Supporting Information for

Determination of Aflatoxin B_1 in *Pixian Douban* based on

Aptamer-Magnetic Solid-Phase extraction

Chaoyi Zeng $^{a,d\#}\!\!$, Chi Xu $^{a\#}\!\!$, Hongyun Tian c , Kun Shao^a, Yaning Song^a, Xiao Yang^a ,

Zhenming Che^{a,b} and Yukun Huang

^a School of Food and Biological Engineering, Chongqing Key Laboratory of Speciality Food Co-

Built by Sichuan and Chongqing, Xihua University, Chengdu 610039, China

^b Key Laboratory of Food Non Thermal Processing, Engineering Technology Research Center of

Food Non Thermal Processing, Yibin Xihua University Research Institute, Yibin 644004, China

^c Shandong Institute of Food and Drug Control, Jinan 250101, China

^d Department of Food Biotechnology, Faculty of Biotechnology, Assumption University, Bangkok 10240, Thailand.

[#] The authors contributed equally in this work.

Figure S1 TEM images of Fe₃O₄ (a) and Fe₃O₄@SiO₂ (b)



Different Morphology of Fe_3O_4 before and after embedding SiO_2 . Figure S1. (a) Transmission Electron Microscope of Fe_3O_4 . (b) Transmission Electron Microscope of $Fe_3O_4@SiO_2$.

Figure S2 Particle size distribution of Fe₃O₄



The data are the characterization of the particle size of the prepared Fe3O4 by nanoparticle size analyzer.

Figure S3 XRD pattern of Fe₃O₄



The crystal form of the prepared nano-magnetic bead Fe_3O_4 was characterized by X-ray diffraction (XRD).

Figure S4 The dispersion stability of $Fe_3O_4@SiO_2$ in deionized water (a) and magnetic response to the applied magnetic field (b)



Figure S4. (a) $Fe_3O_4@SiO_2$ dispersed uniformly in deionized water . Figure S4. (b) In the presence of an external magnetic field, Fe_3O_4 @ SiO₂ can respond quickly and

gather in the direction of magnet, indicating that the Fe_3O_4 surface coated with SiO_2 still maintains strong magnetic properties.



Figure S5 The FT - IR spectra of Fe3O4 (a) and Fe3O4@SiO2 - NH2 (b)

Fourier Transform Infrared Spectroscopy (FT-IR) Analysis of the Surface Groups of Fe_3O_4 and Amino Modified Silica Magnetic Beads.

Figure S6 UV - V is absorption spectra of avidin (a) and aptamer (b) before and after the reaction



(a1) avidin stock solution; (b1) avidin solution after glutaraldehyde; (a2) aptamer solution; (b2) aptamer solution after reacting with avidinized magnetic beads.

Figure S7 Detection of AFB₁ by HPLC-MS/MS. Standard curve of AFB₁(a). MRM chromatogram of AFB₁ in the Pixian Douban sample (b, c)



The linear equations of AFB_1 in blank *Pixian Douban* samples are $y = 183.23x - 95.46(R^2 = 0.99989)$. The mass spectrometric analysis was performed in MRM. For fragmentation of the

 $AFB_1[M+H]^+$ ions is 313 m/z. The detected and quantified fragment ions were: 241 and 269 m/z for AFB_1 .