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## Supporting Information

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**Crystal facet and Na-doping dual engineering of bismuth**

4

**oxychloride ultrathin nanosheets with efficient oxygen activation**

5

**for enhanced photocatalytic performance**

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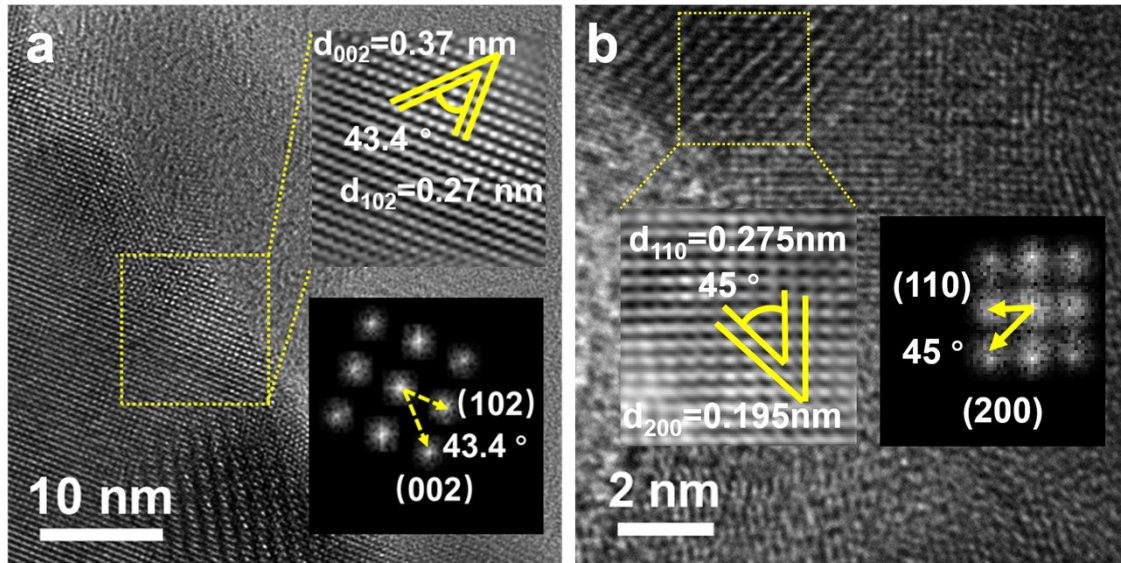
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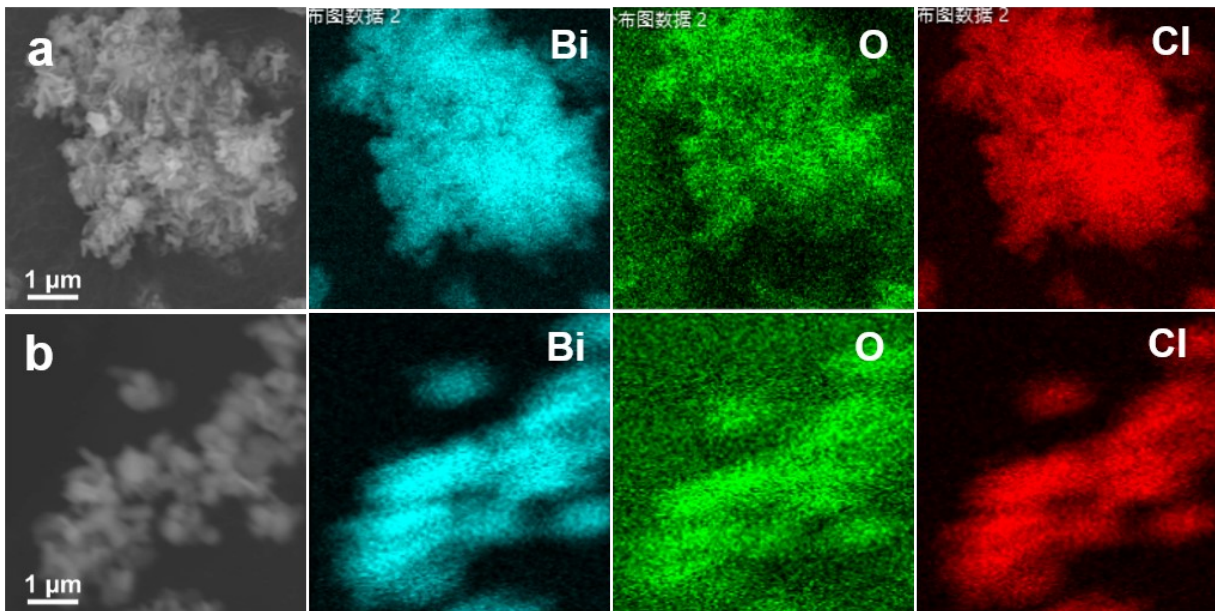


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23 **Fig. S1.** HRTEM images with enlarged view (inset) and SAED patterns (inset) of (a) BOC-  
 24 010 and (b) BOC-001.

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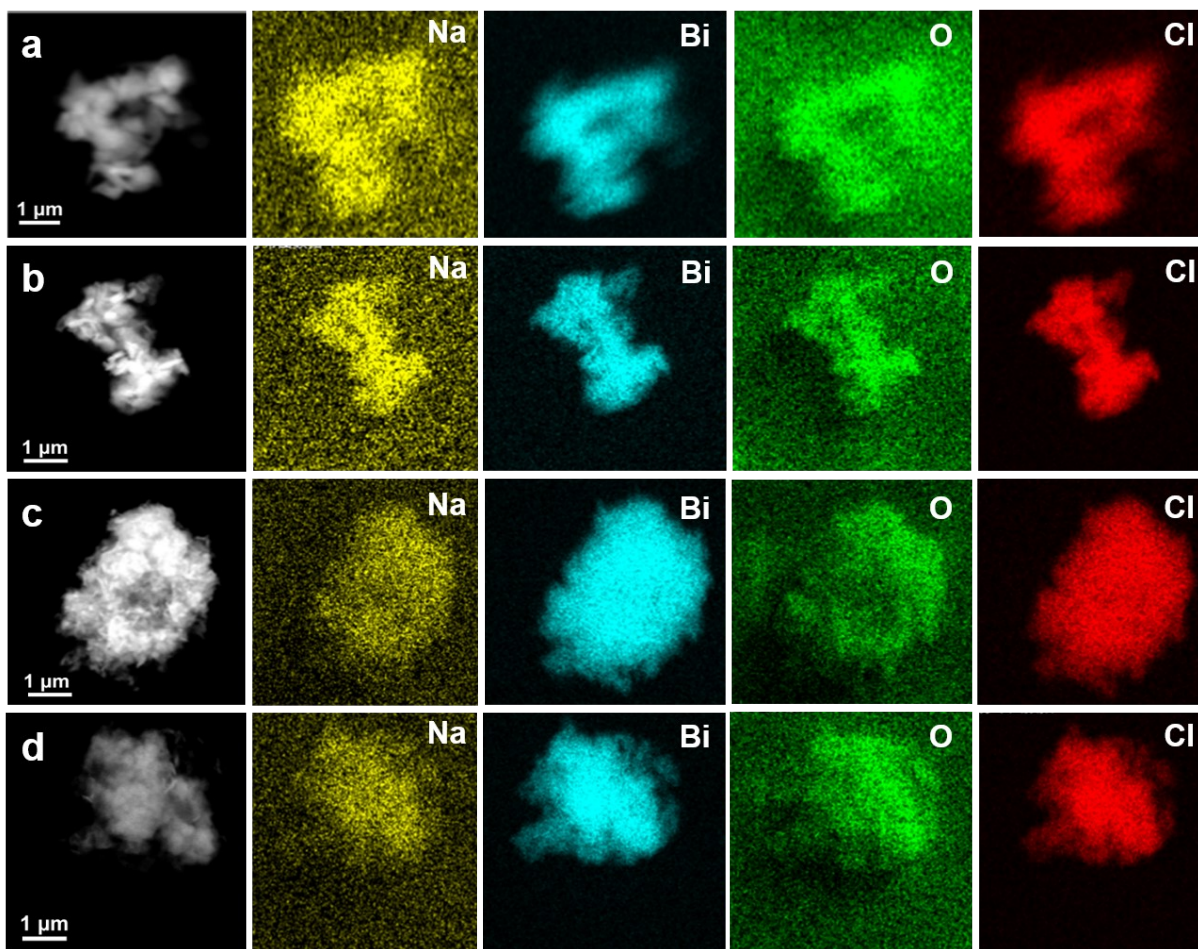
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28 **Fig. S2.** SEM images and corresponding EDS elemental mapping of (a) BOC-010 and (b)  
 29 BOC-001.

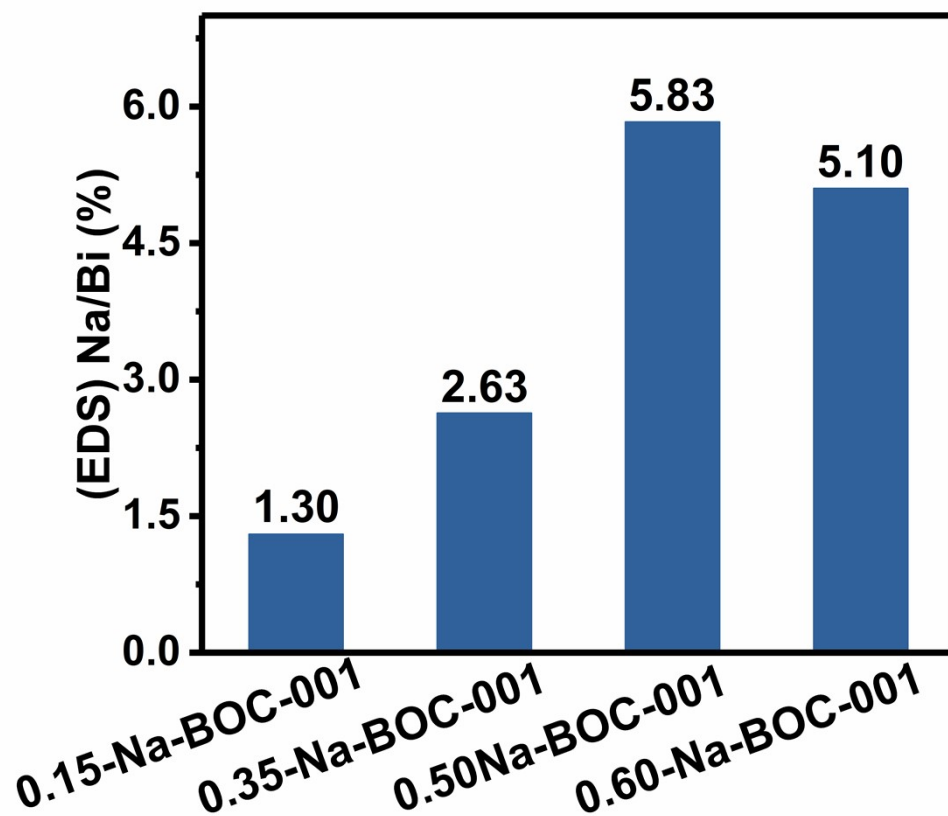
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32 **Fig. S3.** SEM images and corresponding EDS elemental mapping of (a) 0.50-Na-BOC-010, (b)  
 33 0.15-Na-BOC-001, (c) 0.35-Na-BOC-001 and (d) 0.60-Na-BOC-001.

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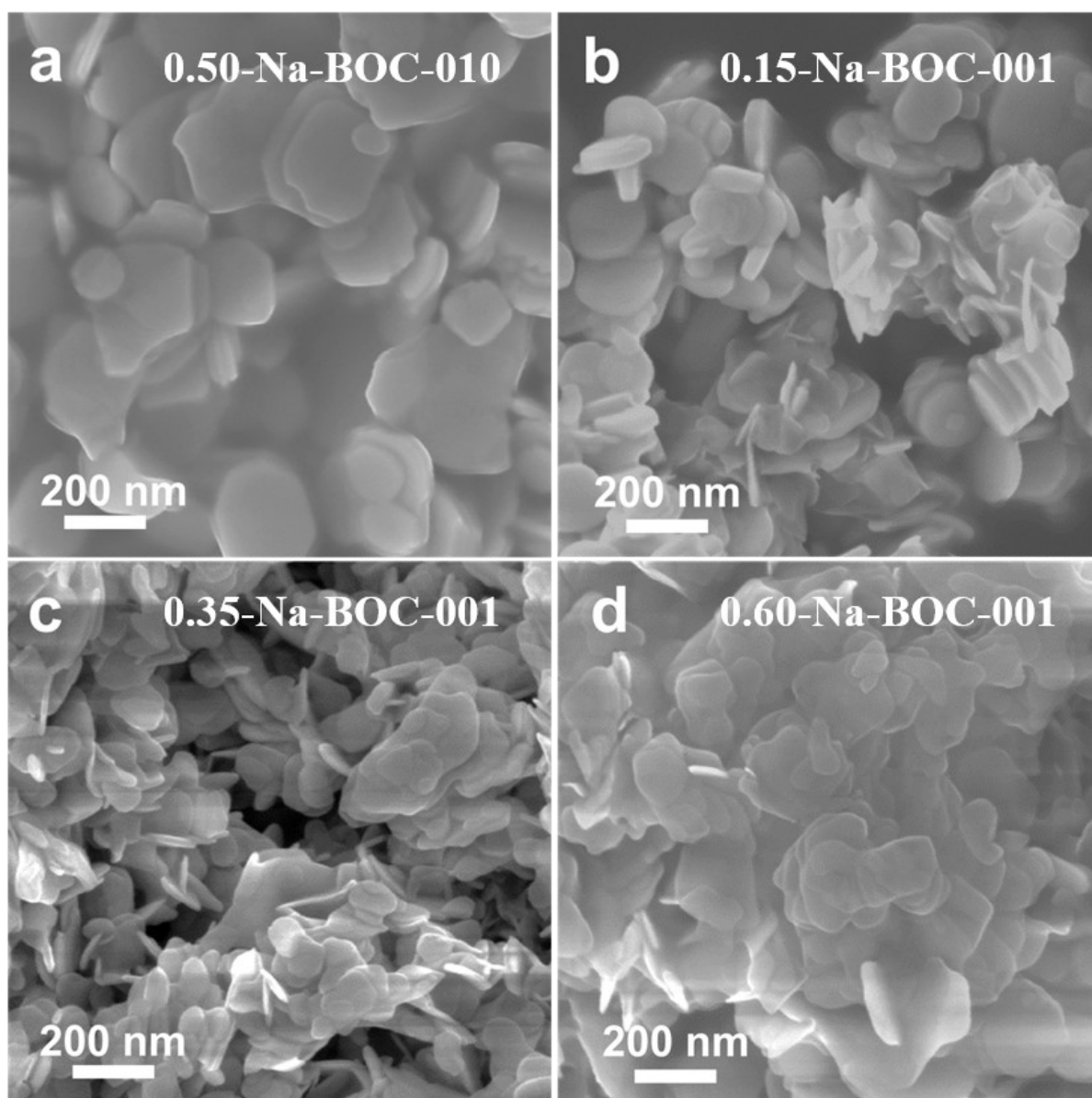


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36 **Fig. S4.** (EDS) Na/Bi ratio of 0.15-Na-BOC-001, 0.35-Na-BOC-001, 0.50-Na-BOC-001 and

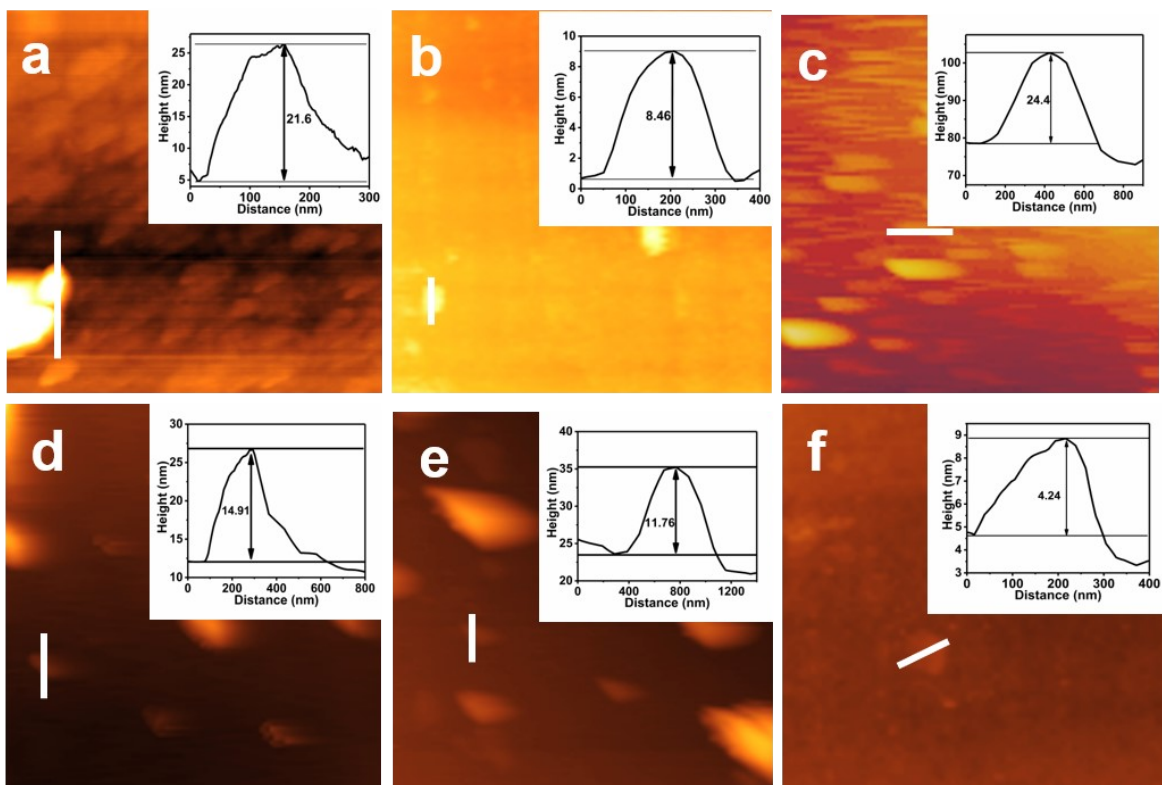
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0.60-Na-BOC-001.



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39 **Fig. S5.** SEM images of (a) 0.50-Na-BOC-010, (b) 0.15-Na-BOC-001, (c) 0.35-Na-BOC-001  
40 and (d) 0.60-Na-BOC-001.



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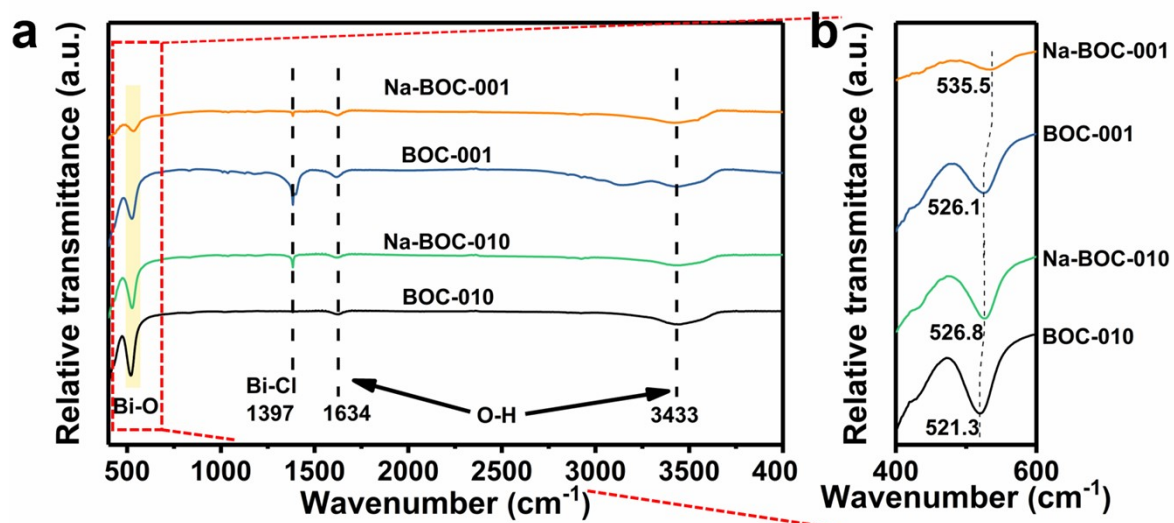
42 **Fig. S6.** AFM images and corresponding height profiles of (a) BOC-010, (b) BOC-001, (c)  
 43 0.50-Na-BOC-010, (d) 0.15-Na-BOC-001, (e) 0.35-Na-BOC-001 and 0.60-Na-BOC-001.

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49 **Fig. S7.** FTIR spectra of BOC-010, Na-BOC-010, BOC-001 and Na-BOC-001 samples.

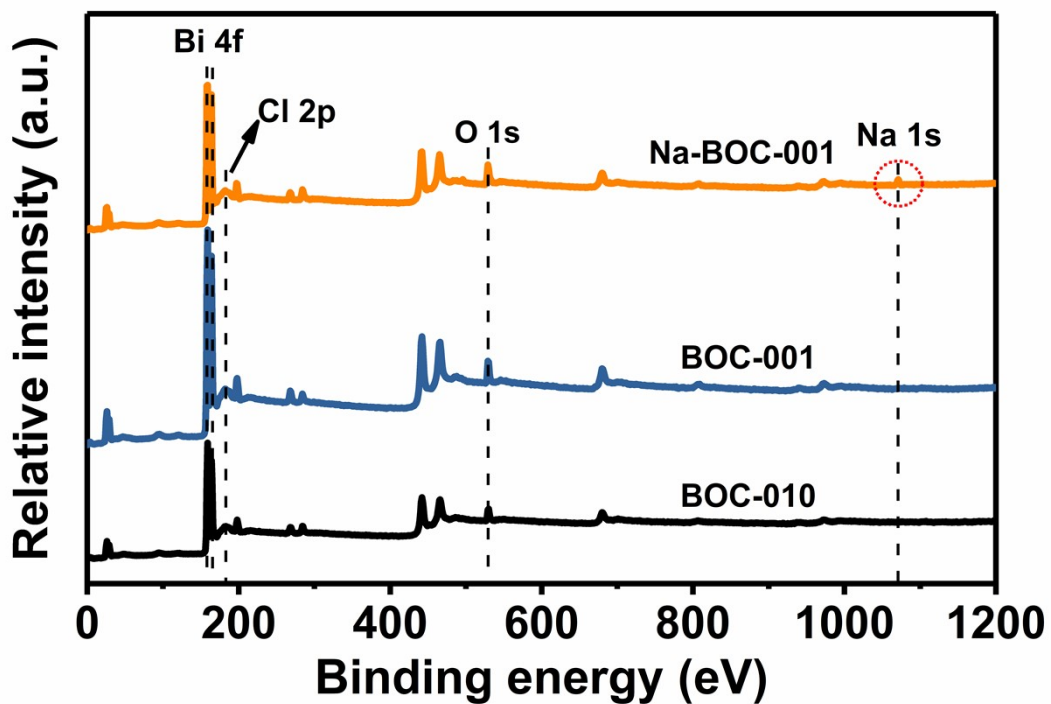
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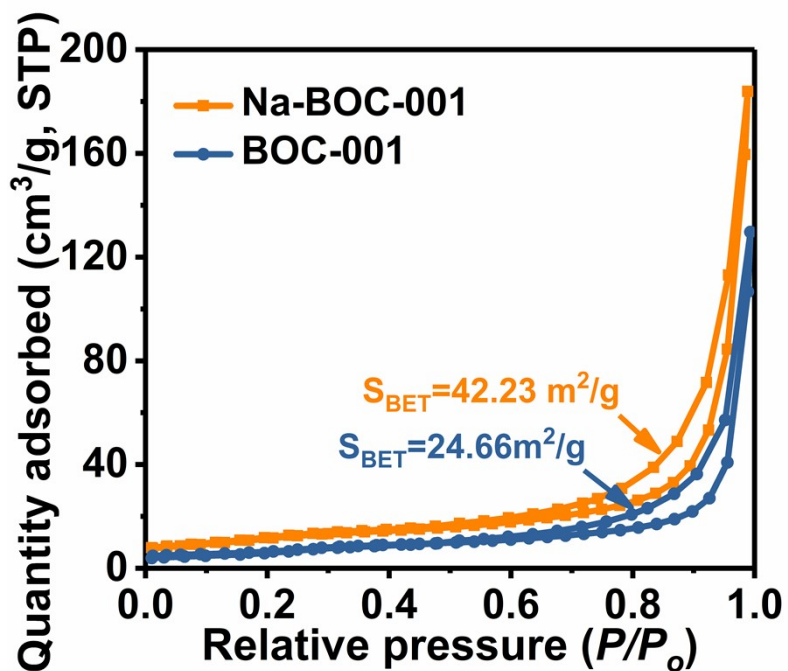
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Fig. S8. Survey XPS spectra of BOC-010, BOC-001 and Na-BOC-001.

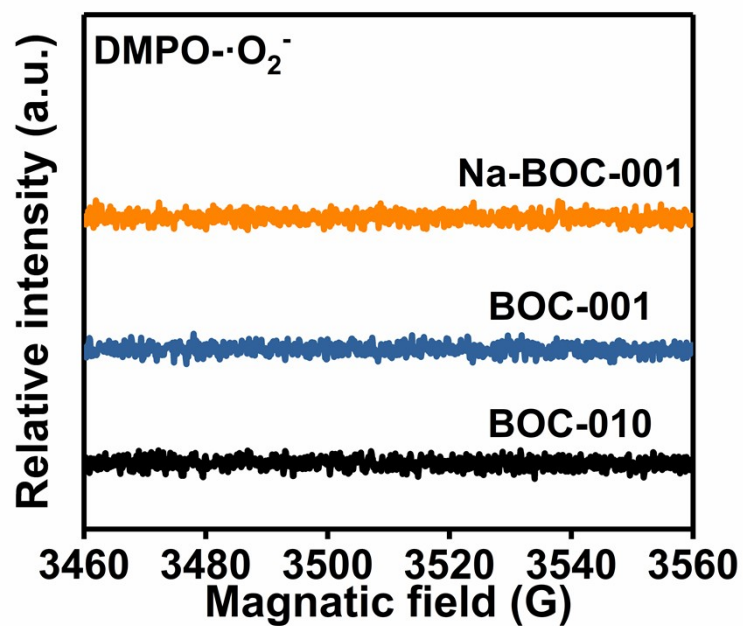


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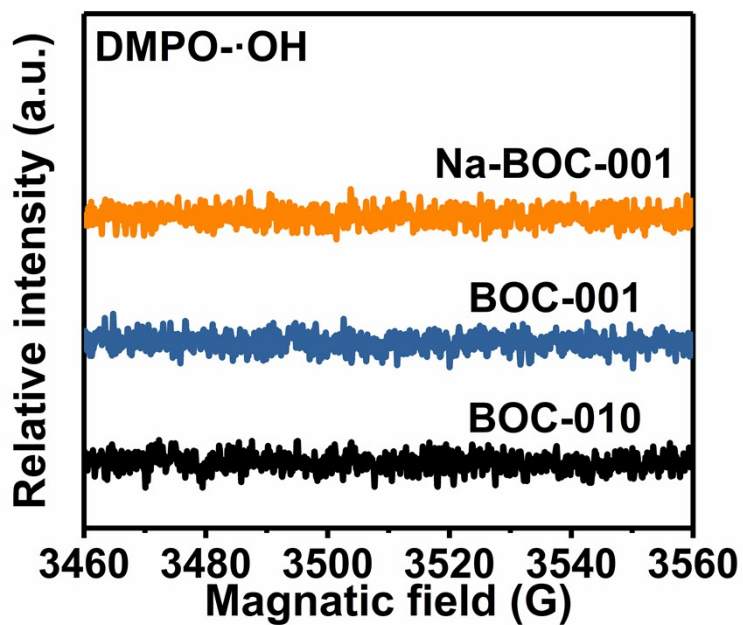
Fig. S9. N<sub>2</sub> adsorption and desorption isotherms of Na-BOC-001 and BOC-001.





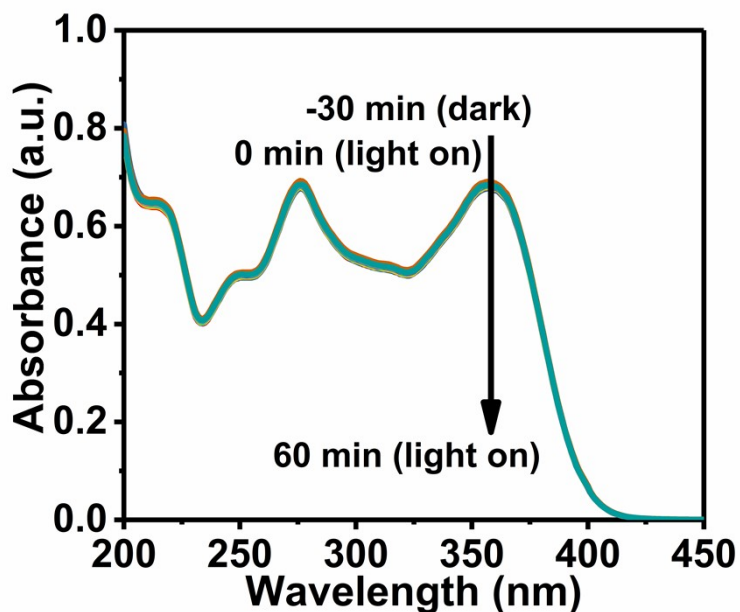
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60 **Fig. S10.** ESR spectra of DMPO-•O<sub>2</sub><sup>-</sup> over BOC-010, BOC-001, and Na-BOC-001 under  
61 darkness.



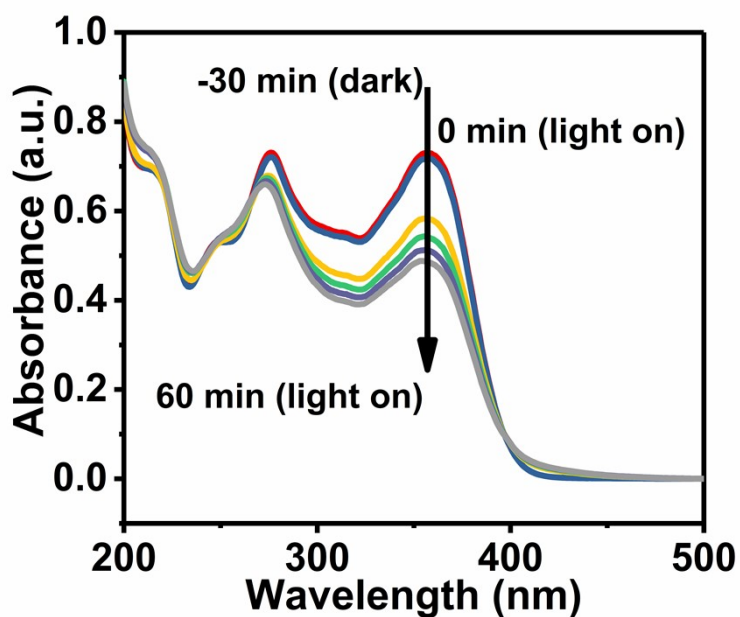
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63 **Fig. S11.** ESR spectra of DMPO-•OH over BOC-010, BOC-001, and Na-BOC-001 under  
64 darkness.



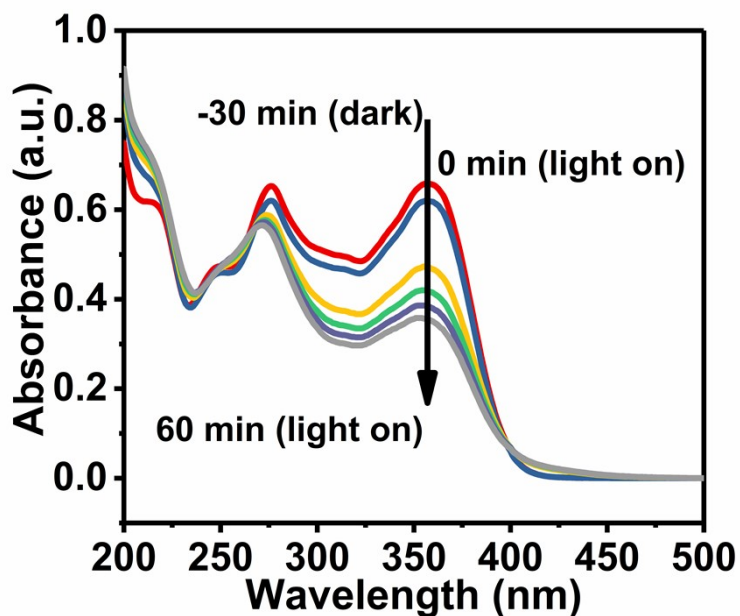
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66 **Fig. S12.** UV-vis absorption spectra of TC collected during TC degradation process **without**  
 67 **catalyst** with irradiation time (Conditions: [TC] = 20 mg L<sup>-1</sup>, [catalyst] = 0 g L<sup>-1</sup>, pH = 6.5, T  
 68 = 25 °C, visible light ( $\lambda > 420$  nm)).



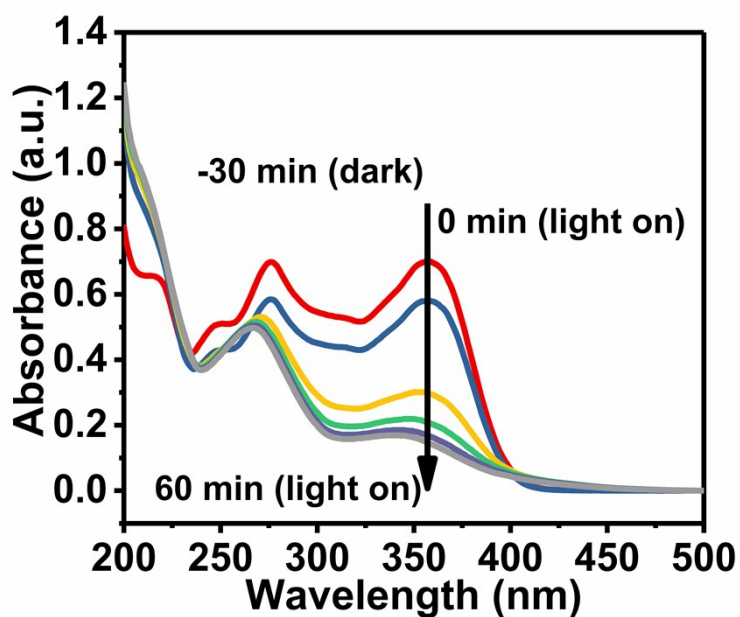
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70 **Fig. S13.** UV-vis absorption spectra of TC collected during TC degradation process over  
 71 **BOC-010** with irradiation time. (Conditions: [TC] = 20 mg L<sup>-1</sup>, [catalyst] = 0.50 g L<sup>-1</sup>, pH =  
 72 6.5, T = 25 °C, visible light ( $\lambda > 420$  nm)).



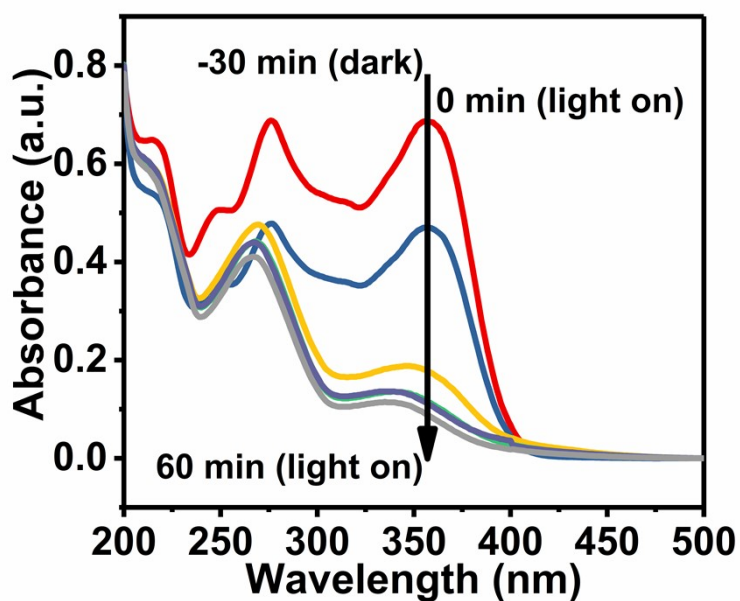
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74 **Fig. S14.** UV-vis absorption spectra of TC collected during TC degradation process over Na-  
 75 **BOC-010** with irradiation time. (Conditions: [TC] = 20 mg L<sup>-1</sup>, [catalyst] = 0.50 g L<sup>-1</sup>, pH =  
 76 6.5, T = 25 °C, visible light ( $\lambda > 420$  nm)).



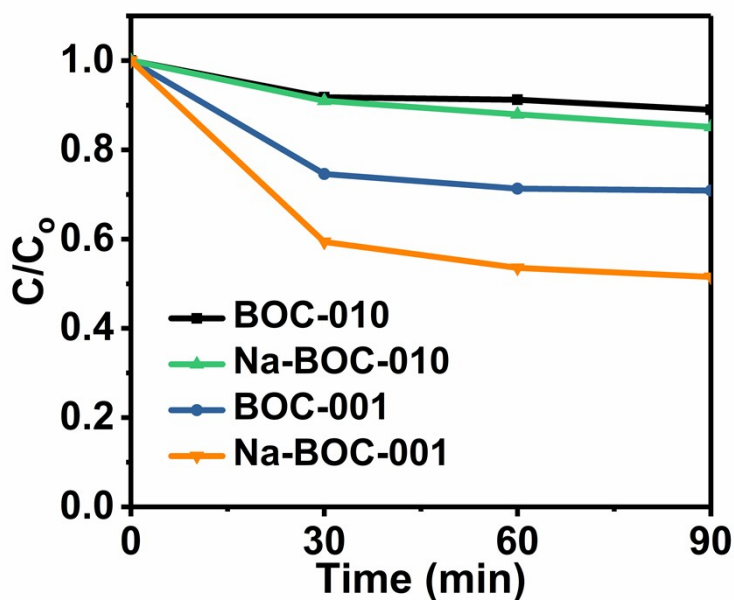
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78 **Fig. S15.** UV-vis absorption spectra of TC collected during TC degradation process over  
 79 **BOC-001** with irradiation time. (Conditions: [TC] = 20 mg L<sup>-1</sup>, [catalyst] = 0.50 g L<sup>-1</sup>, pH =  
 80 6.5, T = 25 °C, visible light ( $\lambda > 420$  nm)).



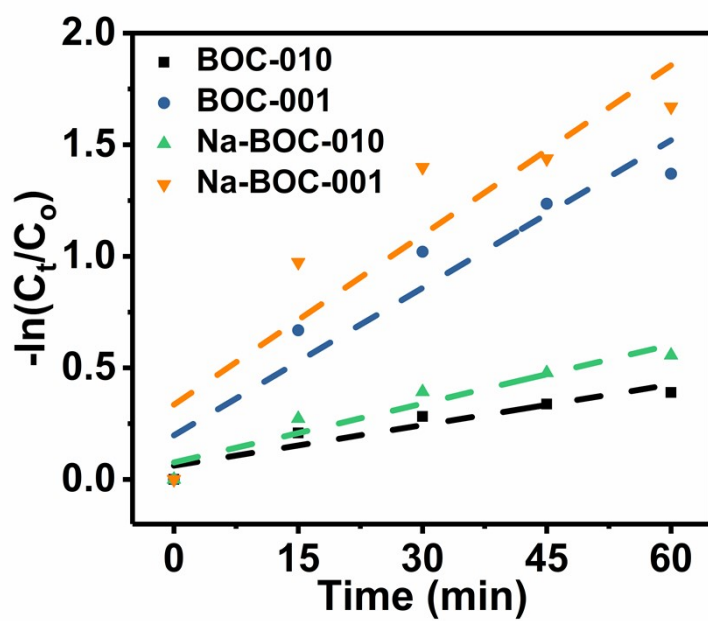
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82 **Fig. S16.** UV-vis absorption spectra of TC collected during TC degradation process over Na-  
 83 **BOC-001** with irradiation time. (Conditions: [TC] = 20 mg L<sup>-1</sup>, [catalyst] = 0.50 g L<sup>-1</sup>, pH =  
 84 6.5, T = 25 °C, visible light ( $\lambda > 420$  nm)).



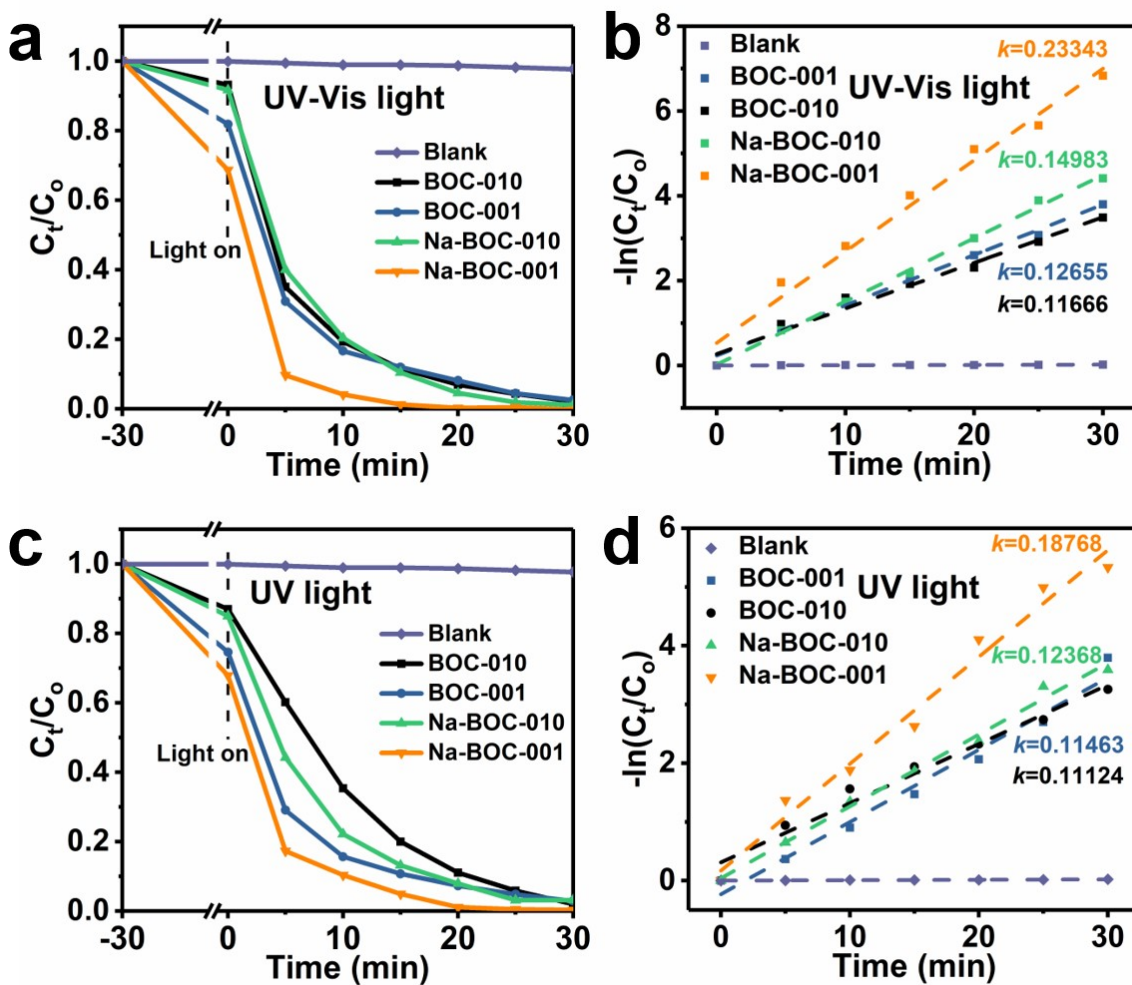
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86 **Fig. S17.** TC adsorption profiles over the samples (BOC-010, Na-BOC-010, BOC-001 and  
 87 Na-BOC-001) under dark condition. (Conditions: [TC] = 20 mg L<sup>-1</sup>, [catalyst] = 0.50 g L<sup>-1</sup>,  
 88 pH = 6.5, T = 25 °C, visible light ( $\lambda > 420$  nm)).



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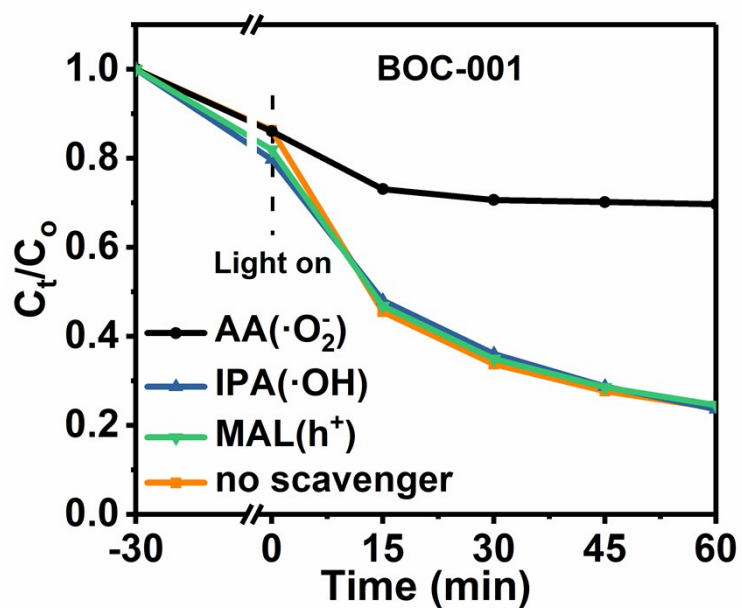
90 **Fig. S18.** Pseudo-first-order kinetic fitting curves and corresponding kinetic constants ( $k$ ) over  
 91 different samples under the irradiation of Vis light.



92

93 **Fig. S19.** (a) Photocatalytic degradation curves of TC and (b) Pseudo-first-order kinetic fitting curves and  
 94 corresponding kinetic constants ( $k$ ) over different samples under the irradiation of UV-Vis light. (c)  
 95 Photocatalytic degradation curves of TC and (d) Pseudo-first-order kinetic fitting curves and corresponding  
 96 kinetic constants ( $k$ ) over different samples under the irradiation of UV light. Conditions: [TC] = 20 mg L<sup>-1</sup>,  
 97 [catalyst] = 0.5 g L<sup>-1</sup>, pH = 6.5, T = 25 °C, UV-Vis light ( $\lambda > 320$  nm), UV light (320 <  $\lambda$  < 420 nm).

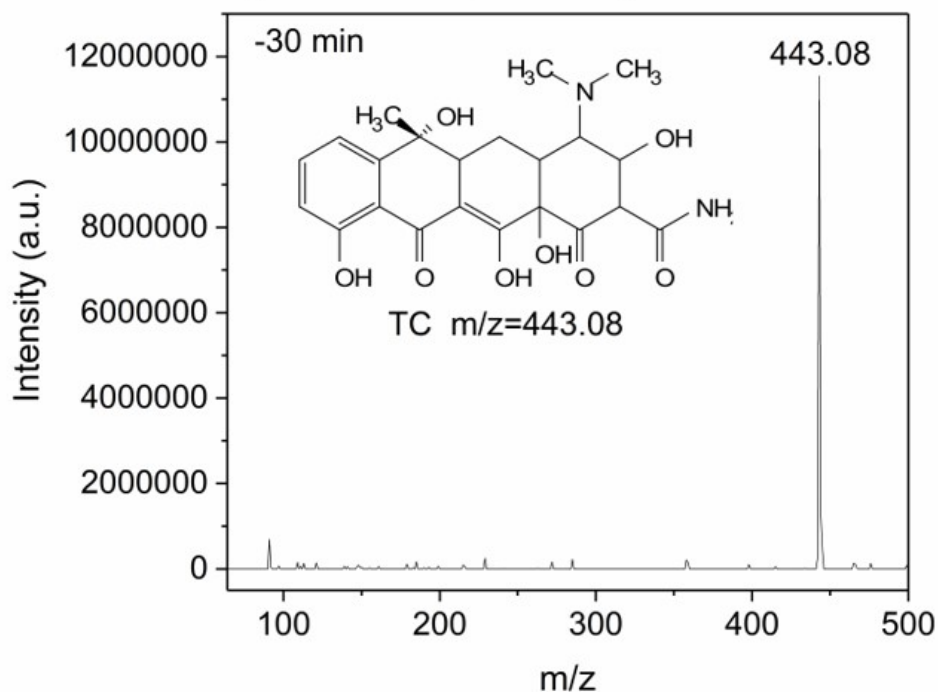
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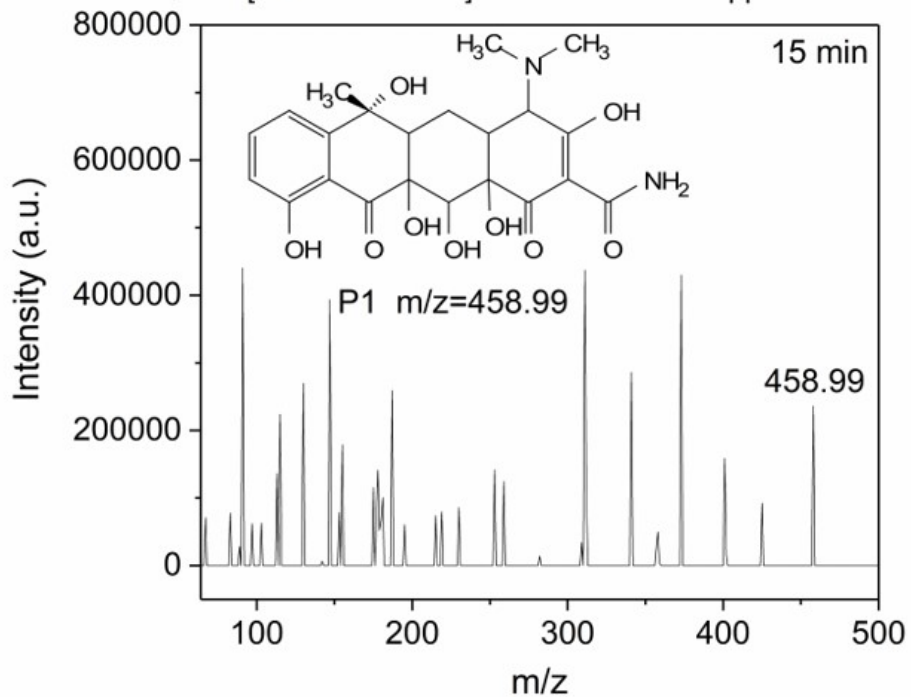
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100 **Fig. S20.** TC removal profiles over **BOC-001** with addition of various radical sacrificial  
 101 agents. Conditions:  $[\text{TC}] = 20 \text{ mg L}^{-1}$ ,  $[\text{catalyst}] = 0.5 \text{ g L}^{-1}$ ,  $\text{pH} = 6.5$ ,  $T = 25 \text{ }^\circ\text{C}$ , visible light  
 102 ( $\lambda > 420 \text{ nm}$ ).

RT: 5.11 AV: 1 NL:3.20E6 C22H24O8N2 RDB: 12.5  
T: -cEIS Q1MS [64.000-500.000] Delta ppm: 2.478



RT: 2.62 AV: 1 NL:4.53E5 C22H26N2O9 RDB: 12 Delta ppm: 0.003  
T: -cEIS Q1MS [64.000-500.000]

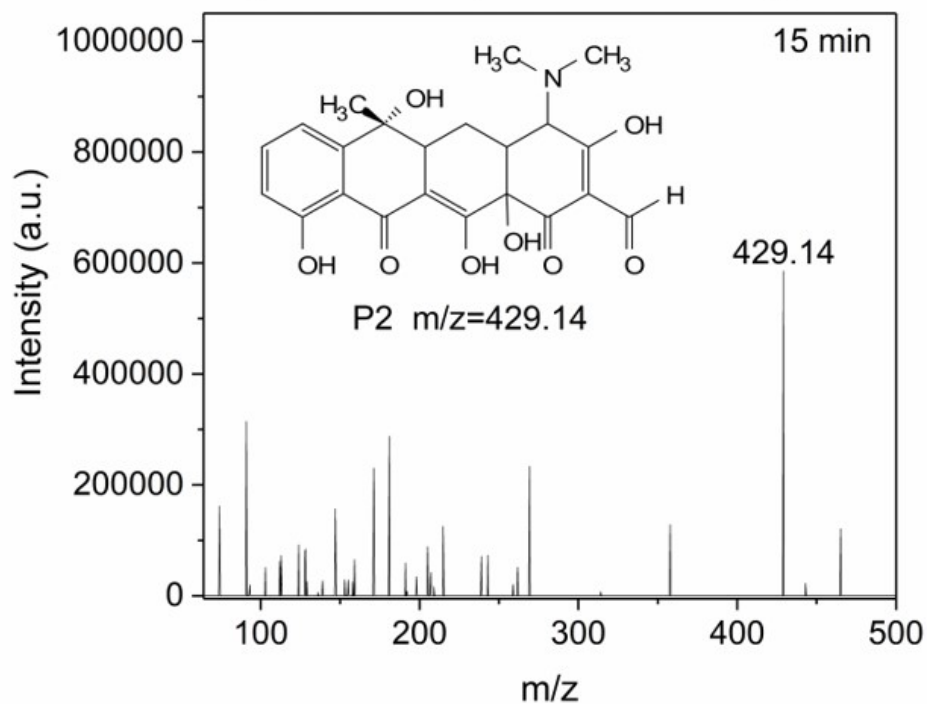


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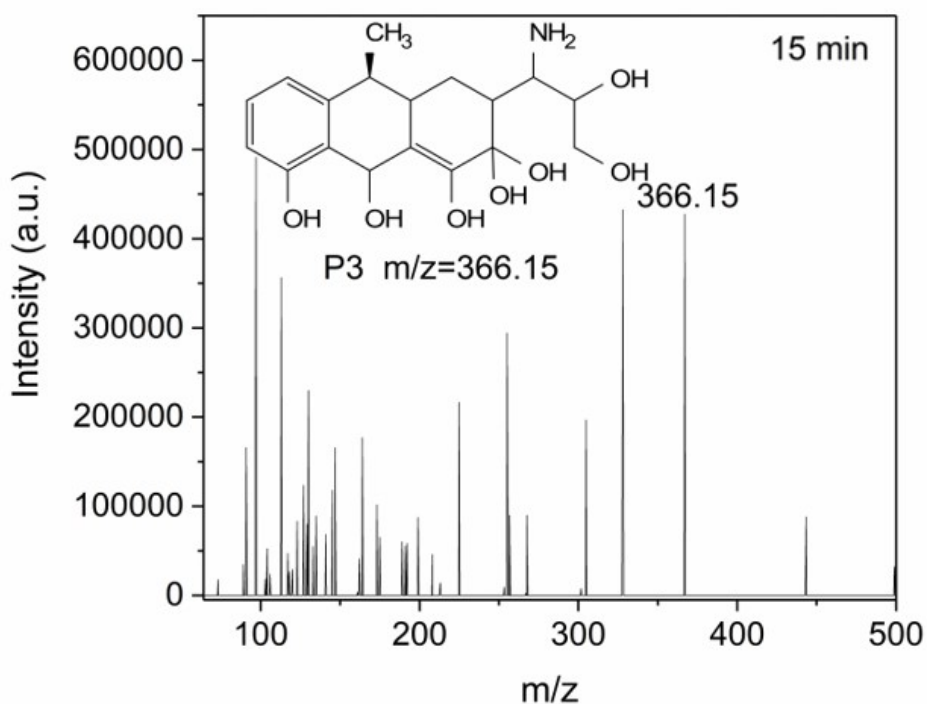
104 **Fig. S21.** MS spectra of identified **TC** and **P1** from solution collected during TC degradation  
105 process over Na-BOC-001 with different irradiation time.



RT: 3.84 AV: 1 NL:5.86E5 C22H23N1O8  
T: -cEIS Q1MS [64.000-500.000] RDB: 12.5 Delta ppm: 1.29

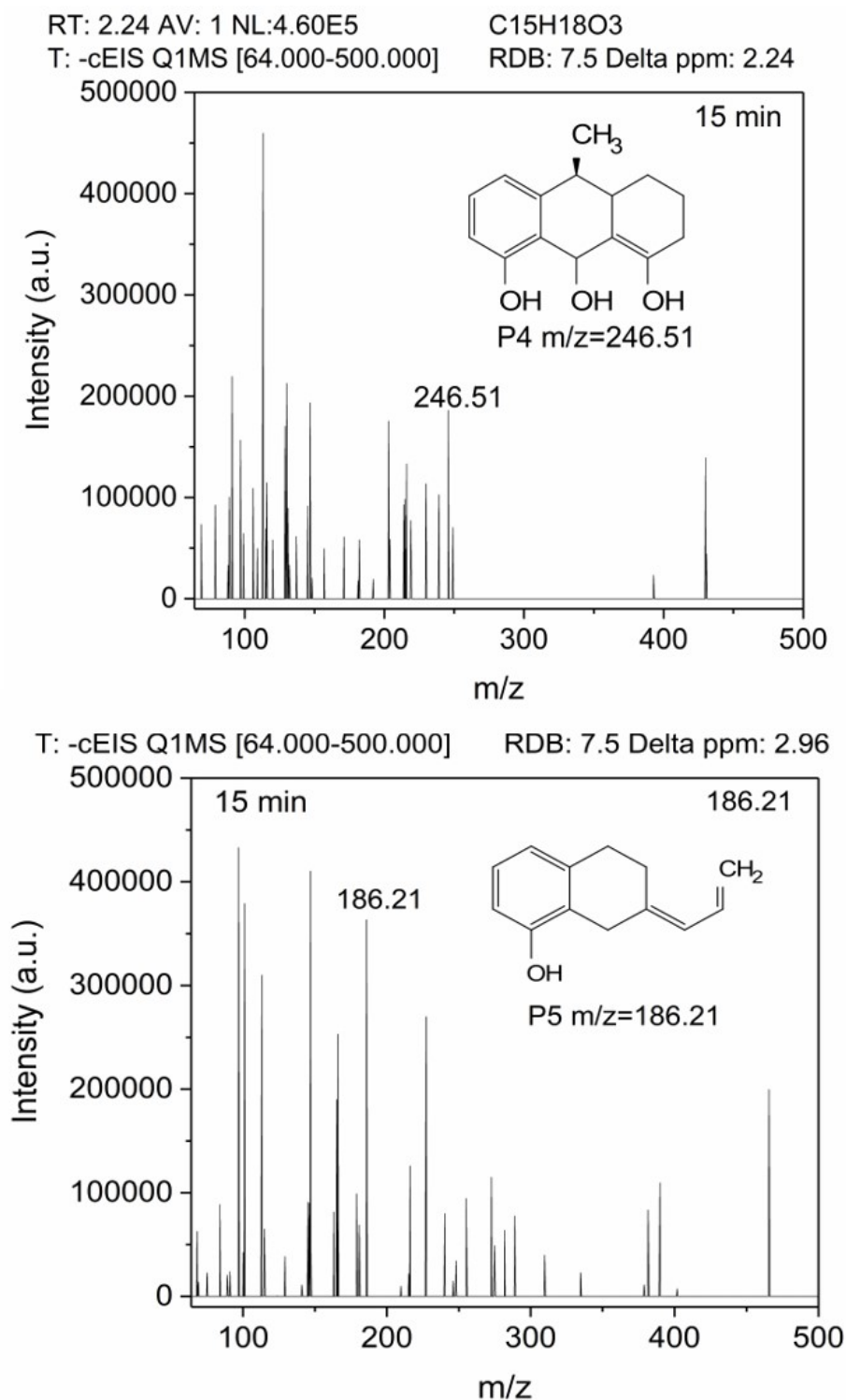


RT: 5.82 AV: 1 NL:2.90E5 C18H25O7N  
T: -cEIS Q1MS [64.000-500.000] RDB: 7.5 Delta ppm: 1.50



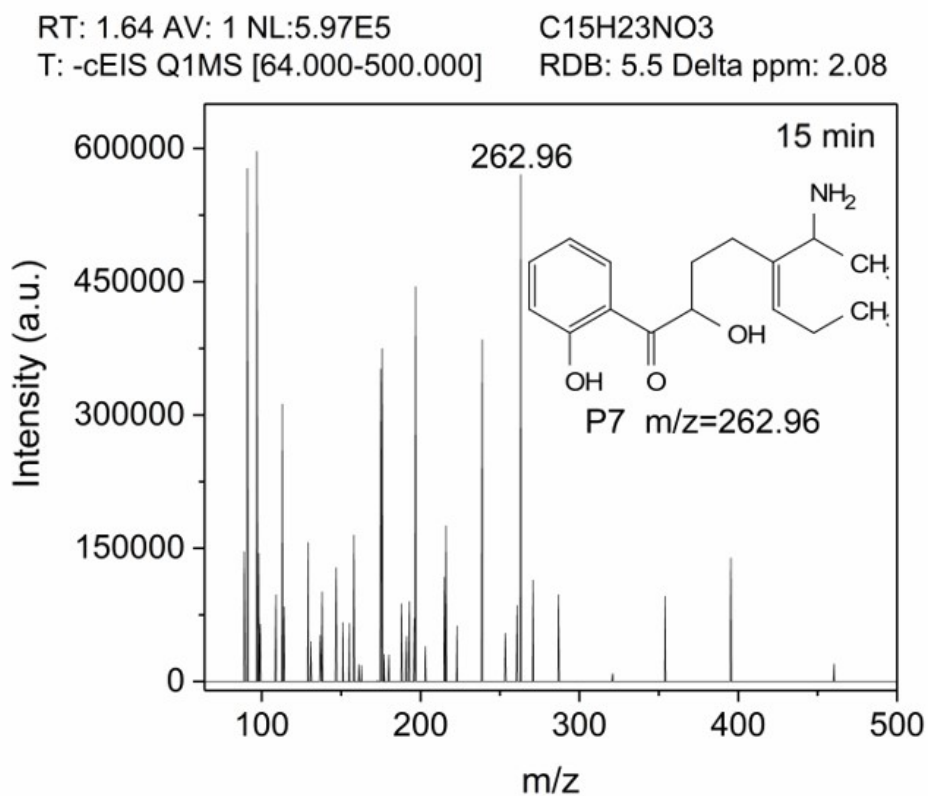
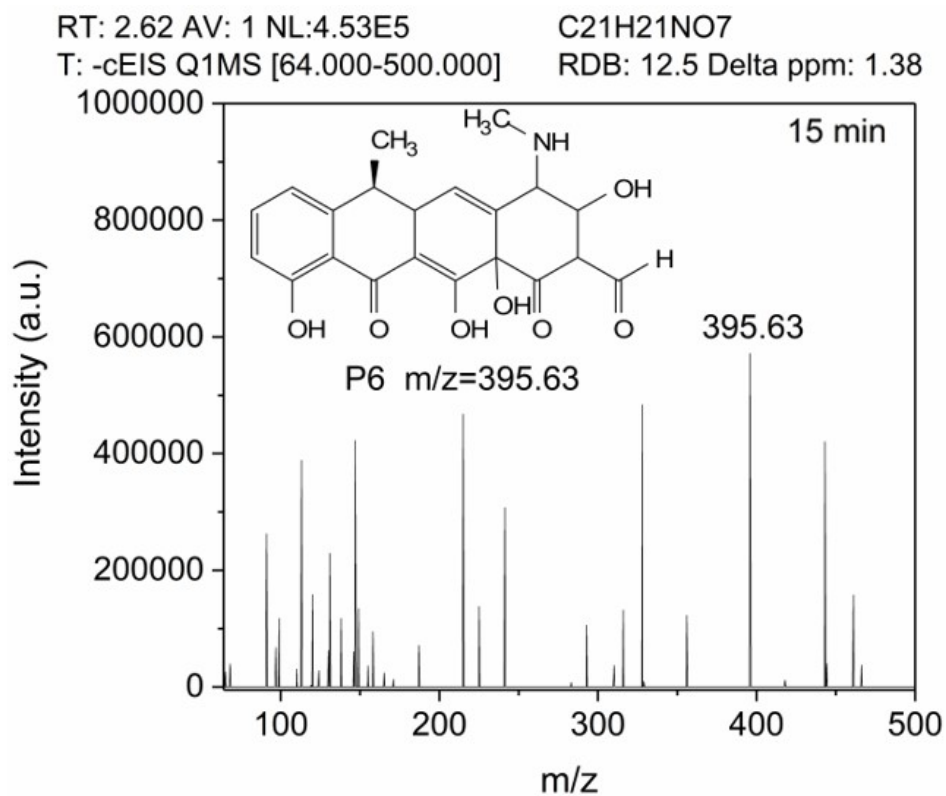
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107 **Fig. S22.** MS spectra of identified **P2** and **P3** from solution collected during TC degradation  
108 process over Na-BOC-001 at irradiation time of 15 min.



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111 **Fig. S23.** MS spectra of identified **P4** and **P5** from solution collected during TC degradation  
112 process over Na-BOC-001 at irradiation time of 15 min.



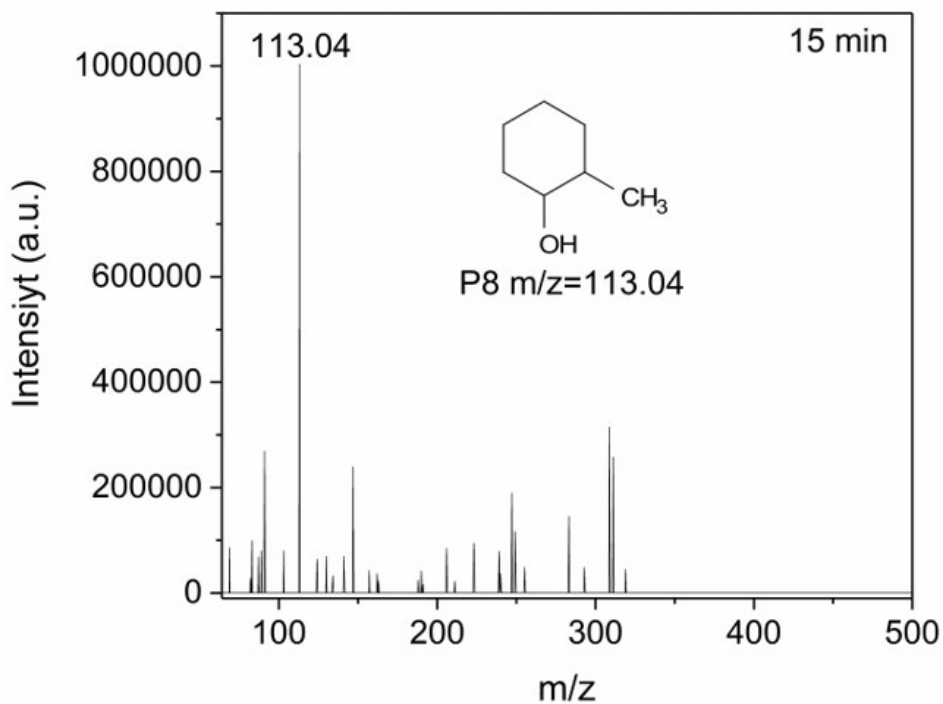
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114 **Fig. S24.** MS spectra of identified **P6** and **P7** from solution collected during TC degradation

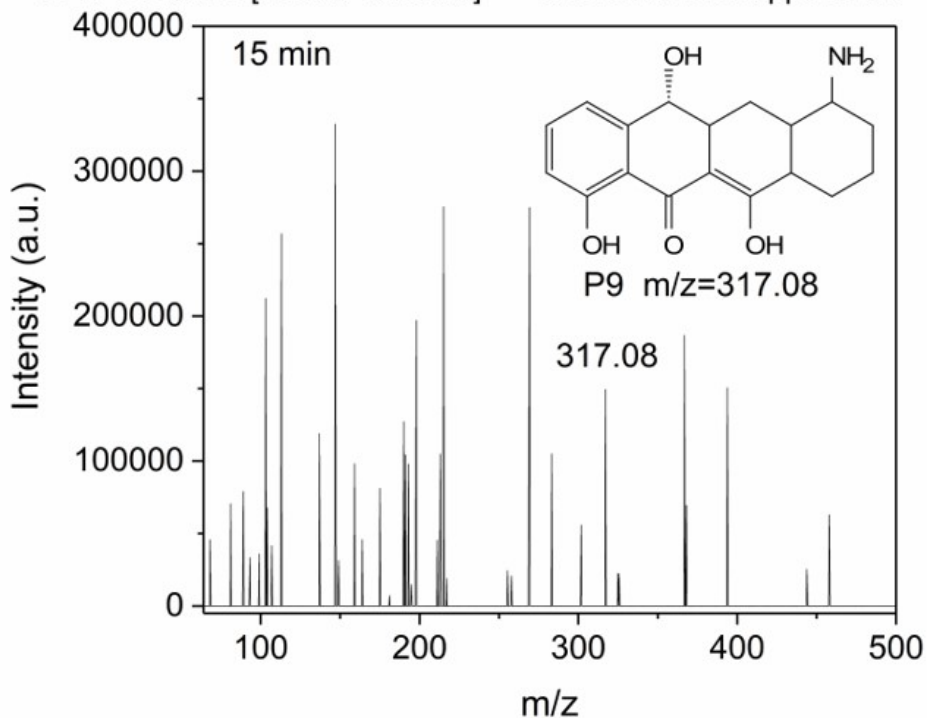
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process over Na-BOC-001 at irradiation time of 15 min.

RT: 8.91 AV: 1 NL: 1.00E6 C7H14O  
T: -cEIS Q1MS [64.000-500.000] RDB: 1.0 Delta ppm: 0.03



RT: 5.23 AV: 1 NL:3.32E5 C18H21O4N  
T: -cEIS Q1MS [64.000-500.000] RDB::15.5 Delta ppm: 1.61

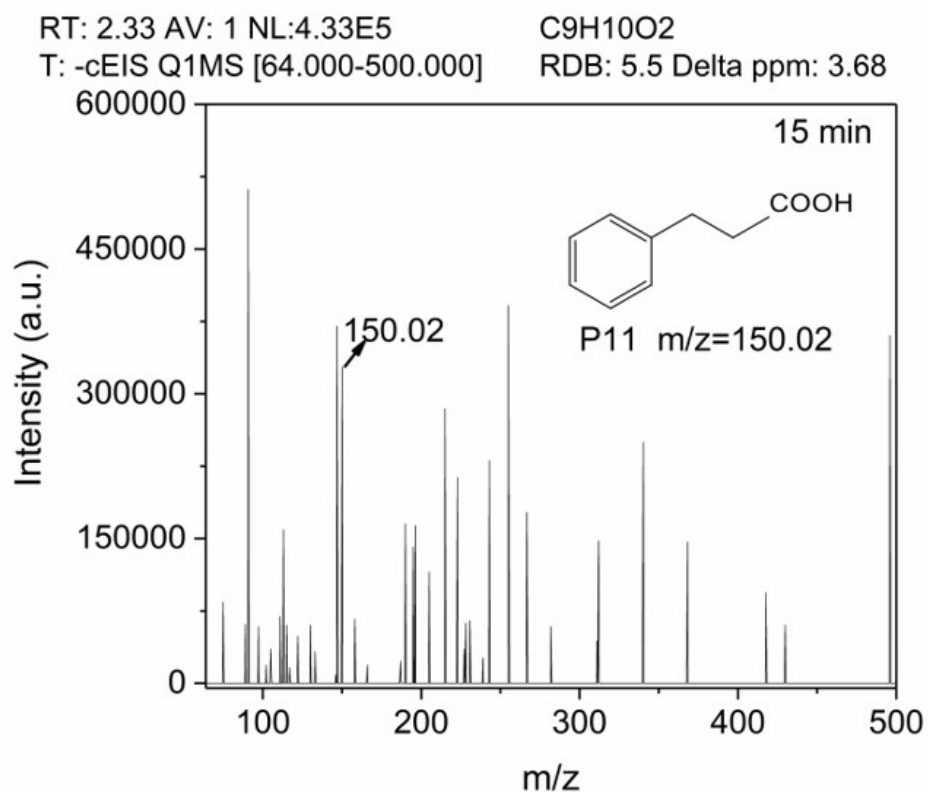
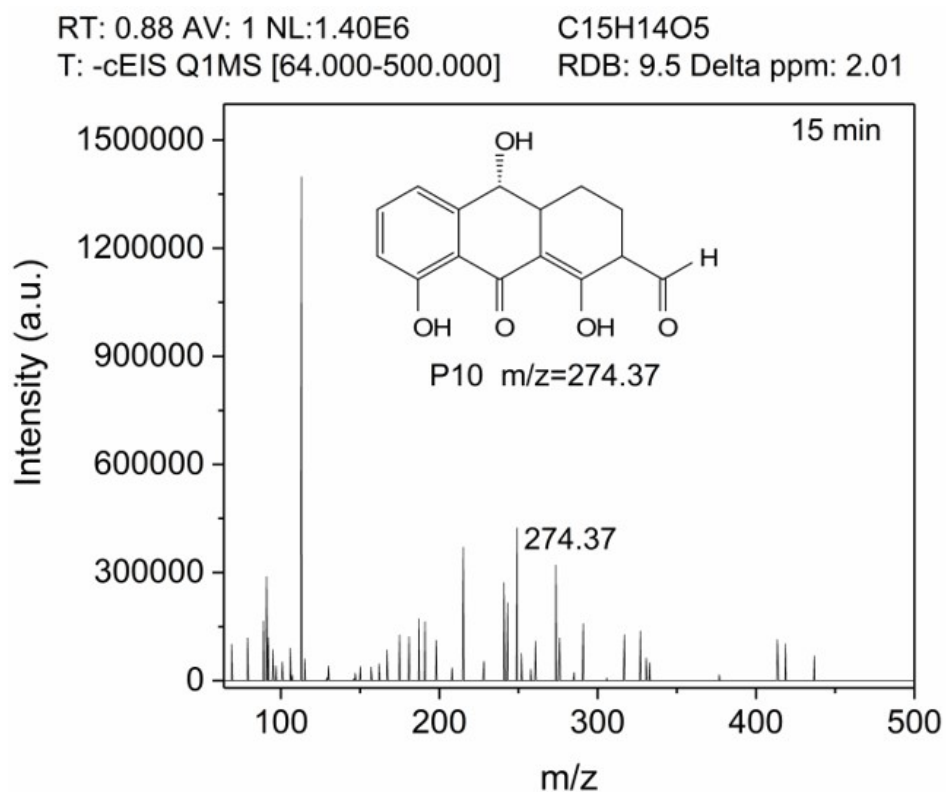


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117 **Fig. S25.** MS spectra of identified **P8** and **P9** from solution collected during TC degradation

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process over Na-BOC-001 at irradiation time of 15 min.

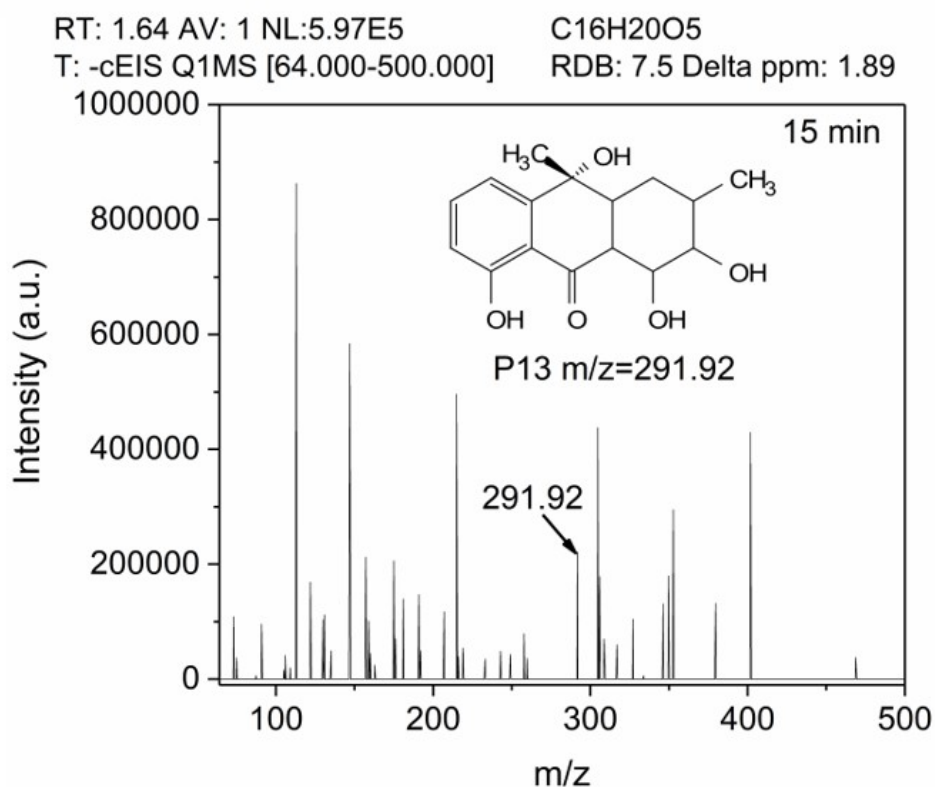
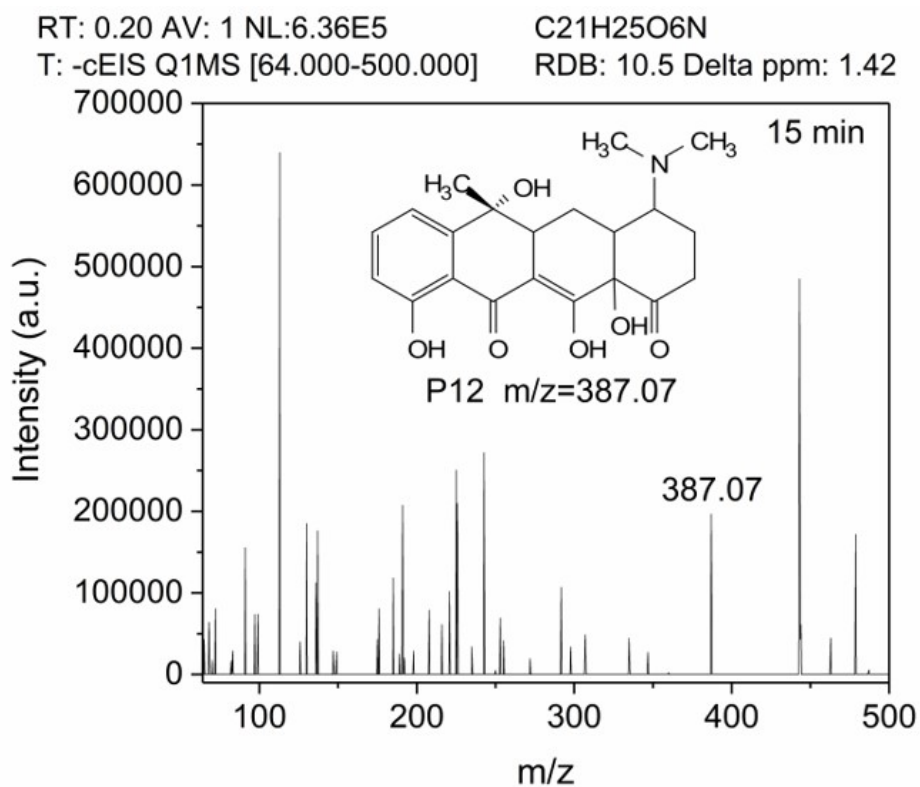


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**Fig. S26.** MS spectra of identified **P10** and **P11** from solution collected during TC degradation process over Na-BOC-001 at irradiation time of 15 min.

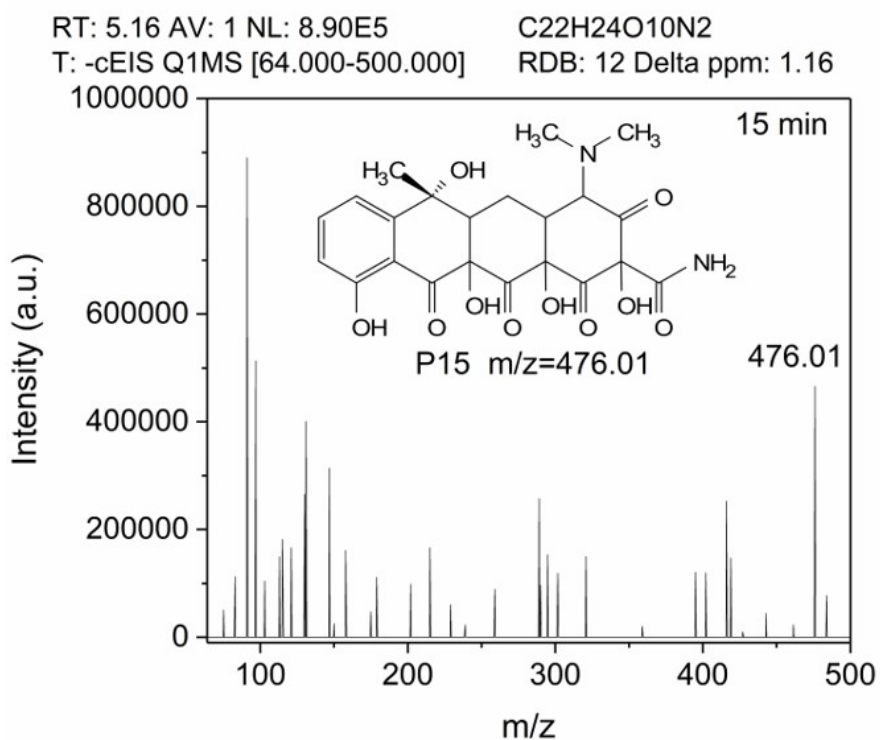
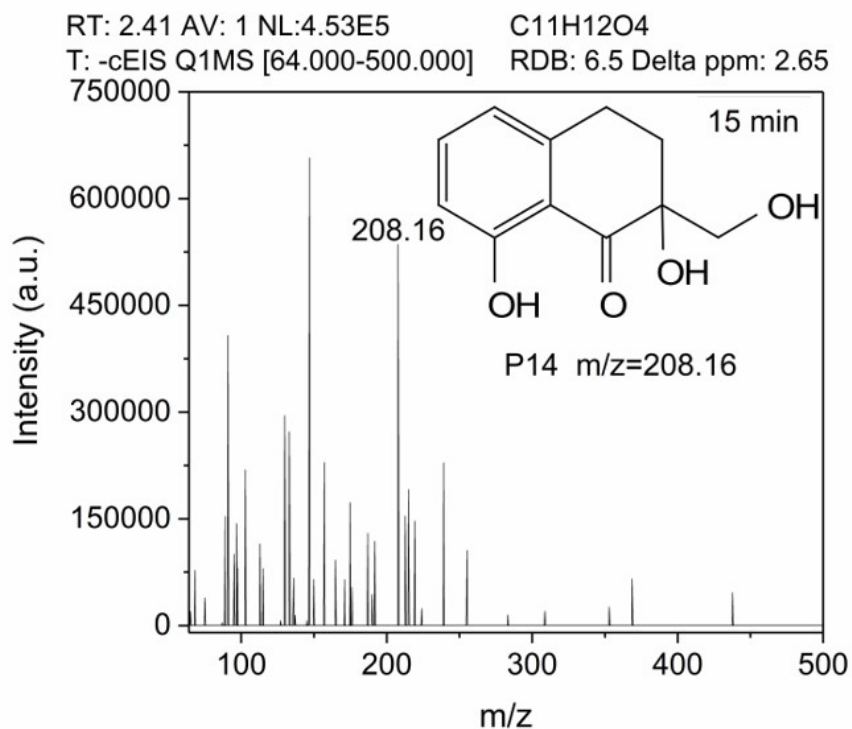


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**Fig. S27.** MS spectra of identified **P12** and **P13** from solution collected during TC degradation process over Na-BOC-001 at irradiation time of 15 min.

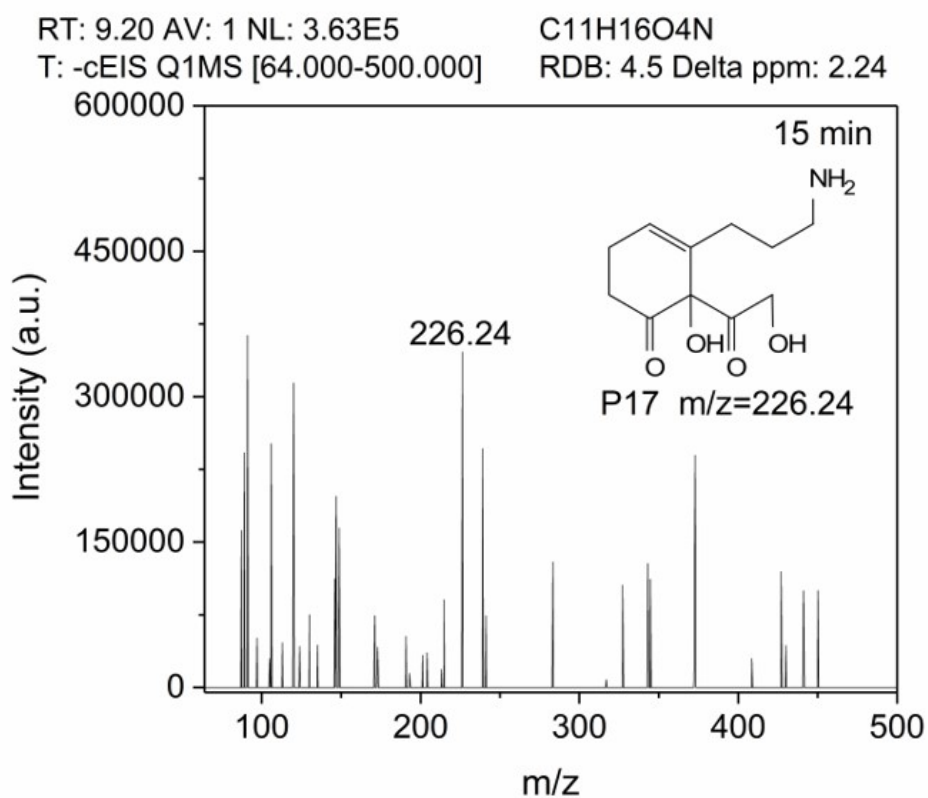
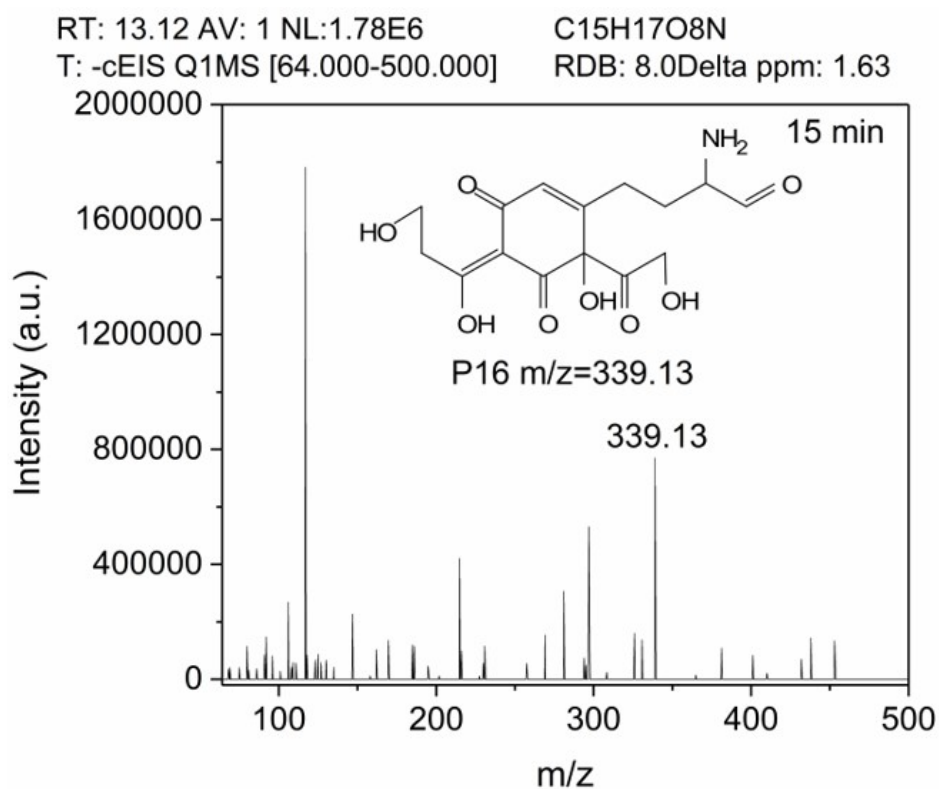


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**Fig. S28.** MS spectra of identified **P14** and **P15** from solution collected during TC degradation process over Na-BOC-001 at irradiation time of 15 min.

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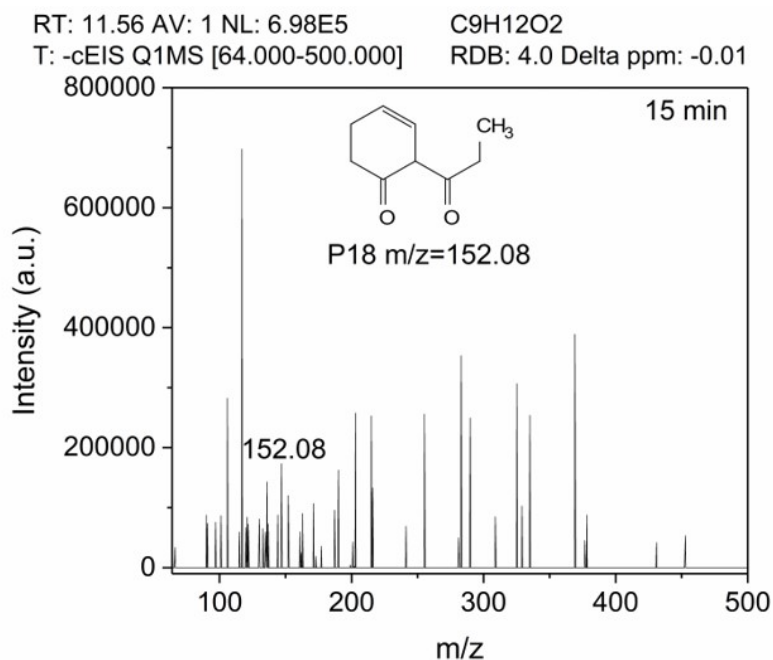
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**Fig. S29.** MS spectra of identified **P16** and **P17** from solution collected during TC degradation process over Na-BOC-001 at irradiation time of 15 min.

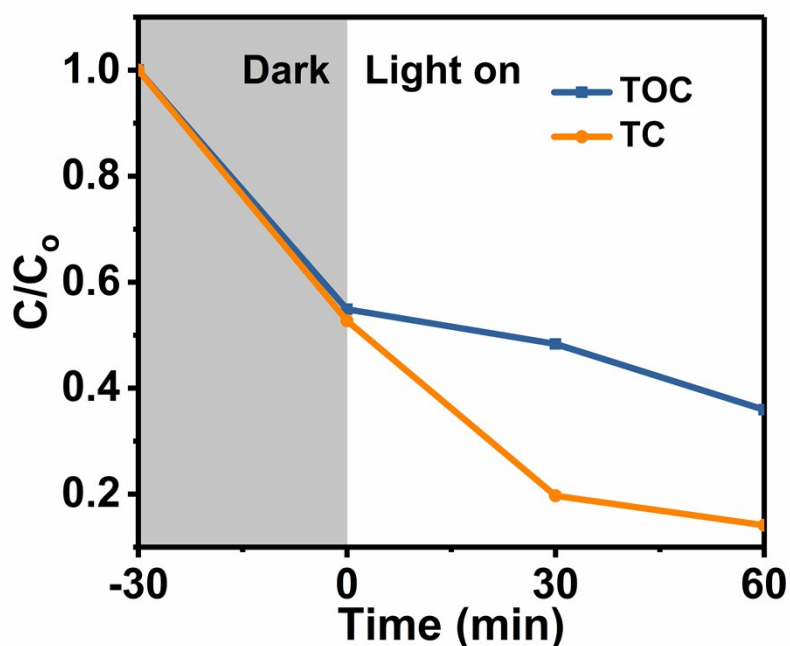
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132 **Fig. S30.** MS spectra of identified **P18** from solution collected during TC degradation process  
133 over Na-BOC-001 at irradiation time of 15 min.



134

135 **Fig. S31.** TOC concentration and TC removal profiles in the TC solution versus time during  
136 the photocatalysis process over Na-BOC-001. Conditions: [TC] = 20 mg L<sup>-1</sup>, [catalyst] = 0.5 g  
137 L<sup>-1</sup>, pH = 6.5, T=25 °C, visible light ( $\lambda > 420$  nm).

138 **Table S1** Elemental composition of as-prepared samples detected by EDS.

Sample	Relative Content (at. %)				Na/Bi (%)
	Na	Bi	O	Cl	
BOC-010		11.71	76.47	11.81	
BOC-001		22.16	58.91	18.93	
0.50-Na-BOC-010	0.20	6.77	86.44	6.59	<b>2.95</b>
0.15-Na-BOC-001	0.11	8.45	83.26	8.18	<b>1.30</b>
0.35-Na-BOC-001	0.32	12.16	76.73	10.79	<b>2.63</b>
0.50-Na-BOC-001	1.56	26.77	48.79	21.87	<b>5.83</b>
0.60-Na-BOC-001	0.53	10.39	79.22	9.86	<b>5.10</b>

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140

141 **Table S2** Relative percent content of lattice oxygen, oxygen vacancy and surface adsorbed  
 142 oxygen peaks in the as-prepared samples by O 1s XPS spectral fitting.

Sample	Lattice oxygen ( $O_{\alpha}$ )			Oxygen vacancy ( $O_{\beta}$ )			Surface adsorbed oxygen ( $O_{\varepsilon}$ )		
	Peak (eV)	Area	* $O_{\alpha}/O_T$ (%)	Peak (eV)	Area	* $O_{\beta}/O_T$ (%)	Peak (eV)	Area	* $O_{\varepsilon}/O_T$ (%)
BOC-010	529.9	44140	71.0	531.3	11295	18.2	533.1	6721	10.8
BOC-001	529.9	86677	75.3	531.3	19399	16.9	533.1	8914	7.8
Na-BOC-001	529.7	70814	70.9	531.1	6986	7.0	532.7	22042	22.1

143 \*  $O_T = O_{\alpha} + O_{\beta} + O_{\varepsilon}$

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145

146 **Table S3** Kinetic fitting data of TC degradation over as-prepared samples under the  
 147 irradiation of UV-Vis light.

Samples	Kinetic Equation	Kinetics Rate ( $k$ , min <sup>-1</sup> ) *	R <sup>2</sup>
BOC-010	$y=0.27303+0.10756x$	0.11666	0.97839
BOC-001	$y=0.23589+0.11869x$	0.12655	0.98763
Na-BOC-010	$y=0.02439+0.14902x$	0.14983	0.99751
Na-BOC-001	$y=0.53512+0.21560x$	0.23343	0.98061

148  $*\ln(C_0/C_t)=k_{app}t$  <sup>1</sup>

149

150 **Table S4** Kinetic fitting data of TC degradation over as-prepared samples under the  
 151 irradiation of UV light.

Samples	Kinetic Equation	Kinetics Rate ( $k$ , min <sup>-1</sup> ) *	R <sup>2</sup>
BOC-010	$y=0.31041+0.10089x$	0.11124	0.96718
BOC-001	$y=0.31041+0.10089x$	0.11463	0.97752
Na-BOC-010	$y=0.04282+0.12225x$	0.12368	0.99253
Na-BOC-001	$y=0.17090+0.18198x$	0.18768	0.98114

152  $*\ln(C_0/C_t)=k_{app}t$  <sup>1</sup>

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156

157 **Table S5** Kinetic fitting data of TC degradation over as-prepared samples under the  
 158 irradiation of Vis light.

Samples	Kinetic Equation	Kinetics Rate ( $k$ , min <sup>-1</sup> )*	R <sup>2</sup>
BOC-010	$y=0.06176+0.00605x$	0.00708	0.89510
BOC-001	$y=0.19743+0.02205x$	0.02534	0.90903
Na-BOC-010	$y=0.07620+0.00878x$	0.01005	0.92054
Na-BOC-001	$y=0.33510+0.02536x$	0.03095	0.82498

159 \* $\ln(C_0/C_t)=k_{app}t$  <sup>1</sup>

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161

162 **Table S6** Kinetic fitting data of TC degradation over as-prepared samples.

Photocatalysts	Contaminant	Light sources	Dosage (g L <sup>-1</sup> )	C <sub>0</sub> (mg L <sup>-1</sup> )	Volume (mL)	Time (min)	Degradation rates (%)	Ref.
Na-BOC-001	TC	Vis	0.5	20	100	60	87.2	This work
WO <sub>3</sub> /Bi <sub>12</sub> O <sub>17</sub> Cl <sub>2</sub>	TC	Vis	0.8	20	80	180	63.2	1
BiOCl/CAU-17	TC	Xenon	0.1	10	100	90	85.5	2
NH <sub>2</sub> -MIL-125 (Ti)/BiOCl	TC	Vis	0.5	20	100	120	78.0	3
P-BiOCl	TC	Xenon	0.5	20	100	30	81.0	4

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