

Supporting Information

Melamine derived N rich C entrapped Au nanoparticles for sensitive and selective monitoring of dopamine from blood samples

Faria Shakeel^a, Muhammad Waseem Fazal^b, Anam Zulfiqar^c, Farhan Zafar^b, Naeem Akhtar^{b*}, Arsalan Ahmed^b, Hafiz Badaruddin Ahmad^{a*}, Safeer Ahmed^d, Asad Syed^e, Ali H. Bahkali^e, Muhammad Abdullah^f and Zahid Shafiq^{a,g*}

^a*Institute of Chemical Sciences, Bahauddin Zakariya University, Multan 60800, Pakistan*

^b*Interdisciplinary Research Centre in Biomedical Materials (IRCBM), COMSATS University Islamabad, Lahore Campus, Lahore 54000, Pakistan*

^c*Department of Biochemistry, Bahauddin Zakariya University, 60800, Multan, Pakistan*

^d*Department of Chemistry, Quaid-i-Azam University, 45320, Islamabad, Pakistan*

^e*Department of Botany and Microbiology, College of Science, King Saud University, P.O. 2455, Riyadh, 11451, Saudi Arabia*

^f*Department of Chemistry, University of Massachusetts Amherst, 710 North Pleasant Street, Amherst, Massachusetts 01003, United States*

^g*Department of Pharmaceutical & Medicinal Chemistry, An der Immenburg 4, D-53121 Bonn, Germany*

Dr. Naeem Akhtar: naeemkhan@cuilahore.edu.pk

Prof. Dr. Zahid Shafiq: zahidshafiq@bzu.edu.pk

Dr. Badaruddin Ahmad: hafizbadar@googlemail.com

S-1 Instruments

The electrochemical measurements were performed with a Gamry electrochemical workstation (Inreface-1010). A conventional three-electrode cell was used at room temperature (24°C), graphitic lead pencil electrode (GPE) as a working electrode, platinum wire as a counter electrode, and Ag/AgCl electrode as reference electrode, respectively. Electrochemical impedance spectroscopy (EIS) was performed in 5mM $[\text{Fe}(\text{CN})_6]^{4-/3-}$ as a redox probe. The morphology of the modified electrode was examined using a transmission electron microscopy (TEM) system (F200×, The Thermo Scientific, Waltham, MA USA). Raman spectroscopy was performed using a EnSpectr R532 Raman spectrometer (Enhanced Spectrometry, Inc., Torrance, CA, USA).

S-2 Synthesis of benzimidazolium-1-acetate ionic liquid (IL)

Benzimidazolium-1-acetate ionic liquid (IL) was synthesized by conventional neutralization of benzimidazole with acetic acid as reported in literature ¹. Briefly, 15 mM of benzimidazole was dissolved into 10 mL of methanol in a round bottom flask under classical conditions and the solution was continuously stirred for 2h at 65°C. Next, 0.95 mL of acetic acid solution was added dropwise into the above prepared solution and kept under reflux overnight. Finally, the resulting light-yellow mixture was heated at 100°C in a rotary evaporator for 3h and upon cooling at room temperature, the IL product was made and used for analytical purposes.

S-3 NMR analyses of benzimidazolium-1-acetate ionic liquid (IL)

The purities of BAIL were verified by NMR.

^1H NMR: $\delta = 1.97$ (s, 3H), 5.11 (s, 1H), 7.23 –7.25 (m, 2H), 7.58 –7.6 (m, 2H), 8.23 (s, 1H)

²(Figure S1).

^{13}C NMR: 21.5 (methyl peak), 115.7, 122.1, 138.4, 142.3, 172.5 (carbonyl carbon of acetate ion)³ (Figure S2).

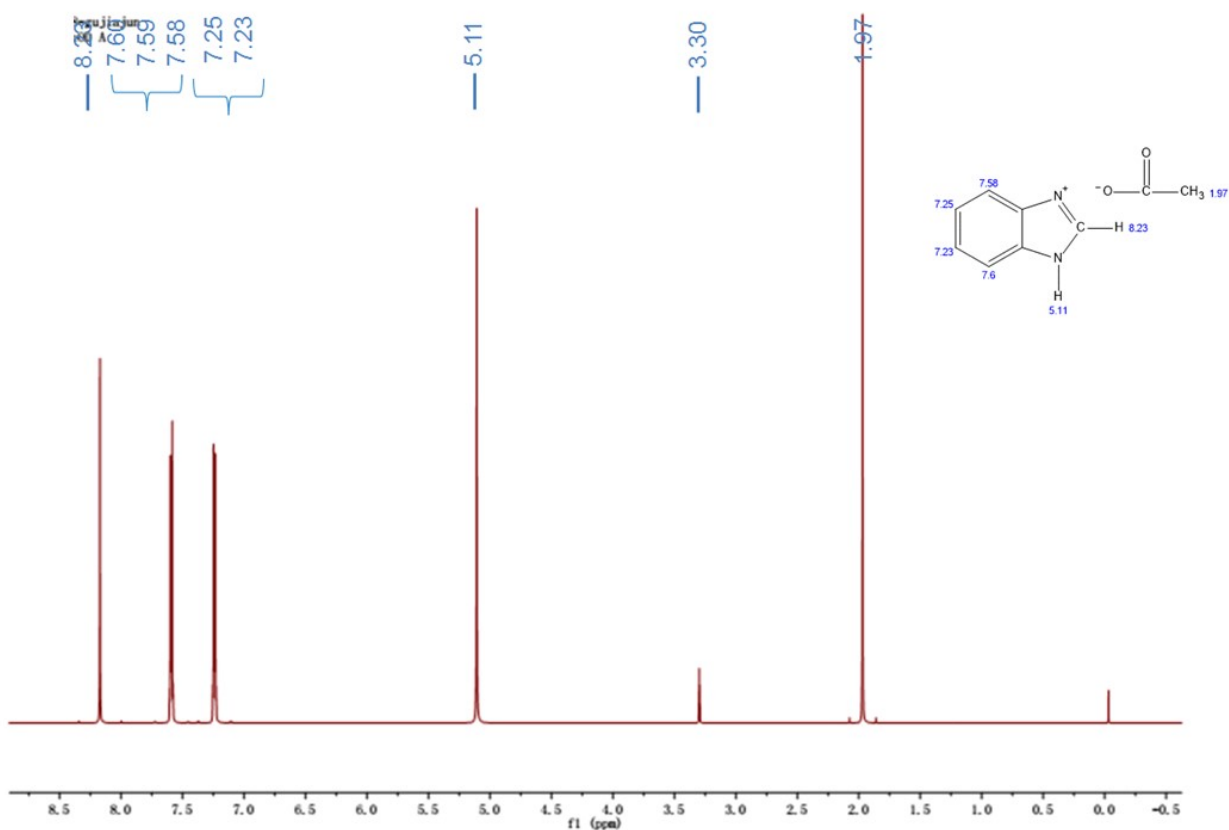


Figure S1. ^1H NMR of IL

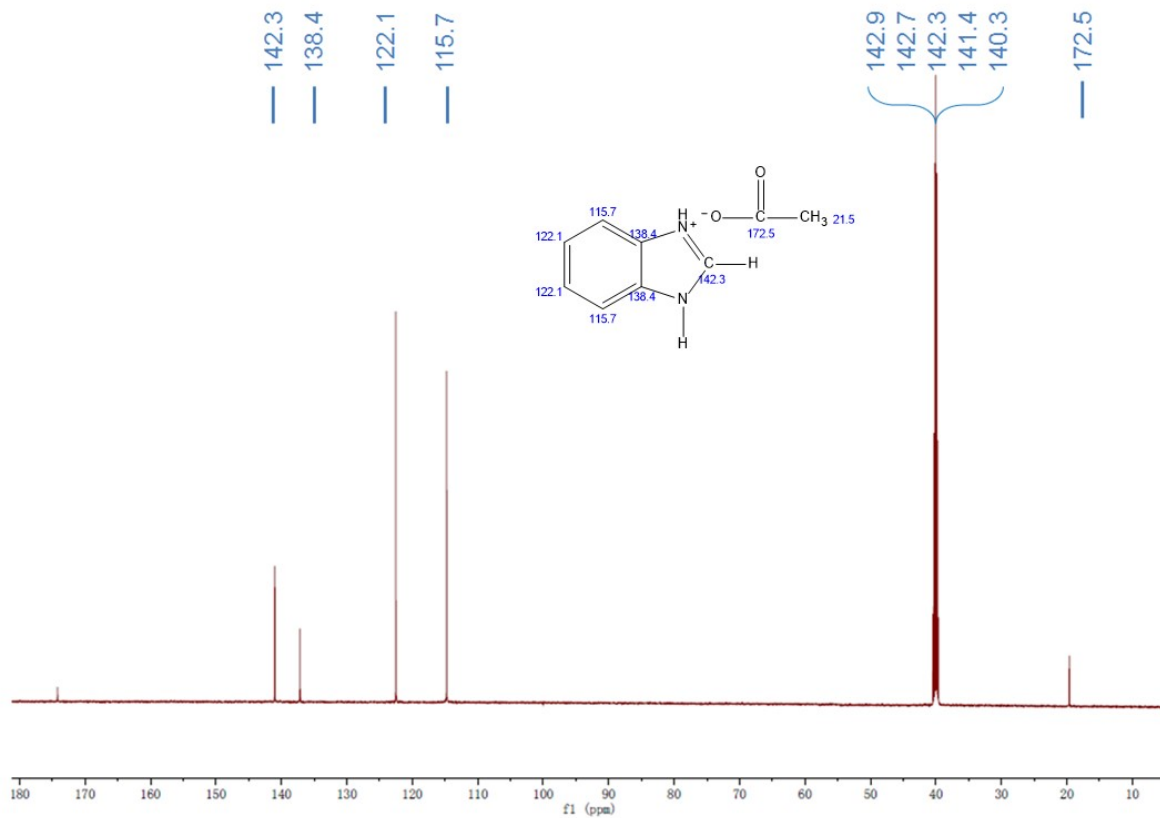


Figure S2. ¹³C NMR of IL

S-4 Fourier transforms infrared (FTIR) spectroscopy analyses of IL

FTIR spectroscopy has been employed to analyze the chemical structure of as-synthesized IL. The band at 2981 cm^{-1} is assigned to the alkyl group. The peak around 3112 cm^{-1} corresponds to alkenyl C—H stretching. The band at 1701 and 1619 cm^{-1} are due to the occurrence of C=O bonds and C=N, respectively. Also, the absence of stretching vibrations around 3000 cm^{-1} (O—H) verifies the presence of acetate ion.

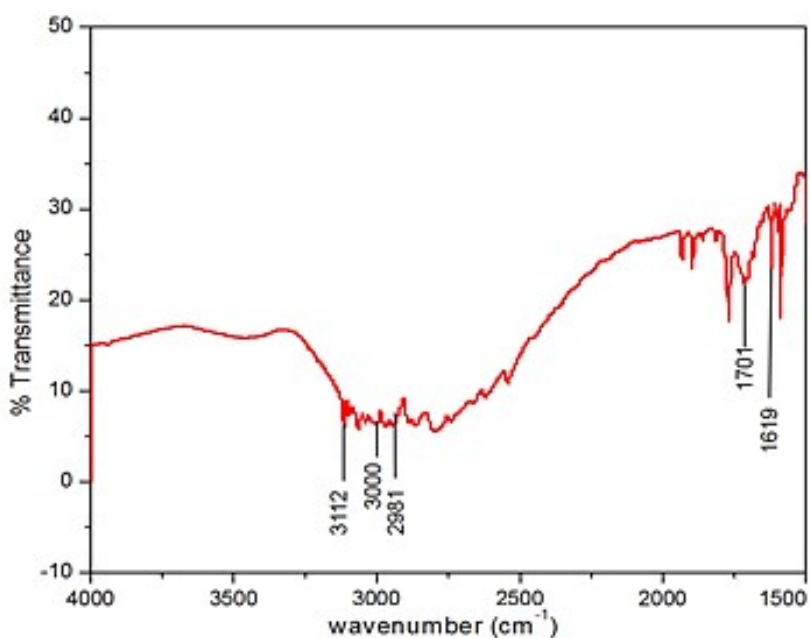


Figure S3 FTIR analysis of benzimidazolium-1-acetate ionic liquid (IL)

Scherer equation used to calculate the crystalline size

$$d = k \cdot \lambda / \beta \cdot \cos\theta$$

Where,

d = Crystallite size (nm)

β (the peak width of the diffraction peak or full width at the half maximum: FWHM)

$\cos\theta$ = (θ is the half of diffraction angle).

Scherer constant k = 0.9

λ = 0.15406 nm X-ray wavelength

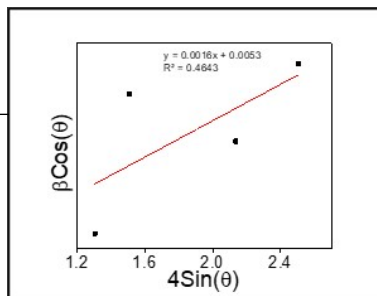
Williamson-Hull equation to calculate the strain

$$\beta \cos(\theta) = (k\lambda/d) + (\eta \sin(\theta))$$

η = Strain factor

Table S1 Table showing the calculation for average crystalline size and strain.

Parameter		Peak position	FWHM		Sherer	X-axis	Y-axis	Intercepts	Slope	Micro-strain
K	Lambda (A)	2 θ	β	Crystalline size (nm)	Average size (nm)	4 Sin θ	$\beta \cos\theta$	C=K A/D	m	*10 ³
Au-NPs										
0.9	0.154	38.1167	0.3945	21.29714	17.19782	1.306124	0.006508	0.0053	0.002	1.6
		44.3051	0.55006	15.58759		1.508296	0.008892			
		64.5425	0.54746	17.15554		2.135712	0.008079			
		77.5342	0.69046	14.75099		2.504625	0.009396			



A = constant

Absorption edge was observed for the transitions for $n = 2$

Table S2 Table showing the calculation for band gap.

Wavelength (nm)	Absorbance (a.u.)	Absorbance (a.u.)	Absorption coefficient (α) cm^{-1}	Absorption coefficient (α) cm^{-1}	Energy (eV)	$(\alpha h\nu)^2$	$(\alpha h\nu)^2$
698	0.029	0.025	0.066787	0.057575	1.776504	0.118647	0.102282
696	0.029	0.026	0.066787	0.059878	1.781609	0.118988	0.106679
694	0.029	0.026	0.066787	0.059878	1.786744	0.119331	0.106987
692	0.029	0.027	0.066787	0.062181	1.791908	0.119676	0.111423
690	0.029	0.028	0.066787	0.064484	1.797101	0.120023	0.115884
688	0.03	0.028	0.06909	0.064484	1.802326	0.124523	0.116221
686	0.03	0.029	0.06909	0.066787	1.80758	0.124886	0.120723
684	0.03	0.029	0.06909	0.066787	1.812865	0.125251	0.121076
682	0.03	0.03	0.06909	0.06909	1.818182	0.125618	0.125618
680	0.03	0.03	0.06909	0.06909	1.823529	0.125988	0.125988
678	0.031	0.031	0.071393	0.071393	1.828909	0.130571	0.130571
676	0.031	0.031	0.071393	0.071393	1.83432	0.130958	0.130958
674	0.031	0.032	0.071393	0.073696	1.839763	0.131346	0.135583
672	0.031	0.033	0.071393	0.075999	1.845238	0.131737	0.140236
670	0.032	0.034	0.073696	0.078302	1.850746	0.136393	0.144917
668	0.032	0.034	0.073696	0.078302	1.856287	0.136801	0.145351
666	0.032	0.035	0.073696	0.080605	1.861862	0.137212	0.150075

664	0.033	0.036	0.075999	0.082908	1.8674 7	0.1419 26	0.1548 28
662	0.033	0.037	0.075999	0.085211	1.8731 12	0.1423 55	0.1596 1
660	0.033	0.038	0.075999	0.087514	1.8787 88	0.1427 86	0.1644 2
658	0.034	0.039	0.078302	0.089817	1.8844 98	0.1475 6	0.1692 6
656	0.034	0.041	0.078302	0.094423	1.8902 44	0.1480 1	0.1784 83
654	0.035	0.042	0.080605	0.096726	1.8960 24	0.1528 29	0.1833 95
652	0.035	0.043	0.080605	0.099029	1.9018 4	0.1532 98	0.1883 37
650	0.036	0.045	0.082908	0.103635	1.9076 92	0.1581 63	0.1977 04
648	0.037	0.046	0.085211	0.105938	1.9135 8	0.1630 58	0.2027 21
646	0.037	0.048	0.085211	0.110544	1.9195 05	0.1635 63	0.2121 9
644	0.038	0.05	0.087514	0.11515	1.9254 66	0.1685 05	0.2217 17
642	0.038	0.051	0.087514	0.117453	1.9314 64	0.1690 3	0.2268 56
640	0.039	0.053	0.089817	0.122059	1.9375	0.1740 2	0.2364 89
638	0.04	0.055	0.09212	0.126665	1.9435 74	0.1790 42	0.2461 83
636	0.041	0.058	0.094423	0.133574	1.9496 86	0.1840 95	0.2604 27
634	0.042	0.06	0.096726	0.13818	1.9558 36	0.1891 8	0.2702 57
632	0.043	0.063	0.099029	0.145089	1.9620 25	0.1942 97	0.2846 68
630	0.044	0.066	0.101332	0.151998	1.9682 54	0.1994 47	0.2991 71
628	0.045	0.069	0.103635	0.158907	1.9745 22	0.2046 3	0.3137 65
626	0.046	0.072	0.105938	0.165816	1.9808 31	0.2098 45	0.3284 53
624	0.047	0.075	0.108241	0.172725	1.9871 79	0.2150 94	0.3432 36
622	0.048	0.078	0.110544	0.179634	1.9935 69	0.2203 77	0.3581 13
620	0.049	0.081	0.112847	0.186543	2	0.2256 94	0.3730 86
618	0.05	0.085	0.11515	0.195755	2.0064	0.2310	0.3927

					72	45	77
616	0.051	0.088	0.117453	0.202664	2.0129 87	0.2364 31	0.4079 6
614	0.052	0.092	0.119756	0.211876	2.0195 44	0.2418 53	0.4278 93
612	0.054	0.096	0.124362	0.221088	2.0261 44	0.2519 75	0.4479 56
610	0.055	0.1	0.126665	0.2303	2.0327 87	0.2574 83	0.4681 51
608	0.057	0.104	0.131271	0.239512	2.0394 74	0.2677 24	0.4884 78
606	0.058	0.109	0.133574	0.251027	2.0462 05	0.2733 2	0.5136 53
604	0.06	0.113	0.13818	0.260239	2.0529 8	0.2836 81	0.5342 65
602	0.061	0.118	0.140483	0.271754	2.0598 01	0.2893 67	0.5597 59
600	0.063	0.123	0.145089	0.283269	2.0666 67	0.2998 51	0.5854 23
598	0.065	0.129	0.149695	0.297087	2.0735 79	0.3104 04	0.6160 33
596	0.067	0.134	0.154301	0.308602	2.0805 37	0.3210 29	0.6420 58
594	0.069	0.14	0.158907	0.32242	2.0875 42	0.3317 25	0.6730 65
592	0.072	0.145	0.165816	0.333935	2.0945 95	0.3473 17	0.6994 58
590	0.074	0.151	0.170422	0.347753	2.1016 95	0.3581 75	0.7308 71
588	0.077	0.157	0.177331	0.361571	2.1088 44	0.3739 63	0.7624 97
586	0.08	0.163	0.18424	0.375389	2.1160 41	0.3898 59	0.7943 38
584	0.082	0.17	0.188846	0.39151	2.1232 88	0.4009 74	0.8312 88
582	0.085	0.176	0.195755	0.405328	2.1305 84	0.4170 73	0.8635 85
580	0.089	0.183	0.204967	0.421449	2.1379 31	0.4382 05	0.9010 29
578	0.092	0.189	0.211876	0.435267	2.1453 29	0.4545 44	0.9337 91
576	0.095	0.197	0.218785	0.453691	2.1527 78	0.4709 95	0.9766 96
574	0.099	0.204	0.227997	0.469812	2.1602 79	0.4925 37	1.0149 25
572	0.103	0.211	0.237209	0.485933	2.1678 32	0.5142 29	1.0534 21

570	0.107	0.219	0.246421	0.504357	2.1754 39	0.5360 74	1.0971 98
568	0.111	0.227	0.255633	0.522781	2.1830 99	0.5580 72	1.1412 82
566	0.116	0.234	0.267148	0.538902	2.1908 13	0.5852 71	1.1806 33
564	0.12	0.242	0.27636	0.557326	2.1985 82	0.6076	1.2253 27
562	0.125	0.25	0.287875	0.57575	2.2064 06	0.6351 69	1.2703 38
560	0.13	0.257	0.29939	0.591871	2.2142 86	0.6629 35	1.3105 72
558	0.135	0.265	0.310905	0.610295	2.2222 22	0.6909	1.3562 11
556	0.14	0.272	0.32242	0.626416	2.2302 16	0.7190 66	1.3970 43
554	0.146	0.28	0.336238	0.64484	2.2382 67	0.7525 9	1.4433 24
552	0.152	0.287	0.350056	0.660961	2.2463 77	0.7863 58	1.4847 67
550	0.158	0.294	0.363874	0.677082	2.2545 45	0.8203 7	1.5265 12
548	0.164	0.301	0.377692	0.693203	2.2627 74	0.8546 32	1.5685 62
546	0.17	0.307	0.39151	0.707021	2.2710 62	0.8891 44	1.6056 89
544	0.176	0.313	0.405328	0.720839	2.2794 12	0.9239 09	1.6430 89
542	0.182	0.318	0.419146	0.732354	2.2878 23	0.9589 32	1.6754 96
540	0.187	0.322	0.430661	0.741566	2.2962 96	0.9889 25	1.7028 55
538	0.192	0.326	0.442176	0.750778	2.3048 33	1.0191 42	1.7304 18
536	0.197	0.33	0.453691	0.75999	2.3134 33	1.0495 84	1.7581 86
534	0.201	0.332	0.462903	0.764596	2.3220 97	1.0749 06	1.7754 66
532	0.205	0.334	0.472115	0.769202	2.3308 27	1.1004 18	1.7928 77
530	0.208	0.334	0.479024	0.769202	2.3396 23	1.1207 35	1.7996 42
528	0.209	0.334	0.481327	0.769202	2.3484 85	1.1303 89	1.8064 59
526	0.211	0.333	0.485933	0.766899	2.3574 14	1.1455 45	1.8078 99
524	0.211	0.33	0.485933	0.75999	2.3664	1.1499	1.7984

					12	18	5
522	0.211	0.327	0.485933	0.753081	2.3754 79	1.1543 24	1.7889 28
520	0.209	0.323	0.481327	0.743869	2.3846 15	1.1477 8	1.7738 41
518	0.208	0.318	0.479024	0.732354	2.3938 22	1.1466 98	1.7531 25
516	0.205	0.313	0.472115	0.720839	2.4031 01	1.1345 4	1.7322 49
514	0.202	0.307	0.465206	0.707021	2.4124 51	1.1222 87	1.7056 54
512	0.199	0.3	0.458297	0.6909	2.4218 75	1.1099 38	1.6732 73
510	0.195	0.293	0.449085	0.674779	2.4313 73	1.0918 93	1.6406 39
508	0.191	0.286	0.439873	0.658658	2.4409 45	1.0737 06	1.6077 48
506	0.187	0.278	0.430661	0.640234	2.4505 93	1.0553 75	1.5689 53
504	0.182	0.27	0.419146	0.62181	2.4603 17	1.0312 32	1.5298 5
502	0.178	0.262	0.409934	0.603386	2.4701 2	1.0125 86	1.4904 36
500	0.173	0.255	0.398419	0.587265	2.48	0.9880 79	1.4564 17
498	0.169	0.247	0.389207	0.568841	2.4899 6	0.9691 1	1.4163 91
496	0.165	0.24	0.379995	0.55272	2.5	0.9499 88	1.3818
494	0.161	0.234	0.370783	0.538902	2.5101 21	0.9307 1	1.3527 09
492	0.158	0.228	0.363874	0.525084	2.5203 25	0.9170 81	1.3233 82
490	0.154	0.222	0.354662	0.511266	2.5306 12	0.8975 12	1.2938 16
488	0.151	0.217	0.347753	0.499751	2.5409 84	0.8836 35	1.2698 59
486	0.148	0.213	0.340844	0.490539	2.5514 4	0.8696 43	1.2515 81
484	0.146	0.209	0.336238	0.481327	2.5619 83	0.8614 36	1.2331 52
482	0.144	0.205	0.331632	0.472115	2.5726 14	0.8531 61	1.2145 7
480	0.142	0.202	0.327026	0.465206	2.5833 33	0.8448 17	1.2017 82
478	0.14	0.199	0.32242	0.458297	2.5941 42	0.8364 03	1.1888 88

476	0.139	0.197	0.320117	0.453691	2.6050 42	0.8339 18	1.1818 84
474	0.137	0.195	0.315511	0.449085	2.6160 34	0.8253 87	1.1748 22
472	0.136	0.193	0.313208	0.444479	2.6271 19	0.8228 35	1.1676 99
470	0.135	0.191	0.310905	0.439873	2.6382 98	0.8202 6	1.1605 16
468	0.134	0.19	0.308602	0.43757	2.6495 73	0.8176 63	1.1593 74
466	0.133	0.189	0.306299	0.435267	2.6609 44	0.8150 45	1.1582 21
464	0.133	0.188	0.306299	0.432964	2.6724 14	0.8185 58	1.1570 59
462	0.132	0.187	0.303996	0.430661	2.6839 83	0.8159 2	1.1558 87
460	0.132	0.186	0.303996	0.428358	2.6956 52	0.8194 67	1.1547 04
458	0.131	0.185	0.301693	0.426055	2.7074 24	0.8168 11	1.1535 11
456	0.131	0.185	0.301693	0.426055	2.7192 98	0.8203 93	1.1585 71
454	0.13	0.185	0.29939	0.426055	2.7312 78	0.8177 17	1.1636 74
452	0.13	0.184	0.29939	0.423752	2.7433 63	0.8213 35	1.1625 05
450	0.13	0.184	0.29939	0.423752	2.7555 56	0.8249 86	1.1676 72
448	0.13	0.184	0.29939	0.423752	2.7678 57	0.8286 69	1.1728 85
446	0.129	0.184	0.297087	0.423752	2.7802 69	0.8259 82	1.1781 45
444	0.129	0.184	0.297087	0.423752	2.7927 93	0.8297 02	1.1834 52
442	0.129	0.184	0.297087	0.423752	2.8054 3	0.8334 57	1.1888 07
440	0.129	0.184	0.297087	0.423752	2.8181 82	0.8372 45	1.1942 1
438	0.13	0.185	0.29939	0.426055	2.8310 5	0.8475 88	1.2061 83
436	0.13	0.185	0.29939	0.426055	2.8440 37	0.8514 76	1.2117 16
434	0.13	0.185	0.29939	0.426055	2.8571 43	0.8554	1.2173
432	0.13	0.185	0.29939	0.426055	2.8703 7	0.8593 6	1.2229 36
430	0.13	0.186	0.29939	0.428358	2.8837	0.8633	1.2352

					21	57	65
428	0.131	0.186	0.301693	0.428358	2.8971 96	0.8740 64	1.2410 37
426	0.131	0.187	0.301693	0.430661	2.9107 98	0.8781 67	1.2535 67
424	0.131	0.187	0.301693	0.430661	2.9245 28	0.8823 1	1.2594 8
422	0.132	0.188	0.303996	0.432964	2.9383 89	0.8932 58	1.2722 16
420	0.132	0.188	0.303996	0.432964	2.9523 81	0.8975 12	1.2782 75
418	0.132	0.189	0.303996	0.435267	2.9665 07	0.9018 06	1.2912 23
416	0.133	0.19	0.306299	0.43757	2.9807 69	0.9130 07	1.3042 95
414	0.133	0.19	0.306299	0.43757	2.9951 69	0.9174 17	1.3105 96
412	0.134	0.191	0.308602	0.439873	3.0097 09	0.9288 02	1.3238 9
410	0.134	0.192	0.308602	0.442176	3.0243 9	0.9333 33	1.3373 13
408	0.135	0.193	0.310905	0.444479	3.0392 16	0.9449 07	1.3508 68
406	0.135	0.193	0.310905	0.444479	3.0541 87	0.9495 62	1.3575 22
404	0.136	0.194	0.313208	0.446782	3.0693 07	0.9613 31	1.3713 11
402	0.137	0.195	0.315511	0.449085	3.0845 77	0.9732 18	1.3852 37
400	0.137	0.196	0.315511	0.451388	3.1	0.9780 84	1.3993 03

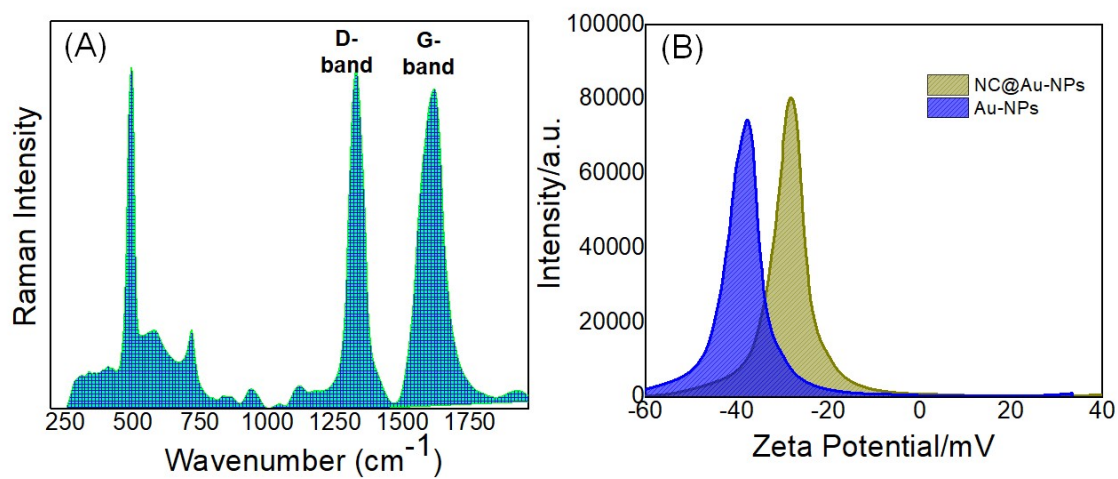


Figure S4 (A) Raman spectra of NC@Au-NPs. (B) Zeta potential of NC@Au-NPs and Au-NPs.

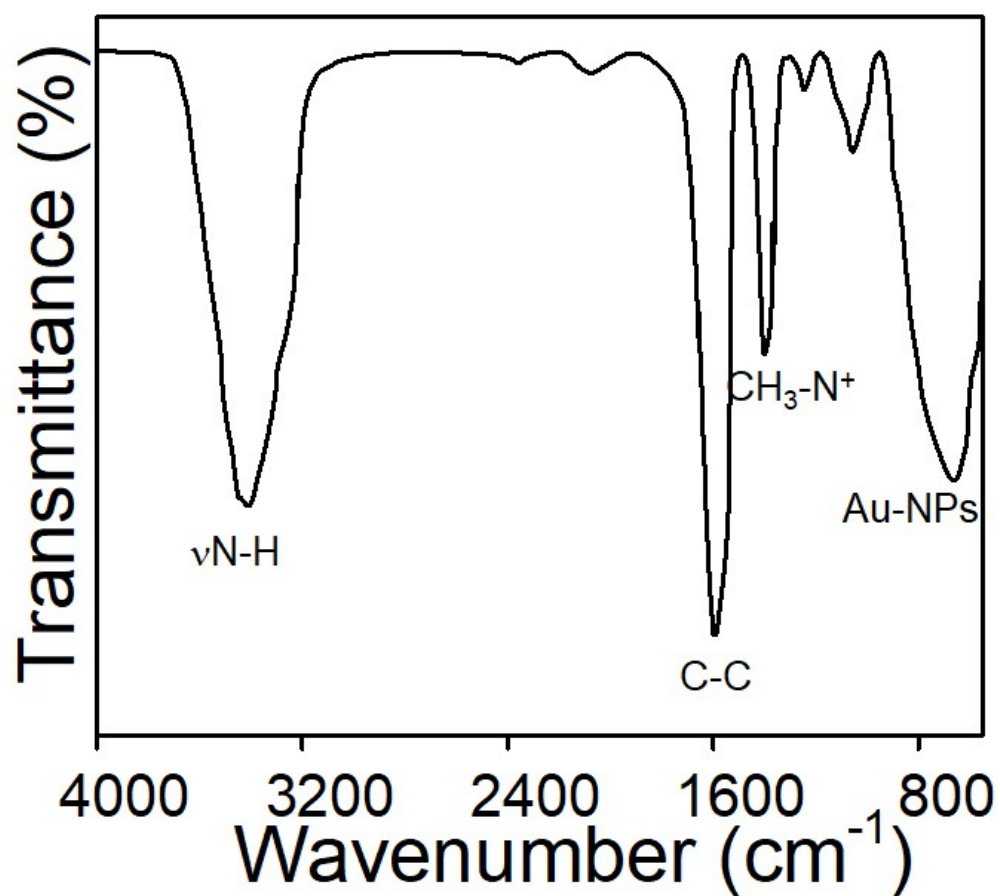


Figure S5 FTIR spectra of IL modified NC@Au-NPs.

Table S3 Full width at Half Maxima (FWHM) of (A) NC@Au-NPs and (B)Au-NPs.

(A) NC@Au-NPs

		Value	Standard Error
Book1_C	y0	0.10659	0.00641
	xc	528.60563	1.94352
	w	67.65688	4.75435
	A	18.95404	1.48874
	sigma	33.82844	2.37717
	FWHM	79.65989	5.59782
	Height	0.22353	0.01201

(B) Au-NPs

		Value	Standard Error
Book1_B	y0	0.07522	0.00414
	xc	520.5291	2.27891
	w	63.87246	5.45418
	A	9.96513	0.93388
	sigma	31.93623	2.72709
	FWHM	75.20407	6.42181
	Height	0.12448	0.00823

$$\text{FWHM}=0.5346wL+\text{sqrt}(0.2166*wL^2+wG^2)$$

whereas,

wL and wG (widths of the Lorentzian and Gaussian contributions) are given by the Origin software after the fitting.

Table S4 Table showing the peak current and peak potential difference of each electrode calculated from figure S2.

<i>Electrode</i>	<i>Anodic current</i>	<i>Ep</i>	<i>Ec</i>	<i>Peak Potential difference</i>
<i>Au-NPS</i>	0.096270791	0.445678228	0.026025712	0.419652517
<i><u>NC@Au-NPs</u></i>	0.2329252	0.430804696	0.022319052	0.408485644
<i>Au-NPs/PDA</i>	0.102208367	0.446608261	0.012255572	0.458863833

<i>NC@Au-NPs/PDA</i>	0.26041408	0.348232281	0.067018129	0.281214152
<i>Au-NPs/PDA-IL</i>	0.124572978	0.359583405	0.085441434	0.274141971
<u><i>NC@Au-NPs/PDA-IL</i></u>	0.32852887	0.310595934	0.102511091	0.208084843

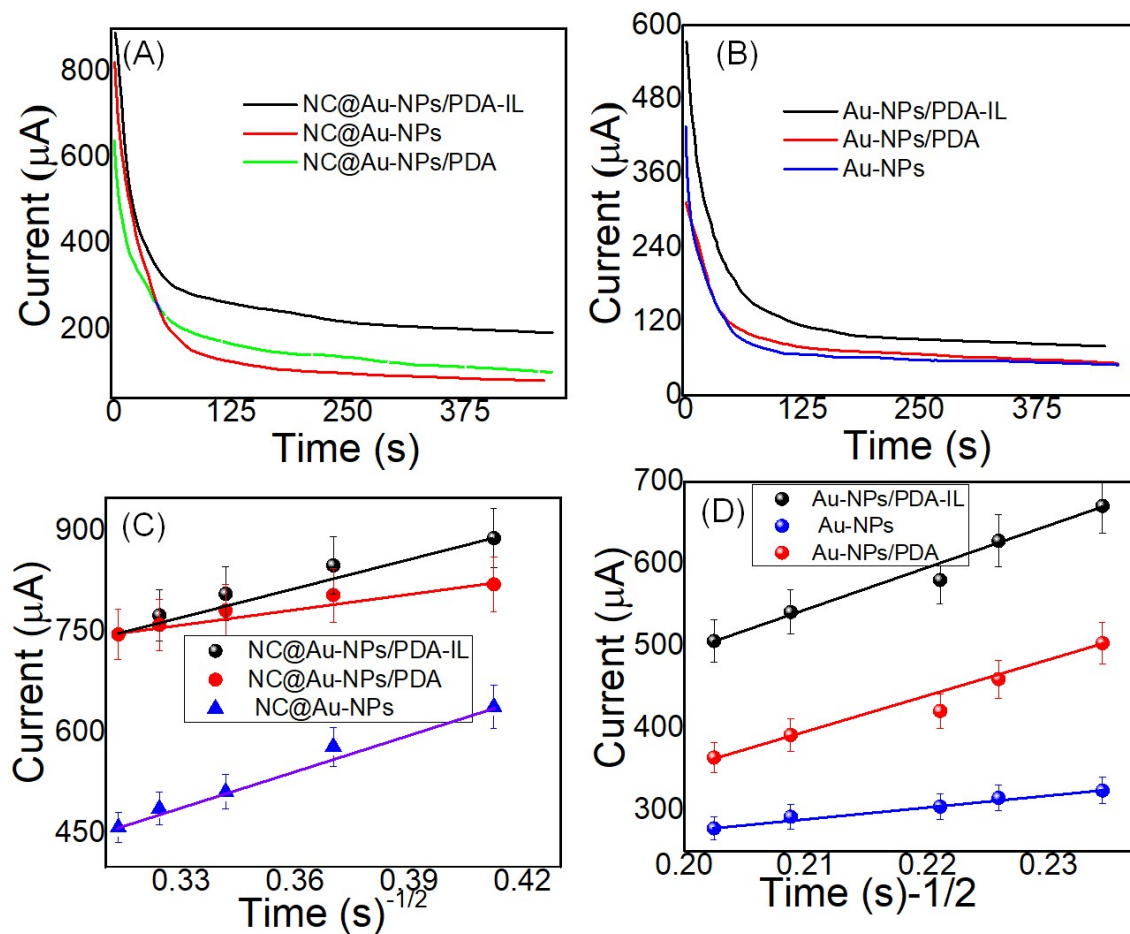


Figure S6. Chronoamperometric curve of (A) NC@Au-NPs, NC@Au-NPs/PDA, NC@Au-NPs/PDA-IL electrodes and (B) Au-NPs, Au-NPs/PDA, Au-NPs/PDA-IL in PBS against 1 mM of $[\text{Fe}(\text{CN})_6]^{3-}$. Electrochemical impedance spectroscopy (EIS) of NC@Au-NPs (C) and Au (D) based electrodes by using 2mM of $[\text{Fe}(\text{CN})_6]^{3-}$ as a redox probe. (C & D) Dependence of IC/IL on the $(\text{time})^{1/2}$ derived from their respective chronoamperogram data.

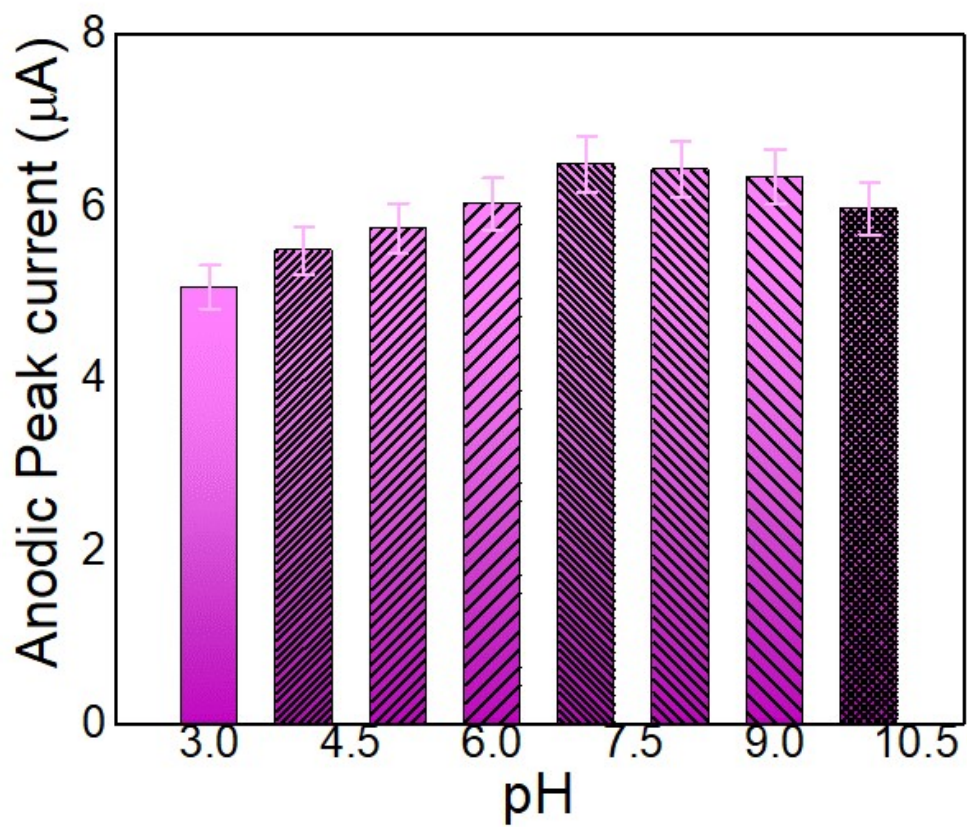


Figure S7. Optimization parameters pH at which the sensitive behavior of NC@Au-NPs/PDA/IL electrode was determined.

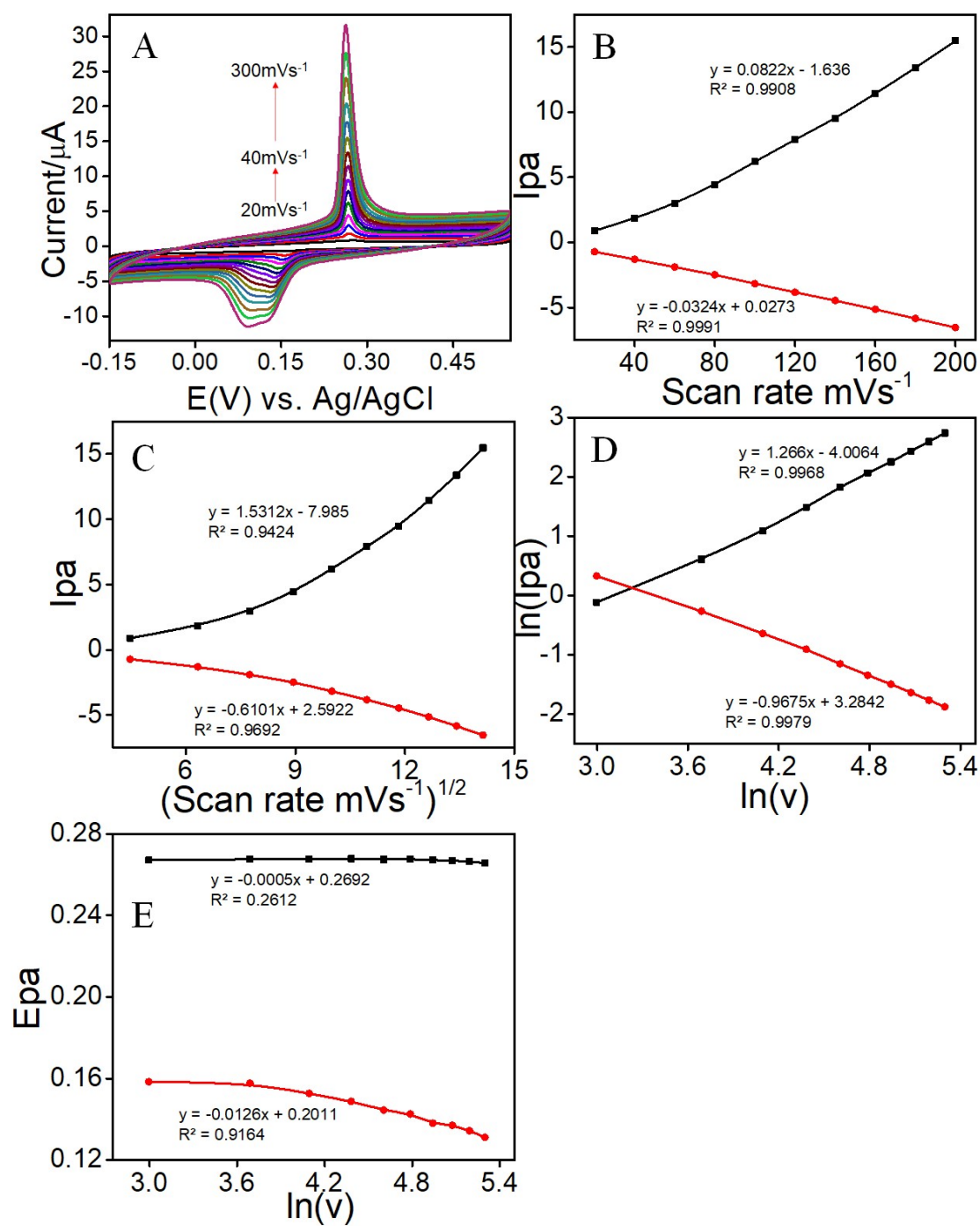


Figure S8. (A) cyclic voltammetry of Au-NPs/PDA-IL electrode at the scan rate ranging from 5 to 200 mV/s in the presence of 5 μM DA in phosphate buffer (pH 7.0) and its corresponding calibration plots (B-C) scan rate (mV/s) vs Current(μA) and square root scan rate (mV/s) vs Current(μA), respectively and (D-

E) log of scan rate (mV/s) vs oxidizing potential(mV) and log of scan rate (mV/s) vs log of Current(μ A), respectively.

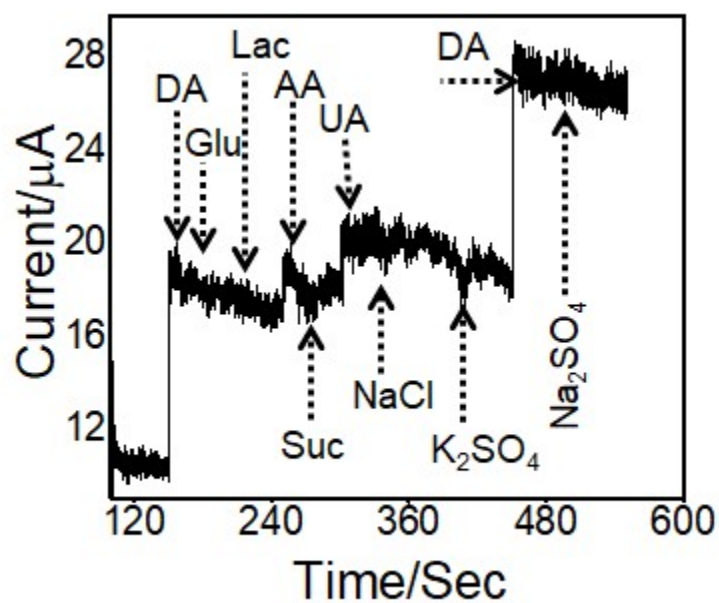


Figure S9: Selective amperometric response of the NC@Au-NPs/PDA/IL electrode exposed towards simultaneous monitoring of DA and co-existing electroactive species such as organic (DA, AA, UA, & Lac) and inorganic (KCl, NaCl, Na₂SO₄, & K₂SO₄) species in 0.1M PBS (pH;7) electrolyte.

Table S5 comparison of the electrochemical determination of NC@Au-NPs/PDA/IL modified electrode towards DA detection with the reported data.

Sr. #	Modified Electrode	Applied Potential (V)	Detection Limit (μM)	Linear Range (μM)	Ref.
1	RGO/Au nanoplate	-0.2–6	1.4	6.8–41	S ⁴
2	RGO-AuNPs-CSHMs	-0.2–8	0.3	1–200	S ⁵
3	AgNPs/rGO	-1–1	5.4	10–800	S ⁶
4	RGO/AuNPs	-1.5–1	0.137	0.14–700	S ⁷
5	PANI-AuNPs	-0.4–1.4	16	0.270–0.361	S ⁸
6	Au NPs/N-doped CN	-0.3–0.4	0.007	0.02–700	S ⁹
7	NC@Au-NPs/PDA/IL	-0.6-0.6	0.002	0.05-1	This work

Table S6. Determination of DA in DA hydrochloride injection using NC@Au-NPs/PDA/IL electrode.

DA sample	Added (μM)	Found [a] (μM)	Recovery (%)	RSD [b] (%)
1	0	1.78	-	-
2	2	3.75	87.7	3.7
3	2	4.7	92.8	3.65

1, 2 and 3 diluted DA hydrochloride injection sample, [a] Standard deviation, [b] Relative standard deviation (n = 3).

References:

1. A. S. Saleemi, M. Hafeez, A. Munawar, N. Akhtar, W. Abbas, M. E. Mazhar, Z. Shafiq, A. P. Davis and S.-L. Lee, *Journal of Materials Chemistry C*, 2020, **8**, 12984-12992.
2. N. Touj, A. S. Al-Ayed, M. Sauthier, L. Mansour, A. H. Harrath, J. Al-Tamimi, I. Özdemir, S. Yaşar and N. Hamdi, *RSC Advances*, 2018, **8**, 40000-40015.
3. Y. Abbas, S. Ali, M. Basharat, W. Zou, F. Yang, W. Liu, S. Zhang, Z. Wu, N. Akhtar and D. Wu, *ACS Applied Nano Materials*, 2020, **3**, 11383-11390.
4. C. Wang, J. Du, H. Wang, C. e. Zou, F. Jiang, P. Yang and Y. Du, *Sensors and Actuators B: Chemical*, 2014, **204**, 302-309.
5. X. Liu, L. Xie and H. Li, *Journal of Electroanalytical Chemistry*, 2012, **682**, 158-163.
6. B. Kaur, T. Pandiyan, B. Satpati and R. Srivastava, *Colloids and Surfaces B: Biointerfaces*, 2013, **111**, 97-106.
7. C.-S. Lee, S. H. Yu and T. H. Kim, *Nanomaterials*, 2018, **8**.
8. S. Mahalakshmi and V. Sridevi, *Electrocatalysis*, 2021, **12**, 415-435.
9. Z. Shi, X. Wu, Z. Zou, L. Yu, F. Hu, Y. Li, C. Guo and C. M. Li, *Biosensors and Bioelectronics*, 2021, **186**, 113303.