## Metallic boro-carbides of A<sub>2</sub>BC (A=Ti, Zr, Hf and W): A comprehensive theoretical study for thermo-mechanical and optoelectronic applications

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Fig. S1: The orthorhombic crystal structure of Ti<sub>2</sub>BC compound.



Fig. S2: Electronic partial density of states (DOS) for (a)  $Ti_2BC$ , (b)  $Zr_2BC$ , (c)  $Hf_2BC$  and (d)  $W_2BC$  compounds.



**Fig. S3:** Photon energy dependent (a) real part of dielectric function ( $\mathcal{E}_1$ ) and imaginary part of dielectric function ( $\mathcal{E}_2$ ), and (b) refractive index (*n*) and extinction coefficient (*k*) of A<sub>2</sub>BC (A = Ti, Zr, Hf, and W) compounds.

<b>Table S1:</b> Lattice parameters (orthorhombic), $a$ , $b$ and $c$ (all in Å), unit cell volume $V$ (in Å <sup>3</sup> ) of
$A_2BC$ (A = Ti, Zr, Hf, and W) compounds and other available metallic boro-carbides
compounds.

Compound	а	b	С	V	Ref.
Ti <sub>2</sub> BC	3.165	17.587	3.089	169.11	This
	3.169	17.660	3.048	170.57	1
Zr <sub>2</sub> BC	3.431	19.288	3.256	215.52	This
	3.437	19.473	3.273	219.08	1
$Hf_2BC$	3.420	19.127	3.256	213.04	This
	3.388	19.019	3.226	207.91	1
W <sub>2</sub> BC	3.115	17.480	3.068	167.25	This
	3.125	17.578	3.086	169.54	1
V <sub>2</sub> BC	2.999	16.770	2.956	148.70	1
	2.998	16.647	2.948	147.17	2
Nb <sub>2</sub> BC	3.227	18.345	3.149	186.41	1
	3.224	18.239	3.141	184.71	2
Mo <sub>2</sub> BC	3.119	17.580	3.082	169.01	1
	3.137	17.702	3.094	171.86	2
	3.094	17.768	3.091	169.93	3
	3.086	17.350	3.047	163.14	4
Ta <sub>2</sub> BC	3.221	18.246	3.140	184.51	1
	3.267	18.445	3.180	191.69	2

Compound	$C_{11}$	$C_{12}$	$C_{13}$	$C_{23}$	$C_{22}$	$C_{33}$	$C_{44}$	$C_{55}$	$C_{66}$	СР	Ref.
Ti <sub>2</sub> BC	417	132	108	92	379	468	147	208	159	-15	This
$Zr_2BC$	374	124	96	96	325	408	121	175	130	+03	This
$Hf_2BC$	420	130	128	108	373	440	132	206	146	-02	This
$W_2BC$	563	257	240	258	532	572	184	291	189	+73	This
$V_2BC$	474	148	159	115	450	525	161	257	146	-13	2
$Nb_2BC$	482	154	156	134	488	504	140	251	130	+14	2
Mo <sub>2</sub> BC	520	209	218	193	545	547	176	265	180	+33	2
	551	211	204	210	566	553	168	241	182	+43	3
Ta <sub>2</sub> BC	503	157	181	153	468	528	134	258	125	+23	2

**Table S2:** The calculated elastic stiffness constants  $C_{ij}$  (GPa) and Cauchy pressure (CP) (GPa) of A<sub>2</sub>BC (A = Ti, Zr,Hf and W) compounds with other metallic boro-carbides.

**Table S3:** Shear anisotropy factors ( $A_i$ ) where i stand for 1, 2 and 3; bulk modulus (GPa)  $B_a$ ,  $B_b$  and  $B_c$  along *a*-, *b*- and *c*-directions, respectively; anisotropy factors in compressibility ( $A_B$ ) and shear moduli ( $A_G$ ) (in %), and universal anisotropy index,  $A^U$  for A<sub>2</sub>BC (A = Ti, Zr, Hf and W) compounds. The values for other metallic boro-carbides are given for comparison.

Compound	$A_1$	$A_2$	$A_3$	$B_a$	$\mathbf{B}_{\mathbf{b}}$	$B_{c}$	$A_{\rm B}$	$A_{\rm G}$	$A^{\mathrm{U}}$	Ref.
Ti <sub>2</sub> BC	0.87	1.25	1.195	685	561	691	0.234	1.54	0.16	This
Zr <sub>2</sub> BC	0.82	1.30	1.15	624	503	622	0.22	1.47	0.15	This
$Hf_2BC$	0.87	1.38	1.10	716	564	698	0.23	1.31	0.14	This
$W_2BC$	1.12	1.98	1.30	1066	1015	1099	0.015	2.92	0.30	This
$V_2BC$	0.94	1.38	0.93	809	663	833	0.251	2.134	0.22	2
Nb <sub>2</sub> BC	0.83	1.38	0.78	799	763	801	0.012	2.622	0.26	2
Mo <sub>2</sub> BC	1.11	1.50	1.11	939	942	972	0.005	1.672	0.17	2
Ta <sub>2</sub> BC	0.80	1.50	0.76	861	719	915	0.227	3.247	0.34	2

	Atomic population							Bond overlap population			
Compound	Atom	S	р	d	Total	Charge (e)	EVC (e)	Bond	Bond number	Bond length $d^{\mu}(\text{\AA})$	Bond population
									$n^{\mu}$		$P^{\mu}$
Ti <sub>2</sub> BC	С	1.48	3.24	0.00	4.72	-0.72		B-B	2	1.78777	1.47
	В	1.02	2.62	0.00	3.63	-0.63		C-Ti(I)	4	2.07903	0.25
	Ti	2.05	6.60	2.65	11.30	0.70	3.30	C-Ti(II)	4	2.10977	0.24
	Ti	2.13	6.62	2.59	11.34	0.66	3.34	C-Ti(III)	4	2.19435	1.51
								B-Ti(I)	4	2.32277	0.89
								B-Ti(II)	4	2.32689	-0.13
								Ti-Ti(I)	2	2.89955	-0.25
	С	1.50	3.29	0.00	4.79	-0.79		B-B	2	1.87731	1.42
Zr <sub>2</sub> BC	В	1.05	2.59	0.00	3.63	-0.63		C-Zr(I)	4	2.29162	0.35
	Zr	2.13	6.47	2.68	11.29	0.71	3.29	C-Zr(II)	4	2.29402	0.30
	Zr	2.20	6.50	2.60	11.29	0.71	3.29	C-Zr(III)	4	2.36560	1.42
								B-Zr(I)	4	2.49658	-0.03
								B-Zr(II)	4	2.52756	0.94
								Zr-Zr	2	3.16471	-0.19
Hf <sub>2</sub> BC	С	1.56	3.32	0.00	4.89	-0.89		B-B	2	1.9007	1.43
	В	1.18	2.64	0.00	3.81	-0.81		C-Hf(I)	4	2.27847	0.41
	Hf	0.35	-0.05	2.81	3.11	0.89	3.11	C-Hf(II)	4	2.30038	0.48
	Hf	0.43	0.02	2.73	3.19	0.81	3.19	C-Hf(III)	4	2.36171	1.51
								B-Hf(I)	4	2.49895	-0.12
								B-Hf(II)	4	2.50748	1.14
								Hf-Hf	2	3.12378	-0.24
W <sub>2</sub> BC	С	1.38	3.30	0.00	4.68	-0.68		B-B	2	1.82386	1.20
	В	0.92	2.55	0.00	3.47	-0.47		C-W(I)	4	2.10329	0.14
	W	2.29	6.50	4.70	13.49	0.51	5.49	C-W(II)	4	2.16423	0.22
	W	2.36	6.49	4.51	13.36	0.64	5.36	C-W(III)	4	2.19304	1.54
								B-W(I)	4	2.31895	0.98
								W-W	2	2.95597	-0.49
								C-C	2	2.96839	-0.19

**Table S4:** Mulliken atomic and bond overlap populations of  $A_2BC$  (A = Ti, Zr, Hf and W) compounds. The EVC stands for effective valence charge.

## References

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