

AN INSIGHT INTO THE PHOTODEGRADATION MECHANISM OF BISPHENOL-A BY OXYGEN DOPED MESOPOROUS CARBON NITRIDE UNDER VISIBLE IRRADIATION AND DFT CALCULATION

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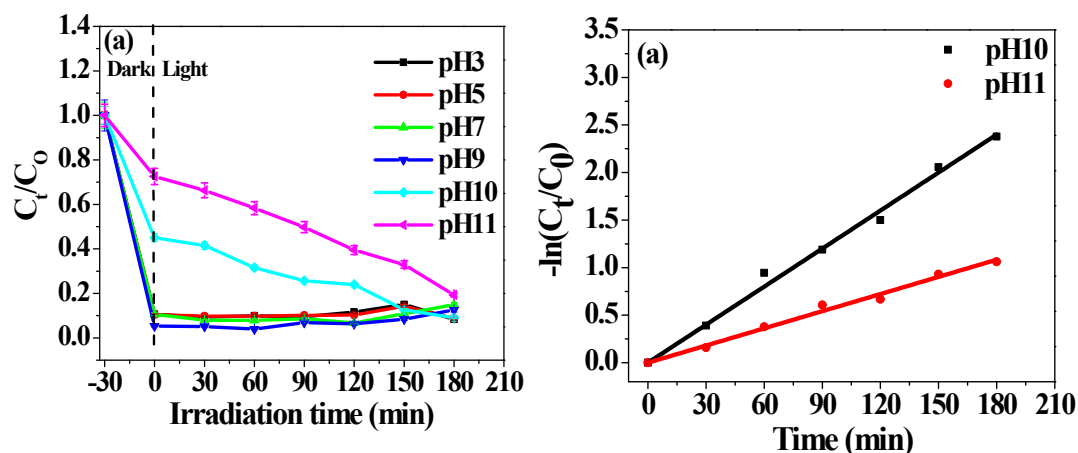


Fig. S1 The plot of (a) effect of initial pH of BPA and (b) pseudo-first-order kinetics of BPA photodegradation by O-MCN nanocomposite in different pH media on the removal of BPA ([BPA] = 20 mg L⁻¹; [O-MCN] = 50 mg L⁻¹).

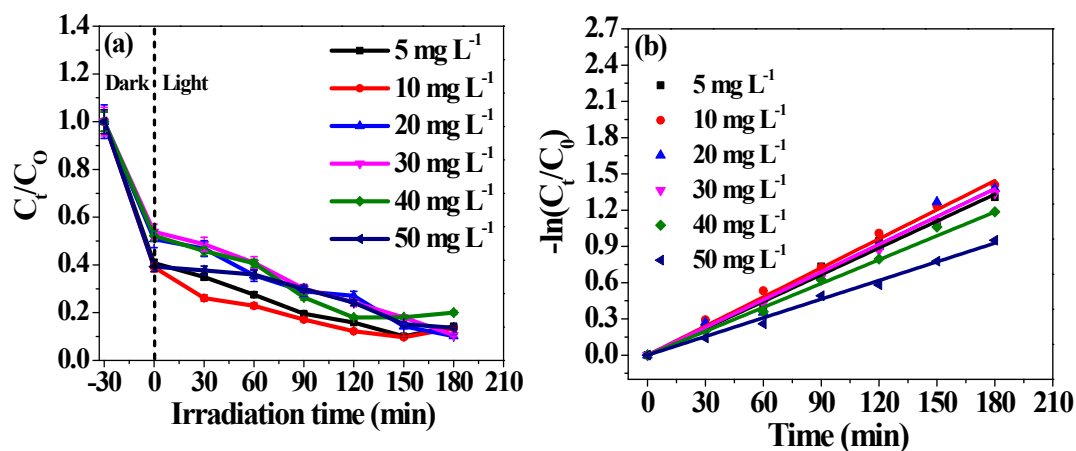


Fig. S2 The plot of (a) effect of initial concentration of BPA and (b) pseudo-first order kinetics of BPA photodegradation by O-MCN nanocomposite under different initial concentration of BPA solution. ([BPA] = 5, 10, 20, 30 40 and 50 mg L⁻¹; [O.2O-MCN] = 50 mg L⁻¹; pH = 10).

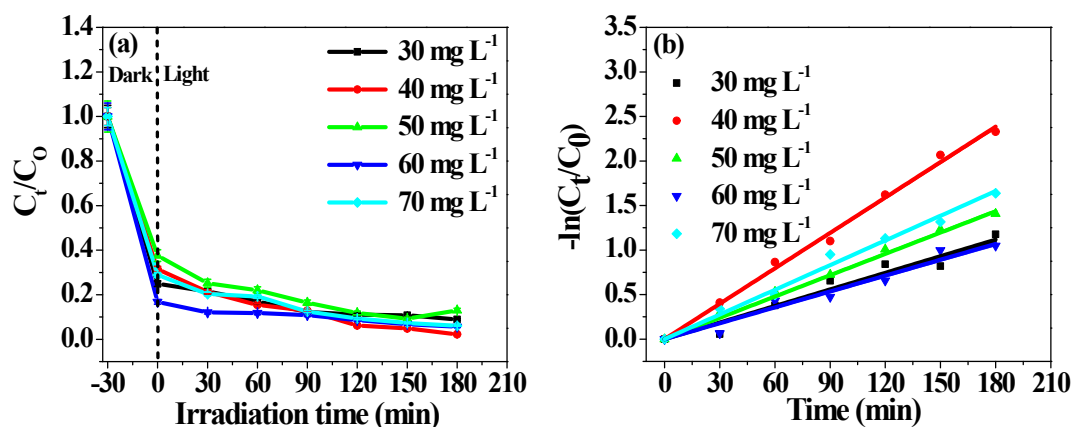


Fig. S3 The plot of (a) effect of catalyst dosage on BPA and (b) pseudo-first order kinetics of BPA photodegradation using different dosages of O-MCN nanocomposite. ([BPA] = 10 mg L⁻¹; [O-MCN] = 30, 40, 50, 60 and 70 mg L⁻¹; pH= 10).

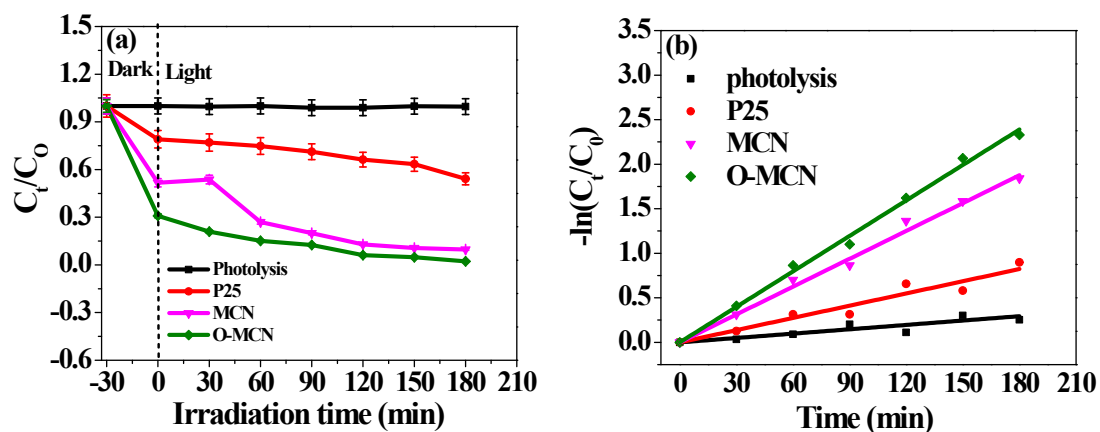


Fig. S4 The plot of (a) the effect of different synthesized photocatalyst and (b) pseudo-first order kinetics of BPA photodegradation using different synthesized against P25 on degradation of BPA ([BPA] = 10 mg L⁻¹; [catalyst] = 40 mg L⁻¹; pH= 10).