

Supplementary Information

Synthesis, crystal structure, and properties of methyl-substituted coronene amide analogue

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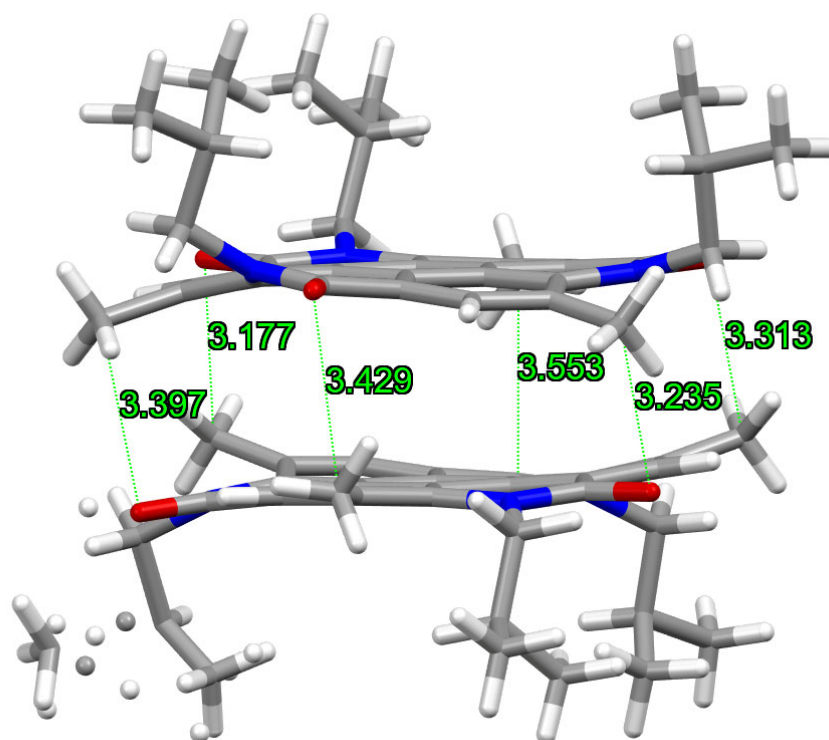


Fig. S1. Multiple CH₃...O contacts (with distance (Å) between C and O) in the stacking molecules.

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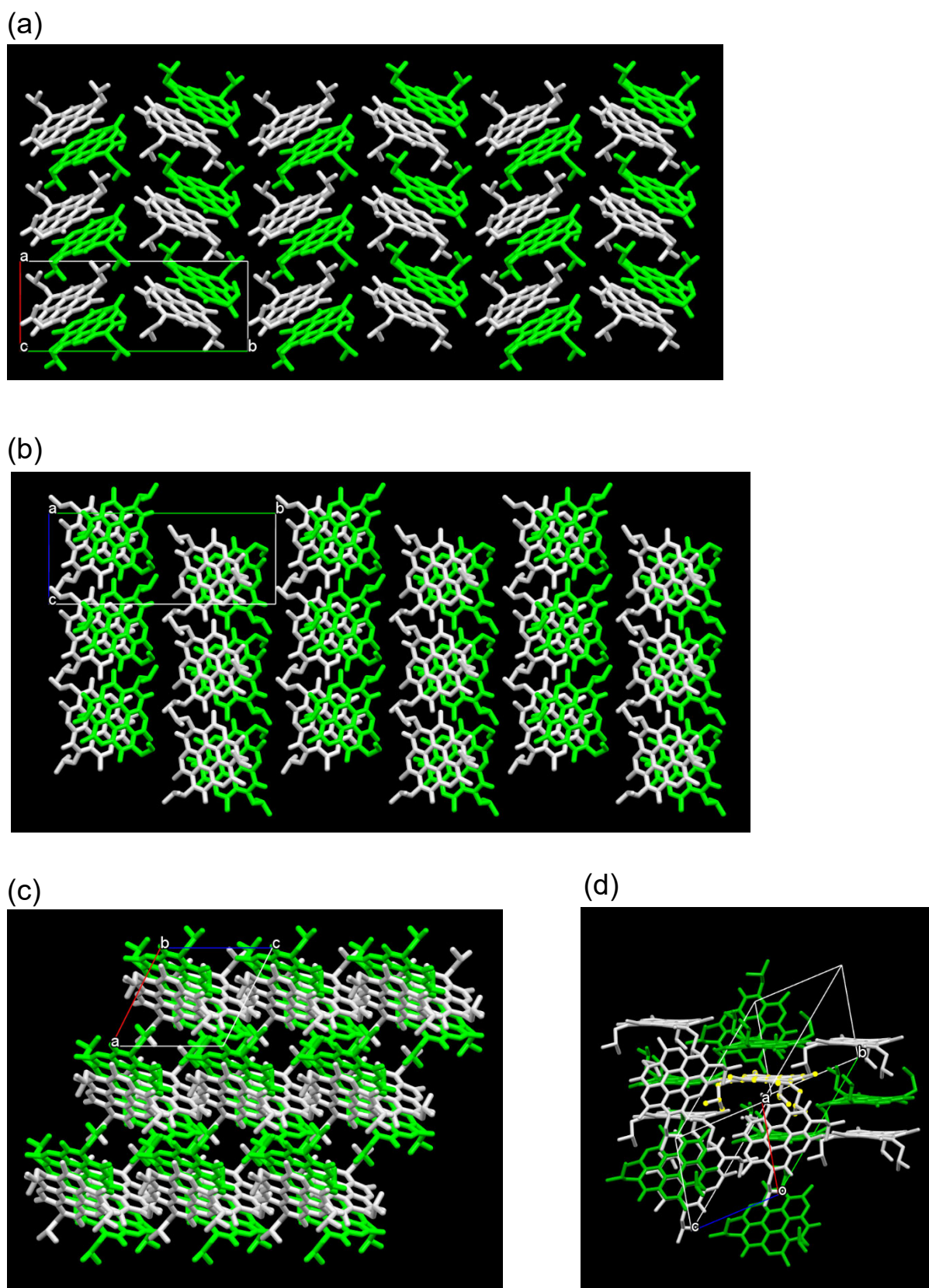
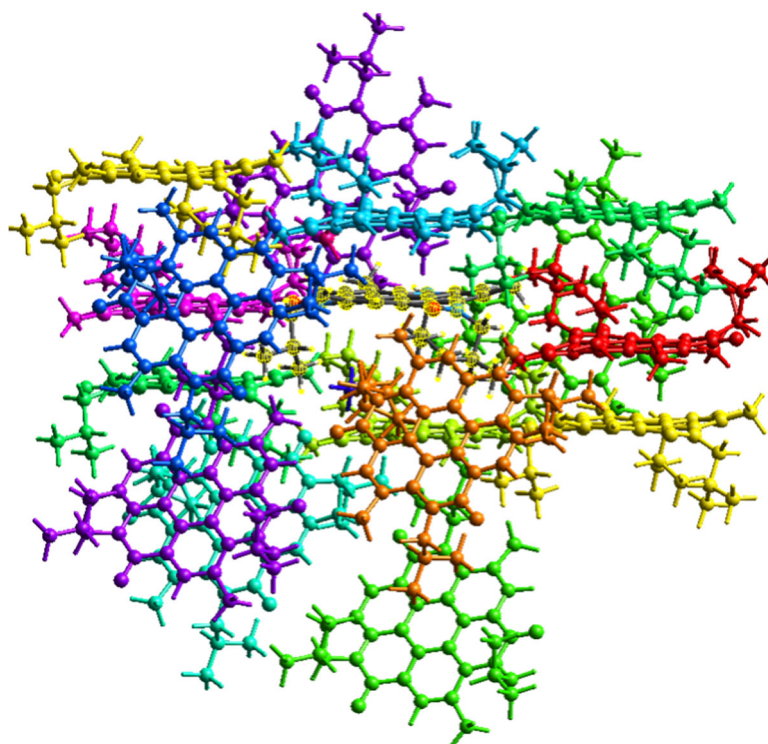


Fig. S2. Packing structure of **3** with symmetry operation viewed down (a) *c*-axis, (b) *b*-axis, and (c) *a*-axis. (d) Packing structure around one of **3** (the same direction as Fig. S3).



Crystal Atoms Surface **Energies**

Interaction Energies (kJ/mol)

R is the distance between molecular centroids (mean atomic position) in Å.

Total energies, only reported for two benchmarked energy models, are the sum of the four energy components, scaled appropriately (see the scale factor table below)

	N	Symop	R	Electron Density	E_ele	E_pol	E_dis	E_rep	E_tot
	1	-	10.40	HF/3-21G	-29.0	-15.4	-63.2	36.9	-66.5
	1	-	10.58	HF/3-21G	-10.4	-8.0	-63.3	33.7	-45.5
	2	x, y, z	11.94	HF/3-21G	-0.4	-0.7	-17.1	8.6	-9.3
	1	-	8.29	HF/3-21G	-9.5	-6.0	-90.3	41.5	-61.3
	2	-x, y+1/2, -z	15.42	HF/3-21G	0.1	-0.0	-2.8	0.1	-2.4
	2	x, y, z	11.45	HF/3-21G	-4.4	-2.2	-23.3	10.7	-18.3
	1	-	12.28	HF/3-21G	-4.9	-3.7	-16.5	8.5	-15.3
	1	-	4.93	HF/3-21G	-38.2	-12.0	-200.4	96.2	-149.3
	1	-	12.66	HF/3-21G	0.2	-1.0	-15.1	7.3	-8.2
	1	-	5.62	HF/3-21G	0.0	nan	0.0	0.0	nan
	2	-x, y+1/2, -z	14.37	HF/3-21G	0.0	-0.5	0.0	0.0	-0.3
	1	-	11.47	HF/3-21G	-25.3	-13.6	-34.0	35.4	-36.6
	1	-	9.14	HF/3-21G	0.3	-0.5	-13.8	6.0	-7.6

← stacking pair

Scale factors for benchmarked energy models
See Mackenzie et al. IUCrJ (2017)

Energy Model	k_ele	k_pol	k_disp	k_rep
CE-HF ... HF/3-21G electron densities	1.019	0.651	0.901	0.811
CE-B3LYP ... B3LYP/6-31G(d,p) electron densities	1.057	0.740	0.871	0.618

Fig. S3. The interaction energies of various pairwise molecules estimated using CE-HF/3-21G in CrystalExplorer 17.5.¹ Total interaction energy includes electrostatic, polarization, dispersion, and exchange-repulsion terms. The viewing direction is the same as in Fig S2d.

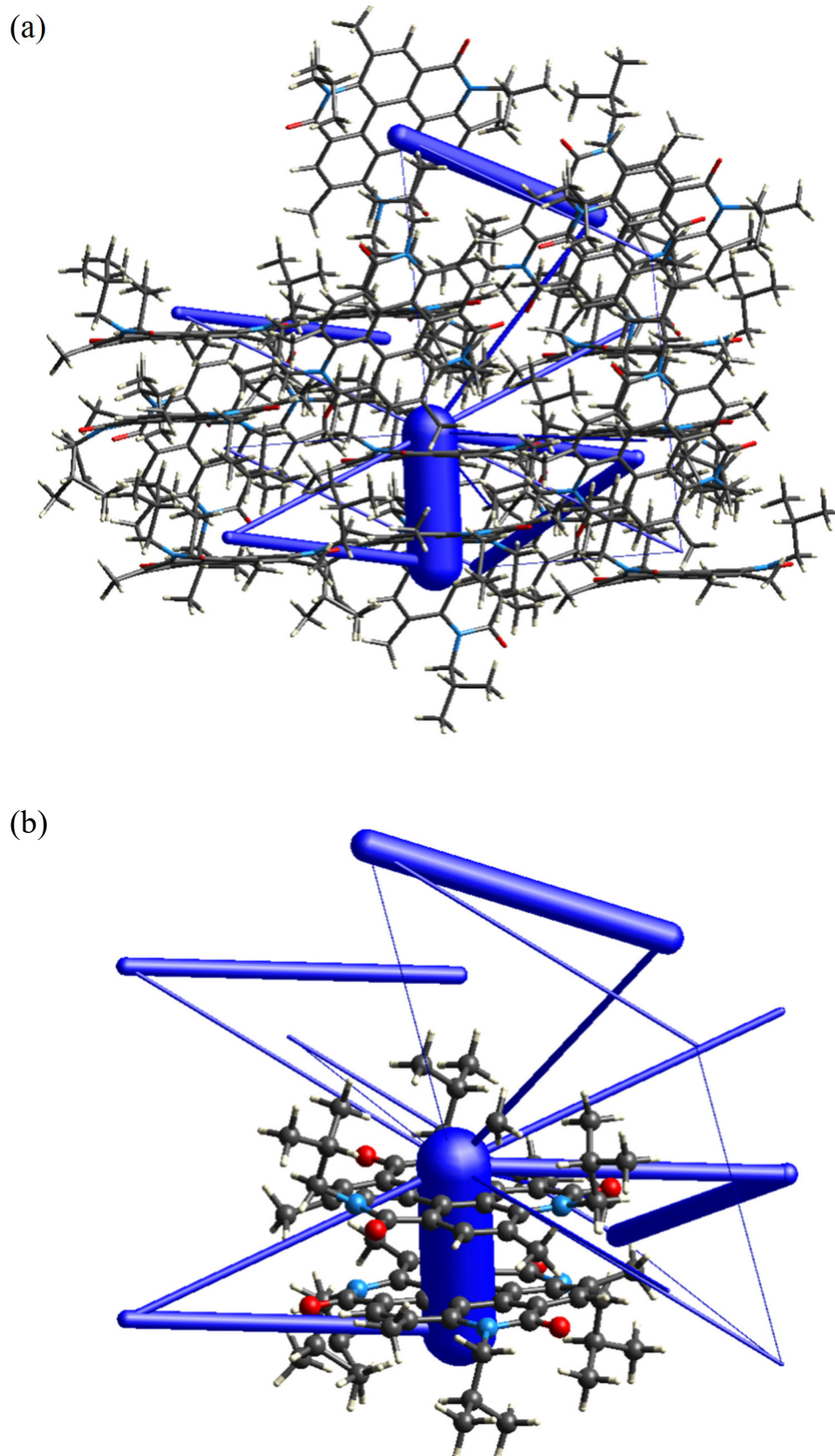


Fig. S4. Energy frameworks of **3** in (a) total interaction strengths and (b) extracted dimers. Energy frameworks were constructed from pairwise intermolecular interaction energy calculations (at crystal geometry).

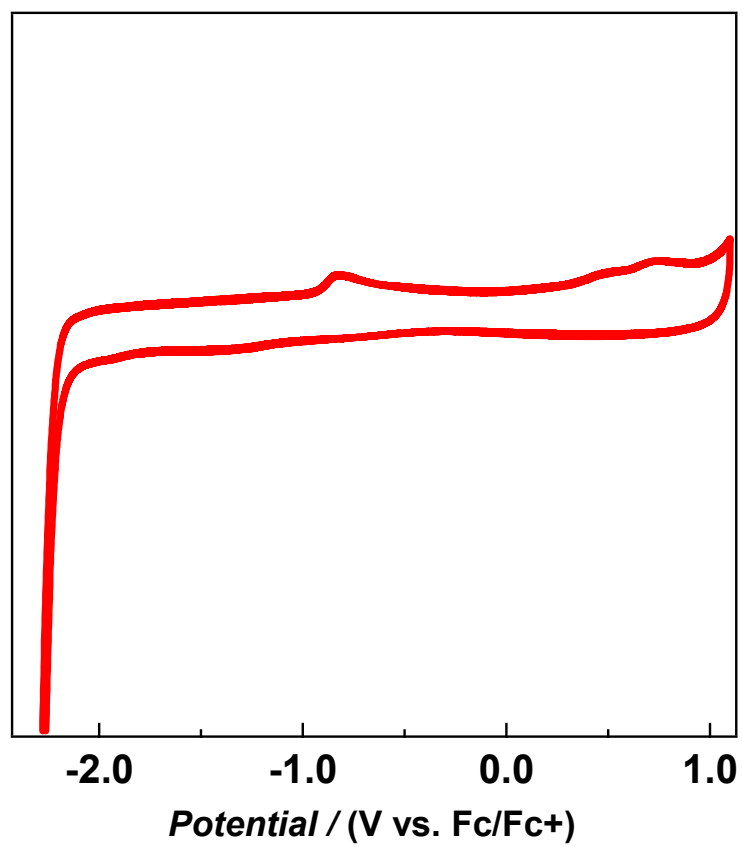


Fig. S5. Cyclic voltammogram of **3** measured in CH₂Cl₂ (0.1 M *n*-Bu₄NPF₆) at a scan rate of 0.1 V/s.

Table S1. Crystallographic data and refinement parameters for **3**.

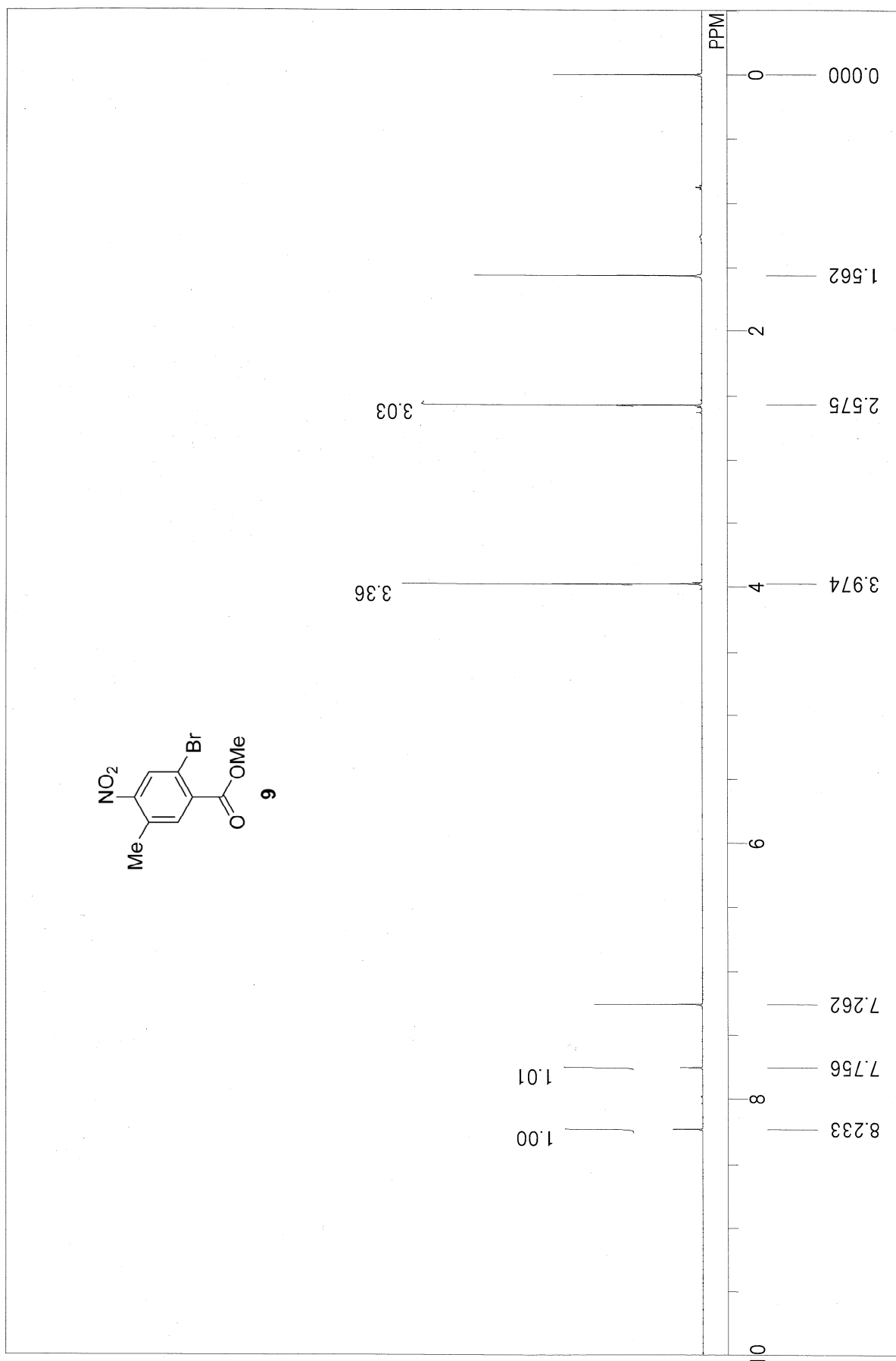
One of isobutyl groups showed positional disorder. Two possible positions having different orientations were assumed with 50%/50% probabilities for C37A, C39A and C37B, C39B, respectively.

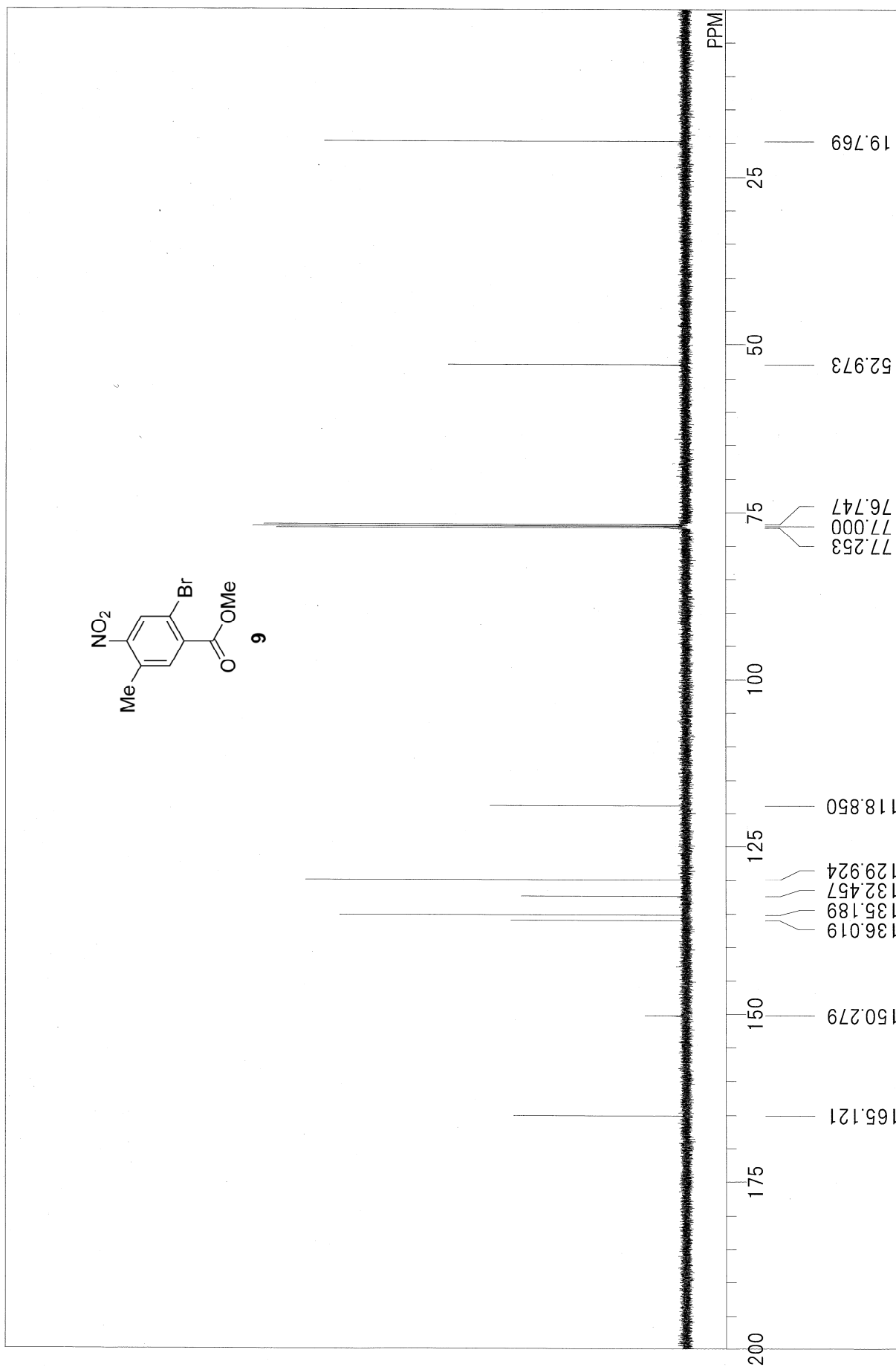
Formula	C ₃₆ H ₃₉ N ₃ O ₃
M_r	561.72
Crystal system	Monoclinic
Space group	$P2_1$
$a / \text{\AA}$	11.26590(10)
$b / \text{\AA}$	25.57580(10)
$c / \text{\AA}$	11.44520(10)
α / deg	90
β / deg	116.5650(10)
γ / deg	90
$V / \text{\AA}^3$	2949.60(5)
Z	4
Z'	2
Temperature / K	93
Goodness-of-fit on F^2	1.041
$R_1 [I > 2\sigma(I)]$ on F	0.0383
wR_2 (all data) on F^2	0.1035
Reflection collected (all data)	39543
No. of reflections [$I > 2\sigma(I)$]	10048
R_{int}	0.0225
Abs. corr.	multi-scan
Flack parameter	0.55(3)
Radiation type	CuK α ($\lambda = 1.54187 \text{\AA}$)
T_{max}	0.838
T_{min}	0.981
$2\theta_{\text{max}}$	68.228
μ / mm^{-1}	0.637
CCDC No.	2130884

References

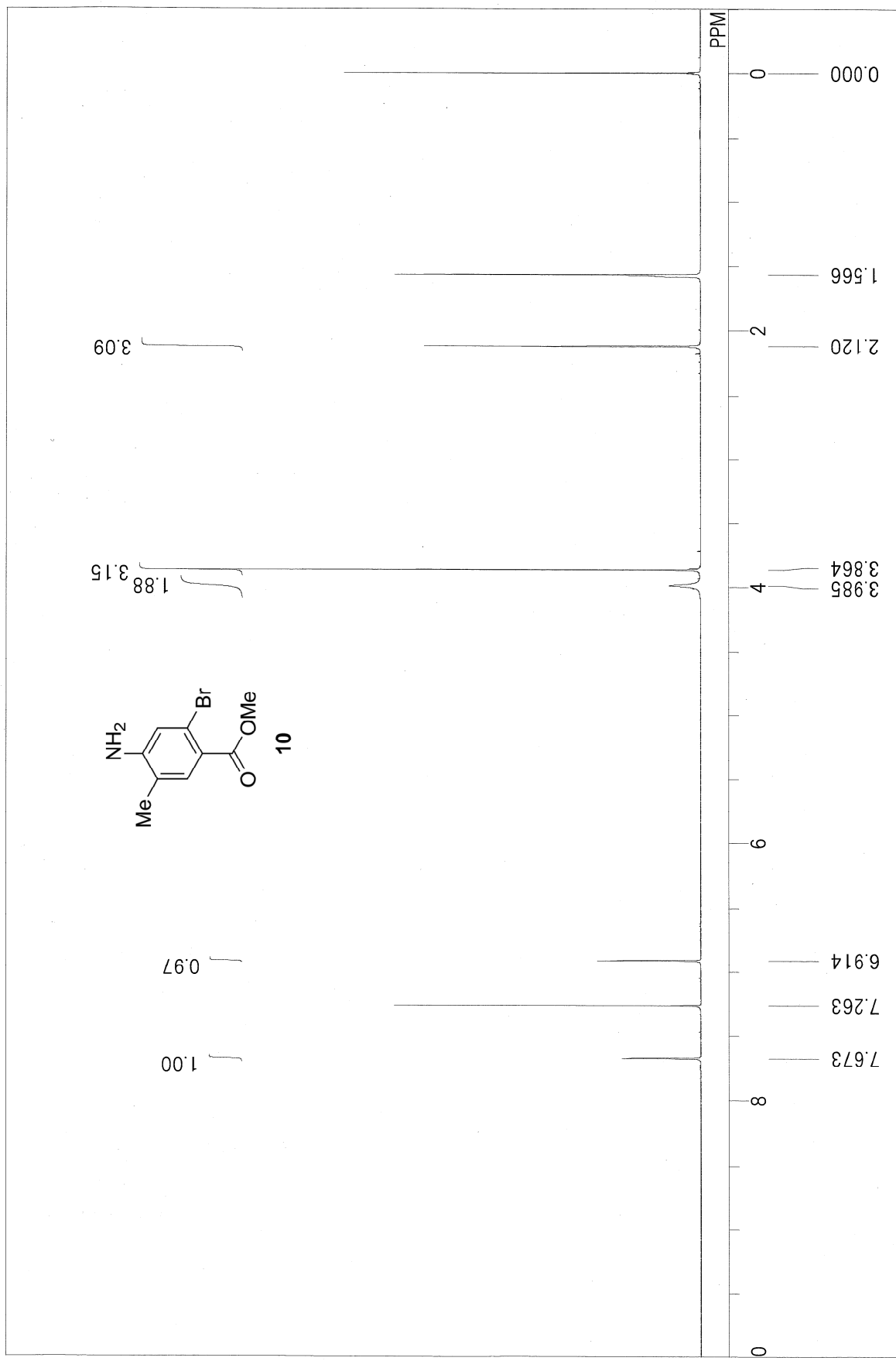
- 1 Spackman, P. R.; Turner, M. J.; McKinnon, J. J.; Wolff, S. K.; Grimwood, D. J.; Jayatilaka, D.; Spackman, M. A. *J. Appl. Cryst.* **2021**, *3*, 1006–1011.

^1H NMR of **9** (CDCl_3 , 500 MHz)

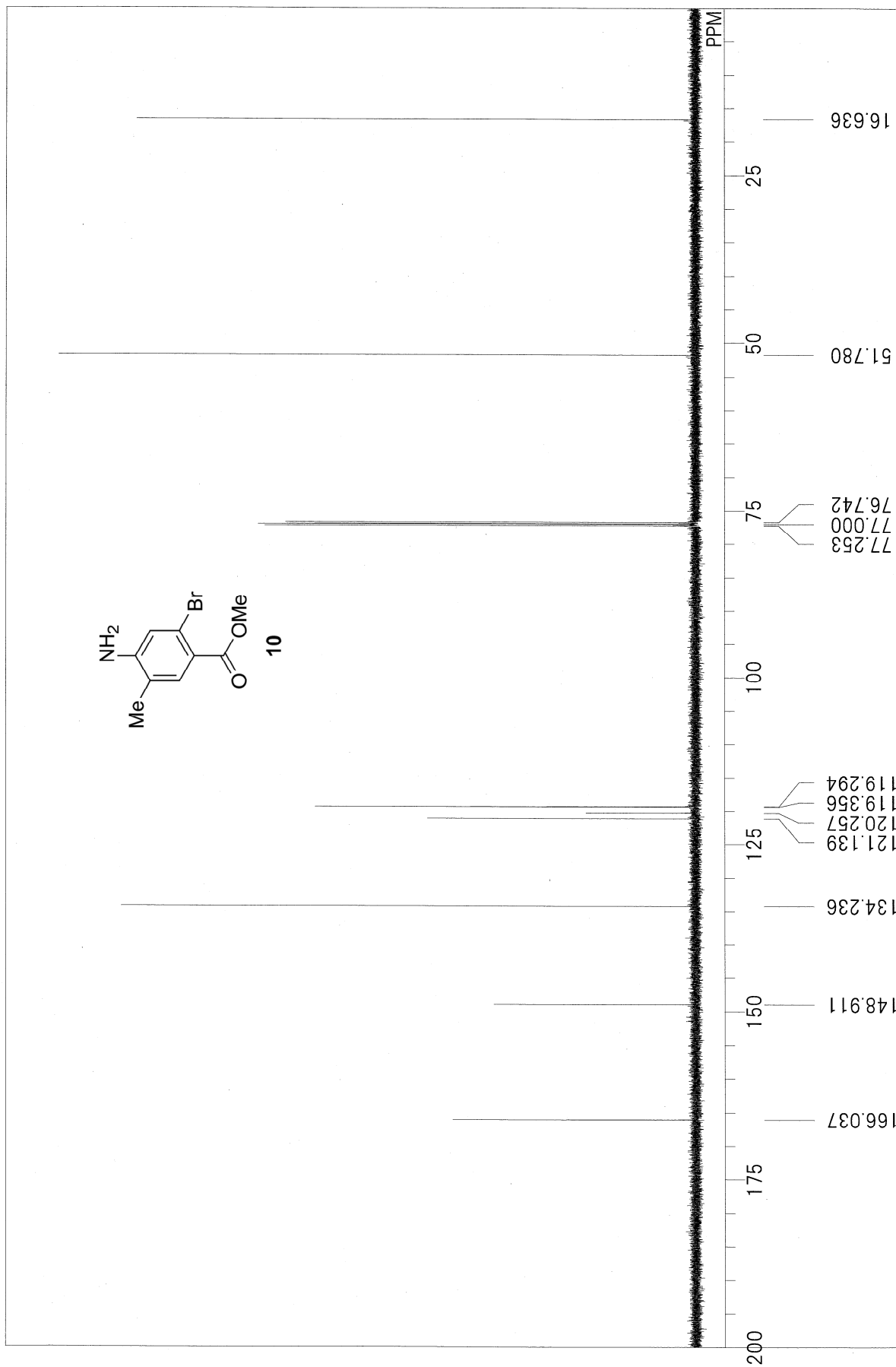




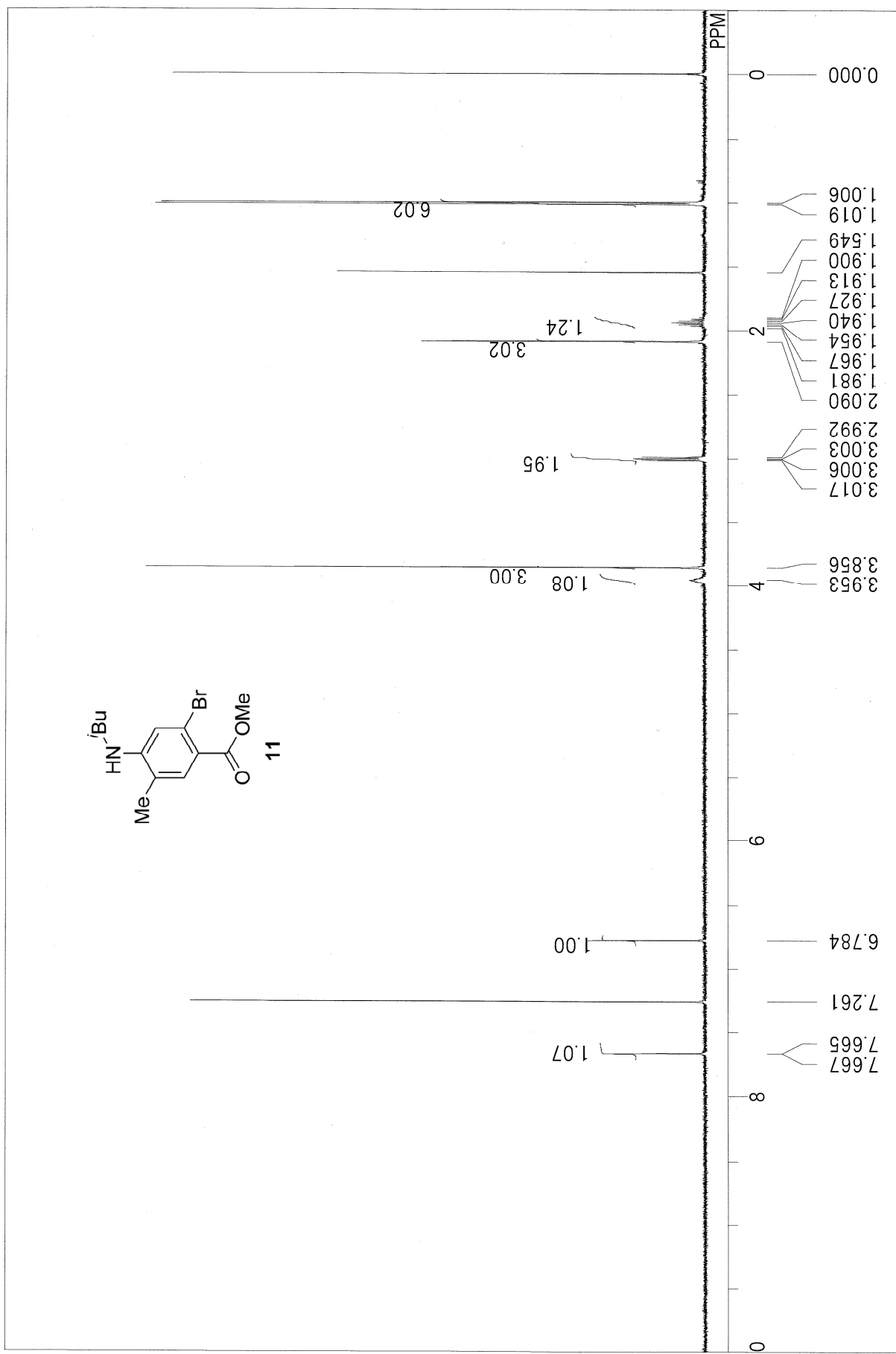
^1H NMR of **10** (CDCl_3 , 500 MHz)



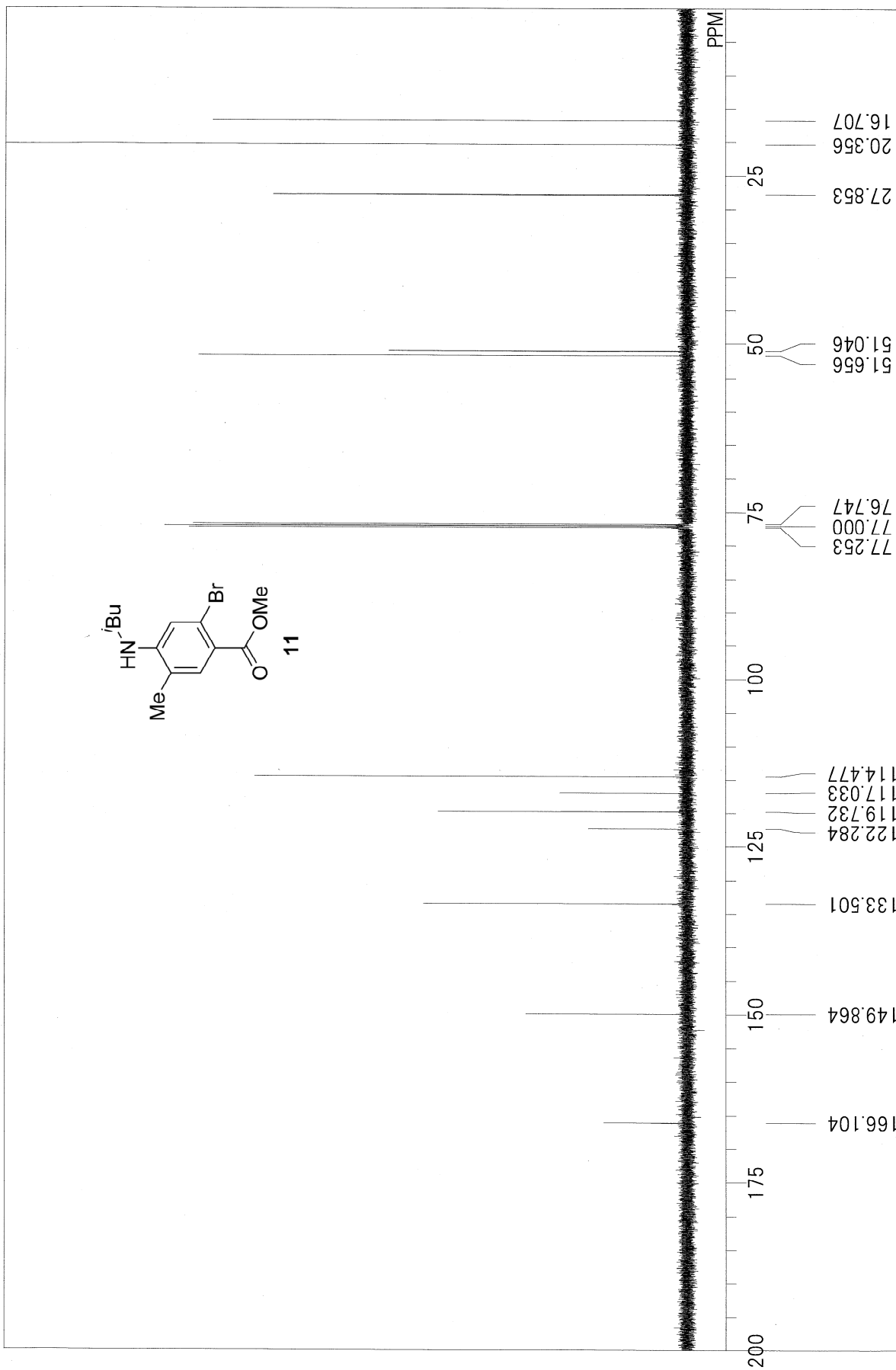
¹³C NMR of **10** (CDCl₃, 126 MHz)



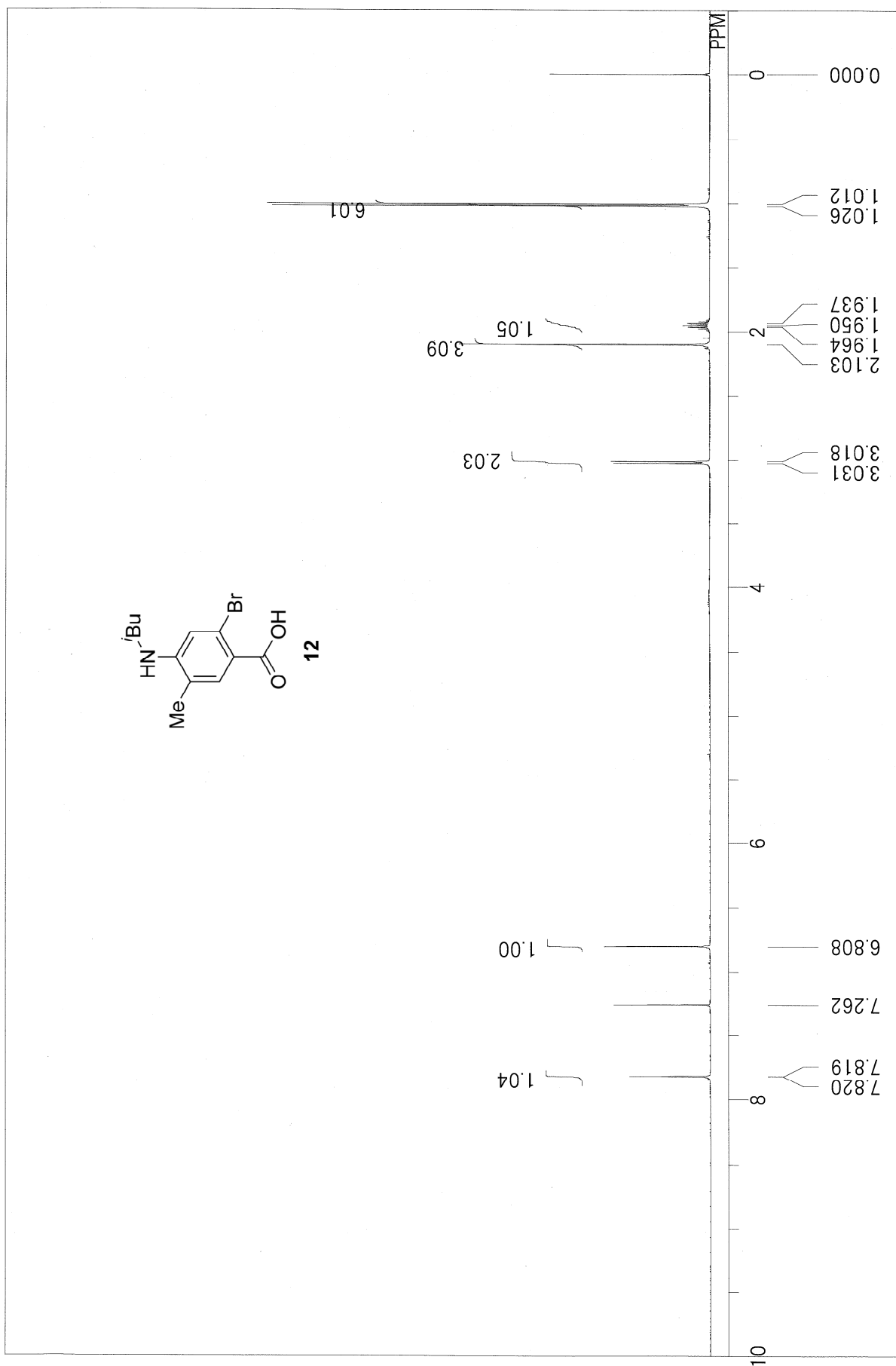
¹H NMR of **11** (CDCl₃, 500 MHz)



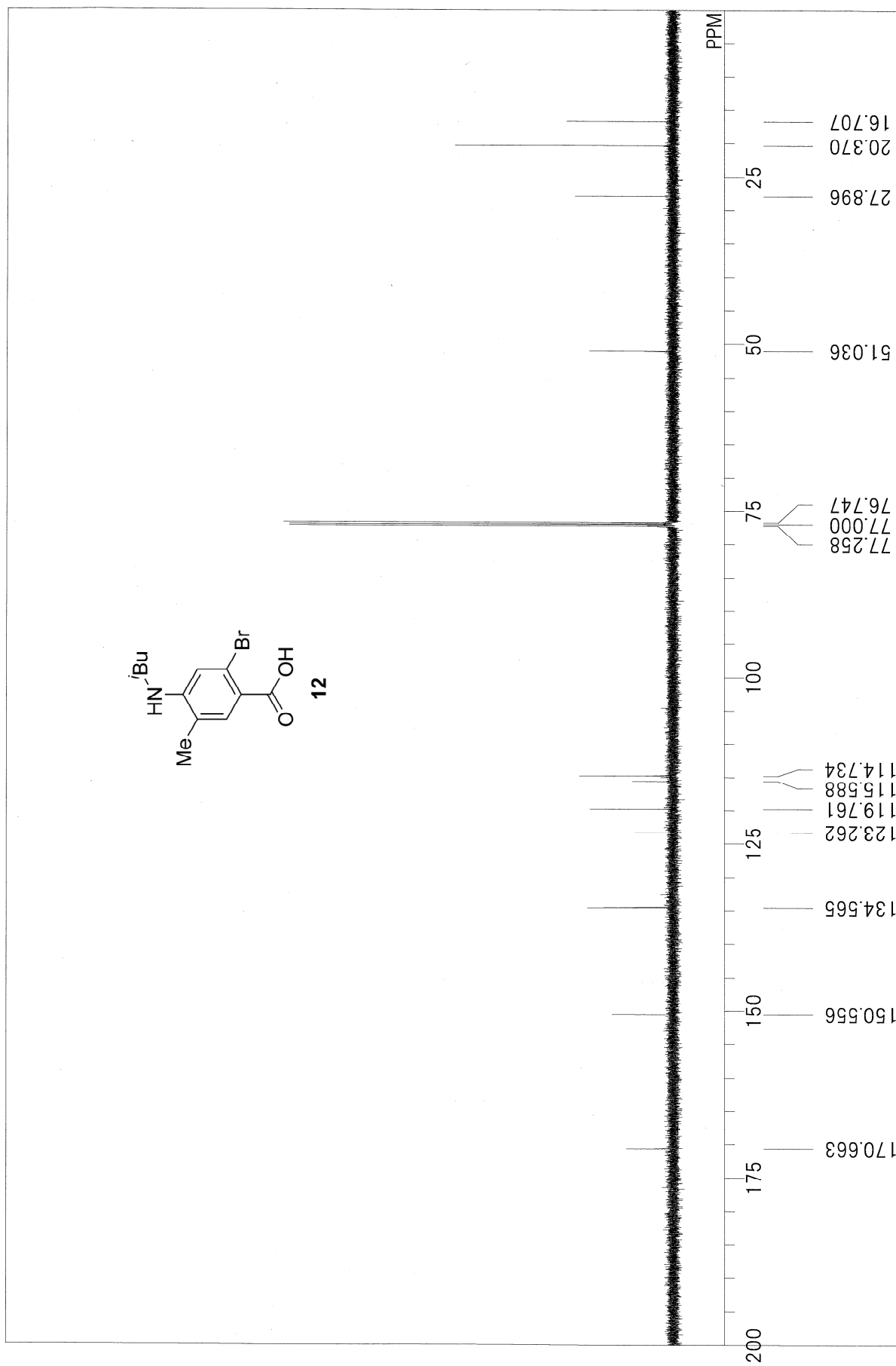
^{13}C NMR of **11** (CDCl_3 , 126 MHz)



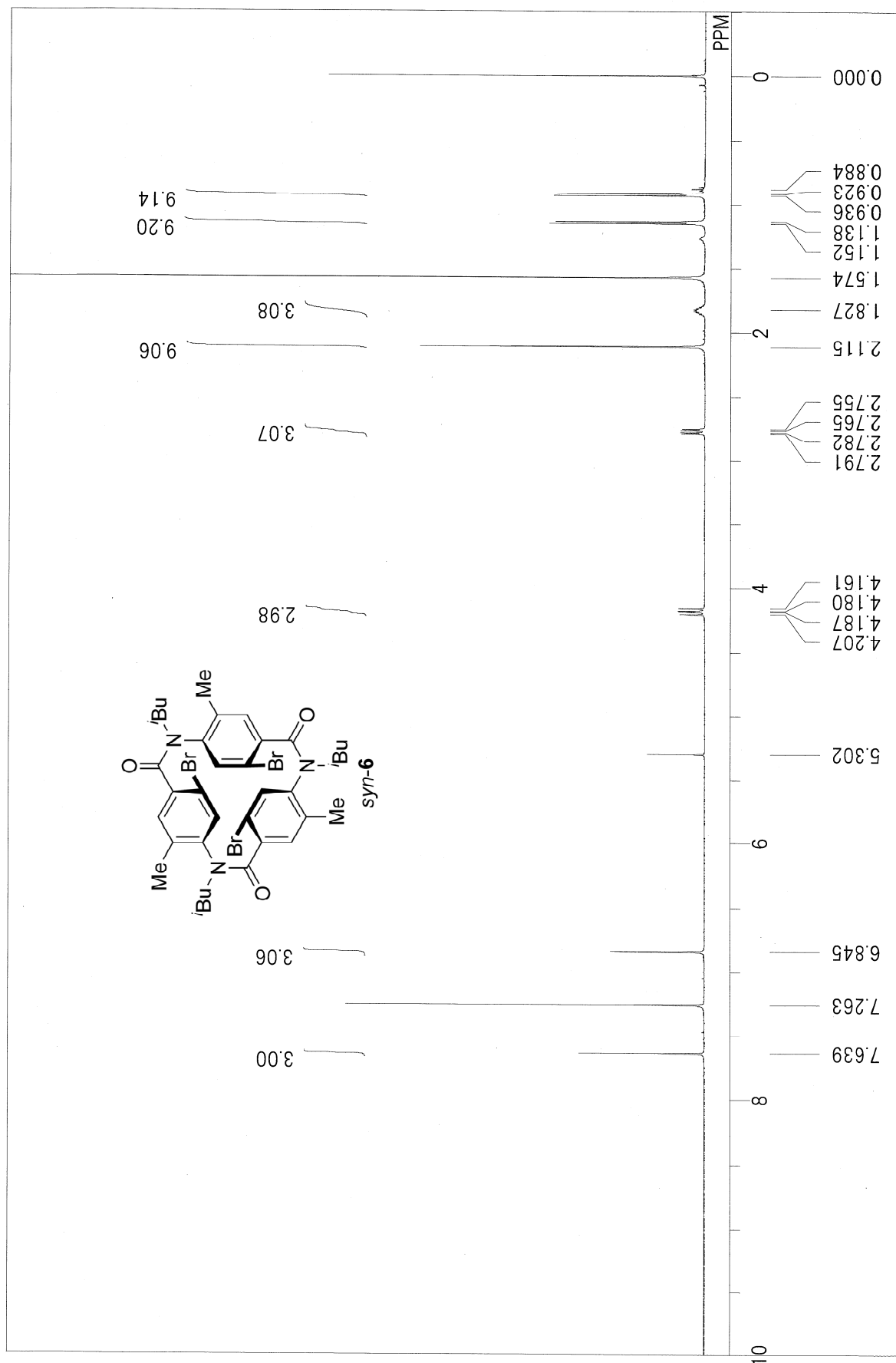
^1H NMR of **12** (CDCl_3 , 500 MHz)



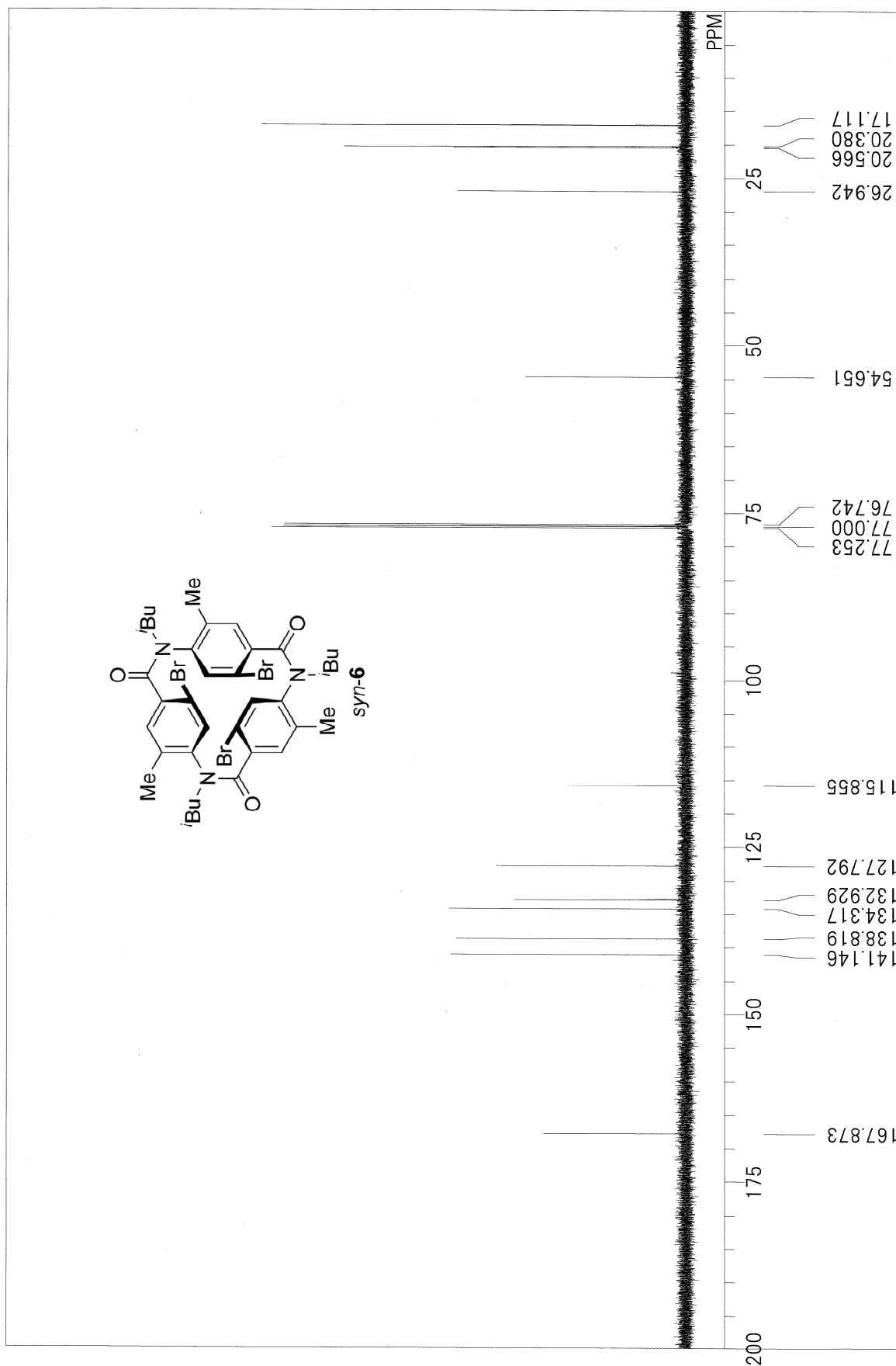
^{13}C NMR of **12** (CDCl_3 , 126 MHz)



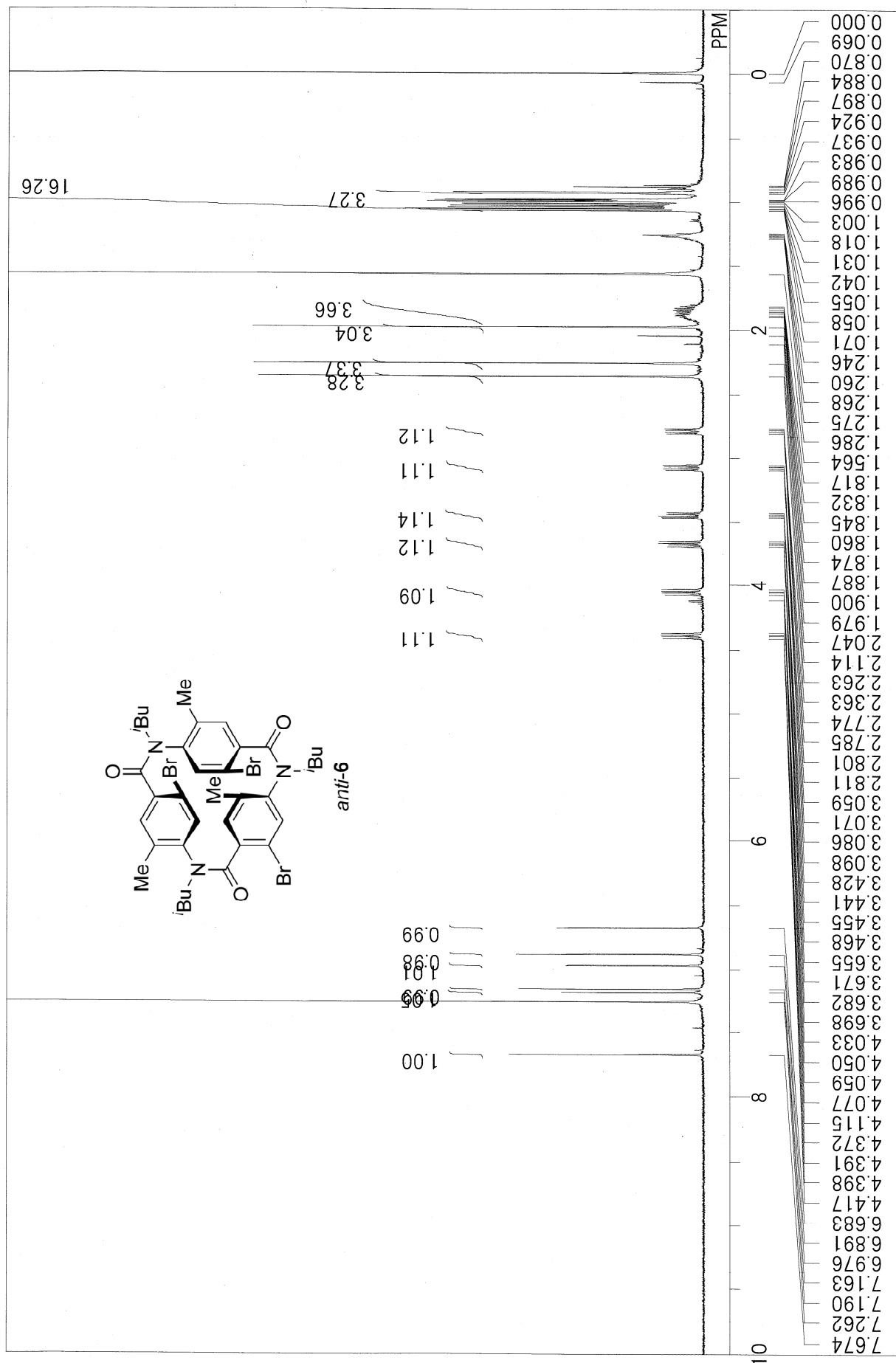
^1H NMR of *syn-9* (CDCl_3 , 500 MHz)

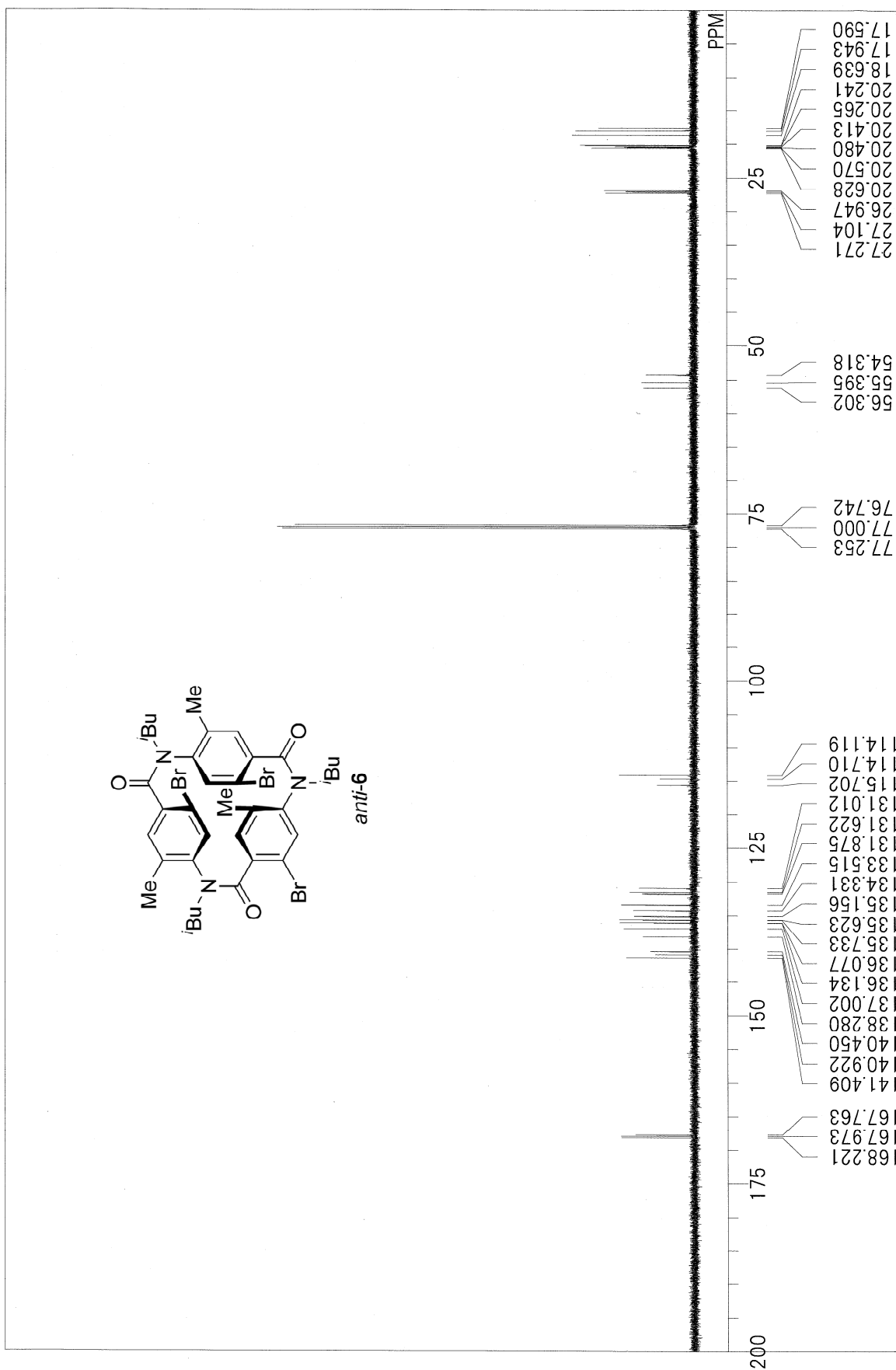


¹³C NMR of *syn-6* (CDCl₃, 126 MHz)

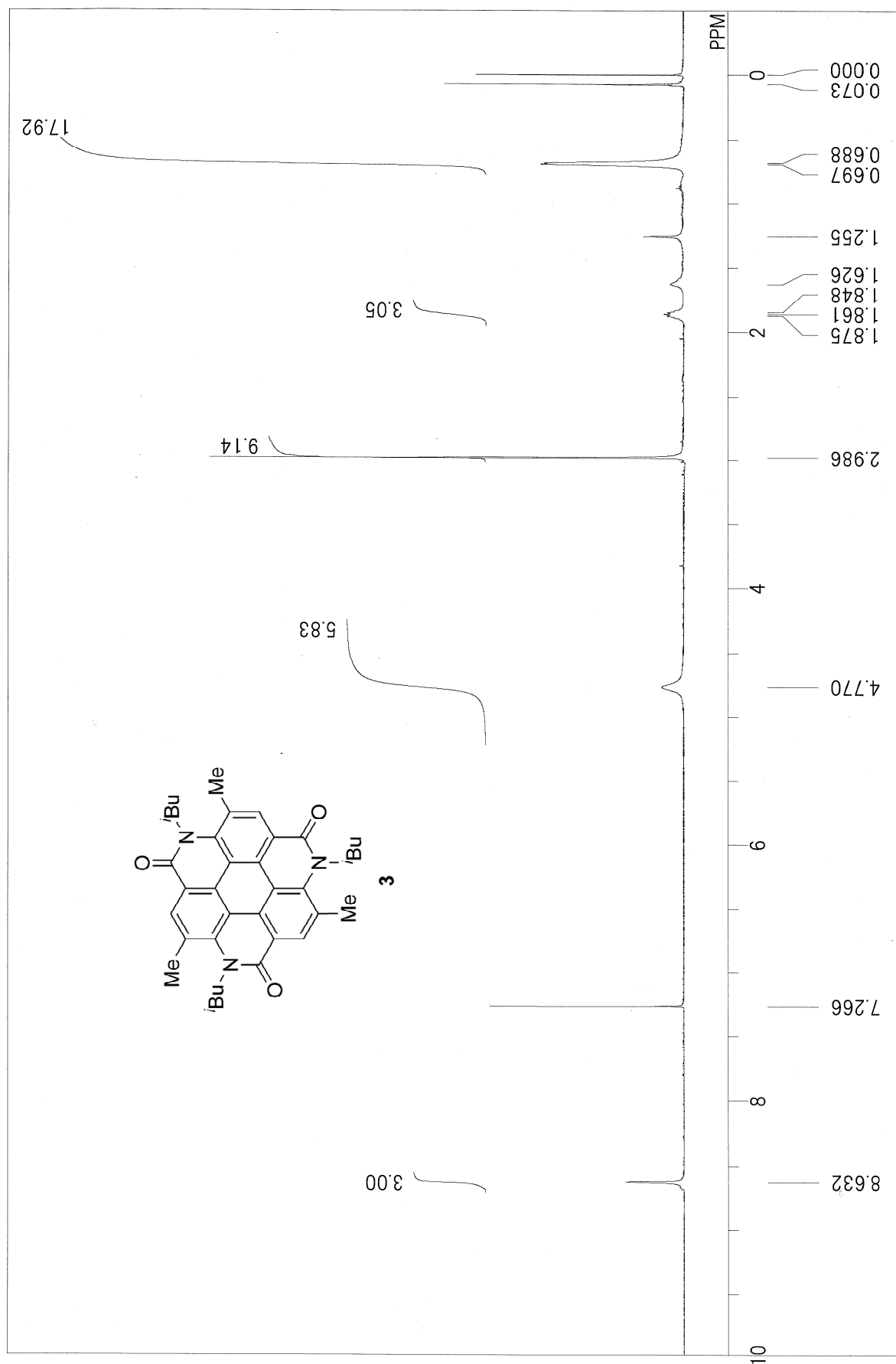


¹H NMR of *anti*-6 (CDCl₃, 500 MHz)

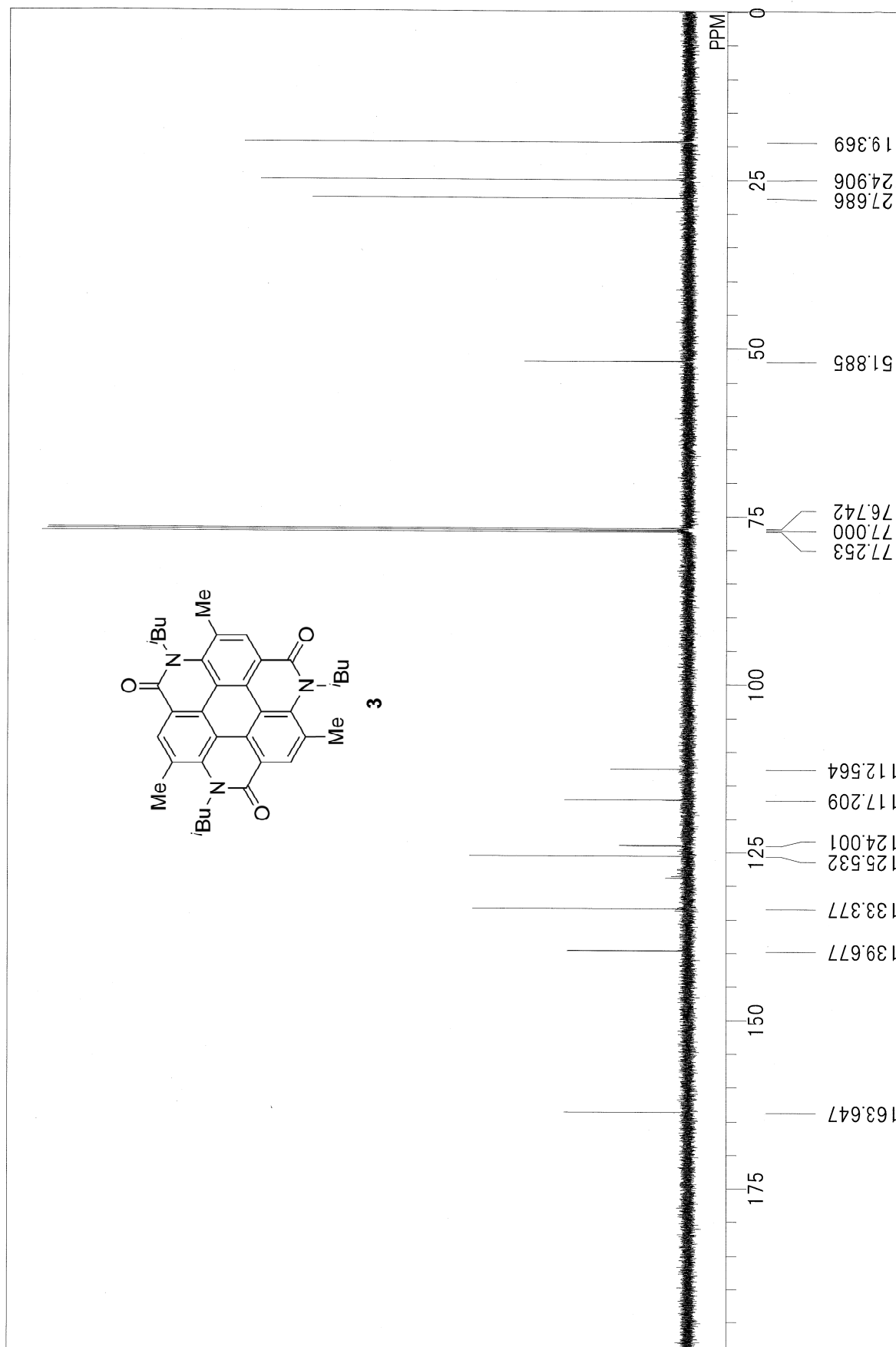




^1H NMR of **3** (CDCl_3 , 500 MHz)



^{13}C NMR of **3** (CDCl_3 , 126 MHz)



¹H NMR of **13** (CDCl₃, 500 MHz)

