

## Supporting information

# Hydrangea-like nanosheets of Co(OH)<sub>2</sub>@NiFe-LDH/NF as efficient electrocatalyst for oxygen evolution reactions

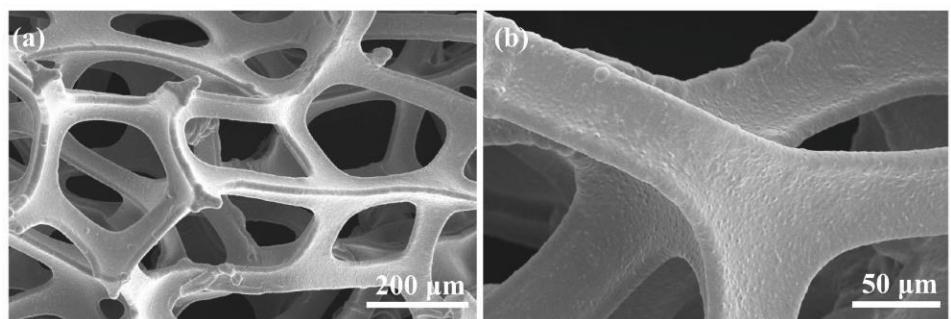
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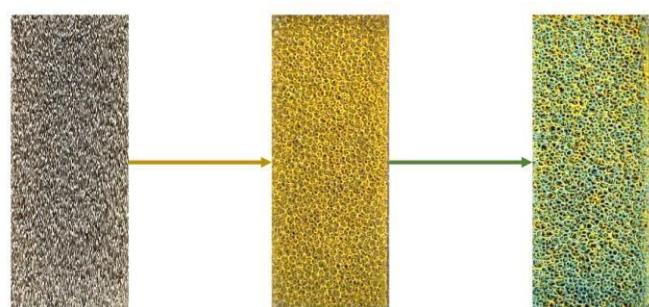
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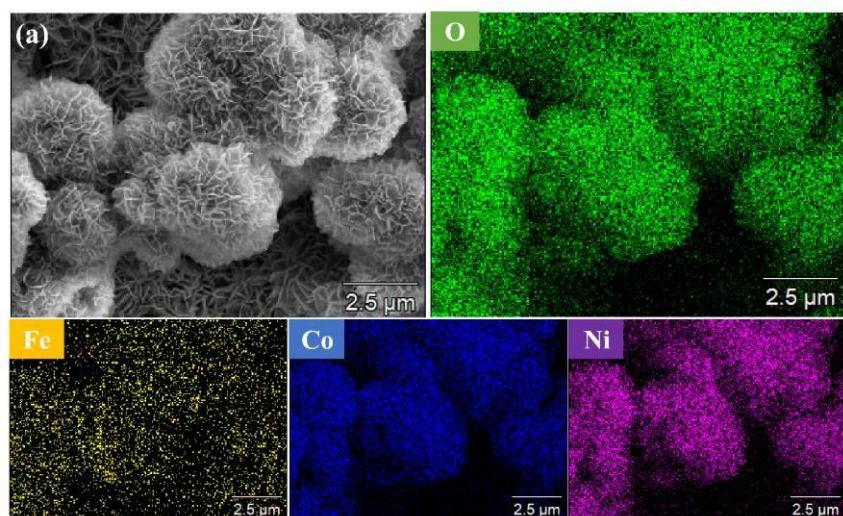
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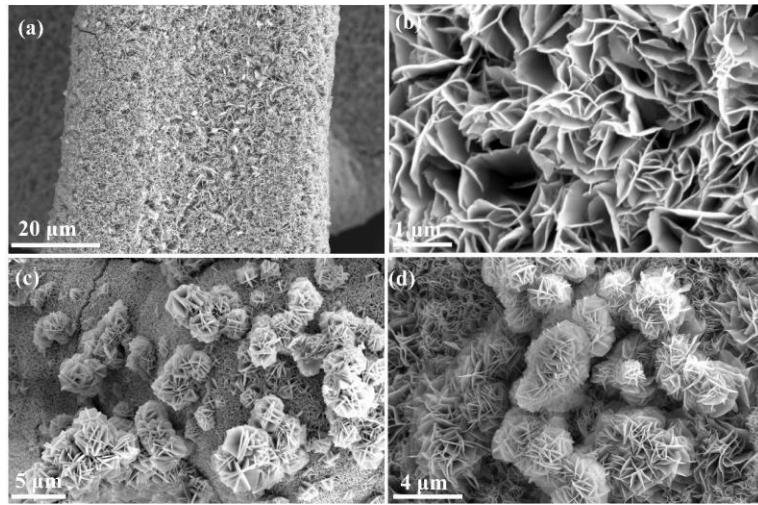
**Fig. S1** (a, b) SEM images of bare NF skeleton at low and high magnification.



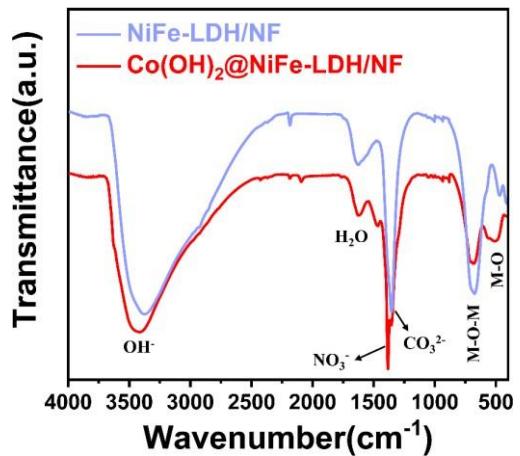
**Fig. S2** The photos of bare NF, NiFe-LDH/NF and Co(OH)<sub>2</sub>@NiFe-LDH/NF.



**Fig. S3** EDS mapping images of the of Co(OH)<sub>2</sub>@NiFe-LDH/NF.



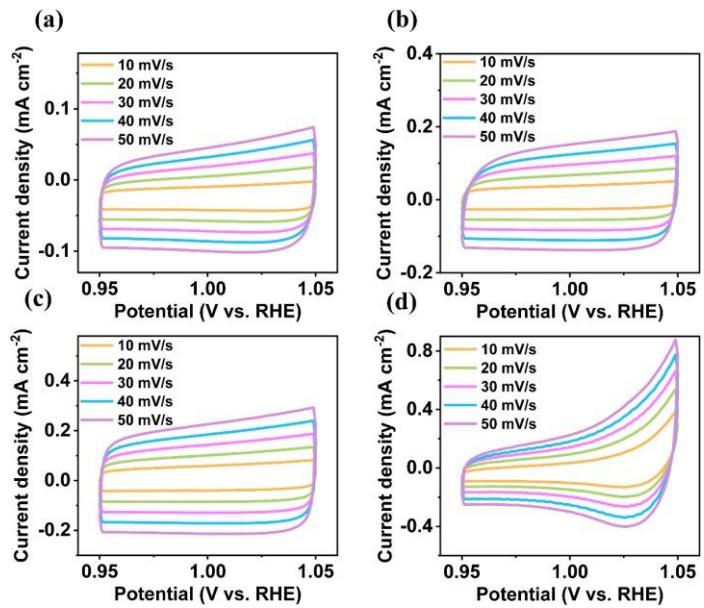
**Fig. S4** SEM images of (a, b)  $\text{Co}(\text{OH})_2/\text{NF}$  at low and high magnification, (c) NiFe-LDH/NF and (d)  $\text{Co}(\text{OH})_2@\text{NiFe-LDH}/\text{NF}$ .



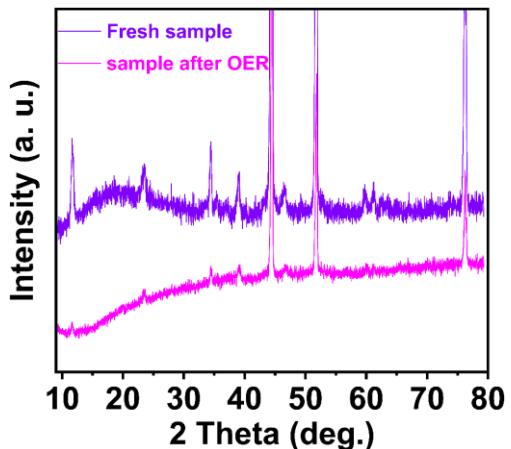
**Fig. S5** FT-IR spectra of NiFe-LDH/NF and  $\text{Co}(\text{OH})_2@\text{NiFe-LDH}/\text{NF}$ .

**Table. S1** The simulated series resistance ( $R_s$ ) and charge transfer resistance ( $R_{ct}$ ) values based on the fitting models.

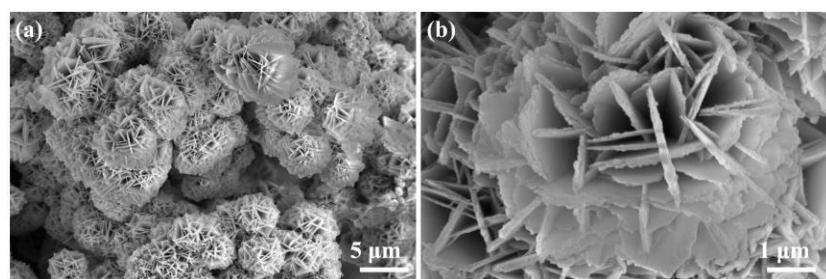
Samples	$R_s$ ( $\Omega$ )	$R_{ct}$ ( $\Omega$ )
bare NF	1.293	76.57
$\text{Co}(\text{OH})_2/\text{NF}$	1.367	11.13
NiFe-LDH/NF	1.413	1.08
$\text{Co}(\text{OH})_2@\text{NiFe-LDH}/\text{NF}$	1.325	0.63



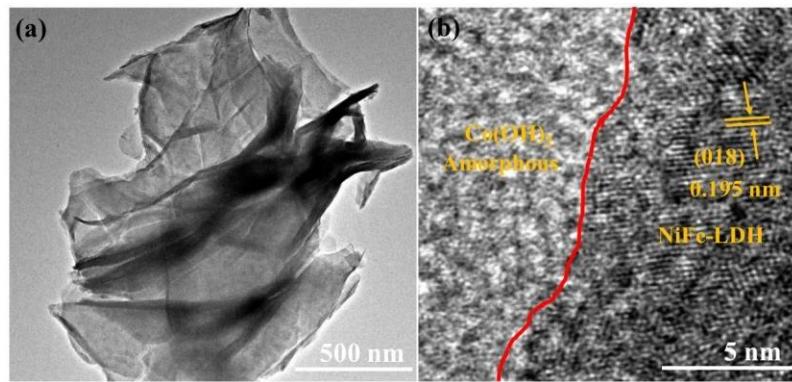
**Fig. S6** Cyclic voltammograms of (a) bare NF, (b) Co(OH)<sub>2</sub>/NF, (c) NiFe-LDH/NF and (d) Co(OH)<sub>2</sub>@NiFe-LDH/NF with different scan rates.



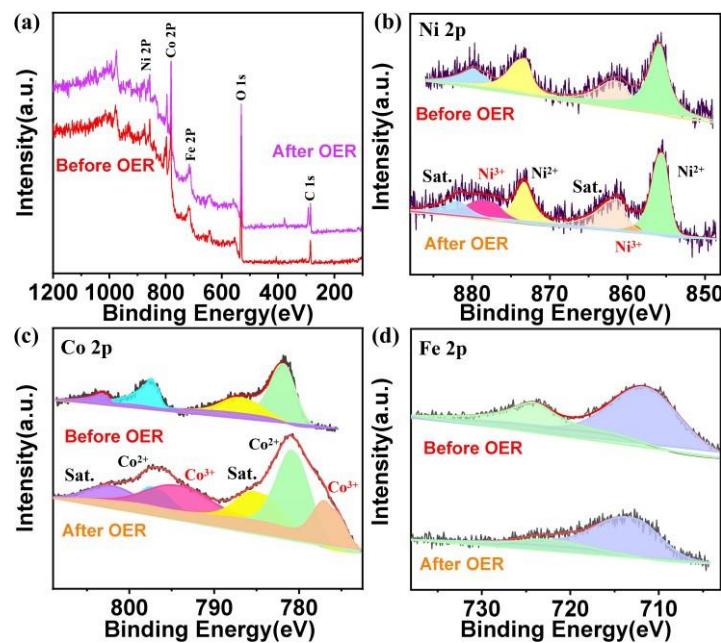
**Fig. S7** XRD patterns of Co(OH)<sub>2</sub>@NiFe-LDH/NF before and after OER testing.



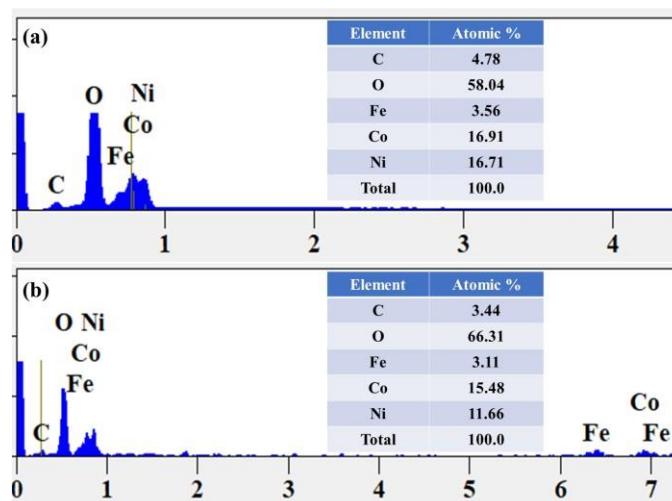
**Fig. S8** SEM images of after OER testing.



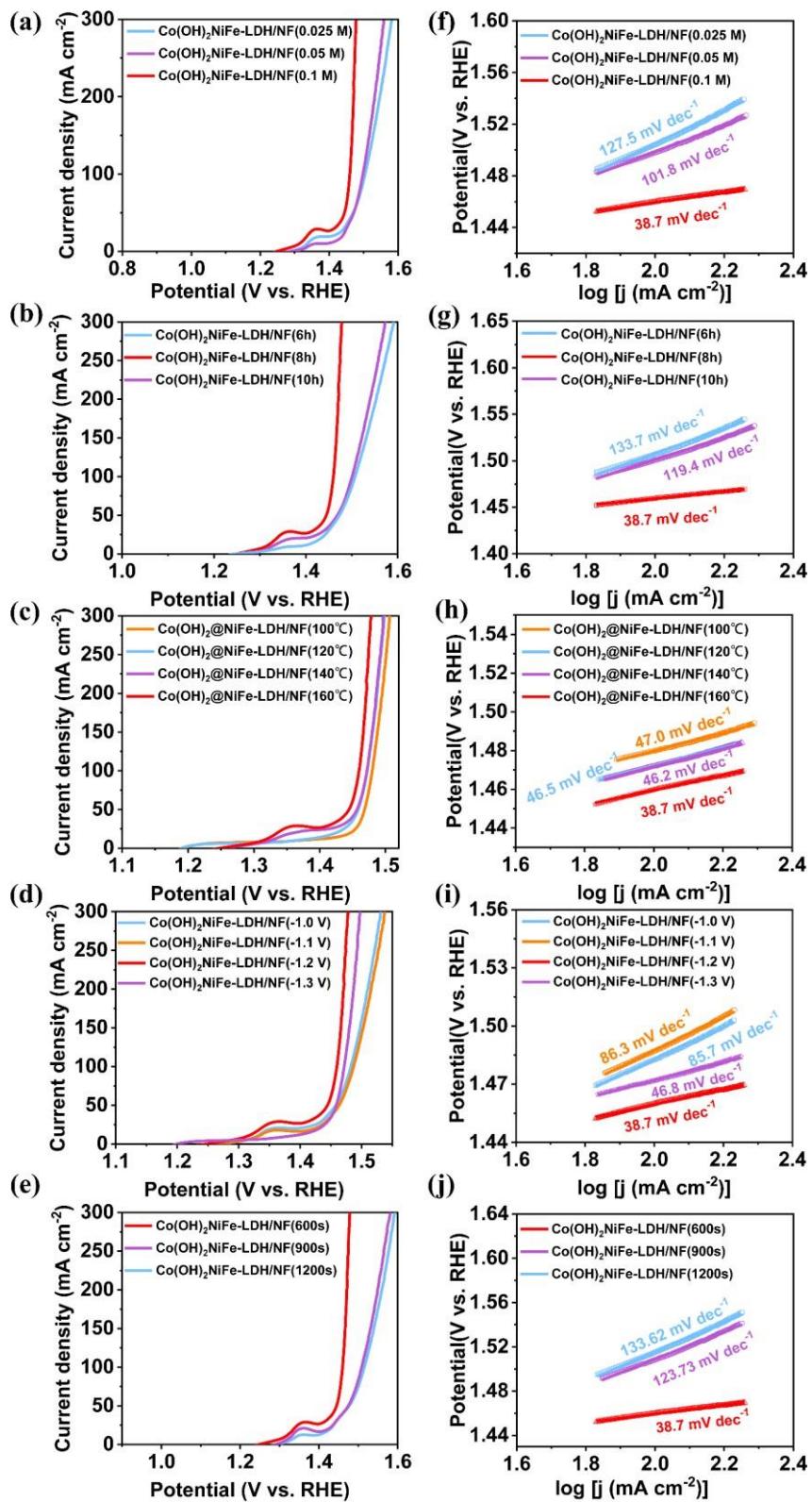
**Fig. S9** TEM images of  $\text{Co}(\text{OH})_2$ @NiFe-LDH/NF after OER testing.



**Fig. S10** XPS spectra of  $\text{Co}(\text{OH})_2$ @NiFe-LDH/NF before and after OER test.



**Fig. S11** EDS of  $\text{Co}(\text{OH})_2$ @NiFe-LDH/NF (a) before and (b) after OER test.



**Fig. S12** a-e (Left) and f-j (Right) Polarization curves and Tafel plots of different conditions.

**Table. S2** A comparison of the catalytic OER performance of recently reported catalysts in 1 M KOH solution.

Catalysts	Electrolyte	J / (mA m <sup>-2</sup> )	η <sub>j</sub> / (mV)	Ref
Co(OH) <sub>2</sub> @NiFe-LDH/NF	1M KOH	50 100	209 229	This work
NiSe@CoFe LDH/NF	1M KOH	100	236	1
Ni-Fe-W LDH/NF	1M KOH	100	247	2
NiFeCoP/NF	1M KOH	100	244.2	3
FeOOH/NiFe/NF	1M KOH	100	290	4
NiCoP@NiMn LDH/NF	1M KOH	100	293	5
NiFe-60/Co <sub>3</sub> O <sub>4</sub> @NF	1M KOH	50	221	6
cMOF/LDH	1M KOH	50	217	7
CoMoP/NiFe-LDH/NF	1M KOH	50	225	8
NiFe-LDH-Vo@NiCu	1M KOH	50	244	9
Nb-NiFe-LDH	1M KOH	50	242	10
NiFe-PO <sub>x</sub> /NF	1M KOH	50	247	11
Ni <sub>3</sub> S <sub>2</sub> -NiFe LDHs /NF	1M KOH	50	230	12
CoO@NiFe-LDH/NF	1M KOH	20	225	13
Ru/NiFe LDH-F/NF	1M KOH	10	230	14
NiFeMn-LDH	1M KOH	10	310	15
CoFeV LDH/NF	1M KOH	10	242	16
CoO-Co <sub>4</sub> N@NiFe-LDH	1M KOH	10	231	17

## References

1. F. Nie, Z. Li, X. Dai, X. Yin, Y. Gan, Z. Yang, B. Wu, Z. Ren, Y. Cao and W. Song, *Chem. Eng. J.*, 2022, **431**, 134080.
2. L. Wu, L. Yu, F. Zhang, D. Wang, D. Luo, S. Song, C. Yuan, A. Karim, S. Chen and Z. Ren, *J. Mater. Chem. A*, 2020, **8**, 8096-8103.
3. J. Cen, L. Wu, Y. Zeng, A. Ali, Y. Zhu and P. K. Shen, *ChemCatChem*, 2021, **13**, 4602-4609.
4. J. Chi, H. Yu, G. Jiang, J. Jia, B. Qin, B. Yi and Z. Shao, *J. Mater. Chem. A*, 2018, **6**, 3397-3401.
5. P. Wang, J. Qi, X. Chen, C. Li, W. Li, T. Wang and C. Liang, *ACS Appl. Mater. Interfaces*, 2019, **12**, 4385-4395.
6. J. Lv, L. W, R. Li, K. Zhang, D. Zhao, Y. Li, X. Li, X. Huang, and G. Wang, *ACS Catal.*, 2021, **11**, 14338-14351.
7. Y. Wang, L. Yan, K. Dastafkan, C. Zhao, X. Zhao, Y. Xue, J. Huo, S. Li and Q. Zhai, *Adv. Mater.*, 2021, **33**, e2006351.
8. W. Mai, Q. Cui, Z. Zhang, K. Zhang, G. Li, L. Tian and W. Hu, *ACS Appl. Energy Mater.*, 2020, **3**, 8075-8085.
9. H. Su, J. Jiang, N. Li, Y. Gao and L. Ge, *Chem. Eng. J.*, 2022, **446**, 137226.
10. Y. Zhou, F. Wang, S. Dou, Z. Shi, B. Dong, W. Yu, H. Zhao, F. Wang, J. Yu and Y. Chai, *Chem. Eng. J.*, 2022, **427**, 131643.
11. S. Song, J. Zang, S. Zhou, H. Gao, X. Tian, Y. Yuan, W. Li and Y. Wang, *Electrochim. Acta*, 2021, **392**, 138996.
12. S. Wu, S. Liu, X. Tan, W. Zhang, K. Cadien and Z. Li, *Chem. Eng. J.*, 2022, **442**, 136105.
13. Z. Wang, J. Zhang, Q. Yu, H. Yang, X. Chen, X. Yuan, K. Huang and X. Xiong, *Chem. Eng. J.*, 2021, **410**, 130123.
14. Y. Wang, P. Zheng, M. Li, Y. Li, X. Zhang, J. Chen, X. Fang, Y. Liu, X. Yuan, X. Dai and H. Wang, *Nanoscale*, 2020, **12**, 9669-9679.
15. Z. Lu, L. Qian, Y. Tian, Y. Li, X. Sun and X. Duan, *Chem. Commun. (Camb)*, 2016, **52**, 908-911.
16. Y. Hu, Z. Wang, W. Liu, L. Xu, M. Guan, Y. Huang, Y. Zhao, J. Bao and H. Li, *ACS Sustain. Chem. Eng.*, 2019, **7**, 16828-16834.
17. B. Chen, M. Humayun, Y. Li, H. Zhang, H. Sun, Y. Wu and C. Wang, *ACS Sustain. Chem. Eng.*, 2021, **9**, 14180-14192.