

Supporting Information

A selective hybrid fluorescent sensor for fructose detection based on a pheyboronic acid and BODIPY-based hydrophobicity probe

Gengo Kashiwazaki,^a Ryo Watanabe,^a Akihiro Nishikawa,^a Koyori Kawamura,^b Takashi Kitayama,^{a*} Takao Hibi^{b*}

^aMajor in Advanced Bioscience, Graduate School of Agriculture, Kindai University, 3327-204, Nakamachi, Nara, Nara 631-8505, Japan

^b4-1-1 Matsuoka-Kenjojima, Eiheiiji, Fukui 910-1195, Japan.

To whom correspondence should be addressed. E-mail: kitayama@nara.kindai.ac.jp, and hibi@fpu.ac.jp

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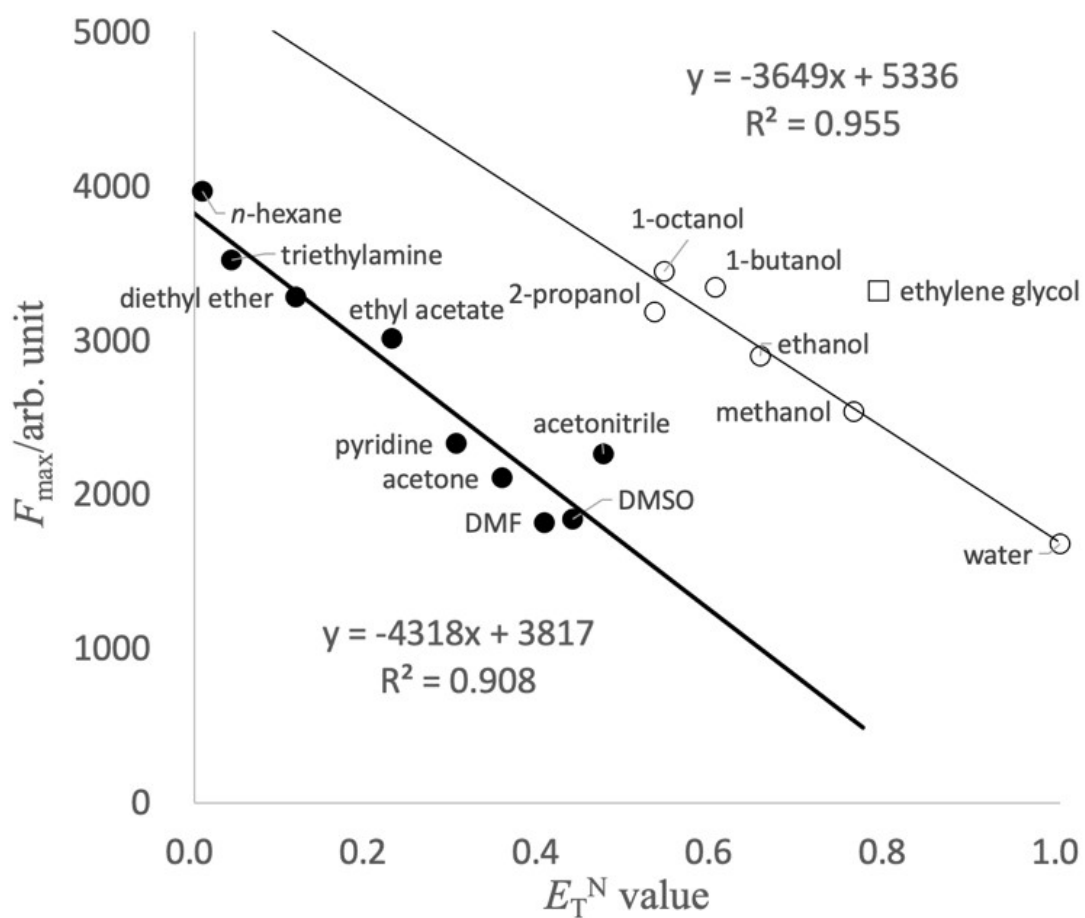


Figure S1. A correlation plot of the fluorescence intensities of HPSensor2 vs. the E_T^N value of the respective solvent.

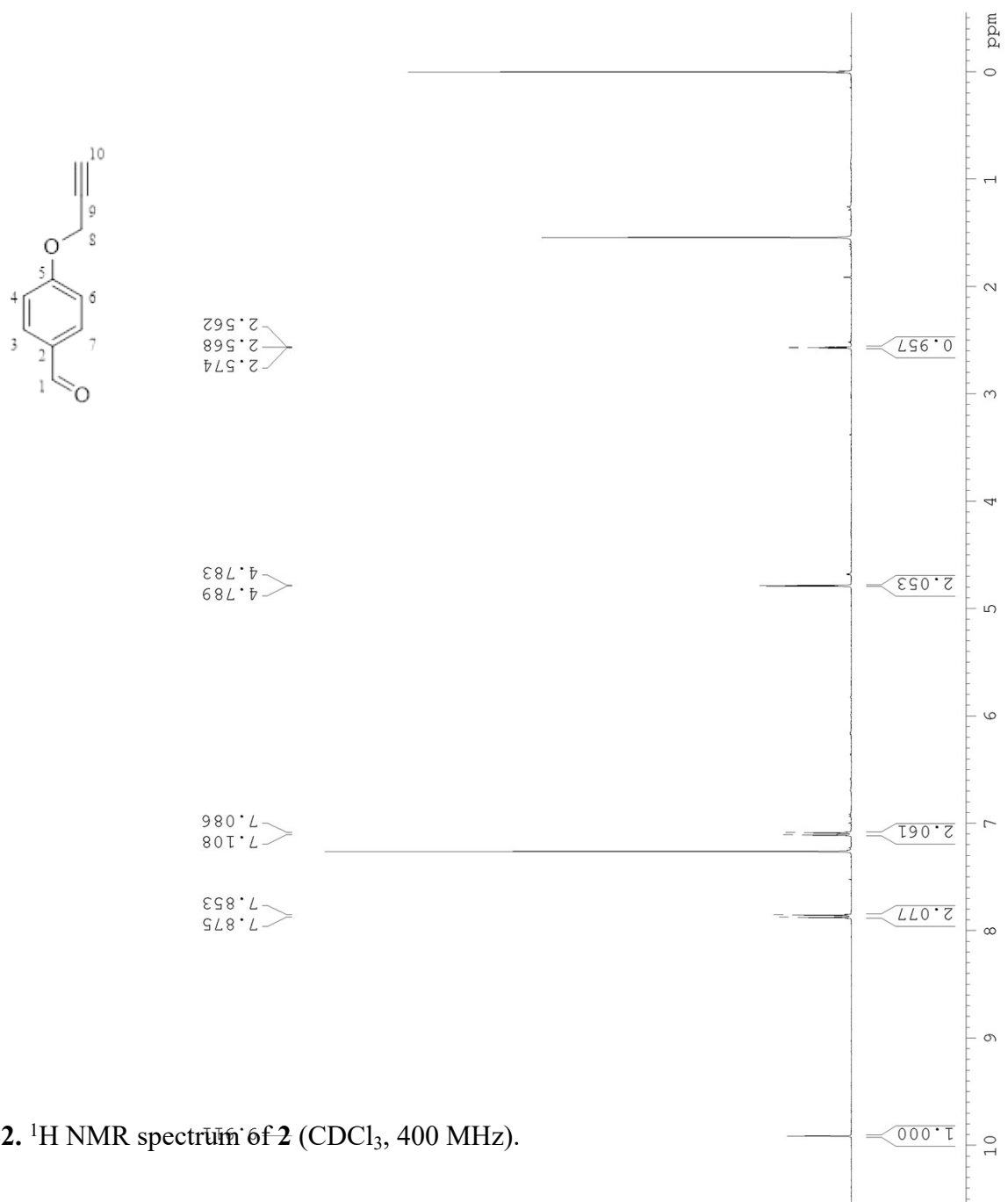


Figure S2. ^1H NMR spectrum of **2** (CDCl₃, 400 MHz).

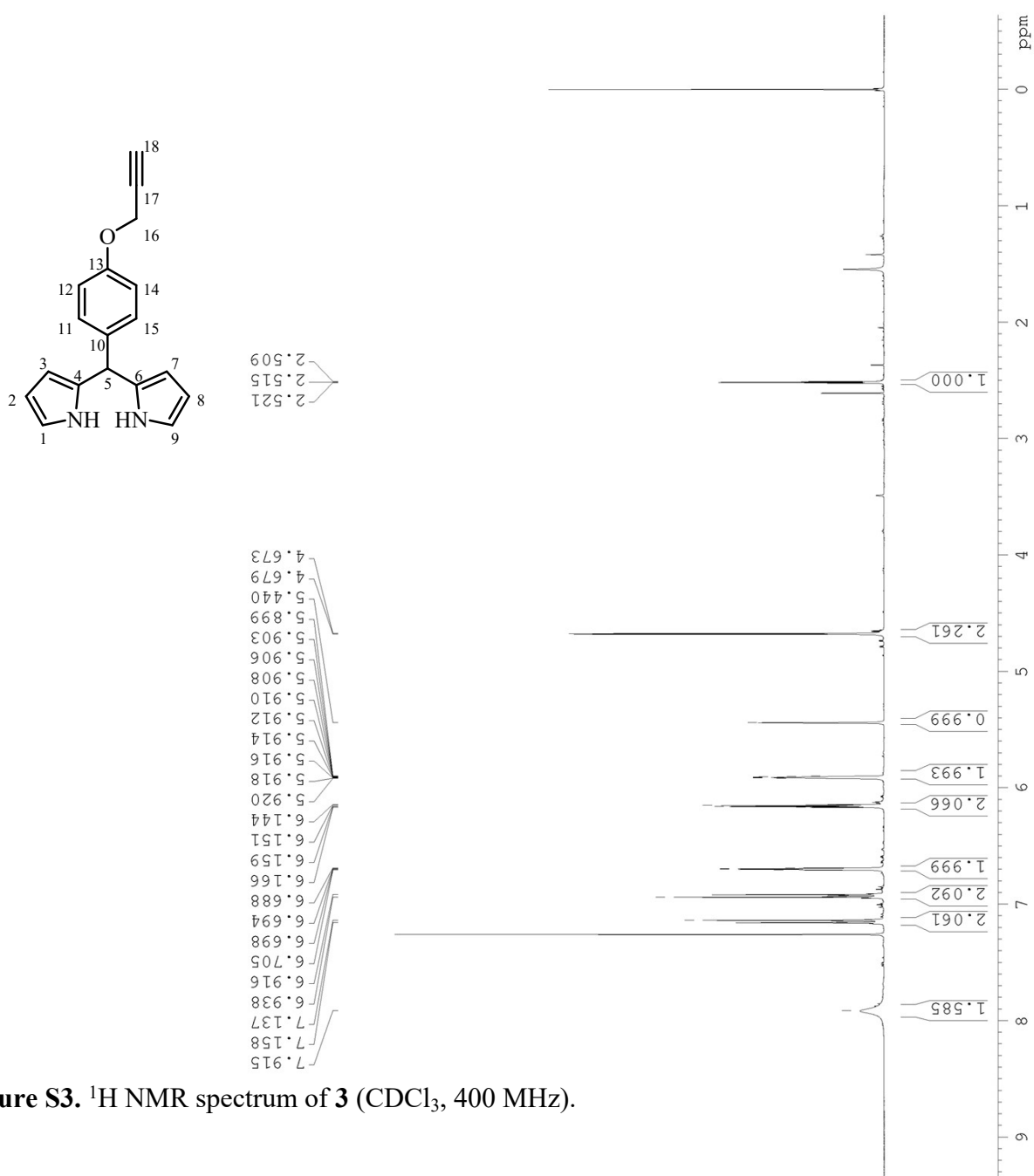


Figure S3. ^1H NMR spectrum of **3** (CDCl₃, 400 MHz).

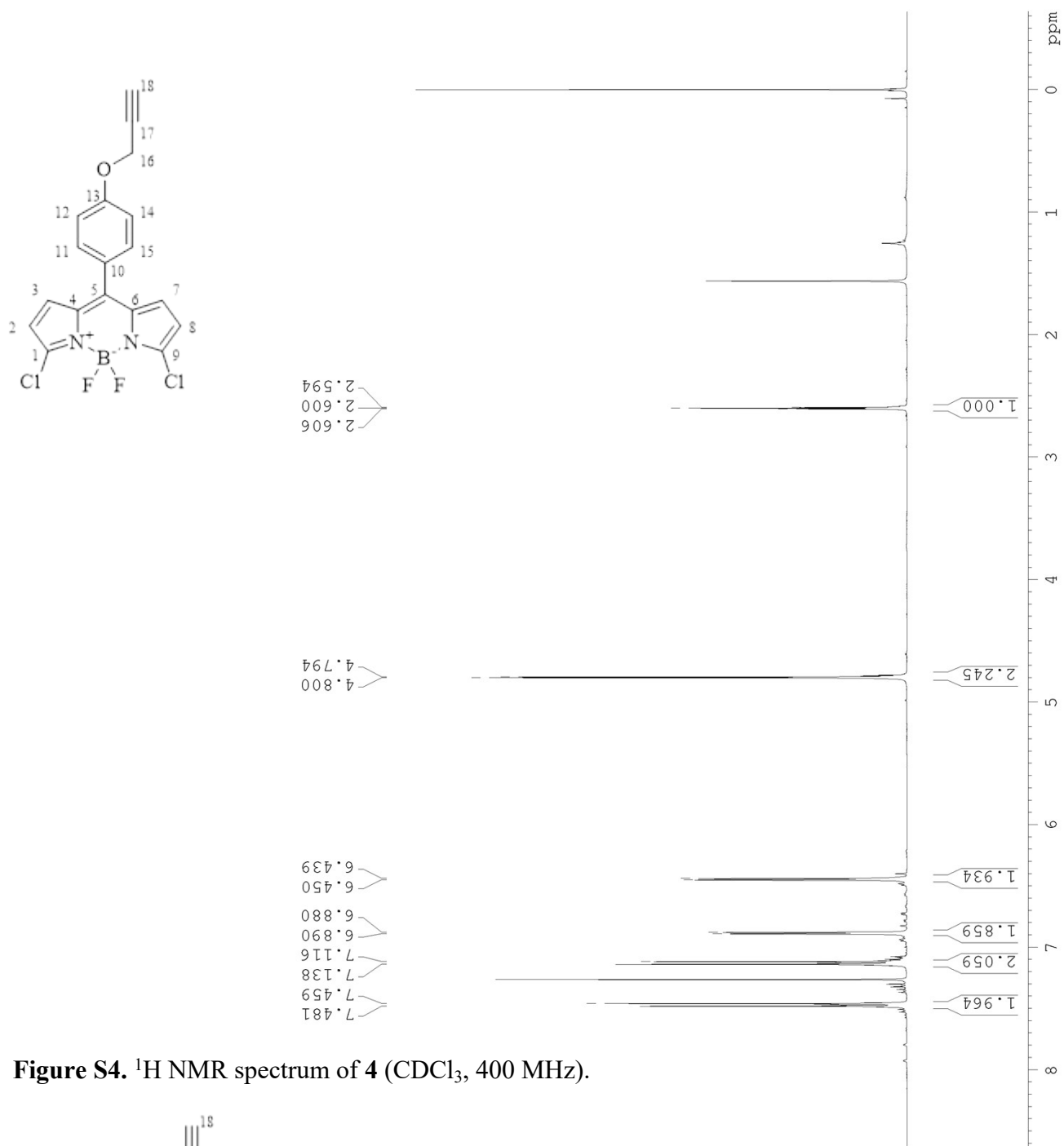
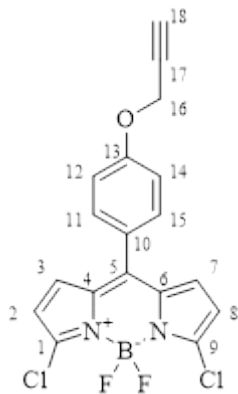


Figure S4. ¹H NMR spectrum of **4** (CDCl₃, 400 MHz).



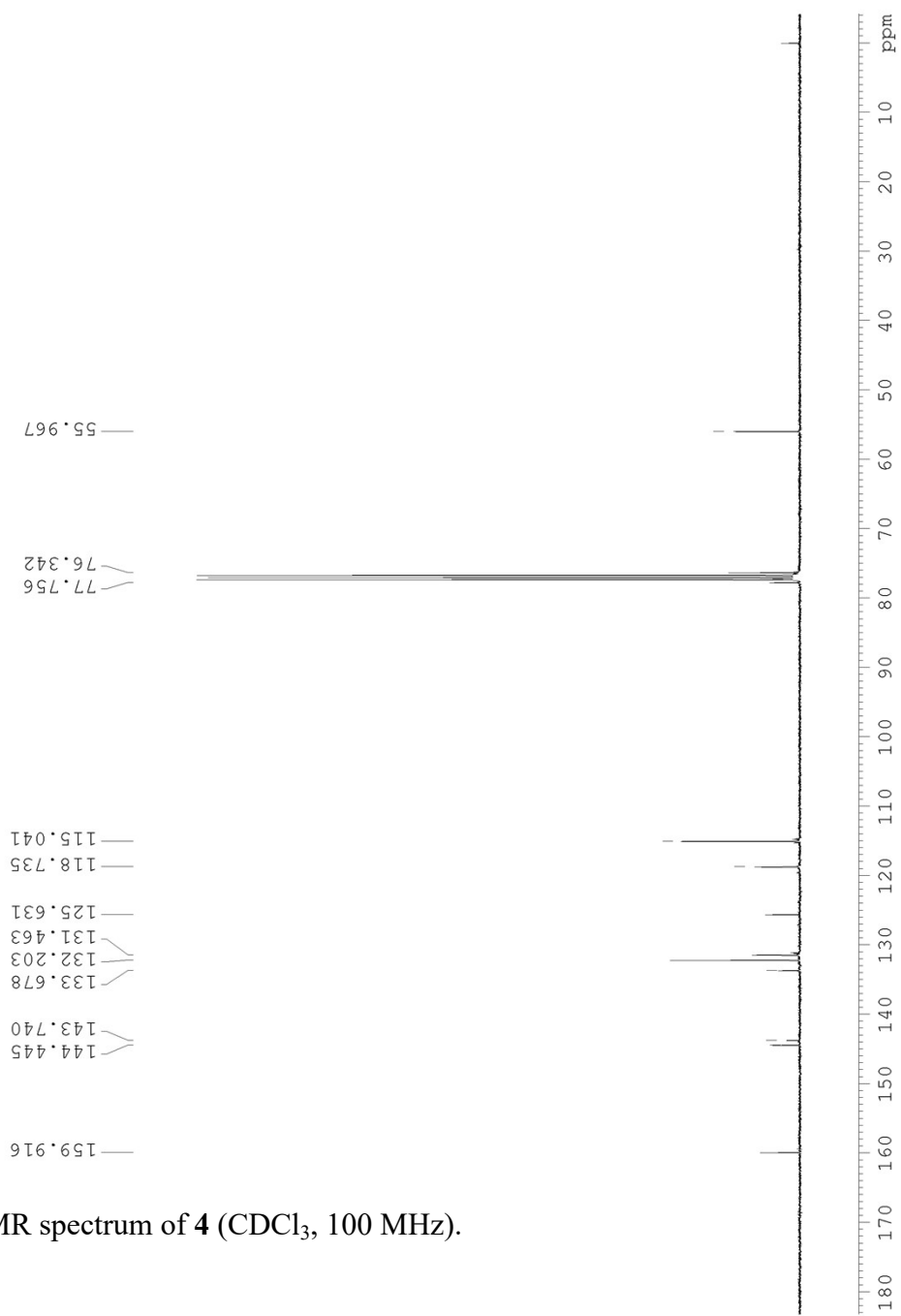


Figure S5. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of **4** (CDCl₃, 100 MHz).

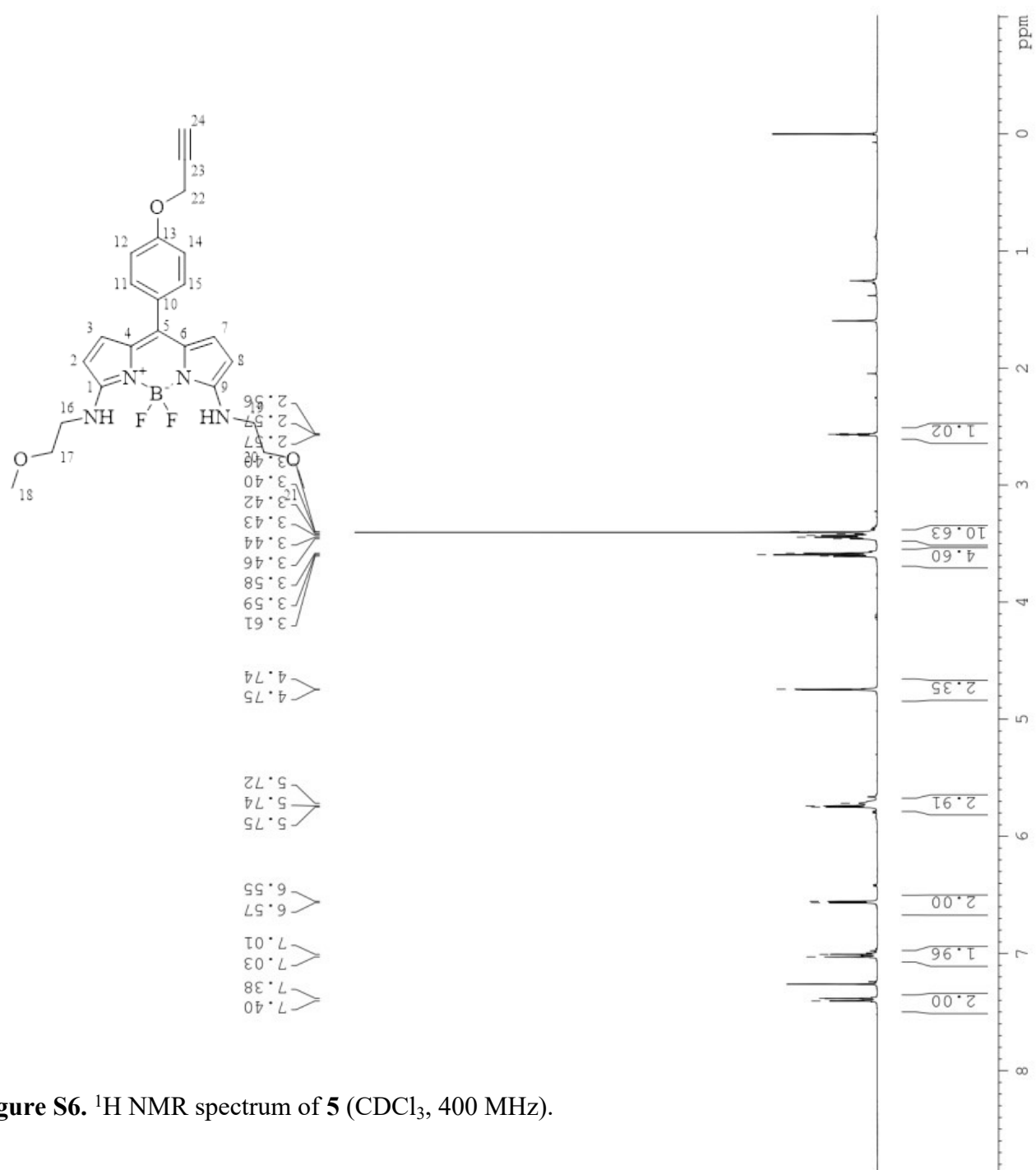
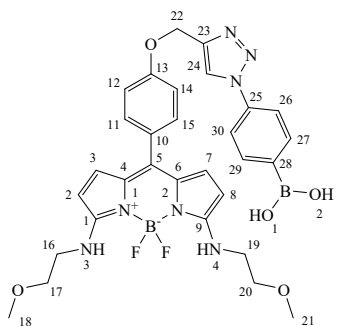
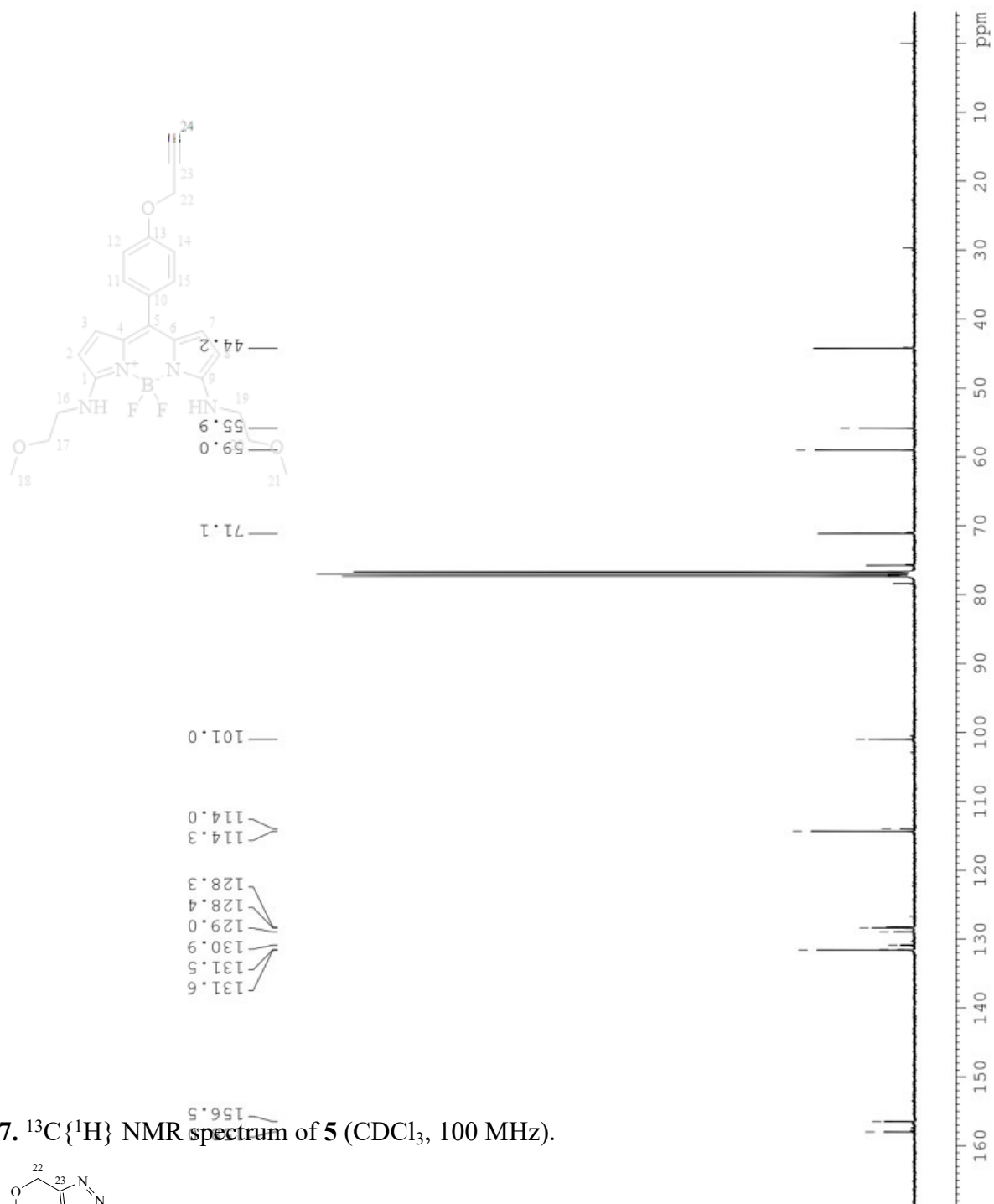


Figure S6. ^1H NMR spectrum of **5** (CDCl_3 , 400 MHz).



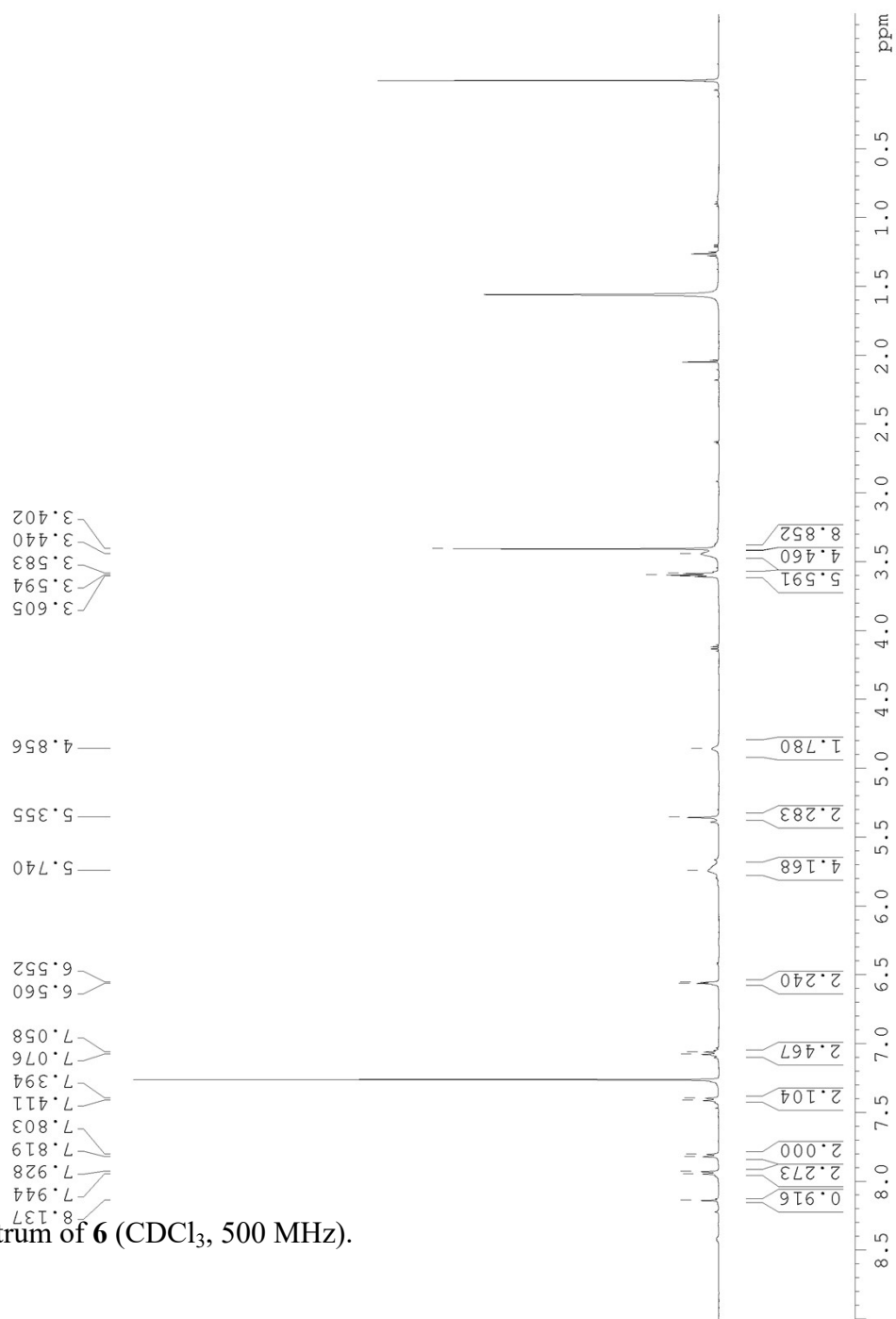


Figure S8. ^1H NMR spectrum of **9** (CDCl_3 , 500 MHz).

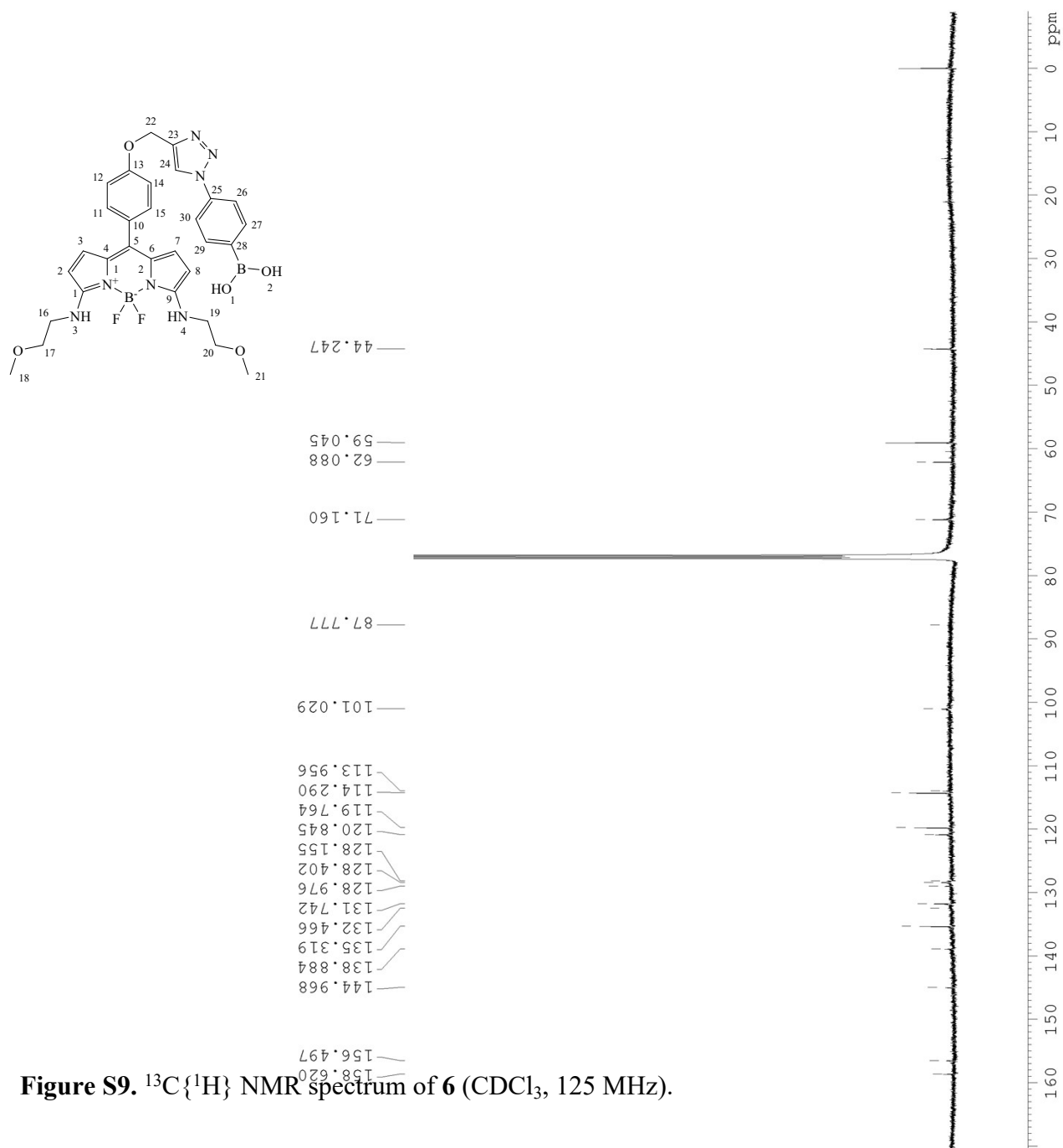


Figure S9. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of **6** (CDCl₃, 125 MHz).

Table S1: Effect of interference substances on fructose assay

sample	$\Delta F \pm SE (n = 3)$
2 mM D-fructose	361 \pm 29
+ 1 mM uric acid	389 \pm 10
+ 1 mM ascorbic acid*	424 \pm 15
+ 0.5%(v/v) ethanol	375 \pm 7
+ 0.5 mg/mL hen egg lysozyme	383 \pm 10
+ 0.5 mg/mL chicken ovalbumin*	892 \pm 13

*ANOVA followed by Dunnett's test showed that the changes after the addition of interference substances are statistically significant ($P < 0.05$).