

## Supporting Information

### Silver-functionalized Bismuth oxide nanoparticles ( $\text{AgBi}_2\text{O}_3$ ) for superior electrochemical detection of Glucose, $\text{NO}_2^-$ and $\text{H}_2\text{O}_2$

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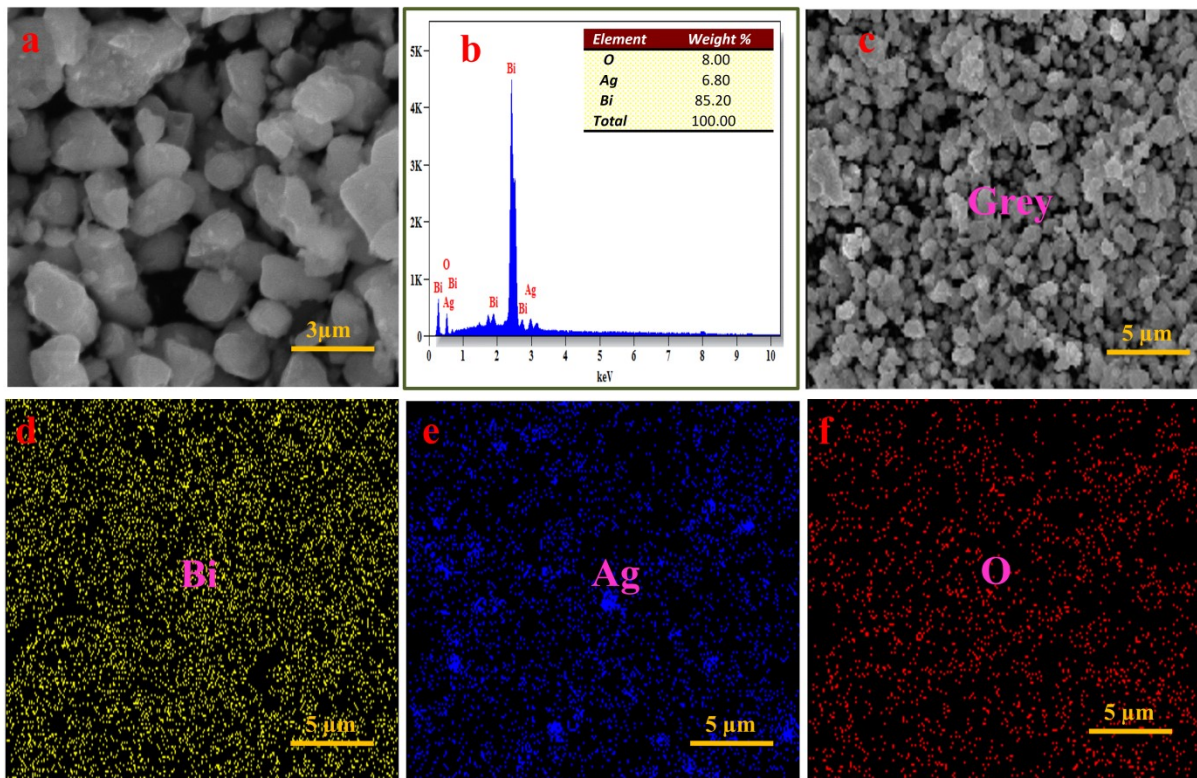
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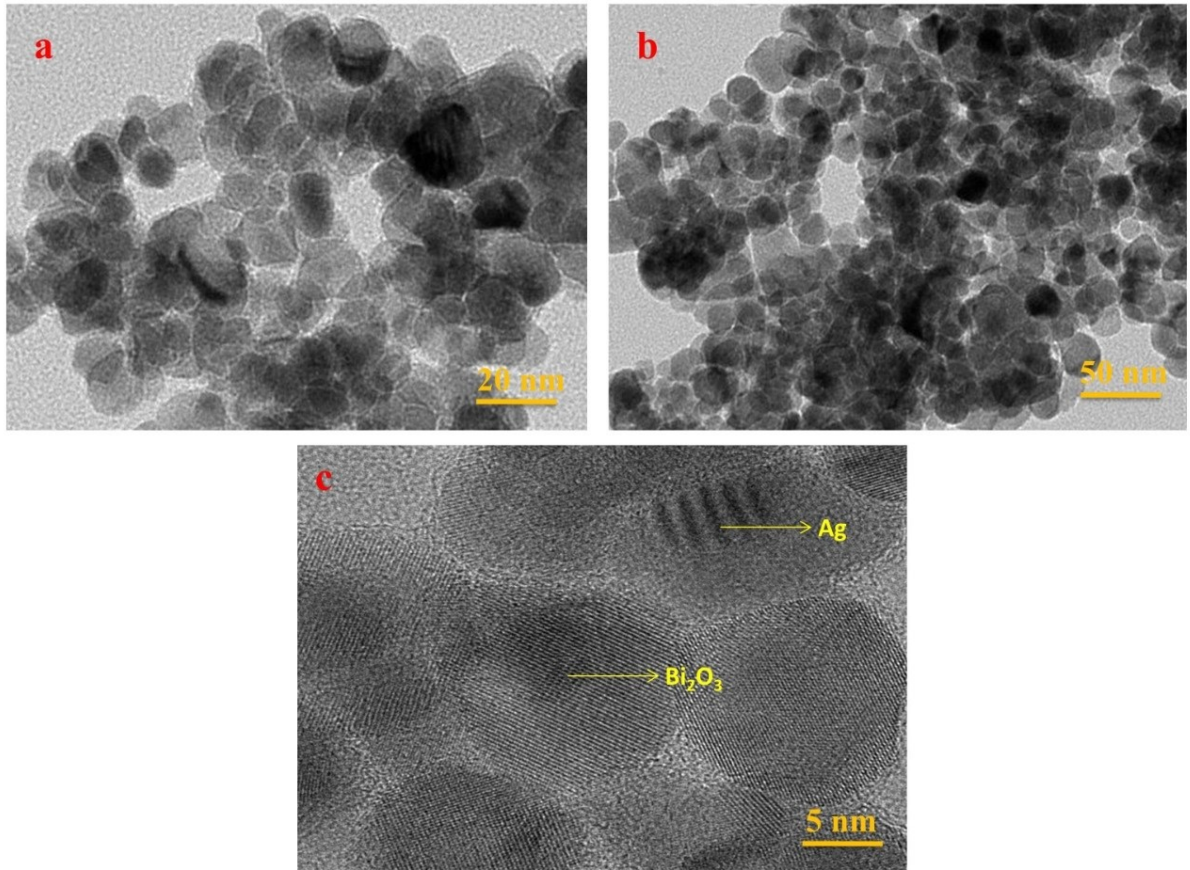
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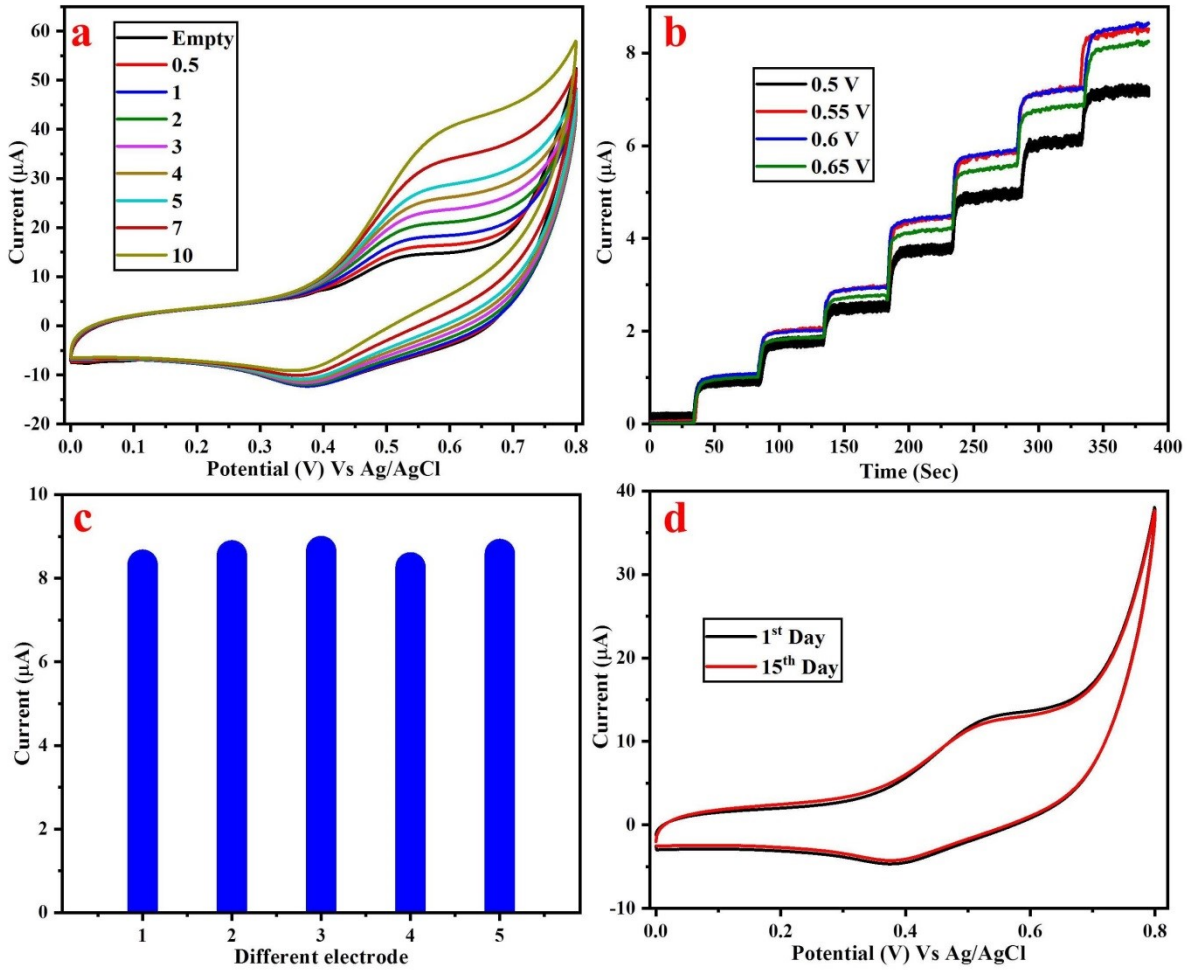
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2 **Fig. S1.** a) HR-SEM image of SBO NPs, b) EDS spectrum of SBO NPs, c) Mapping analysis  
 3 of gray spectrum, d) Mapping of Bi, e) Mapping of Ag, and f) Mapping of O.



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**Fig. S2.** HR-TEM image of SBO NPs a) 20nm, b) 50 nm, c) IFFT image.



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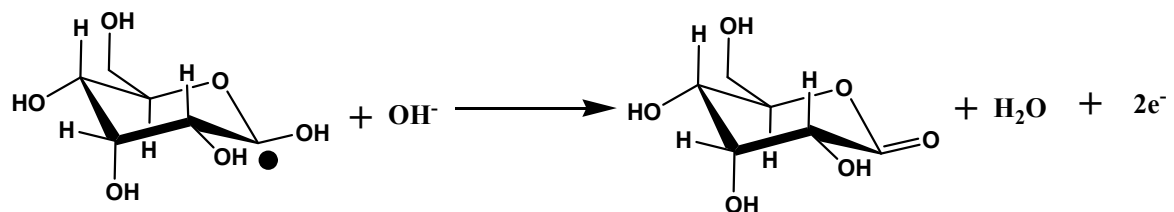
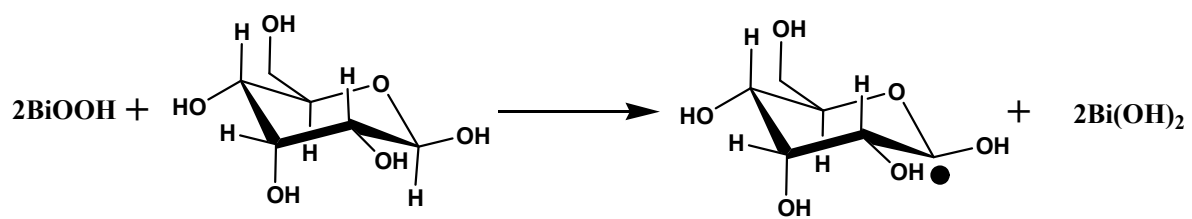
2 **Fig. S3.** a) CVs recorded at the SBO-GC electrode with different concentration of glucose. b)  
 3 Amperometric *i-t* curve for different potential in addition of 0.5 and 1 mM of glucose in  
 4 NaOH solution. c) Reproducibility study for SBO-GCE for five different electrodes. d) CV  
 5 plot of stability study recorded before and after 15 days with 1 mM of glucose.

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**Scheme S1.** Electrochemical mechanism for detection of glucose

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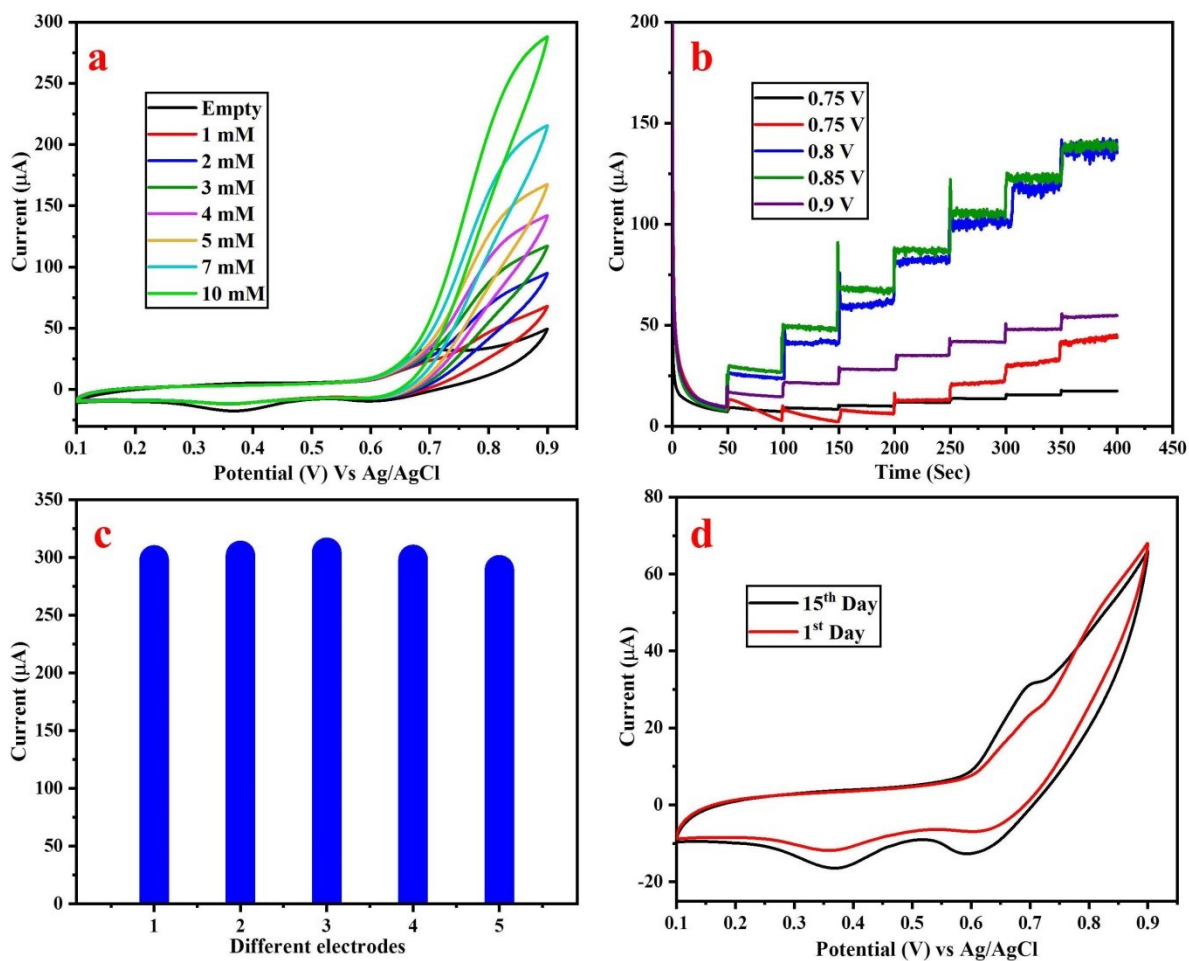
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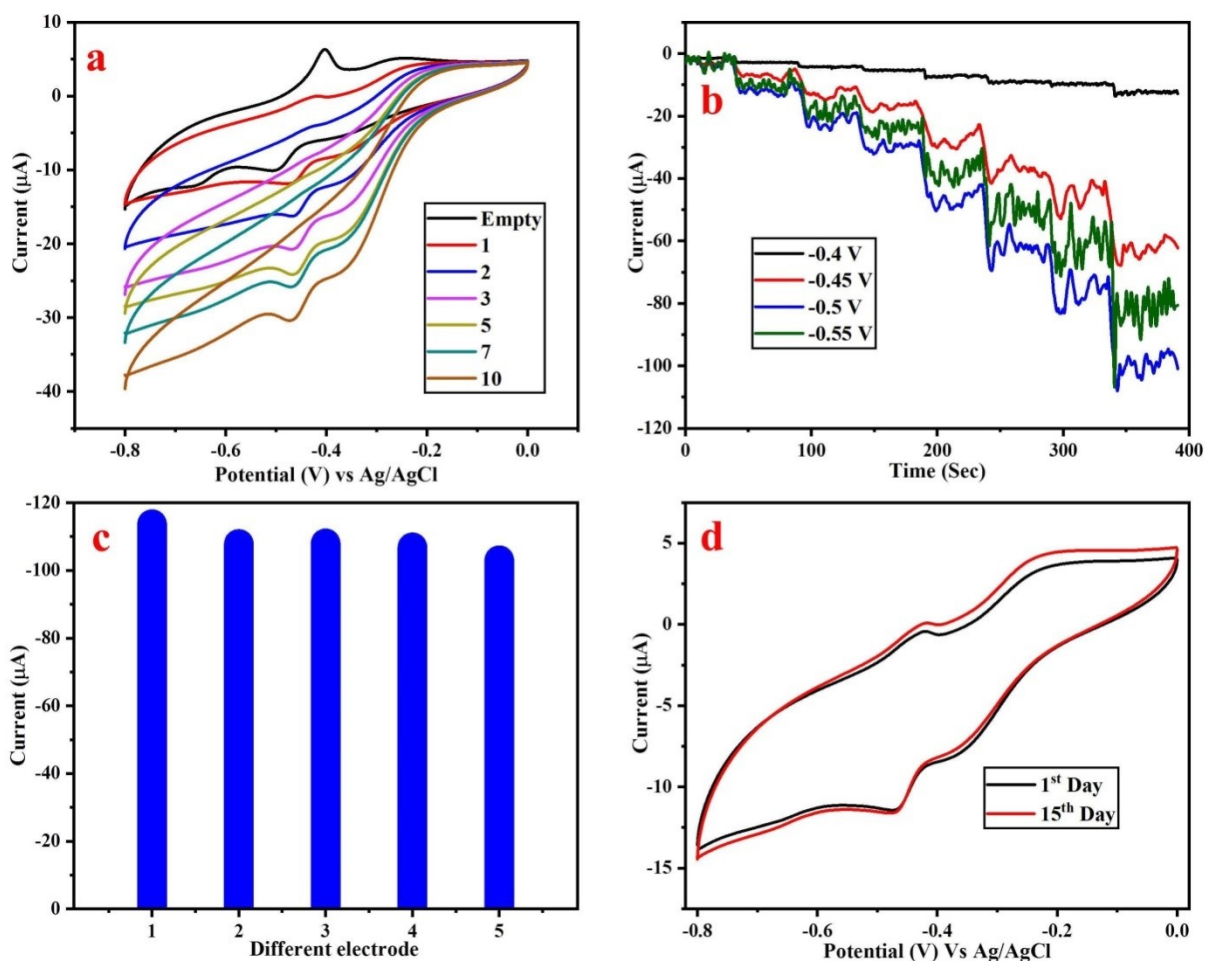
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2 **Fig. S4.** a) CVs recorded at the SBO-GC electrode with different concentration of  $\text{NO}_2^-$ . b)  
 3 Amperometric  $i-t$  curve for different potential in addition of 0.5 mM of  $\text{NO}_2^-$  in 0.1 M NaOH  
 4 solution. c) Reproducibility study for SBO-GCE for five different electrodes. d) CV plot of  
 5 stability study recorded before and after 15 days with 1 mM of  $\text{NO}_2^-$ .



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2 **Fig. S5.** a) CVs recorded at the SBO-GC electrode with different concentration of glucose. b)  
 3 Amperometric  $i-t$  curve for different potential in addition of 0.5 and 1 mM of  $\text{H}_2\text{O}_2$  in 0.1 M  
 4 NaOH solution. c) Reproducibility study for SBO-GCE for five different electrodes. d) CV  
 5 plot of stability study recorded before and after 15 days with 1 mM of  $\text{H}_2\text{O}_2$ .

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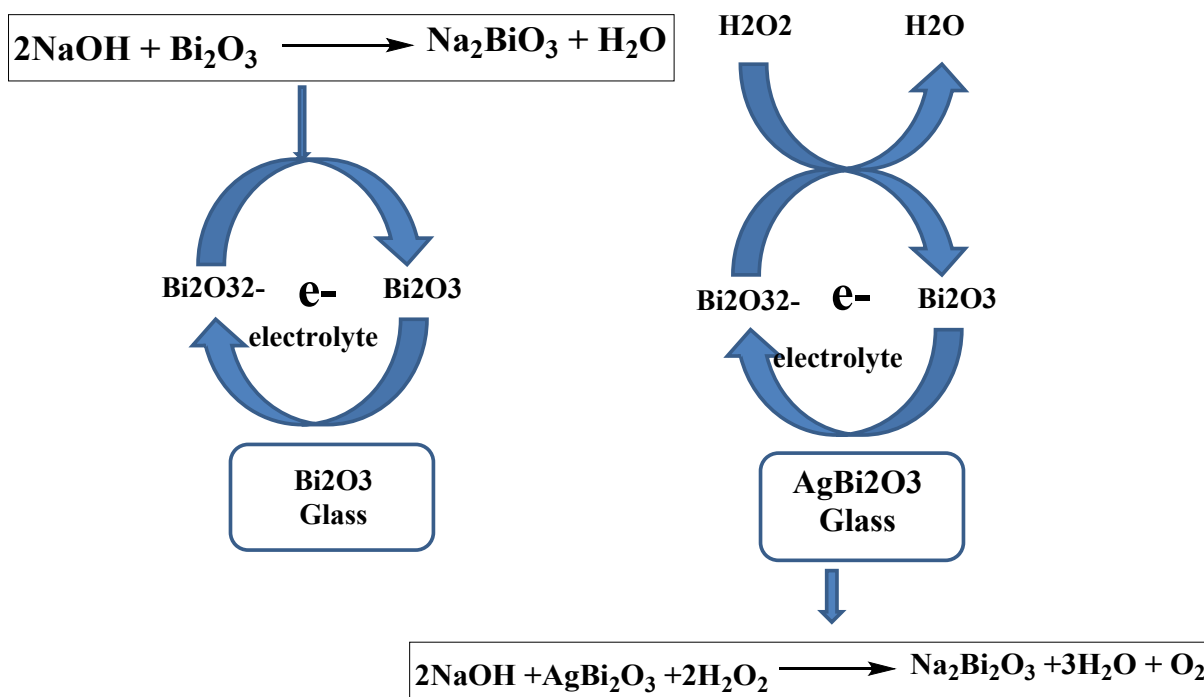
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3 **Fig. S6.** The proposed mechanism for the reduction of  $\text{H}_2\text{O}_2$  on the SBO-GC electrode

4 **Effect of temperature**

5 The amperometric response was recorded in optimum working environments (in 0.1

6 M NaOH), at various temperatures (20 to 40 °C), which are shown in **Fig. S7**. The steady

7 state current increases from 20 to 30 °C, after 30 °C it was decreased in the amperometric

8 response. This drop was an effect of the loss of activity initiated by the higher temperatures.

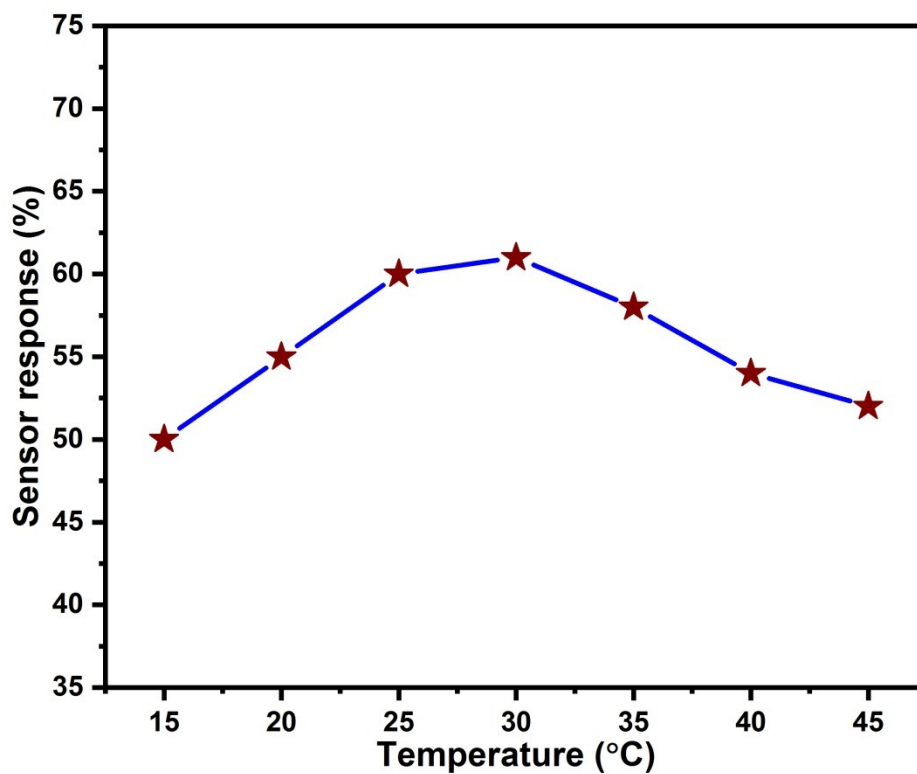
9 Conversely, the amperometric response of the biosensor at 25 and 30 °C was very similar to

10 each other (**Fig. S7**). Therefore, in order to keep the fabricated electrode more stable, 25 °C,

11 which is very close to the growth temperature of the bacterium (28 °C), was employed in

12 further studies.





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2 **Fig. S7.** The thermal effect on the fabricated  $\text{AgBi}_2\text{O}_3$ -GCE sensing response to addition of  
3 glucose in electrolyte solution

1 **Table S1.** Comparison of the Analytical Performance for Different Nanomaterial-Based on Electrochemical Sensing of Glucose

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Sensor Materials	Detection potential (V)	Sensitivity ( $\mu\text{A mM}^{-1} \text{cm}^{-2}$ )	LOD	Linear range	Refs.
Ag/CNT/Ch/ITO	-0.51	135.9	0.1 $\mu\text{M}$	0.5 – 50 $\mu\text{M}$	1
Ag-PANI/rGO	0.5	2.7664	0.79 $\mu\text{M}$	0.1 $\mu\text{M}$ – 50 $\mu\text{M}$	2
PmAPNFs/AgNPs/GCE	0.34	17.45	0.062 $\mu\text{M}$	0.1–8.0	3
GOD/nano-BiOx	0.5	51	0.4 $\mu\text{M}$	1 $\mu\text{M}$ – 1.5 mM	4
SPCE/GNR/Bi <sub>2</sub> O <sub>3</sub>	0.6	64.81	0.07 mM	0.28 - 1.70 mM	5
BiOCl-G NHS	0.5	1.878	0.22 mM	0.5 – 2 mM	6
FTO nanoCuBi <sub>2</sub> O <sub>4</sub>  CuO	0.55	330	0.7 $\mu\text{M}$	Upto 8 mM	7
HO-BiONO <sub>3</sub> - GCE	0.3	8.2	0.12 $\mu\text{M}$	5 $\mu\text{M}$ – 2.1 mM	8
SBO-GCE	0.55	2.153	0.87 $\mu\text{M}$	1 $\mu\text{M}$ – 5.848 mM	Present work

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1 **Table S2.** Comparison of the Analytical Performance for Different Nanomaterial-Based on Electrochemical Sensing of NO<sub>2</sub><sup>-</sup>

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Sensor Materials	Detection potential (V)	Sensitivity ( $\mu\text{A mM}^{-1} \text{cm}^{-2}$ )	LOD	Linear range	Refs.
Ag-GCE	1.0	1642.27	0.046	1 $\mu\text{M}$ – 6 mM	9
Ag-SO <sub>3</sub> -NU-902	1.1	-	9.1	Upto 2 mM	10
AgNPs/MWCNTs/GCE	0.85	0.19	0.095	1 $\mu\text{M}$ - 100 $\mu\text{M}$	11
rGO/AgNPs/poly(PyY)	0.86	13.5	0.012	1 $\mu\text{M}$ - 1000 $\mu\text{M}$	12
Ag-P(MMA-co-AMPS)- GCE	0.9	104.6	0.2	1 $\mu\text{M}$ - 100 mM	13
Bi <sub>2</sub> Se <sub>3</sub> @MWNTs- COOH/CE	0.8	223	0.002	0.01 $\mu\text{M}$ - 7 mM	14
SBO-GCE	0.85	22	1.8	1 $\mu\text{M}$ – 5.848 mM	Present work

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1 **Table S3.** Comparison of the Analytical Performance for Different Nanomaterial-Based on Electrochemical Sensing of H<sub>2</sub>O<sub>2</sub>

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Sensor Materials	Detection potential (V)	Sensitivity ( $\mu\text{A mM}^{-1} \text{cm}^{-2}$ )	LOD	Linear range	Refs.
C $\mu$ F/Ag NPs-Naf	-0.35	21.93	0.485 $\mu\text{M}$	0.10 – 80 mM	15
AgNPs/Ox-pTTBA/MWCNT	-0.6	-	0.24 $\mu\text{M}$	10 – 260 $\mu\text{M}$	16
PpyNFs-AgNPs-rGO/GCE	-0.75	-	1.099 $\mu\text{M}$	0.1 – 5 mM	17
NF/HRP/Bi <sub>2</sub> O <sub>3</sub> – MWCNT/GCE	-0.3	26.54	-	8.34 – 28.88	18
BiNDs/GaN	-0.7	60	5 $\mu\text{M}$	0.01 – 1 mM	19
CuBi <sub>2</sub> O <sub>4</sub>	-0.6	280	0.38 mM	-	20
CPE/BiFeO <sub>3</sub>	-0.8	0.142	0.080 $\mu\text{M}$	0.0002-0.05	21
SBO-GCE	-0.5	1.72	1.15 $\mu\text{M}$	2 $\mu\text{M}$ – 6.847 mM	Present work

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