# **Supporting Information**

## Controlling Elasticity of Polyacrylonitrile Fibers *via* Ionic Liquids Containing Cyano-based Anions

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#### **Experimental Section**

#### Materials.

Polyacrylonitrile (PAN) powder ( $M_n = 52,290$ ) was obtained from Carbon and Composites Group at Oak Ridge National Laboratory. 1-ethyl-3-methylimidazolium chloride ([C<sub>2</sub>mim]Cl, 98%, Sigma-Aldrich), 1-ethyl-3-methylimidazolium bromide ([C<sub>2</sub>mim]Br, 98.5%, Sigma-Aldrich), 1ethyl-3-methylimidazolium dicyanamide ([C<sub>2</sub>mim]DCA, 98.5%, Sigma-Aldrich), 1-ethyl-3methylimidazolium triacyanomethanide ([C<sub>2</sub>mim]TCM, 98%, TCI Chemicals) were used as received without further purification, and their structures can be found in **Fig. 1**.

#### **Procedures.**

#### Preparation and Fabrication of PAN/ILs Fibers through Melt-Spinning

Dry white PAN powder was mixed with the ionic liquid in a 20 mL glass vial. The PAN/ILs mixture (with 30/70 PAN/ILs weight ratio) was purged with argon for 1 minute and heated to 160 °C until a homogeneous melt was formed.

Melt-spinning experiments were performed with an Atlas Laboratory Melt Extruder and take-up system (TUS) following the previous work.<sup>1</sup> The rotor temperature was set to 150 °C, the header temperature was set to 160 °C, and the rotational speed was set to 90 rpm. The polymer thread (extrudate) was wrapped around the take-up spool with tweezers when the polymer composite melts were extruded from the orifice. Once the thread was wrapped around the spool, the take-up speed was maintained at 60 feet/minute.

#### Ionic liquid Recovery

The ionic liquids in the polymer composite were removed by contacting the as-spun fibers with deionized water for  $\sim$ 24 h and then dried under 100 °C for 24 h.

#### Characterization.

Size Exclusion Chromatography (SEC). Number-average molecular weights  $(M_n)$  and molecular weight distributions  $(M_w/M_n)$  of PAN powders were determined by SEC in DMF as an eluent at 35 °C.

**Thermogravimetric Analysis (TGA).** TGA with TA Instruments 2950 was used to investigate the thermal decomposition behaviors of ionic liquids, PAN powders, and the PAN/ILs polymer composite fibers in the N<sub>2</sub> atmosphere. The data were analyzed with TA Universal Analysis. The heating procedure involved four steps: (1) jump to 120 °C; (2) hold at 120 °C for 10 min; (3) ramp up at a rate of 20 °C/min to 800 °C; (4) hold for 2 min. The TGA plots were normalized to the total weight after holding at 120 °C.

**Differential scanning calorimetry (DSC)**. The glass transition temperature ( $T_g$ ), crystallization temperature ( $T_c$ ) and melting temperature ( $T_m$ ) of ionic liquids, PAN powders, and the PAN/ILs fibers were measured by differential scanning calorimetry (DSC) with TA Instrument QA-100. The same procedure was run three times, each involving the following steps: (1) Equilibrate at 25.00 °C, (2) Isothermal for 1.00 min, (3) Ramp 20.00 °C/min to -60.00 °C, (4) Isothermal for 1.00 min, (5) Ramp 20.00 °C/min to 150.00 °C, (6) Isothermal for 1.00 min, (7) Ramp 20.00 °C/min to -60.00 °C, (10) Isothermal for 1.00 min, (11) Ramp 20.00 °C/min to -60.00 °C, (12) Isothermal for 1.00 min, (13) Ramp 20.00 °C/min to 180.00 °C, (14) Isothermal for 1.00 min, (15) Jump to 25.00 °C. The DSC data were analyzed with a TA Universal Analysis instrument, and  $T_g$ ,  $T_c$ , and  $T_m$  were directly acquired.

**Scanning and Electron Microscopy (SEM).** SEM was carried out using a ZEISS AURIGA CROSSBEAM FIB scanning electron microscope to survey the surface morphologies of the PAN/ILs fibers before and after the removal of ionic liquids by deionized water.

**Mechanical Analysis.** The as-spun and washed PAN/ILs fibers were tested in the tensile mode by using an Instron 5943 universal testing machine. The fibers were fixed on paper mounts with a 25.4 mm gauge length and 10 mm/min speed at room temperature. 10 tests were performed for each PAN/ILs fiber to determine the average elastic modulus, tensile strength, and elongation at break values.

**X-ray diffraction (XRD).** XRD analysis using Cu  $K\alpha$  ( $\lambda = 0.1542$  nm) of the PAN/ILs fibers was conducted using a PANalytical Empyrean X-ray diffractometer, operated at 45 kV and 40 mA (scanned at 0.026° per step).

**Optical Microscope**. The photographs of PAN/IL fibers were obtained using a Nikon Eclipse TE2000 Inverted Microscope.

### **Supporting data:**





**Fig. S2** (a) TGA curve of pure PAN, (b) The plot of derivative weight vs. temperature of pure PAN, (c) DSC curve of pure PAN, (d) The plot of derivative heat flow vs. temperature of pure PAN.



**Fig. S3** (a) TGA curves of pure ILs, (b) The plots of derivative weight vs. temperature of pure ILs, (c) DSC curves of pure ILs, (d) The plots of derivative heat flow vs. temperature of pure ILs, black:  $[C_2mim]Cl$ , red:  $[C_2mim]Br$ , blue:  $[C_2mim]DCA$ , olive:  $[C_2mim]TCM$ .

Table 51. Thermal properties of pure 1 125 and 1 Alv					
Entry	$T_d^{\ a}$	Carbon yield (%) at 800 °C <sup>a</sup>	$T_{c}^{\ b}$	$T_{\rm m}^{\   { m b}}$	
[C <sub>2</sub> mim]Cl	320	0	-14.5	69.5	
[C <sub>2</sub> mim]Br	360	0	-	65.5	
[C <sub>2</sub> mim]DCA	315	11.5	-47.5	-5.5	
[C <sub>2</sub> mim]TCM	370	31.5	-	-	
Pure PAN	295	33.1	-	-	

Table S1. Thermal properties of pure PILs and PAN

<sup>*a*</sup> The decomposition temperature ( $T^{d}$ ) and the carbon yield are determined by TGA. <sup>*b*</sup> The crystallization temperature and the melting temperature are determined by DSC.



**Fig. S4** Strain-stress curves of PAN/ILs fibers: (a) PAN/[C<sub>2</sub>mim]Cl, (b) PAN/[C<sub>2</sub>mim]Br, (c) PAN/[C<sub>2</sub>mim]DCA, (d) PAN/[C<sub>2</sub>mim]TCM.



Fig. S5 Strain-stress curves of washed PAN/ILs fibers: (a) washed PAN/ $[C_2mim]Cl$  fibers, (b) washed PAN/ $[C_2mim]TCM$  fibers.

Table 52. Weenamear properties of TARVIES moets				
Entry	Young's modulus (MPa) <sup>a</sup>	Elongation at break (%) <sup>a</sup>		
PAN/[C <sub>2</sub> mim]Cl	$4425\pm502$	$81 \pm 27$		
PAN/[C <sub>2</sub> mim]Br	$3596\pm\ 503$	$71 \pm 15$		
PAN/[C <sub>2</sub> mim]DCA	$2280\pm765$	$292\pm 63$		
PAN/[C <sub>2</sub> mim]TCM	$1314\pm387$	$209 \pm 19$		
Washed PAN/[C <sub>2</sub> mim]Cl	$8943\pm4487$	$49 \pm 10$		
Washed PAN/[C <sub>2</sub> mim]TCM	$8188\pm2180$	$22 \pm 14$		

Table S2. Mechanical properties of PAN/ILs fibers

<sup>*a*</sup> Determined by Instron.



Fig. S6 (a) TGA curves of washed PAN/ILs fibers, (b) The plots of derivative weight vs. temperature of washed PAN/ILs fibers, (c) DSC curves of washed PAN/ILs fibers, (d) The plots of derivative heat flow vs. temperature of washed PAN/ILs fibers, black: PAN/[C<sub>2</sub>mim]Cl, olive: PAN/[C<sub>2</sub>mim]TCM.



Fig. S7 Photographs of PAN/ILs fibers: (a)  $PAN/[C_2mim]Cl$ , (b)  $PAN/[C_2mim]Br$ , (c)  $PAN/[C_2mim]DCA$ , (d)  $PAN/[C_2mim]TCM$ .

#### Reference

1. H. J. Martin, H. Luo, H. Chen, C.-L. Do-Thanh, L. T. Kearney, R. Mayes, A. K. Naskar and S. Dai, ACS Applied Materials & Interfaces, 2020, 12, 8663-8673.