## **Supporting information**

## Preparation of Ni-Fe-based transition metal phosphide as efficient electrocatalyst for the oxygen evolution reaction

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Fig. S1 FT-IR spectra of Ni<sub>2</sub>P-Fe<sub>2</sub>P/NF and Ni-Fe PBA/NF.



Fig. S2 TGA curve of Ni-Fe PBA precursor measured under a  $N_{\rm 2}$  atmosphere.



Fig. S3 EDX spectrum and corresponding elemental compositions of the Ni<sub>2</sub>P-  $Fe_2P/NF$ .





**Fig. S5** Under different conditions, the LSV curves and Tafel slopes of mixed-metal phosphides.



Fig. S6 The cyclic voltammograms (CVs) measurements with various scan rates for (a) NF, (b) Ni(OH)<sub>2</sub>/NF, (c) Ni-Fe PBA/NF, (d)Ni<sub>2</sub>P-Fe<sub>2</sub>P/NF.



Fig. S7 The XRD patterns of fresh sample and sample after OER.



Fig. S8 The high resolution XPS of Ni 2p before (down) and after (up) OER test.



Fig. S9 The high resolution XPS of P 2p before (down) and after (up) OER test.



Fig. S10 The high resolution XPS of Fe 2p before (down) and after (up) OER test.



**Fig. S11** The SEM images of (a,b)bare NF, (c,d) Ni<sub>2</sub>P-Fe<sub>2</sub>P/NF and (e,f) Ni<sub>2</sub>P-Fe<sub>2</sub>P/NF after OER test.

Table S1 Average mass loading of catalysts on NF substrates.

Samples	Mass loading (mg cm <sup>-2</sup> )
Ni(OH)2/NF	3.19
Ni-Fe PBA/NF	2.02
Ni <sub>2</sub> P-Fe <sub>2</sub> P/NF	2.63

**Table S2** The simulated series resistance ( $R_s$ ) and charge transfer resistance( $R_{ct}$ ) values based on the fitting models.

Samples	<i>R</i> s (Ω)	$R_{ m ct}\left(\Omega ight)$
NF	1.258	5.68
Ni(OH) <sub>2</sub> /NF	1.276	5.50
Ni-Fe PBA/NF	1.497	1.18
Ni <sub>2</sub> P-Fe <sub>2</sub> P/NF	1.288	0.23

**Table S3** The activity comparison of PBA precursors on the surface of NF forOER in 1.0 M KOH solution.

Catalysis	Electrolyte	Overpotential	Tafel slope	Ref
Ni <sub>2</sub> P-Fe <sub>2</sub> P/NF	1 M KOH	222 mV	40.9 mV dec <sup>-1</sup>	This
		100 mA cm <sup>-2</sup>		work
NiFeP@NiP@NF	1 M KOH	252 mV	56 mV dec <sup>-1</sup>	1
		100 mA cm <sup>-2</sup>		
NiFe2O4@NPNiFePB	1 M KOH	304 mV	80 mV dec <sup>-1</sup>	2
A/NF		100 mA cm <sup>-2</sup>		
Ni(OH)2@Ni3S2/NF	1 M KOH	210 mV	62.0 mV dec <sup>-1</sup>	3
		100 mA cm <sup>-2</sup>		
NiSe <sub>2</sub> /CoSe/NF	1 M KOH	270 mV	69.77 mV dec <sup>-1</sup>	4
		100 mA cm <sup>-2</sup>		
Ni-Fe-P@NC/NF	1 M KOH	245 mV	81.0 mV dec⁻¹	5
		100 mA·cm⁻²		
(Mo, Fe)P <sub>2</sub> O <sub>7</sub> @NF	1 M KOH	250 mV	40.0 mV dec-1	6
		100 mA cm <sup>-2</sup>		
NiFeP <sub>x</sub> /NF	1 M KOH	258 mV	29 mV dec <sup>-1</sup>	7
		100 mA cm <sup>-2</sup>		
FeCoNi-LDH/NF	1.0 M KOH	230 mV	45.76 mV dec <sup>-1</sup>	8

	+ 0.5 M	100 mA cm <sup>-2</sup>		
	urea			
FeCoNiN/NF	1 M KOH	267 mV	60mV dec <sup>-1</sup>	9
		50 mA cm <sup>-2</sup>		
Ni <sub>3</sub> S <sub>2</sub> @MIL-	1 М КОН	236 mV	14.8 mV dec <sup>-1</sup>	10
53(NiFeCo)/NF		50 mA cm <sup>-2</sup>		
MOF CoFeP	1 М КОН	140 mV	40 mV dec <sup>-1</sup>	11
		50 mA cm <sup>-2</sup>		
NiFeP@NC/NF	1 M KOH	286 mV	69 mV dec <sup>-1</sup>	12
		20 mA cm <sup>-2</sup>		
Ru–MnFeP/NF	1 M KOH	191 mV	69 mV dec <sup>-1</sup>	13
		20 mA cm <sup>-2</sup>		
PBA-SMo /NF-10 h	1 M KOH	252 mV	64 mV dec <sup>-1</sup>	14
		20 mA cm <sup>-2</sup>		
CuFe-P/NF	1 M KOH	231 mV	63.0 mV dec <sup>-1</sup>	15
		10 mA cm <sup>-2</sup>		
Fe-CoP/NF	1 M KOH	190 mV	92 mV dec <sup>-1</sup>	16
		10 mA cm <sup>-2</sup>		
NixCo <sub>3-x</sub> O <sub>4</sub> /NF	1 M KOH	287 mV	88 mV dec <sup>-1</sup>	17
		10 mA cm <sup>-2</sup>		
CoFePBA@NiCoFe-	1 M KOH	228 mV	36 mV dec <sup>-1</sup>	18
LTH/NF		10 mA cm <sup>-2</sup>		
(NiCo)₂P/NF	1 M KOH	162 mV	135 mV dec <sup>-1</sup>	19
		10 mA cm <sup>-2</sup>		

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