

## Supplemental files

### A portable smartphone-based detection of glyphosate based on inhibiting peroxidase-like activity of heptanoic acid and Prussian blue decorated Fe<sub>3</sub>O<sub>4</sub> nanoparticles

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**Tables:**

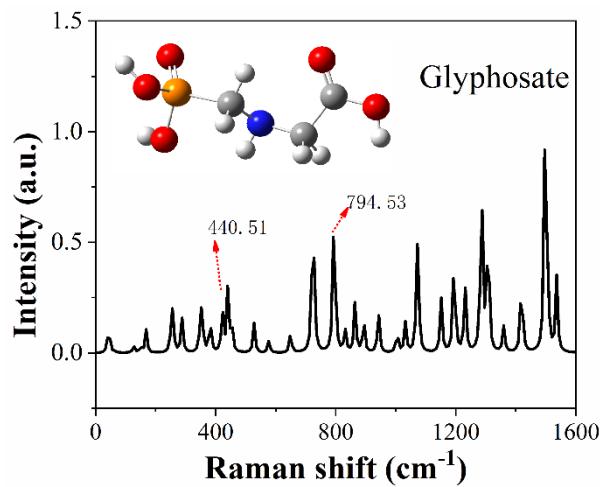
**Table S1** The precipitation efficiencies of glyphosate by co-precipitation method.

Samples	Spiked glyphosate ( $\mu\text{g mL}^{-1}$ )	Precipitation efficiencies (%)	RSD (%)
Tobacco 1	12.5	3.24	3.33
	50	3.11	2.14
Tobacco 2	12.5	2.47	4.12
	50	2.91	3.42
Water 1	12.5	1.87	4.38
	50	2.44	2.67
Water 2	12.5	2.64	2.87
	50	2.75	4.09

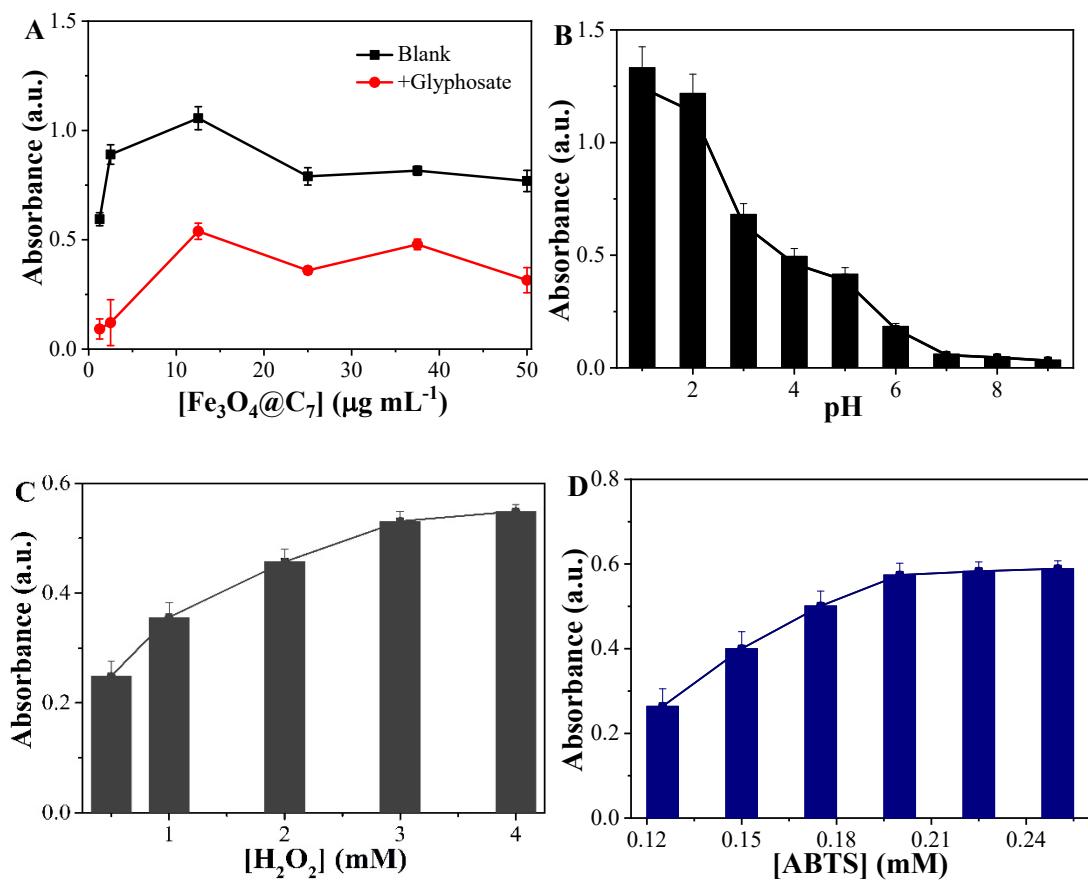
**Table S2** Determination of glyphosate in spiked tobacco samples (n=6).

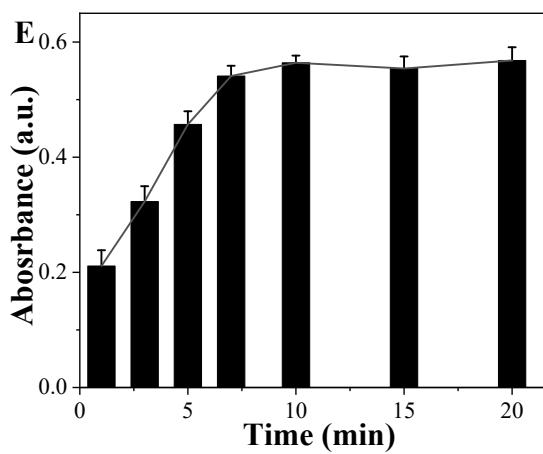
Tobacco samples	Spiked ( $\mu\text{g mL}^{-1}$ )	Smartphone color sensing platform			GC-MS		
		Detected ( $\mu\text{g mL}^{-1}$ )	Recovery (%)	RSD (%)	Detected ( $\mu\text{g mL}^{-1}$ )	Recovery (%)	RSD (%)
1	0	0	-	-	0	-	-
	12.5	11.18	89.44	3.53	11.65	93.2	3.23
	50	48.55	97.1	4.67	48.54	97.08	4.32
2	0	0	-	-	0	-	-
	12.5	11.57	92.56	1.89	11.64	93.12	2.65
	50	48.34	96.68	4.34	47.87	95.74	5.21
3	0	0	-	-	0	-	-
	12.5	11.83	94.64	5.38	11.03	88.24	3.59
	50	46.56	93.12	3.53	48.32	96.64	4.07

**Figures:**



**Fig. S1** The standard Raman spectra of glyphosate.





**Fig. S2** Effect of the concentrations of  $\text{Fe}_3\text{O}_4@\text{C}_7$  (A), pH (B) and substrate concentrations of  $\text{H}_2\text{O}_2$  (C) and ABTS (D).