## **Electronic Supplementary Information (ESI)**

## A sensitive Non-Enzymatic Electrochemical Glucose Sensor based on ZnO/Co<sub>3</sub>O<sub>4</sub>/Reduced Graphene Oxide nanocomposite

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Table S1 Absorbance maximum wavelengths and hand gaps of as synthesized sensors

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Samples	Absorbance	Maximum Wavelength (nm)	Band gap (Eg) (eV)
Co <sub>3</sub> O <sub>4</sub>	1.83	358	3.46
ZnO	2.25	378	3.28
Co <sub>3</sub> O <sub>4</sub> /ZnO	2.27	406	3.05
Co <sub>3</sub> O <sub>4</sub> /ZnO/rGO	2.30	495	2.50

Table S2. Average crystallite size (D) of the as-synthesized nanoparticles

Sample	2θ (Degree)	$\beta$ (Radians)	D (nm)
Co <sub>3</sub> O <sub>4</sub>	37.31	0.003142	44.9
ZnO NPs	36.27	0.008378	21.9
ZnO/Co <sub>3</sub> O <sub>4</sub>	36.37	0.006981	25.2
ZnO/Co <sub>3</sub> O <sub>4</sub> /rGO	36.36	0.010123	17.5

Order of average crystallite size (D) of as-synthesized sensors is:  $Co_3O_4 < ZnO/Co_3O_4 < ZnO/NPs < ZnO/Co_3O_4/rGO$ 

Table S3. Summary of electrochemical Parameters of modified electrodes in 2mM of K3Fe  $(CN)_6$  at a scan rate of 50 mVs<sup>-1</sup>

Electrodes	i <sub>pa</sub>	E <sub>pa</sub>	i <sub>pc</sub>	Epc	i <sub>pa</sub> /i <sub>pc</sub>	$\Delta_{E_p}$	Eo
	$(\mu A)$	(mV)	$(\mu A)$	(mv)	(A)	(mv)	(mV)
Bare GCE	6.08	250.93	-6.5	160.23	0.935	90.7	220.58
Co <sub>3</sub> O <sub>4</sub>	10.67	255.4	-11.6	167.8	0.920	87.6	216.55

ZnO	15.63	262.99	-16.6	170.8	0.942	92.19	223.89
ZnO/Co <sub>3</sub> O <sub>4</sub>	21.4	265.29	-20.96	177.7	1.021	87.59	216.545
ZnO/Co <sub>3</sub> O <sub>4</sub> /rGO	24.6	270.34	-24.56	184.8	1.002	85.54	220.34



Figure S1. Cyclic Voltammograms of (a)  $2mM K_3Fe(CN)_6$  in 0.1 M KCl on glassy carbon electrode at scan rates 10, 25, 50, 75 and 100 mVs<sup>-1</sup>, (b) CV of supporting electrolyte 0.1 M KCl and 2 mM K3[Fe(CN)6]/0.1M KCl at scan rate of 50 mVs<sup>-1</sup>.



Figure. S2 Anodic current of  $ZnO/Co_3O_4/GCE$  (a) at different concentration of NaOH (0.005M up to 0.2M) upon addition of 2 mM glucose at scan rate of 50 mVs<sup>-1</sup> (b) the plot of peak current versus respective concentration of NaOH.



Figure **S3**. Current –time graph of (A) the repetability four different measurement of a same sansor by injection of 0.5 mM glucose in 100 s interval (B) the reproducibility of the sensor by injection of 2mM glucose at 50 s interval.

Table S4. Repeatability and reproducibility of the ZnO/Co<sub>3</sub>O<sub>4</sub>/rGO nanocomposite sensor

2 mM	Response current (
glucose	$\mu A_{)}$
1	109.86
2	105.86
3	113.238
4	111.86
Mean	110.2045
RSD	2.91%

Sensor	Response current (
	$\mu A_{)}$
1	149.54
2	154.58
3	158.58
4	147.58
Mean	152.57
RSD	3.26%

Concentration (mM)	Response current $(\mu A)$
0.0001	10
0.05	15
0.01	21
0.1	36
0.5	85
0.8	120
1.5	175
3	290
6	530
10	890

Table **S5**. The current response of different concentration of glucose using  $ZnO/Co_3O_4$  /rGO electrode