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Enhancement of photovoltaic performance in ferrocenyl π -extended multi donor- π acceptor (D-D'- π -A) dyes using chenodeoxycholic acid as a dye co-adsorbent for dye sensitized solar cells

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Figure S2. $^{13}\mathrm{C}$ NMR spectrum of 3, 3 (4-methoxyphenyl) ferrocenyl acrylaldehyde in CDCl₃ at 25 °C





Figure S3. ¹H NMR spectrum of dye 1 in CDCl₃ at 25 $^{\circ}$ C













Figure S12. Theoretically calculated absorption spectra for dyes **1** and **2**. The absorption spectra were obtained by TD-DFT calculation with B3LYP/6-31+G(d,p)/LanL2TZf level of theory. The spectra were visualized in Gauss View 6.1.

Table	S1 .	Selected	transitions	obtained	from	TD-DFT	calculation	with	B3LYP/6-
31+G(d	l,p)/L	anL2TZf le	evel of theory	7.					

Entry	λ	Oscillator	Energy	Selected Major Transitions ^a
	(nm)	strength, f	(eV)	
	497	0.4317	2.49	$H-2 \rightarrow L (79\%)$
	441	0.2014	2.81	$H-4 \rightarrow L (32\%)$
Dye 1	699	0.0580	1.77	$H \rightarrow L (86\%)$
	350	0.0314	3.54	$H-6 \rightarrow L (39\%)$
	531	0.0261	2.33	$H-3 \rightarrow L (56\%)$
	484	0.0175	2.55	$H \rightarrow L+3 (32\%)$
	513	0.5953	2.41	$H-2 \rightarrow L (65\%)$
	431	0.2158	2.87	$H-3 \rightarrow L (63\%)$
	640	0.1014	1.93	$H \rightarrow L (58\%)$
	356	0.0745	3.47	$H \rightarrow L+1 (84\%)$
Dye 2	406	0.0290	3.06	$H-4 \rightarrow L (54\%)$
	339	0.0198	3.65	$H-6 \rightarrow L (43\%)$
	485	0.0100	2.55	$ \text{H-1} \rightarrow \text{L+6} (32\%)$
	661	0.0068	1.87	$ \text{H-1} \rightarrow \text{L} (71\%)$

^a H = HOMO; L = LUMO; only contributions above 10% are included.

Table S2. Density surfaces of the HOMO-2 orbitals frontier involved in electronic of transitions chromophores and 2 which is 1 derived from B3LYP/6-

31+G(d,p)/LanL2TZf level of theory using isosurface value of 0.02 au.

Orbitals	Dye 1	Dye 2
НОМО-5		
НОМО-4		
НОМО-3		

HOMO-1	ن ن ن ن ن ن ن ن ن ن ن ن ن ن ن ن ن ن ن	
НОМО		
LUMO		A Contraction
LUMO+1		A CONTRACT
LUMO+2		
LUMO+3		
LUMO+4		

