

Supporting Information

Effect of Temperature on Structural, Dynamical, and Electronic Properties of Sc_2Te_3 from First-Principles Calculations.

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Table S1 Average coordination numbers of different atomic pairs in the Sc_2Te_3 system at 300 and 773 K.

		With Sc	With Te	Total
300 K	Sc	0.003	5.634	5.637
	Te	3.756	0.003	3.759
773 K	Sc	0.029	5.346	5.375
	Te	3.564	0.026	3.590

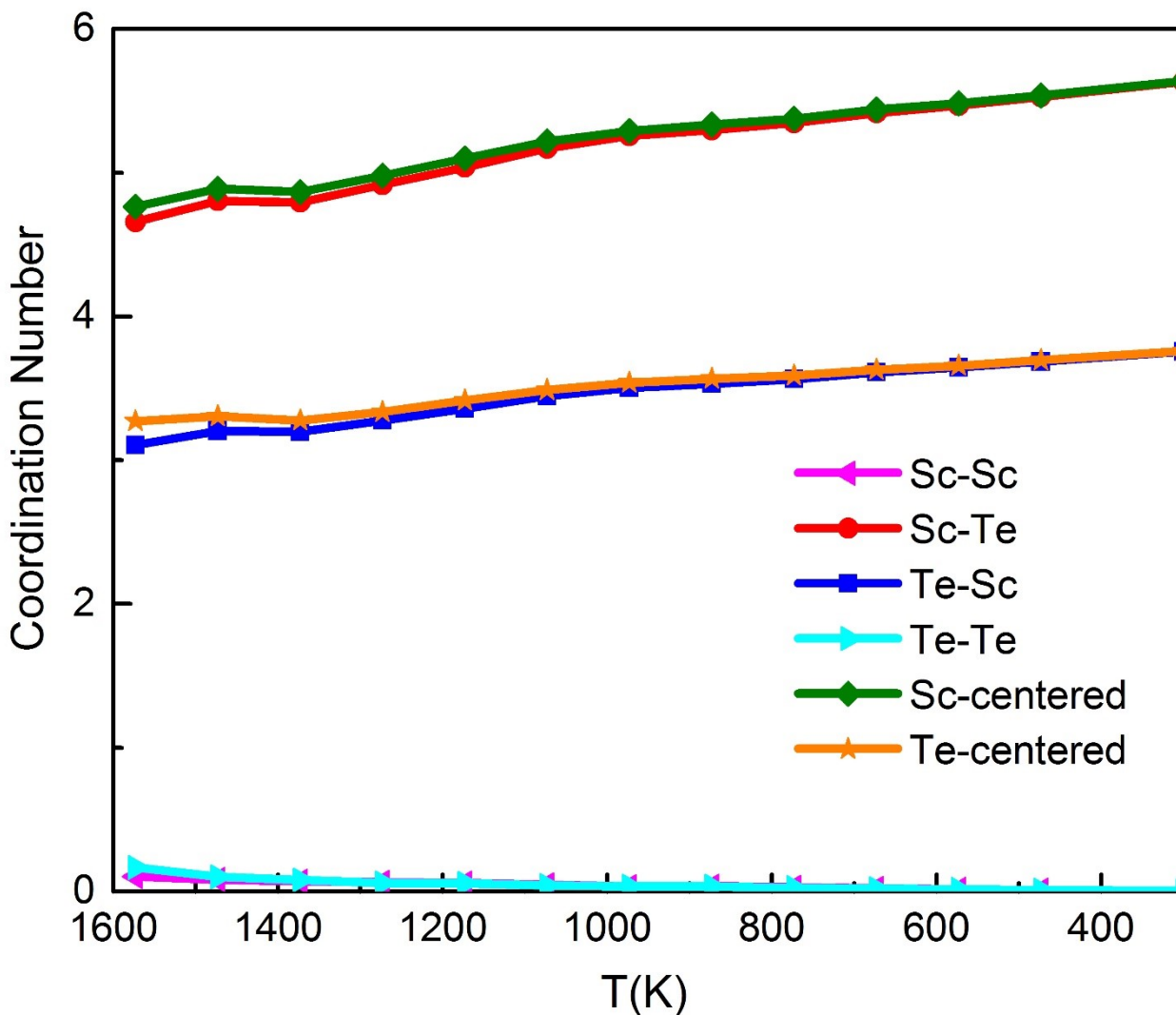


Figure S1 The A-B partial coordination number of the Sc_2Te_3 system as a function of temperatures. Here, A is the central atom and B is the coordination atom. The green and orange solid lines labeled as Sc-centered and Te-centered represent the total CNs of the Sc and Te atoms, respectively. The pink, red, blue, and cyan solid lines are labeled as Sc-Sc, Sc-Te, Te-Sc, and Te-Te partial CNs, respectively. A uniform cutoff distance of 3.3 Å is used.

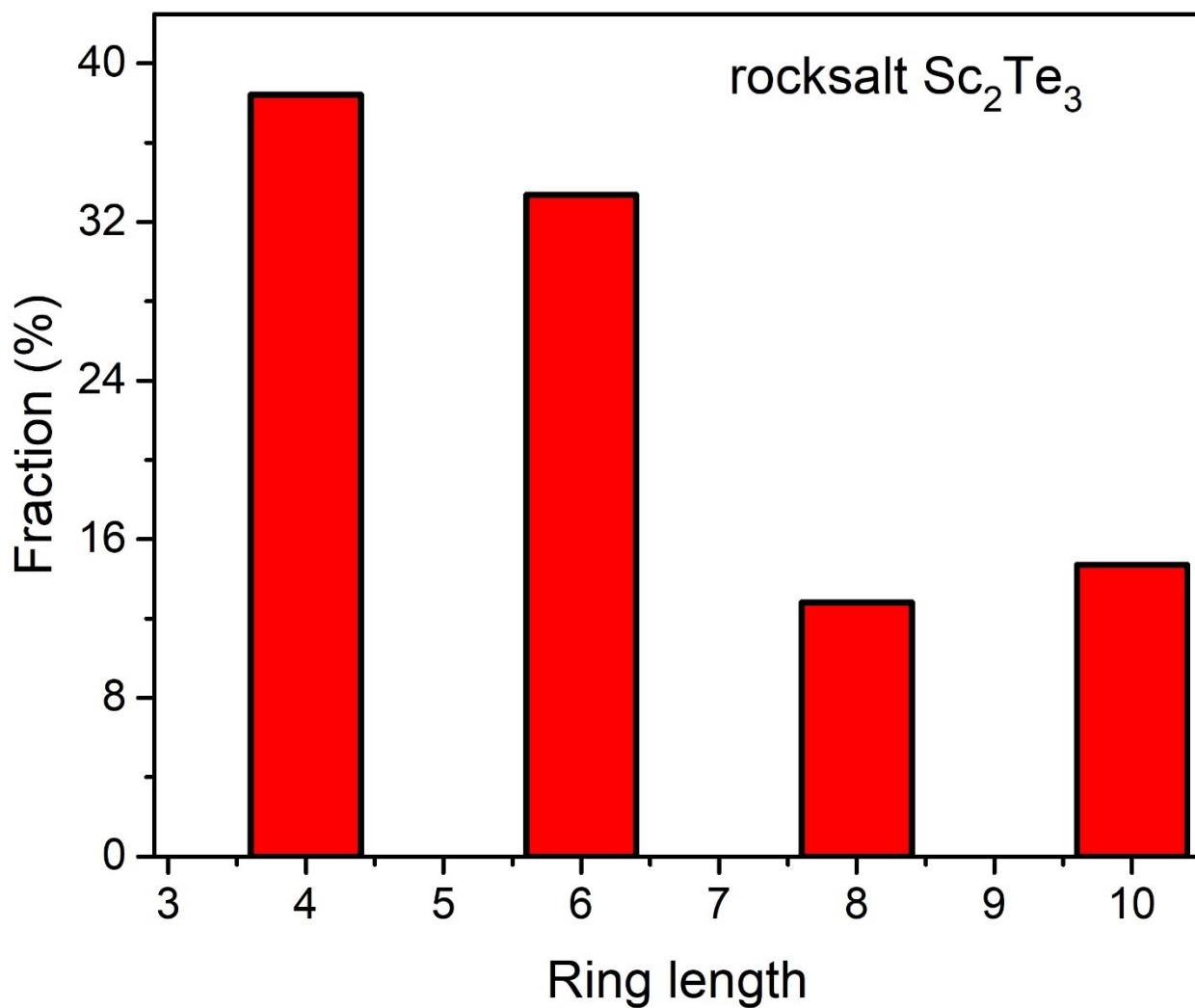


Figure S2 The primitive ring distributions in the cubic rocksalt Sc_2Te_3 structure.

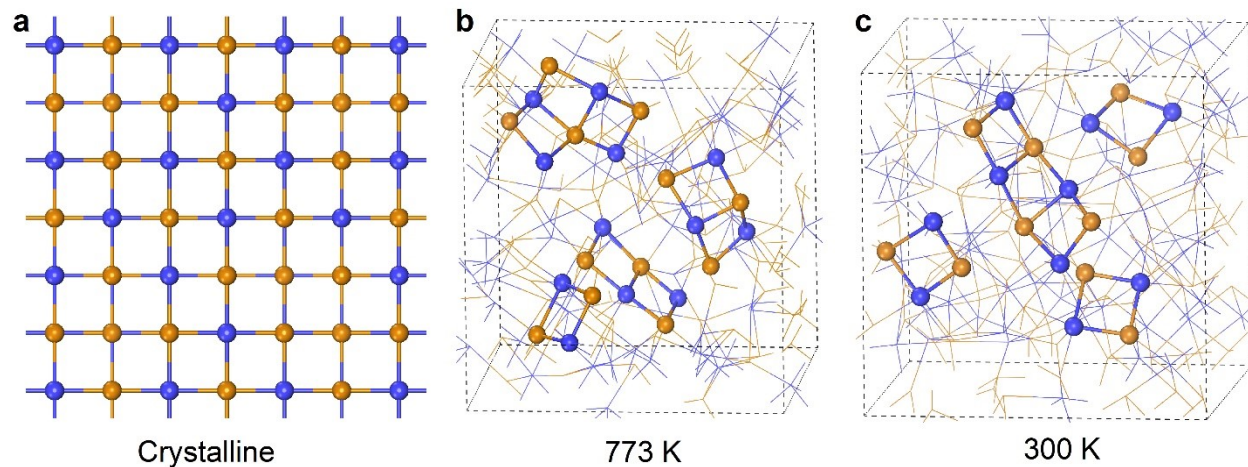


Figure S3 (a) The rocksalt-like crystal structures of Sc_2Te_3 with $1/3$ vacancies on the cationic sub-lattice. Four-membered rings as the defining structural motifs are displayed. (b)-(c) The snapshots of typical four-membered primitive rings (ABAB squares) in the Sc_2Te_3 system at 773 K and 300 K, respectively, resembling the crystalline phase. The Sc and Te atoms are rendered in blue and orange balls, respectively.

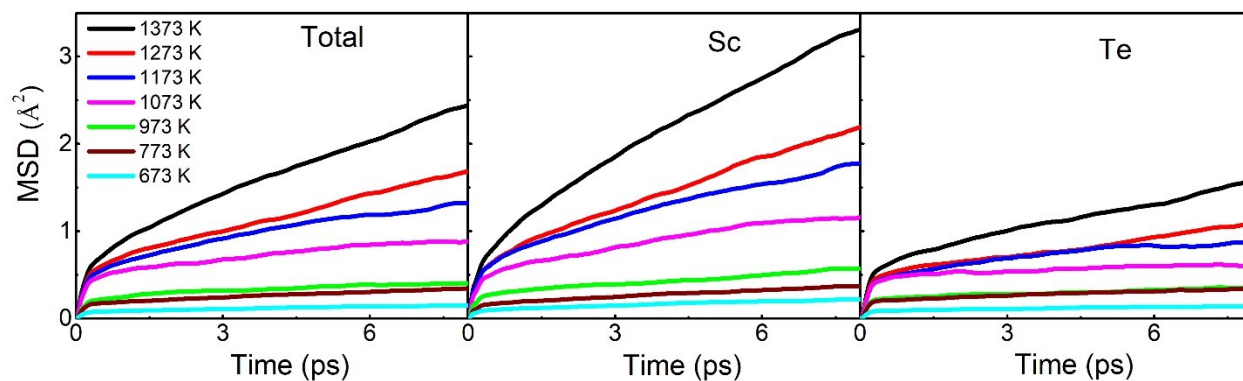


Figure S4 The Mean-squared displacement (MSD) of the Sc_2Te_3 system as a function of time at different temperatures.

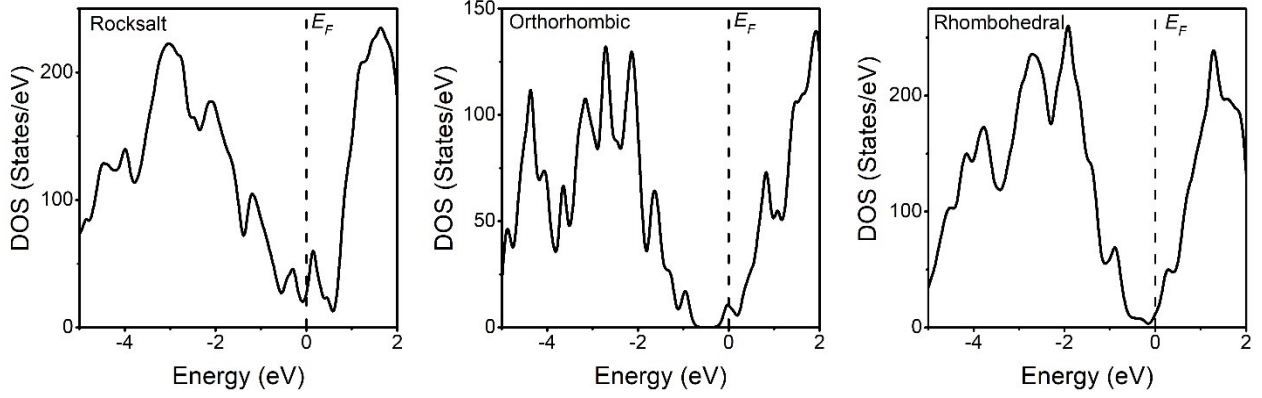


Figure S5 The total density of states (DOS) of crystalline phases of Sc₂Te₃ using the mBJLDA functional. (a)-(c) The total DOSs of the rocksalt, orthorhombic, and rhombohedral phases of the Sc₂Te₃ structure, respectively. The dashed vertical line is the Fermi level (E_F), corresponding to the zero point energy.