A…H15A: 2.29,				
	H87C…S9: 2.83, H87C…C58: 2.66				
	Se2…H4: 3.09, C4…H13B: 2.79, H13C…H13C: 2.33,				
rac-1c	H13A…H15A: 2.29				
	Se1…Se5: 3.55, Se2…Se4: 3.78, Se2…Se5: 3.65, Se5…Se8: 3.66,				
11	Se6…Se9: 3.77, Se9…Se4: 3.50, Se6…C32: 3.60, Se1…H39B: 2.88,				
rac-1d	Se3…H4: 2.90, C38…H42C: 2.50, C42…H38A: 2.84, C43…H39A:				
	2.86, H37B…H16: 2.29, H43C…H39A: 2.35, H42C…H38B: 2.36				

Table S8. Intermolecular interactions (Å) of *rac*-1(a)–(d)

\_

	Ring	Dihedral	Ring	Dihedral	Ring	Dihedral
	plane	angles/°	plane	angles/°	plane	angles/°
rac-1a	AB	6.36	A'B'	2.49	A″B″	5.67
	BC	8.60	B'C'	7.77	B″C″	9.17
	CD	9.73	C'D'	10.63	C″D″	9.25
	DO	14.41	D'O'	16.15	D″O″	16.13
	AD″	46.69	A'D	43.43	A″D′	50.80
rac-1b	AB	4.48	A'B'	5.68	A″B″	6.16
	BC	6.75	B'C'	8.17	B″C″	8.65
	CD	9.84	C'D'	9.34	C″D″	8.70
	DO	17.13	D'O'	18.35	D″O″	15.62
	AD″	44.39	A'D	49.64	A″D′	45.64
rac-1c	AB	5.69	A'B'	5.69	A″B″	5.69
	BC	7.89	B'C'	7.89	B″C″	7.89
	CD	9.23	C'D'	9.23	C″D″	9.23
	DO	15.98	D'O'	15.98	D″O″	15.98
	AD″	44.63	A'D	44.63	A″D′	44.63
rac-1d	AB	4.29	A'B'	7.41	A″B″	4.34
	BC	6.00	B'C'	5.35	B″C″	10.92
	CD	10.45	C'D'	6.31	C″D″	8.74
	DO	21.46	D'O'	14.97	D″O″	17.71
	AD″	52.60	A'D	44.95	A″D′	44.95

	Carbon atoms	Torsion	Carbon atoms	Torsion	Carbon atoms	Torsion
Carbon atoms		angles/°	Carbon atoms	angles/°	Carbon atoms	angles/°
rac-1a	C(5-6-9-10)	1.65	C(34-33-30-28)	4.83	C(40-39-22-20)	0.58
	C(6-9-10-15)	14.66	C(33-30-28-29)	11.47	C(39-22-20-21)	12.78
	C(9-10-15-16)	26.06	C(30-28-29-14)	28.08	C(22-20-21-24)	27.61
	C(10-15-16-17)	20.45	C(28-29-14-13)	16.59	C(20-21-24-25)	20.85
rac-1b	C(2-4-6-8)	0.84	C(24-22-20-18)	2.40	C(36-34-32-30)	2.39
	C(4-6-8-9)	11.66	C(22-20-18-14)	10.52	C(34-32-30-26)	15.59
	C(6-8-9-13)	21.58	C(20-18-14-25)	22.65	C(32-30-26-10)	21.78
	C(8-9-13-15)	19.72	C(18-14-25-27)	15.68	C(30-26-10-11)	15.73
rac-1c	C(11-10-7-6)	2.77	C(11-10-7-6)	2.77	C(11-10-7-6)	2.77
	C(10-7-6-1)	12.58	C(10-7-6-1)	12.58	C(10-7-6-1)	12.58
	C(7-6-1-2)	27.15	C(7-6-1-2)	27.15	C(7-6-1-2)	27.15
	C(6-1-2-3)	20.49	C(6-1-2-3)	20.49	C(6-1-2-3)	20.49
rac-1d	C(11-10-7-6)	6.00	C(23-22-19-18)	8.51	C(35-34-31-30)	0.83
	C(10-7-6-1)	13.39	C(22-19-18-13)	21.11	C(34-31-30-25)	9.68
	C(7-6-1-26)	24.12	C(19-18-13-2)	26.01	C(31-30-25-14)	32.80
	C(6-1-26-27)	25.17	C(18-13-2-3)	14.22	C(30-25-14-15)	19.97

Table S10. Torsion angles of the internal helix of *rac*-1(a)–(d)

## 6. References

- J. Suffert, Simple Direct Titration of Organolithium Reagents Using N-Pivaloyl-otoluidine and/or N-pivaloyl-o-benzylaniline, J. Org. Chem., 1989, 54, 509–510.
- [2] G. M. Sheldrick, SADABS, University of gottingen: Germany, 1996.
- [3] G. M. Sheldrick, SHELXTL, version. 5.1; Bruker analytical X-ray systems, Inc.: Madison, WI, 1997.