Electronic supplementary information (ESI)

Stability & Durability of Self-Driven Photo-detective Parameters based on $\text{Sn}_{1-\beta}$ **Sb**_β**Se** (β = 0, 0.05, 0.10, 0.15, 0.20) **Ternary** Alloys **Single Crystals**

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Sample $Sn_{\beta}Sb_{1-\beta}Se$	Compound		Crystal Growth Process			
	Temperature	Time (h)	Temperature	Heating rate		
	(K)		Source Zone	Growth Zone	(k/h)	
$\beta = 0$	1073	160	1073	1023	20	
$\beta = 0.05$	1093	160	1093	1043	20	
$\beta = 0.10$	1093	160	1093	1043	20	
$\beta = 0.15$	1093	160	1093	1043	20	
$\beta = 0.20$	1093	160	1093	1043	20	

Table. S1.The growth parameters fixed for crystal growth of Sn_{1-8} Sb_B Se ($\beta = 0, 0.05, 0.10,$) 0.15, 0.20)

Table. 1 presents the growth parameters of the Sn β Sb_{1-β}Se (β = 0, 0.05, 0.10, 0.15 and 0.20) crystals. Which confirms the high temperature and prolonged time durations of the growth of materials.

Figure. S1. (a) The XPS full range spectra of SnSe crystal for (b) Se 3d3/2 & 3d5/2 (c) Se 3p3/2 & 3p5/2 and (d) Sn 3d5/2 & 3d3/2.

Figure 1a indicates the full range spectra of the SnSe ternary alloys. Where binding energy peaks of Se and Sn are observed to be matching with the reported articles. Figure 1b indicates the peaks at \sim 496 eV and \sim 487 eV correspond to Sn 3d3/2 and Sn 3d5/2 doublets, which is associated with the Sn+4 oxidation state. The peak at ~160.84 eV and ~166.45 eV corresponds to Se 3p3/2 and 3p1/2 doublet state depicted in Fig. 3c. While Figure 1d indicates the singlet peaks of Se $3d3/2$ and Se $3d5/2$ at \sim 56... eV and \sim 55... eV corresponding to 3d5/2 and 3d3/2, respectively, which is attributed to the Sn+4 oxidation state.¹

Figure. S2. The XPS Spectra of Sb0.05Sn0.95Se crystal.

Figure. S3. The XPS Spectra of Sb0.10Sn0.90Se crystal.

Figure. S4. TheXPS Spectra of Sb0.15Sn0.85Se crystal.

Figure. S5. TheXPS Spectra of Sb0.20Sn0.8Se crystal.

(1)

Figure 2 to 5 show the XPS spectra obtained to analyze chemical states and to confirm the formation of the prepared Sn Sb Se ternary alloys.

Figure 2 to 5 (b) indicates the peaks of Sb at **~540 eV, ~535 eV** and ~530 eV correspond to Sb 3d3/2, Sb+3 and Sb 3d5/2 (tabulated in manuscript table 1), which is associated with the Sb+3 oxidation state. Other peak of O 1s presents oxygen states.

XRD equations:

The structural parameters including crystallite size (τ) for the most preferred orientation were assessed using the 'Debye-Scherrer' formula given in below:

$$
D = \frac{0.9 \lambda}{\beta_{2\theta} \cos \theta}
$$

Where ' λ ' is the wavelength of the X-Ray source, ' $\beta_{2\theta}$ ' is the broadening of the diffraction line measured at half of its maximum intensity (FWHM) and 'θ' is the Bragg's angle in degrees.

The crystallite size was calculated by resolving the most intense peak from the XRD graphs. The micro-strain evaluated in the crystals was evaluated from the equation:

$$
\varepsilon = \frac{\beta_{2\theta} \cos \theta}{4}
$$
 (2)

Where ' $\beta_{2\theta}$ ' is the broadening of the diffraction line measured at half of its maximum intensity (FWHM) and θ is Bragg's angle in degrees.

The dislocation density (δ) defined as the length of dislocation lines per unit volume of the crystals is calculated for all planes using the equation:

$$
\delta = \frac{15 * \varepsilon}{a * D} \tag{3}
$$

Where 'ε' is the micro-strain, obtain from further equations, a is the lattice parameters and 'D' is the crystallinity size obtained from the 'Debye-Scherrer' formula.2,3

Sample $Sn1$. $_{\beta}Sb_{\beta}Se$	Lattice Parameter				Volume		2θ		Crystallite	Micro Strain ε	Dislocation Density $\delta \times$
	A	\bf{B}	$\mathbf C$	c/a	$(\AA)^3$	FWHM	(Degree)	d	size t (nm	$(lin-2m-4)$	10^{14} $(\text{lin } m-4)$
$\beta = 0$	11.490	4.440	4.135	0.35988	210.95	0.4000	31.074	0.287572	35.964	0.09635	3.507
$\beta = 0.05$	11.485	4.435	4.130	0.35960	210.366	0.4100	31.095	0.287446	35.088	0.09875	3.684
$\beta = 0.10$	11.480	4.430	4.125	0.35932	209.783	0.4200	31.108	0.287389	34.254	0.10115	3.868
$\beta = 0.15$	11.475	4.425	4.125	0.35948	209.455	0.4300	31.112	0.287236	33.458	0.10356	4.055
$\beta = 0.20$	11.460	4.410	4.110	0.35864	207.714	0.4500	31.138	0.287001	14.387	0.10837	9.862

Table.S3. TheXRD parameters of $Sn_{1-\beta}$ Sb_β Se ($\beta = 0, 0.05, 0.10, 0.15, 0.20$) crystals.

Figure. S6. The optical microscopic images of $\text{Sn}_{1-\beta}$ Sb_β Se ($\beta = 0, 0.05, 0.10, 0.15, 0.20$) crystals.

Figure 5 shows the optical microscopic images of the Sn Sb Se ternary alloys. Here fig. 1a shows the layered surfaces, 1b shows the flat region of as-grown crystals, 1c shows the layered surfaces, 1d shows the many layered grown crystals and 1e shows the layered surface of the as-grown crystals. There are cleaved surfaces to show the layered and flat region to fabricate the devices of these samples.

UV VIS equations:

The optical absorption (α) was calculated using lambert law as shown in the equation:

$$
\ln\left(\frac{I_0}{I}\right) = 2.303 A = \alpha t
$$
\n
$$
\alpha = \frac{2.303 * A}{t}
$$
\n(4)

(5)

Where, α is the absorption, I₀ and I are the intensity of incident and transmitted light, respectively t is the thickness of the crystals and A is the optical absorbance.

The band gap of SnSe thin film can be determined by the below equation:

$$
(\alpha h v)^n = A (h v - E_g) \tag{6}
$$

Where E_g is the band gap, A is the constant that comes from the Fermi-Golden rules, and n depends on the type of the band transition of the materials.

The indirect band gap was calculated using the Davis-mott model (for $n = \frac{1}{2}$). The indirect energy gap is given generally by the below equation:

$$
E_g = \frac{E_1 + E_4}{2} = \frac{E_2 + E_3}{2}
$$

$$
(9)
$$

The phonon energies are given below equations:

$$
E_{p1} = \frac{E_4 - E_1}{2} \text{ and } E_{p2} = \frac{E_3 - E_2}{2}
$$
 (10)

where E'g is the energy band gap due to absorption and emission of phonon energies, and Ep1 and Ep2 are the absorbed and emitted phonon energies.

The steepness constant is determined usually using the equation shown below

$$
\sigma = \frac{K_B T}{E_U} \tag{11}
$$

where ' K_B ' is the Boltzmann constant, ' E_U ' the Urbach energy, and 'T' the temperature in Kelvin (K) .³⁻⁷

Figure. S7. (a-j) The energy stack and urbach energy of $\text{Sn}_{1-\beta}$ Sb_β Se ($\beta = 0, 0.05, 0.10, 0.15,$ 0.20) crystals.

Pulse photodetectors equations:

The detector parameters such as photocurrent (I_{ph}) , sensitivity (S), responsivity (R λ), and specific detectivity (D) are evaluated using the following equations:

$$
I_{ph} = I_{ill} - I_{dark} \tag{12}
$$

$$
S = \frac{I_{ph}}{I_{dark}}
$$
 (13)

$$
R_{\lambda} = \frac{I_{ph}}{P.A}
$$
 (14)

$$
D = \frac{R_{\lambda} S^{1/2}}{\left(2eI_{dark}\right)^{1/2}}
$$
\n(15)

Where I_{ill} = current under illuminated condition and I_{dark} = current under dark condition) is the photocurrent, P is the illumination intensity, A is the effective area of the photodetector, λ is the wavelength of the light source, h is Planck's constant, e is the elementary electronic charge (1.6 x 10^{19} C), c is the speed of light in vacuum, k is the wavelength of incident radiation, and I_{dark} is the dark current.^{8–10}

References

- 1 C. Yamamoto, X. He, T. Katase, K. Ide, Y. Goto, Y. Mizuguchi, A. Samizo, M. Minohara, S. Ueda, H. Hiramatsu, H. Hosono and T. Kamiya, *Adv Funct Mater*, , DOI:10.1002/adfm.202008092.
- 2 H. S. Jagani, S. U. Gupta, K. Bhoraniya, M. Navapariya, V. M. Pathak, G. K. Solanki and H. Patel, *Mater Adv*, , DOI:10.1039/d1ma01005k.
- 3 K. Patel, G. Solanki, K. Patel, V. Pathak and P. Chauhan, *European Physical Journal B*, , DOI:10.1140/epjb/e2019-100306-8.
- 4 V. Dixit, S. Nair, J. Joy, C. U. Vyas, G. K. Solanki, K. D. Patel and V. M. Pathak, *Mater Sci Semicond Process*, 2018, **88**, 1–9.
- 5 D. A. Dholakia, Solanki. G. K., S. G. Patel and M. K. Agarwal, *SCIENTIA IRANICA*, 2003, **10**, 373– 382.
- 6 A M Elkorashy, *J. Phys. Chem. Solids.* .
- 7 V. R. K. P. G. V.M. Koshkin, *Sov. Phys. Semicond.*
- 8 A. Patel, K. Patel, C. Limberkar, K. D. Patel, G. K. Solanki and V. M. Pathak, *Journal of Materials Science: Materials in Electronics*, 2022, **33**, 8417–8425.
- 9 K. Patel, P. Chauhan, A. B. Patel, G. K. Solanki, K. D. Patel and V. M. Pathak, *ACS Appl Nano Mater*, 2020, **3**, 11143–11151.
- 10 P. Chauhan, A. B. Patel, G. K. Solanki, H. K. Machhi, C. K. Sumesh, S. S. Soni, V. Patel and V. M. Pathak, *Journal of Physical Chemistry C*, 2021, **125**, 14729–14740.