Benzothiazole appended 2,2'-(1,4-phenylene)diacetonitrile for the calorimetric and fluorescence detection of cyanide ions

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Fig S1 ¹H NMR spectra of PDBT.



Fig S2 ¹³C NMR spectra of PDBT.



Fig S3 HRMS spectra of PDBT.



Fig S4. The linear plot of absorption intensity against the concentration of CN^{-} from 1 to 20 μ M in ACN/H₂O (1:1, v/v).



Fig S5. Job's plot for determining the binding stoichiometry of PDBT and CN⁻ ions.



Fig S6 HRMS spectra of PDBT+CN-.



Fig S7. Optimized structures of PDBT and PDBT+CN⁻.

	λ max(nM)	Oscillatory	ΔE,Energy(eV)	Selected major
Entry		strength		transitions
	434.72	1.8799	18.7282	H → L (47%)
	370.21	0.0733	17.7472	H → L ₂ (42%)
PDBT	329.02	0.3664	14.9453	H ₁ →L ₁ (30%)
	322.42	0.0675	16.3412	H₃ → L (36%)
	317.54	0.0310	15.5306	H₅—►L (32%)
	312.49	0.1017	18.2697	H ₁ →L ₃ (45%)
	474.08	0.8540	18.7559	H—▶L (47%)
	402.98	0.0809	13.9161	H —→L₁(26%)
PDBT+ CN-	387.84	0.0018	18.4784	H₄ → L (46%)
	351.68	0.0003	18.8697	H ₂ → L ₁ (48%)
	303.12	0.1183	8.0216	H ₃ → L ₂ (8.6%)
	302.85	0.0168	9.8015	$H_1 \longrightarrow L_2$ (12%)

Table S1. Oscillator strengths of important transitions for the PDBT and PDBT + CN⁻.















Fig S8. Individual electronic spectra of PDBT with various selective anions.



Fig S9. Standard deviation calculation spectra of PDBT.